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Supporting Information for:

# Isolating elusive 'Al( $\mu$ -O)M' intermediates in CO<sub>2</sub> reduction by bimetallic

# Al–M complexes (M = Zn, Mg)

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#### **General Experimental Procedures**

All manipulations were performed under dry argon using standard Schlenk-line techniques, or in a conventional nitrogen-filled glovebox. Solvents were dried over appropriate drying agents and degassed prior to use. NMR spectra were recorded using a Jeol JNM-ECZ500S 500 MHz spectrometer equipped with a ROYAL digital auto tune probe S, operating at 500.1 (<sup>1</sup>H), 125.7 (<sup>13</sup>C), 194.4 (<sup>7</sup>Li) and 130.3 (<sup>27</sup>Al) MHz. Spectra were recorded at 298 K (unless stated otherwise) and proton and carbon chemical shifts were referenced internally to residual solvent resonances. Elemental analyses were performed by the EA Services Team at the London Metropolitan University, UK. Coupling constants are quoted in Hz. Al(NON<sup>Dipp</sup>)I<sup>S1</sup> and Mg(BDI<sup>Mes</sup>)I(OEt<sub>2</sub>)<sup>S2</sup> were prepared according to the literature procedures. [Li(THF)<sub>2</sub>][Zn(BDI<sup>Mes</sup>)Cl<sub>2</sub>] was made according to the procedure for [Li(Et<sub>2</sub>O)<sub>2</sub>][Zn(BDI<sup>Dipp</sup>)Cl<sub>2</sub>]<sup>S3</sup> using BDI<sup>Mes</sup>H in THF solvent.

### Synthesis of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (1-Zn)

A bright yellow toluene solution of  $[K{Al(NON^{Dipp})}]_2$  (644 mg, 0.59 mmol) was added to  $[Li(THF)_2][Zn(BDI^{Mes})Cl_2]$  (729 mg, 1.17 mmol) in a Schlenk flask to give a dark brown/black suspension. The resulting solution was stirred for 18 hours after which time the volatiles were removed. The product was extracted in hexane and then filtered through celite to give a colourless solution. Crystallisation was achieved from the storage of the hexane solution at – 30 °C for 18 hours. Yield 650 mg, 61 %.

Elemental Analysis for  $C_{58}H_{83}AIN_4OSi_2Zn$  (1000.78)\*. Calculated: C, 69.60; H, 8.36; N, 5.60. Found C, 69.92; H, 8.83; N, 5.09. \* = toluene solvated sample

<sup>1</sup>H NMR (500 MHz,  $C_7D_8$ , 293 K):  $\delta$  7.07-6.98\* (m, 6H,  $C_6H_3$ ), 6.68 (s, 2H,  $C_6H_2$ ), 4.81 (s, 1H, BDI  $\gamma$ -CH), 3.69 (sept, J = 6.5, 4H, CHMe<sub>2</sub>), 2.26 (s, 6H, p-Me), 1.96 (br s, 6H, o-Me), 1.31 (s, 6H, BDI-Me), 1.27 (br m, 12H, o-Me and CHMe<sub>2</sub>), 1.24 (br d, 12H, CHMe<sub>2</sub>), 0.68 (br s, 6H, CHMe<sub>2</sub>), 0.46, -0.17 (br s, 6H, SiMe<sub>2</sub>). \*overlap with  $C_7D_8$ .

<sup>1</sup>H NMR (500 MHz,  $C_7D_8$ , 213 K):  $\delta$  7.07-7.00\* (m, 6H,  $C_6H_3$ ), 6.69 (s, 2H,  $C_6H_2$ ), 4.82 (s, 1H, BDI  $\gamma$ -CH), 3.79, 3.72 (br sept, 2H, CHMe<sub>2</sub>), 2.29 (s, 6H, *p*-Me), 2.02 (br d, 6H, CHMe<sub>2</sub>), 1.36 (s, 6H, BDI-Me), 1,34\* (br m, 12H, *o*-Me and CHMe<sub>2</sub>), 1.27 (br d, 12H, CHMe<sub>2</sub>), 0.62 (br d CHMe<sub>2</sub>), 0.57, -0.06 (s, 6H, SiMe<sub>2</sub>).

<sup>13</sup>C{<sup>1</sup>H} NMR (151 MHz, C<sub>6</sub>D<sub>6</sub>): δ 166.9 (s, 1H, BDI-CMe), 146.4, 142.8, 137.9, 133.5, 129.3, 128.6, 125.7, 124.0, 124.0 ( $C_6$ H<sub>3</sub>,  $C_6$ H<sub>2</sub>), 97.3 (BDI γ-CH), 27.7 (CHMe<sub>2</sub>), 26.5, 23.5 (CHMe<sub>2</sub>), 21.4 (*o*-Me), 20.9 (BDI-Me), 18.6 (*p*-Me), 4.2, 0.7 (SiMe<sub>2</sub>).

**Figure S1** <sup>1</sup>H NMR spectrum (500 MHz, C<sub>7</sub>D<sub>8</sub>, 293 K) of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (**1-Zn**).



**Figure S2** <sup>1</sup>H NMR spectrum (500 MHz, C<sub>7</sub>D<sub>8</sub>, 213 K) of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (**1-Zn**).





# **Figure S3** Stacked <sup>1</sup>H NMR spectra (500 MHz, C<sub>7</sub>D<sub>8</sub>) of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (**1-Zn**).









#### Synthesis of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (1-Mg)

A bright yellow toluene solution of  $[K{Al(NON^{Dipp})}]_2$  (317 mg, 0.58 mmol) was added to a white suspension of Mg(BDI<sup>Mes</sup>)I(OEt<sub>2</sub>) (323 mg, 0.58 mmol) in toluene (~ 5 mL) and stirred for 18 hours. The resulting suspension was filtered through celite to give a pale orange solution and the volatiles reduced *in vacuo* (~ 2 mL). The product separated from solution upon standing after ~ 0.5 hr as a white solid. The supernatant was decanted from the white powder and transferred to a scintillation vial. A second crop of the product was obtained as colourless crystals from the storage of the supernatant at -30 °C for 18 hours. Yield 298 mg, 59 %.

Elemental Analysis for  $C_{51}H_{75}AIMgN_4OSi_2$  (867.55). Calculated: C, 70.60; H, 8.71; N, 6.46. Found C, 69.19; H, 9.09; N, 6.04.

<sup>1</sup>H NMR (500 MHz,  $C_7D_8$ , 293 K):  $\delta$  7.02 – 6.98\* (m, 6H,  $C_6H_3$ ), 6.69 (s, 4H,  $C_6H_2$ ), 4.72 (s, 1H, BDI  $\gamma$ -CH), 3.75 (sept, J = 6.8, 4H, CHMe<sub>2</sub>), 2.26 (s, 6H, p-Me), 1.69 (s, 12H, o-Me), 1.31 (s, 6H, BDI-*Me*), 1.26, 0.93 (d, J = 6.8, 12H, CH*Me*<sub>2</sub>), 0.19 (br s, 12H, Si*Me*<sub>2</sub>). \*overlap with  $C_7D_8$ .

<sup>1</sup>H NMR (500 MHz,  $C_7D_8$ , 213 K):  $\delta$  7.07 – 7.00\* (m, 6H,  $C_6H_3$ ), 6.70 (s, 4H,  $C_6H_2$ ), 4.73 (s, 1H, BDI  $\gamma$ -CH), 3.82 (br sept, 4H, CHMe<sub>2</sub>), 2.29 (s, 6H, *p*-Me), 1.98, 1.43 (s, 6H, *o*-Me), 1.36 (br d, 6H, CH*Me*<sub>2</sub>), 1.31 (s, 6H, BDI-*Me*), 1.27, 1.23, 0.68 (br d, 6H, CH*Me*<sub>2</sub>), 0.59, 0.01 (br s, 6H, Si*Me*<sub>2</sub>). \*overlap with  $C_7D_8$ .

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>): δ 169.3 (BDI-CMe), 145.9, 144.6, 143.2, 133.4, 131.4, 130.0, 123.7, 123.6 (C<sub>6</sub>H<sub>3</sub>, C<sub>6</sub>H<sub>2</sub>), 96.5 (BDI γ-CH), 27.4 (CHMe<sub>2</sub>), 26.6, 23.7 (CH*Me*<sub>2</sub>), 23.3 (BDI-*Me*), 20.9 (*o*-*Me*), 18.5 (*p*-*Me*), 2.4 (Si*Me*<sub>2</sub>).





**Figure S7** <sup>1</sup>H NMR spectrum (500 MHz, C<sub>7</sub>D<sub>8</sub>, 213 K) of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (**1-Mg**).





Figure S8 Stacked <sup>1</sup>H NMR spectra (500 MHz, C<sub>7</sub>D<sub>8</sub>) of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (1-Mg).



Figure S9  ${}^{13}C{}^{1}H$  NMR spectrum (126 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (1-Mg).



**Figure S10** Thermal displacement plot (30 % probability) of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (**1-Mg**).

# Synthesis of (NON<sup>Dipp</sup>)Al( $\mu$ -O<sub>2</sub>C)Zn(BDI<sup>Mes</sup>) (2)

A solution of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (**1-Zn**, 160 mg, 0.18 mmol) in C<sub>6</sub>D<sub>6</sub> was transferred to a J. Youngs NMR tube. The solution was degassed and <sup>13</sup>CO<sub>2</sub> (~1 atm.) was administered to the reaction vessel (condensed from lecture bottle at –196 °C). Reaction progress was monitored by <sup>13</sup>C NMR spectroscopy and shown to be complete after 2 days at room temperature. The solvent was removed *in vacuo* and the residue dissolved in toluene. Crystallisation was achieved from a toluene solution stored at –30 °C for 18 hours. Yield 100 mg, 60 %. Elemental Analysis for C<sub>52</sub>H<sub>75</sub>AlN<sub>4</sub>O<sub>3</sub>Si<sub>2</sub>Zn (*952.66*). Calculated: C, 65.55; H, 7.93; N, 5.88. Found C, 64.60; H, 8.23; N, 5.59.

<sup>1</sup>H NMR (600 MHz,  $C_6D_6$ ) :  $\delta$  7.02 – 6.99 (m, 6H,  $C_6H_3$ ), 6.73 (s, 4H,  $C_6H_2$ ), 4.83 (s, 1H, BDI  $\gamma$ -CH), 3.84 (sept, J = 6.8, 4H, CHMe<sub>2</sub>), 2.28 (s, 6H, p-Me), 1.81 (s, 12H, o-Me), 1.44 (s, 6H, BDI-Me), 1.34 (d, J = 6.8, 12H, CHMe<sub>2</sub>), 1.09 (d, J = 6.8, 12H, CHMe<sub>2</sub>), 0.38 (s, 12H, SiMe<sub>2</sub>).

<sup>13</sup>C{<sup>1</sup>H} NMR (151 MHz, C<sub>6</sub>D<sub>6</sub>): 222.2 (<sup>13</sup>CO<sub>2</sub>), 169.1 (BDI–CMe), 146.2, 143.3, 140.6, 134.3, 130.2, 129.9, 123.7, 123.4 (C<sub>6</sub>H<sub>3</sub>, C<sub>6</sub>H<sub>2</sub>), 96.6 (BDI γ-CH), 27.7 (CHMe<sub>2</sub>), 25.5, 24.8 (CH*Me*<sub>2</sub>), 22.3 (BDI-*Me*), 21.2 (*p*-Me), 18.2 (*o*-Me), 2.3 (Si*Me*<sub>2</sub>).

**Figure S11** <sup>1</sup>H NMR spectrum (500 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -O<sub>2</sub>C)Zn(BDI<sup>Mes</sup>) (2).



**Figure S12** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of (NON<sup>Dipp</sup>)Al( $\mu$ -O<sub>2</sub>C)Zn(BDI<sup>Mes</sup>) (2).





**Figure S13** Thermal displacement plot (30 % probability) of (NON<sup>Dipp</sup>)Al–Zn(BDI<sup>Mes</sup>) (2).

# Synthesis of $\{(NON^{Dipp})AI(\mu-CO_3)\}_2\{Mg(THF)_4\}$ (3)

A solution of  $(NON^{Dipp})AI-Mg(BDI^{Mes})$  (107 mg, 0.12 mmol) was prepared in toluene-D<sub>8</sub> and transferred to a J. Youngs tap NMR tube. The solution was degassed and <sup>13</sup>CO<sub>2</sub> (~1 atm) was administered to the reaction vessel (condensed from lecture bottle at -196 °C). An insoluble white precipitate formed upon warming the solution to room temperature. The solvent was removed *in vacuo* and the white powder dissolved in THF. Crystallisation was achieved from a THF solution stored at -30 °C for 18 hours. Yield 81 mg, 81%.

Elemental Analysis for C<sub>74</sub>H<sub>124</sub>Al<sub>2</sub>MgN<sub>4</sub>O<sub>12</sub>Si<sub>4</sub> (*1452.28*). Calculated: C, 61.19; H, 8.61; N, 3.86. Found C, 60.60; H, 8.80; N, 3.73.

The observed data are a better fit for the formula calculated for loss of one equivalent of THF,  $C_{70}H_{116}AI_2MgN_4O_{11}Si_4$  (1380.81). Calculated: C, 60.91; H, 8.47; N, 4.06. We therefore suggest the results indicate partial loss of THF during sample preparation and analysis.

<sup>1</sup>H NMR (600 MHz, THF-D<sub>8</sub>):  $\delta$  6.90 (d, J = 7.8, 4H, C<sub>6</sub>H<sub>3</sub>), 6.83 (dd, J = 8.2, 6.8, 2H, C<sub>6</sub>H<sub>3</sub>), 3.77 (sept, J = 6.8, 4H, CHMe<sub>2</sub>), 3.62 (s, 12H, THF), 1.78 (s, 12H, THF), 1.17 (d, J = 6.8, 6H, CHMe<sub>2</sub>), 1.16 (d, J = 6.8, 6H, CHMe<sub>2</sub>), 1.12 (d, J = 6.8, 12H, CHMe<sub>2</sub>), 0.11 (s, 12H, SiMe<sub>2</sub>).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, THF-D<sub>8</sub>):  $\delta$  164.1 (<sup>13</sup>CO<sub>2</sub>), 147.1, 142.5, 125.9, 123.8, 123.6, 123.5 ( $C_6H_3$ ), 68.3 (THF), 28.3 (CHMe<sub>2</sub>), 26.4 (THF), 25.9, 25.0, 24.2 (CHMe<sub>2</sub>), 2.5 (SiMe<sub>2</sub>).

NMR reactions were carried out in  $THF-D_8$  to prevent precipitation of the product and to provide an insight into the reaction mechanism.

# (NON<sup>Dipp</sup>)AI-Mg(BDI<sup>Mes</sup>) + <sup>13</sup>CO<sub>2</sub>:

Using the general method detailed above. [(NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>)] (38 mg, 0.04 mmol). Yield 28 mg, 79 %.

# (NON<sup>Dipp</sup>)Al=O-Mg(BDI<sup>Mes</sup>) + <sup>13</sup>CO<sub>2</sub>:

Using the ge	eneral method	detailed above	e. [(NON <sup>Dipp</sup>	)AI=O-Mg(BDI <sup>Mes</sup> )]	(62 mg, 0.07	' mmol).
Yield	55		mg,	97		%.

**Figure S14** <sup>1</sup>H NMR spectrum (500 MHz,  $D_8$ -THF) of {(NON<sup>Dipp</sup>)Al( $\mu$ -CO<sub>3</sub>)}<sub>2</sub>{Mg(THF)<sub>4</sub>} (**3**).



**Figure S15** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum (126 MHz,  $D_8$ -THF) of {(NON<sup>Dipp</sup>)Al( $\mu$ -CO<sub>3</sub>)}<sub>2</sub>{Mg(THF)<sub>4</sub>} (**3**).







**Figure S17**  ${}^{13}C{}^{1H}$  NMR spectra (126 MHz, D<sub>8</sub>-THF) of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) +  ${}^{13}CO_2$ .



**Figure S18** <sup>1</sup>H NMR spectra (500 MHz,  $D_8$ -THF) of (NON<sup>Dipp</sup>)Al–( $\mu$ -O)Mg(BDI<sup>Mes</sup>) + <sup>13</sup>CO<sub>2</sub>.



**Figure S19** <sup>13</sup>C{<sup>1</sup>H} NMR spectra (126 MHz,  $D_8$ -THF) of (NON<sup>Dipp</sup>)Al–( $\mu$ -O)Mg(BDI<sup>Mes</sup>) + <sup>13</sup>CO<sub>2</sub>.





**Figure S20** Thermal displacement plot (30 % probability) of  $\{(NON^{Dipp})Al(\mu-CO_3)\}_2\{Mg(THF)_4\}$  (3).

### Synthesis of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Zn(BDI<sup>Mes</sup>) (4-Zn)

A solution of  $(NON^{Dipp})AI-Zn(BDI^{Mes})$  (206 mg, 0.23 mmol) in toluene (ca. 5 mL) was transferred to an ampule. The solution was degassed and nitrous oxide (~1 atm.) was administered to the reaction vessel. The mixture was allowed to stir at room temperature for 72 hours to give a pale-yellow solution. The solvent was reduced *in vacuo* (ca. 2 mL) and transferred to a vial. Crystallisation was achieved from a toluene solution stored at -30 °C for 18 hours. Yield 87 mg, 42 %.

Accurate elemental analysis could not be obtained for this compound, likely due to high oxygen and moisture sensitivity and possible decomposition during sample preparation and analysis.

<sup>1</sup>H NMR (500 MHz,  $C_6D_6$ ):  $\delta$  7.09 – 7.03 (m, 6H,  $C_6H_3$ ), 6.73 (s, 4H), 4.70 (s, 1H), 3.81 (sept,  $J = 6.9, 4H, CHMe_2$ ), 2.29 (s, 6H, *p-Me*), 1.73 (s, 12H, *o-Me*), 1.37 (s, 6H, BDI-*Me*), 1.32 (d,  $J = 6.8, 12H, CHMe_2$ ), 0.96 (d,  $J = 6.8, 12H, CHMe_2$ ), 0.25 (s, 12H, Si $Me_2$ ).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  169.3 (BDI-CMe), 145.2, 144.3, 142.4, 133.6, 131.2, 129.8, 129.3, 128.6, 123.8, 123.4 (C<sub>6</sub>H<sub>3</sub>, C<sub>6</sub>H<sub>2</sub>), 95.2 (BDI-CH), 28.1 (CHMe<sub>2</sub>), 25.0, 23.0 (CH*Me*<sub>2</sub>), 22.8 (BDI-C*Me*), 21.0 (*p*-*Me*), 18.2 (*o*-*Me*), 1.7 (Si*Me*<sub>2</sub>).

**Figure S21** <sup>1</sup>H NMR spectrum (500 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Zn(BDI<sup>Mes</sup>) (**4-Zn**).



**Figure S22** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum (126 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Zn(BDI<sup>Mes</sup>) (**4-Zn**).







#### Synthesis of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Mg(BDI<sup>Mes</sup>) (4-Mg)

A solution of (NON<sup>Dipp</sup>)Al–Mg(BDI<sup>Mes</sup>) (222 mg, 0.26 mmol) in toluene (ca. 5 mL) was transferred to an ampule. The solution was degassed and nitrous oxide (~1 atm.) was administered to the reaction vessel. The mixture was allowed to stir at room temperature for 72 hours to give a yellow solution. The solvent was reduced *in vacuo* (ca. 2 mL) and transferred to a scintillation vial. Crystallisation was achieved from a toluene solution stored at –30 °C for 18 hours. Yield 103 mg, 46%.

Elemental Analysis for C<sub>51</sub>H<sub>75</sub>AlMgN<sub>4</sub>O<sub>2</sub>Si<sub>2</sub> (*883.55*). Calculated: C, 69.32; H, 8.55; N, 6.34. Found C, 68.61; H, 8.74; N, 5.88.

<sup>1</sup>H NMR (500 MHz,  $C_7D_8$ ):  $\delta$  7.04 – 6.97 (m, 6H,  $C_6H_3$ ), 6.71 (s, 4H,  $C_6H_2$ ), 4.68 (s, 1H, BDI-CH), 3.77 (sept, J = 6.8, 4H, CHMe<sub>2</sub>), 2.28 (s, 6H, *p*-Me), 1.71 (s, 12H, *o*-Me), 1.38 (s, 6H, BDI-Me), 1.30 (d, J = 6.8, 12H, CHMe<sub>2</sub>), 0.92 (d, J = 6.8, 12H, CHMe<sub>2</sub>), 0.19 (s, 12H, SiMe<sub>2</sub>).

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz,  $C_7D_8$ ):  $\delta$  170.1 (BDI-CMe), 145.2 ( $C_6H_3$ ), 144.1 ( $C_6H_2$ ), 142.3 ( $C_6H_3$ ), 133.2, 130.9, 129.8 ( $C_6H_2$ ), 123.8, 123.2 ( $C_6H_3$ ), 95.3 (BDI-CH), 28.0 (CHMe<sub>2</sub>), 25.1, 23.0 (CHMe<sub>2</sub>), 22.9 (BDI-Me), 20.9 (*p*-Me), 18.2 (*o*-Me), 1.6 (SiMe<sub>2</sub>).

**Figure S24** <sup>1</sup>H NMR spectrum (500 MHz,  $C_7D_8$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Mg(BDI<sup>Mes</sup>) (**4-Mg**).



**Figure S25** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum (126 MHz,  $C_7D_8$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -O)Mg(BDI<sup>Mes</sup>) (**4-Mg**).







# Synthesis of (NON<sup>Dipp</sup>)Al(µ-CO<sub>3</sub>)Zn(BDI<sup>Mes</sup>) (5)

A solution of  $(NON^{Dipp})Al(\mu-O)Zn(BDI^{Mes})]$  (56 mg, 0.06 mmol) in toluene-D<sub>8</sub> was transferred to a J. Youngs NMR tube. The solution was degassed and <sup>13</sup>CO<sub>2</sub> (~1 atm.) was administered to the reaction vessel (condensed from lecture bottle at –196 °C). The solvent was removed *in vacuo* and the resulting white residue dissolved in hexane. Crystallisation was achieved by slow evaporation from a hexane solution stored at room temperature. Yield 52 mg, 89%. Accurate elemental analysis could not be obtained for this compound, likely due to high

analysis.

<sup>1</sup>H NMR (500 MHz,  $C_6D_6$ ):  $\delta$  7.11 (s, 6H,  $C_6H_3$ ), 6.73 (s, 4H,  $C_6H_2$ ), 4.70 (s, 1H, BDI-CH), 4.01 (hept, J = 6.8 Hz, 4H, CHMe<sub>2</sub>), 2.25 (s, 6H, *p*-Me), 1.75 (s, 12H, *o*-Me), 1.39 (d, J = 6.8 Hz, 12H, CHMe<sub>2</sub>), 1.37 (s, 6H, BDI-Me), 1.10 (d, J = 6.8 Hz, 12H, CHMe<sub>2</sub>), 0.38 (s, 12H, SiMe<sub>2</sub>).

oxygen and moisture sensitivity and possible decomposition during sample preparation and

<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  170.9 (BDI-CMe), 168.4 (<sup>13</sup>CO<sub>2</sub>), 167.0, 146.6, 142.5, 141.3, 134.9, 131.0, 129.9, 124.7, 123.8, 123.7 (C<sub>6</sub>H<sub>3</sub>, C<sub>6</sub>H<sub>2</sub>), 96.2 (BDI-CH), 27.9 (CHMe<sub>2</sub>), 25.8, 24.8 (CH*Me*<sub>2</sub>), 22.8 (BDI-*Me*), 21.0 (*p*-*Me*), 18.1 (*o*-*Me*), 2.2 (Si*Me*<sub>2</sub>).

**Figure S27** <sup>1</sup>H NMR spectrum (500 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -CO<sub>3</sub>)Zn(BDI<sup>Mes</sup>) (5).



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**Figure S28** <sup>13</sup>C{<sup>1</sup>H} NMR spectrum (126 MHz,  $C_6D_6$ ) of (NON<sup>Dipp</sup>)Al( $\mu$ -CO<sub>3</sub>)Zn(BDI<sup>Mes</sup>) (5).





**Figure S29** Thermal displacement plot (30 % probability) of (NON<sup>Dipp</sup>)Al( $\mu$ -CO<sub>3</sub>)Zn(BDI<sup>Mes</sup>) (5).

# **Computational Details**

DFT calculations were run with Gaussian 16 (C.01).<sup>54</sup> The Al, Si, P, Mg and Zn centres were described with the Stuttgart RECPs and associated basis sets,<sup>55</sup> and the 6-31G\*\* basis set was used for all other atoms (BS1).<sup>56</sup> A polarization function was also added to Al ( $\zeta d = 0.190$ ), Si ( $\zeta d = 0.284$ ) and P ( $\zeta d = 0.387$ ).<sup>57</sup> Initial BP86 optimizations<sup>58</sup> were performed using the 'grid = ultrafine' option, with all stationary points being fully characterized via analytical frequency calculations as minima or transition states (all positive eigenvalues or one imaginary eigenvalue respectively). All energies were recomputed with a larger basis set featuring 6-311++G\*\* basis sets on all atoms. Corrections for the effect of benzene ( $\varepsilon = 2.2706$ ) solvent were employed using the polarizable continuum model and BS1.<sup>59</sup> Single-point dispersion corrections to the BP86 results employed Grimme's D3 parameter set with Becke-Johnson damping as implemented in Gaussian.<sup>510</sup>

The Quantum Theory of Atoms in Molecules (QTAIM, AIMALL program)<sup>S11</sup> and Natural Bonding Orbital (NBO7<sup>S12</sup>) analyses were performed on the BP86-optmised geometries of **1**-Mg, **4-Mg**, **1-Zn**, **4-Zn**, **[Al(NC<sub>2</sub>N<sup>Dipp</sup>)]-Mg**, **[Al(**<sub>xanth</sub>**NON**<sup>Dipp</sup>)]-Mg, **[Al(**<sub>xanth</sub>**NON**<sup>Dipp</sup>)]-Zn and **[Al(**<sub>xanth</sub>**NON**<sup>Dipp</sup>)]-Li. The QTAIM topological analyses used wavefunction files obtained at the BP86/6-311++G<sup>\*\*</sup> level, whilst NBO analyses were carried out with NBO 7 within Gaussian 16 (C.01) at the same methodology level as the QTAIM calculations. Contour plots were generated in the AIMStudio package, using critical point (CP) visualisation threshold values of 0.025 *e* bohr<sup>-3</sup> (solid line BCP = strong) and 0.02 *e* bohr<sup>-3</sup> (dashed line BCP = weak).

# Table S1 Computed QTAIM BCP data, interatomic distances (in Å) and WBIs for 1-Mg, 1-Zn, 4-Mg, 4-Zn, [Al(NC<sub>2</sub>N<sup>Dipp</sup>)]–Mg, [Al(xanthNON<sup>Dipp</sup>)]–Mg, [Al(xanthNON<sup>Dipp</sup>)]–Ii. Mg, [Al(xanthNON<sup>Dipp</sup>)]–Zn and [Al(xanthNON<sup>Dipp</sup>)]–Li.

Asterisk denotes BCP is under the threshold of 0.02 for visualization in the corresponding Laplacian contour plot.

Complex	ВСР	ρ(r)	∇²ρ(r)	ε	G(r)	V(r)	H(r)	DI(A B)	R <sub>AB</sub>	WBI
1-Mg	Al3 - Mg81	0.03083	0.028309	0.04386	0.013605	-0.02013	-0.00653	0.381463	2.80	0.6015
4 Ма	Al3 - O6	0.104281	0.874522	0.023981	0.218567	-0.2185	0.000064	0.428464	1.69	0.4255
4-1VIg	Mg4 - 06	0.065697	0.560808	0.009136	0.125653	-0.1111	0.014549	0.270464	1.84	0.0749
[AI(NC <sub>2</sub> N <sup>Dipp</sup> )]–Mg	Al1 - Mg86	0.029404	0.023232	0.003636	0.012067	-0.01833	-0.00626	0.377625	2.85	0.6589
[Al( <sub>Xanth</sub> NON <sup>Dipp</sup> )]–Mg	Al1 - Mg114	0.032735	0.033421	0.003409	0.015283	-0.02221	-0.00693	0.386029	2.76	0.6268
1-Zn	Zn1 - Al3	0.054633	-0.06299	0.120399	0.008707	-0.03316	-0.02446	0.705544	2.52	0.7401
4.7n	Al4 - 06	0.101311	0.858249	0.027501	0.213088	-0.21161	0.001475	0.403974	1.70	0.3715
4-211	Zn1 - 06	0.115768	0.659099	0.017047	0.184404	-0.20403	-0.01963	0.71041	1.82	0.1988
[Al( <sub>Xanth</sub> NON <sup>Dipp</sup> )]–Zn	Zn1 - Al2	0.05736	-0.05187	0.119682	0.014576	-0.04212	-0.02755	0.637282	2.48	0.7442
[AI( <sub>Xanth</sub> NON <sup>Dipp</sup> )]–Li	Al1 - Li144*	0.017811	0.032456	0.045272	-0.01102	0.009567	-0.00145	0.109863	2.78	0.3097

Complex	Atom	Vol(r)	Loc(r)	q(A) <sub>AIM</sub>	q(A) <sub>NBO7</sub>	q(A) <sub>NB07</sub> <sup>†</sup>
1.04a	Al3	182.3954	92.29855	0.989647	0.98503	
T-IAIB	Mg81	77.70651	95.4012	1.513973	1.38126	
	Al3	40.303809	94.014683	2.387243	2.21056	
4-Mg	06	187.46068	93.238026	-1.63577	-1.69546	
	Mg4	53.42904	96.38614	1.699277	1.81141	
	Al1	181.83053	92.216446	1.001614	1.13864	
	Mg86	79.430356	95.476229	1.517331	1.27820	
	Al1	175.28208	92.398796	1.055461	0.99398	0.78822
	Mg114	71.942566	95.396063	1.513374	1.35405	1.48422
1.7n	Al3	85.203333	91.765124	1.812006	1.28794	
1-211	Zn1	185.05725	96.114087	0.331426	1.00385	
	Al4	39.965208	94.026105	2.388986	2.23603	
4-Zn	06	162.67005	91.140775	-1.46555	-1.64700	
	Zn1	114.48646	96.31231	1.176645	1.64002	
	Al2	73.27507	92.407916	1.987558	1.26191	0.95981
	Zn1	177.25782	96.152346	0.226972	1.00110	1.26128
	Al1	208.35093	93.204108	0.929222	0.79477	0.68705
	Li144	30.512036	92.875442	0.878518	0.71647	0.72517

Table S2Selected QTAIM and NBO atomic data for 1-Mg, 1-Zn, 4-Mg, 4-Zn, [Al(NC2NDipp)]-Mg, [Al(xanthNONDDipp)]-Mg, [Al(xanthNONDDipp)]-Zn and [Al(xanthNONDDipp)]-Li.

<sup>+</sup> Data taken from reference 11: M. M. D. Roy, J. Hicks, P. Vasko, A. Heilmann, A.-M. Baston, J. M. Goicoechea and S. Aldridge, *Angew. Chem. Int. Ed.*, 2021, **60**, 22301-22306.

# Table S3NBO atomic charges for[Al(xanthNON<sup>Dipp</sup>)]-Mg, [Al(xanthNON<sup>Dipp</sup>)]-Zn and[Al(xanthNON<sup>Dipp</sup>)]-Li at different levels of theory.

Complex	q(A) <sub>NBO7</sub>			
OPT I	METHOD =	BP	PBE0/SVP <sup>†</sup>	
SP I	BP86/BS2	BP86/6-31G**	PBE0/SVP <sup>†</sup>	
	Al1	0.99398	0.80498	0.78822
[AI( <sub>Xanth</sub> INON <sup>-,pp</sup> )]-IVIg	Mg114	1.35405	1.40365	1.48422
	Al2	1.26191	0.97654	0.95981
	Zn1	1.00110	1.17592	1.26128
	Al1	0.79477	0.65654	0.68705
	Li144	0.71647	0.73735	0.72517

<sup>+</sup> Data taken from reference 11: M. M. D. Roy, J. Hicks, P. Vasko, A. Heilmann, A.-M. Baston, J. M. Goicoechea and S. Aldridge, *Angew. Chem. Int. Ed.*, 2021, **60**, 22301-22306.

**Figure S30** QTAIM contour plot of the Laplacian ( $\nabla^2 \rho(r)$ ) of **1-Mg**. (a) complete molecule; (b) focus on the Al-Mg motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S31** QTAIM contour plot of the Laplacian ( $\nabla^2 \rho(r)$ ) of **4-Mg**. (a) complete molecule; (b) focus on the Al-O-Mg motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S32** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of **[Al(NC<sub>2</sub>N<sup>Dipp</sup>)]–Mg**. (a) complete molecule; (b) the Al-Mg motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S33** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of  $[Al(_{xanth}NON^{Dipp})]-Mg$ . (a) complete molecule; (b) focus on the Al-Mg motif). Bond critical points (BCPs) shown as small green spheres.





(b)



**Figure S34** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of **1-Zn**. (a) complete molecule; (b) focus on the Al-Zn motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S35** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of **4-Zn**. (a) complete molecule; (b) focus on the Al-O-Zn motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S36** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of  $[Al(x_{anth}NON^{Dipp})]$ –Zn. (a) complete molecule; (b) focus on the Al-Zn motif. Bond critical points (BCPs) shown as small green spheres.



(b)



**Figure S37** QTAIM contour plot of the Laplacian  $(\nabla^2 \rho(r))$  of **[Al(<sub>Xanth</sub>NON<sup>Dipp</sup>)]–Li**. (a) complete molecule; (b) focus on the Al-Li motif. Bond critical points (BCPs) shown as small green spheres.





(b)



Table S4Central bonding NLMO compositions in 1-Mg, 1-Zn, [Al(NC2NDipp)]-Mg,<br/>[Al(xanthNONDipp)]-Mg, [Al(xanthNONDipp)]-Zn and [Al(xanthNONDipp)]-Li.

Complex	Bond	Atom	Contribution	Contribution to NLMO in % ( <i>s, p, d</i> character)			
			overall	s orbital	p orbital	d orbital	
1 Ma		Al	75.84	78.24	21.69	0.07	
T-IAIR	AI-IVIg	Mg	24.16	94.18	4.89	0.93	
		AI	70.09	85.42	14.51	0.07	
	AI-IVIg	Mg	29.91	93.73	5.42	0.85	
	Al-Mg	AI	73.63	76.33	23.59	0.09	
		Mg	26.37	96.42	3.16	0.41	
1.7n	AL 70	AI	57.27	81.90	18.02	0.09	
1-211	AI-ZII	Zn	42.73	95.52	3.40	1.08	
	Al 7n	Al	56.61	77.49	22.41	0.10	
	AI-ZII	Zn	43.39	96.41	2.57	1.02	
	AL 11	AI	n/a	n/a	n/a	n/a	
	Al-Li	Li	n/a	n/a	n/a	n/a	

Note: for  $[Al(_{xanth}NON^{Dipp})]-Li$ , no formal Al-Li bond was identified in the NBO search, only a strong donor-acceptor interaction in the second-order perturbation energy analysis, where  $\Delta E[LP_{Al} \rightarrow nLi^*] = 42.1$  kcal/mol.

**Table S5**Selected donor acceptor interaction energies (in kcal mol<sup>-1</sup>),  $\Delta E^{(2)}$ , for **4-Mg** and**4-Zn** resulting from second-order perturbation energy analysis of the Fock<br/>matrix in NBO basis.

	4-Mg			4-Zn	
Donor NBO	Acceptor NBO	ΔE <sup>(2)</sup>	Donor NBO	Acceptor	ΔE <sup>(2)</sup>
(Unit)	(Unit)	(kcal/mol)	(Unit)	NBO (Unit)	(kcal/mol)
LP <sub>06</sub> (1)	n <sub>Al4</sub> * (2)	37.7	LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	50.9
LP <sub>06</sub> (1)	n <sub>Al4</sub> * (2)	20.8	LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	26.1
LP <sub>06</sub> (1)	n <sub>Al4</sub> * (2)	8.1	LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	8.3
LP <sub>06</sub> (1)	n <sub>Al4</sub> * (2)	6.0	LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	7.6
LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	5.4	LP <sub>06</sub> (1)	n <sub>A/3</sub> * (2)	6.7
LP <sub>06</sub> (1)	<i>n<sub>Zn1</sub></i> * (3)	49.3	LP <sub>06</sub> (1)	n <sub>Mg4</sub> * (3)	16.2
LP <sub>06</sub> (1)	n <sub>Zn1</sub> * (3)	9.0			

**Figure S38** NBOs of **4-Mg** involved in the donor acceptor interactions as out lined in Table S5. Arene substituents on both the [AI] and [Mg] complexes, and all hydrogen atoms have been removed for clarity. Al = pink, Mg = orange.



**Figure S39** NBOs of **4-Zn** involved in the donor acceptor interactions as out lined in Table S5. Arene substituents on both the [AI] and [Zn] complexes, and all hydrogen atoms have been removed for clarity. Al = pink, Zn = blue.



# **Crystallographic Details**

Crystals were covered in inert oil and suitable single crystals were selected under a microscope and mounted on an Agilent SuperNova diffractometer fitted with an EOS S2 detector. Data were collected at 120 K or 150 K using focused microsource Cu K $\alpha$  radiation at 1.54184 Å. Intensities were corrected for Lorentz and polarisation effects and for absorption using multi-scan methods.<sup>513</sup> Space groups were determined from systematic absences and checked for higher symmetry. All structures were solved using direct methods with SHELXS,<sup>514</sup> refined on  $F^2$  using all data by full matrix least-squares procedures with SHELXL-97,<sup>515</sup> within the WinGX<sup>516</sup> program. Non-hydrogen atoms were refined with anisotropic displacement parameters. Hydrogen atoms were placed in calculated positions or manually assigned from residual electron density where appropriate, unless otherwise stated. The functions minimized were  $\Sigma w(F2o-F2c)$ , with  $w = [\sigma 2(F2o) + aP2 + bP] - 1$ , where P = [max(Fo)2 + 2F2c]/3. The isotropic displacement parameters are 1.2 or 1.5 times the isotropic equivalent of their carrier atoms.

# Additional Information:

(*NON*<sup>*Dipp*</sup>)*Al*–*Zn*(*BDI*<sup>*Mes*</sup>) (**1-***Zn*): The molecule is located on a C2-axis that passes through the Al-Zn bond. The unit cell contains poorly defined toluene solvate molecules that have been treated as a diffuse contribution to the overall scattering without specific atom positions by SQUEEZE/PLATON. Details are given in the .cif.

 $(NON^{Dipp})Al(\mu-O_2C)Zn(BDI^{Mes})$  (2): The toluene solvate is on an inversion centre and was modelled as a full molecule at 50% occupancy. One of the ortho-Me groups of a Mes substituent is disordered and was modelled over two positions with SADI restraints applied.

 $\{(NON^{Dipp})AI(\mu-CO_3)\}_2\{Mg(THF)_4\}\ (3)$ : One of the methylene groups of a coordinated THF is disordered and was modelled over two positions. The oxygen atom of the THF solvate is also disordered and was modelled over two positions.

 $(NON^{Dipp})Al(\mu-O)Zn(BDI^{Mes})$  (4-Zn): The aromatic group of one of the NON-ligands is disordered and the C6-ring and one of the iPr substituents were modelled over two positions. In addition, the asymmetric unit contains 4 poorly defined toluene solvate molecules. These were treated as a diffuse contribution to the overall scattering without specific atom positions by SQUEEZE/PLATON. Details are given in the .cif.

 $(NON^{Dipp})Al(\mu-O)Mg(BDI^{Mes})$  (**4-Mg**): The asymmetric unit contains 4 poorly defined toluene solvate molecules. These were treated as a diffuse contribution to the overall scattering without specific atom positions by SQUEEZE/PLATON. Details are given in the .cif.

	1-Zn	1-Mg	2
Empirical formula	C <sub>58</sub> H <sub>85</sub> AlN <sub>4</sub> OSi <sub>2</sub> Zn	C <sub>51</sub> H <sub>75</sub> AlMgN <sub>4</sub> OSi <sub>2</sub>	$C_{215}H_{308}Al_4N_{16}O_{12}Si_8Zn_4$
CCDC Number	2183690	2183691	2183692
M <sub>r</sub>	1000.81	867.62	3902.97
Т [К]	120.0(1)	150(2)	150.0(1)
Crystal size [mm]	0.48 × 0.13 × 0.11	0.22 × 0.13 × 0.08	$0.20 \times 0.14 \times 0.11$
Crystal system	Monoclinic	Monoclinic	Triclinic
Space group	C2/c (No.15)	$P2_1/n$ (alternative No.14)	P <sup>1</sup> (No. 2)
a [Å]	19.79296(16)	11.5558(2)	14.3252(2)
<i>b</i> [Å]	12.46075(9)	18.8726(3)	19.0153(3)
<i>c</i> [Å]	26.05702(19)	24.3018(4)	22.8866(3)
α[°]	90	90	73.2612(13)
β[°]	97.9035(7)	90.5318(13)	86.3624(13)
γ[°]	90	90	71.2352(15)
V [ų]	6365.53(8)	5299.69(15)	5649.87(16)
Ζ	4	4	1
D <sub>calc.</sub> [mg m <sup>-3</sup> ]	1.044	1.088	1.147
Absorption coefficient [mm <sup>-1</sup> ]	1.302	1.162	1.482
heta range for data collection [°]	4.2039 to 72.176	3.6378 to 73.4215	3.6679 to 73.6941
Reflections collected	20014	34093	78880
Independent reflections	6202 [ <i>R</i> <sub>int</sub> 0.015]	10464 [ <i>R</i> <sub>int</sub> 0.034]	22585 [ <i>R</i> <sub>int</sub> 0.031]
Reflections with $l > 2\sigma(l)$	5876	8804	20053
Data/restraints/parameters	6202 / 0 / 283	10464 / 0 / 561	22585 / 3 / 1250
Final <i>R</i> indices $[I > 2\sigma(I)]$	$R_1 = 0.031$ , w $R_2 = 0.088$	$R_1 = 0.040$ , w $R_2 = 0.102$	$R_1 = 0.035, wR_2 = 0.092$
Final R indices (all data)	$R_1 = 0.033, wR_2 = 0.089$	$R_1 = 0.049, wR_2 = 0.109$	$R_1 = 0.040, wR_2 = 0.096$
GOOF on F <sup>2</sup>	1.02	1.04	1.018
Largest diff. peak/hole [e.Å <sup>-3</sup> ]	0.27 and –0.26	0.26 and –0.21	0.57 and –0.39

# Table S6Crystal structure and refinement data for 1-Zn, 1-Mg and 2

	3	4-Zn	4-Mg	5
Empirical formula	C <sub>78</sub> H <sub>132</sub> Al <sub>2</sub> MgN <sub>4</sub> O <sub>13</sub> Si <sub>4</sub>	$C_{130}H_{182}AI_2N_8O_4Si_4Zn_2$	$C_{130}H_{182}AI_2Mg_2N_8O_4Si_4$	$C_{52}H_{75}AIN_4O_4Si_2Zn$
CCDC Number	2183693	2183694	2183695	2183696
M <sub>r</sub>	1524.5	2217.89	2135.77	968.69
Т [К]	120.0(1)	150.0(1)	150.0(1)	150.0(1)
Crystal size [mm]	$0.18 \times 0.12 \times 0.08$	0.48 × 0.37 × 0.12	$0.40 \times 0.24 \times 0.12$	$0.31 \times 0.13 \times 0.08$
Crystal system	Triclinic	Monoclinic	Monoclinic	Monoclinic
Space group	P <sup>1</sup> (No. 2)	P2 <sub>1</sub> /c (No. 14)	P2 <sub>1</sub> /c (No. 14)	P2 <sub>1</sub> /c (No. 14)
a [Å]	16.8558(7)	19.0790(2)	19.0122(4)	14.56239(17)
<i>b</i> [Å]	17.0958(8)	26.4548(3)	26.6902(5)	15.0196(2)
<i>c</i> [Å]	19.1000(6)	26.1164(3)	26.0942(6)	25.1167(3)
<i>α</i> [°]	115.826(4)	90	90	90
β[°]	93.517(3)	96.4642(10)	96.081(2)	97.8850(10)
γ[°]	114.756(4)	90	90	90
V [ų]	4294.2(3)	13098.0(2)	13166.0(5)	5441.60(11)
Ζ	2	4	4	4
$D_{\text{calc.}}$ [mg m <sup>-3</sup> ]	1.179	1.133	1.078	1.182
Absorption coefficient [mm <sup>-1</sup> ]	1.382	1.331	1.031	1.55
heta range for data collection [°]	3.6353 to 73.3746	3.4672 to 73.3138	3.4573 to 73.4317	3.747 to 73.24
Reflections collected	62418	91143	87564	40320
Independent reflections	17097 [ <i>R</i> <sub>int</sub> 0.055]	26101 [ <i>R</i> <sub>int</sub> 0.031]	26221 [ <i>R</i> <sub>int</sub> 0.043]	10851 [ <i>R</i> <sub>int</sub> 0.027]
Reflections with $l > 2\sigma(l)$	13500	20965	18454	9420
Data/restraints/parameters	17097 / 28 / 963	26101/0/1162	26221/0/1139	10851/0/597
Final <i>R</i> indices $[I > 2\sigma(I)]$	$R_1 = 0.078, wR_2 = 0.210$	$R_1 = 0.051, wR_2 = 0.154$	$R_1 = 0.068, wR_2 = 0.193$	$R_1 = 0.034, wR_2 = 0.088$
Final R indices (all data)	$R_1 = 0.095, wR_2 = 0.223$	$R_1 = 0.063, wR_2 = 0.162$	$R_1 = 0.089, wR_2 = 0.211$	$R_1 = 0.041, wR_2 = 0.093$
GOOF on F <sup>2</sup>	1.059	1.082	0.947	1.012
Largest diff. peak/hole [e.Å-3]	1.24 and –0.53	0.36 and –0.47	0.37 and –0.67	0.36 and –043

Table S7Crystal structure and refinement data for 3, 4-Zn, 4-Mg and 5

# References

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Cartesian Coordinates and Computed Energies (in Hartree	es) fo	or Calculated	Structures	
1-Mg	С	-3.53794	-1.65750	-0.85828
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Enthalpy 298K = -2292.660484	С	-0.78099	0.48603	-4.16003
$E_{ree} = 298K = -2292 848804$	н	-1 58516	0 99878	-4 71723
Lewest Frequency = $11.4634$ cm <sup>-1</sup>	ц	0 14059	0.53070	-1 76550
Lowest frequency = $11.4034$ cm	п	0.14050	1 05120	-4.70559
Second Frequency = 18.7/12 cm <sup>2</sup>	п	-0.01302	1.05129	-3.22364
~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	C	-1.32116	-1.///26	-5.16547
Si 0.61978 -3.80684 1.40893	Н	-1.58598	-2.83006	-4.96975
Si -0.70238 -3.79769 -1.40324	Н	-0.38797	-1.76173	-5.75603
Al -0.01121 -0.97725 -0.00204	Н	-2.11995	-1.34927	-5.79697
0 -0.04683 -4.48180 0.00324	С	-3.52841	-2.23290	0.55821
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C 4.71237 -1.28065 1.40769	Н	-4.08230	-0.35888	1.56572
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C 4.77703 -0.76192 2.70728	Н	-5.68712	-2.38877	1.00749
н 5.73434 -0.43204 3.12551	Н	-4.66327	-3.54481	1.88246
C 3 60668 -0 68234 3 46796	н	-5 01931	-3 80314	0 15115
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C = 3.47492 = 2.30560 = 0.56142	N	_1 /20013	3 25049	0.47107
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$\begin{array}{c} 11 \\ 1.04051 \\ -1.02051 \\ -2.75087 \\ -2.7788 \\ -2.75087 \\ -$	и П	7.10609	2 57620	-2 30600
$\begin{array}{c} 0.000001 - 4.02001 - 2.70007 \\ 1 1.00102 - 2.20704 - 2.50410 \end{array}$	п	7.40090	2.57059	1 56661
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Н	-6.81658	0.65673	2.37012
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Н	-1.66120	4.15764	3.24710

#### 1-Zn

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Enthalpy $298K = -2519.032787$
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Al	-0.00002	-0.89320	-0.00000
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Η	-6.86157	0.52529	2.45778
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Η	-3.03053	3.43500	-1.78520
тт	1 27716	2 20240	2 11100
п	-4.3//40	2.30340	-2.11109
Η	-2.73732	1.68547	-1.83277
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<u>о</u> т	0.000/0	5.70405	1.10040
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п	-2.30032	-2.94309	0.03049
С	-3.20199	-1.15593	1.65000
н	-2 26792	-0 59972	1 43903
11	2.20192	0.39972	1.45905
Н	-3.12214	-1.57194	2.67065
Н	-4.01615	-0.41292	1.64092
<u> </u>	1 (0000	2 10050	1 01000
C	-4.68889	-3.10058	T.01200
Н	-5.58744	-2.47311	1.15031
ប	_1 50010	_3 61001	1 07000
п	-4.30210	-2.01091	1.9/330
Η	-4.92747	-3.86917	0.25785
C	0 65531	-3 91613	-2 76551
<u> </u>	0.000034	2.21013	2.70004
Н	1.52996	-3.26765	-2.59929
н	0 24675	-3 68800	-3 76340
11	0.240/0	5.00000	5.10542
H	1.00391	-4.96337	-2.77189
С	-2.18366	-4.71327	-1.87931
	1 07510		1.07.551
Н	-1.8/512	-5./5647	-2.06521
Н	-2.66239	-4.32779	-2.79423
	2 02650	1 71740	1 07557
Н	-2.93658	-4./1/48	-1.0/55/
Ν	1.42959	3.00126	-0.48253
C	1 00140	1 22250	_0 20200
C	1.22142	4.33333	-0.38265
С	2.36040	5.29758	-0.69180
IJ	3 22007	5 10614	_0 03700
п	3.2200/	5.10014	-0.03/82
Н	0 0 0 0 1 0	6.33940	-0.55009
11	2.03812		
IJ	2.03812	5 17000	-1 77664
Н	2.03812	5.17980	-1.72664
H C	2.03812 2.72286 2.74434	5.17980 2.60072	-1.72664 -0.92015
H C C	2.03812 2.72286 2.74434 3.03132	5.17980 2.60072 2.51511	-1.72664 -0.92015 -2.30728
H C C	2.03812 2.72286 2.74434 3.03132	5.17980 2.60072 2.51511	-1.72664 -0.92015 -2.30728

НССНССНННСНННСННН	4.56317 5.37190 5.05344 5.83650 3.75749 1.96229 2.34199 1.60194 1.07658 6.77027 7.52268 7.04094 6.86163 3.46351 3.03063 4.37751 2.73735	2.14640 -3.78318 1.96125 -1.78057 2.03734 -0.41305 1.85406 0.33265 2.35155 0.03848 2.79617 -3.33853 2.61662 -4.35725 3.84003 -3.29187 2.15799 -3.18440 1.60587 -2.24139 1.84941 -1.47305 2.14326 -3.16612 0.52506 -2.45767 2.45505 1.51692 3.43503 1.78512 2.30338 2.11105 1.68551 1.83274	
<b>4-M</b> SCF Enth Enth	(BP86) H (BP86) H nalpy OK nalpy 298	Energy = -2369.18621984 = -2368.064569 BK = -2367.987402	
Lowe	est Frequend	$a = 12.300 \cdot 10.5272$ $a = 12.300 \cdot 10.5272$ $a = 12.300 \cdot 10.5272$ $a = 12.300 \cdot 10.5272$	
Si - Al Ma -	0.71052 -0.66462 0.00784	-4.09718 1.41074 -4.10404 -1.40983 -1.34683 0.00054 2.19201 0.00024	
0 0 - N N -	0.02315 -0.00177 1.06476 -1.03693	-4.74411 0.00192 0.34769 0.00181 -2.37150 1.11106 -2 38199 -1 11172	
N - N	-1.47629 1.43434	3.54758 0.44354 3.56429 -0.44340	
C C	2.24200 3.47880 4.62431	-1.79979 1.00007 -1.25139 1.60600	
H C H	5.57283 4.57855 5.48379	-1.25473 1.05882 -0.70814 2.89648 -0.29584 3.35563	
C H C	3.36582 3.32557	-0.70135 3.59370 -0.27016 4.60079 -1.24007 3.03094	
C H	3.55246	-2.34482 -0.42647 -3.08566 -0.52792	
C H H	3.27693 2.31015 3.26169	-1.21755 -1.44960 -0.71357 -1.26089 -1.61190 -2.48165	
H C	4.05109 4.87841	-0.43285 -1.38897 -3.05877 -0.75626	
H H H	4.81456 5.11614	-2.35728 -0.79222 -3.53872 -1.74830 -3.83887 -0.01317	
C H C	0.88537 0.14940	-1.16609 3.82109 -1.78826 3.28600	
н н -	1.04666	0.97202 4.31670 0.33526 4.37372	
H C H	0.16746 1.01735 1.41199	0.62433 2.79394 -1.71380 5.25795 -2.74427 5.26396	
H H	0.03397	-1.72006 5.76027 -1.09663 5.87421	
C H H	2.29069 2.03716 2.78874	-5.05612 1.80933 -6.11250 2.00205 -4.65301 2.70639	

	2 00040	F 00000	0 07400
н	3.00840	-5.02308	0.9/439
С	-0.56130	-4.38219	2.78996
ц	_1 /7200	-2 70516	2 62121
п	-1.4/590	-3./0510	2.02434
Н	-0.15689	-4.12150	3.78234
u	_0 8519/	-5 11678	2 80070
п	-0.03194	-3.44070	2.00919
С	0.61472	-4.37596	-2.78473
u	1 52060	-3 769/2	-2 61637
п	1.32000	-3.70942	-2.01037
Н	0.21068	-4.11938	-3.77835
ц	0 91619	-5 13716	-2 80376
11	0.91049	5.45/40	2.00570
С	-2.23373	-5.07868	-1.81385
н	-1 96915	-6 13277	-2 00428
	1.00010	0.102//	2.00120
Н	-2./3189	-4.68140	-2./1344
Н	-2.95522	-5.05163	-0.98196
~	2.20025	1 00077	1 71070
C	-2.22025	-1.822//	-1./19/8
С	-3.45707	-1.83656	-1.00118
C	1 60010	1 20100	1 60764
C	-4.00040	-1.30100	-1.00/04
Η	-5.55692	-1.31445	-1.06063
C	-4 56837	-0 75824	-2 89849
C	4.50057	0.75024	2.00040
Н	-5.47801	-0.35622	-3.35807
C	-3 35567	-0 73858	-3 59554
	2 220007	0 20700	1 (0000
н	-3.32000	-0.30/68	-4.60292
С	-2.17525	-1.26392	-3.03216
C	-3 52/02	-2 20107	0 12557
C	5.52492	2.3010/	0.42007
Η	-2.70507	-3.11485	0.52701
C	-3 25990	-1 25171	1 44830
~	5.25550	1.201/1	1.44050
Н	-2.29819	-0./382/	1.25902
Н	-3.24034	-1.64565	2.48042
TT	1 01176	0 47465	1 20702
н	-4.041/0	-0.4/465	1.30/03
С	-4.84378	-3.10847	0.75613
н	-5 70297	-2 41520	0 79240
	5.70257	2.41520	0.75240
Н	-4.77481	-3.58760	1.74821
Н	-5.07439	-3.89098	0.01332
C	0 07000	1 17502	2 02107
C	-0.07000	-1.1/392	-3.02107
Η	-0.12732	-1.78912	-3.28581
C	-0 33807	0 27563	-3 83140
~	0.55007	0.27505	5.05140
Н	-1.05534	0.95976	-4.31982
Н	0.62113	0.34192	-4.37514
TT	0 17220	0 (0070	2 70624
н	-0.1/320	0.62372	-2./9624
С	-0.99520	-1.72653	-5.25825
н	-1 37762	-2 76160	-5 26334
	1.01102	1 70100	5.20001
Н	-0.01168	-1./2168	-5./6032
Н	-1.68000	-1.11801	-5.87526
C	_1 27162	1 07015	0 33000
C	-1.2/103	4.0/915	0.33990
С	-0.03202	5.47514	-0.00001
Н	-0.03829	6.56746	-0.00003
~	1 01420	4 00242	0 22000
C	1.21439	4.89342	-0.33989
С	-2.43297	5.83098	0.58988
н	-3 26586	5 61460	-0 10128
	3.20500	5.01100	0.10120
Н	-2.12584	6.87785	0.45325
Н	-2.84012	5.71288	1.60812
C	2 26170	E 0E0E1	0 50052
C	2.364/8	5.85854	-0.58953
Η	3.20002	5.65161	0.10169
н	2 04564	6 90180	-0 45275
	2.01001	0.90100	0.15275
Н	2.77336	5.74532	-1.60776
С	-2.78133	3.09020	0.84726
Ċ	-2 70664	2 70520	_0 12502
C	-3.12004	2.10522	-0.13283
С	-4.99276	2.25442	0.28285
Н	-5.72508	1.96632	-0.48115
~	5.72000	1.0002	1 6410210
С	-5.34188	2.16227	1.64179
С	-4.37475	2.52912	2.59632
U	-1 62020	2 46012	3 66370
п	-4.02038	∠.40013	3.003/2
С	-3.09716	2.98666	2.22748
С	-3.38571	2.77465	-1.60842
	0.070071	2.77100	1 00012
Н	-2.97039	3.75667	-1.89410
Н	-4.27619	2.58092	-2.22645
IJ	-2 63570	2 00000	_1 00500
11	2.033/9	2.00990	1.00002
С	-6.70427	1.65581	2.06856
Н	-7.45238	1.78738	1.26940
	. = = = •		

Н Н С Н Н	-7.06916 -6.67479 -2.08301 -1.76887 -1.16248	2.18238 0.57823 3.36304 4.41948 2.75863	2.96675 2.31549 3.28624 3.21196 3.19765
Η	-2.49365	3.20364	4.29613
С	2.74457	3.12182	-0.84693
С	3.69382	2.74710	0.13640
С	4.96500	2.31060	-0.28192
Η	5.70018	2.03039	0.48227
С	5.31562	2.22283	-1.64080
С	4.34484	2.57923	-2.59552
Η	4.59169	2.51342	-3.66285
С	3.06204	3.02232	-2.22702
С	3.35155	2.81246	1.60886
Η	2.92640	3.79023	1.89475
Η	4.24359	2.62727	2.22724
Η	2.60906	2.04025	1.88477
С	6.68369	1.73149	-2.06707
Η	7.03857	2.25487	-2.97108
Η	6.66853	0.65136	-2.30400
Н	7.43224	1.88063	-1.27139
С	2.04412	3.38753	-3.28609
Н	1.13067	2.77235	-3.19832
Н	2.45709	3.23352	-4.29586
Η	1.71763	4.44018	-3.21135

#### 4-Zn

SCF (BP86) Energy = $-2595.52413223$
Enthalpy $0K = -2594.402860$
Enthalpy $298K = -2594.325484$
Free Energy 298K = -2594.520986
Lowest Frequency = $9.2236 \text{ cm}^{-1}$
Second Frequency = $12.3800 \text{ cm}^{-1}$

Zn	0.22167	2.14119	0.01983
Si	-1.09935	-4.05732	1.36480
Si	0.27434	-4.12834	-1.45170
Al	-0.14386	-1.35757	0.00611
0	-0.42439	-4.74116	-0.03300
0	0.02163	0.32965	0.07333
Ν	-1.30300	-2.30284	1.08154
Ν	0.80845	-2.45225	-1.12780
Ν	1.78621	3.25650	0.45576
Ν	-1.08488	3.54522	-0.43471
С	-2.44611	-1.65208	1.67835
С	-3.66246	-1.54314	0.93338
С	-4.77831	-0.92728	1.52936
Н	-5.71239	-0.84836	0.96355
С	-4.72197	-0.42158	2.83475
Н	-5.60503	0.04419	3.28603
С	-3.52859	-0.52251	3.55732
Н	-3.47987	-0.12109	4.57621
С	-2.38383	-1.13269	3.00574
С	-3.74158	-2.05175	-0.50620
Н	-3.00742	-2.87208	-0.59498
С	-5.12069	-2.61815	-0.89765
Н	-5.89379	-1.83093	-0.95078
Н	-5.46621	-3.38122	-0.17957
Н	-5.06893	-3.08596	-1.89614
С	-3.31415	-0.94360	-1.49685
Н	-2.31234	-0.53604	-1.26102
Н	-4.00911	-0.08739	-1.45384
Н	-3.29284	-1.32294	-2.53438
С	-1.09619	-1.17819	3.82720
Н	-0.39594	-1.84082	3.29271
С	-1.30588	-1.75178	5.24472
Н	-1.77923	-2.74821	5.21396

Η	-1.94851	-1.09950	5.86239
Η	-0.33853	-1.84743	5.76866
С	-0.44131	0.22062	3.88800
Η	-1.10846	0.95114	4.38018
H	-0.22188	0.58243	2.86852
Н	0.5055/	0.18/49	4.45608
C	-2.76458	-4.88538	1.70797
н	-2.60408	-5.96263	1.884/0
п	-3.43709	-4.///00	2 60204
C	0 10518	-4 47005	2 77156
н	1.06889	-3.94920	2.64262
Н	0.30505	-5.55555	2.77666
Н	-0.30392	-4.19484	3.75793
С	-1.05532	-4.24883	-2.79995
Н	-1.89489	-3.56183	-2.60120
Η	-1.45780	-5.27600	-2.83243
Η	-0.64818	-4.01093	-3.79695
С	1.74136	-5.22618	-1.91508
Η	1.38233	-6.25311	-2.09906
Η	2.49181	-5.26398	-1.10966
H	2.23901	-4.86606	-2.83036
C	3.23263	-2.72550	0.48364
н	2.34423	-3.3/841	0.55151
U U	4.40239 5 20174	-2 96775	0.03300
п	4 64592	-4 37560	0.93290
н	4.30940	-4.04795	1.84557
C	3.03989	-1.57801	1.50301
Н	2.14754	-0.96285	1.27831
Н	3.89962	-0.88609	1.48309
Η	2.93468	-1.97186	2.52998
С	1.71427	4.60161	0.35590
С	0.54275	5.31454	0.01184
Η	0.65134	6.40131	0.01453
С	-0.74780	4.84912	-0.33027
С	2.96338	5.42694	0.62416
H	3.78380	5.12060	-0.04/16
п u	2 76654	5.27747	1.05240
C	-1 81156	5 90707	-0 58121
Н	-2.67106	5.76062	0.09524
Н	-2.21159	5.84551	-1.60709
Н	-1.40642	6.91688	-0.42261
С	3.02832	2.66237	0.87689
С	3.94793	2.19952	-0.09557
С	5.16347	1.64013	0.34111
Η	5.87899	1.29013	-0.41271
С	5.48247	1.51648	1.70521
C	4.53589	1.961/2	2.64/03
п	4.75800	2 52960	3./1/00
C	3 63365	2.32000	-1 57048
н	2 76758	1 67538	-1 83989
Н	4.48502	1.96492	-2.17884
Н	3.38405	3.34028	-1.86896
С	6.78934	0.89326	2.15041
Η	6.66415	-0.18009	2.38563
Η	7.18077	1.37907	3.06028
Η	7.56029	0.96788	1.36576
С	2.31251	2.98892	3.30360
H	1.34418	2.47023	3.19092
H	2.09678	4.07004	3.23148
H	2.68895	2./8/91	4.31936
C	-2.42330	3.21U10 3.16572	-0.04001
C	-4.06574	2.85539	-2.59894
H	-4.31854	2.83241	-3.66659

С	-5.06539	2.57761	-1.64797
С	-4.70667	2.60269	-0.28821
Н	-5.46348	2.37662	0.47260
С	-3.39952	2.90717	0.13525
С	-1.69699	3.44887	-3.28012
Н	-0.83799	2.76167	-3.18663
Н	-1.28627	4.47163	-3.20299
Н	-2.11794	3.33165	-4.29154
С	-6.47476	2.22808	-2.07917
Н	-6.57134	1.14947	-2.30378
Н	-6.76672	2.77516	-2.99148
Н	-7.20859	2.46379	-1.29069
С	-3.04631	2.91337	1.60522
Н	-2.60259	3.87366	1.92234
Н	-2.30942	2.12379	1.84000
Н	-3.93621	2.71970	2.22357
С	2.06389	-2.02497	-1.69772
С	2.12035	-1.47764	-3.01388
С	3.36894	-1.09579	-3.54503
Н	3.41167	-0.67408	-4.55608
С	4.55048	-1.24661	-2.81172
Н	5.51319	-0.95598	-3.24658
С	4.49118	-1.77912	-1.51715
Н	5.41608	-1.89774	-0.94325
С	3.26955	-2.17541	-0.94254
С	0.85881	-1.25721	-3.84731
Η	0.02988	-1.75126	-3.31415
С	0.95631	-1.87795	-5.25737
Η	-0.00065	-1.76448	-5.79676
Η	1.73653	-1.39013	-5.86834
Η	1.19818	-2.95366	-5.21091
С	0.51536	0.24723	-3.92919
Η	-0.41398	0.40891	-4.50394
Η	0.37143	0.66190	-2.91632
Н	1.32417	0.81571	-4.42254

[Al(<sub>Xanth</sub>NON<sup>Dipp</sup>)]-Mg SCF (BP86) Energy = -3019.54944840 Enthalpy 0K = -3018.142253 Enthalpy 298K = -3018.054531 Free Energy 298K = -3018.268668 Lowest Frequency = 10.9818 cm<sup>-1</sup> Second Frequency = 15.6475 cm<sup>-1</sup>

Al	-0.15293	-0.11342	-0.21954
0	-1.93190	0.29661	0.75727
Ν	-0.55448	1.66111	-0.96845
Ν	-1.12585	-1.82172	-0.50368
С	-2.54431	1.50467	0.33612
С	-2.85110	-0.78308	0.76997
С	-1.77829	2.23028	-0.59465
С	-2.38200	-1.92540	0.10069
С	-2.35906	3.44381	-1.03497
Н	-1.79695	4.05984	-1.74298
С	-3.82130	1.81942	0.78243
С	-3.64702	3.84451	-0.60424
С	-0.63718	-3.01002	-1.15387
С	-3.26777	-3.03659	0.10868
Н	-2.94013	-3.96025	-0.37222
С	0.42211	2.55937	-1.52709
С	-4.36998	3.02481	0.29071
Н	-5.37421	3.31417	0.60314
С	1.05731	3.52391	-0.68436
С	-4.13672	-0.58940	1.26593
С	-4.55756	-2.94000	0.67397
С	-4.50410	0.82953	1.75453
С	0.06440	-4.00222	-0.40316
С	0.75586	2.49795	-2.91306

~	E E 2 0 1 7	4 14051	0 00007
C	-2.2301/	-4.14351	0.68287
С	-0.88947	-3.21811	-2.54501
C	-4 24193	5 16052	-1 16117
~	1.21100	1 20002	1 00000
С	-4.99112	-1.70852	1.22336
Η	-6.01067	-1.61745	1.60664
C	0 70107	2 60010	0 01000
C	0./910/	3.00910	0.01909
Η	0.10974	2.78526	1.08978
C	1 97542	4 42823	-1 25317
C	1.97942		1.2001/
Н	2.45416	5.17664	-0.61094
C	0 33663	-3 85101	1 09298
	0.05000	0.001010	1 000200
Н	0.02821	-2.81819	1.36/44
С	0.07666	1.50044	-3.85275
ц	_0 15005	0 50775	-3 25010
п	-0.13903	0.59775	-3.23040
С	-3.92198	1.04625	3.18445
н	-4 14451	2 06860	3 53563
	7.17751	2.00000	0.00000
Н	-2.82918	0.90893	3.19822
Н	-4.36961	0.32585	3.89084
~	0 40420	C 17C74	1 05424
C	0.49438	-5.1/5/4	-1.05434
Η	1.02580	-5.94169	-0.47818
C	_1 60002	-2 21174	-3 30300
<u> </u>	1.00002	2.211/4	5.50200
Η	-1.62801	-1.23950	-2.85927
C	1,69084	3.41775	-3.42778
	1 04000	2 20000	1 10014
Н	1.94936	3.38063	-4.49014
С	-0.43404	-4.40741	-3.14787
TT	0 62020	1 57511	1 01007
п	-0.02020	-4.5/514	-4.2123/
С	-6.02959	1.03678	1.80204
н	-6 49259	0 89299	0 81209
	0.15255	0.05255	0.01200
Н	-6.2/115	2.05233	2.15699
Н	-6.49388	0.33086	2.51048
0	2 20047	C 24007	0 00010
C	-3.30947	0.3488/	-0.80812
Η	-3.71886	7.29221	-1.21247
н	-2 29826	6 21622	-1 22710
	2.29020	0.21022	1.22710
Н	-3.21009	6.46112	0.28542
С	-5.91021	-4.48311	2.14881
11	C 20241	2 (2002	0 (5451
н	-6.39241	-3.62992	2.65451
Η	-5.01565	-4.75781	2.73385
н	-6 61385	-5 33465	2 17750
11	0.01000	5.55405	2.17750
С	1.82821	-4.05125	1.43670
Н	2 15809	-5 08671	1 23965
	2.10009	0.00071	1.20000
Н	2.01251	-3.84902	2.50568
Η	2.48379	-3.38695	0.84632
C	2 20150	/ 38510	-2 61/35
	2.29139	4.30310	2.01455
Н	3.00729	5.09688	-3.03996
С	-5.64239	5,46582	-0.58453
	6 01665	6 41561	1 00/10
п	-0.01003	0.41301	-1.00419
Η	-5.61992	5.57295	0.51393
н	-6 37374	4 67991	-0 84101
<u> </u>	0 0 0 0 0 0 0	E 20400	0 11 - 10
C	0.25052	-3.38408	-2.41548
Η	0.58849	-6.30471	-2.90360
C	0 10720	4 93655	1 21777
	0.10/00		1.21///
Н	0.74927	5.80366	0.97798
Н	-0.09193	4.95762	2.30360
ц.	_0 05074	5 06260	0 60034
п	-0.030/4	5.00200	0.09034
С	-3.17504	-2.60594	-3.47434
н	-3.28640	-3.58532	-3.97437
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Н	-3./3556	-1.82866	-4.06360
Н	-3.63842	-2.67459	-2.47769
C	2 00120	3 11625	1 62022
C	2.09120	J. HIUJJ	1.02900
Η	2.59270	2.46846	1.35865
н	1.88274	3.38638	2.71181
 T 7	2 01207	4 00054	1 44040
н	∠.8138/	4.23054	1.44349
С	-4.91666	-5.40776	0.04154
н	-5 64582	-6 23543	0 08369
	0.04002	0.20040	0.00009
Η	-4.00704	-5.73843	υ.57185
Η	-4.65732	-5.24748	-1.01901
C	_1 10071	_2 02041	-1 70060
	1.100/1	2.02941	
Н	-0.01826	-1.82316	-4./7/518
Η	-1.60266	-1.18859	-5.30875

Н	-1.26036	-2.92409	-5.42761
С	-1.26701	2.06612	-4.37187
Н	-1.95180	2.30488	-3.54245
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С	-6.81463	-3.77895	-0.10828
н	-7 52255	-4 62733	-0 10400
ц	-6 57/96	-3 53750	-1 15797
п	-0.37490	-3.33730	-1.13/9/
н	-/.33256	-2.90790	0.32643
С	-4.3/068	5.05314	-2./0426
Н	-5.03704	4.21988	-2.98626
Н	-3.39377	4.87915	-3.18479
Н	-4.79112	5.98585	-3.12182
С	-0.54101	-4.81098	1.92946
Н	-1.61268	-4.63379	1.74395
ц	-0 35215	-4 66991	3 00855
ц	_0 31970	-5 96514	1 60307
п	-0.31070	-J.00314	1.00397
C	0.97029	1.068/9	-5.03239
Н	0.47482	0.27064	-5.60879
Н	1.94685	0.69000	-4.68559
Н	1.15295	1.90098	-5.73560
Mg	2.37658	-0.42995	0.82780
Ν	4.17610	-1.00900	-0.02484
N	3.12008	-0.55509	2.77125
C	4 47493	-0 67259	-1 39378
c	2 59674	0.07200	2 05257
C	2.30074	0.22047	1 (7007
Ĉ	5.1/941	0.52655	-1.6/82/
С	4.14464	-1.3960/	3.0443/
С	4.02917	-1.50745	-2.45075
С	5.08129	-1.72248	0.67436
С	4.97732	-1.98931	2.06506
Н	5.73643	-2.66920	2.45954
С	1.31618	-0.07447	4.40906
C	3 31268	1 35790	4 32387
c	1 21720	_1 12176	-3 77666
	4.51/50	-1.13170	-3.77000
н	3.98406	-1./883/	-4.59032
C	5.02316	0.04466	-4.08685
С	5.44326	0.85793	-3.01947
Н	5.99502	1.78195	-3.23338
С	6.31545	-2.26625	-0.03126
Н	6.04609	-2.81745	-0.94740
Н	6.89168	-2.92963	0.63028
Н	6.97424	-1.43917	-0.34956
C	4 46003	-1 74066	4 49511
U U	4 02501	-0 00033	5 02275
11	4.92301 E 1E071	0.09032	1 5 6 2 2 7 5
н 	5.15071	-2.59503	4.55405
Н	3.53863	-1.98584	5.04941
С	4.65168	1.73290	3.72556
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Н	5.45213	1.02262	4.00244
С	3.27106	-2.78423	-2.16815
н	2.30807	-2.58862	-1.66228
ч	3 83482	-3 46678	-1 50778
и П	2 04516	-2 22515	-3 10030
п	3.04510	-3.32313	-3.10030
C .	2./36/6	2.14/65	5.34/43
H	3.32090	3.02017	5.70119
С	0.80200	0.74684	5.42954
Н	-0.17874	0.50038	5.85573
С	5.63866	1.44201	-0.56375
Н	6.24692	2.26949	-0.96288
Н	6.23725	0.91099	0.19662
Н	4.77935	1,89041	-0.03139
C	0 52350	-1 26836	3 92907
U U	-0 30530	_1 /1020	1 525/1
11 U	1 11=10	_2 10000	2 07044
п	1.11319	-2.19929	3.9/044
н	0.2035/	-1.14659	2.8/638
С	5.29868	υ.43579	-5.52383

Н	6.21763	1.03958	-5.60962
Н	4.47157	1.03941	-5.94136
Η	5.41134	-0.45081	-6.17023
С	1.50474	1.86124	5.92028
С	0.94420	2.70822	7.04349
Н	-0.15781	2.74181	7.01416
Η	1.31792	3.74458	6.99396
Н	1.23059	2.30590	8.03328

[Al (NC<sub>2</sub>N<sup>pipp</sup>)]-Mg SCF (BP86) Energy = -2532.99685646 Enthalpy 0K = -2531.660673 Enthalpy 298K = -2531.577132 Free Energy 298K = -2531.377132Free Energy 298K = -2531.777438Lowest Frequency = 17.3343 cm<sup>-1</sup> Second Frequency = 21.0452 cm<sup>-1</sup>

Al	1.38182	-0.40161	0.17023
Si	4.01193	-1.61817	-1.31803
Si	3.29954	-1.90735	2.50937
Ν	1.74449	-1.33647	1.77342
N	2.91473	-0.23664	-0.94345
С	0.58742	-1.55912	2.61598
С	0.15985	-0.56603	3.55354
C	-0.1218/	-2.80190	2.545/1
U U	0.93976	0.72831	3./902/
п С	3 19740	0.70857	-1 65853
C	2 52518	1 31389	-2 88248
C	-0.98714	-0.80763	4.33834
Н	-1.31297	-0.04204	5.05220
С	1.44017	0.42850	-3.49704
Н	1.27294	-0.41416	-2.80288
С	0.33468	-3.92118	1.60917
Η	1.40145	-3.73688	1.39047
С	-1.25349	-2.99771	3.35946
Н	-1.78973	-3.95155	3.30814
С	0.10823	1.19333	-3.66086
H	-0.22028	1.65583	-2./1155
н u	-0 69415	2.01853	-4.38916
п С	4 20879	1 88884	-4.02030
С	-1.69774	-2.00822	4.24642
Н	-2.57951	-2.18094	4.87263
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Н	0.62866	0.88549	5.96585
Н	2.05066	-0.08439	5.51826
Н	2.10124	1.69755	5.37931
С	1.87684	-0.15436	-4.86102
Η	2.11259	0.64827	-5.58196
H	2.77318	-0.78879	-4.76471
Н	1.0/115	-0./6853	-5.300/3
	2.98986	-3.14101	-1.83/04
п ц	2.20117	-2 98043	-2 81393
Н	3 64285	-4 02851	-1 92419
C	0.11529	1.98717	3.45592
Н	-0.21416	1.99281	2.40338
Н	-0.78825	2.06072	4.08598
Н	0.71058	2.90142	3.62464
С	5.10633	-2.17169	0.15030
Η	5.86930	-2.83774	-0.30462
Н	5.67246	-1.30469	0.53773
С	2.87971	2.48474	-3.58410
H	2.36546	2.71391	-4.52504
C	4.36539	-0.49308	3.22185
н u	4.JLJL9 3 01500	0.334/0	Z.JIJIZ A 12740
11	2.77703	0.0/00/	7.17/42

Н	5.36137	-0.88950	3.49053
С	4.38813	-2.91246	1.29948
Н	5.13937	-3.38710	1.96524
Н	3.78632	-3.75254	0.90318
С	4.96392	1.65024	0.12507
Н	4.61510	0.68058	0.51513
С	-0.42168	-3.85985	0.26713
Н	-1.50490	-4.01446	0.40909
Н	-0.29401	-2.87277	-0.21381
Н	-0.05530	-4.63064	-0.43391
С	4.53374	3.03799	-1.92814
Н	5.31757	3.70745	-1.55463
С	3.88278	3.34482	-3.12761
Н	4.15271	4.24026	-3.69746
С	2.93693	-3.07082	3.97642
Н	2.26348	-2.61009	4./16/8
H	2.48520	-4.02394	3.65814
н ~	3.89047	-3.30299	4.48340
C	5.1/935	-1.16/25	-2.75320
H	4.62/51	-0.84434	-3.65022
H	5.8/436	-0.35478	-2.48/84
н	5.77871	-2.05496	-3.02182
C II	4.62960	2.73174	1.1//51
H	4.904/9	3./3934	0.81870
п	5.33244	2.74455	2 11677
п С	0 21550	-5 22162	2.110//
U U	0.21550	-5.39254	2.22390
п u	_0 93732	-5 64099	2 24005
п u	-0.83732	-6 07462	2.54903
C	6 49352	1 56136	-0 07526
Н	6.76788	0.77719	-0.80028
Н	6.91195	2.51524	-0.44258
н	6.99410	1.32894	0.88150
Μα	-1.25997	0.39997	-0.53463
N	-2.83368	-0.54564	-1.59589
Ν	-2.22733	2.27427	-0.48480
С	-3.11587	2.56810	-1.45935
С	-2.03198	3.28431	0.53285
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С	-2.97581	3.41112	1.59391
С	-2.65860	-2.87992	-2.49959
С	-3.61410	1.63734	-2.40576
Н	-4.26342	2.08011	-3.16500
С	-0.89784	4.14778	0.46919
С	-1.75501	-2.46367	-3.66139
Н	-1.70160	-1.36079	-3.66482
С	-3.00847	-4.24041	-2.38362
Н	-2.62506	-4.95206	-3.12331
С	-3.99600	-2.41735	-0.47285
С	-0.76233	5.15047	1.44958
Н	0.09666	5.82606	1.40630
С	-4.17802	2.47691	1.75189
Н	-4.21783	1.81794	0.86734
С	-4.56979	-0.43013	-3.38782
Η	-4.01835	-0.87256	-4.23485
Н	-5.13699	-1.24856	-2.91785
Н	-5.27898	0.30684	-3.79162
С	-1.69597	5.29875	2.48216
H	-1.56849	6.08703	3.23171
С	-2./8256	4.42309	2.55499
TT	2 5 2 2 2 5		
H	-3.50395	4.52505	3.37363
H C	-3.50395	4.52505	3.37363
H C H	-3.50395 -3.83684 -4.10261	4.52505 -4.69815 -5.75841	3.37363 -1.35574 -1.28839
H C H C	-3.50395 -3.83684 -4.10261 0.15832	4.52505 -4.69815 -5.75841 4.00959	3.37363 -1.35574 -1.28839 -0.63027
H C H C H	-3.50395 -3.83684 -4.10261 0.15832 0.26090	4.52505 -4.69815 -5.75841 4.00959 2.92653	3.37363 -1.35574 -1.28839 -0.63027 -0.84574

НСНННСНСНННСНННСНННСНННСИНННСИННСИН	$\begin{array}{c} -4.97244\\ -2.29754\\ -2.26782\\ -3.34113\\ -1.67965\\ -4.59359\\ -3.89375\\ -3.72758\\ -4.64620\\ -3.05069\\ -4.01398\\ -0.32302\\ 0.12001\\ -0.31948\\ 0.3112\\ -5.94371\\ -5.83403\\ -6.67161\\ -6.37140\\ -4.01355\\ -3.10606\\ -3.94153\\ -4.88413\\ -4.7687\\ -5.55758\\ -3.83198\\ -5.07327\\ -5.51785\\ -5.57855\\ -5.66055\\ -6.36530\\ 1.54539\\ 1.57304\\ 1.83815\\ 2.30451\\ -0.25004\\ -1.15732\\ -0.43491\\ \end{array}$	$\begin{array}{c} -4.14465 & 0.39269 \\ -2.92370 & -5.03493 \\ -4.02306 & -5.13167 \\ -2.60833 & -5.20127 \\ -2.50726 & -5.84950 \\ -1.45507 & 0.55784 \\ -0.60372 & 0.65884 \\ 3.96175 & -1.56099 \\ 4.01769 & -0.95081 \\ 4.74904 & -1.20092 \\ 4.17653 & -2.60246 \\ -3.00316 & -3.46111 \\ -2.62693 & -2.52596 \\ -4.10619 & -3.41003 \\ -2.70297 & -4.29800 \\ -0.87578 & 0.6966 \\ -0.30207 & -0.86405 \\ -1.68732 & -0.10899 \\ -0.19995 & 0.83148 \\ 1.58165 & 3.00174 \\ 0.95731 & 2.94677 \\ 2.19386 & 3.91823 \\ 0.91231 & 3.11724 \\ -2.08825 & 1.95239 \\ -2.86140 & 1.95787 \\ -2.54141 & 2.31346 \\ -1.31665 & 2.67911 \\ 3.24500 & 1.82680 \\ 3.86535 & 2.73826 \\ 3.91456 & 0.96212 \\ 2.53794 & 1.85282 \\ 4.52082 & -0.19838 \\ 5.62414 & -0.14029 \\ 4.12332 & 0.78718 \\ 4.21148 & -0.93359 \\ 4.70132 & -1.95243 \\ 4.26037 & -2.39275 \\ 5.77786 & -1.78618 \\ \end{array}$
[A SC En En Fr Lo Se	<b>1</b> ( <sub>Xanth</sub> NON <sup>D,</sup> F (BP86) 1 thalpy 0K thalpy 29 ee Energy west Freq cond Freq	<pre>ipp)]-Zn Energy = -3245.92561827 = -3244.517730 8K = -3244.430191 298K = -3244.642189 uency = 12.4975 cm<sup>-1</sup> uency = 19.3366 cm<sup>-1</sup></pre>
Znl O N N N C C C H C C C C H C C C H C	-2.09530 0.08788 1.89298 0.43709 0.98747 -3.93878 -2.62420 2.50114 1.69676 2.27648 1.68621 3.59633 4.35191 5.37850 3.80776 4.52499 4.11139 4.94964 5.99106 4.47238	-0.45945 0.87180 -0.07035 -0.23009 0.24744 0.67807 1.73504 -0.88043 -1.76430 -0.69544 -0.92798 0.15103 -0.75286 2.82614 1.47159 0.28998 2.25856 -0.55287 3.48200 -0.96519 4.14515 -1.60417 3.83507 -0.59362 2.95785 0.21601 3.21227 0.48281 1.73899 0.67729 0.68341 1.55032 -0.69516 0.98922 -1.81154 0.80582 -1.75721 1.13295 -2.99060 0.18234

С	3.15093	-3.03331	-0.31104
	0 70000	0.01007	0 0 0 0 1 0
н	2./8828	-3.9100/	-0.85049
С	2.27591	-1.92258	-0.17119
~	0 00000		0.55054
С	2./95/9	-0.842/2	0.559/4
С	5.43223	-4.19491	0.03170
0	( (77()	2 7 7 7 6 0	0 70010
C	0.0//02	-3./0/00	-0./9012
Н	7.22740	-2.94274	-0.30713
тт	C 20CE1	2 12120	1 70004
н	0.38031	-3.43130	-1./9994
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C	5.88401	-4.00590	1.43984
Н	6.58026	-5.51977	1.35580
ц	5 01022	_1 00703	2 01300
п	J.01052	-4.90/03	2.04500
Н	6.39949	-3.86341	1.99349
C	1 77116	-5 20202	_0 60745
C	4.//410	-3.39303	-0.00/45
Н	4.45887	-5.13527	-1.71302
ы	3 80330	-5 76866	-0 14046
11	5.09250	5.70000	0.14040
Н	5.49780	-6.22328	-0.76101
C	6 05362	0 86971	1 53053
C	0.03302	0.000/1	1.55055
Н	6.54318	0.11473	2.16778
н	6 32753	1 85782	1 93581
11	0.52755	1.05/02	1.95501
Н	6.46311	0.78421	0.51069
С	4 02098	0 81551	3 01966
	0.00001	0.01001	0.01000
Н	2.92861	0.69026	3.08525
н	4 27743	1 81070	3 42181
	1.27710	1.01070	0.12101
Н	4.495/5	0.04/12	3.65404
С	4.18925	5.16294	-1.12403
~	4 0 0 0 0 0	F 10004	0 07510
C	4.23065	5.12094	-2.0/510
Н	3.22509	4.98759	-3.10714
тт	1 0 ( 1 0 0	1 20001	2 02057
н	4.86190	4.28801	-3.02957
Н	4.64802	6.06269	-3.07499
C	2 20542	6 25265	0 66016
C	5.30343	0.33303	-0.00910
Н	3.26721	6.41939	0.43192
U	2 27042	6 26210	_1 02011
п	2.2/042	0.20240	-1.03011
Н	3.71534	7.30482	-1.05401
C	5 62579	5 /1335	-0 613/1
C	5.02579	J.4135J	-0.01341
Н	5.99735	6.37382	-1.01041
ы	6 32118	1 62603	-0 9/5/7
11	0.52440	4.02005	0.94947
Н	5.66613	5.47052	0.48827
C	-0 553/1	2 60106	-1 31013
C	0.0041	2.09190	1.31013
С	-1.08347	3.62631	-0.36745
C	-2 02040	4 58152	-0 80836
~	2.02010	1.00102	0.00050
Н	-2.42221	5.30427	-0.08886
C	-2 45031	4 62101	-2 13776
	2.10001	5.05040	2.10000
Н	-3.18050	5.3/042	-2.46226
С	-1.94042	3.69198	-3.05128
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н	-2.28252	3./23/1	-4.08962
С	-0.99042	2.72485	-2.66773
C	_0 40150	1 70210	-2 71020
C	-0.40130	1./0319	-3./1930
Η	-0.17290	0.82572	-3.21465
C	-1 36766	1 48268	-4 88180
<u> </u>	1.50700	1.40200	4.00100
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н	-0.93381	0.71618	-5.54414
**	1 54010	0.000010	5.51414
Н	-1.54810	2.3/611	-5.50608
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Н	U.//100	3.31961	-4./5339
Н	1.37161	1,65837	-5.01135
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н	1.6/331	∠.48444	-3.45/16
С	-0.68527	3.63460	1.10983
	0.00000	0 70044	1 00170
н	0.00680	2./9344	1.281/6
С	-1.91125	3.41150	2.01807
	2 64700	1 22005	1 01 000
н	-2.64/93	4.22825	т. Атеа8
Н	-1.60811	3.35426	3.07689
ы	_2 /2110	2 16500	1 76510
п	-2.42118	∠.40089	7. \QJT8
С	0.04700	4.93643	1.50626
TT	0 0 5 4 0 0	E 00E01	0 00004
п	0.95429	J.U0521	0.89994
Н	0.34606	4.90088	2.56867
U	_0 60174	5 01000	1 27177
Η	-0.60474	5.81833	1.37177

~	0 (1707	0 0 0 1 0 0	0 04007
C	0.01/2/	-2.96100	-2.8428/
С	0.13637	-4.09555	-3.52626
Η	0.24710	-4.15101	-4.61388
C	-0 46666	-5 15971	-2 84625
	0.10000	6 02402	2.01020
н	-0.82620	-0.03493	-3.39830
С	-0.59777	-5.09790	-1.45599
Η	-1.06142	-5.93356	-0.91962
C	-0 14079	-3 98398	-0 72330
ä	0.20010	4.00040	0.72550
C	-0.28818	-4.00343	0./9/44
Η	0.03505	-3.01663	1.16728
С	0.62558	-5.07143	1.44284
н	0 35310	-6 08677	1 10274
11	0.55510	C.00077	2 54200
н	0.52///	-5.04955	2.54280
Η	1.68441	-4.89768	1.19186
С	-1.75279	-4.21621	1.23621
н	-2 42416	-3 44691	0 81993
U	_1 03500	-1 17496	2 33606
п	-1.03399	-4.1/490	2.33000
Н	-2.13481	-5.2019/	0.915/5
С	1.33517	-1.86444	-3.63171
Н	1.26722	-0.93405	-3.03888
С	0 69851	-1 60409	-5 01199
	0 07/10	_2 //100	_5 70005
п	0.0/412	-2.44199	-3.10995
H	1.14807	-0.70776	-5.47238
Η	-0.39006	-1.44735	-4.93840
С	2.83915	-2.19000	-3.80428
н	3 34308	-2 31537	-2 83326
11 TT	2 24040	1 27000	4 25476
н	3.34843	-1.3/920	-4.354/6
Η	2.96704	-3.12433	-4.38062
С	-4.31936	-0.49108	-1.16720
С	-4 93382	0 77708	-1 33223
ĉ	_5 31313	1 19346	-2 62437
C	-3.31313	1.10340	-2.02437
Н	-5./9821	2.16004	-2./4563
С	-5.09407	0.37844	-3.75627
С	-4.46335	-0.86449	-3.56510
н	-4 28136	-1 51352	-4 43112
	4.00100	1 21 674	2 20277
C	-4.06361	-1.310/4	-2.29211
С	-3.40100	-2.66431	-2.13490
Η	-3.27251	-3.15855	-3.11044
Н	-3.98282	-3.34309	-1.48566
н	-2 40281	-2 57557	-1 67215
C	-5 50103	0 94455	-5 13964
<u> </u>	-3.30193	0.04455	-5.15004
Н	-6.3/001	1.52355	-5.09614
Η	-5.76514	-0.00526	-5.79069
Н	-4.68202	1.39457	-5.63702
С	-5 18069	1 68347	-0 14651
ц	-5 70306	2 55169	_0 /2051
п	-3.79300	2.33100	-0.43651
Н	-4.22938	2.0/189	0.26051
Η	-5.69281	1.16169	0.68026
С	-6.16380	-2.01860	0.25069
Н	-6.77428	-1.11909	0.05666
н	-6 73216	-2 68201	0 01010
п	0.13210	2.00294	0.91019
Н	-6.02896	-2.51793	-0./2276
С	-4.83085	-1.63115	0.87463
С	-4.62141	-2.02413	2.21949
н	-5,38511	-2.68601	2.63495
C	-3 65724	-1 56521	3 1/210
C	-3.03/24	-1.000001	J.14318
С	-3.82693	-2.02522	4.58609
Η	-2.85996	-2.33625	5.01516
Н	-4.53034	-2.86967	4.64214
н	-4,21317	-1.21886	5,23256
C	-1 05/0/	-0 06756	3 20000
ć	-1.90404	1 00051	3.09000
С	-2.60327	1.02351	4.54120
С	-1.92677	1.70823	5.56820
Н	-2.43438	2.54808	6.05994
С	-0.62925	1.35537	5,97831
č	-0 01160	0 27231	5 32010
	0.01109	0.27201	5.52019
Н	0.99450	-0.03/12	5.63826
С	-0.64579	-0.44482	4.29729

С	0.05432	-1.61756	3.65367
Н	-0.56365	-2.53170	3.68180
Н	1.01255	-1.83117	4.15340
Н	0.26759	-1.42039	2.58751
С	0.08873	2.13028	7.06318
Н	-0.62160	2.58279	7.77519
Н	0.69287	2.95350	6.63815
Н	0.77774	1.48439	7.63309
С	-3.99759	1.45917	4.14272
Н	-4.77822	0.78298	4.53804
Н	-4.12089	1.46822	3.04751
Н	-4.21430	2.46877	4.52726

# [Al(XanthNON<sup>Dipp</sup>)]-Li

SCF (BP86) Energy = -2490.06405633 Enthalpy 0K = -2488.838766 Enthalpy 298K = -2488.763651 Free Energy 298K = -2488.953241 Lowest Frequency = 10.0289 cm<sup>-1</sup> Second Frequency = 11.5685 cm<sup>-1</sup>

Al	0.56100	-0.29901	0.20918
0	-1.34336	0.52827	0.97064
Ν	0.63424	1.70127	-0.17997
Ν	-0.87311	-1.68938	-0.24116
С	-1.51465	1.91516	0.80559
С	-0.40666	2.54369	0.19979
С	-0.51085	3.94961	0.05119
Н	0.33494	4.48924	-0.38587
С	-1.68747	4.64692	0.41737
С	-2.78682	3.92838	0.93247
Н	-3.71215	4.45091	1.17851
С	-2.71713	2.53034	1.13222
С	-3.85318	1.64097	1.67812
С	-3.76452	0.29100	0.93455
С	-4.86235	-0.49850	0.52292
Н	-5.87334	-0.13503	0.71127
С	-4.66887	-1.73090	-0.13767
С	-3.35254	-2.18468	-0.39226
Н	-3.18782	-3.13139	-0.91600
С	-2.21742	-1.42242	-0.01558
С	-2.49851	-0.21330	0.65662
С	-5.86224	-2.58722	-0.62666
С	-7.22697	-1.96767	-0.25152
Н	-7.36668	-0.97300	-0.70901
Н	-7.34423	-1.86493	0.84132
Н	-8.04204	-2.61751	-0.61501
С	-5.80144	-2.71375	-2.17249
Н	-6.64294	-3.32652	-2.54411
Н	-4.86411	-3.18919	-2.50564
Н	-5.86228	-1.72078	-2.65023
С	-5.79128	-4.00285	0.00292
Н	-6.63260	-4.62526	-0.35250
Н	-5.84798	-3.94820	1.10388
Н	-4.8550/	-4.52217	-0.26007
C	-5.23316	2.30345	1.50086
H	-5.45538	2.505//	0.44024
H	-5.28084	3.23342	2.05508
н	-0.02866	1 20062	1.90/63
	-3.60946	1.38962	3.19//1
п u	-4.30933	2 24414	3 75105
п u	-2 62881	0 91777	3 37062
C 11	_1 7/000	6 1 9 1 0 9	0 21260
C	-0 60883	6 85022	1 04300
н	-0 62943	7 94719	1 90920
Н	0 38750	6 49305	0 73367
Н	-0.72555	6.63430	2.11914
**	5.,2000	5.05.50	- •

0	1 54501	C E2000	1 20200
C	-1.54521	6.52099	-1.28269
Η	-2.34090	6.06369	-1.89567
Η	-0.57649	6.15616	-1.66246
н	-1 57745	7 61473	-1 43886
~	2.00520	6 70000	1.15000
C	-3.08520	6./8960	0.6/422
Η	-3.93496	6.38741	0.09587
Η	-3.06843	7.88300	0.52276
н	-3 27761	6 60383	1 74516
~	1 04226	2 22202	0 60000
Ĉ	1.04320	2.33302	-0.60666
С	2./9/94	2.//91/	0.35915
С	3.97778	3.40950	-0.08350
Н	4,70404	3.76528	0.65712
C	4 23629	3 59868	-1 44684
		1 1 0 0 0 5	1 77140
Н	5.15452	4.10095	-1.//143
С	3.30688	3.14675	-2.39240
Η	3.50794	3.30142	-3.45841
С	2,10976	2.51232	-2.00183
C	1 00962	2 09171	-3 06530
	1.09002	2.001/1	5.00555
Н	0.416/2	1.35980	-2.58164
С	1.75534	1.38657	-4.27536
Н	2.39133	2.07818	-4.85666
н	0 97862	1 01114	-4 96436
ц	2 27070	0 52100	-3 06526
п	2.3/9/0	0.55169	-3.90320
С	0.24262	3.28085	-3.53822
Η	-0.29869	3.74109	-2.69658
Н	-0.50194	2.95544	-4.28647
н	0 87722	4 05570	-4 00589
0	0.07722	2 (101)	1 0 0 1 1 0
C	2.55/11	2.01010	1.80112
Η	1.64844	2.00243	1.97653
С	3.71418	1.87153	2.55919
Н	3.89351	0.89122	2.08314
н	3 47789	1 70546	3 62565
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н	4.65921	2.44201	2.51030
С	2.30182	3.98050	2.54550
Η	3.18005	4.64508	2.45306
Н	2.09641	3,84250	3,62208
н	1 43564	4 49164	2 09611
 	1.4004	2 00101	2.00011
C	-0.51063	-3.02191	-0.60316
С	-0.07540	-3.28394	-1.94525
С	0.28159	-4.59361	-2.31786
Н	0,60082	-4.79286	-3.34566
C	0 22487	-5 65067	-1 40158
	0.22407	6 6 6 7 2 9	1 70705
н	0.49512	-0.00/28	-1.70795
С	-0.16585	-5.38644	-0.08647
Η	-0.19185	-6.20707	0.64252
С	-0.52541	-4.09124	0.35149
C	-0 92038	-4 02132	1 84230
ц	-0 27536	_1 79756	2 32100
п	-0.27550	-4.70750	2.52100
С	-0.65684	-2./1260	2.61112
Η	0.33206	-2.28378	2.36961
Η	-0.69604	-2.91818	3.69706
н	-1,41440	-1.94326	2.39636
C	-2 37602	-1 10506	2 07307
C	-2.37002	-4.49590	2.07307
Н	-2.58154	-4.5843/	3.15593
Η	-2.55635	-5.48056	1.60982
Н	-3.09413	-3.77851	1.64581
С	-0.06072	-2.16741	-2.99246
ц	0 20214	-1 22280	-2 46300
11 C	0.20014	1.23209	4 11000
C	0.9/445	-2.38481	-4.11263
Η	1.98221	-2.58414	-3.70782
Η	1.03358	-1.48758	-4.75062
Н	0.70263	-3.22941	-4.77112
C	-1 46789	-1 95825	-3 60067
	1 20222	1.90020	1 1 2 1 2 2
Н	-1./9//9	-2.86987	-4.13138
Η	-1.45832	-1.12686	-4.32800
Η	-2.21267	-1.72547	-2.82365
0	4.79925	-0.98169	-0.10405
0	3,48466	-2.86208	1.99389
0	J. 10100	2.00200	±•))))))

С	5.02147	-0.83543	-2.57161
Н	5.95016	-1.42850	-2.62934
Н	5.04033	-0.10861	-3.40137
Н	4.16445	-1.51130	-2.72932
С	4.88277	-0.08473	-1.24960
Н	3.94603	0.49491	-1.22141
Н	5.71520	0.62469	-1.09608
С	6.06456	-1.53554	0.33024
Н	6.61628	-1.92878	-0.54511
Н	5.78617	-2.38863	0.96981
С	6.91998	-0.53834	1.11300
Η	7.21873	0.32768	0.50012
Η	7.84387	-1.03449	1.45781
Н	6.37671	-0.16267	1.99522
С	3.39667	-4.49667	0.21203
Η	2.68808	-3.89998	-0.38845
Η	3.19426	-5.55471	-0.01857
Η	4.42940	-4.26623	-0.09849
С	3.19866	-4.25372	1.70060
Н	2.15327	-4.47130	1.99488
Н	3.87142	-4.89826	2.29852
С	3.27994	-2.50713	3.38545
Н	2.31202	-2.92586	3.72278
Н	3.18358	-1.40758	3.39043
С	4.42684	-2.95314	4.29213
Н	4.53094	-4.05023	4.32377
Н	4.23690	-2.60938	5.32332
Н	5.38461	-2.52108	3.95848
Li	3.04533	-1.45362	0.64391