## Supporting Information

# 2-Guanidyl Pyridine PNA Nucleobase for Triple-Helical Hoogsteen Recognition of Cytosine in Double-Stranded RNA 

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## General Synthetic Procedures

Solvents and reagents were obtained from commercial suppliers and were used without further purification unless stated otherwise. Tetrahydrofuran (THF) and dichloromethane (DCM) were dried by passing over activated alumina. Pyridine and diisopropylethylamine (DIPEA) were dried by refluxing over $\mathrm{CaH}_{2}$ followed by distillation. Reactions were carried out under an atmosphere of nitrogen using a Schlenk line unless otherwise states. Analytical thin layer chromatography (TLC) used Silacycle 60 Å silica gel F254 plates ( 0.25 mm ) and visualization was aided by UV light, iodine, or ninhydrin stain. Column chromatography used Silacycle P60 230-400 mesh silica gel as the stationary phase. NMR spectra were obtained using Bruker AM 400 with the chemical shift ( $\delta$ ) reported in parts per million (ppm) relative to TMS or to the solvent peak [dimethyl sulfoxide (DMSO)-d6 or $\mathrm{CDCl}_{3}$ ] as a reference. $\mathrm{LC} / \mathrm{MS}$ analysis was performed using Shimadzu LCMS 2020 system with an ESI ionization source using a gradient of 0-100\% $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid) in 18 minutes. For compounds 1,3 , and 4 , a $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ without formic acid was used to avoid Boc-deprotection. The column used for all small molecules was a Cosmosil 2.5 C18-MS-II, $120 \AA$ Å, $2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.

## Synthesis of PNA monomers

5-Oxo-5-((6-((2,2,10,10-tetramethyl-4,8-dioxo-3,9-dioxa-5,7-diazaundecan-6-ylidene)amino)pyridin-2yl)amino)pentanoic acid (1). 2,6-Diaminopyridine ( $109 \mathrm{mg}, 1 \mathrm{mmol}$ ) was added to round bottom flask, purged with nitrogen gas, and dissolved in dry pyridine ( $2.5 \mathrm{~mL}, 0.4 \mathrm{M}$ ). Glutaric anhydride ( $110 \mathrm{mg}, 0.96$ $\mathrm{mmol})$ was added and the solution was heated at $50^{\circ} \mathrm{C}$ for 6 hours. The solution was cooled to room temperature, 1,3-di-Boc-2-trifluoromethylsulfonyl guanidine ( $391 \mathrm{mg}, 1 \mathrm{mmol}$ ) was added, the vessel
was purged with nitrogen, and heated at $45^{\circ} \mathrm{C}$ for 18 hours. Pyridine was partially removed by rotary evaporation and the residual oil was redissolved in dichloromethane ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(2.5 \mathrm{~mL})$ and deionized water $(5 \mathrm{~mL})$ to remove remaining pyridine. The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified via silica gel chromatography ( $0-8 \% \mathrm{MeOH} / \mathrm{DCM}$ ) to afford the title compound $\mathbf{1}$ as a white foam ( $197 \mathrm{mg}, 42 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.15$ in $5 \%$ MeOH/DCM. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 1.55(18 \mathrm{H}, \mathrm{s}), 2.08(2 \mathrm{H}, \mathrm{p}, \mathrm{J}=6.8 \mathrm{~Hz}), 2.53(2 \mathrm{H}, \mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz})$, $2.65(2 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}), 7.77(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=8.2 \mathrm{~Hz}), 8.05(2 \mathrm{H}, \mathrm{m}), 9.71(1 \mathrm{H}, \mathrm{s}) 10.6-11.6(2 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C}$ NMR (101 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ : 20.1, $28.1,32.4,35.9,110.3,112.4,141.4,148.4,150.2,153.0,171.8,188.7$. Mass spec (ESI) m/z: $[M+H]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{~N}_{5} \mathrm{O}_{7} 466.5$; found 466.2.

## Benzyl N-(2-(()(9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-((6-((2,2,10,10-tetramethyl-4,8-dioxo-3,9-dioxa-5,7-diazaundecan-6-ylidene)amino)pyridin-2-

yl)amino)pentanoyl)glycinate (3). 5-Oxo-5-((6-((2,2,10,10-tetramethyl-4,8-dioxo-3,9-dioxa-5,7-diazaundecan-6-ylidene)amino)pyridin-2-yl)amino) pentanoic acid $\mathbf{1}$ ( $250 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) and EDC• HCl ( $230 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) were added to a round bottom flask. The vessel was purged with nitrogen gas and charged with 1:1 mixture of dry dimethyl formamide (DMF)/pyridine ( $2.5 \mathrm{~mL}, 0.2 \mathrm{M}$ ). The reaction mixture was stirred at room temperature until solids dissolve ( $\sim 1$ hour), cooled to $0^{\circ} \mathrm{C}$, and PNA backbone $\mathbf{2}^{1}$ ( $230 \mathrm{mg}, 0.53 \mathrm{mmol}$ ) was added. The reaction mixture was allowed to warm up to room temperature and stirred for 6 hours. The reaction mixture was diluted with ethyl acetate and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$ and deionized water ( 5 mL ) to remove remaining pyridine. The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified via silica gel chromatography ( $50-75 \%$ ethyl acetate/hexanes) to afford the title compound $\mathbf{3}$ as a white foam ( 362 $\mathrm{mg}, 77 \%) . \mathrm{R}_{\mathrm{f}}=0.3$ in $60 \%$ ethyl acetate/hexanes. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 1.55(18 \mathrm{H}, \mathrm{s}), 2.04(2.3 \mathrm{H}$, m), $2.37(2.7 \mathrm{H}, \mathrm{m}), 2.48(1.1 \mathrm{H}, \mathrm{t}) 2.8-3.1(0.5 \mathrm{H}, \mathrm{bm}), 3.37(1.8 \mathrm{H}, \mathrm{m}), 3.55(1.7 \mathrm{H}, \mathrm{m}) 3.85(0.3 \mathrm{H}, \mathrm{bs}), 4.09-$ $4.20(2.8 \mathrm{H}, \mathrm{m}), 4.40(1.7 \mathrm{H}, \mathrm{d}), 4.62(0.2 \mathrm{H}, \mathrm{bs}), 5.16-5.22(2.0 \mathrm{H}, \mathrm{d}), 5.57(0.3 \mathrm{H}, \mathrm{t}), 5.83(0.5 \mathrm{H}, \mathrm{bs}) 7.27-7.41$ $(9.1 \mathrm{H}, \mathrm{m}), 7.57-7.60(2.0 \mathrm{H}, \mathrm{d}), 7.65(0.9 \mathrm{H}, \mathrm{q}, \mathrm{J}=8.3 \mathrm{~Hz}), 7.74(2.0 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}), 7.87(1.1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.9$ $\mathrm{Hz}), 7.94(0.7 \mathrm{H}, \mathrm{bs}), 8.10(1.0 \mathrm{H}, \mathrm{bs}), 10.6(0.9 \mathrm{H}, \mathrm{bs}), 11.5(0.9 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 20.4$, $20.7,24.7,28.1,31.3,31.7,36.1,36.3,36.7,39.4,47.2,47.3,47.9,48.9,49.2,50.7,66.7,66.9,67.3,67.6$, 80.1, 84.0, 109.5, 109.6, 111.5, 119.9, 120.0, 125.0, 125.1, 127.0, 127.1, 127.6, 127.7. 128.4, 128.5, $128.7,128.7,134.9,135.2,140.6,141.3,143.8,144.0,149.5,152.8,156.6,156.7,169.5,170.2,171.2$, 173.1, 173.6. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of $\mathbf{3}$ was further confirmed with LCMS (ESI) $\mathrm{m} / \mathrm{z}$ : $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{47} \mathrm{H}_{55} \mathrm{~N}_{7} \mathrm{O}_{10} 879.0$; found 878.3.

## N-(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-((6-((2,2,10,10-tetramethyl-4,8-

 dioxo-3,9-dioxa-5,7-diazaundecan-6-ylidene)amino)pyridin-2-yl)amino)pentanoyl)glycine (4). Prior to the reaction, dry THF was degassed by purging with nitrogen gas for 30 minutes. A round bottom flask was charged with benzyl N -(2-((( $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl}) m e t h o x y)$ carbonyl)amino)ethyl)-N-(5-oxo-5-((6-((2,2,10,10-tetramethyl-4,8-dioxo-3,9-dioxa-5,7-diazaundecan-6-ylidene)amino)pyridin-2yl)amino) pentanoyl)glycinate $3(362 \mathrm{mg}, 0.41 \mathrm{mmol})$ and $10 \% \mathrm{Pd} / \mathrm{C}(75 \mathrm{mg}, 20 \mathrm{wt} \%)$ and purged with nitrogen gas. Degassed THF ( $1.25 \mathrm{~mL}, 0.33 \mathrm{M}$ ) was added followed by purging the mixture with $\mathrm{H}_{2}$ gas using a balloon for 3 minutes. The reaction was stirred at high speed under an $\mathrm{H}_{2}$ atmosphere (balloon) for 3 hours. Once complete, the reaction mixture was diluted with methanol and filtered through a celite plug to remove the $\mathrm{Pd} / \mathrm{C}$. The solvent was removed via rotary evaporation to give pure 4 ( 316 mg , $95 \%)$ as a white foam. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.54(18 \mathrm{H}, \mathrm{m}), 1.98-2.04(2.3 \mathrm{H}, \mathrm{bs}), 2.37-2.45(3.8 \mathrm{H}$, m), 3.41-3.58 (3.4H, m), 3.87-3.96 (1.8H, m), 4.16-4.23 (1.0H, m), 4.33 (1.4H, d), 4.43 (0.2H, bs), 4.57 (0.2H, bs), $6.40(0.3 H, b s), 6.52(0.2 H, b s), 6.69(0.5 H, b s), 7.01(0.2 H, b s), 7.23-7.31(2.5 H, m), 7.32-7.39$ $(2.1 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}), 7.53-7.65(3.1 \mathrm{H}, \mathrm{m}), 7.73(2.0 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}), 7.85(1.0 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.8 \mathrm{~Hz}), 8.06(1 \mathrm{H}, \mathrm{d})$, 8.37-8.41 (0.7H, m), 8.53-8.64 (0.3H, m) 10.56 (0.8H, bs), $11.53(0.8 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, CDCl $\left.{ }_{3}\right) \delta$ : $20.6,22.4,22.6,28.1,31.6,32.1,36.1,36.6,39.2,39.7,44.6,47.2,47.2,49.7,51.1,53.4,66.7,66.9,80.1$, $83.9,109.7,111.4,119.9,120.0,125.1,127.0,127.7,140.5,141.2,143.9,144.1,148.6,149.7,152.7$, 156.7, 156.8, $171.4,171.9,173.5,174.1,175.0$. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 4 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{40} \mathrm{H}_{49} \mathrm{~N}_{7} \mathrm{O}_{10} 788.8$; found 788.3.

Tert-butyl (7-chloro-1,8-naphthyridin-2-yl)carbamate (6). In an oven dried round bottom flask, 7-chloro-1,8-naphthyridin-2-amine ${ }^{1} 5(250 \mathrm{mg}, 1.39 \mathrm{mmol})$ was dissolved in THF ( $4.2 \mathrm{~mL}, 0.33 \mathrm{M}$ ). The flask was purged with nitrogen gas and cooled to $-78{ }^{\circ} \mathrm{C}$ in a dry ice/acetone bath. 2 M NaHMDS in THF ( $1.39 \mathrm{~mL}, 2 \mathrm{eq}$ ) was added. After 5 minutes of stirring, a solution of $\mathrm{Boc}_{2} \mathrm{O}(303 \mathrm{mg}, 1 \mathrm{eq}) \mathrm{in} \mathrm{THF}(1.0 \mathrm{~mL})$

[^0]was added and the reaction mixture was stirred for 18 hours. The solution was partially concentrated and diluted with DCM. The organic layer was then washed with $5 \%$ aqueous $\mathrm{AcOH}(3 \mathrm{~mL})$, saturated aqueous $\mathrm{NaHCO}_{3}(3 \mathrm{~mL})$, water $(3 \mathrm{~mL})$, and brine $(3 \mathrm{~mL})$. The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting solid was purified via silica gel chromatography (10-50\% ethyl acetate/hexanes) to afford the title compound 6 as a white solid ( $197 \mathrm{mg}, 65 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.25,20 \%$ ethyl acetate/hexanes. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.58(9 \mathrm{H}, \mathrm{s}), 7.38(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.3 \mathrm{~Hz}), 7.62(1 \mathrm{H}, \mathrm{s}), 8.06(1 \mathrm{H}$, $\mathrm{d}, \mathrm{J}=8.4 \mathrm{~Hz}), 8.15(2 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.9 \mathrm{~Hz}), 8.37(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 28.2,82.2$, $113.9,118.6,121.6,138.7,152.0,154.2,154.8$. Mass spec (ESI) $\mathrm{m} / \mathrm{z}$ : $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{CIN}_{3} \mathrm{O}_{2} 280.1$ and 282.1; found 280.1 and 282.1.

Tert-butyl (7-azido-1,8-naphthyridin-2-yl)carbamate (7). Tert-butyl (7-chloro-1,8-naphthyridin-2$\mathrm{yl})$ carbamate $6(264 \mathrm{mg}, 0.94 \mathrm{mmol})$ and sodium azide ( $67 \mathrm{mg}, 1.03 \mathrm{mmol}$ ) were combined in a round bottom flask. DMF ( $2.3 \mathrm{~mL}, 0.4 \mathrm{M}$ ) was added to the flask and the reaction mixture was heated to $60^{\circ} \mathrm{C}$ for 18 hours. After 18 hours, solvent was removed by rotary evaporation and the resulting oil was purified by silica gel column chromatography (20-50\% ethyl acetate/hexanes) to afford the tittle compound 7 as a pale yellow solid ( $235 \mathrm{mg}, 87 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.12$ in $20 \%$ ethyl acetate/hexanes. ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.54(9 \mathrm{H}, \mathrm{s}), 7.88(2 \mathrm{H}, \mathrm{q}, \mathrm{J}=16.7,9.2 \mathrm{~Hz}), 8.24(1 \mathrm{H}, \mathrm{s}), 8.31(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.9 \mathrm{~Hz}), 8.46(1 \mathrm{H}, \mathrm{d}$, $\mathrm{J}=8.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, CDCl 3 ) $\delta: 28.1,82.5,111.3,113.4,114.5,132.0,139.9,140.6,149.4$, 151.7, 153.5. Mass spec (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{6} \mathrm{O}_{2}$ 287.3; found 287.2.

Tert-butyl (7-amino-1,8-naphthyridin-2-yl)carbamate (8). Tert-butyl (7-azido-1,8-naphthyridin-2$\mathrm{yl})$ carbamate $7(223 \mathrm{mg}, 0.78 \mathrm{mmol})$ and tri-n-butylphosphonium tetrafluoroborate ( $253 \mathrm{mg}, 0.87$ mmol ) were combined in a round bottom flask. The solids were dissolved in 9:1 mixture of THF/water (3 mL 0.25 M ) followed by the addition of triethylamine ( $121 \mu \mathrm{~L}, 0.87 \mathrm{mmols}$ ). The reaction was heated to $65^{\circ} \mathrm{C}$ for 6 hours. The reaction was diluted with methanol, filtered through a 3 cm celite plug, and concentrated. The compound was purified by silica gel chromatography ( $2-6 \% \mathrm{MeOH} / \mathrm{DCM}$, containing $0.1 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to afford the tittle compound 8 as a white solid ( $198 \mathrm{mg}, 97 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.35$ in $5 \% \mathrm{MeOH} / \mathrm{DCM}$. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.59(9 \mathrm{H}, \mathrm{s}), 6.82(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.7 \mathrm{~Hz}), 7.81(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.8 \mathrm{~Hz}), 7.93(2 \mathrm{H}, \mathrm{m})$, $8.45(1 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 28.2,82.0,109.6,111.4,113.7,138.5,138.8,152.6,153.6$, 154.0, 159.3. Mass spec (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2}$ 261.3; found 261.2



Benzyl N-(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(4-((7-((tert-butoxycarbonyl)amino)-1,8-naphthyridin-2-yl)amino)-4-oxobutanoyl)glycinate (9). Benzyl (2-(()(9H-fluoren-9-
yl)methoxy)carbonyl)amino)ethyl)glycinate $\mathbf{2}$ ( $233 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) and succinic anhydride ( $55 \mathrm{mg}, 0.55$ mmol ) were combined in a round bottom flask. The flask was purged with nitrogen gas and cooled to 0 ${ }^{\circ} \mathrm{C}$. DMF ( $2 \mathrm{~mL}, 0.25 \mathrm{M}$ ) added followed by slow addition of distilled triethylamine ( $153 \mu \mathrm{~L}, 1.1 \mathrm{mmol}$ ). The reaction mixture was stirred for 18 hours. HBTU ( $180 \mathrm{mg}, 0.47 \mathrm{mmol}$ ) was added, and solution was stirred for 30 minutes. The reaction was cooled to $0{ }^{\circ} \mathrm{C}$ and tert-butyl (7-amino-1,8-naphthyridin-2$\mathrm{yl})$ carbamate 8 was added followed by slow addition of diisopropylethylamine ( $187 \mu \mathrm{~L}, 1.1 \mathrm{mmol}$ ). The reaction was warmed to room temperature and stirred for 18 hours. The reaction was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water ( 5 mL ), and brine ( 5 mL ). The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography (50-100\% ethyl acetate/hexanes then $1 \% \mathrm{MeOH} /$ ethyl acetate) to afford the title compound 9 as a white solid ( $134 \mathrm{mg}, 35 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.2$ in $70 \%$ ethyl acetate/hexanes. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right)$ ס: 1.39-1.45 (4.7H, s), 1.55-1.61 (6.5H, s), 2.63-2.73 (1.2H, m), $2.83(1.6 \mathrm{H}, \mathrm{s}), 2.91(0.1 \mathrm{H}, \mathrm{s}), 2.98$ $(0.1 \mathrm{H}, \mathrm{s}), 3.07(1.1 \mathrm{H}, \mathrm{m}), 3.40(0.9 \mathrm{H}, \mathrm{m}), 3.45(0.8 \mathrm{H}, \mathrm{q}), 3.51(0.2 \mathrm{H}, \mathrm{d}, \mathrm{J}=5.4 \mathrm{~Hz}), 3.63(1.4 \mathrm{H}, \mathrm{m}), 3.73$ $(0.7 \mathrm{H}, \mathrm{t}), 3.97(0.4 \mathrm{H}, \mathrm{s}), 4.14(0.7 \mathrm{H}, \mathrm{s}), 4.22-4.29(1.1 \mathrm{H}, \mathrm{m}), 4.34(0.6 \mathrm{H}, \mathrm{s}), 4.42(0.7 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.1 \mathrm{~Hz}), 4.48$ ( $0.7 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.5 \mathrm{~Hz}$ ), $4.66(0.4 \mathrm{H}, \mathrm{d}, \mathrm{J}=5.5 \mathrm{~Hz}), 5.21(2.0 \mathrm{H}, \mathrm{m}), 7.18(0.8 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}), 7.25-7.38(7.2 \mathrm{H}$, $m), 7.55(0.7 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.0 \mathrm{~Hz}), 7.60(1.2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.4 \mathrm{~Hz}), 7.70-7.79(2.2 \mathrm{H}, \mathrm{m}), 7.91-8.06(1.9 \mathrm{H}, \mathrm{m}), 8.18-8.30$ $(1.3 \mathrm{H}, \mathrm{m}), 8.39(0.5 \mathrm{H}, \mathrm{t}, \mathrm{J}=8.9 \mathrm{~Hz}), 8.77(0.4 \mathrm{H}, \mathrm{bs}), 9.71(0.3 \mathrm{H}, \mathrm{bs}), 10.05(0.1 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 28.1,28.2,32.2,32.3,47.2,47.3,49.1,49.3,66.6,67.3,67.6,81.7,112.0,112.2,113.1,120.0$, $125.1,125.3,126.9,127.1,127.6,127.7,128.5,128.7,134.9,138.5,138.7,141.3,143.9,152.3,154.4$, $154.7,156.8,169.2,173.3$. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 9 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{43} \mathrm{H}_{44} \mathrm{~N}_{6} \mathrm{O}_{8} 773.9$; found 773.4.

N-(2-((( $9 \mathrm{H}-$ fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(4-((7-((tert-butoxycarbonyl)amino)-1,8-naphthyridin-2-yl)amino)-4-oxobutanoyl)glycine (10). THF was degassed via bubbling nitrogen for 1 hour prior to reaction. Benzyl N-(2-((( $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)-N-(4-((7-((tert-butoxycarbonyl)amino)-1,8-naphthyridin-2-yl)amino)-4-oxobutanoyl)glycinate 9 ( $130 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) and $10 \% \mathrm{Pd} / \mathrm{C}(24 \mathrm{mg}, 20 \mathrm{wt} \%$ ) were combined in a round bottom flask. The flask was purged with nitrogen gas and THF ( $1 \mathrm{~mL}, 0.17 \mathrm{M}$ ) was added via syringe. The flask was purged with hydrogen gas and the reaction mixture was stirred under a hydrogen atmosphere (balloon) for 4 hours. The reaction mixture was diluted with methanol ( 1 mL ) and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / \mathrm{C}$. After filtration, solvent was removed by rotary evaporation giving pure PNA monomer 10 as a white foam (112 mg, 97\%). ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 1.45-1.52(9 \mathrm{H}, \mathrm{s}), 2.29(0.3 \mathrm{H}, \mathrm{s}), 2.50(0.2 \mathrm{H}, \mathrm{bs}), 2.70-2.96$ (2.7H, m), 2.99-3.13 (0.6H, bs), 3.21-3.39 (0.7H, s), 3.47 (1.1H, bs), $3.60(1.0 \mathrm{H}, \mathrm{bs}), 3.81-4.34(3.2 \mathrm{H}, \mathrm{m})$, $4.58(0.4 \mathrm{H}, \mathrm{m}), 6.50-6.91(0.5 \mathrm{H}, \mathrm{bs}), 7.00(0.2 \mathrm{H}, \mathrm{s}), 7.19(1.4 \mathrm{H}, \mathrm{m}), 7.26-7.37(1.14 \mathrm{H}, \mathrm{m}), 7.46-7.57(1.6 \mathrm{H}$, $\mathrm{m}), 7.61-7.84(3.2 \mathrm{H}, \mathrm{m}), 8.02-8.22(1.6 \mathrm{H}, \mathrm{m}), 10.56(0.5 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 28.2,30.3$, $31.9,32.4,39.2,39.4,47.1,48.5,49.6,50.7,53.4,56.0,66.7,67.0,81.6,112.1,113.2,116.5,119.8$, $119.8,125.1,125.2,125.5,127.0,127.6,128.3,135.8,138.4,141.1,141.3,143.9,151.5,152.4,154.1$, 154.6, 156.8, 172.6, 173.5, 173.7, 174.7. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 3 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{36} \mathrm{H}_{38} \mathrm{~N}_{6} \mathrm{O}_{8}$ 683.7; found 683.3.



Benzyl N-(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-((7-((tert-butoxycarbonyl)amino)-1,8-naphthyridin-2-yl)amino)-5-oxopentanoyl)glycinate (11). Benzyl (2-(()(9H-fluoren-9-
yl)methoxy)carbonyl)amino)ethyl)glycinate $2(230 \mathrm{mg}, 0.49 \mathrm{mmol})$ and glutaric anhydride ( $62 \mathrm{mg}, 0.58$ $\mathrm{mmol})$ were combined in a round bottom flask. The flask was purged with nitrogen gas and cooled to $0^{\circ}$ C. DMF ( $2 \mathrm{~mL}, 0.25 \mathrm{M}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $83 \mu \mathrm{~L}$,
$0.48 \mathrm{mmol})$. The reaction was stirred for 18 hours. HATU ( $220 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was added, and the solution was stirred for 30 minutes. The reaction was cooled to $0{ }^{\circ} \mathrm{C}$ and tert-butyl (7-amino-1,8-naphthyridin-2-yl)carbamate 8 was added followed by slow addition of diisopropylethylamine ( $187 \mu \mathrm{~L}$, $0.96 \mathrm{mmol})$. The reaction was warmed to room temperature and stirred 18 hours. The reaction was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water ( 5 mL ), and brine (5 mL ). The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography (50-100\% ethyl acetate/hexanes then $1 \% \mathrm{MeOH} / \mathrm{ethyl}$ acetate) to afford the title compound 11 as a white solid ( $135 \mathrm{mg}, 36 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.2$ in $70 \%$ ethyl acetate/hexanes. ${ }^{1} \mathrm{H}$ NMR (400 MHz, CDCl $)_{3}$ ) $: 1.53(9 \mathrm{H}, \mathrm{m}), 2.08(2.1, \mathrm{~m}), 2.25-2.39(1.0 \mathrm{H}, \mathrm{m}), 2.40-2.58(2.7 \mathrm{H}, \mathrm{m}), 2.65$ $(0.6 \mathrm{H}, \mathrm{t}), 3.04(0.3 \mathrm{H}, \mathrm{bd}), 3.26-3.53(2.2 \mathrm{H}, \mathrm{m}), 3.55-3.65(1.5 \mathrm{H}, \mathrm{m}), 3.80(0.2 \mathrm{H}, \mathrm{bs}), 3.89(0.2 \mathrm{H}, \mathrm{bs}), 4.11$ $(1.1 \mathrm{H}, \mathrm{s}), 4.13-4.27(1.6 \mathrm{H}, \mathrm{m}), 4.40(1.5 \mathrm{H}, \mathrm{t}), 4.67(0.3 \mathrm{H}, \mathrm{m}), 5.05(0.4 \mathrm{H}, \mathrm{s}), 5.19(2.1 \mathrm{H}, \mathrm{m}), 7.28-7.42$ $(8.5 \mathrm{H}, \mathrm{m}), 7.58(2.5 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.4 \mathrm{~Hz}), 7.73(1.9 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.4 \mathrm{~Hz}), 7.84-7.93(0.3 \mathrm{H}, \mathrm{m}), 7.96-8.10(2.0 \mathrm{H}, \mathrm{m})$, 8.17-8.39 ( $2 \mathrm{H}, \mathrm{m}$ ), $8.59(0.5 \mathrm{H}, \mathrm{bs}), 8.84(0.2 \mathrm{H}, \mathrm{bs}), 8.90(0.2 \mathrm{H}, \mathrm{bs}), 9.04(0.1 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 20.7,28.2,31.2,32.6,36.5,39.4,39.6,47.2,48.0,49.0,49.2,50.9,66.7,66.9,67.3,67.6,79.2$, 81.8, 112.1, 113.1, 117.6, 119.9, 120.0, 125.0, 125.2, 127.0, 127.6, 127.7, 128.4, 128.5, 128.6, 128.7, $128.7,129.0,134.9,135.2,138.6,138.8,141.3,143.8,143.9,152.2,153.6,153.8,153.8,153.9,156.6$, 170.2, 171.9, 173.1, 173.7. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 11 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{44} \mathrm{H}_{46} \mathrm{~N}_{6} \mathrm{O}_{8} 787.9$; found 787.5.

## N-(2-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-((7-((tert-butoxycarbonyl)amino)-1,8-

 naphthyridin-2-yl)amino)-5-oxopentanoyl)glycine (12). THF was degassed via bubbling nitrogen for 1 hour prior to reaction. Benzyl N-(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-((7-((tert-butoxycarbonyl)amino)-1,8-naphthyridin-2-yl)amino)-5-oxopentanoyl)glycinate 11 ( $135 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) and $10 \% \mathrm{Pd} / \mathrm{C}$ ( $25 \mathrm{mg}, 20 \mathrm{wt} \%$ ) were combined in a round bottom flask. The flask was purged with nitrogen gas then THF ( $0.5 \mathrm{~mL}, 0.34 \mathrm{M}$ ) was added via syringe. The flask was purged with hydrogen gas, and the reaction mixture was stirred under a hydrogen atmosphere (balloon) for 4 hours. The reaction mixture was diluted with methanol $(0.5 \mathrm{~mL})$ and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / \mathrm{C}$. After filtration, solvent was removed by rotary evaporation giving pure PNA monomer 12 as a white foam (118 mg, 97\%). ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 1.43-1.57(9 \mathrm{H}, \mathrm{m}), 1.84-2.24(2.6 \mathrm{H}, \mathrm{m}), 2.29(0.4 \mathrm{H}, \mathrm{s})$, 2.42-2.72 (3.5H, m), $3.04(0.5 \mathrm{H}, \mathrm{d}), 3.21(0.4 \mathrm{H}, \mathrm{bs}), 3.36-3.67(3.0 \mathrm{H}, \mathrm{m}), 3.92-4.25(2.9 \mathrm{H}, \mathrm{m}), 4.31(1.3 \mathrm{H}$, m), 4.60 (0.5H, d), $6.23(0.5 \mathrm{H}, \mathrm{bd}), 6.67(0.3 \mathrm{H}, \mathrm{bd}), 7.00(0.2 \mathrm{H}, \mathrm{s}), 7.21(1.2 \mathrm{H}, \mathrm{t}), 7.26-7.41(4.6 \mathrm{H}, \mathrm{m}), 7.50$$(1.7 \mathrm{H}, \mathrm{d}), 7.59-7.77(2.4 \mathrm{H}, \mathrm{m}), 7.78-7.88(0.8 \mathrm{H}, \mathrm{m}), 7.89-7.98(0.9 \mathrm{H}, \mathrm{m}), 8.04(0.3 \mathrm{H}, \mathrm{bs}), 8.16(1.0 \mathrm{H}, \mathrm{t})$, 8.27 ( $0.8 \mathrm{H}, \mathrm{bs}$ ), 10.39 ( $0.7 \mathrm{H}, \mathrm{bs}$ ). ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 20.5,28.2,30.3,31.2,34.2,36.3,39.6$, $47.2,48.6,49.5,50.0,52.0,53.4,66.7,81.8,82.0,112.1,112.3,113.4,113.6,116.8,119.8,119.9,125.0$, $125.1,125.5,126.9,127.5,127.6,135.8,138.9,139.1,141.1,143.8,143.9,152.3,154.6,154.8,156.6$, 156.8, 174.1. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 12 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{37} \mathrm{H}_{40} \mathrm{~N}_{6} \mathrm{O}_{8}$ 697.8; found 697.5.


## Benzyl N-(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(4-oxo-4-(pyridin-2-ylamino)

butanoyl)glycinate (13). Benzyl (2-((() $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)glycinate 2 (230 $\mathrm{mg}, 0.49 \mathrm{mmol}$ ) and succinic anhydride ( $62 \mathrm{mg}, 0.62 \mathrm{mmol}$ ) were combined in a round bottom flask. The flask was purged with nitrogen gas and cooled to $0^{\circ} \mathrm{C}$. DMF ( $2 \mathrm{~mL}, 0.25 \mathrm{M}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $104 \mu \mathrm{~L}, 0.6 \mathrm{mmol}$ ). The reaction mixture was stirred for 18 hours. HBTU ( $274 \mathrm{mg}, 0.72 \mathrm{mmol}$ ) was added and the reaction mixture was stirred for 30 minutes. The reaction mixture was cooled to $0^{\circ} \mathrm{C}$ and 2 -aminopyridine ( $68 \mathrm{mg}, 0.72 \mathrm{mmol}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $146 \mu \mathrm{~L}, 0.86 \mathrm{mmol}$ ). The reaction mixture was warmed to room temperature and stirred 18 hours. The reaction was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water ( 5 mL ), and brine ( 5 mL ). The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography (50-100\% ethyl acetate/hexanes then $1 \% \mathrm{MeOH} /$ ethyl acetate) to afford the title compound 13 as a white solid ( $147 \mathrm{mg}, 51 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.55$ in ethyl acetate. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 2.60(0.8 \mathrm{H}, \mathrm{t}), 2.77$ $(3.8 \mathrm{H}, \mathrm{m}), 3.07(0.3 \mathrm{H}, \mathrm{bs}), 3.26(0.3 \mathrm{H}, \mathrm{bs}), 3.38(1.7 \mathrm{H}, \mathrm{m}), 3.57(1.6 \mathrm{H}, \mathrm{bs}), 3.87(0.2 \mathrm{H}, \mathrm{bs}), 4.14(2.5 \mathrm{H}, \mathrm{m})$, $4.32(0.6 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.2 \mathrm{~Hz}), 4.39(0.8 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.0 \mathrm{~Hz}), 4.62(0.3 \mathrm{H}, \mathrm{m}), 5.16(2.0 \mathrm{H}, \mathrm{m}), 5.84(0.3 \mathrm{H}, \mathrm{t}), 6.01$ $(0.4 \mathrm{H}, \mathrm{t}), 6.95(0.8 \mathrm{H}, \mathrm{t}, \mathrm{J}=6.0 \mathrm{~Hz}), 7.23-7.40(8.5 \mathrm{H}, \mathrm{m}), 7.57(2.7 \mathrm{H}, \mathrm{m}), 7.71(1.8 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.6 \mathrm{~Hz}), 8.13$
$(0.9 \mathrm{H}, \mathrm{t}, \mathrm{J}=8.4 \mathrm{~Hz}), 8.23(0.9 \mathrm{H}, \mathrm{m}), 8.95(0.6 \mathrm{H}, \mathrm{m}), 9.12(0.3 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 27.7,28.1$, $28.4,32.1,32.3,38.6,39.1,39.4,47.2,48.1,49.1,49.2,50.7,53.5,66.7,67.0,67.2,67.7,114.1,119.5$, $119.9,120.0,125.1,127.1,127.7,128.3,128.5,128.7,128.7,134.9,135.2,138.2,141.3,143.9,144.0$, $147.7,147.7,151.5,151.6,156.7156 .8,169.3,170.1,170.9,171.0,172.7,173.3$. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 13 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{35} \mathrm{H}_{34} \mathrm{~N}_{4} \mathrm{O}_{6} 607.7$; found 607.3.

## N-(2-((() $9 \mathrm{H}-$ fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(4-oxo-4-(pyridin-2-ylamino)butanoyl)

 glycine (14). Methanol was degassed via bubbling nitrogen gas for 1 hour prior to reaction. Meanwhile, $10 \% \mathrm{Pd} / \mathrm{C}(19 \mathrm{mg}, 40 \mathrm{wt} \%)$ and benzyl N -(2-((( $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)-N-(4-oxo-4-(pyridin-2-ylamino)butanoyl)glycinate $13(38 \mathrm{mg}, 0.06 \mathrm{mmol})$ were combined in a round bottom flask. The flask was purged with nitrogen gas and degassed methanol ( $250 \mu \mathrm{~L}, 0.24 \mathrm{M}$ ) was added. The reaction mixture was purged with hydrogen gas and stirred under a hydrogen atmosphere (balloon) for 1.5 hours. The reaction mixture was diluted with methanol $(250 \mu \mathrm{~L})$ and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / \mathrm{C}$. After filtration, the compound was adsorbed on celite and filtered through a 3 cm silica plug using DCM to $50 \% \mathrm{MeOH} / \mathrm{DCM}$ as eluent. The solvent was then removed, solids were redissolved in DCM, and filtered through 5 cm of celite in a pipette to remove residual silica. Solvent was removed via rotary evaporation affording the tittle compound 14 as a pale yellow waxy solid ( 24 mg , $77 \%) . \mathrm{R}_{\mathrm{f}}=0.3$ in $20 \% \mathrm{MeOH} / \mathrm{DCM} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 2.38-2.83(3.7 \mathrm{H}, \mathrm{m}), 3.00(0.4 \mathrm{H}, \mathrm{m}), 3.20$ $(0.2 H, b s), 3.35(1.7 \mathrm{H}, \mathrm{m}), 3.50(1.2 \mathrm{H}, \mathrm{m}), 3.80(0.3 \mathrm{H}, \mathrm{bs}), 3.88-4.21(2.2 \mathrm{H}, \mathrm{m}), 4.31(1.2 \mathrm{H}, \mathrm{m}), 4.54(0.1 \mathrm{H}$, d), $4.99(0.2 \mathrm{H}, \mathrm{s}), 5.90(0.1 \mathrm{H}, \mathrm{m}), 6.05(0.1 \mathrm{H}, \mathrm{bs}), 6.19(0.4 \mathrm{H}, \mathrm{m}), 6.97(1.0 \mathrm{H}, \mathrm{m}), 7.23(0.9 \mathrm{H}, \mathrm{m}), 7.33$ $(2.8 \mathrm{H}, \mathrm{m}), 7.54(1.8 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}), 7.69(2.8 \mathrm{H}, \mathrm{m}), 8.05(0.9 \mathrm{H}, \mathrm{m}), 8.21(0.9 \mathrm{H}, \mathrm{d}, \mathrm{J}=8.4 \mathrm{~Hz}), 10.41(0.7 \mathrm{H}$, m). ${ }^{13} \mathrm{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 30.3,32.0,32.2,36.7,39.5,47.2,49.4,49.8,66.8,67.0,115.1,119.3$, $119.9,125.1,127.0,127.6,127.7,128.6,129.0,139.9,141.2,143.9,145.2,151.3,156.7,156.8,171.9$, 172.0, 172.7, 173.5, 175.0. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 14 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{6} 517.6$; found 517.3.

## Benzyl N-(2-((( $(9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-2-ylamino)

pentanoyl)glycinate (15). Benzyl (2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)glycinate $\mathbf{2}$ (230 $\mathrm{mg}, 0.49 \mathrm{mmol}$ ) and glutaric anhydride ( $61 \mathrm{mg}, 0.53 \mathrm{mmol}$ ) were combined in a round bottom flask. The flask was purged with nitrogen gas and cooled to $0^{\circ} \mathrm{C}$. DMF ( $2 \mathrm{~mL}, 0.25 \mathrm{M}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $80 \mu \mathrm{~L}, 0.46 \mathrm{mmol}$ ). The reaction mixture was stirred for 18 hours. HBTU ( $219 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was added and the reaction mixture was stirred for 30 minutes. The reaction was cooled to $0^{\circ} \mathrm{C}$ and 2 -aminopyridine ( $54 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $125 \mu \mathrm{~L}, 0.72 \mathrm{mmol}$ ). The reaction was stirred 18 hours allowing for passive warming to room temperature. The reaction was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water ( 5 mL ), and brine ( 5 mL ). The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography ( $50-100 \%$ ethyl acetate/hexanes then $1 \% \mathrm{MeOH} /$ ethyl acetate) to afford the title compound 15 as a clear waxy solid ( $95 \mathrm{mg}, 32 \%$ ). $\mathrm{R}_{\mathrm{f}}=0.65$ in $5 \% \mathrm{MeOH}$ /ethyl acetate. ${ }^{1} \mathrm{H} \mathrm{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta: 2.03(2.6 \mathrm{H}, \mathrm{m}), 2.14-2.33(1.6 \mathrm{H}, \mathrm{m}), 2.39(1.7 \mathrm{H}, \mathrm{q}, \mathrm{J}=6.0 \mathrm{~Hz}), 2.48(1 \mathrm{H}, \mathrm{t}, \mathrm{J}=6.6 \mathrm{~Hz}), 2.79(0.5 \mathrm{H}$, s), $3.33(1.9 \mathrm{H}, \mathrm{m}), 3.52(1.5 \mathrm{H}, \mathrm{m}), 3.81(0.2 \mathrm{H}, \mathrm{bs}), 4.12(2.8 \mathrm{H}, \mathrm{m}), 4.37(1.4 \mathrm{H}, \mathrm{t}, \mathrm{J}=5.8 \mathrm{~Hz}), 4.60(0.2 \mathrm{H}, \mathrm{d})$, $5.14(2.0 \mathrm{H}, \mathrm{m}), 5.69(0.3 \mathrm{H}, \mathrm{t}), 5.89(0.5 \mathrm{H}, \mathrm{t}), 6.97(0.8 \mathrm{H}, \mathrm{m}), 7.23-7.39(8.1 \mathrm{H}, \mathrm{m}), 7.56(1.8 \mathrm{H}, \mathrm{m}), 7.63$ $(0.9 \mathrm{H}, \mathrm{q}, \mathrm{J}=7.7 \mathrm{~Hz}), 7.72(1.2 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.6 \mathrm{~Hz}), 8.19(1.7 \mathrm{H}, \mathrm{m}), 8.68(0.8 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta: 20.5,20.8,21.1,28.4,29.3,31.3,31.7,31.8,36.1,36.2,38.6,39.4,39.5,47.2,47.3,47.8,49.0,49.2$, $50.7,56.0,60.4,66.7,66.9,67.3,67.6,114.1,119.6,119.9,120.0,125.0,125.1,127.1,127.7,127.7$, $128.4,128.5,128.7,128.7,134.9,135.2,138.3,141.3,143.8,144.0,147.7,151.5,156.6,156.7,169.4$, $170.2,171.5,173.3,173.7$. Because the NMR spectra were complicated by mixtures of rotamers, the
identity and purity of 15 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{36} \mathrm{H}_{36} \mathrm{~N}_{4} \mathrm{O}_{6} 621.7$; found 621.3.

## N-(2-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-2-ylamino)pentanoyl)

 glycine (16). Methanol was degassed via bubbling nitrogen gas for 1 hour prior to reaction. Meanwhile, 10\% Pd/C (19 mg, $40 \mathrm{wt} \%)$ and benzyl N -(2-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-2-ylamino)pentanoyl)glycinate 15 were combined in a round bottom flask. The flask was purged with nitrogen gas and methanol $(250 \mu \mathrm{~L})$ and DMF $(20 \mu \mathrm{~L})$ were added. The reaction mixture was purged with hydrogen gas and stirred under a hydrogen atmosphere (balloon) for 1.5 hours. The reaction mixture was diluted with of methanol $(270 \mu \mathrm{~L})$ and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / \mathrm{C}$. After filtration, the compound was adsorbed on celite and filtered through a 3 cm silica plug using DCM to $50 \% \mathrm{MeOH} / \mathrm{DCM}$ as eluent. The solvent was removed, solids were redissolved in DCM, and the solution was filtered through 5 cm of celite in a pipette to remove residual silica. Solvent was removed via rotary evaporation affording the tittle compound 16 as a pale yellow waxy solid ( 23 $\mathrm{mg}, 75 \%) . \mathrm{R}_{\mathrm{f}}=0.3$ in $20 \% \mathrm{MeOH} / \mathrm{DCM} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 1.95(3.0 \mathrm{H}, \mathrm{bs}), 2.37(4.4 \mathrm{H}, \mathrm{m}), 3.10-$ $3.59(4.5 \mathrm{H}, \mathrm{m}), 3.94(2.1 \mathrm{H}, \mathrm{m}), 4.12(1.1 \mathrm{H}, \mathrm{m}), 4.27(1.5 \mathrm{H}, \mathrm{s}), 4.45(0.2 \mathrm{H}, \mathrm{bs}), 6.08(0.2 \mathrm{H}, \mathrm{m}), 6.34(0.6 \mathrm{H}$, $m), 6.90(1.0 \mathrm{H}, \mathrm{m}), 7.22(1.7 \mathrm{H}, \mathrm{m}), 7.30(2.5 \mathrm{H}, \mathrm{m}), 7.44-7.62(3.2 \mathrm{H}, \mathrm{m}), 7.67(2.7 \mathrm{H}, \mathrm{m}), 8.11(2.1 \mathrm{H}, \mathrm{m})$, $9.94(0.8 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta: 20.4,20.5,20.8,21.1,31.5,31.9,32.0,36.2,39.4,47.2$, 49.3, 66.8, 114.7, 114.9, 119.4, 119.9, 125.1, 125.2, 127.0, 127.7, 139.2, 141.2, 143.9, 144.0, 146.3, 151.5, 156.7, 156.9, 172.3, 173.7, 174.1, 174,9. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 16 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{6} 531.6$; found 531.3.


## Benzyl N-(2-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-3-ylamino)

 pentanoyl)glycinate (17). Benzyl (2-((() $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)glycinate 2 (190 $\mathrm{mg}, 0.41 \mathrm{mmol}$ ) and glutaric anhydride ( $62 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) were combined in a round bottom flask. The flask was nitrogen purged and cooled to $0^{\circ} \mathrm{C}$. DMF ( $2 \mathrm{~mL}, 0.2 \mathrm{M}$ ) added followed by slow addition of distilled diisopropylethylamine ( $78 \mu \mathrm{~L}, 0.45 \mathrm{mmol}$ ). The reaction was stirred for 18 hours, HBTU ( 154 mg , $0.41 \mathrm{mmol})$ was added, and the reaction mixture was stirred 30 minutes. The reaction was cooled to $0^{\circ}$ C and 3-aminopyridine ( $118 \mathrm{mg}, 1.25 \mathrm{mmol}$ ) was added followed by slow addition of diisopropylethylamine ( $78 \mu \mathrm{~L}, 0.45 \mathrm{mmol}$ ). The reaction was warmed to room temperature and stirred for 18 hours. The reaction mixture was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water $(5 \mathrm{~mL})$, and brine $(5 \mathrm{~mL})$. The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography (50-100\% ethyl acetate/hexanes then $2 \% \mathrm{MeOH} /$ ethyl acetate) to afford the title compound 17 as a clear waxy solid (86 $\mathrm{mg}, 34 \%) \mathrm{R}_{\mathrm{f}}=0.25$ in $5 \% \mathrm{MeOH} /$ ethyl acetate. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 2.00(2.2 \mathrm{H}, \mathrm{m}), 2.26-2.42$ $(2.5 \mathrm{H}, \mathrm{m}), 2.47(1.1 \mathrm{H}, \mathrm{t}, \mathrm{J}=6.7 \mathrm{~Hz}), 2.89(0.4 \mathrm{H}, \mathrm{m}), 3.04(0.3 \mathrm{H}, \mathrm{bd}), 3.33(1.9 \mathrm{H}, \mathrm{m}), 3.47(2.6 \mathrm{H}, \mathrm{m}), 3.81$ ( $0.4 \mathrm{H}, \mathrm{d}, \mathrm{J}=15.9 \mathrm{~Hz}$ ), 4.03-4.25 (2.6H, m), 4.37 (1.6H, m), 4.60 (0.3H, d, J = 35.5 Hz$), 5.15(2.0 \mathrm{H}, \mathrm{m}), 5.72$ ( $0.1 \mathrm{H}, \mathrm{d}$ ), $5.87(0.2 \mathrm{H}, \mathrm{m}), 6.07(0.5 \mathrm{H}, \mathrm{m}), 7.20(1.1 \mathrm{H}, \mathrm{m}), 7.23-7.41(8.7 \mathrm{H}, \mathrm{m}), 7.55(1.9 \mathrm{H}, \mathrm{m}), 7.72$ (1.9H, $d, J=7.4 \mathrm{~Hz}), 8.16(1.0 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.4 \mathrm{~Hz}), 8.27(0.8 \mathrm{H}, \mathrm{bs}), 8.64(0.8 \mathrm{H}, \mathrm{bs}), 9.13(0.8 \mathrm{H}, \mathrm{s}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 19.0,20.7,21.1,31.2,31.8,36.0,36.1,39.2,39.4,47.2,47.2,47.8,48.9,49.2,50.4,50.7,66.7$, $66.8,67.3,67.7,120.0,123.8,125.0,127.1,127.8,128.3,128.5,128.6,128.7,128.7,134.8,135.1,141.0$, 141.3, 143.8, 143.9, 144.5, 156.7, 156.8, 169.5, 170.1, 172.0, 173.7, 174.0. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 17 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{36} \mathrm{H}_{36} \mathrm{~N}_{4} \mathrm{O}_{6}$ 621.7; found 621.3.
## N -(2-((()9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-3-ylamino)pentanoyl)

 glycine (18). Methanol was degassed via bubbling nitrogen gas for 1 hour prior to reaction. Meanwhile, $10 \% \mathrm{Pd} / \mathrm{C}(19 \mathrm{mg}, 40 \mathrm{wt} \%)$ and benzyl N -(2-((( 9 H -fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-3-ylamino)pentanoyl)glycinate 17 ( $104 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) were combined in a round bottom flask. The flask was purged with nitrogen gas and methanol ( $500 \mu \mathrm{~L}, 0.33 \mathrm{M}$ ) and DMF ( $40 \mu \mathrm{~L}$ ) were added. The reaction mixture was purged with hydrogen gas and stirred under a hydrogen atmosphere (balloon) for 1.5 hours. The reaction mixture was diluted with methanol ( $540 \mu \mathrm{~L}$ ) and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / C$. Solvent was removed by rotary evaporation affording the tittle compound 18 as a clear waxy solid ( $89 \mathrm{mg}, 99 \%$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.90(2.6 \mathrm{H}, \mathrm{d}), 2.32(3.4 \mathrm{H}$, $m), 2.78(1.1 \mathrm{H}, \mathrm{bs}), 3.11-3.54(3.0 \mathrm{H}, \mathrm{m}), 3.95(1.4 \mathrm{H}, \mathrm{m}), 4.08(0.7 \mathrm{H}, \mathrm{m}), 4.24(1.5 \mathrm{H}, \mathrm{m}), 4.41(0.2 \mathrm{H}, \mathrm{bs})$, $4.64(0.5 \mathrm{H}, \mathrm{bs}), 6.17(0.2 \mathrm{H}, \mathrm{bs}), 6.34(0.4 \mathrm{H}, \mathrm{bs}), 6.95(0.2 \mathrm{H}, \mathrm{bs}), 7.17(2.1 \mathrm{H}, \mathrm{m}), 7.27-7.41(2.1 \mathrm{H}, \mathrm{m}), 7.49$ $(2.0 H, m), 7.69(2.0 H, m), 8.02-8.28(1.9 H, m), 8.50-8.72(0.8 H, m), 8.88(0.1 H, b s), 9.59(0.2 H, b s), 9.68$ $(0.4 \mathrm{H}, \mathrm{bs}), 9.84(0.5 \mathrm{H}, \mathrm{bs}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 19.6,20.7,21.0,31.1,31.9,35.5,35.9,36.1$, 39.3, 39.8, 47.1. 47.9, 49.2, 49.7, 51.7, 66.6, 66.8, 120.0, 124.3, 125.0, 125.1, 127.0, 127.3, 127.7, 127.9, $136.5,139.1,139.9,141.2,141.5,142.1,142.8,143.8,143.9,156.7,156.9,157.2,172.3,172.5,173.1$, 173.5, 173.7, 173.8, 174.3. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 18 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{6} 531.6$; found 531.2.

## Benzyl N-(2-((() 9 H -fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-4-

ylamino)pentanoyl)glycinate (19). Benzyl (2-((()(9H-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)glycinate $\mathbf{2}$ ( $190 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) and glutaric anhydride ( $62 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) were combined in a round bottom
flask. The flask was purged with nitrogen gas and cooled to $0^{\circ} \mathrm{C}$. DMF ( $2 \mathrm{~mL}, 0.2 \mathrm{M}$ ) was added followed by slow addition of distilled diisopropylethylamine ( $78 \mu \mathrm{~L}, 0.45 \mathrm{mmol}$ ). The reaction was stirred for 18 hours, HBTU ( $154 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) was added, and the reaction mixture was stirred 30 minutes. The reaction was cooled to $0^{\circ} \mathrm{C}$ and 4 -aminopyridine ( $118 \mathrm{mg}, 1.25 \mathrm{mmol}$ ) was added followed by slow addition of diisopropylethylamine ( $78 \mu \mathrm{~L}, 0.45 \mathrm{mmol}$ ). The reaction was warmed to room temperature and stirred 18 hours. The reaction was diluted with ethyl acetate ( 5 mL ) and washed with $5 \%$ aqueous $\mathrm{AcOH}(5 \mathrm{~mL})$, water $(5 \mathrm{~mL})$, and brine $(5 \mathrm{~mL})$. The organic layer was separated, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The resulting oil was then purified by silica gel chromatography (50-100\% ethyl acetate/hexanes then $2 \% \mathrm{MeOH} /$ ethyl acetate) to afford the title compound 19 as a clear waxy solid (86 $\mathrm{mg}, 34 \%) \mathrm{R}_{\mathrm{f}}=0.25$ in $5 \% \mathrm{MeOH} / \mathrm{ethyl}$ acetate. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.89-2.12(2.6 \mathrm{H}, \mathrm{m})$, 2.21$2.51(3.3 \mathrm{H}, \mathrm{m}), 2.83-3.11(0.6 \mathrm{H}, \mathrm{m}), 3.23-3.56(3.7 \mathrm{H}, \mathrm{m}), 3.83(0.4 \mathrm{H}, \mathrm{bd}, \mathrm{J}=14.8 \mathrm{~Hz}), 4.00-4.22(2.3 \mathrm{H}, \mathrm{m})$, $4.39(1.4 \mathrm{H}, \mathrm{m}), 4.59(0.4 \mathrm{H}, \mathrm{d}, \mathrm{J}=40.2 \mathrm{~Hz}), 5.15(2.0 \mathrm{H}, \mathrm{m}), 5.61(0.4 \mathrm{H}, \mathrm{m}), 5.80(0.5 \mathrm{H}, \mathrm{m}), 7.23-7.41(7.8 \mathrm{H}$, m), 7.44-7.62 (3.7H, m), $7.71(1.8 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.4 \mathrm{~Hz}), 8.41(1.6 \mathrm{H}, \mathrm{bs}), 9.09(0.8 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta: 18.9,20.6,21.0,30.9,31.0,31.7,36.3,36.4,39.3,39.4,47.2,47.2,47.9,49.0,49.3 .50 .5,50.7$, $66.7,66.8,67.4,67.7,113.6,120.0,124.9,125.0,127.1,127.8,128.3,128.6,128.7,128.8,134.7,135.0$, 141.3, 143.7, 143.8, 145.8, 150.1, 156.6, 169.5, 170.1, 172.3, 173.7, 174.0. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of 19 was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{36} \mathrm{H}_{36} \mathrm{~N}_{4} \mathrm{O}_{6}$ 621.7; found 621.3.

## N -(2-((( $(9 \mathrm{H}$-fluoren-9-yl)methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-4-ylamino)pentanoyl)

 glycine (20). Methanol was degassed via bubbling nitrogen gas for 1 hour prior to reaction. Meanwhile, $10 \% \mathrm{Pd} / \mathrm{C}(19 \mathrm{mg}, 40 \mathrm{wt} \%)$ and benzyl N -(2-((( $9 \mathrm{H}-\mathrm{fluoren}-9-\mathrm{yl})$ methoxy)carbonyl)amino)ethyl)-N-(5-oxo-5-(pyridin-4-ylamino)pentanoyl)glycinate $19(43 \mathrm{mg}, 0.07 \mathrm{mmol})$ were combined in a round bottom flask. The flask was purged with nitrogen gas and methanol ( $250 \mu \mathrm{~L}, 0.28 \mathrm{M}$ ) and DMF ( $20 \mu \mathrm{~L}$ ) were added. The reaction mixture was purged with hydrogen gas and stirred under a hydrogen atmosphere (balloon) for 1.5 hours. The reaction mixture was diluted with methanol ( $270 \mu \mathrm{~L}$ ) and filtered through a 3 cm celite plug to remove $\mathrm{Pd} / \mathrm{C}$. Solvent was then removed by rotary evaporation affording the tittle compound 20 as a clear waxy solid ( $22 \mathrm{mg}, 60 \%$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}(400 \mathrm{MHz}, \mathrm{DMSO}) \delta: 1.79(2.2 \mathrm{H}, \mathrm{p}, \mathrm{J}=7.3 \mathrm{~Hz}$ ), $2.24(1.7 \mathrm{H}, \mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}), 2.36(3.1 \mathrm{H}, \mathrm{m}), 2.72(0.3 \mathrm{H}, \mathrm{s}), 2.88(0.3 \mathrm{H}, \mathrm{s}), 2.95(0.3 \mathrm{H}, \mathrm{bs}), 3.12(2.0 \mathrm{H}, \mathrm{m}), 3.22$ $(0.5 \mathrm{H}, \mathrm{m}), 3.34(1.8 \mathrm{H}, \mathrm{m}), 3.54(0.4 \mathrm{H}, \mathrm{bs}), 3.80(2.2 \mathrm{H}, \mathrm{m}), 4.21(1.8 \mathrm{H}, \mathrm{m}), 4.30(0.9 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.0 \mathrm{~Hz}), 7.30$ $(2.0 H, q, J=7.3 H z), 7.39(2.5 H, m), 7.57(2.1 H, m), 7.65(2.3 H, t, J=8.0 H z), 7.86(2.2 H, m), 8.37(2.0 H$, m), 10.45 (1.0H, m). ${ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO) $\delta: 20.9,21.1,31.2,32.1,36.1,36.4,38.8,47.2,47.4$,48.1, 52.6, 65.9, 110.2, 113.6, 113.6, 120.5, 120.6, 121.8, 125.5, 125.7, 127.5, 127.8, 128.0, 129.4, 137.9, 139.9, 141.1, 141.2, 144.3, 144.4, 146.3, 146.3, 150.7, 156.6, 156.7, 171.9, 172.3, 172.8, 173.0. Because the NMR spectra were complicated by mixtures of rotamers, the identity and purity of $\mathbf{1 2}$ was further confirmed with LCMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{6} 531.6$; found 531.2.

## Synthesis of PNA oligomers

The PNAs used in this study (Figures 2 and 3) were synthesized on an Expedite 8909 synthesizer at 2 $\mu$ mol scale using NovaSyn TG Sieber resin (Novabiochem) and following methods previously developed in our group. ${ }^{2}$ Commercial PNA-T-monomer was purchased from Link Technologies. M monomer was synthesized using procedures previously described by our group. ${ }^{3}$ For newly reported monomers as single modifications in PNA, manual coupling was employed using the same solutions prepared for the Expedite. First, the resin is washed twice with 1.0 mL dry DMF. Meanwhile, a solution of monomer is prepared in NMP at 0.2 M concentration using $0.1 \mathrm{~mL} \mathrm{NMP}, 0.1 \mathrm{~mL}$ base solution, and 0.1 mL activator solution. After washing the resin, half of monomer solution is loaded onto resin and left for 30 minutes with sporadic agitation using the two-syringe push/pull method. The resin is washed once with 1.0 mL DMF then the remaining monomer solution is loaded and reacted for 30 minutes. The resin is then washed with 1.0 mL DMF then 1.0 mL capping solution is added and left for 10 minutes with sporadic agitation. The resin is washed one final time with 1.0 mL DMF then returned to the Expedite for the remainder of the synthesis. PNAs containing multiple V-base modifications were performed entirely on the Expedite. PNAs were cleaved from the solid support using 0.6 mL of $20 \% \mathrm{~m}$-cresol in TFA for 2 h using two-syringe pull-push method. Crude PNA (separated in three Eppendorf tubes, $200 \mu \mathrm{~L}$ in each) was precipitated by the addition of chilled diethyl ether ( $\sim 1.0 \mathrm{~mL}$ ) and incubation at $-20^{\circ} \mathrm{C}$ for 15 minutes followed by the centrifugation (15000 rpm, 30 minutes). Ethyl ether was then removed using a pipette and the PNA was washed once more with 1.0 mL ethyl ether and centrifuged ( $15000 \mathrm{rpm}, 10$ minutes). The ethyl ether was again decanted and the crude PNA, white solid, was dissolved in 1.25 mL $0.1 \%$ formic acid in water. Samples were then analyzed by LC/MS to confirm the synthesis, purified by HPLC, and finally reanalyzed by LC/MS to confirm their purity. Yields were quantified by absorbance at 260 nm using molar extinction coefficient of $8560 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ for $\mathrm{T}, 806 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ for $\mathrm{M}, 3770 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ for ${ }^{\mathrm{DA}} \mathrm{N}$ and CR1, $3772 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ for CR2-5, and $4200 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ for V . The final three values were determined by acylation of respective heterocycles, removal of Boc protecting groups (if necessary), and subsequent

[^1]serial dilution in water with absorbance measurements at 260 nm . After PNA quantification, solutions were lyophilized and redissolved in nuclease-free DI water to a final concentration of $240 \mu \mathrm{M}$.

## RNA Purification

RNA hairpins were purchased from Dharmacon. Prior to use, RNA was deprotected in accordance with manufacturer instruction and purified on reverse phase HPLC using a gradient of acetonitrile in 50 mM aqueous triethyl ammonium acetate buffer ( pH 7.4 ). Post purification, a final injection of the collected material confirmed its purity. Absorbance at 260 nm was used to quantify the RNA. After, RNA was lyophilized and redissolved in nuclease-free water to a concentration of $240 \mu \mathrm{M}$.

Table S1: Molecular weights and deconvoluted masses of PNAs.

| PNA(monomer) | Mass calcd | $\mathbf{M} / \mathbf{Z}$ calcd (M+6, $\mathbf{M}+5, \mathbf{M}+4 \ldots)$ | $\mathbf{M} / \mathbf{Z}$ found (M+6, $\mathbf{M}+5, \mathbf{M}+4 . .)$. |
| :---: | :---: | :---: | :---: |
| PNA1 X $={ }^{\text {DA }} \mathrm{N}$ | 2456.9 | $410.5,492.4,615.2,820.0,1229.5$ | $410.6,492.5,615.3,820.1,1229.6$ |
| PNA1 X = CR1 | 2471.0 | $412.8,495.2,618.7,824.6,1236.5$ | $412.9,495.3,618.8,824.7,1236.5$ |
| PNA1 X = CR2 | 2390.9 | $399.5,479.2,598.7,798.0,1196.4$ | $399.6,479.3,598.8,798.0,1196.7$ |
| PNA1 X = CR3 | 2404.7 | $401.8,481.9,602.1,802.6,1203.4$ | $401.9,482.1,602.3,802.8,1203.6$ |
| PNA1 X = CR4 | 2404.7 | $401.8,481.9,602.1,802.6,1203.4$ | $402.0,482.1,602.4,802.8,1203.6$ |
| PNA1 X = CR5 | 2404.7 | $401.8,481.9,602.1,802.6,1203.4$ | $401.9,482.0,602.3,802.7,1203.5$ |
| PNA1 X = V | 2461.9 | $411.3,493.4,616.5,821.6,1232.0$ | $411.4,493.5,616.6,821.8,1231.8$ |
| PNA2 | 2543.1 | $424.8,509.6,636.8,848.7,1272.5$ | $425.0,509.8,636.9,848.8,1272.7$ |
| PNA3 | 2543.1 | $424.8,509.6,636.8,848.7,1272.5$ | $424.8,509.6,636.8,848.8,1272.8$ |
| PNA4 | 2624.2 | $438.4 .525 .8,657.1,875.7,1313.1$ | $438.5,526.0,657.1,875.9,1313.3$ |



Figure S1. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 $\mathrm{X}={ }^{\mathrm{DA}} \mathrm{N}$.


Figure S2. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 X = CR1.


Figure S3. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 X = CR2.


Figure S4. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 X = CR3.


Figure S5. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 X = CR4.


Figure S6. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 X = CR5.


Figure S7. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA1 $\mathrm{X}=\mathrm{V}$.


Figure S8. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA2.


Figure S9. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA3.


Figure S10. Crude HPLC (top) and purified LC/MS (middle/bottom) data of PNA4.

## UV-melting experiments

UV-melting experiments were performed on a Shimadzu UV-2600 spectrophotometer equipped with TMSPC-8 temperature controllers. A temperature ramp rate of $1.0^{\circ} \mathrm{C}$ was used in all cases. Absorbance was monitored at 300 nm for triplex-formation involving PNAs $(18 \mu \mathrm{M})$. Experiments were performed in a buffer containing 50 mM potassium phosphate, $2 \mathrm{mM} \mathrm{MgCl} 2,90 \mathrm{mM} \mathrm{KCl}, 10 \mathrm{mM} \mathrm{NaCl}$ at pH 7.4 .

Solutions were prepared as follows. Both PNA and RNA combined from $240 \mu \mathrm{M}$ stock solutions in water, frozen, and lyophilized. Buffer ( $230 \mu \mathrm{~L}, 18 \mu \mathrm{M}$ ) is added and the pellet is dissolved by vortexing. The solution is then heated to $95{ }^{\circ} \mathrm{C}$ for 5 minutes and then snap-cooled to $4^{\circ} \mathrm{C}$ and left for 5 minutes. Samples were then melted three time, typically from $10^{\circ} \mathrm{C}$ to $95^{\circ} \mathrm{C}$. Five of the replicates were then used to determine average and standard deviation in melting temperature ( $T_{\mathrm{m}}$ ).

Table S2. Replicated Melting data for PNA1 X $={ }^{\mathrm{DA}} \mathrm{N}$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 40.2 | 50.5 | 38.9 | 35.2 |
| Melting | 41.2 | 50.3 | 38.7 | 35.2 |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | 40.8 | 49.8 | 39.3 | 35.1 |
|  | 40.5 | 51.3 | 38.7 | 35.6 |
|  | 41.1 | 51.1 | 39.8 | 34.7 |
| Average | 40.8 | 50.6 | 39.1 | 35.2 |
| Std. Dev | 0.4 | 0.6 | 0.5 | 0.3 |

Table S3. Melting data for PNA1 X = CR1.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 32.2 | 40.8 | 51.2 | 35.6 |
| Melting | 31.5 | 40.8 | 51.1 | 35.7 |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | 32.7 | 40.2 | 51.4 | 35.3 |
|  | 32.2 | 40.4 | 50.7 | 35.8 |
|  | 32.3 | 40.7 | 50.1 | 35.0 |
| Average | 32.2 | 40.6 | 50.9 | 35.5 |
| Std. Dev | 0.4 | 0.2 | 0.5 | 0.3 |

Table S4. Melting data for PNA1 $\mathrm{X}=\mathrm{CR} 2$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | NM | 55.5 | 39.9 | 34.0 |
| Melting | NM | 55.9 | 40.2 | 34.3 |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | NM | 55.3 | 40.2 | 33.4 |
|  | NM | 55.4 | 40.6 | 34.4 |
| Average | NM | 55.3 | 40.5 | 33.7 |
| Std. Dev | - | 55.5 | 40.3 | 34.0 |
|  | - | 0.3 | 0.3 | 0.4 |

*NM - no melting

Table S5. Melting data for PNA1 $\mathrm{X}=\mathrm{CR} 3$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 30.8 | 39.1 | 51.6 | 34.6 |
| Melting | 29.9 | 39.7 | 51.7 | 34.6 |
| Temperatures ( ${ }^{\circ} \mathrm{C}$ ) | 30.6 | 39.1 | 51.3 | 34.2 |
|  | 30.6 | 39.0 | 51.9 | 34.7 |
| Average | 29.6 | 38.8 | 51.0 | 34.6 |
| Std. Dev | 30.3 | 39.1 | 51.5 | 34.5 |
|  | 0.5 | 0.3 | 0.3 | 0.2 |

Table S6. Melting data for PNA1 $\mathrm{X}=\mathrm{CR4}$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 32.0 | 36.5 | 51.3 | 34.2 |
| Melting | 31.8 | 37.2 | 51.0 | 34.5 |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | 31.3 | 37.0 | 51.4 | 34.6 |
|  | 31.8 | 37.1 | 50.8 | 34.7 |
| Average | 31.9 | 37.8 | 51.6 | 34.6 |
| Std. Dev | 31.8 | 37.1 | 51.2 | 34.5 |
|  | 0.3 | 0.5 | 0.3 | 0.2 |

Table S7. Melting data for PNA1 $\mathrm{X}=\mathrm{CR5}$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 32.8 | 37.7 | 52.7 | 35.3 |
| Melting | 32.0 | 37.8 | 52.2 | 35.6 |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | 32.4 | 37.9 | 52.1 | 35.8 |
|  | 33.0 | 37.8 | 52.6 | 35.7 |
|  | 32.7 | 37.8 | 52.7 | 36.0 |
| Average | 32.6 | 37.8 | 52.5 | 35.7 |
| Std. Dev | 0.4 | 0.1 | 0.3 | 0.3 |

Table S8. Melting data for PNA1 $\mathrm{X}=\mathrm{V}$.

|  | HRP1 (G) | HRP2 (A) | HRP3 (C) | HRP4 (U) |
| :---: | :---: | :---: | :---: | :---: |
|  | 35.8 | 42.3 | 60.1 | 36.5 |
|  | 36.4 | 41.6 | 59.7 | 36.9 |
|  | 36.3 | 42.3 | 60.8 | 37.1 |
|  | 36.2 | 42.8 | 60.3 | 37.4 |
| Std. Dev | 36.4 | 42.6 | 60.4 | 37.0 |

Table S9. Melting data for PNA2.

|  | HRP5 | HRP6 | HRP7 |
| :---: | :---: | :---: | :---: |
| Melting | 49.3 | 40.8 | NM |
|  | 48.7 | 40.9 | NM |
|  | 49.1 | 39.5 | NM |
|  | 49.3 | 40.2 | NM |
| Average | 49.2 | 39.1 | NM |
| Std. Dev | 49.1 | 40.1 | - |
|  | 0.2 | 0.8 | - |

Table S10. Melting data for PNA3.

|  | HRP5 | HRP6 | HRP7 |
| :---: | :---: | :---: | :---: |
|  | 29.6 | 42.7 | NM |
| Melting | 30.5 | 42.5 | NM |
| Temperatures $\left({ }^{\circ} \mathrm{C}\right)$ | 30.2 | 42.5 | NM |
|  | 29.6 | 43.2 | NM |
|  | 30.2 | 41.8 | NM |
| Average | 30.0 | 42.5 | - |
| Std. Dev | 0.4 | 0.5 | - |

## Isothermal Titration Calorimetry

Isothermal titration calorimetry experiments were done on a MicroCal ITC200 instrument at $25{ }^{\circ} \mathrm{C}$ in 50 mM potassium phosphate buffer ( pH 7.4 ) containing $2 \mathrm{mM} \mathrm{MgCl} 2,90 \mathrm{mM} \mathrm{KCl}, 10 \mathrm{mM} \mathrm{NaCl}$. In a typical experiment $2.45 \mu \mathrm{~L}$ aliquots of PNA solution were sequentially injected from a $40 \mu \mathrm{~L}$ rotating syringe ( 750 rmp ) into $200 \mu \mathrm{~L}$ of RNA or DNA hairpin solution. Standard concentration of PNA was $75 \mu \mathrm{M}$ and stand concentration of RNA was $10 \mu \mathrm{M}$. Experiments which deviate from these values have concentrations listed on their figures. Results were analyzed using MicroCal PEAQ-ITC software.


Figure S11. ITC experiment of PNA1 X = CR3 vs HRP3.


Figure S12. ITC experiment of PNA1 $\mathrm{X}=\mathrm{V}$ vs HRP1.


Figure S13.: ITC experiment of PNA1 $\mathrm{X}=\mathrm{V}$ vs HRP2.


Figure S14. ITC experiment of PNA1 $X=V$ vs HRP3.


Figure S15. ITC experiment of PNA1 $\mathrm{X}=\mathrm{V}$ vs HRP4.


Figure S16. ITC experiment of PNA2 vs HRP5.


Figure S17. ITC experiment of PNA2 vs HRP7.


Figure S18. ITC experiment of PNA3 vs HRP5.


Figure S19. ITC experiment of PNA3 vs HRP6.


Figure S20. ITC experiment of PNA4 vs HRP5.


Figure S21. ITC experiment of PNA4 vs HRP7.

Table S11. PNA1 $X=$ CR3 binding affinity and thermodynamic data obtained by ITC

| Name | $\mathrm{Kd}(\mathrm{M})$ | $\mathrm{Ka}\left(\mathrm{M}^{-1}\right)$ | $\Delta \mathrm{H}(\mathrm{kcal} / \mathrm{mol})$ | $-\mathrm{T} \Delta \mathrm{S}(\mathrm{kcal} / \mathrm{mol})$ | N |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PNA1 X = CR3 vs HRP3 01 | $1.3 \mathrm{E}-07$ | $7.8 \mathrm{E}+06$ | -34.6 | 25.2 | 1.0 |
| PNA1 X = CR3 vs HRP3 02 | $1.2 \mathrm{E}-07$ | $8.2 \mathrm{E}+06$ | -34.7 | 25.2 | 1.0 |
| PNA1 X = CR3 vs HRP3 03 | $1.3 \mathrm{E}-07$ | $7.7 \mathrm{E}+06$ | -33.4 | 24.0 | 0.9 |
| Average | $1.3 \mathrm{E}-07$ | $7.9 \mathrm{E}+06$ | -34.2 | 24.8 | 1.0 |
| Std. Dev. | $4.0 \mathrm{E}-09$ | $2.6 \mathrm{E}+05$ | 0.7 | 0.7 | 0.06 |

Table S12. PNA1 $\mathrm{X}=\mathrm{V}$ binding affinity and thermodynamic data obtained by ITC

| Name | $\mathrm{Kd}(\mathrm{M})$ | $\mathrm{Ka}\left(\mathrm{M}^{-1}\right)$ | $\Delta \mathrm{H}(\mathrm{kcal} / \mathrm{mol})$ | $-\mathrm{T} \Delta \mathrm{S}(\mathrm{kcal} / \mathrm{mol})$ | N |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PNA1 X = V vs HRP1 01 | $1.2 \mathrm{E}-06$ | $8.5 \mathrm{E}+05$ | -11.0 | 2.9 | 1.0 |
| PNA1 X = V vs HRP1 02 | $9.4 \mathrm{E}-07$ | $1.1 \mathrm{E}+06$ | -9.1 | 0.9 | 1.2 |
| PNA1 X = V vs HRP1 03 | $7.9 \mathrm{E}-07$ | $1.3 \mathrm{E}+06$ | -11.7 | 3.3 | 0.9 |
| Average | $9.7 \mathrm{E}-07$ | $1.1 \mathrm{E}+06$ | -10.6 | 2.4 | 1.0 |
| Std. Dev. | $2.0 \mathrm{E}-07$ | $2.1 \mathrm{E}+05$ | 1.3 | 1.3 | 0.13 |
| PNA1 X = V vs HRP2 01 | $5.8 \mathrm{E}-07$ | $1.7 \mathrm{E}+06$ | -33.4 | 24.9 | 1.0 |
| PNA1 X = V vs HRP2 02 | $5.7 \mathrm{E}-07$ | $1.8 \mathrm{E}+06$ | -25.8 | 17.3 | 1.0 |
| PNA1 X = V vs HRP2 03 | $6.6 \mathrm{E}-07$ | $1.5 \mathrm{E}+06$ | -21.5 | 13.0 | 1.0 |
| Average | $6.0 \mathrm{E}-07$ | $1.7 \mathrm{E}+06$ | -26.9 | 18.4 | 1.0 |
| Std. Dev. | $5.0 \mathrm{E}-08$ | $1.3 \mathrm{E}+05$ | 6.0 | 6.0 | 0.0 |
| PNA1 X = V vs HRP3 01 | $7.0 \mathrm{E}-08$ | $1.4 \mathrm{E}+07$ | -33.5 | 23.8 | 1.0 |
| PNA1 X = V vs HRP3 02 | $7.0 \mathrm{E}-08$ | $1.4 \mathrm{E}+07$ | -36.3 | 26.5 | 0.9 |
| PNA1 X = V vs HRP3 03 | $7.2 \mathrm{E}-08$ | $1.4 \mathrm{E}+07$ | -39.3 | 29.5 | 0.9 |
| Average | $7.1 \mathrm{E}-08$ | $1.4 \mathrm{E}+07$ | -36.4 | 26.6 | 0.9 |
| Std. Dev. | $1.4 \mathrm{E}-09$ | $2.8 \mathrm{E}+05$ | 2.9 | 2.9 | 0.05 |
| PNA1 X = V vs HRP4 01 | $1.2 \mathrm{E}-06$ | $8.4 \mathrm{E}+05$ | -16.9 | 8.8 | 0.9 |
| PNA1 X = V vs HRP4 02 | $8.7 \mathrm{E}-07$ | $1.1 \mathrm{E}+06$ | -14.1 | 5.8 | 0.9 |
| PNA1 X = V vs HRP4 03 | $1.2 \mathrm{E}-06$ | $8.9 \mathrm{E}+05$ | -16.8 | 8.7 | 0.9 |
| Average | $1.1 \mathrm{E}-06$ | $9.6 \mathrm{E}+05$ | -15.9 | 7.7 | 0.9 |
| Std. Dev. | $1.7 \mathrm{E}-07$ | $1.7 \mathrm{E}+05$ | 1.6 | 1.7 | 0.02 |

Table S13. PNA2 binding affinity and thermodynamic data obtained by ITC

| Name | $\mathrm{Kd}(\mathrm{M})$ | $\mathrm{Ka}\left(\mathrm{M}^{-1}\right)$ | $\Delta \mathrm{H}(\mathrm{kcal} / \mathrm{mol})$ | $-\mathrm{T} \Delta \mathrm{S}(\mathrm{kcal} / \mathrm{mol})$ | N |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PNA2 vs HRP5 01 | $2.1 \mathrm{E}-07$ | $4.9 \mathrm{E}+06$ | -29.3 | 20.2 | 1.0 |
| PNA2 vs HRP5 02 | $2.1 \mathrm{E}-07$ | $4.7 \mathrm{E}+06$ | -33.9 | 24.8 | 1.0 |
| PNA2 vs HRP5 03 | $2.1 \mathrm{E}-07$ | $4.7 \mathrm{E}+06$ | -33.4 | 24.3 | 0.9 |
| Average | $2.1 \mathrm{E}-07$ | $4.8 \mathrm{E}+06$ | -32.2 | 23.1 | 0.9 |
| Std. Dev. | $4.7 \mathrm{E}-09$ | $1.1 \mathrm{E}+05$ | 2.5 | 2.5 | 0.05 |

Table S14. PNA3 binding affinity and thermodynamic data obtained by ITC

| Name | $\mathrm{Kd}(\mathrm{M})$ | $\mathrm{Ka}\left(\mathrm{M}^{-1}\right)$ | $\Delta \mathrm{H}(\mathrm{kcal} / \mathrm{mol})$ | $-\mathrm{T} \Delta \mathrm{S}(\mathrm{kcal} / \mathrm{mol})$ | N |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PNA3 vs HRP6 01 | $9.1 \mathrm{E}-07$ | $1.1 \mathrm{E}+06$ | -20.7 | 12.5 | 1.0 |
| PNA3 vs HRP6 02 | $1.0 \mathrm{E}-06$ | $9.7 \mathrm{E}+05$ | -21.9 | 13.8 | 1.0 |
| PNA3 vs HRP6 03 | $1.1 \mathrm{E}-06$ | $9.1 \mathrm{E}+05$ | -22.8 | 14.6 | 1.1 |
| Average | $1.0 \mathrm{E}-06$ | $1.0 \mathrm{E}+06$ | -21.8 | 13.6 | 1.0 |
| Std. Dev. | $9.8 \mathrm{E}-08$ | $9.9 \mathrm{E}+04$ | 1.0 | 1.1 | 0.07 |

## Ab initio calculations

To obtain insights in possible H-bonding interactions of nucleobases in Figure 2, we optimized geometry of triplets formed by CR1 (same as ${ }^{\mathrm{DA}} \mathrm{N}$ ) and CR3 (same as CR2) with C-G base pair using ab initio calculations at the B3LYP/6-31G+(d,p) level of theory. The results (Figure S22A) showed that the CR1 nucleobase formed two H -bonds, one between the $\mathrm{C}=\mathrm{O}$ of linker and $-\mathrm{NH}_{2}$ of cytosine and the second, somewhat longer ( $2.11 \AA$ ) between $-\mathrm{NH}_{2}$ of CR1 and N 7 of guanosine. Most notably, ${ }^{\mathrm{DA}} \mathrm{N}$ was tilted out of plane of C-G duplex by $12^{\circ}$ angle. In contrast, the optimized geometries of $\mathrm{CR} 3 \cdot \mathrm{C}-\mathrm{G}$ triplet converged on being planar (Figure S22B). Steric interactions between pyridine ( H 6 ) and guanosine ( $\mathrm{C}=\mathrm{O}$ ) shifted CR3 towards cytidine and the PNA backbone resulting in only a weak interaction between pyridine N in CR3 and $-\mathrm{NH}_{2}$ of cytidine observed at the distance of $2.42 \AA$ and $127^{\circ}$ angle.




Figure S22. Geometry optimization of $(A) C R 1 \bullet C-G$ and $(B) C R 3 \cdot C-G$ triplets using B3LYP/6-31G+(d, p). Top view shows H-bond distances and side view shows planarity of extended system.

## Molecular Dynamics

Molecular dynamics simulations were carried out with Maestro Version 12.6.144, Release 2020-4. PNAdsRNA triple helix, constructed in our previous NMR structural studies, ${ }^{4}$ was chosen as starting structure for the computational studies on modified PNA complexes with RNA. The sequence of interest was built by extending RNA and PNA strands to match the structure of PNA1-HRP3 triplex. The RNA hairpin loop

[^2]and lysine were added from Maestro templates. Next, nucleobases were changed by hand to match the desired sequence. All extensions were achieved by copying existing pairs or triples to keep initial geometry and distances. Sequence was adjusted in the same way by copying nucleobases from the core strand, except for the V base, which was added manually according to optimized geometry from quantum calculations. Designed dsRNA-PNA triple helix was subjected to 5 ns molecular dynamics in MacroModel, with applying SHAKE to all bonds. Sampled structure with lowest energy was minimized applying AMBER* force field. For the 50 ns molecular dynamics simulation carried out in Desmond, minimized structure was solvated with explicit TIP4PD water molecules, in a orthorhombic box of $195088 \AA^{3}$ with 19 atoms of sodium added as counterions and mimicking 0.15 M NaCl solution. The final system is made by about 17800 atoms. Amber* force field was applied to all atoms.


Figure 23. Triple helix derived from our previous NMR structural studies ${ }^{4}$ (on the left) and structure used in MD simulation (on the right). RNA is colored in red and PNA in blue.


Figure S24. RMSD plot along with the molecular dynamics time for the whole dsRNA-PNA triple helix.


Figure S25. RMSD plot along with the molecular dynamics time for the $\mathrm{V} \bullet \mathrm{C}-\mathrm{G}$ base triplex.

Table S15. Cartesian coordinates and total energies for geometry optimized base triplets $\mathrm{CR} 1 \cdot \mathrm{C}-\mathrm{G}$, $\mathrm{Cr} 3 \cdot \mathrm{C}-\mathrm{G}$, and $\mathrm{V} \cdot \mathrm{C}-\mathrm{G}$ obtained at B3LYP/6-31G+(d,p) level.

Geometry optimized base triple ${ }^{\mathrm{DA}} \mathrm{N}^{*} \mathrm{C}-\mathrm{G}$

| Center <br> Number | Atomic <br> Number | Atomic Type | Coordinates (Angstroms) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | X | Y | Z |
| 1 | 6 | 0 | 2.689518 | 6.273285 | 0.336996 |
| 2 | 7 | 0 | 4.820532 | -3.483439 | -0.142699 |
| 3 | 6 | 0 | 3.812507 | -4.402351 | -0.182042 |
| 4 | 6 | 0 | 2.505720 | -4.027031 | -0.204818 |
| 5 | 6 | 0 | 2.227431 | -2.620151 | -0.187013 |
| 6 | 7 | 0 | 0.970422 | -2.163451 | -0.210763 |
| 7 | 7 | 0 | 3.225414 | -1.716208 | -0.147992 |
| 8 | 6 | 0 | 4.522461 | -2.110274 | -0.122561 |
| 9 | 8 | 0 | 5.472876 | -1.297465 | -0.081494 |
| 10 | 6 | 0 | 6.229096 | -3.892721 | -0.116254 |
| 11 | 7 | 0 | 1.879473 | 5.066057 | 0.219555 |
| 12 | 6 | 0 | 0.503060 | 4.986756 | 0.162386 |
| 13 | 7 | 0 | 0.056095 | 3.754464 | 0.053546 |
| 14 | 6 | 0 | 1.201357 | 2.974178 | 0.036959 |
| 15 | 6 | 0 | 1.373712 | 1.562087 | -0.059221 |
| 16 | 8 | 0 | 0.491716 | 0.685084 | -0.156565 |
| 17 | 7 | 0 | 2.727350 | 1.196412 | -0.033580 |
| 18 | 6 | 0 | 3.787648 | 2.071491 | 0.069104 |
| 19 | 7 | 0 | 5.020777 | 1.528289 | 0.072777 |


| 20 | 7 | 0 | 3.636535 | 3.390107 | 0.158618 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 21 | 6 | 0 | 2.342865 | 3.775521 | 0.139400 |
| 22 | 6 | 0 | -2.452306 | -5.314499 | 0.700869 |
| 23 | 6 | 0 | -2.673847 | -3.882300 | 0.256081 |
| 24 | 8 | 0 | -1.763204 | -3.213304 | -0.233977 |
| 25 | 7 | 0 | -3.950955 | -3.411607 | 0.453418 |
| 26 | 6 | 0 | -4.259904 | 0.153903 | 0.015207 |
| 27 | 7 | 0 | -3.705908 | -1.074132 | 0.227417 |
| 28 | 6 | 0 | -4.497813 | -2.130586 | 0.213380 |
| 29 | 6 | 0 | -5.903407 | -2.075398 | 0.010669 |
| 30 | 6 | 0 | -6.476838 | -0.840200 | -0.191220 |
| 31 | 6 | 0 | -5.665315 | 0.314199 | -0.199292 |
| 32 | 6 | 0 | -6.155269 | 1.635394 | -0.403796 |
| 33 | 7 | 0 | -3.401034 | 1.206608 | 0.031736 |
| 34 | 6 | 0 | -5.287457 | 2.689632 | -0.388420 |
| 35 | 6 | 0 | -3.888905 | 2.433634 | -0.162088 |
| 36 | 7 | 0 | -3.023883 | 3.480657 | -0.127151 |
| 37 | 1 | 0 | 3.314280 | 6.220707 | 1.231381 |
| 38 | 1 | 0 | 4.115800 | -5.442726 | -0.194347 |
| 39 | 1 | 0 | 1.713601 | -4.763480 | -0.236972 |
| 40 | 1 | 0 | 0.799674 | -1.153087 | -0.193391 |
| 41 | 1 | 0 | 0.168261 | -2.787151 | -0.221254 |
| 42 | 1 | 0 | 6.755327 | -3.479586 | -0.978877 |
| 43 | 1 | 0 | -0.110463 | 5.876250 | 0.206996 |
| 44 | 1 | 0 | 2.920666 | 0.176834 | -0.086801 |
| 45 | 1 | 0 | 5.177934 | 0.514864 | 0.016230 |
| 46 | 1 | 0 | 5.806155 | 2.154582 | 0.161494 |
| 47 | 1 | 0 | -3.377285 | -5.863883 | 0.889728 |
| 48 | 1 | 0 | -1.869411 | -5.834608 | -0.062311 |
| 49 | 1 | 0 | -4.627618 | -4.106766 | 0.738895 |
| 50 | 1 | 0 | -6.497467 | -2.982929 | 0.008900 |
| 51 | 1 | 0 | -7.547489 | -0.747090 | -0.351653 |
| 52 | 1 | 0 | 3.328017 | 6.387441 | -0.542150 |
| 53 | 1 | 0 | 2.023510 | 7.132962 | 0.412883 |


| 54 | 1 | 0 | 6.708330 | -3.529753 | 0.795143 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 55 | 1 | 0 | 6.275779 | -4.980954 | -0.145761 |
| 56 | 1 | 0 | -1.862222 | -5.302268 | 1.623457 |
| 57 | 1 | 0 | -5.630516 | 3.707732 | -0.541426 |
| 58 | 1 | 0 | -7.217439 | 1.792234 | -0.571294 |
| 59 | 1 | 0 | -3.353847 | 4.405622 | -0.355735 |
| 60 | 1 | 0 | -2.014274 | 3.346831 | -0.056967 |

E(RB3LYP) $=-1697.68295268$ Hartree

Geometry optimized base triple $\mathrm{Cr} 2(3)^{*} \mathrm{C}-\mathrm{G}$

| Center <br> Number | Atomic <br> Number | Atomic <br> Type | Coordinates (Angstroms) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | X | Y | Z |
| 1 | 6 | 0 | -6.802456 | -2.401654 | 0.022506 |
| 2 | 7 | 0 | 0.257784 | 4.671646 | -0.012094 |
| 3 | 6 | 0 | 1.581913 | 4.336728 | 0.003430 |
| 4 | 6 | 0 | 1.989360 | 3.040348 | 0.012704 |
| 5 | 6 | 0 | 0.967740 | 2.032138 | 0.004486 |
| 6 | 7 | 0 | 1.287928 | 0.732947 | 0.009476 |
| 7 | 7 | 0 | -0.338412 | 2.365767 | -0.008485 |
| 8 | 6 | 0 | -0.722306 | 3.666306 | -0.016207 |
| 9 | 8 | 0 | -1.925199 | 4.014162 | -0.027374 |
| 10 | 6 | 0 | -0.178423 | 6.072027 | -0.023350 |
| 11 | 7 | 0 | -5.344765 | -2.436958 | 0.013756 |
| 12 | 6 | 0 | -4.534608 | -3.556965 | 0.011253 |
| 13 | 7 | 0 | -3.251319 | -3.274663 | 0.004150 |
| 14 | 6 | 0 | -3.207183 | -1.889335 | 0.001946 |
| 15 | 6 | 0 | -2.103819 | -0.985580 | -0.004208 |
| 16 | 8 | 0 | -0.884548 | -1.255396 | -0.009408 |
| 17 | 7 | 0 | -2.522721 | 0.353660 | -0.003806 |


| 18 | 6 | 0 | $-3.834622$ | 0.778649 | 0.002010 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 19 | 7 | 0 | -4.039982 | 2.110215 | 0.002829 |
| 20 | 7 | 0 | -4.869922 | -0.056254 | 0.007959 |
| 21 | 6 | 0 | -4.500676 | -1.354759 | 0.007875 |
| 22 | 6 | 0 | 6.658257 | 0.774831 | 0.029325 |
| 23 | 6 | 0 | 5.317333 | 0.066166 | 0.030283 |
| 24 | 8 | 0 | 4.264704 | 0.707064 | 0.055765 |
| 25 | 7 | 0 | 5.394918 | -1.304925 | -0.000139 |
| 26 | 6 | 0 | 2.173734 | -2.898015 | -0.026042 |
| 27 | 7 | 0 | 3.101083 | -1.921302 | -0.016056 |
| 28 | 6 | 0 | 4.384286 | -2.288958 | -0.009675 |
| 29 | 6 | 0 | 4.801279 | -3.634219 | -0.014164 |
| 30 | 6 | 0 | 3.829513 | -4.625922 | -0.025681 |
| 31 | 6 | 0 | 2.479848 | -4.256355 | -0.031481 |
| 32 | 1 | 0 | -7.162660 | -1.886202 | 0.915923 |
| 33 | 1 | 0 | 2.282127 | 5.163977 | 0.007908 |
| 34 | 1 | 0 | 3.037063 | 2.771945 | 0.025109 |
| 35 | 1 | 0 | 0.542307 | 0.033855 | 0.001944 |
| 36 | 1 | 0 | 2.260011 | 0.425825 | 0.017909 |
| 37 | 1 | 0 | -0.776389 | 6.271376 | -0.914823 |
| 38 | 1 | 0 | -4.960547 | -4.551157 | 0.014861 |
| 39 | 1 | 0 | -1.765507 | 1.064629 | -0.007029 |
| 40 | 1 | 0 | -3.269157 | 2.789997 | -0.009810 |
| 41 | 1 | 0 | -4.995369 | 2.432929 | 0.001356 |
| 42 | 1 | 0 | 7.516527 | 0.099984 | 0.021644 |
| 43 | 1 | 0 | 6.707932 | 1.423057 | -0.850342 |
| 44 | 1 | 0 | 6.334264 | $-1.678876$ | -0.011288 |
| 45 | 1 | 0 | 5.856990 | -3.886610 | -0.008482 |
| 46 | 1 | 0 | 4.120736 | -5.671685 | -0.029496 |
| 47 | 1 | 0 | 1.688842 | -4.998217 | -0.039992 |
| 48 | 1 | 0 | -7.173179 | -1.884130 | -0.865352 |
| 49 | 1 | 0 | -7.175037 | -3.426253 | 0.023510 |
| 50 | 1 | 0 | -0.783651 | 6.284105 | 0.860352 |
| 51 | 1 | 0 | 0.703867 | 6.711561 | -0.024457 |


| 52 | 1 | 0 | 6.714830 | 1.413304 | 0.915420 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 53 | 1 | 0 | 1.139180 | -2.563561 | -0.029923 |

$E($ RB3LYP $)=-1472.60730525$ Hartree

Geometry optimized base triple $\mathrm{V}^{*} \mathrm{C}$ - G

| Center <br> Number | Atomic <br> Number | Atomic Type | Coordinates (Angstroms) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | X | Y | Z |
| 1 | 6 | 0 | -3.798528 | -5.808714 | 0.042843 |
| 2 | 7 | 0 | -3.990482 | 4.193249 | 0.102896 |
| 3 | 6 | 0 | -2.824454 | 4.901166 | 0.092952 |
| 4 | 6 | 0 | -1.614332 | 4.282993 | 0.029930 |
| 5 | 6 | 0 | -1.614287 | 2.851226 | -0.025002 |
| 6 | 7 | 0 | -0.472419 | 2.155490 | -0.086564 |
| 7 | 7 | 0 | -2.767591 | 2.155034 | -0.017736 |
| 8 | 6 | 0 | -3.964521 | 2.789478 | 0.046257 |
| 9 | 8 | 0 | -5.053203 | 2.173445 | 0.057371 |
| 10 | 6 | 0 | -5.292725 | 4.865329 | 0.175712 |
| 11 | 7 | 0 | -2.779547 | -4.764536 | 0.001191 |
| 12 | 6 | 0 | -1.411514 | -4.936894 | -0.028612 |
| 13 | 7 | 0 | -0.745134 | -3.802638 | -0.063634 |
| 14 | 6 | 0 | -1.729147 | -2.827287 | -0.056241 |
| 15 | 6 | 0 | -1.640978 | -1.412578 | -0.083230 |
| 16 | 8 | 0 | -0.600721 | -0.711920 | -0.126782 |
| 17 | 7 | 0 | -2.888026 | -0.794002 | -0.056292 |


| 18 | 6 | 0 | -4.096846 | -1.459235 | -0.013476 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 19 | 7 | 0 | -5.202875 | -0.693427 | 0.002159 |
| 20 | 7 | 0 | -4.195460 | -2.786844 | 0.006365 |
| 21 | 6 | 0 | -2.999952 | -3.408612 | -0.015110 |
| 22 | 6 | 0 | 3.318162 | 4.307322 | -0.229479 |
| 23 | 6 | 0 | 3.513054 | 2.808032 | -0.133904 |
| 24 | 8 | 0 | 2.549159 | 2.039214 | -0.103661 |
| 25 | 7 | 0 | 4.829052 | 2.404435 | -0.082338 |
| 26 | 6 | 0 | 5.086542 | -1.158687 | 0.080188 |
| 27 | 7 | 0 | 4.563066 | 0.065131 | -0.006545 |
| 28 | 6 | 0 | 5.383323 | 1.117775 | 0.013023 |
| 29 | 6 | 0 | 6.776001 | 0.991364 | 0.118458 |
| 30 | 6 | 0 | 7.303829 | -0.295363 | 0.202889 |
| 31 | 6 | 0 | 6.461016 | -1.404819 | 0.185504 |
| 32 | 7 | 0 | 4.197450 | -2.248981 | 0.061100 |
| 33 | 6 | 0 | 2.830713 | -2.243262 | -0.004952 |
| 34 | 7 | 0 | 2.147196 | -1.113631 | -0.076491 |
| 35 | 7 | 0 | 2.206209 | -3.426906 | 0.002191 |
| 36 | 1 | 0 | -4.400004 | -5.710061 | 0.949109 |
| 37 | 1 | 0 | -2.921259 | 5.979555 | 0.138214 |
| 38 | 1 | 0 | -0.694978 | 4.853667 | 0.023409 |
| 39 | 1 | 0 | -0.513006 | 1.133866 | -0.110673 |
| 40 | 1 | 0 | 0.443107 | 2.590612 | -0.085785 |
| 41 | 1 | 0 | -5.893575 | 4.613821 | -0.700310 |
| 42 | 1 | 0 | -0.970435 | -5.923974 | -0.021619 |
| 43 | 1 | 0 | -2.877251 | 0.247223 | -0.057241 |
| 44 | 1 | 0 | -5.158991 | 0.332688 | 0.020166 |
| 45 | 1 | 0 | -6.094798 | -1.160719 | 0.060001 |
| 46 | 1 | 0 | 4.250961 | 4.872686 | -0.274621 |
| 47 | 1 | 0 | 2.726051 | 4.524600 | -1.122849 |
| 48 | 1 | 0 | 5.511452 | 3.150103 | -0.104283 |
| 49 | 1 | 0 | 7.418144 | 1.865044 | 0.134511 |
| 50 | 1 | 0 | 8.376765 | -0.434408 | 0.284541 |
| 51 | 1 | 0 | 6.851403 | -2.414104 | 0.252359 |


| 52 | 1 | 0 | 4.633389 | -3.159786 | 0.114545 |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 53 | 1 | 0 | 1.119251 | -1.120042 | -0.104159 |
| 54 | 1 | 0 | 1.177395 | -3.501821 | -0.030630 |
| 55 | 1 | 0 | 2.732354 | -4.284878 | 0.077221 |
| 56 | 1 | 0 | 2.661253 | -0.229085 | -0.080230 |
| 57 | 1 | 0 | -4.447807 | -5.733980 | -0.832158 |
| 58 | 1 | 0 | -3.302052 | -6.778943 | 0.042450 |
| 59 | 1 | 0 | -5.827887 | 4.548651 | 1.073019 |
| 60 | 1 | 0 | -5.126893 | 5.941681 | 0.209044 |
| 61 | 2.743340 | 4.639669 | 0.639919 |  |  |

$E($ RB3LYP $)=-1677.30121913$ Hartree





Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}, 18$ mins, $100 \% \mathrm{ACN}, 5$ mins, $0.2 \mathrm{~mL} / \mathrm{min}$.





Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, 120 Å $2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$. Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}, 18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.


FmocHN








Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.









Column: Cosmosil 2.5 $\mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.





Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), 18 mins, $0.2 \mathrm{~mL} / \mathrm{min}$.




Column: Cosmosil $2.5 \mathrm{C}_{18}-\mathrm{MS}$-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: 0-100\% ACN/ $\mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.





Column: Cosmosil $2.5 \mathrm{C}_{18}-\mathrm{MS}-\mathrm{II}, 120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.




Column: Cosmosil $2.5 \mathrm{C}_{18}-\mathrm{MS}-\mathrm{II}, 120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}(0.1 \%$ formic acid), 18 mins, $0.2 \mathrm{~mL} / \mathrm{min}$.





Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), 18 mins, $0.2 \mathrm{~mL} / \mathrm{min}$.




Column: Cosmosil $2.5 \mathrm{C}_{18}-\mathrm{MS}-\mathrm{II}, 120 \AA 8.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.





Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), 18 mins, $0.2 \mathrm{~mL} / \mathrm{min}$.




Column: Cosmosil $2.5 \mathrm{C}_{18}-\mathrm{MS}$-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.




Column: Cosmosil $2.5 \mathrm{C}_{18}$-MS-II, $120 \AA 2.5 \mu \mathrm{~m}, 2.0 \times 50 \mathrm{~mm}$.
Solvent gradient: $0-100 \% \mathrm{ACN} / \mathrm{H}_{2} \mathrm{O}$ ( $0.1 \%$ formic acid), $18 \mathrm{mins}, 0.2 \mathrm{~mL} / \mathrm{min}$.


[^0]:    ${ }^{1}$ Feagin, T. A.; Shah, N. I.; Heemstra, J. M. J. Nucleic Acids 2012, article ID 354549.

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