

Supporting Information

Boroxine Template for Macrocyclization and Postfunctionalization

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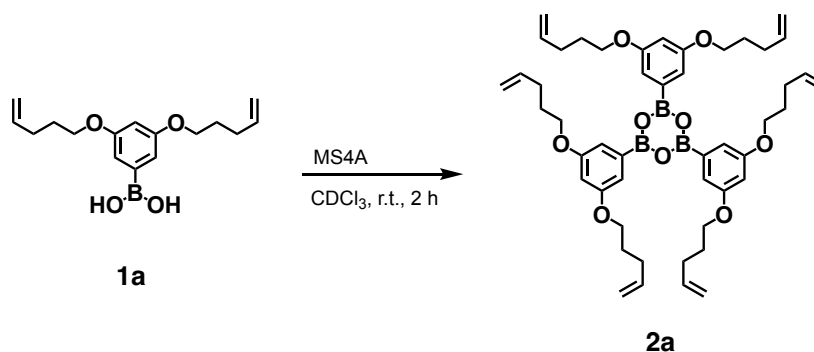
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1. General Methods

^1H NMR spectra were recorded on a Bruker BioSpin AVANCE DPX-400 (400 MHz) spectrometer and Bruker BioSpin AVANCE III (400 MHz) spectrometer and ^{13}C NMR spectra were recorded on a Bruker BioSpin AVANCE 400 M (100 MHz) spectrometer using residual CHCl_3 (^1H , 7.26 ppm), DMSO (^1H , 2.50 ppm), or DMSO- d_6 (^{13}C , 39.5 ppm) as an internal standard. All NMR spectra were recorded at 303 K unless otherwise noted. ^{11}B NMR spectra were recorded on a JEOL ECX-500 (160 MHz) spectrometer using $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (0.00 ppm) as an external standard. High-resolution mass analysis (MALDI⁺) was performed on a JEOL JMS-S3000 mass spectrometer using DCTB (*trans*-2-[3-(4-*tert*-Butylphenyl)-2-methyl-2-propenylidene]malononitrile), DHB (2,5-dihydroxybenzoic acid) and Dithranol as matrix. IR spectra were recorded on a JASCO FT/IR-4600 plus spectrometer. The X-ray analysis data were obtained using a Rigaku R-Axis RAPID charge-coupled device (CCD) apparatus (Cu κ α radiation, $\lambda = 1.54178 \text{ \AA}$). Preparative recycling gel permeation chromatography (GPC) was carried out on a JAI LC-908 equipped with JAIGEL -1HH and -2HH columns (CHCl_3 as an eluent; flow rate: 5.5 mL/min). All reactions except the catalytic hydrogen reduction were carried out under Ar unless otherwise noted. Tetrahydrofuran (THF) was purified by solvent purification system of Glass-Contour. Activated molecular sieves 4 \AA (MS 4A) was added 5 mg per 1 ml of solvent. All commercially available compounds were used without further purification unless otherwise indicated. Dehydrated dichloromethane was purchased from Kanto Chemical Co., Inc. 5-bromo-1-pentene was purchased from Kanto Chemical Co., Inc, 6-bromo-1-hexene was purchased from Oakwood Products, Inc, 7-bromo-1-heptene was purchased from Combi-Blocks, Inc. 5-bromoresorcinol^{S1} and 3,5-bis(pent-5-en-1-yloxy)phenylboronic acid (**1a**)^{S2} were prepared according to the literature procedures.

2. Synthesis of Macrocycle 4

2-1. Boroxine Formation from Boronic acid **1a** to **2a**



To a CDCl₃ (600 μ L) solution of boronic acid **1a** (1.76 mg) in an NMR tube was added MS4A. After standing at room temperature for 2 h, ¹H NMR spectrum of the solution revealed the quantitative formation of boroxine **2a**.

Physical data of **2a**

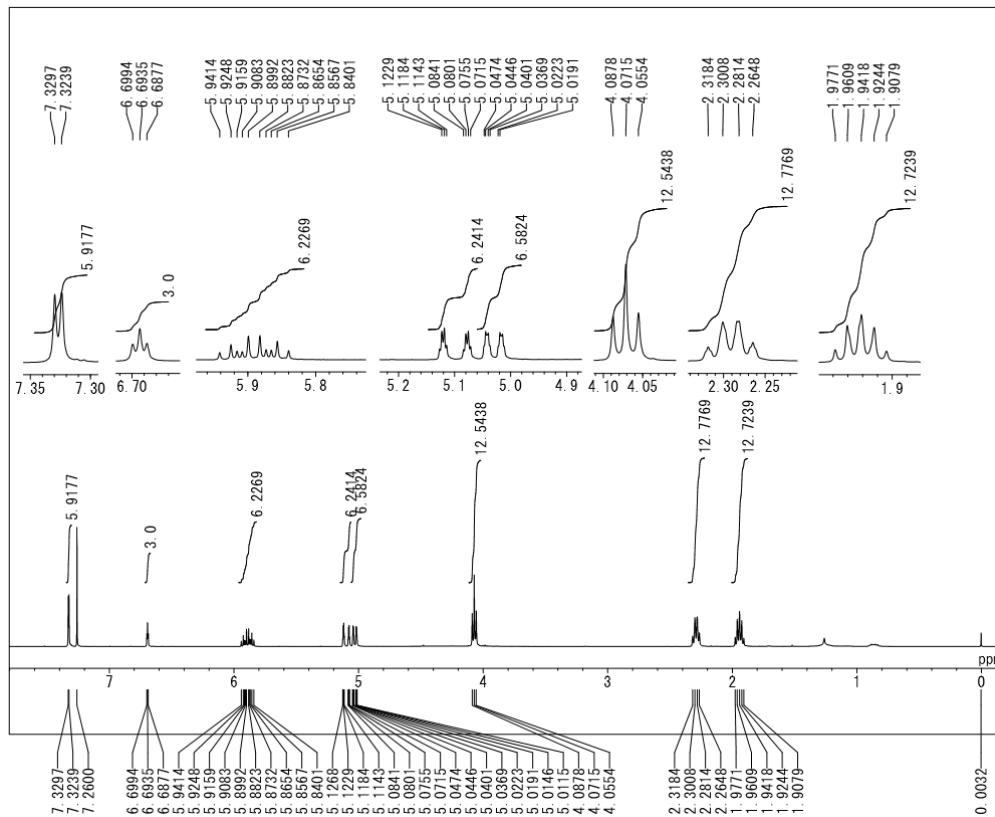
¹H NMR (400 MHz, CDCl₃): δ 7.33 (d, J = 2.3 Hz, 6H), 6.69 (t, J = 2.3 Hz, 3H), 5.89 (ddt, J = 17.0, 10.2, 6.6 Hz, 6H), 5.10 (dq, J = 17.0, 1.6 Hz, 6H), 5.05-5.02 (m, 6H), 4.07 (t, J = 6.5 Hz, 12H), 2.32-2.26 (m, 12H), 1.94 (quin, J = 6.9 Hz, 12H).

¹³C NMR (100 MHz, CDCl₃): δ 160.2, 137.9, 115.4, 113.8, 106.0, 67.5, 30.3, 28.6. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 30.2

HRMS (MALDI⁺): m/z calcd. for C₄₈H₆₃B₃O₉Na: 839.4666, found: 839.4651 [M+Na]⁺.

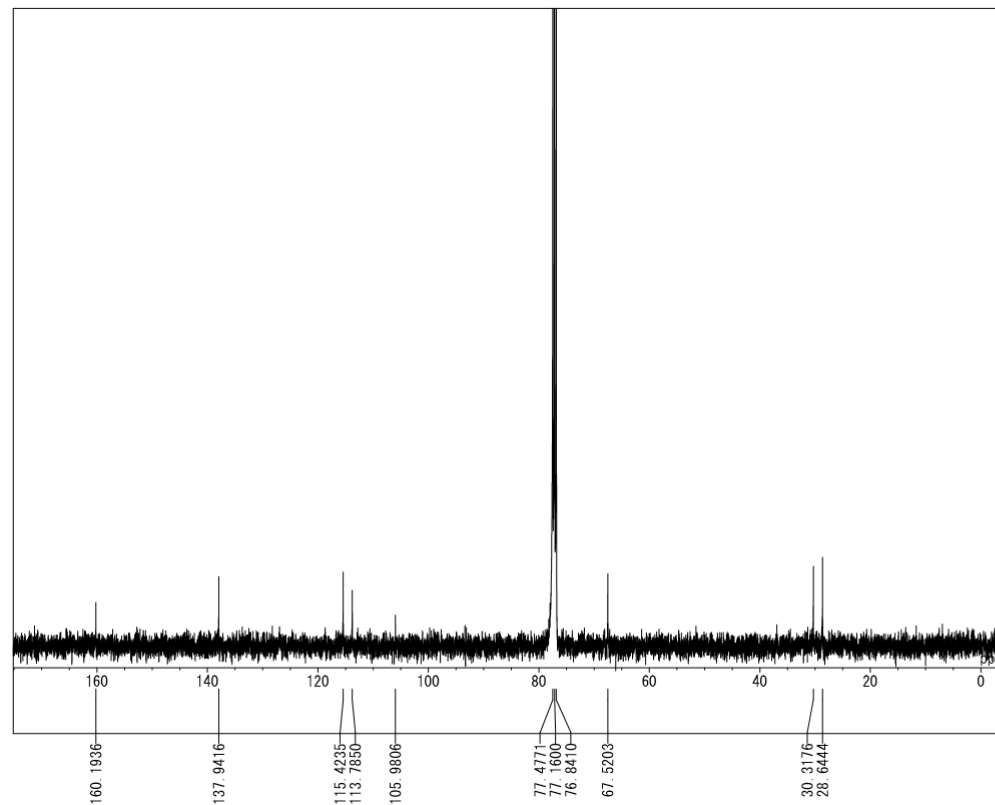
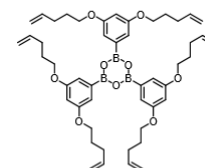
IR (ATR): 3077, 2920, 2872, 1641, 1585, 1433, 1332, 1297, 1162, 1057, 991, 908, 847, 725, 680 cm⁻¹.



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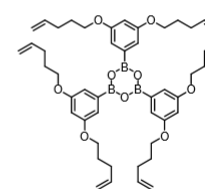
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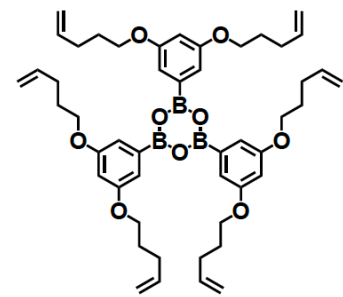
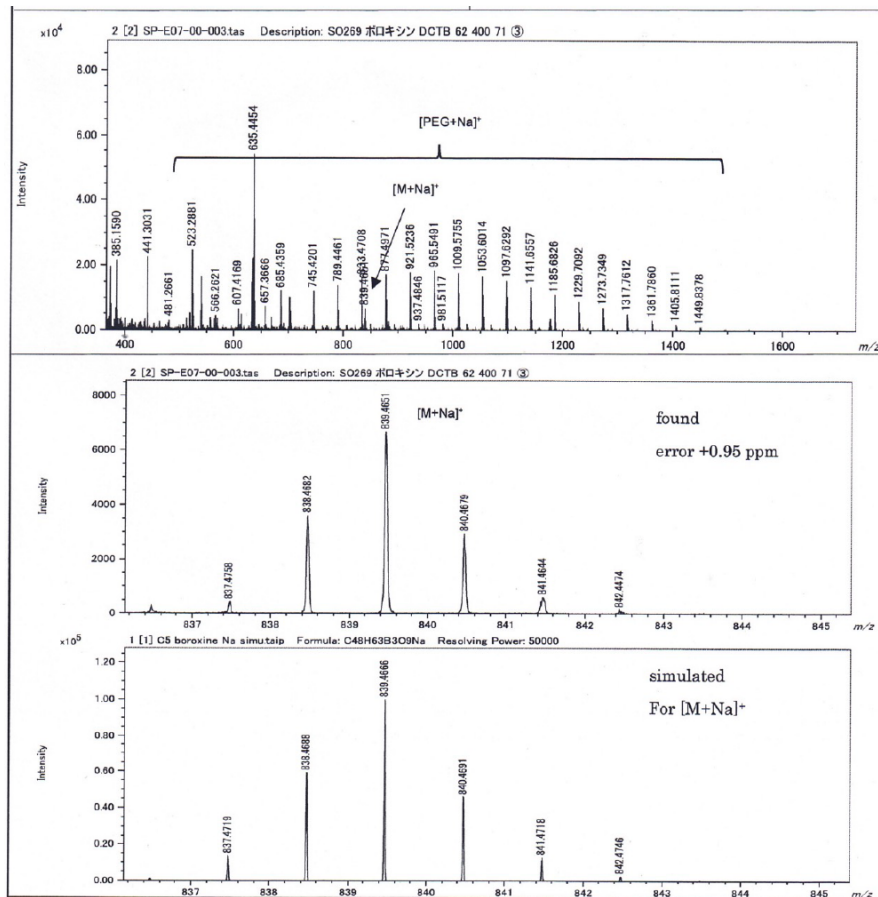
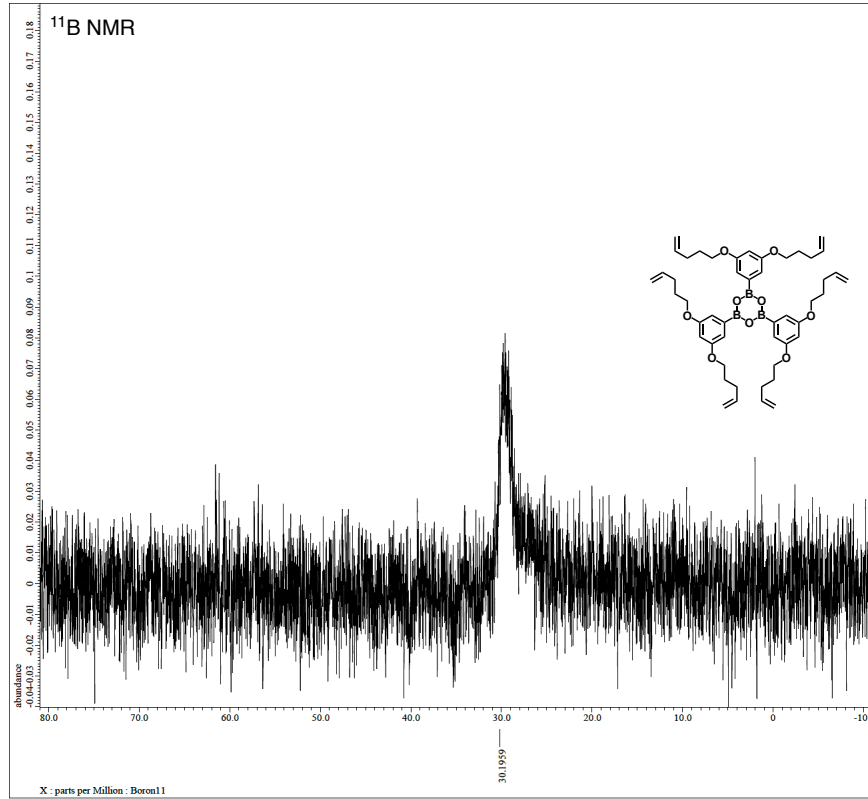


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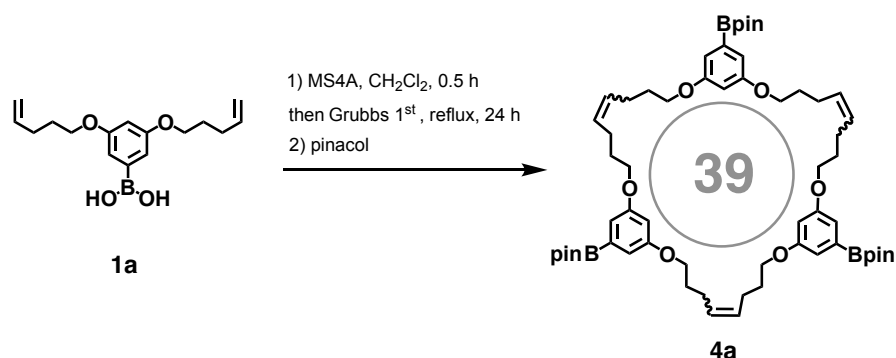
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¹³C NMR





2-2. Synthesis of Macrocycle **4a**



Boronic acid **1a** (69.0 mg, 0.238 mmol) was dissolved in dichloromethane (159 mL) with MS4A and the solution was stirred at room temperature for 0.5 h. Then Grubbs 1st Generation catalyst (19.6 mg, 23.8 μ mol) was added and the resulting mixture was refluxed for 24 h under Ar. The reaction was quenched with ethyl vinyl ether (200 μ L) and the solvent was evaporated to crude **3a**. The crude was treated with pinacol (28.1 mg, 0.238 mmol) in a mixture of methanol and chloroform and MS4A was removed by filtration. The filtrate was evaporated and the resulting crude product was purified by GPC to afford the desired **4a** (26.4 mg, 32%).

Physical data of **4a** (mixture of *cis/trans* isomers)

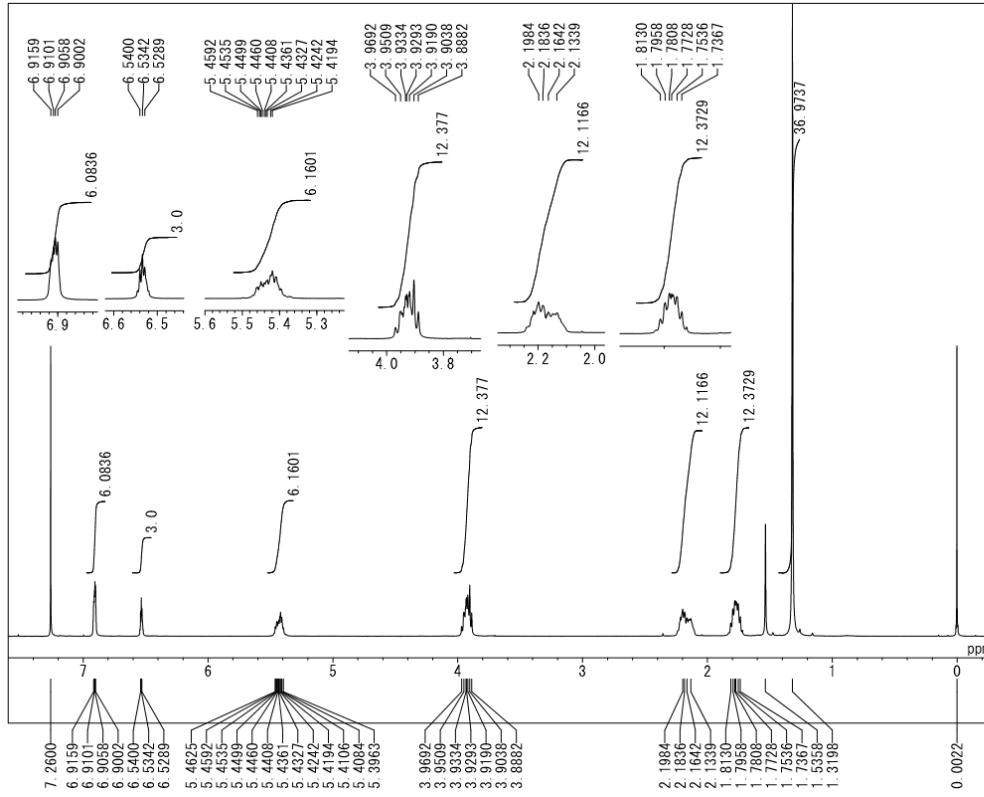
¹H NMR (400 MHz, CDCl₃): δ 6.92-6.90 (m, 6H), 6.54-6.53 (m, 3H), 5.46-5.42 (m, 6H), 3.97-3.89 (m, 12H), 2.20-2.13 (m, 12H), 1.81-1.74 (m, 12H), 1.32 (s, 36H).

¹³C NMR (100 MHz, CDCl₃): δ 160.0, 130.4, 130.3, 130.2, 129.9, 129.8, 129.7, 113.1, 112.9, 112.8, 112.7, 112.6, 112.5, 112.5, 112.3, 105.2, 105.2, 105.1, 105.0, 83.9, 67.3, 67.2, 67.2, 67.2, 29.4, 29.3, 29.3, 29.2, 29.2, 29.2, 29.1, 29.0, 29.0, 28.9, 25.0, 23.8, 23.8, 23.7. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 29.7

HRMS (MALDI⁺): *m/z* calcd. for C₆₀H₈₇B₃O₁₂Na: 1055.6396, found: 1055.6317 [M+Na]⁺.

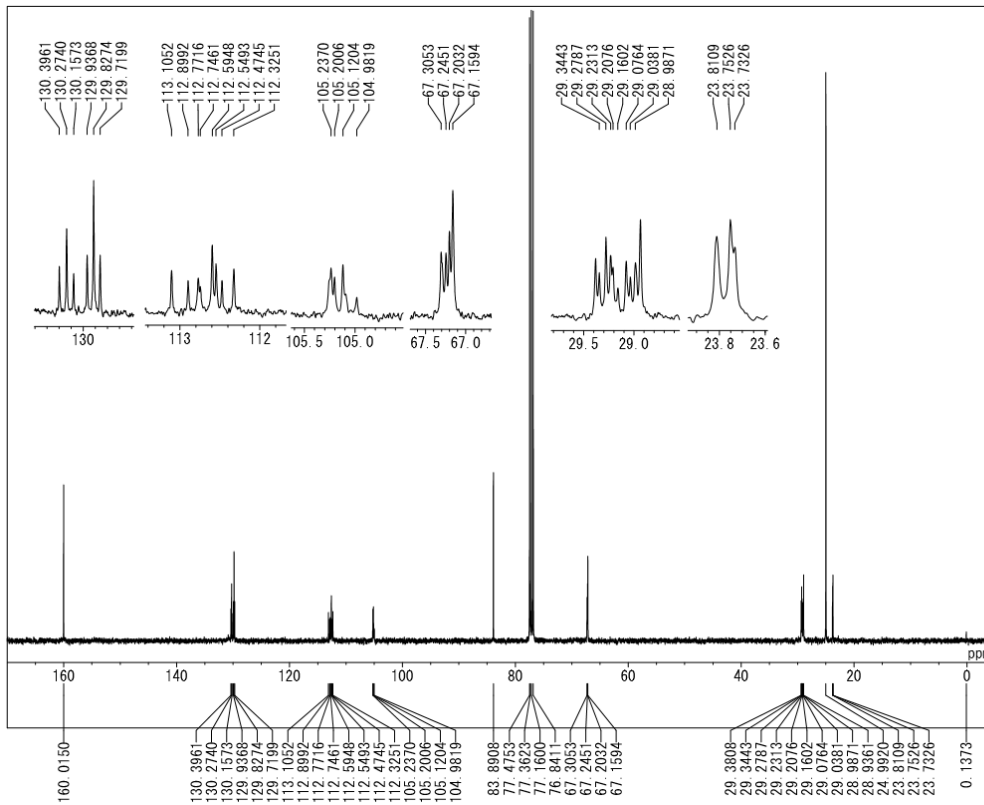
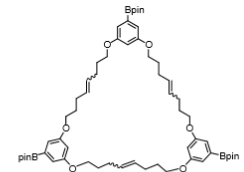
IR (ATR): 2977, 2931, 2871, 1716, 1586, 1469, 1429, 1362, 1308, 1164, 1146, 1056, 969, 851, 706 cm⁻¹.



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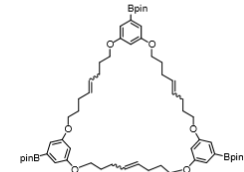
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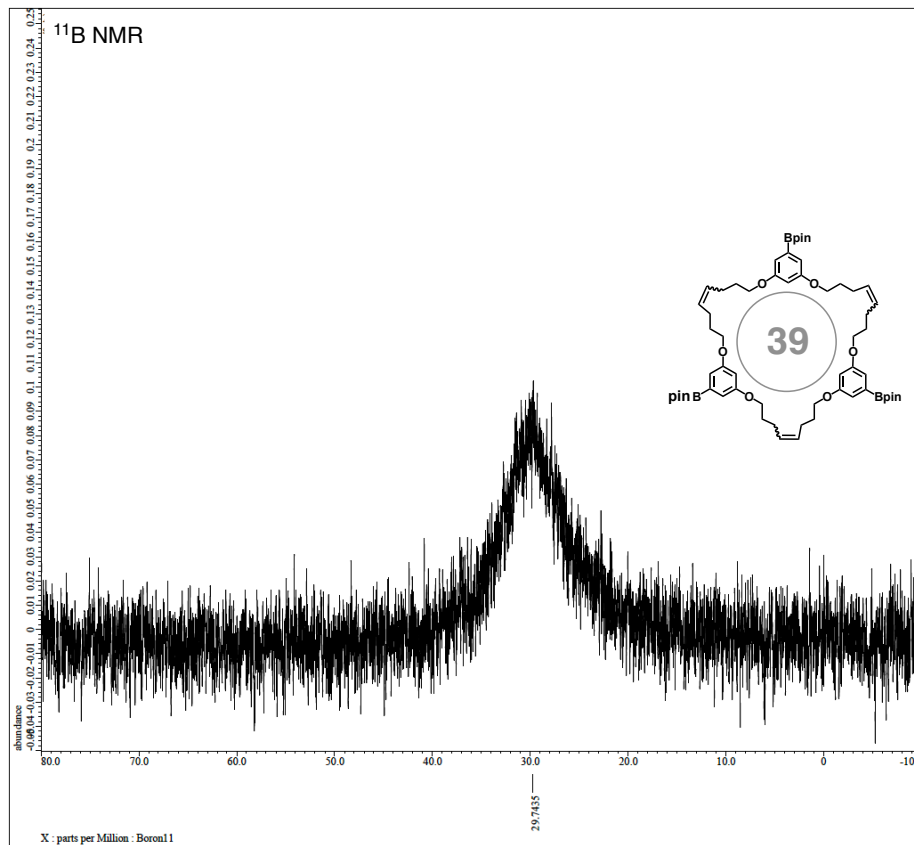


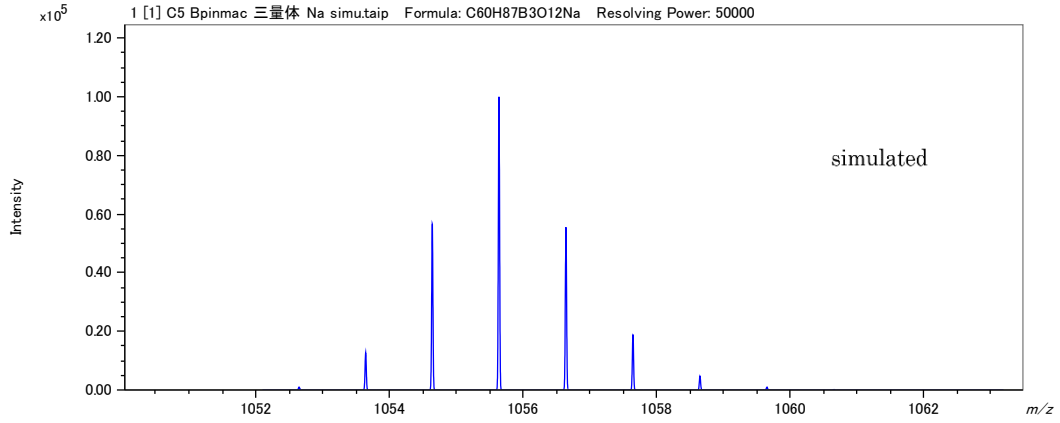
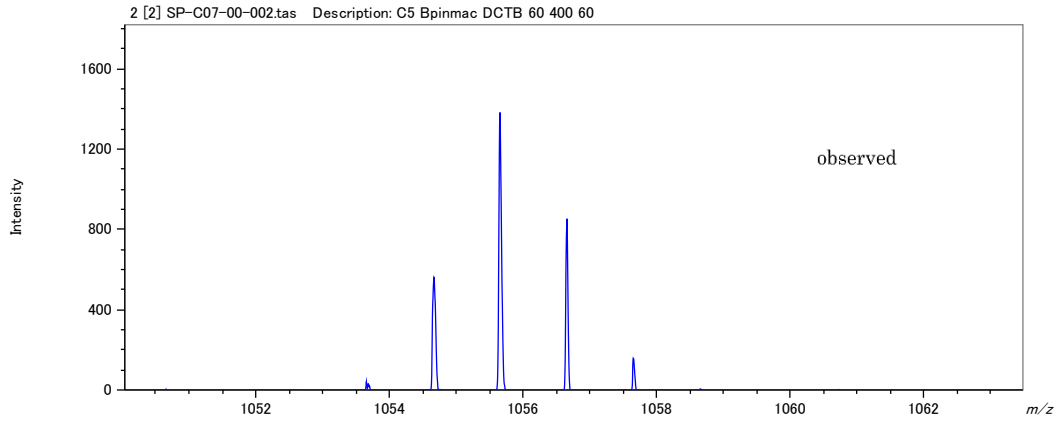
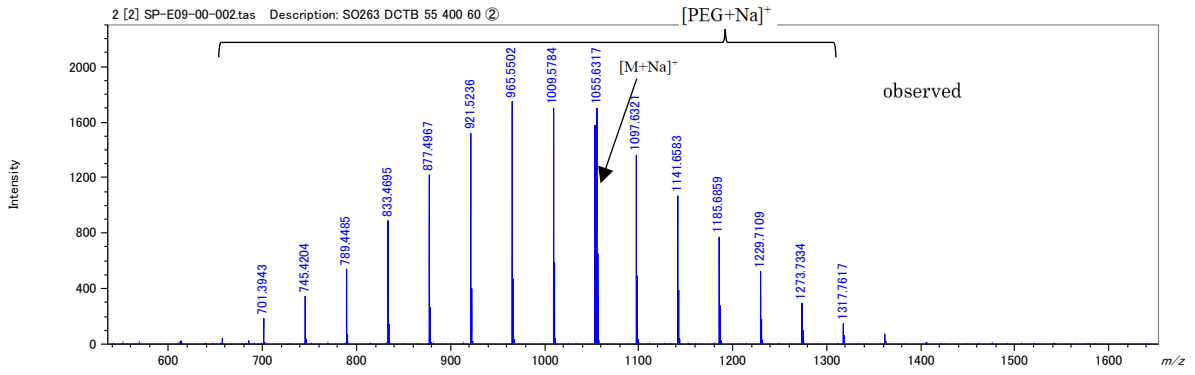
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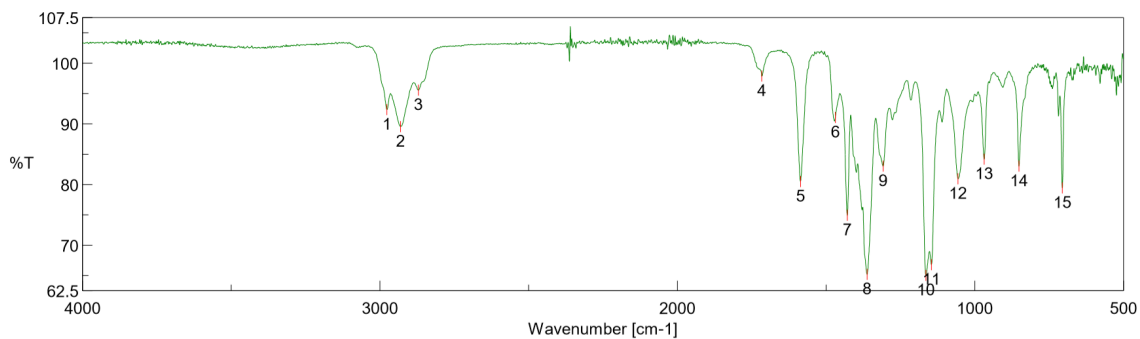
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 RGAIN 114

¹³C NMR





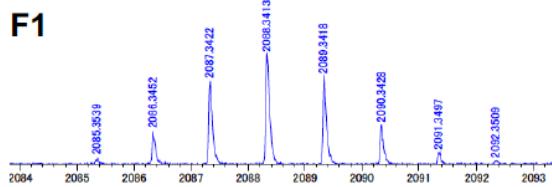
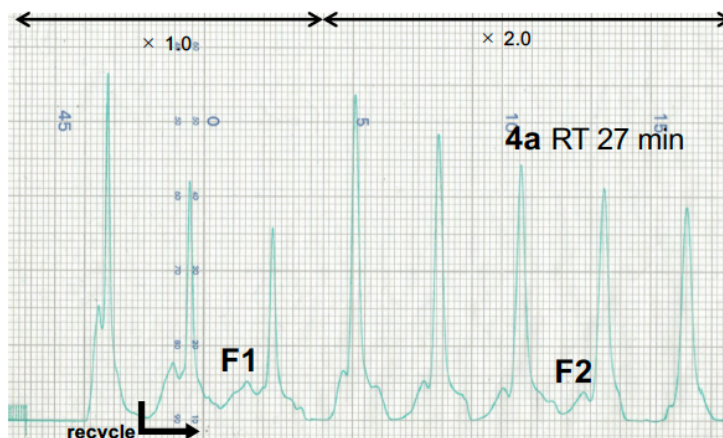




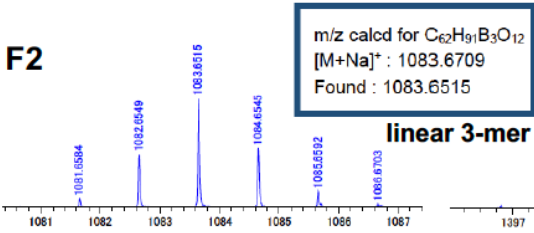
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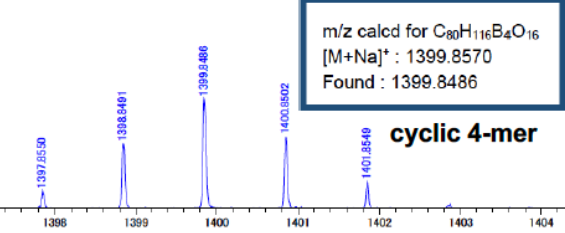
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m/z calcd for C₁₂₀H₁₇₄B₆O₂₄
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 Found : 2088.3413
cyclic 6-mer



m/z calcd for C₆₂H₉₁B₃O₁₂
 [M+Na]⁺ : 1083.6709
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linear 3-mer



m/z calcd for C₈₀H₁₁₆B₄O₁₆
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Fig. S1 GPC traces of crude **4a** and MALDI-TOF MS analysis of fractions F1 and F2.

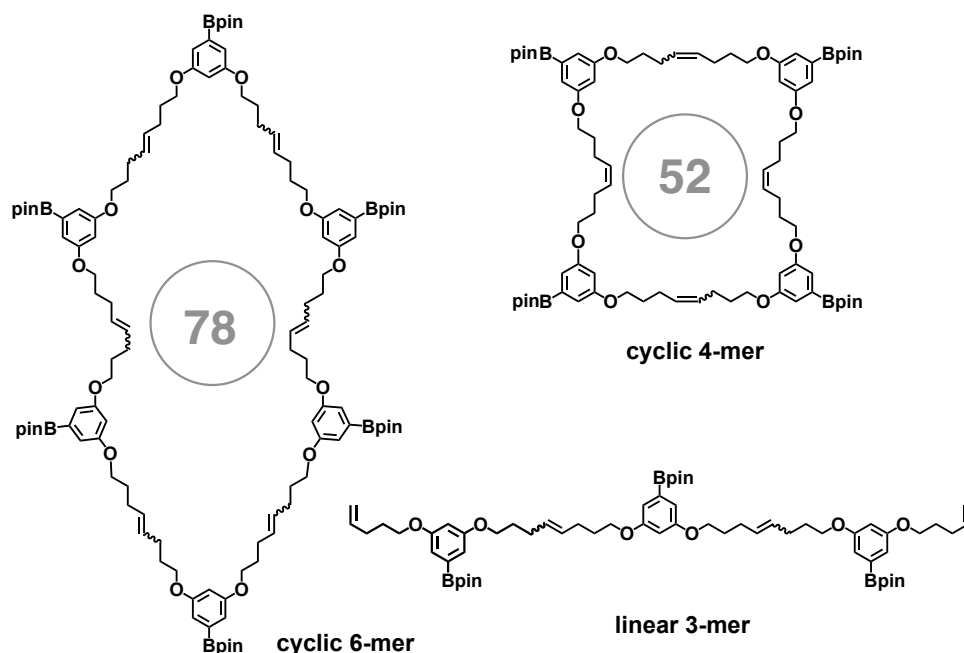
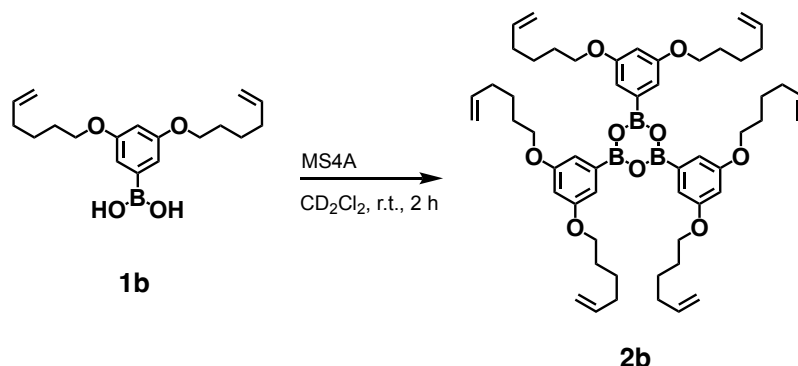


Fig. S2 Chemical structures of cyclic 6-mer, cyclic 4-mer and linear 3-mer.

2-3. Boroxine Formation from Boronic acid **1b** to **2b**



To a CDCl_3 (600 μL) solution of boronic acid **1b** (2.1 mg) in an NMR tube was added MS4A. After standing at room temperature for 2 h, ^1H NMR spectrum of the solution revealed the quantitative formation of boroxine **2b**.

Physical data of **2b**

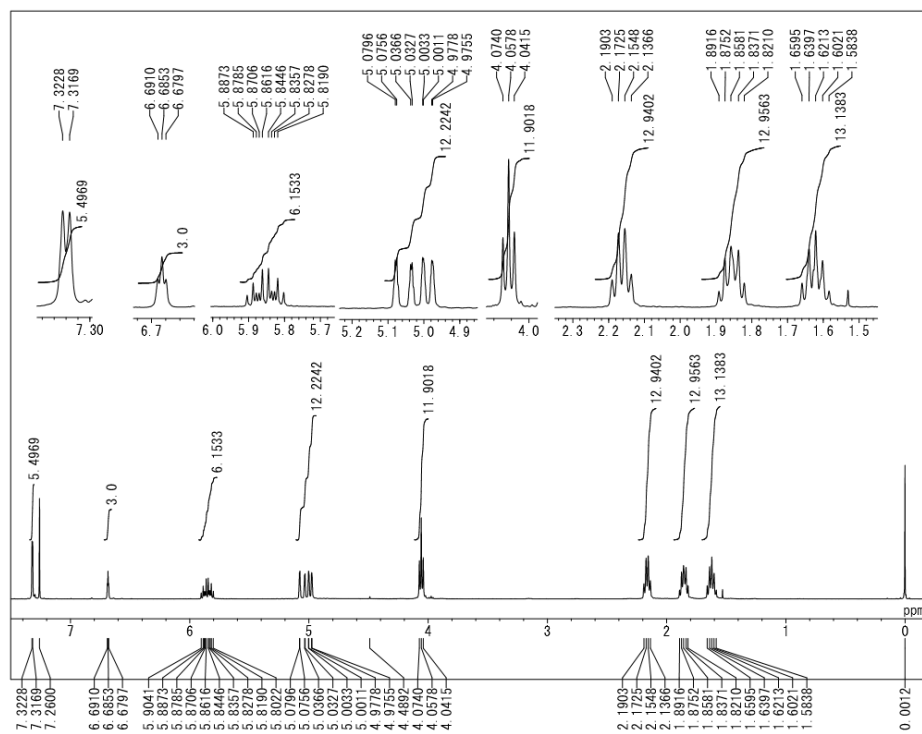
^1H NMR (400 MHz, CDCl_3): δ 7.32 (d, $J = 2.3$ Hz, 6H), 6.69 (t, $J = 2.3$ Hz, 3H), 5.85 (ddt, $J = 17.1, 10.3, 6.7$ Hz, 6H), 5.08-5.03 (m, 6H), 5.00-4.98 (m, 6H), 4.06 (t, $J = 6.5$ Hz, 12H), 2.19-2.14 (m, 12H), 1.89-1.82 (m, 12H), 1.66-1.58 (m, 12H).

^{13}C NMR (100 MHz, CDCl_3): δ 160.2, 138.7, 114.9, 113.7, 105.9, 68.1, 33.6, 28.9, 25.5. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, CDCl_3): δ 28.6

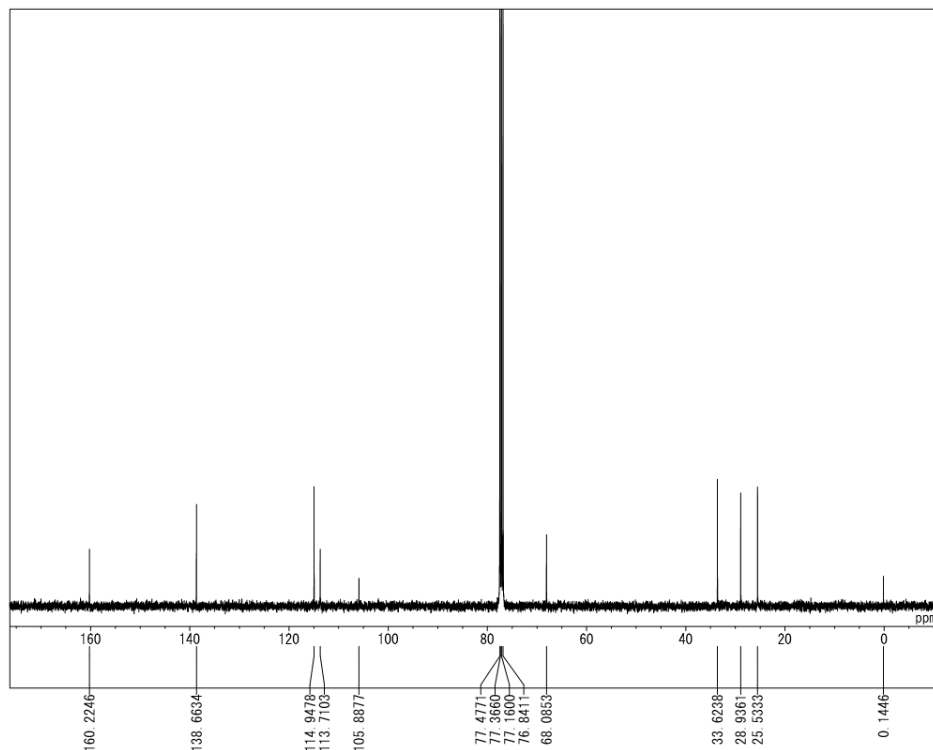
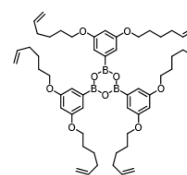
HRMS (MALDI⁺): *m/z* calcd. for C₅₄H₇₅B₃O₉Na: 923.5607, found: 923.5546 [M+Na]⁺.

IR (ATR): 3075, 2942, 2912, 2871, 1641, 1585, 1432, 1393, 1332, 1295, 1168, 1043, 993, 909, 849, 728 678 cm⁻¹.



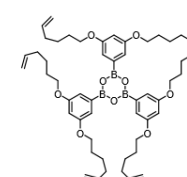
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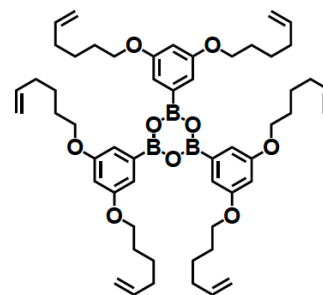
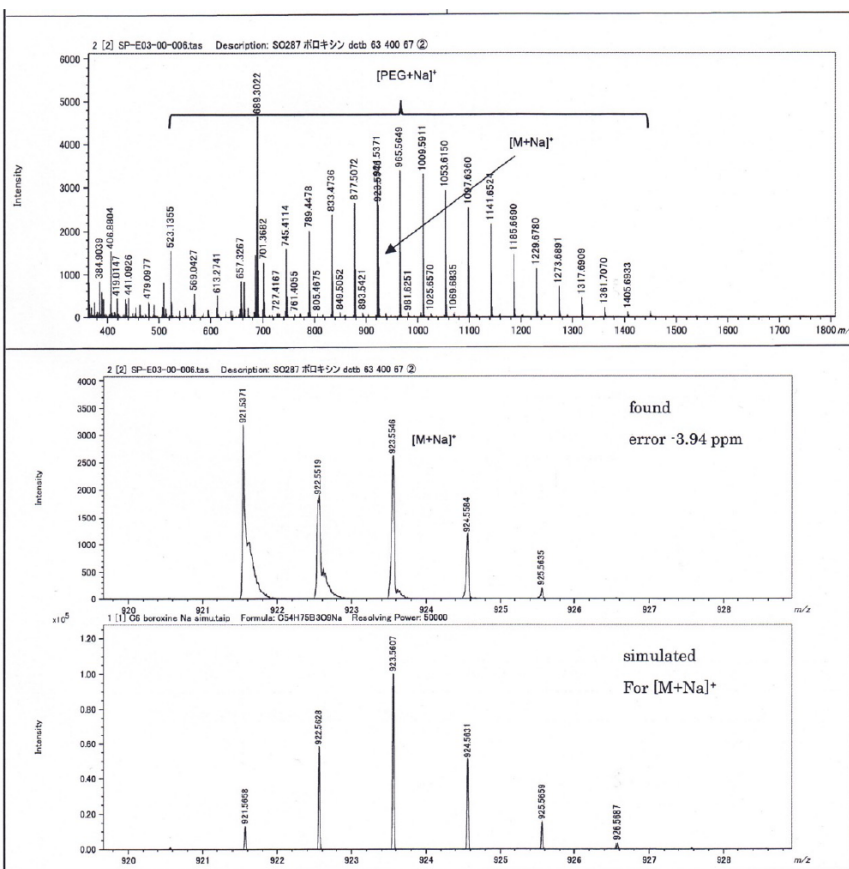
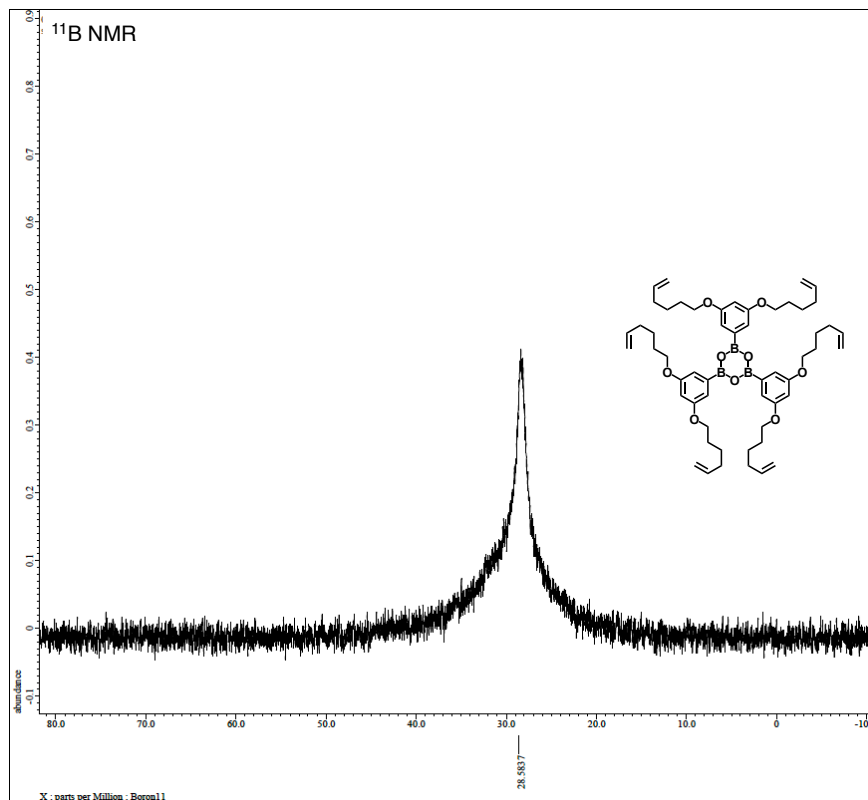
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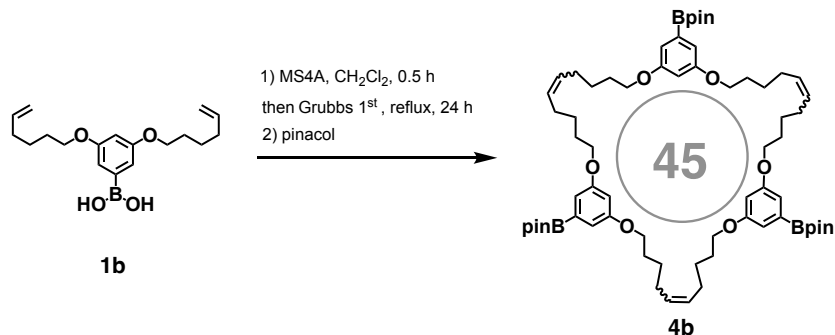
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 RGAIN 72

¹³C NMR





2-4. Synthesis of Macrocycle **4b**



Boronic acid **1b** (79.6 mg, 0.25 mmol) was dissolved in dichloromethane (167 mL) with MS4A and the solution was stirred at room temperature for 0.5 h. Then Grubbs 1st Generation catalyst (20.6 mg, 0.025 mmol) was added and the resulting mixture was refluxed for 24 h under Ar. The reaction was quenched with ethyl vinyl ether (200 μ L) and the solvent was evaporated to crude **3b**. The crude was treated with pinacol (29.5 mg, 0.25 mmol) in a mixture of methanol and chloroform and MS4A was removed by filtration. The filtrate was evaporated and the resulting crude product was purified by GPC to afford the desired **4b** (73.5 mg, 79%).

Physical data of **4b** (*cis/trans* isomers)

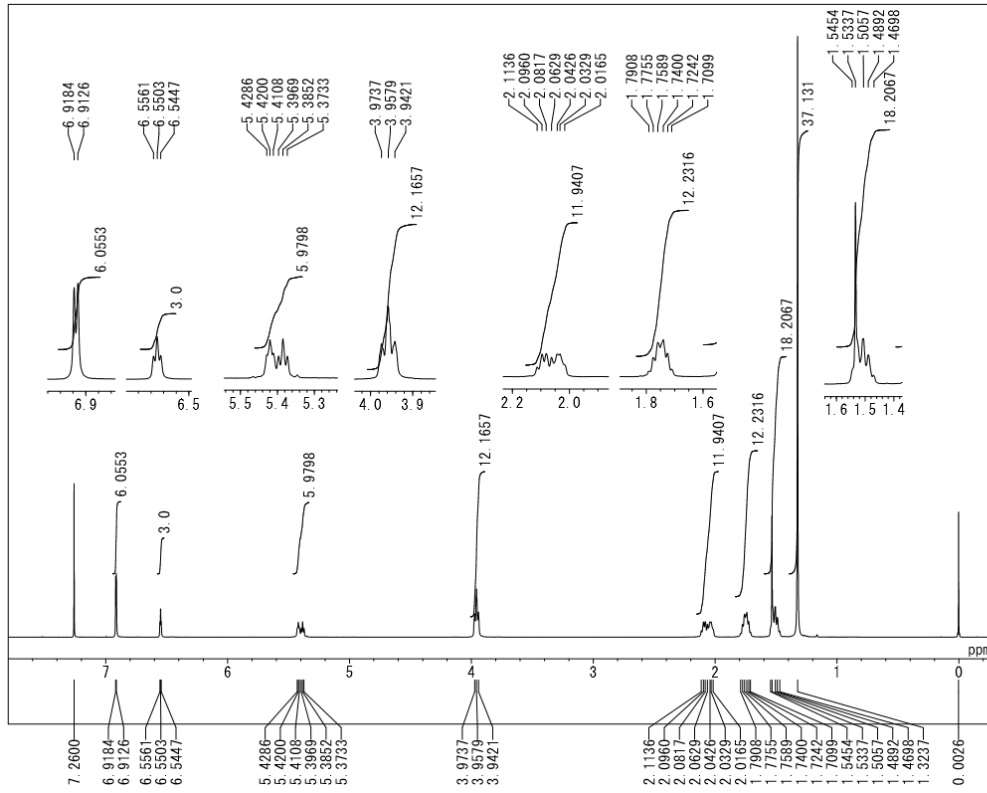
¹H NMR (400 MHz, CDCl₃): δ 6.92 (d, J = 2.3 Hz, 6H), 6.55 (t, J = 2.3 Hz, 3H), 5.43-5.37 (m, 6H), 3.96 (t, J = 6.3 Hz, 12H), 2.11-2.02 (m, 12H), 1.79-1.71 (m, 12H), 1.54-1.47 (m, 12H), 1.32 (s, 36H).

¹³C NMR (100 MHz, CDCl₃): δ 160.1, 130.5, 130.0, 112.6, 105.3, 83.9, 68.0, 32.3, 29.0, 28.9, 27.0, 26.3, 26.1, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 29.7

HRMS (MALDI⁺): m/z calcd. for C₆₆H₉₉B₃O₁₂Na: 1139.7337, found: 1139.7288 [M+Na]⁺.

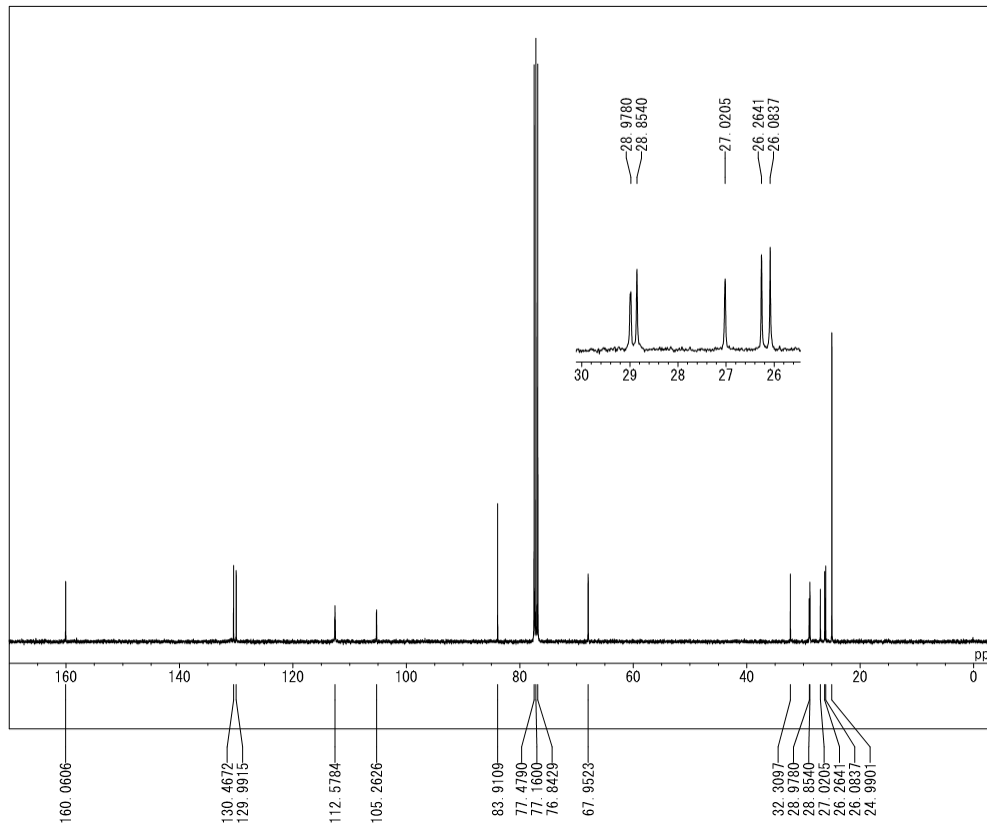
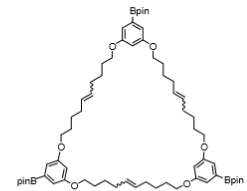
IR (ATR): 2976, 2931, 2867, 1716, 1585, 1428, 1359, 1308, 1145, 1060, 968, 851, 706 cm⁻¹.



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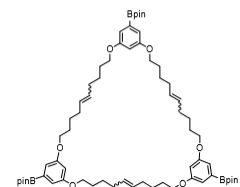
¹H NMR

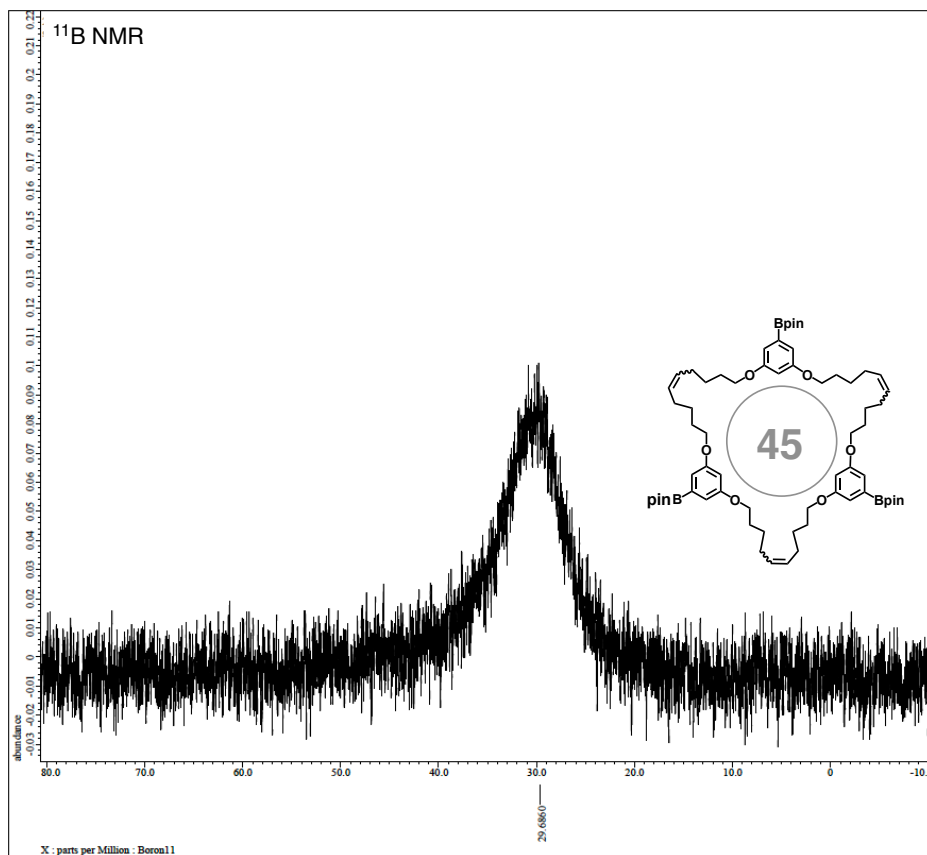


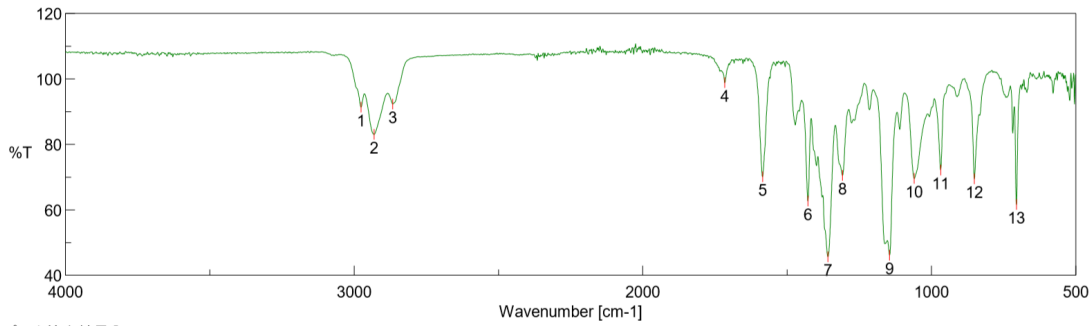
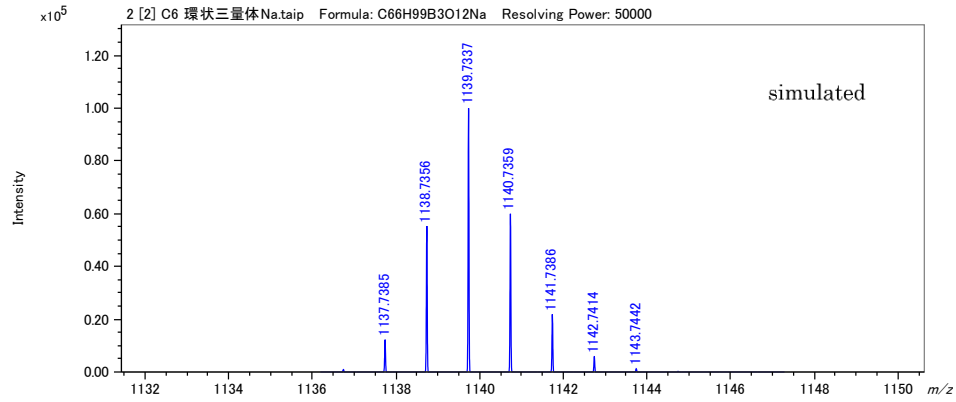
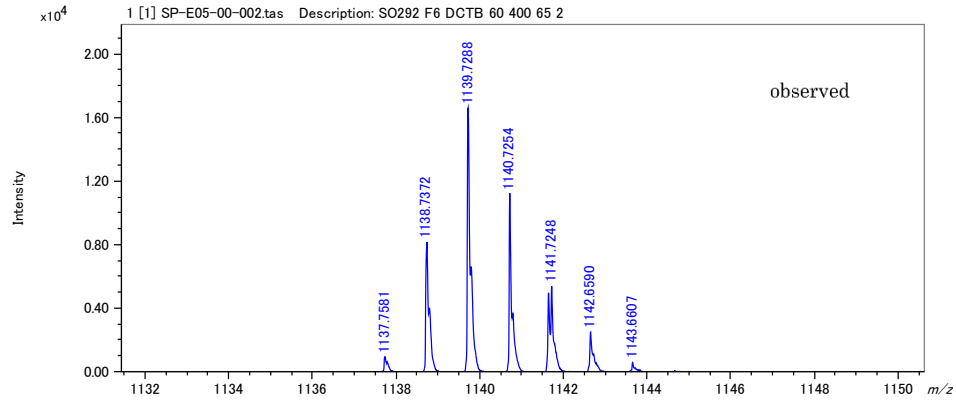
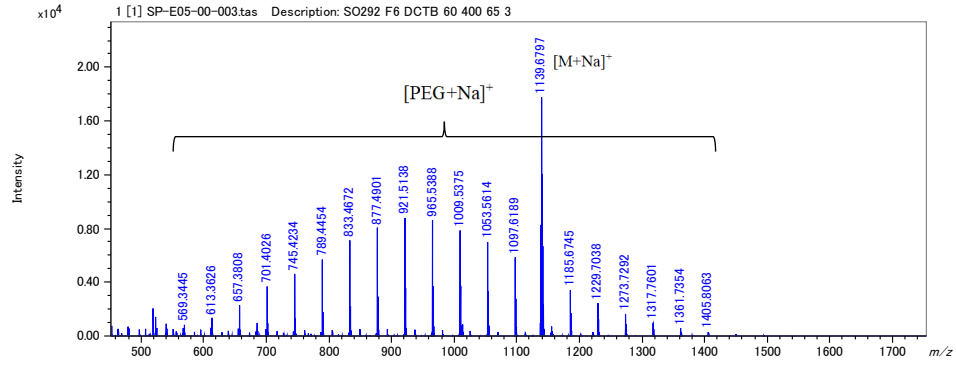
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 BF 0.25 Hz
 RGAIN 57

¹³C NMR







[ピーク検出結果]

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13	705.819	61.6279				12	851.418	69.4343

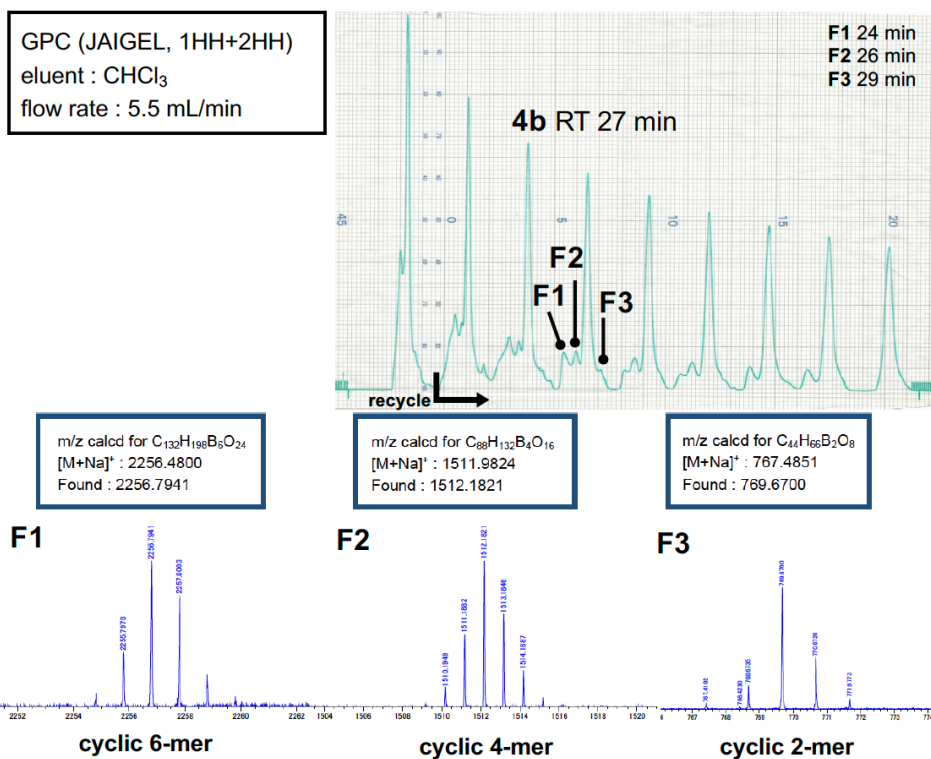


Fig. S3 GPC traces of crude **4b** and MALDI-TOF MS analysis of fractions F1, F2 and F3.

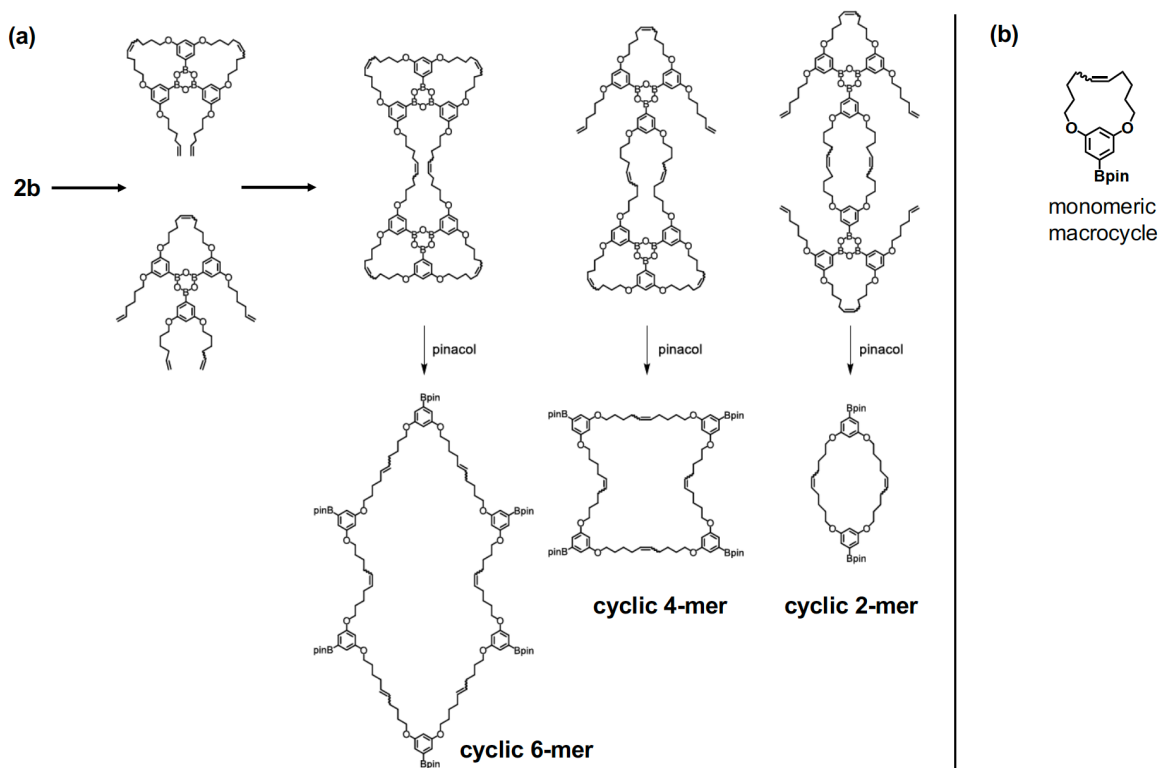


Fig. S4 (a) Chemical structures of cyclic 6-mer, cyclic 4-mer and cyclic 2-mer and their plausible formation mechanism from **2b**. (b) Chemical structure of monomeric macrocycle formed by self-cyclization of **1b**.

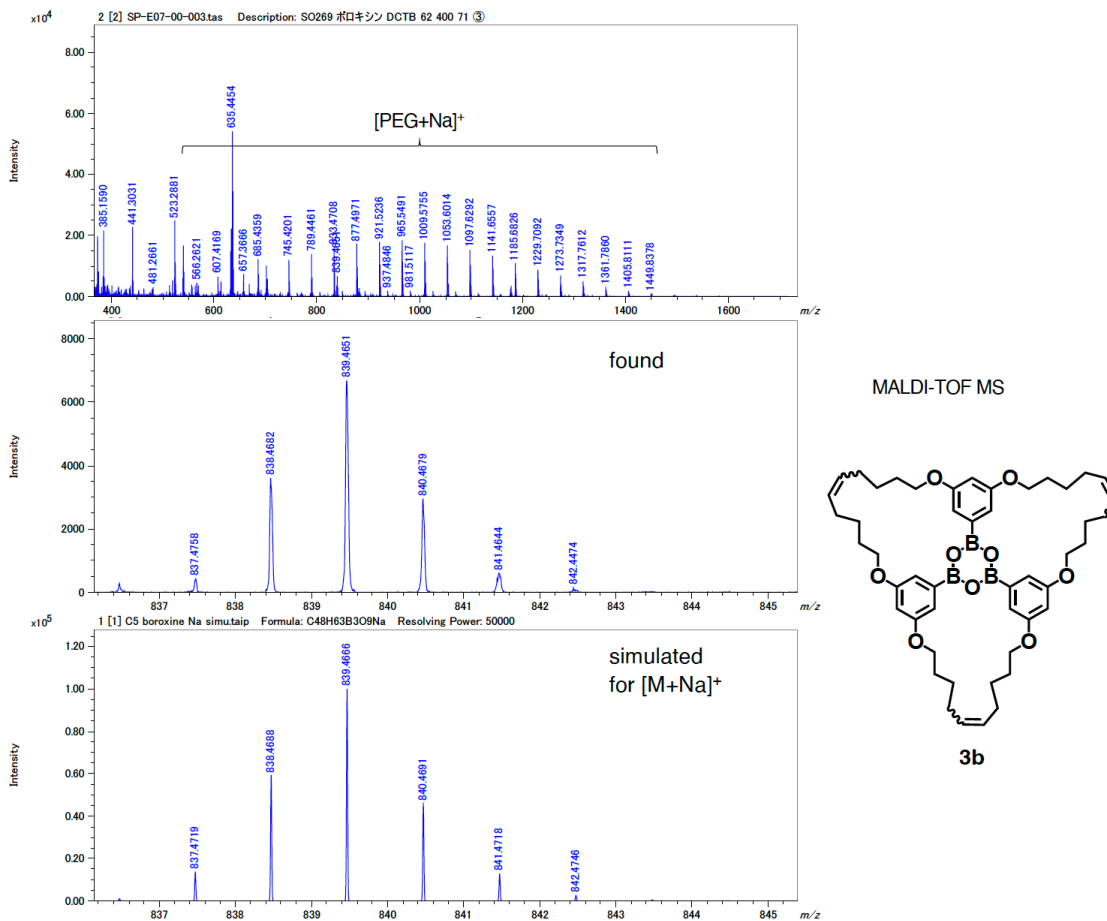
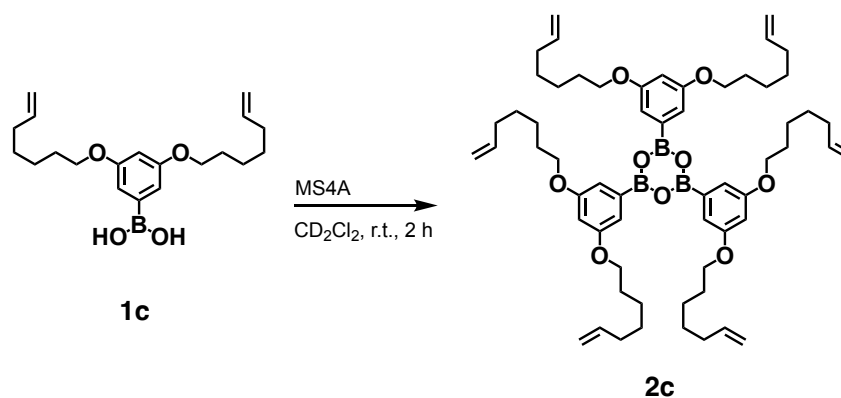


Fig. S5 MALDI-TOF MS spectra of crude **3b**

2-5. Boroxine Formation from Boronic acid **1c** to **2c**



To a $CDCl_3$ (600 μL) solution of boronic acid **1c** (2.4 mg) in an NMR tube was added MS4A. After standing at room temperature for 2 h, 1H NMR spectrum of the solution revealed the quantitative formation of boroxine **2c**.

Physical data of **2c**

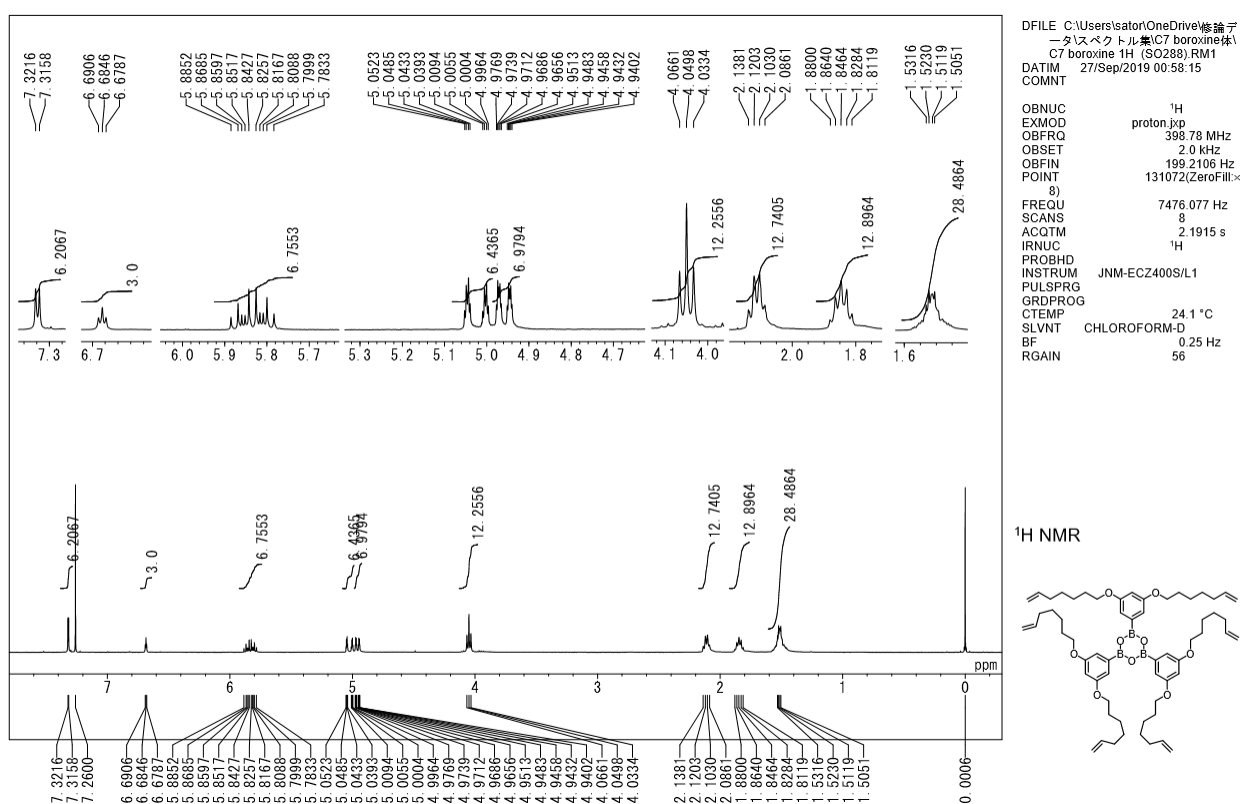
^1H NMR (400 MHz, CDCl_3): δ 7.32 (d, $J = 2.3$ Hz, 6H), 6.68 (t, $J = 2.3$ Hz, 3H), 5.83 (ddt, $J = 17.1, 10.3, 6.7$ Hz, 6H), 5.05-5.00 (m, 6H), 4.98-4.94 (m, 6H), 4.05 (t, $J = 6.5$ Hz, 12H), 2.13-2.09 (m, 12H), 1.85 (quin, $J = 6.8$ Hz, 12H), 1.53-1.51 (m, 24H).

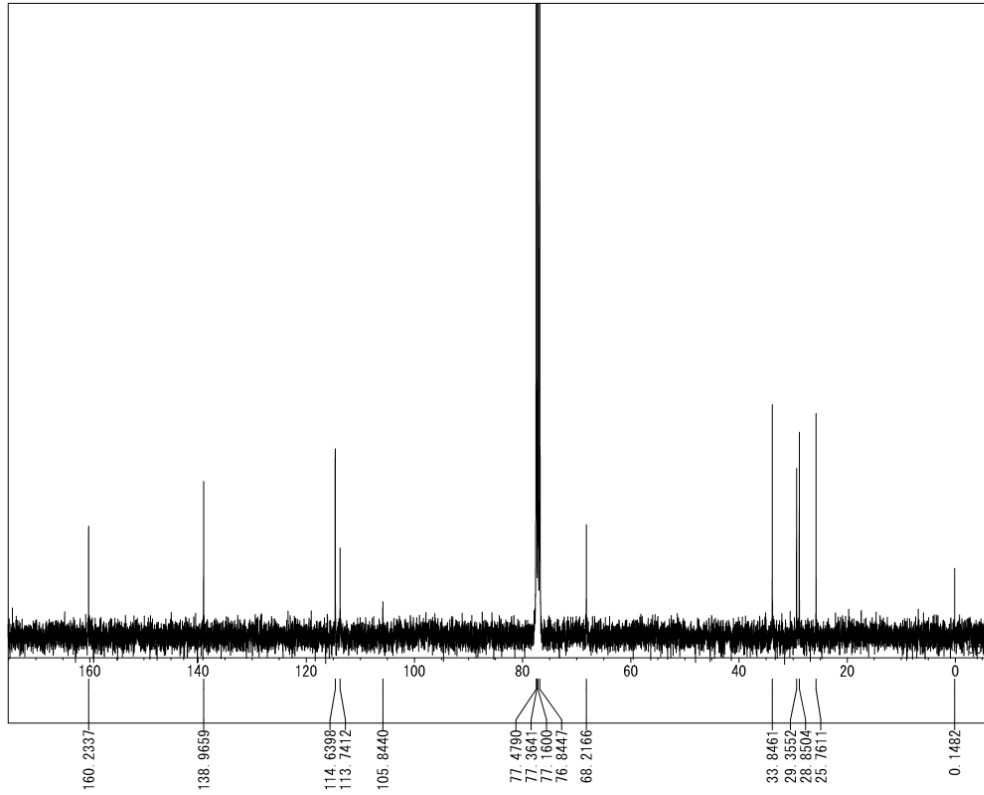
^{13}C NMR (100 MHz, CDCl_3): δ 160.2, 139.0, 114.6, 113.7, 105.8, 68.2, 33.8, 29.4, 28.9, 25.8. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, CDCl_3): δ 29.4

HRMS (MALDI⁺): m/z calcd. for $\text{C}_{60}\text{H}_{87}\text{B}_3\text{O}_9\text{Na}$: 1007.6548, found: 1007.6484 $[\text{M}+\text{Na}]^+$.

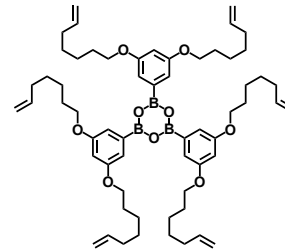
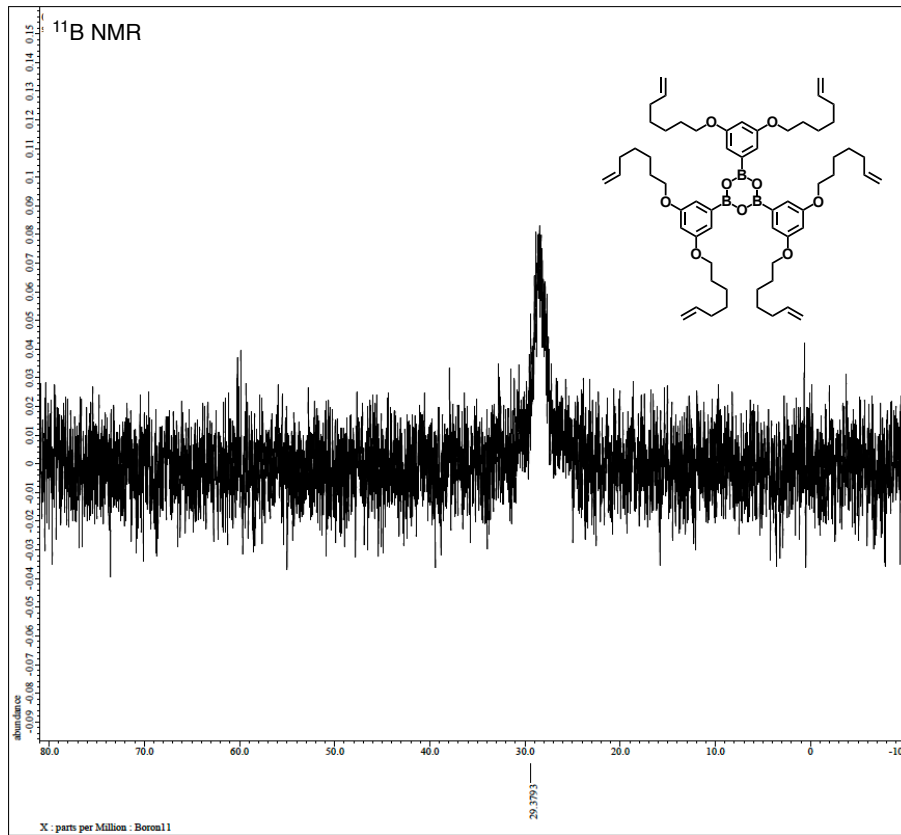
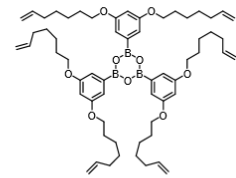
IR (ATR): 3074, 2927, 2856, 1640, 1586, 1462, 1431, 1338, 1259, 1162, 1053, 993, 908, 844, 729, cm^{-1} .

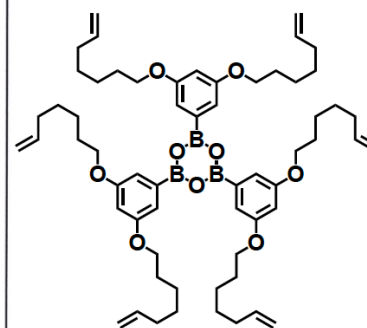
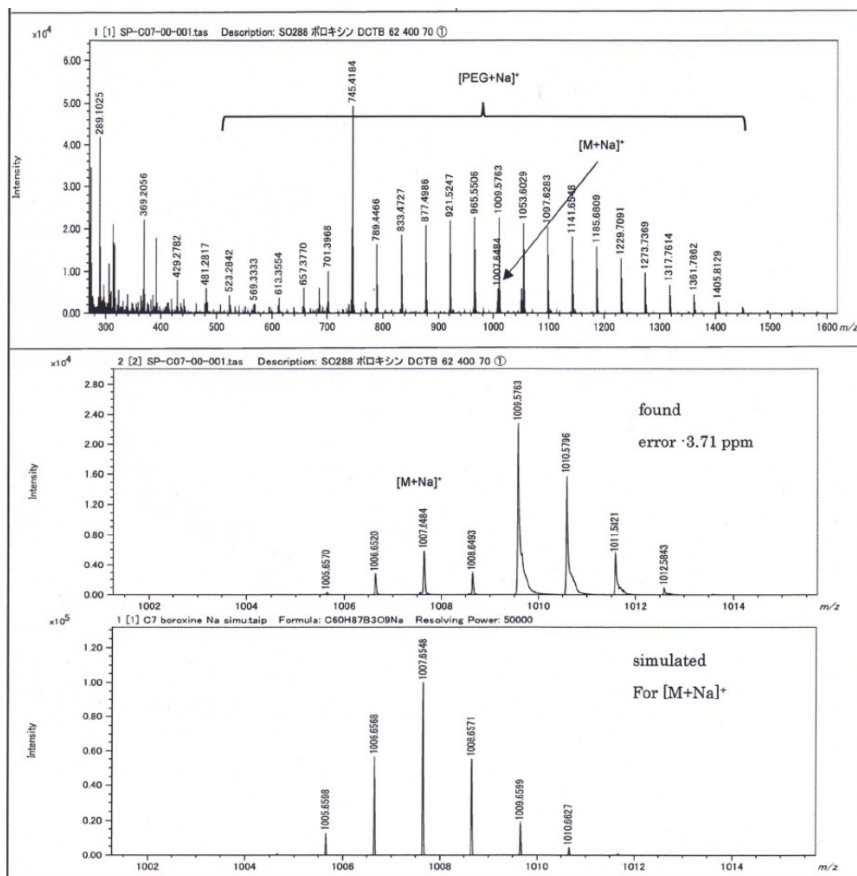




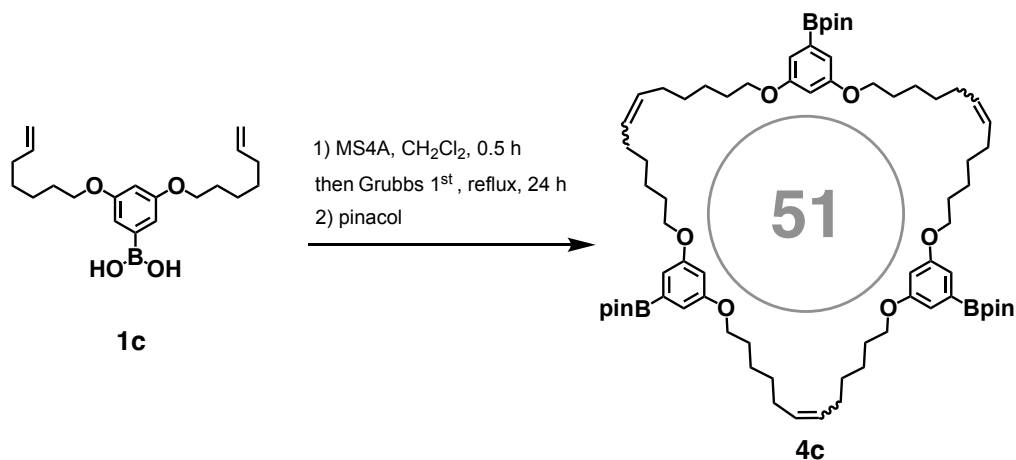
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 BF 0.25 Hz
 RGAIN 81

¹³C NMR





2-6. Synthesis of Macrocycle 4c



Boronic acid **1c** (92.9 mg, 0.268 mmol) was dissolved in dichloromethane (340 mL) with MS4A and stirred at room temperature for 2 h. Then Grubbs 1st Generation catalyst (22.1 mg, 27.0 μ mol) was added and the resulting mixture was refluxed for 40 h. Ethyl vinyl ether (400 μ L) was added and the mixture was stirred at room temperature for 2 h. After MS4A was removed and solvent was evaporated to crude **3c**, the crude

was treated with 3.0 equivalent pinacol (31.7 mg, 270 μmol) in a mixture of methanol and chloroform. After solvent was evaporated, the residue was purified by GPC to afford the desired **4c** (39.2 mg, 38%).

Physical data of **4c** (mixture of *cis/trans* isomers)

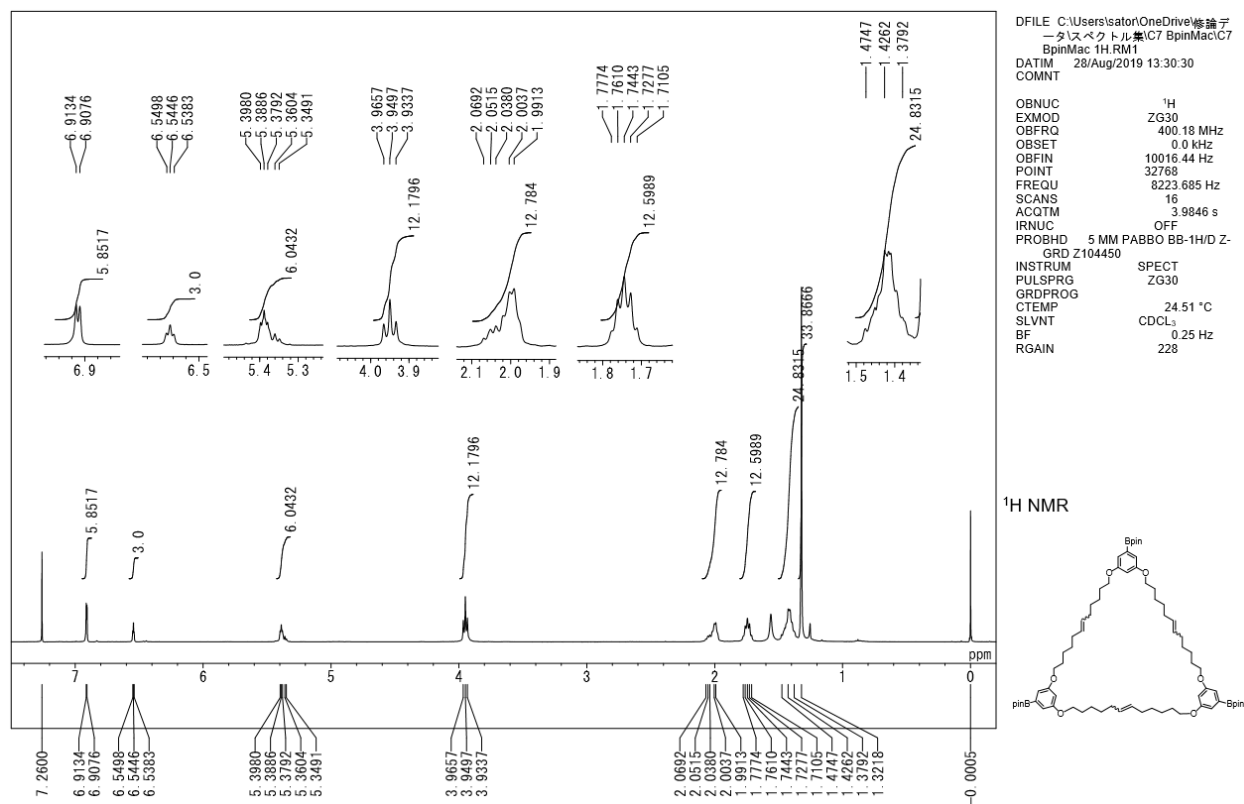
^1H NMR (400 MHz, CDCl_3): δ 6.91 (d, $J = 2.3$ Hz, 6H), 6.54 (t, $J = 2.3$ Hz, 3H), 5.40-5.35 (m, 6H), 3.95 (t, $J = 6.4$ Hz, 12H), 2.07-1.99 (m, 12H), 1.74 (quin, $J = 6.7$ Hz, 12H), 1.47-1.38 (m, 24H), 1.32 (s, 36 H).

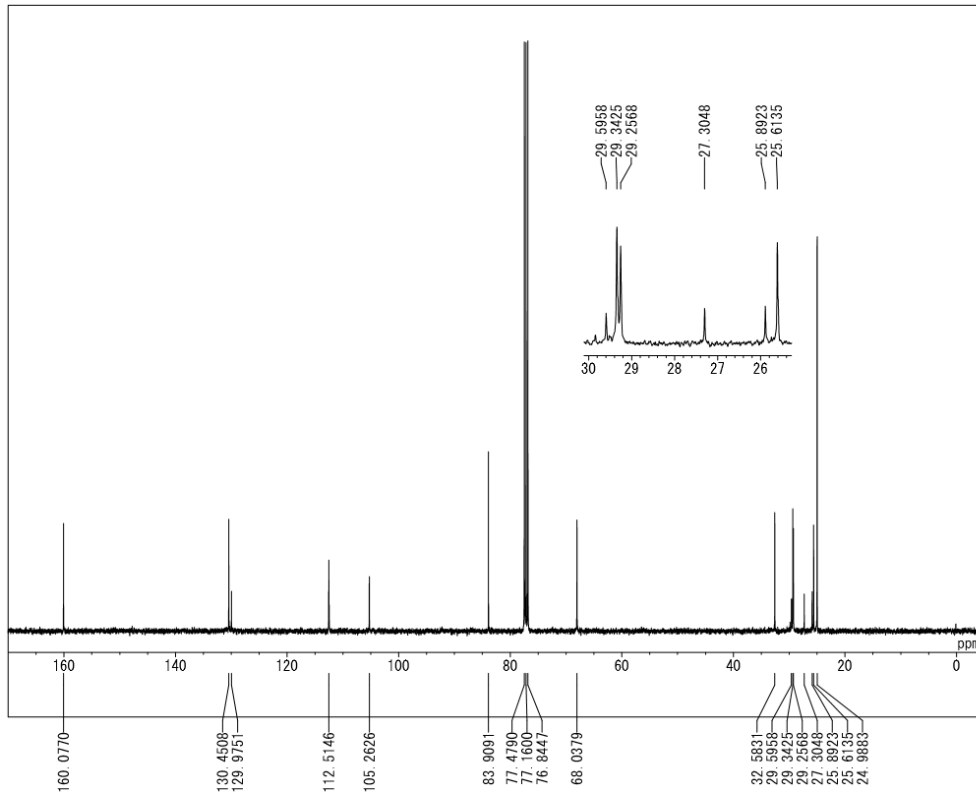
^{13}C NMR (100 MHz, CDCl_3): δ 160.1, 130.5, 130.0, 112.5, 105.3, 83.9, 68.0, 32.6, 29.6, 29.3, 29.3, 27.3, 25.9, 25.6, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, CDCl_3): δ 30.1

HRMS (MALDI $^+$): m/z calcd. for $\text{C}_{72}\text{H}_{111}\text{B}_3\text{O}_{12}\text{Na}$: 1223.8278, found: 1223.8221 $[\text{M}+\text{Na}]^+$.

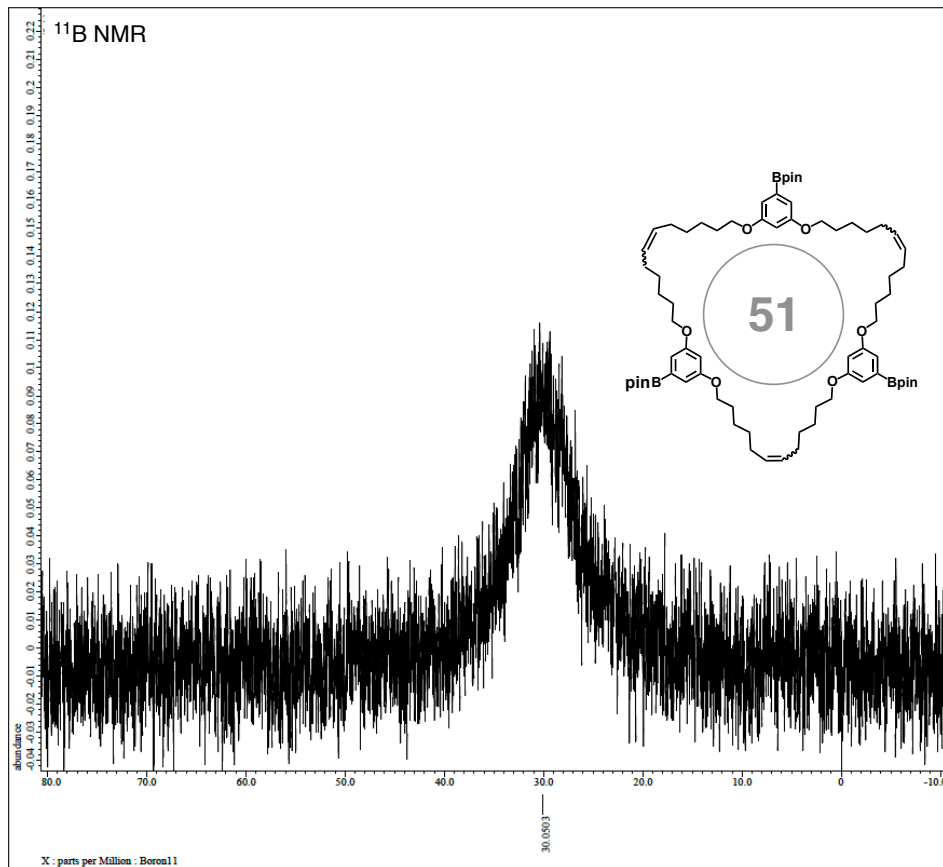
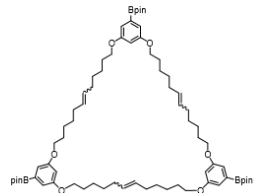
IR (ATR): 2977, 2929, 2855, 1715, 1585, 1428, 1357, 1308, 1144, 1052, 967, 851, 705 cm^{-1} .

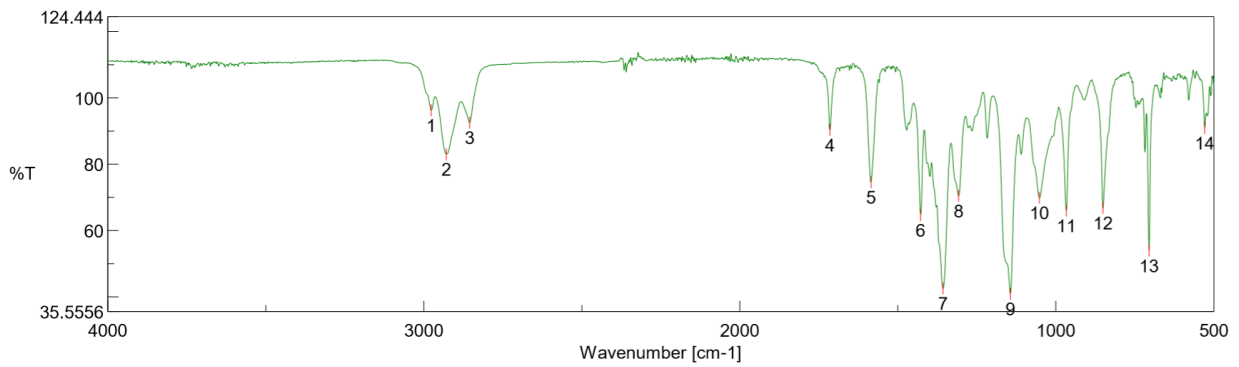
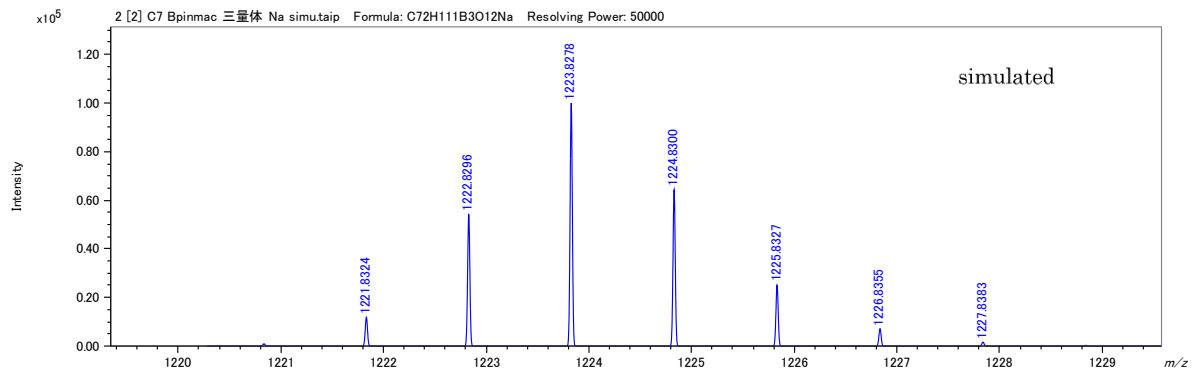
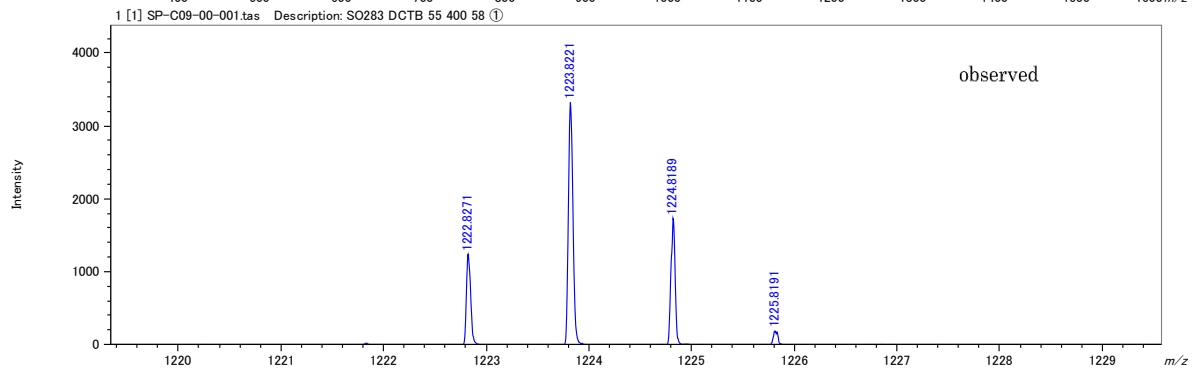
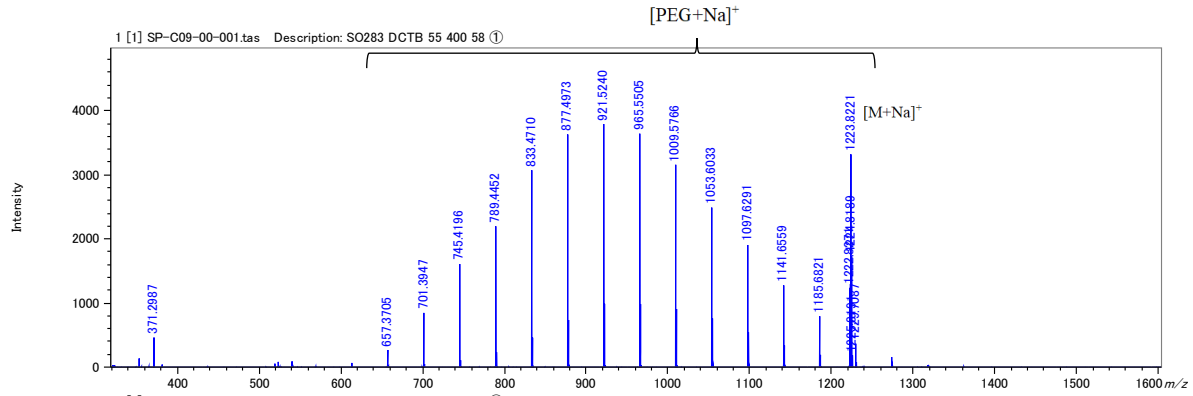




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 RGAIN 64

¹³C NMR





[ピーク検出結果]

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5	1585.2	74.5577	6	1428.03	64.6067	7	1357.16	42.5525	8	1307.5	70.3996
9	1144.06	40.9239	10	1051.98	69.8376	11	967.126	65.9684	12	850.936	66.7748
13	705.337	53.8714	14	529.364	90.8534						

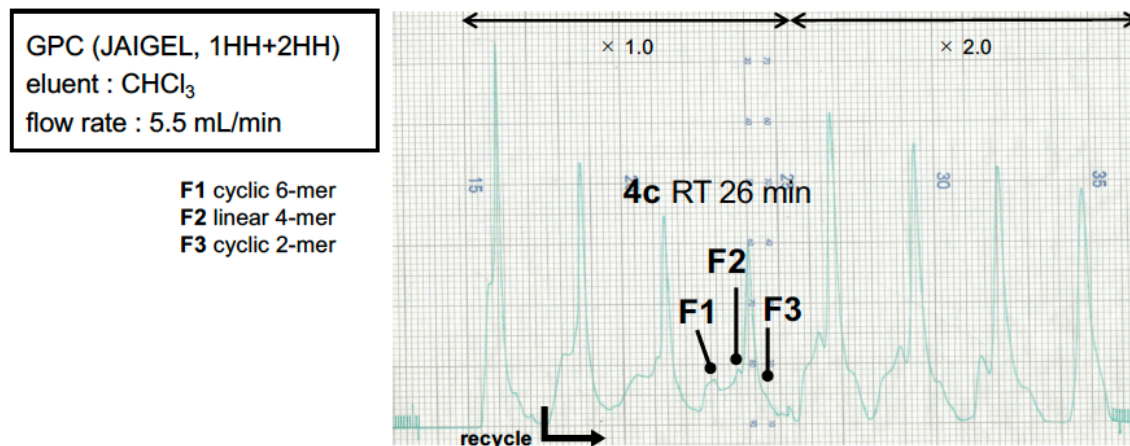
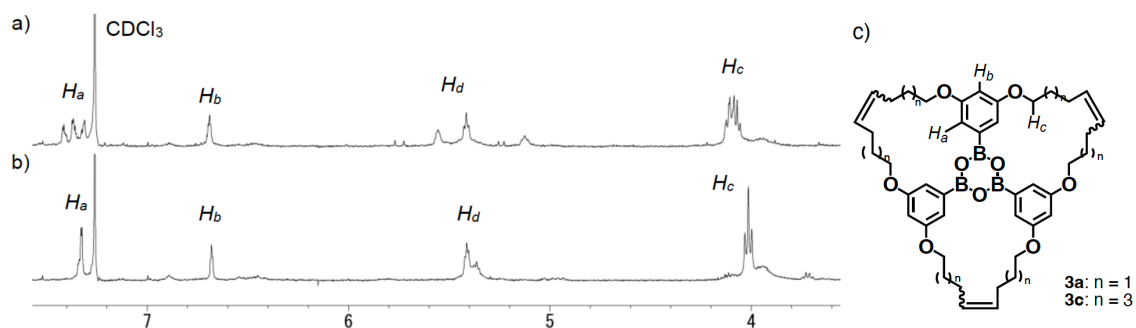


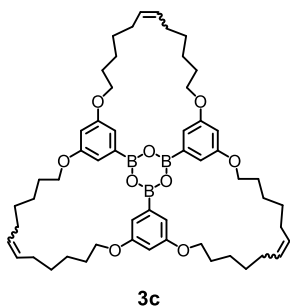
Fig. S6 GPC traces of crude **4c**.

2-7. ¹H NMR spectra of crude **3a** and **3c**



¹H NMR spectra (400 MHz, CDCl₃, r.t.) of a) crude **3a** and b) **3c**. c) Chemical structures of **3a** and **3c**.

2-8. Purification procedure and spectral data of **3c**



In a 50 mL flask, the crude of **3c** (72 mg) was dissolved in water saturated CHCl₃ and stirred at room temperature for 7 h. After that, a white precipitate was obtained (14.8 mg). Diffusion of toluene to a THF solution of the precipitate gave a small amount of solid **3c** including single crystals.

Physical data of **3c** (mixture of *cis/trans* isomers)

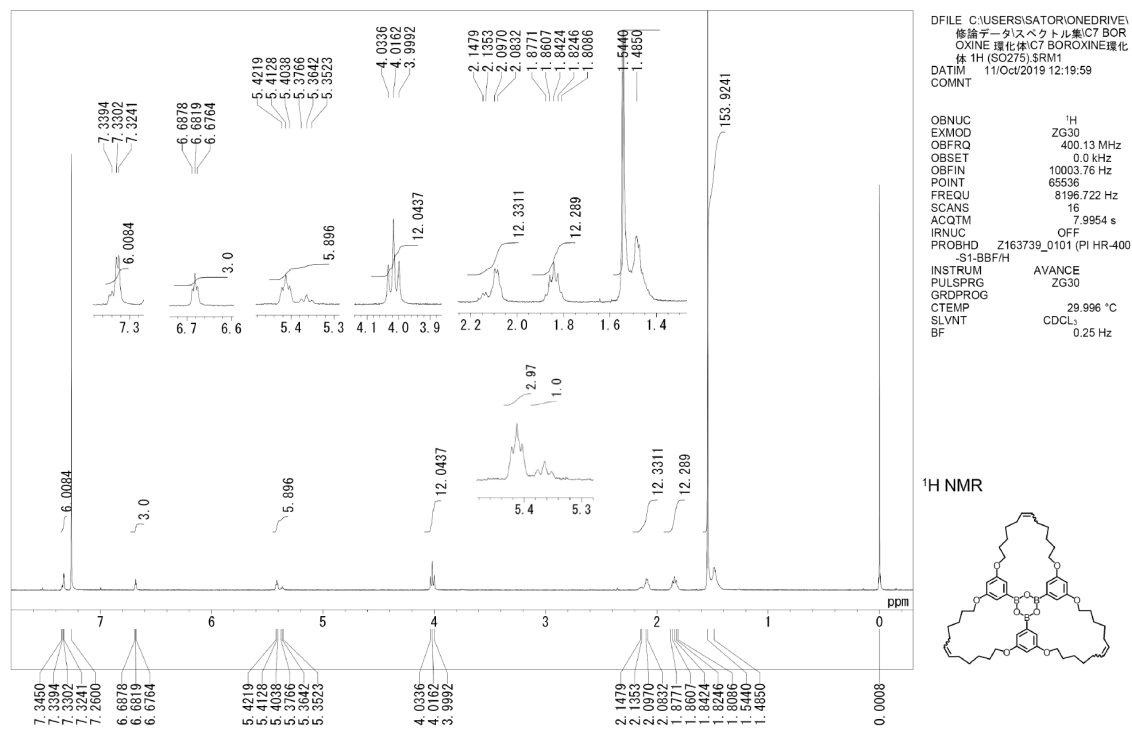
¹H NMR (400 MHz, CDCl₃): δ 7.34-7.33 (m, 6H), 6.68 (t, *J* = 2.3 Hz, 3H), 5.42-5.40, 5.38-5.35 (m, 6H), 4.02 (t, *J* = 6.9 Hz, 12H), 2.15-2.08 (m, 12H), 1.84 (quin, *J* = 6.9 Hz, 12H), 1.48 (m, 24H).

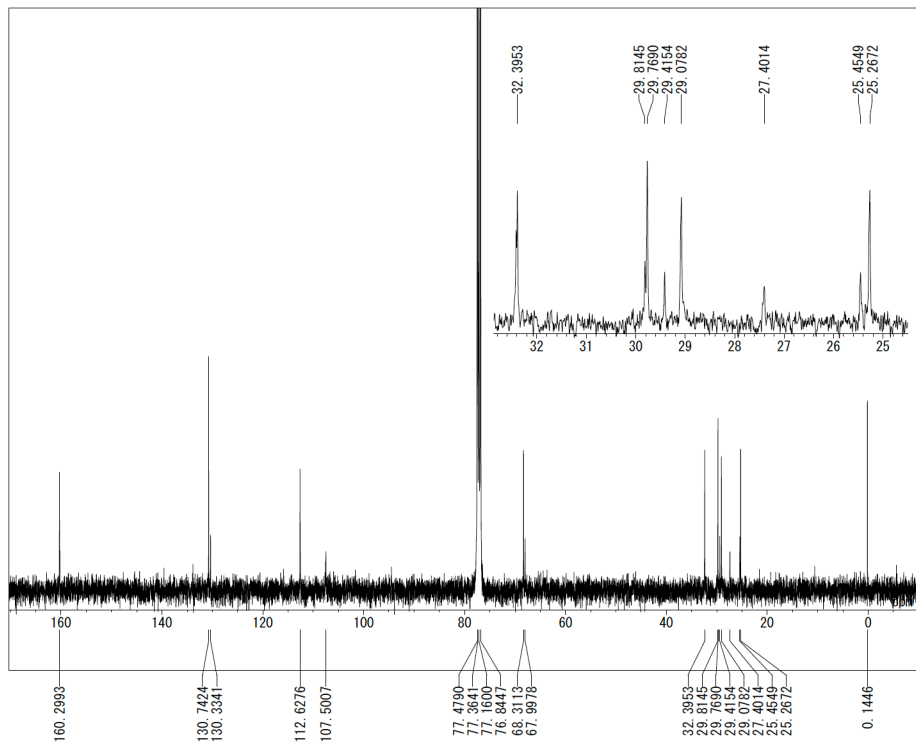
^{13}C NMR (100 MHz, CDCl_3): δ 160.3, 130.7, 130.3, 112.6, 107.5, 68.3, 68.0, 32.4, 29.8, 29.8, 29.4, 29.1, 27.4, 25.5, 25.3. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, CDCl_3): δ 30.3

MS (MALDI $^+$): m/z calcd. for $\text{C}_{54}\text{H}_{75}\text{B}_3\text{O}_9\text{Na}$: 923.5607, found: 923.5546 $[\text{M}+\text{Na}]^+$.

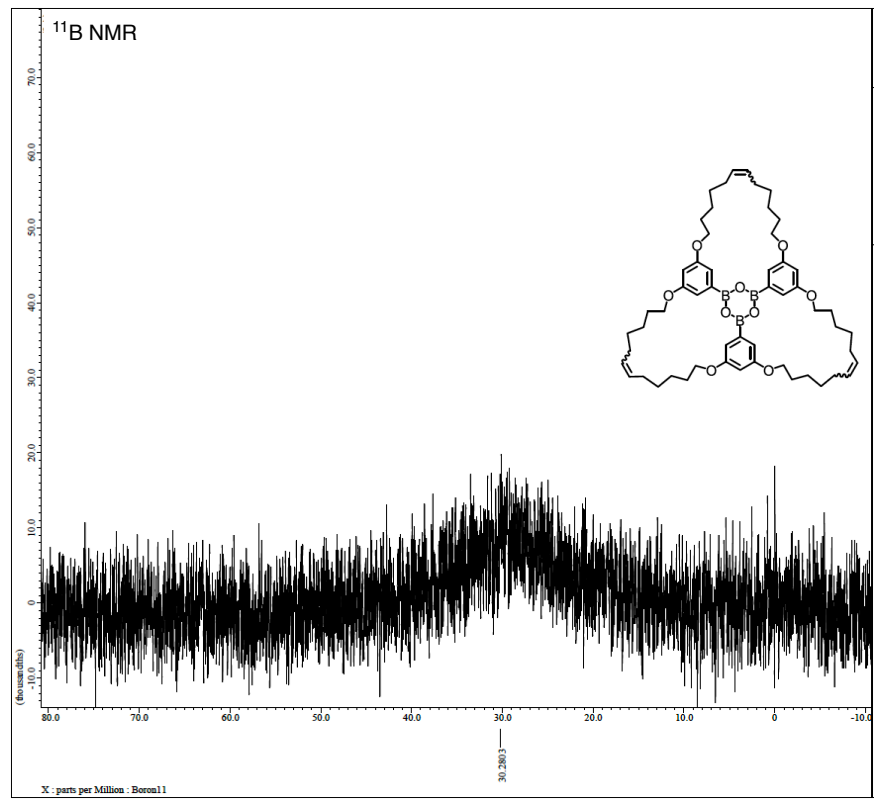
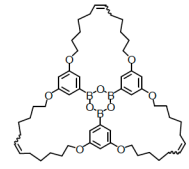
IR (ATR): 2933, 2913, 2851, 1594, 1428, 1322, 1257, 1159, 1059, 963, 846, 736, 670, 582.



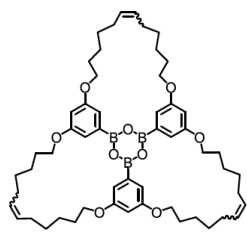


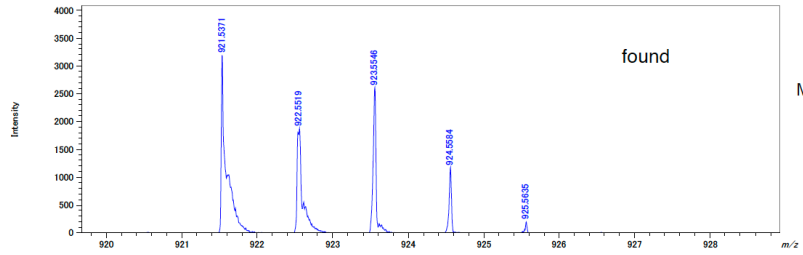
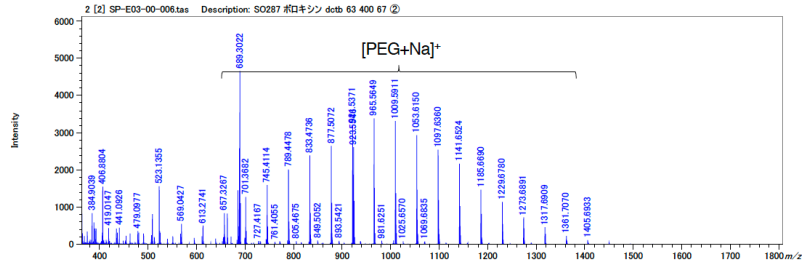
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¹³C NMR

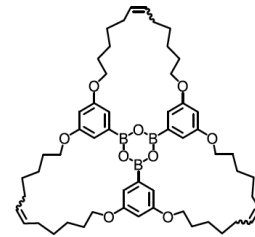


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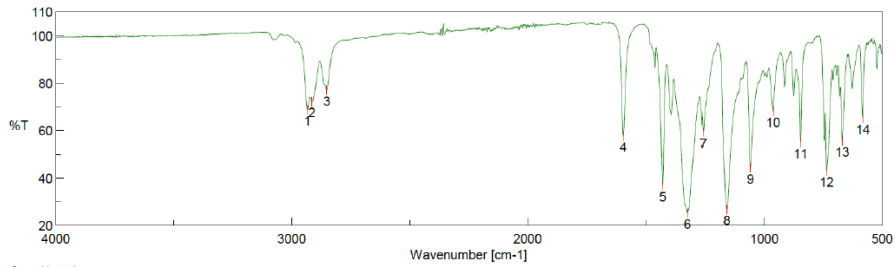
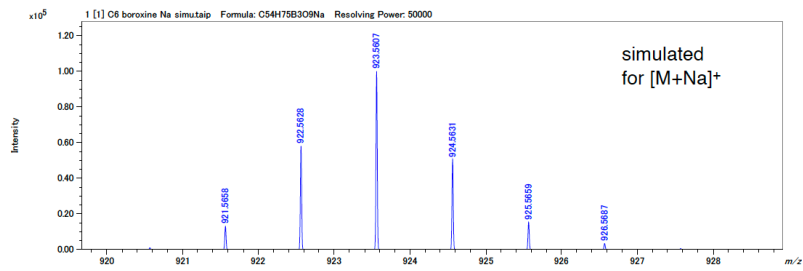




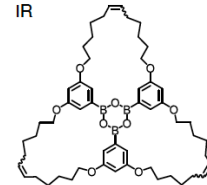
MALDI-TOF MS



3c



IR



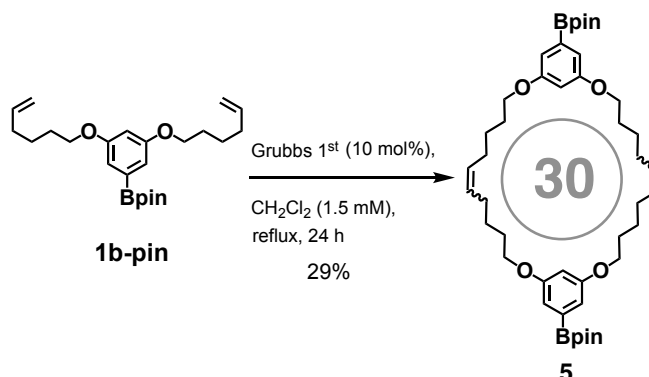
3c

[ピーク検出結果]

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5	1427.55	36.7389	6	1321.96	25.294	7	1257.36	59.3392	8	1159.01	26.726
9	1058.73	44.1204	10	962.787	67.9512	11	846.115	54.852	12	735.71	42.7695
13	669.66	55.4771	14	582.397	64.9391						

3. Control experiments

3-1. Olefin metathesis of boronic ester **1b-pin**



Boronic ester **1b-pin** (5.65 mg, 14 μ mol) was dissolved in dichloromethane (9.0 mL) and allowed to react with Grubbs 1st Generation catalyst (1.15 mg, 1.4 μ mol). The resulting mixture was refluxed for 24 h. Ethyl vinyl ether (50 μ L) was added and the mixture was stirred at room temperature for 2 h. After solvent was evaporated, the residue was purified by GPC to give **5** (3.02 mg, 58%).

Physical data of **5** (mixture of *cis/trans* isomers)

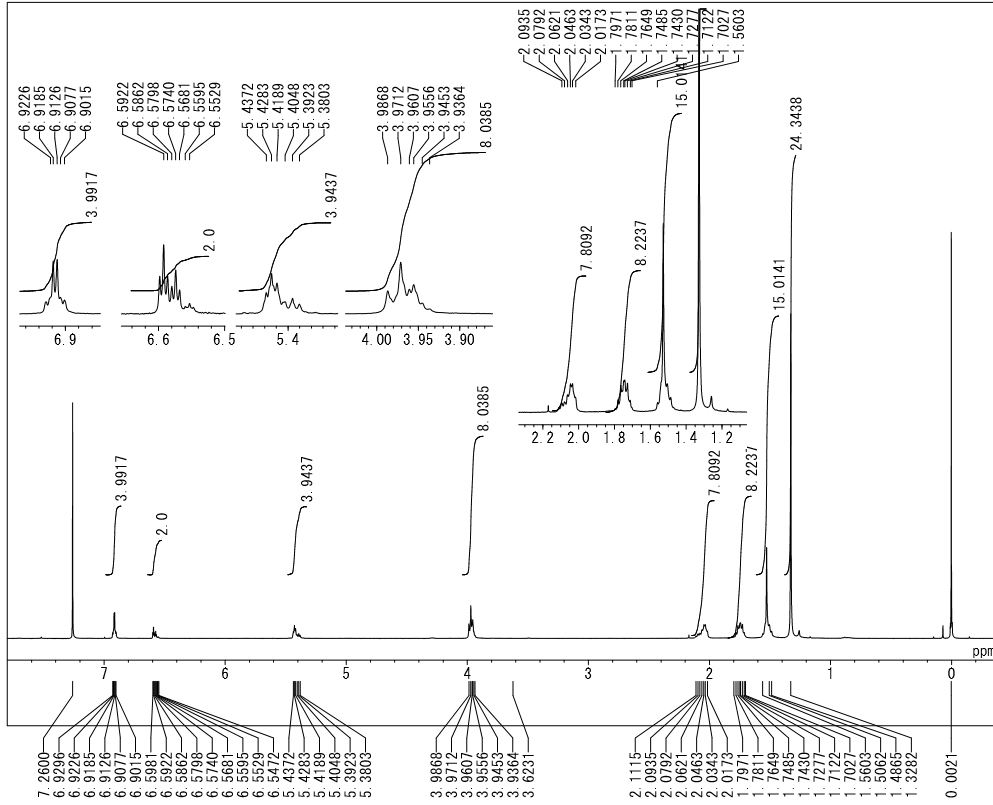
¹H NMR (400 MHz, CDCl₃): δ 6.92-6.90 (m, 4H), 6.60-6.55 (m, 2H), 5.43-5.38 (m, 4H), 3.99-3.94 (m, 8H), 2.11-2.02 (m, 8H), 1.79-1.70 (m, 8H), 1.56-1.49 (m, 8H), 1.33 (s, 24H).

¹³C NMR (100 MHz, CDCl₃): δ 160.1, 130.5, 130.5, 130.0, 130.0, 112.9, 112.2, 112.0, 105.7, 105.4, 83.9, 67.9, 32.2, 32.2, 29.0, 28.8, 28.7, 27.0, 26.3, 26.2, 26.0, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 29.5

HRMS (MALDI⁺): *m/z* calcd. for C₄₄H₆₆B₂O₈Na: 767.4851, found: 767.4839 [M+Na]⁺.

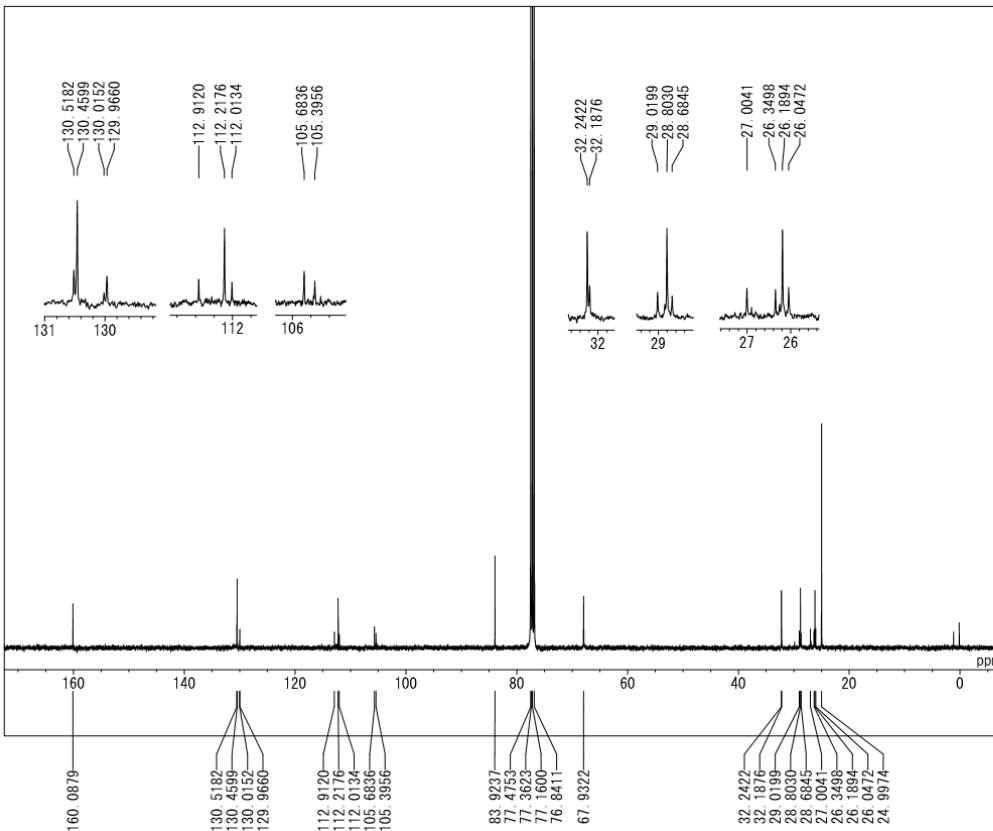
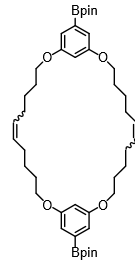
IR (ATR): 2927, 2855, 1727, 1587, 1429, 1361, 1308, 1277, 1164, 1146, 1061, 969, 851, 706, cm⁻¹.



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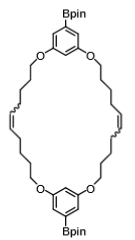
¹H NMR

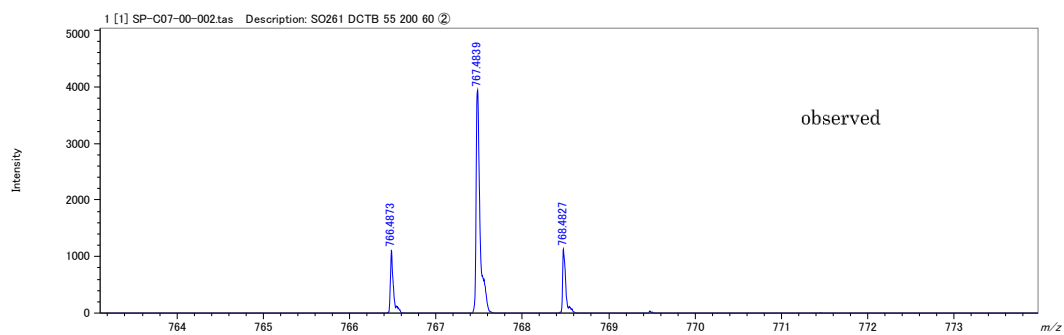
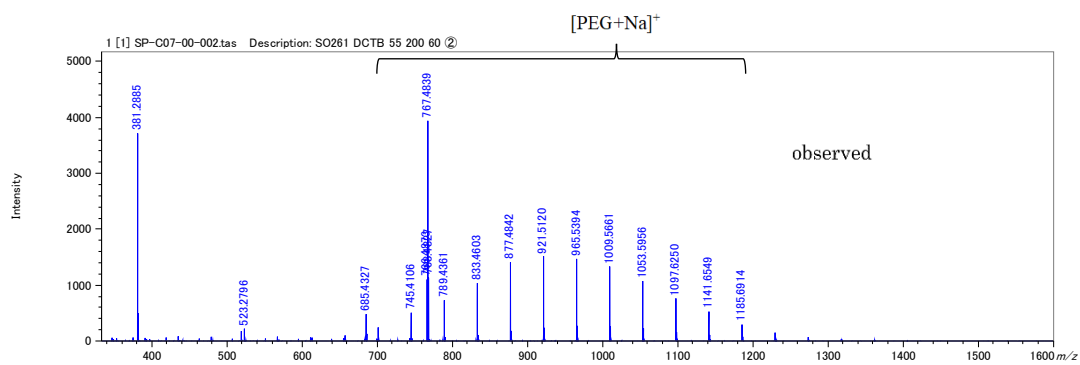
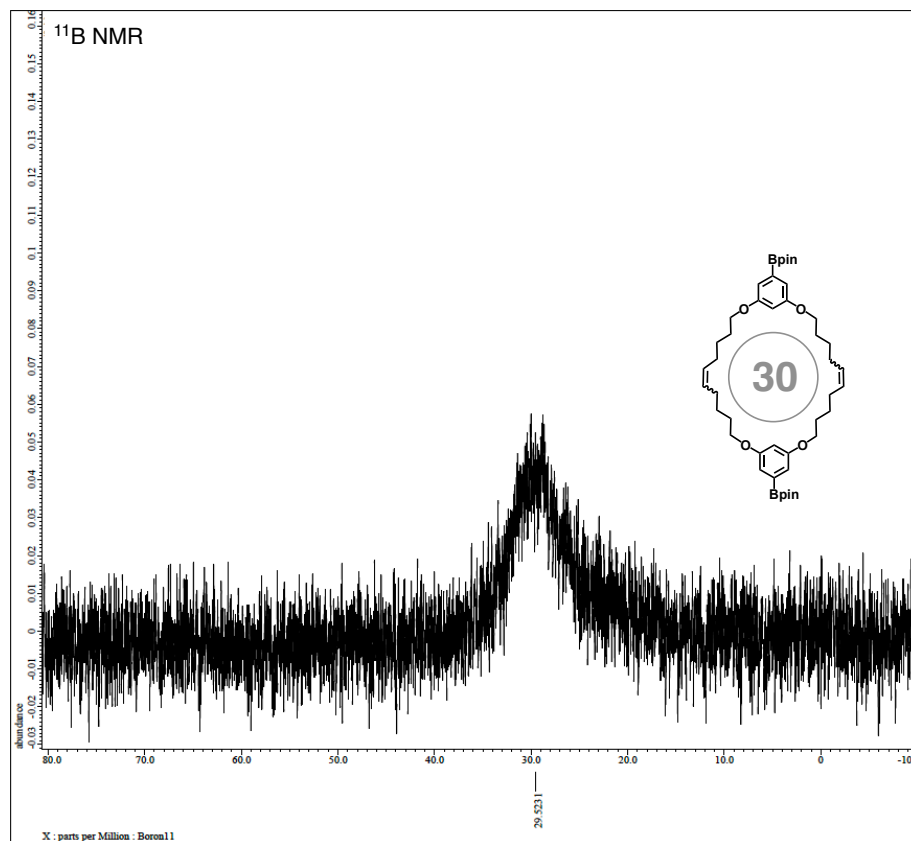


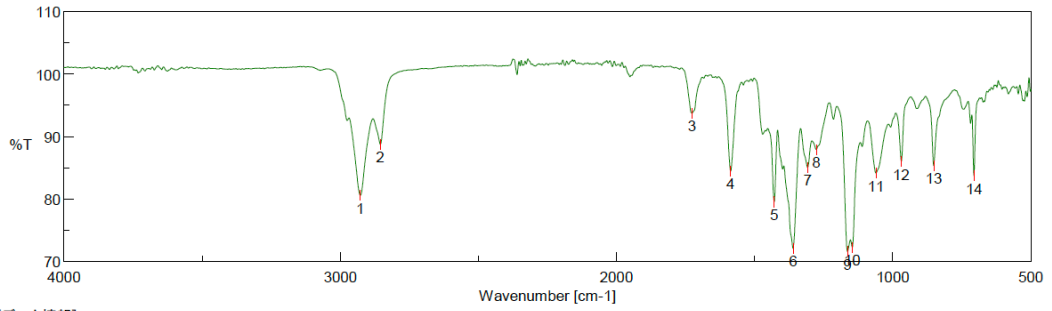
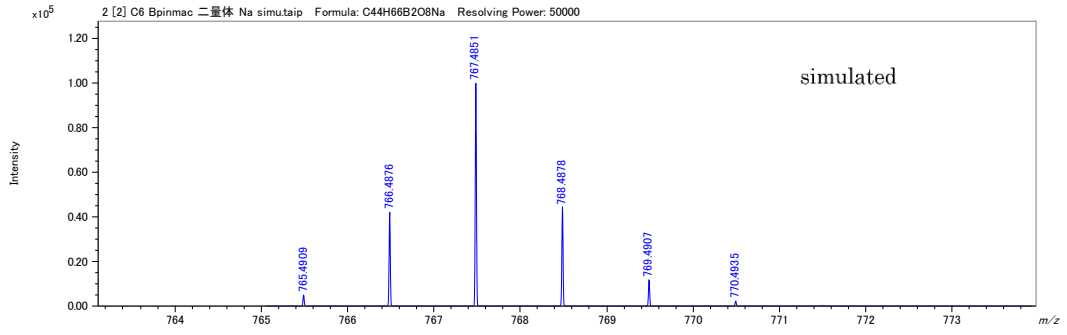
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¹³C NMR







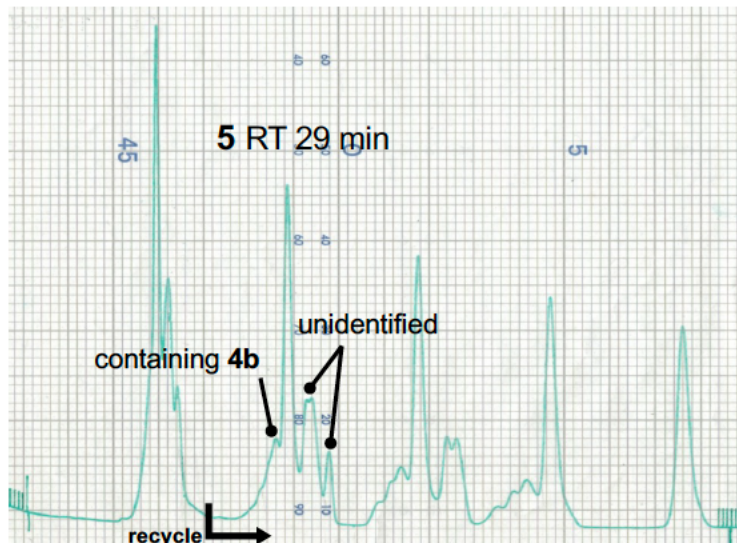
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エンド 4000.6 cm-1
データ間隔 0.964233 cm-1
データ数 3632

[ピーク検出結果]

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5	1428.99	79.5234	6	1360.53	72.0969	7	1308.46	85.0571	8	1276.65	87.9349
9	1163.83	71.5782	10	1146.47	72.2464	11	1060.66	84.1537	12	969.055	86.007
13	851.418	85.3279	14	705.819	83.7107						

GPC (JAIGEL, 1HH+2HH)
 eluent : CHCl₃
 flow rate : 5.5 mL/min



MALDI-TOF MS of crude mixture (Matrix: DCTB)

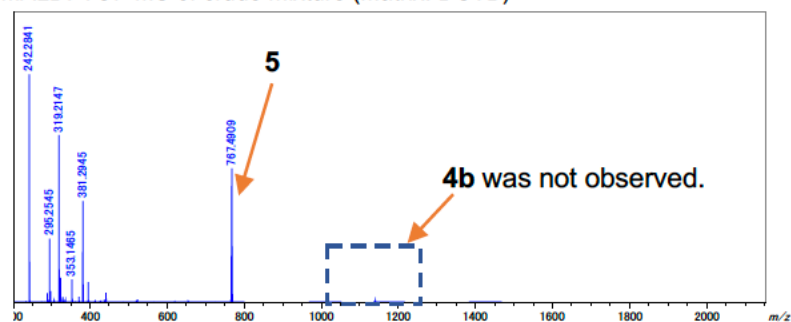
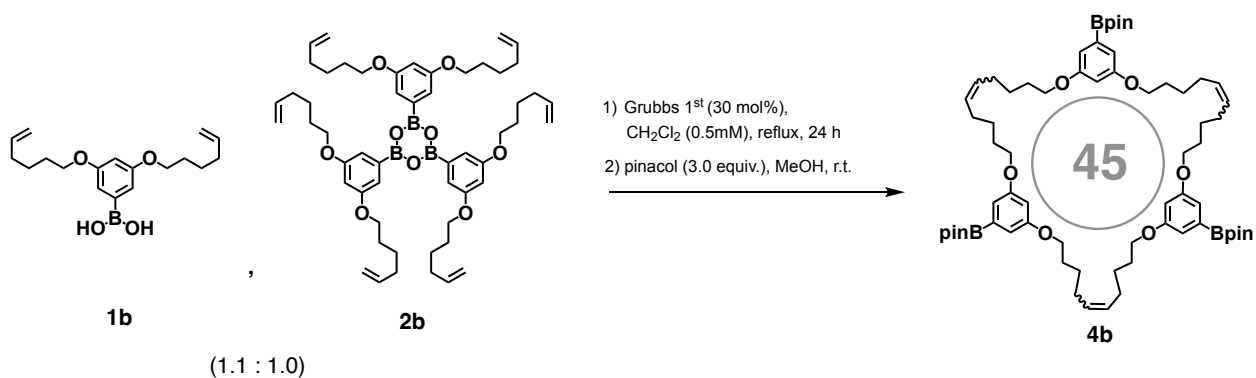


Fig. S7 GPC traces of crude **5** and MALDI-TOF MS analysis of crude mixture.

3-2. Olefin metathesis of a 1.1:1.0 mixture of **1b** and **2b**



A 1.1:1.0 mixture of **1b** and **2b** (1.75 mg) was dissolved in dichloromethane (4.0 mL) and allowed to react with Grubbs 1st Generation catalyst (0.469 mg, 0.57 μmol). The resulting mixture was refluxed for 24 h. Ethyl vinyl ether (25 μL) was added and the mixture was stirred at room temperature for 2 h. After solvent was evaporated, the residue was purified by GPC to give **4b** (1.72 mg, 76%).

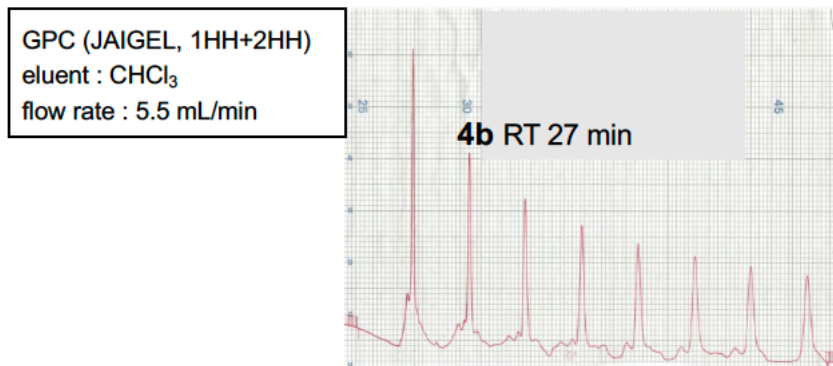
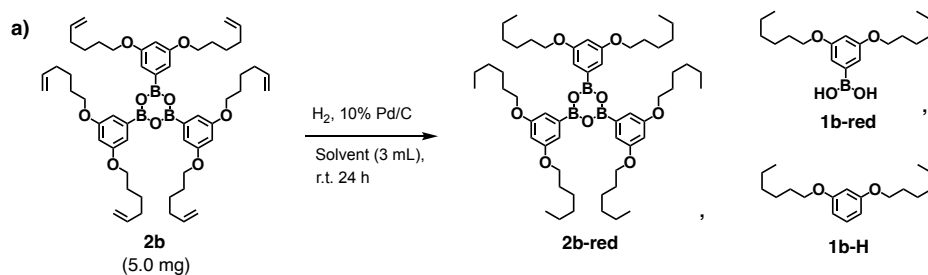


Fig. S8 GPC traces of crude **4b** prepared from a 1.1:1.0 mixture of **1b** and **2b**.

3-3. Hydrogenation of boroxine **2b**



entry	solvent	1b-H
1	EtOAc	31%
2	THF	30%
3*	EtOAc	quant.

*MS4A was added.

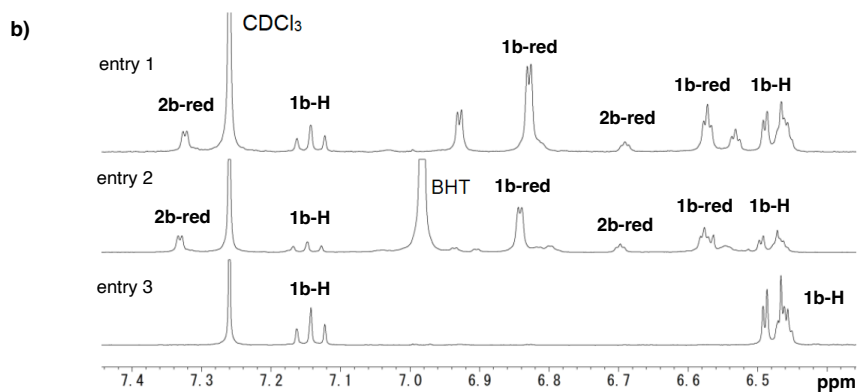
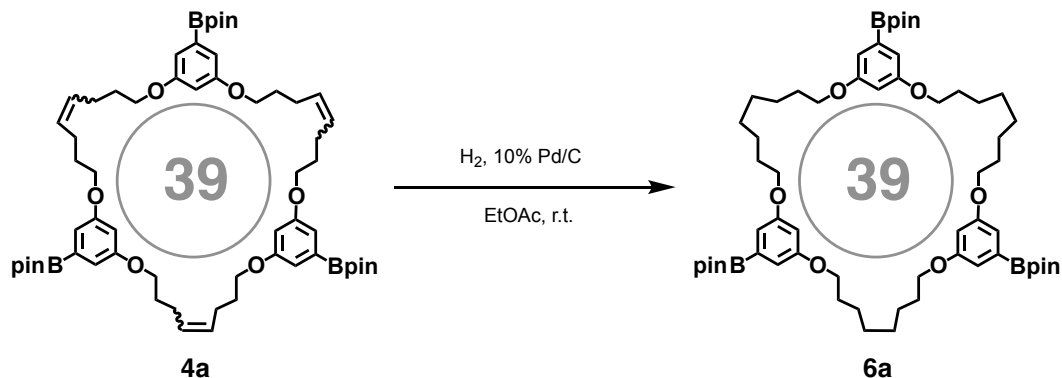


Fig. S9 a) Examination of reaction solvents of hydrogenation of **2b**. b) ¹H NMR spectra (400 MHz, 303 K, CDCl₃) of crude of entry 1, 2, and 3.

4. Post-modification of Macrocycles

4-1. Hydrogenation of Macrocycle **4a** to **6a**



Macrocycle **4a** (1.75 mg, 1.69 μmol) and palladium (10%) on charcoal were mixed in EtOAc (1.0 mL). Under hydrogen, the solution was stirred for 18 h and then filtered through Celite. The solvent was evaporated to afford the desired **6a** quantitatively.

Physical data of **6a**

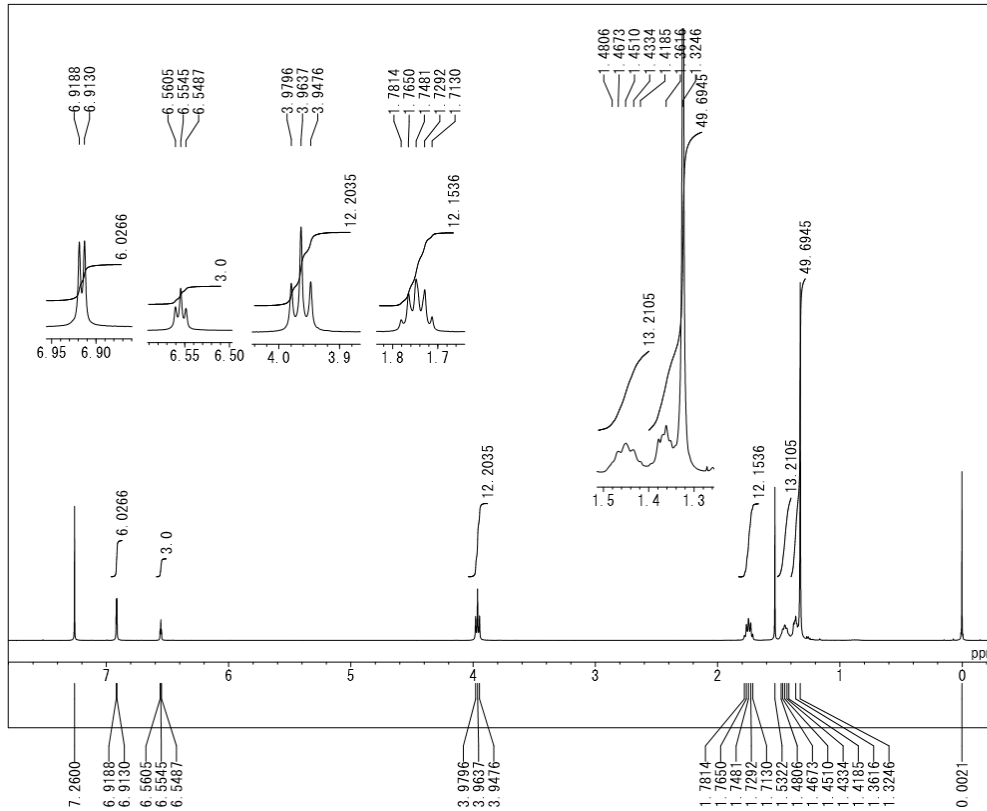
^1H NMR (400 MHz, CDCl_3): δ 6.92 (d, $J = 2.3$ Hz, 6H), 6.55 (t, $J = 2.4$ Hz, 3H), 3.96 (t, $J = 6.4$ Hz, 12H), 1.75 (quin, $J = 6.8$ Hz, 12H), 1.48-1.42 (m, 12H) 1.40-1.36 (m, 12H), 1.32 (s, 36H).

^{13}C NMR (100 MHz, CDCl_3): δ 160.1, 112.6, 105.3, 83.9, 68.0, 29.4, 29.3, 26.0, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, CDCl_3): δ 30.2

HRMS (MALDI $^+$): m/z calcd. for $\text{C}_{60}\text{H}_{93}\text{B}_3\text{O}_{12}\text{Na}$: 1061.6865, found: 1061.6793 $[\text{M}+\text{Na}]^+$.

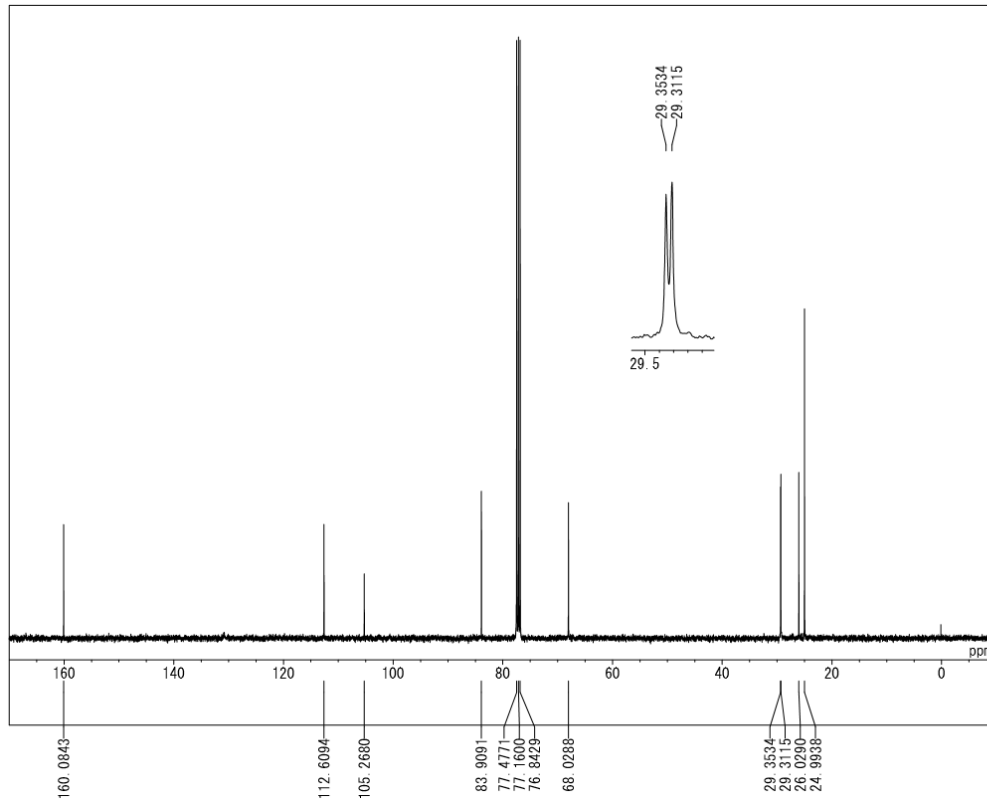
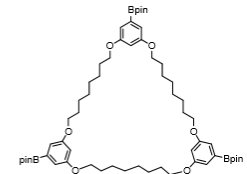
IR (ATR): 2926, 2855, 1715, 1586, 1466, 1429, 1360, 1308, 1263, 1146, 1050, 968, 851, 819, 706 cm^{-1} .



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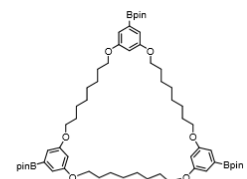
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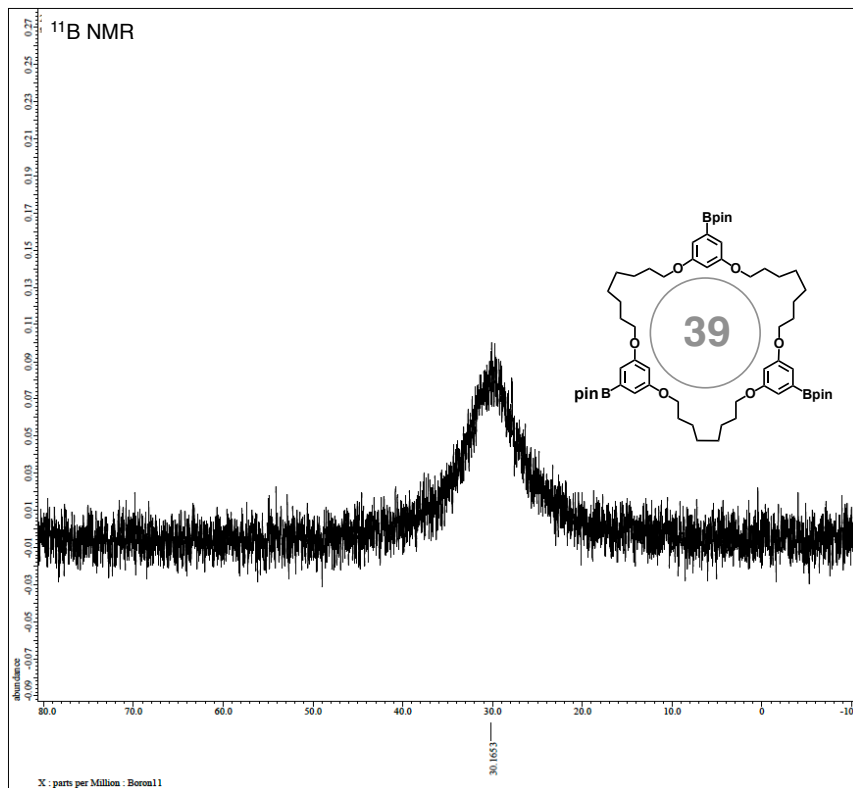


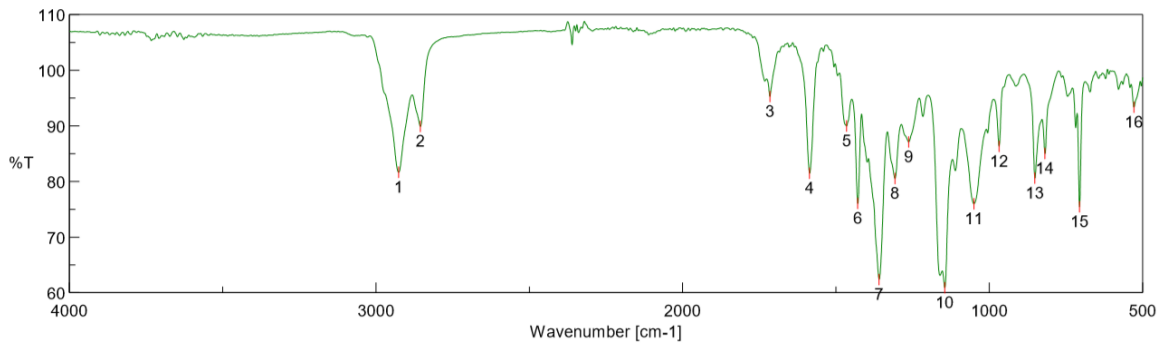
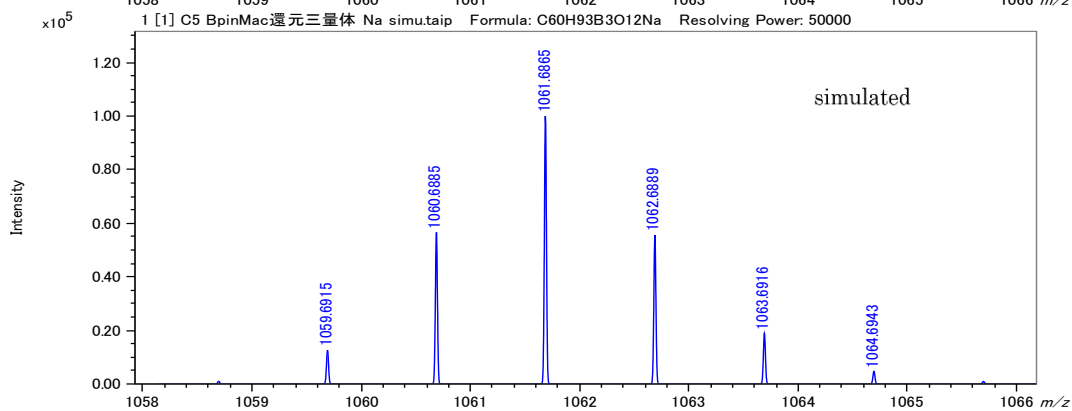
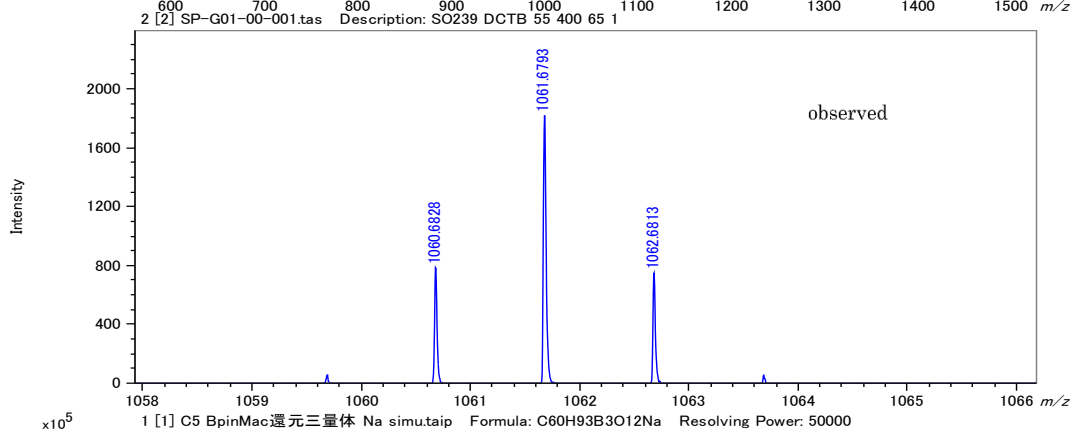
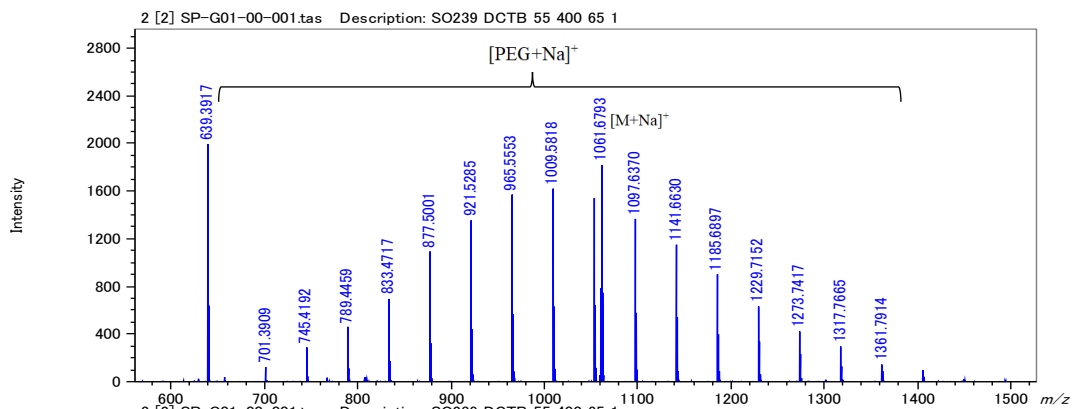
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¹³C NMR



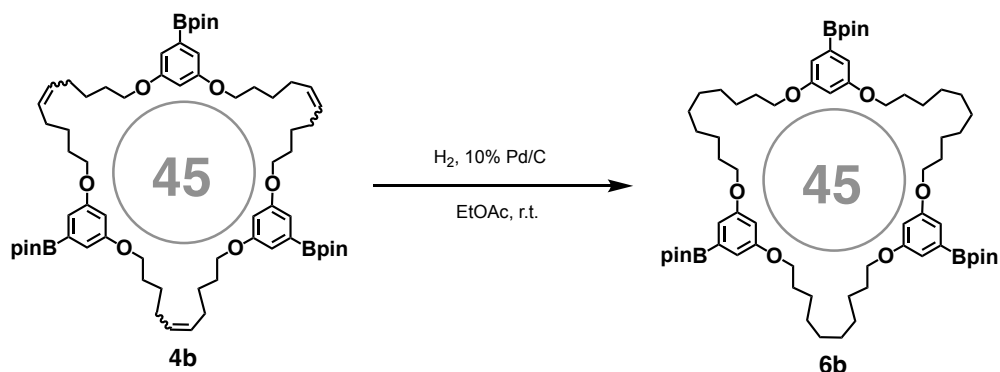




[ピーク検出結果]

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9	1263.15	87.1004	10	1145.51	60.8808	11	1050.05	75.9675
13	851.418	80.5877	14	818.634	85.0107	12	968.09	86.3026
			15	705.819	75.4471	16	528.4	93.3206

4-2. Hydrogenation of Macrocycle **4b** to **6b**



Macrocycle **4b** (72 mg, 64.5 μmol) and palladium (10%) on charcoal were mixed in EtOAc (8.0 mL). Under hydrogen, the solution was stirred for 24 h and then filtered through Celite. The solvent was evaporated to afford the desired **6b** quantitatively.

Physical data of **6b**

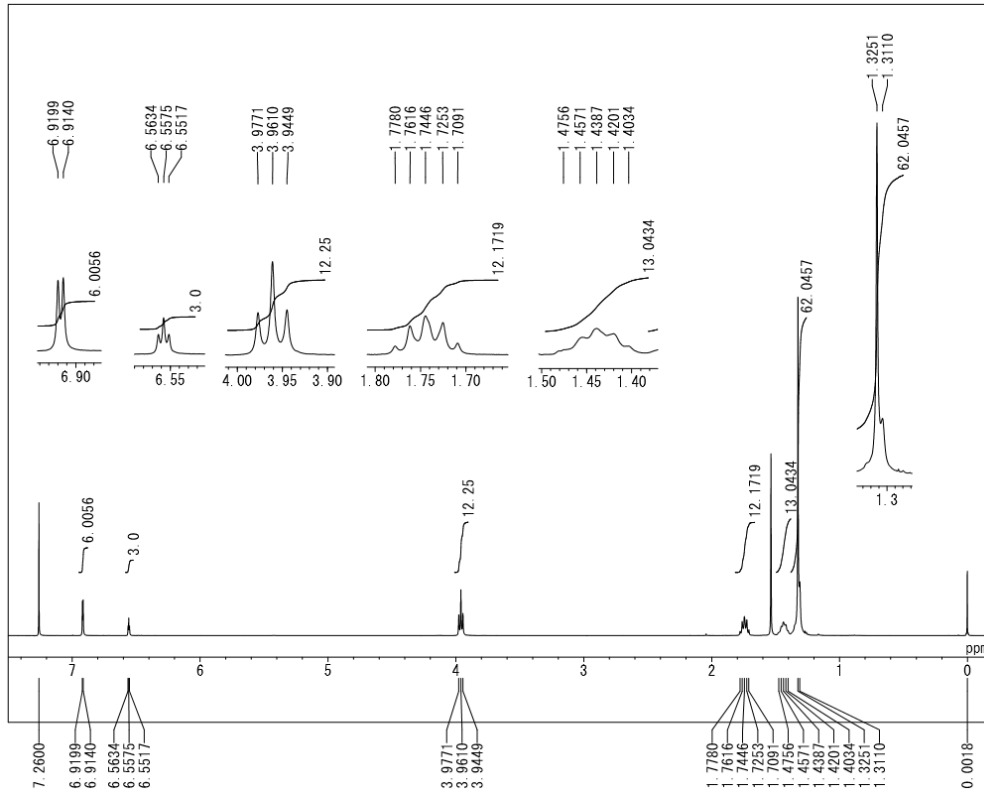
¹H NMR (400 MHz, CDCl₃): δ 6.92 (d, $J = 2.4$ Hz, 6H), 6.56 (t, $J = 2.3$ Hz, 3H), 3.96 (t, $J = 6.4$ Hz, 12H), 1.74 (quin, $J = 6.9$ Hz, 12H), 1.44 (quin, $J = 7.2$ Hz, 12H), 1.33 (s, 36H), 1.33-1.31 (m, 24H).

¹³C NMR (100 MHz, CDCl₃): δ 160.1, 112.6, 105.3, 83.9, 68.1, 29.5, 29.4, 29.4, 26.1, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 30.1

HRMS (MALDI⁺): m/z calcd. for C₆₆H₁₀₅B₃O₁₂Na: 1145.7806, found: 1145.7764 [M+Na]⁺.

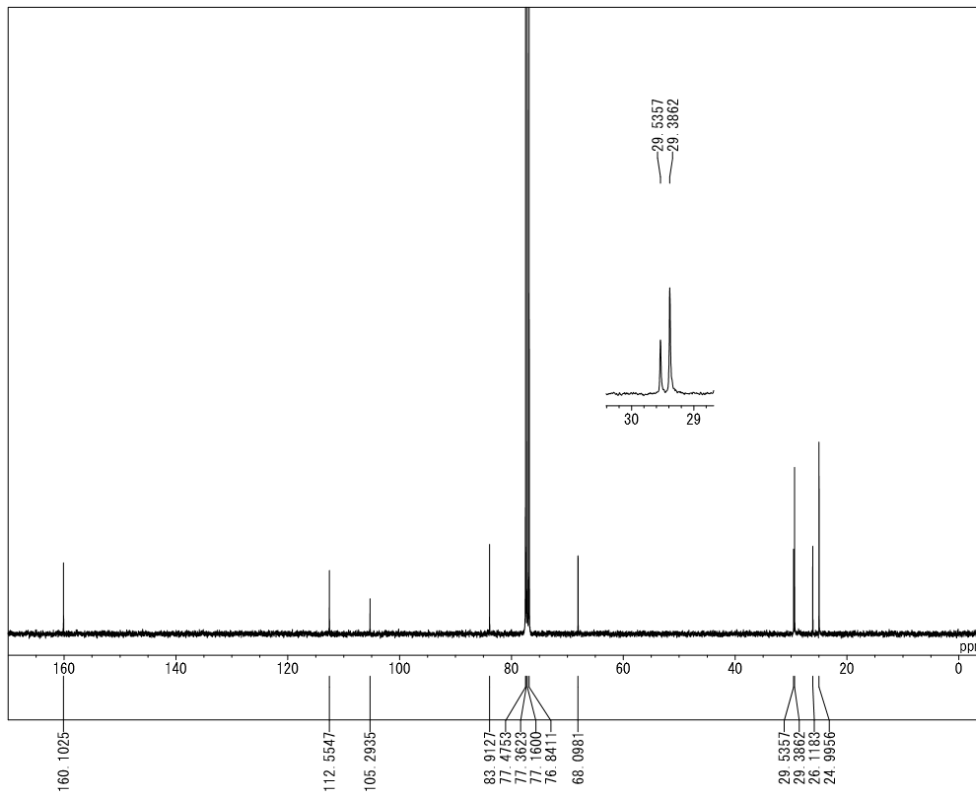
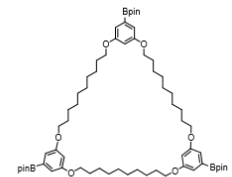
IR (ATR): 2977, 2929, 2855, 1586, 1472, 1429, 1361, 1308, 1165, 1146, 1052, 968, 852, 707 cm⁻¹.



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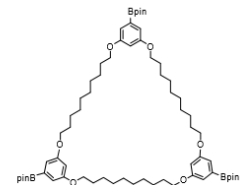
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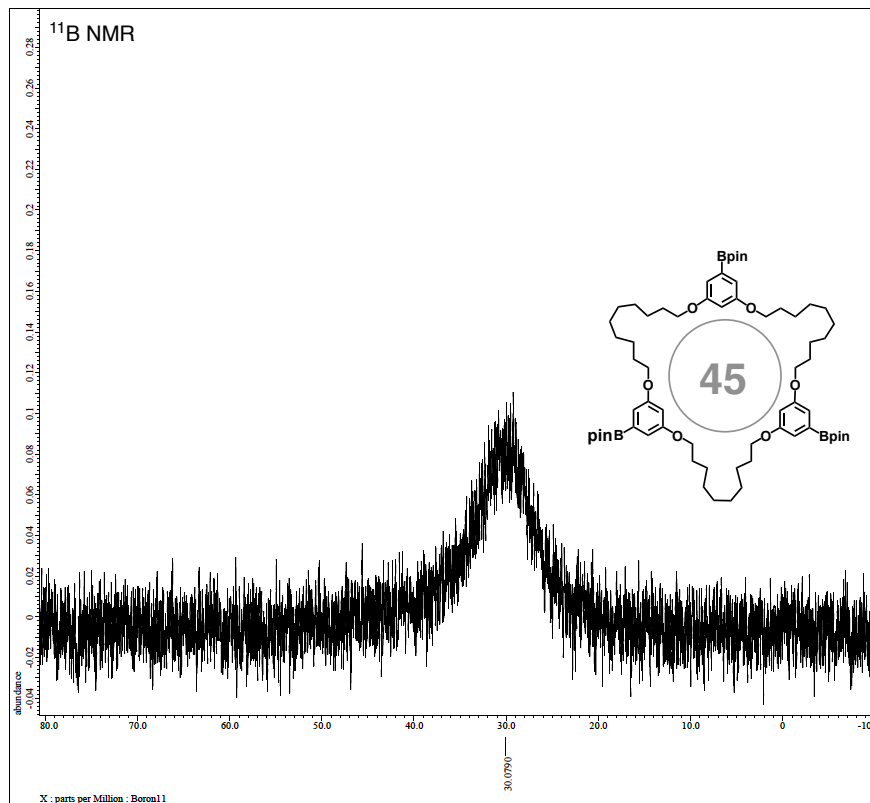


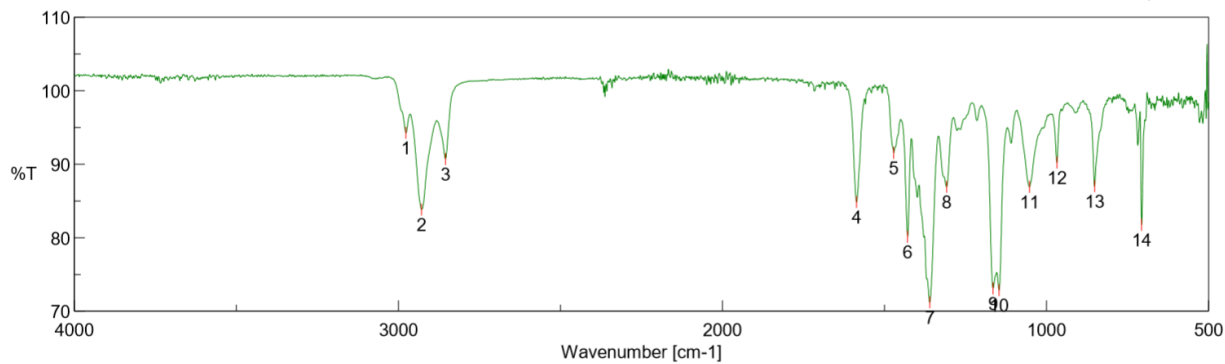
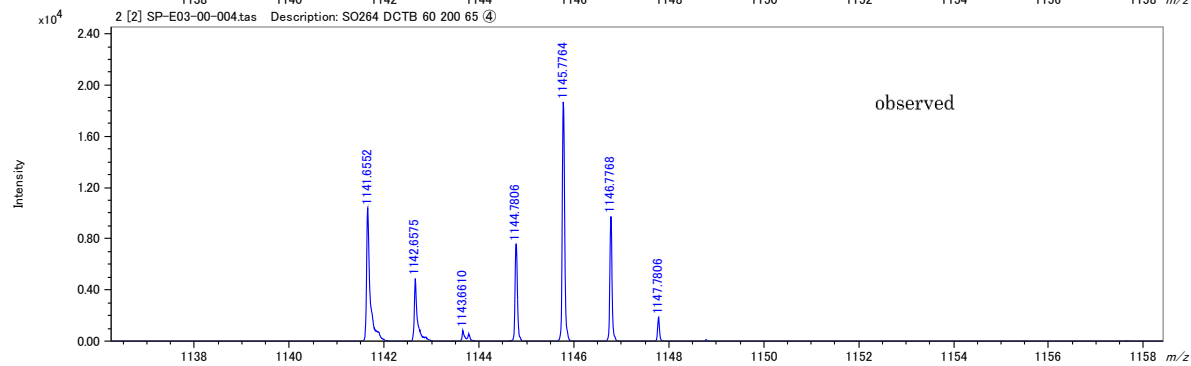
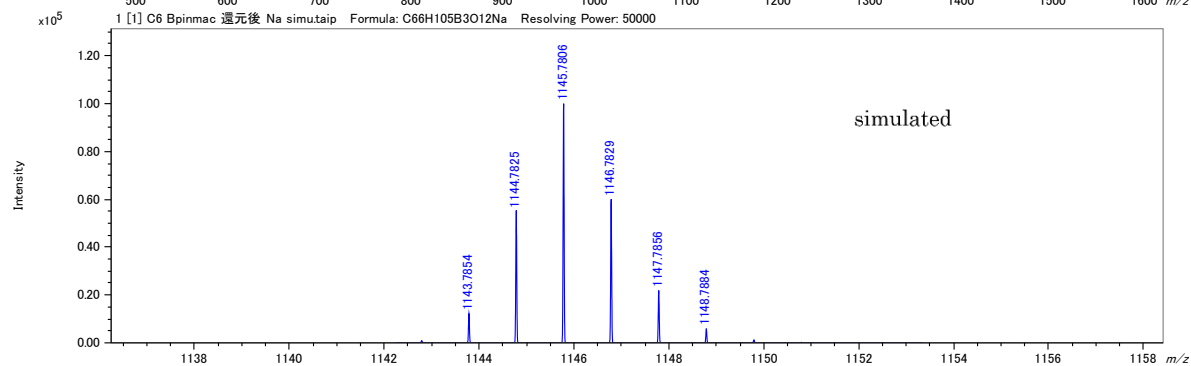
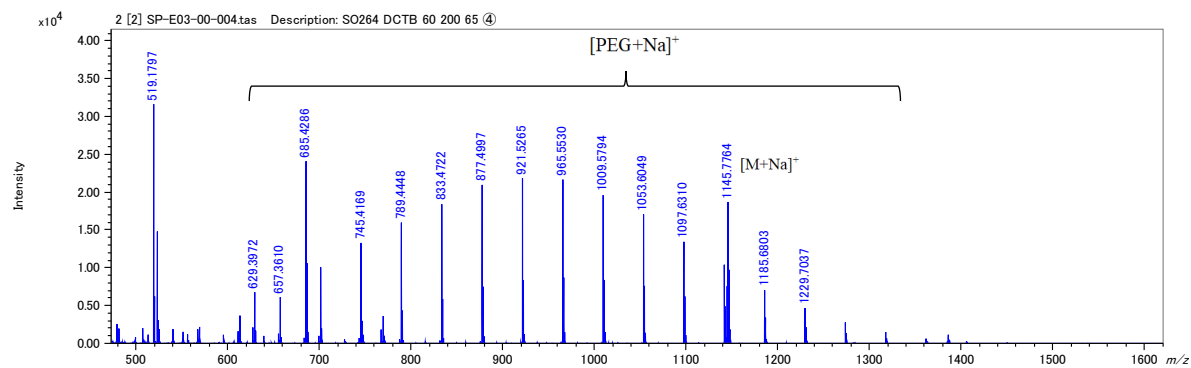
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¹³C NMR



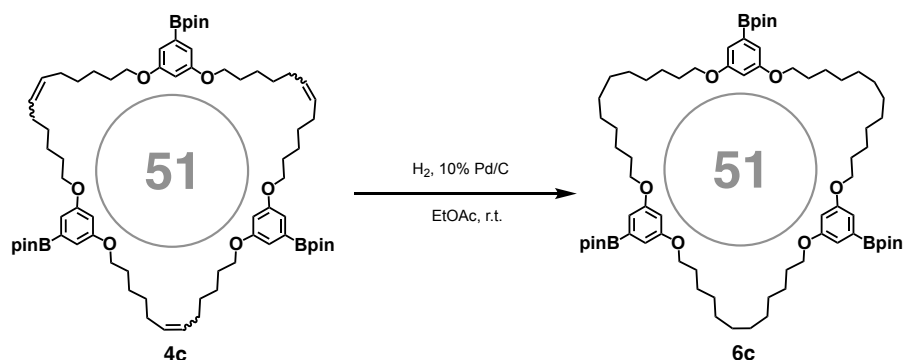




[ピーク検出結果]

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5	1471.9	91.5132	6	1428.99	80.1069	7	1360.53	71.1812	8	1308.46	86.9014
9	1165.28	73.0807	10	1146.47	72.8554	11	1052.46	86.9144	12	968.09	90.2161
13	851.9	86.9928	14	706.783	81.7383						

4-3. Hydrogenation of Macrocycle **4c** to **6c**



Macrocycle **4c** (39.2 mg, 32.6 μmol) and palladium (10%) on charcoal were mixed in EtOAc (8.0 mL). Under hydrogen, the solution was stirred for 19 h and then filtered through Celite. The solvent was evaporated to afford the desired **6c** quantitatively.

Physical data of **6c**

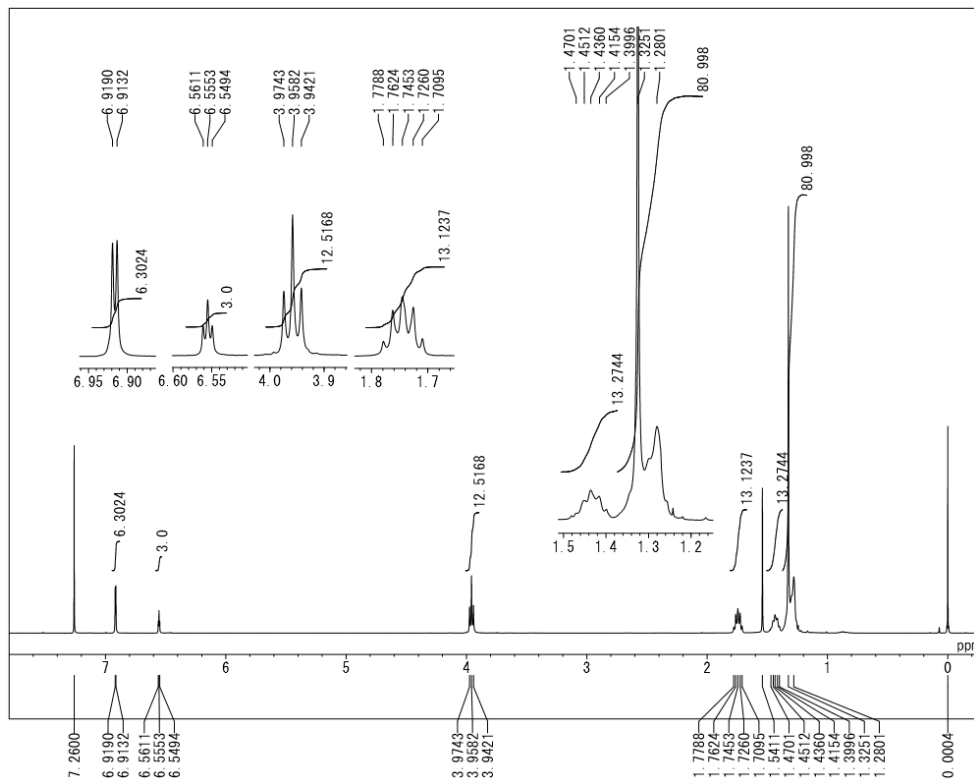
¹H NMR (400 MHz, CDCl₃): δ 6.92 (d, $J = 2.3$ Hz, 6H), 6.56 (t, $J = 2.3$ Hz, 3H), 3.96 (t, $J = 6.4$ Hz, 12H), 1.75 (quin, $J = 6.9$ Hz, 12H), 1.47-1.40 (m, 12H), 1.33 (s, 36H), 1.33-1.28 (m, 36H).

¹³C NMR (100 MHz, CDCl₃): δ 160.1, 112.5, 105.3, 83.9, 68.1, 29.6, 29.6, 29.4, 29.4, 26.1, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 30.4

HRMS (MALDI⁺): m/z calcd. for C₇₂H₁₁₇B₃O₁₂Na: 1229.8747, found: 1229.8778 [M+Na]⁺.

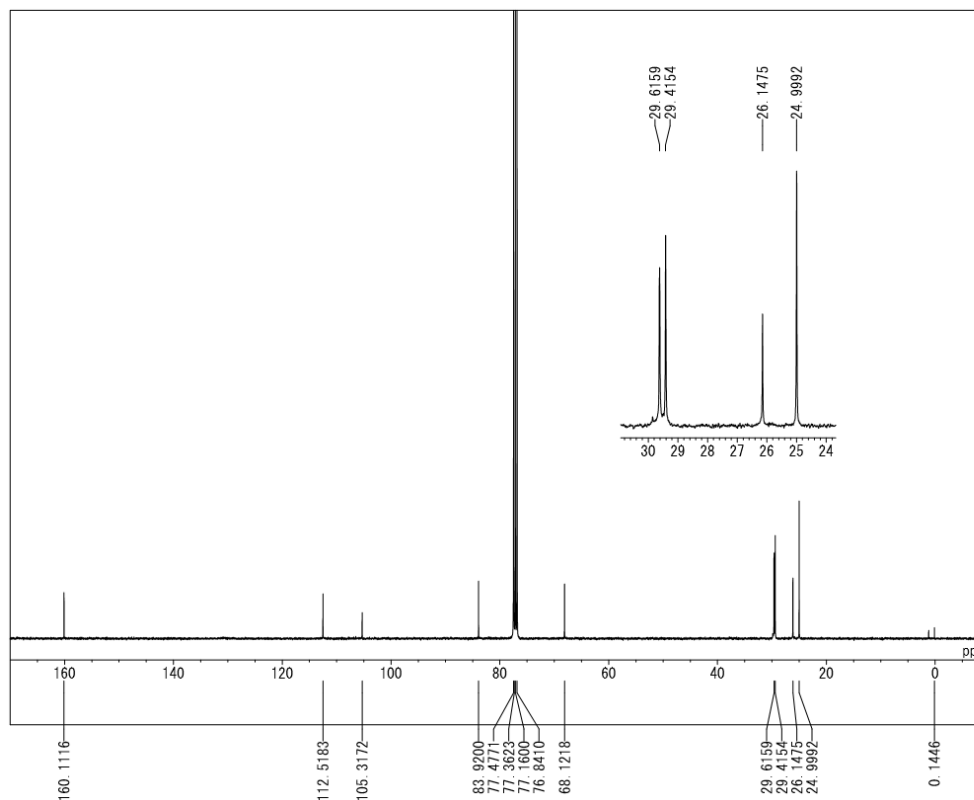
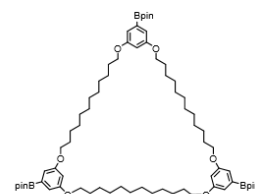
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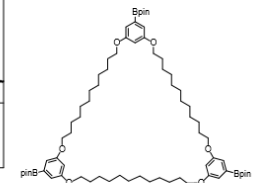
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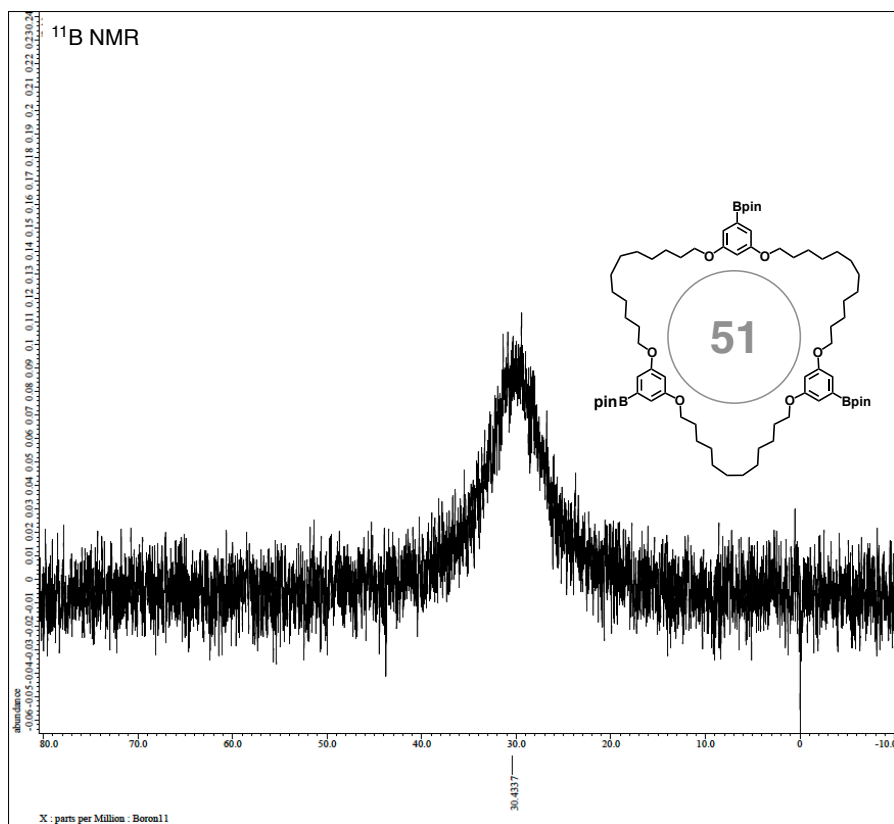


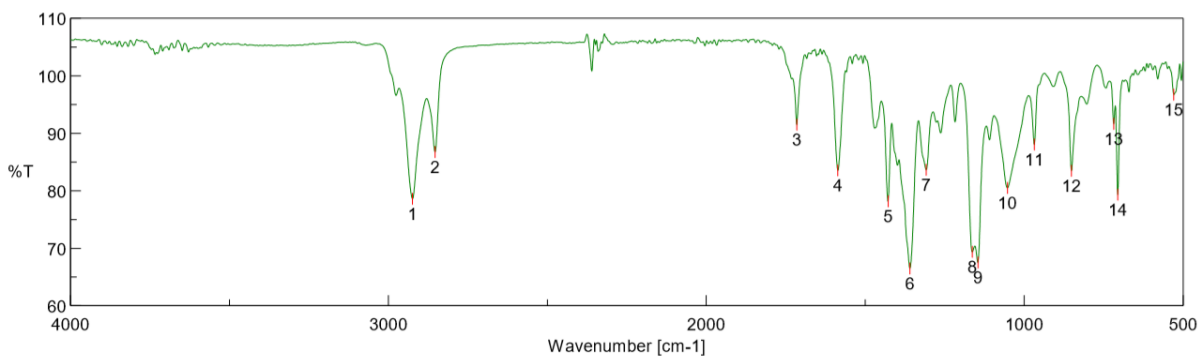
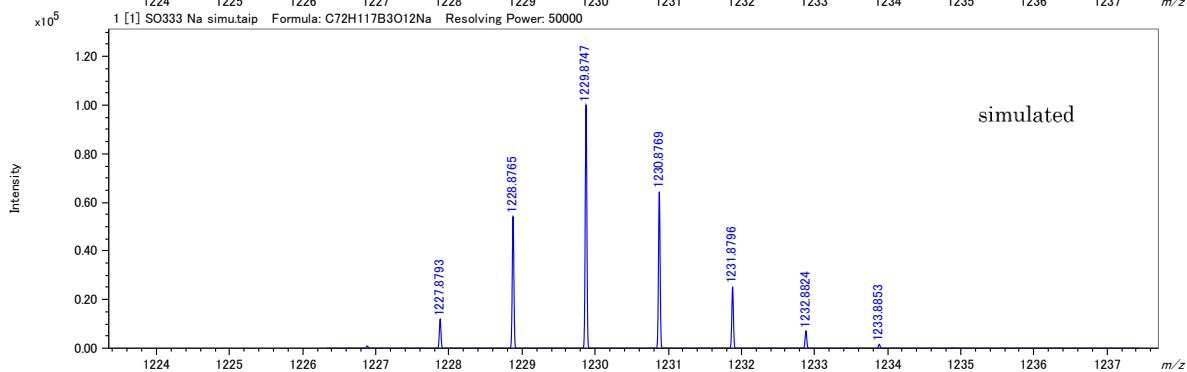
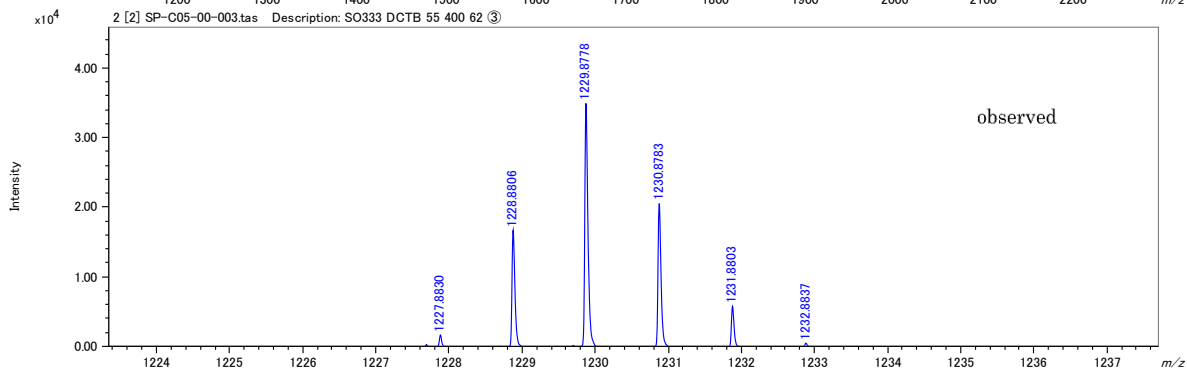
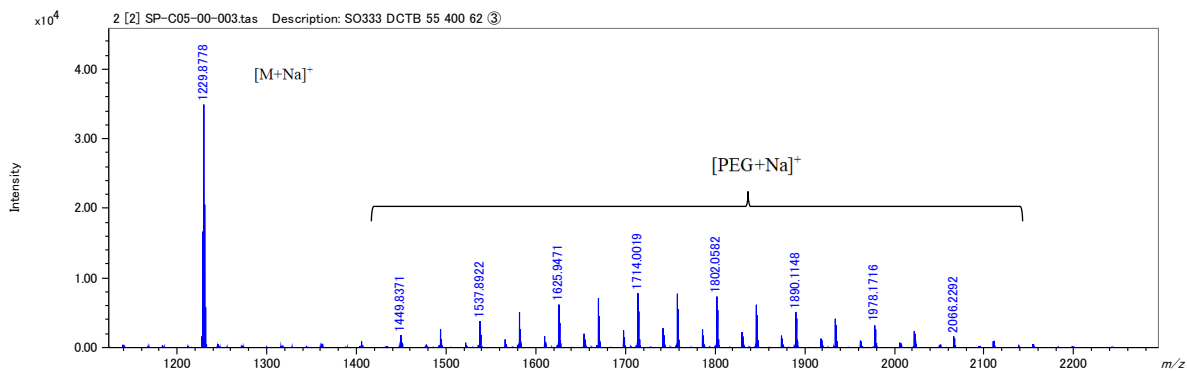
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¹³C NMR



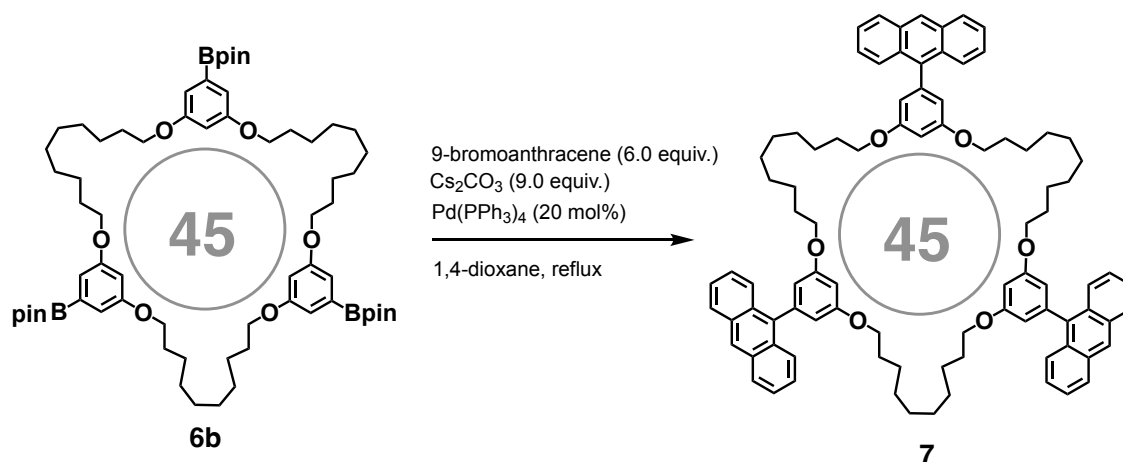




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9	1145.51	67.5313	10	1052.94	80.4953	11	968.09	88.0362	12	851.418	83.4435
13	718.354	91.6253	14	705.819	79.3017	15	529.364	96.7214			

4-4. Suzuki-Miyaura coupling of Macrocycle **6b** and 9-bromoanthracene



Macrocycle **6b** (11.0 mg, 9.8 μmol), 9-bromoanthracene (15.1 mg, 58.7 μmol), Pd(PPh₃)₄ (2.3 mg, 2.0 μmol), Cs₂CO₃ (28.7 mg, 88.1 μmol) were added to a test tube and the test tube was filled with Ar. Degassed 1,4-dioxane (0.5 mL) were added to the test tube and the resulting mixture was refluxed for 24 h. The reaction mixture was treated with water and the organic material was extracted with CHCl₃. The organic layer was washed with brine and dried over Na₂SO₄. The crude product was purified by silica gel column chromatography (CHCl₃/*n*-hexane = 3:2) to afford **7** (8.1 mg, 65% yield).

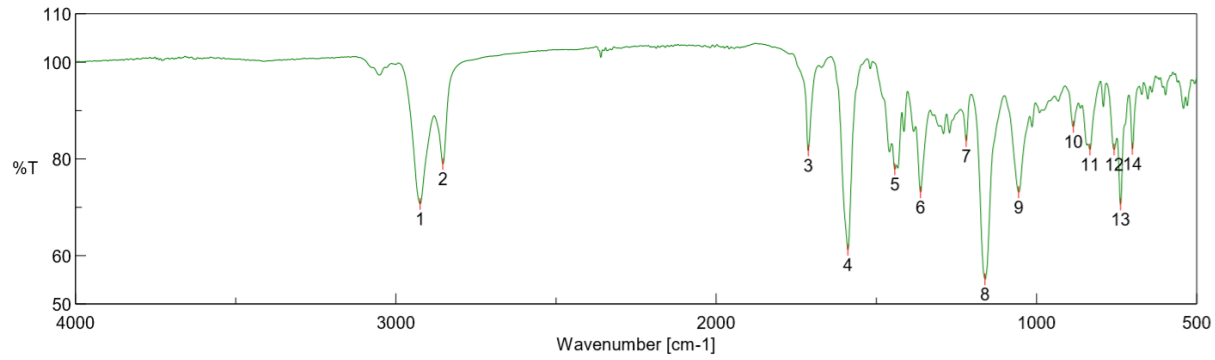
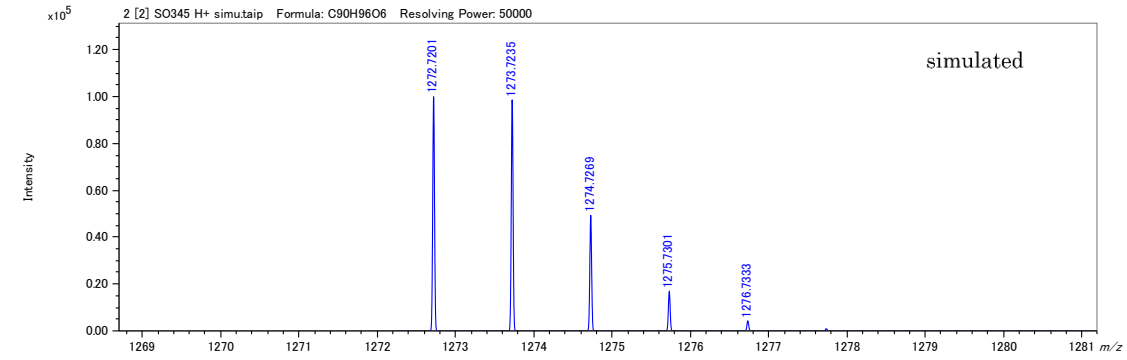
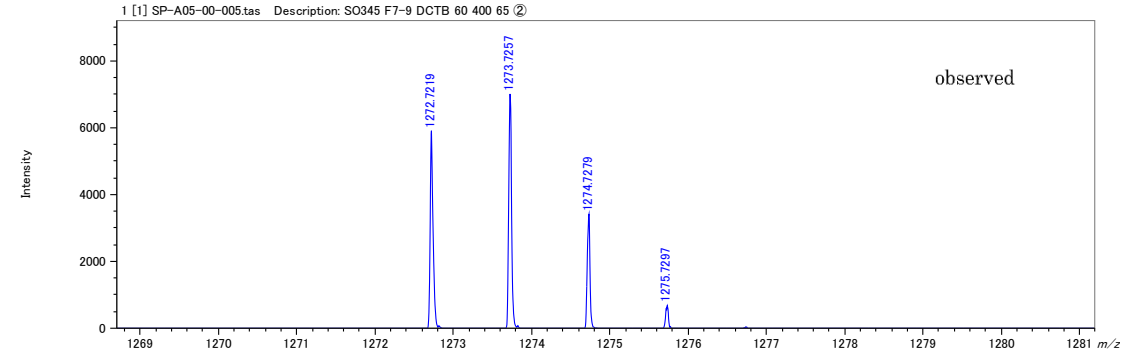
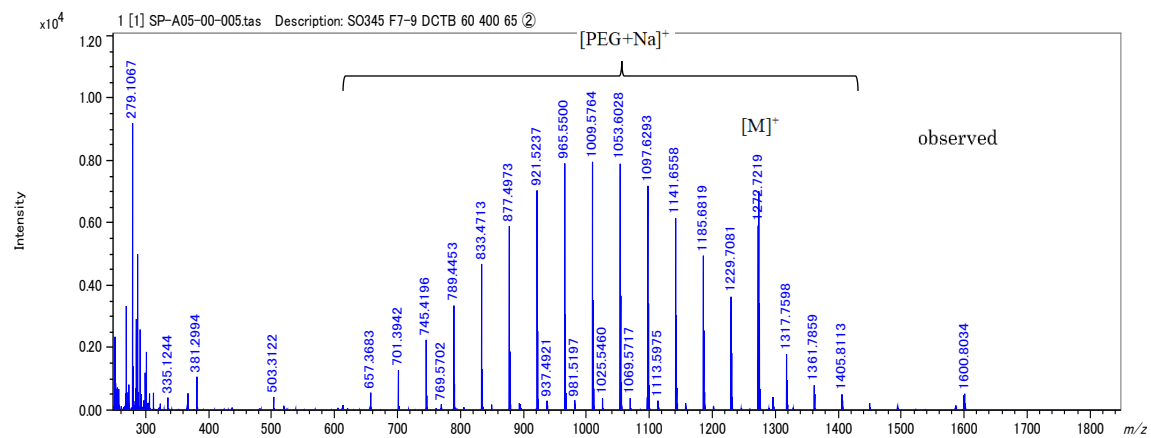
Physical data of **7**

¹H NMR (400 MHz, CDCl₃): δ 8.47 (s, 3H), 8.02 (dd, J = 8.4 Hz, 6H), 7.77 (dd, J = 8.8, 0.8 Hz, 6H), 7.44 (ddt, J = 8.4, 6.6, 1.4 Hz, 6H), 7.35 (ddt, J = 8.8, 6.6, 1.4 Hz, 6H), 6.64 (t, J = 2.3 Hz, 3H), 6.57 (d, J = 2.3 Hz, 6H), 3.98 (t, J = 6.5 Hz, 12H), 1.78 (quin, J = 7.1 Hz, 12H), 1.50-1.44 (m, 12H), 1.38-1.30 (m, 24H).

¹³C NMR (100 MHz, CDCl₃): δ 160.4, 140.8, 137.3, 131.5, 130.1, 128.4, 127.2, 126.6, 125.4, 125.2, 110.0, 100.9, 68.3, 29.5, 29.4, 29.4, 26.1.

HRMS (MALDI⁺): m/z calcd. for C₉₀H₉₆O₆: 1272.7201, found: 1272.7219 [M]⁺.

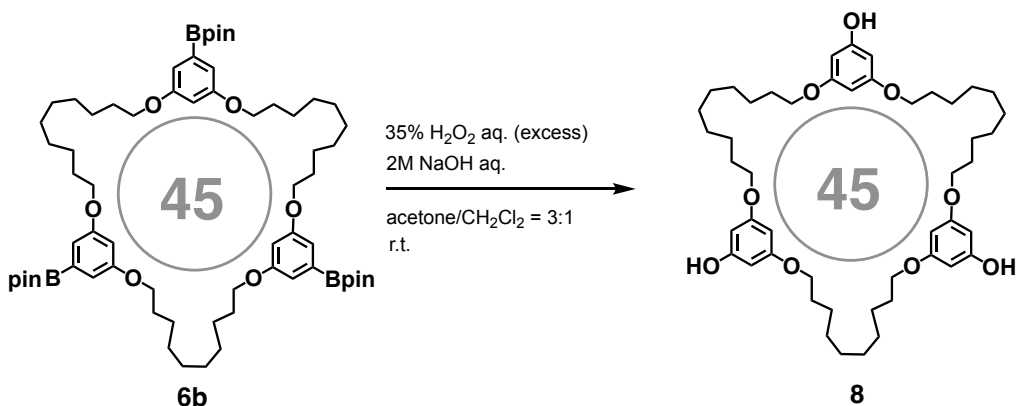
IR (ATR): 2925, 2853, 1712, 1589, 1442, 1362, 1220, 1161, 1056, 885, 833, 758, 738, 701 cm⁻¹.



[ピーク検出結果]

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5	1442.49	77.8298	6	1362.46	73.0868	7	1219.76	83.6846	8	1160.94	55.1092
9	1055.84	73.0851	10	885.166	86.6278	11	833.098	81.941	12	757.888	81.9401
13	737.639	70.5535	14	700.998	82.0879						

4-5. Oxidation of Macrocycle **6b** to **8**



Macrocycle **6b** (17.9 mg, 15.9 μ mol) was dissolved in a mixture of acetone (8.6 mL) and CH₂Cl₂ (2.9 mL). 2 M NaOH aq. (90 equiv.) and 35% H₂O₂ aq. (excess) was added and stirred at room temperature. After 22 h, the reaction mixture was quenched by 1 N HCl aq. and the organic solvents were evaporated. The residue was treated with water and the organic materials was extracted with CHCl₃. The organic layer was washed with brine and dried over MgSO₄. The crude product was washed with hexane to afford **8** (10.1 mg, 80% yield).

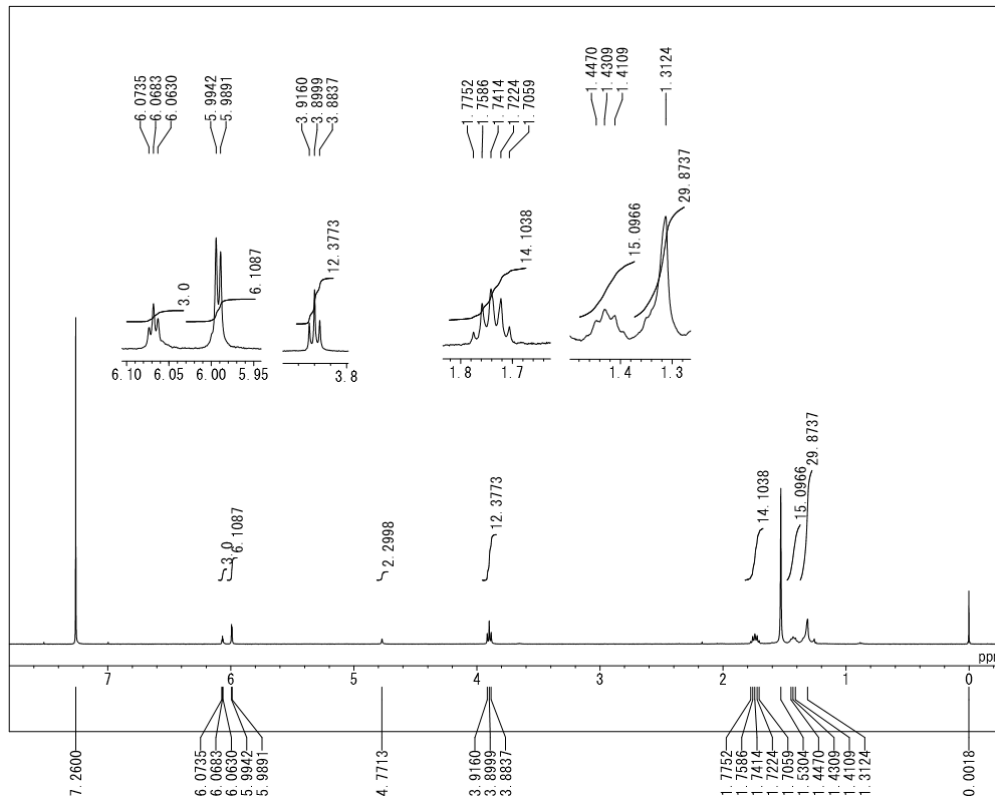
Physical data of **8**

¹H NMR (400 MHz, CDCl₃): δ 6.07 (t, J = 2.1 Hz, 3H), 5.99 (d, J = 2.1 Hz, 6H), 4.77 (s, 3H), 3.90 (t, J = 6.5 Hz, 12H), 1.74 (quin, J = 6.9 Hz, 12H), 1.45-1.41 (m, 12H), 1.38-1.30 (m, 24H).

¹³C NMR (100 MHz, CDCl₃): δ 161.3, 157.4, 94.8, 94.4, 68.1, 29.3, 29.2, 26.0.

HRMS (MALDI⁺): m/z calcd. for C₄₈H₇₂O₉Na: 815.5069, found: 815.5080 [M+Na]⁺.

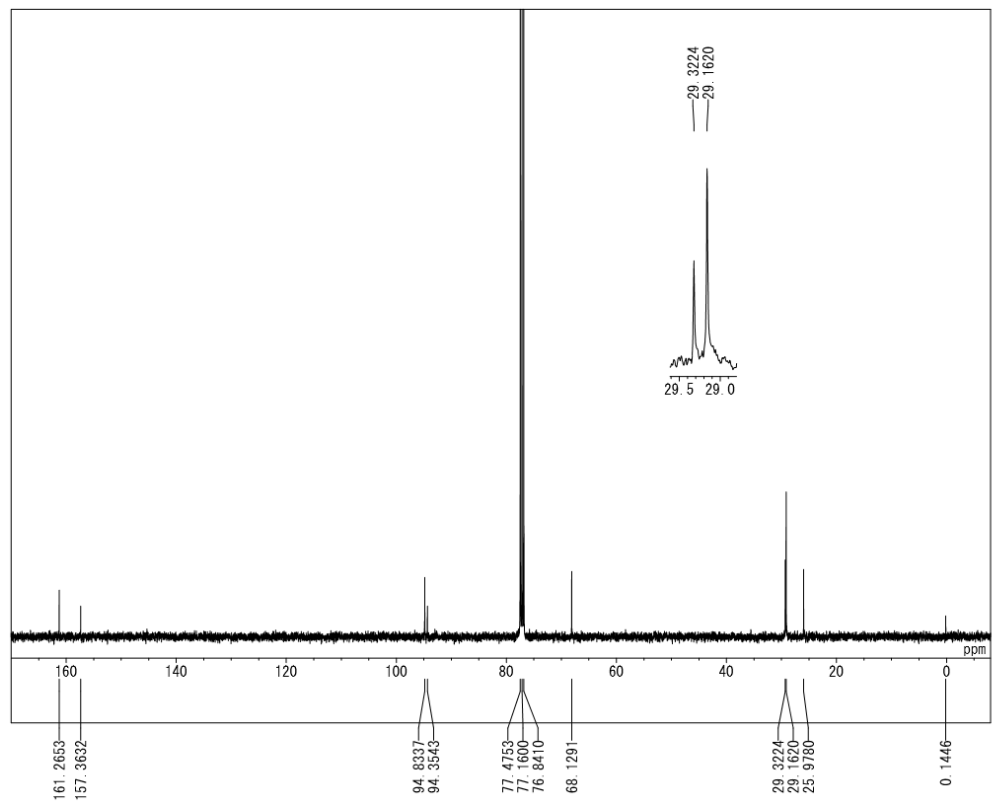
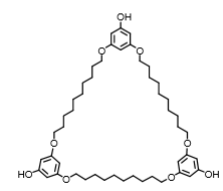
IR (ATR): 3348, 2922, 2851, 1594, 1497, 1463, 1387, 1145, 1092, 821, 593 cm⁻¹.



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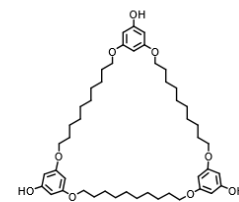
¹H NMR

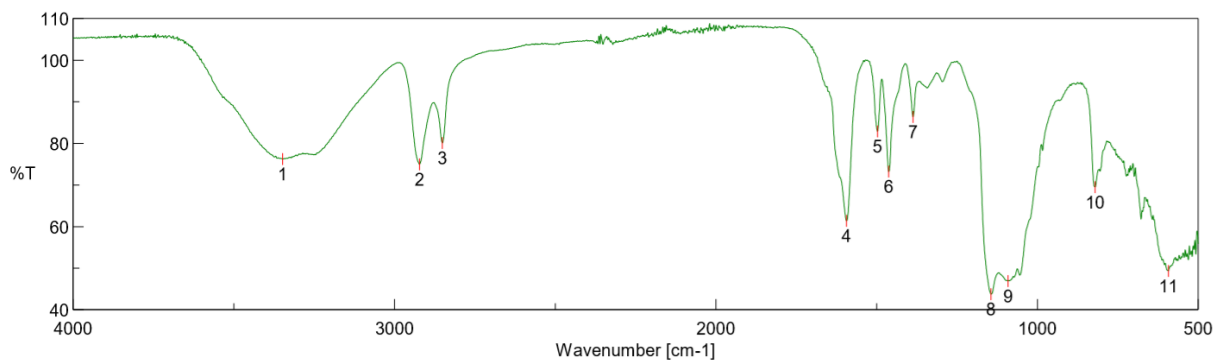
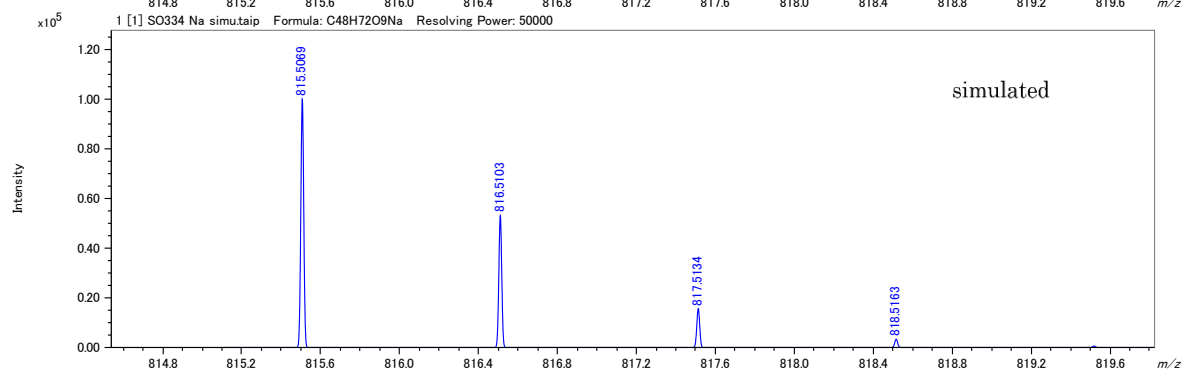
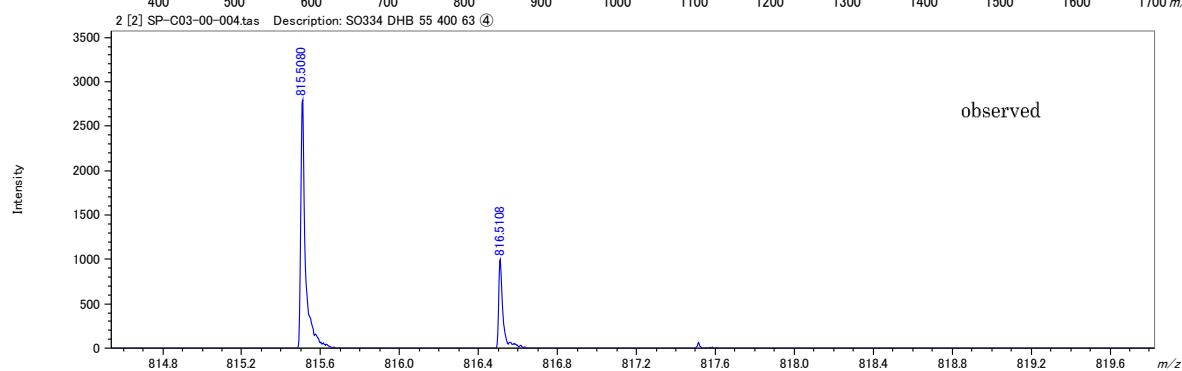
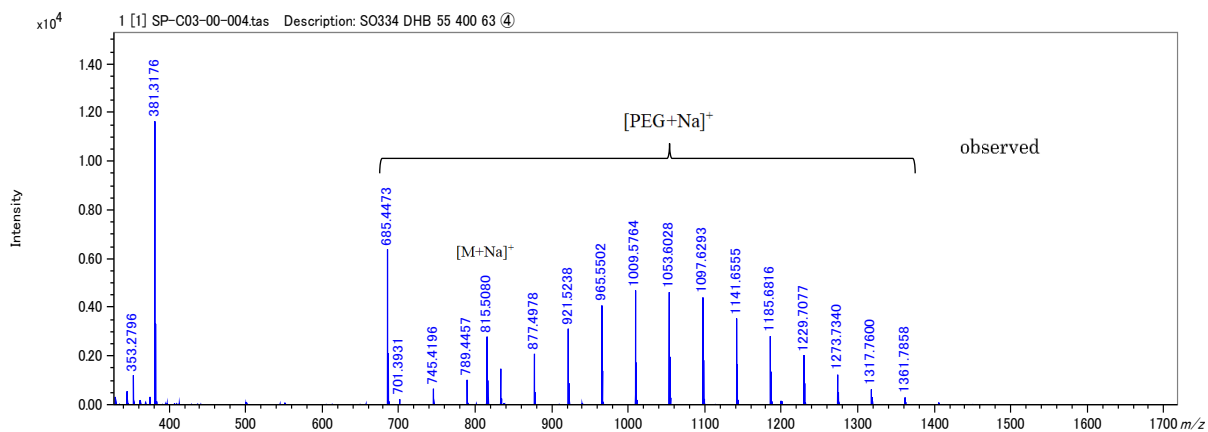


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¹³C NMR

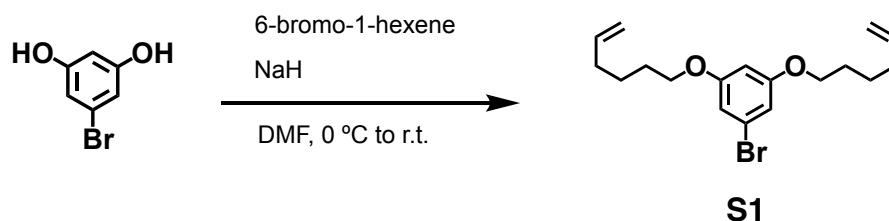




[ピーク検出結果]

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1	3347.82	76.2095	2	2922.11	74.9486	3	2851.24	80.0639	4	1593.88	61.3322
5	1497.45	82.8395	6	1462.74	73.1704	7	1387.05	86.3188	8	1144.55	43.7068
9	1091.51	46.8521	10	820.563	69.4463	11	592.521	49.2967			

5. Synthetic Procedures and Characterization of New Compounds



Under an Ar atmosphere, NaH (60% in oil; 320 mg, 8.0 mmol) was dissolved in dry DMF (13.3 mL) and then 5-bromoresorcinol (756 mg, 4.0 mmol) was added at 0 °C. The mixture was stirred at 0 °C for 1 h, and then 6-bromo-1-hexene (1.6 mL, 12.0 mmol) was added slowly and further stirred at room temperature. After 12 h, water was added and the product was extracted with *n*-hexane/EtOAc = 4:1 and then combined extract was washed with brine and dried over MgSO₄. After solvent was evaporated, the residue was purified by column chromatography (*n*-hexane/EtOAc = 9:1) to give **S1** (1.41 g, quant.).

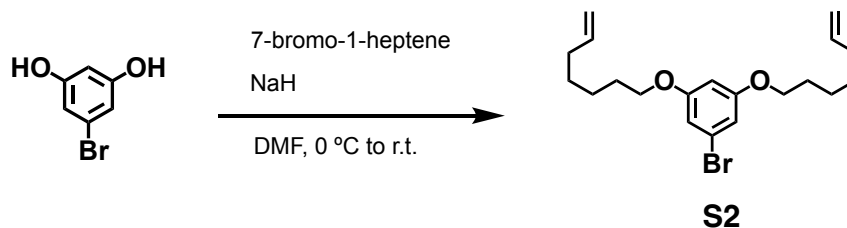
Physical data of **S1**

¹H NMR (400 MHz, CDCl₃): δ 6.64 (d, *J* = 2.2 Hz, 2H), 6.36 (t, *J* = 2.2 Hz, 1H), 5.82 (ddt, *J* = 17.1, 10.2, 6.7 Hz, 2H), 5.06-5.01 (m, 2H), 4.99-4.96 (m, 2H), 3.91 (t, *J* = 6.44 Hz, 4H), 2.15-2.09 (m, 4H), 1.78 (quin, *J* = 7.0 Hz, 4H), 1.59-1.51 (m, 4H).

¹³C NMR (100 MHz, CDCl₃): δ 160.8, 138.6, 123.0, 115.0, 110.4, 100.8, 68.2, 33.5, 28.7, 25.4.

HRMS (MALDI⁺): *m/z* calcd. for C₁₈H₂₅BrO₂Na: 375.0930, found: 375.0929 [M+Na]⁺.

IR (ATR): 3077, 2924, 2871, 1597, 1575, 1451, 1439, 1386, 1278, 1053, 990, 911, 833, 677 cm⁻¹.



Under an Ar atmosphere, NaH (60% in oil; 240 mg, 6.0 mmol) was dissolved in dry DMF (10.0 mL) and then 5-bromoresorcinol (567 mg, 3.0 mmol) was added at 0 °C. The mixture was stirred at 0 °C for 1 h, and then 7-bromo-1-heptene (1.37 mL, 9.0 mmol) was added slowly and further stirred at room temperature. After 22 h, water was added and the product was extracted with hexane/EtOAc = 4:1 and then combined extract was washed by brine and dried over MgSO₄. The solvent was evaporated to give **S2** (1.14 g, quant.).

Physical data of **S2**

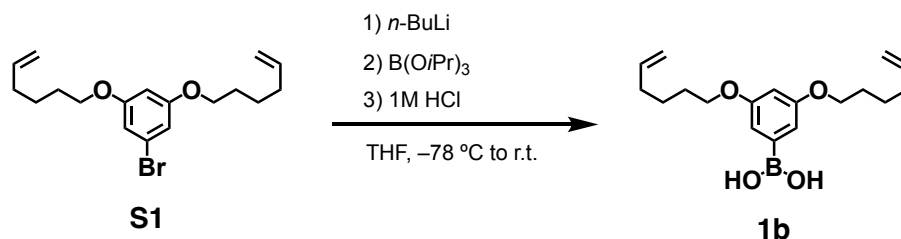
¹H NMR (400 MHz, CDCl₃): δ 6.63 (d, *J* = 2.2 Hz, 2H), 6.36 (t, *J* = 2.2 Hz, 1H), 5.82 (ddt, *J* = 17.1, 10.3,

6.8 Hz, 2H), 5.03-4.98 (m, 2H), 4.97-4.94 (m, 2H), 3.90 (t, $J = 6.5$ Hz, 4H), 2.10-2.06 (m, 4H), 1.76 (quin, $J = 6.9$ Hz, 4H), 1.47-1.43 (m, 8H).

^{13}C NMR (100 MHz, CDCl_3): δ 160.9, 138.9, 123.0, 114.6, 110.4, 100.8, 68.3, 33.8, 29.1, 28.7, 25.6.

HRMS (MALDI⁺): m/z calcd. for $\text{C}_{20}\text{H}_{29}\text{BrO}_2\text{Na}$: 403.1243, found: 403.1258 $[\text{M}+\text{Na}]^+$.

IR (ATR): 3075, 2926, 2857, 1597, 1575, 1438, 1387, 1278, 1165, 1052, 991, 911, 831, 676 cm^{-1} .



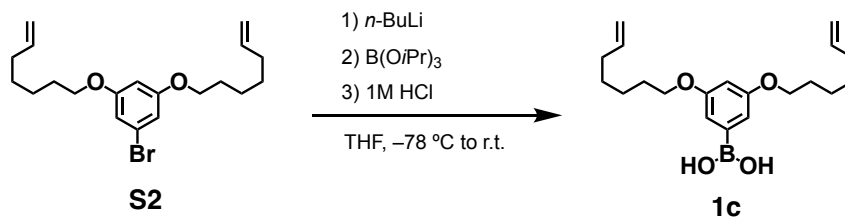
Under an Ar atmosphere, **S1** (700 mg, 19.8 mmol) was dissolved in dry THF (6.6 mL) and then *n*-BuLi (1.57 M in hexane, 1.39 mL, 2.18 mmol) was added dropwise to the solution at -78 °C. The mixture was stirred at -78 °C for 1 h, and then B(O*i*Pr)₃ (500 μL , 2.18 mmol) was added and warmed up to room temperature over 2 h. After warmed up, 1N HCl aq. was added and the product was extracted with *tert*-butyl methyl ether and then combined extract was washed with brine and dried over MgSO_4 . The solvent was evaporated and the solid was washed with CH_3CN to give **1b** (397 mg, 63%).

Physical data of **1b**

^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 7.99 (s, 2H), 6.92 (d, $J = 2.2$ Hz, 2H), 6.48 (t, $J = 2.2$ Hz, 1H), 5.82 (ddt, $J = 17.1, 10.3, 6.7$ Hz, 2H), 5.06-5.01 (m, 2H), 4.98-4.96 (m, 2H), 3.93 (t, $J = 6.5$ Hz, 4H), 2.11-2.05 (m, 4H), 1.70 (quin, $J = 6.9$ Hz, 4H), 1.53-1.46 (m, 4H).

^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 159.2, 138.6, 114.9, 112.0, 103.1, 67.1, 32.8, 28.2, 24.8. The boron-bound carbons were not detected due to quadrupole relaxation.

^{11}B NMR (160 MHz, $\text{DMSO}-d_6$): δ 28.5



Under an Ar atmosphere, **S2** (1.07 g, 2.8 mmol) was dissolved in dry THF (9.3 mL) and then *n*-BuLi (1.57 M in hexane, 1.96 mL, 3.08 mmol) was added dropwise to the solution at -78 °C. The mixture was stirred at -78 °C for 1 h, and then B(O*i*Pr)₃ (775 μL , 3.36 mmol) was added and warmed up to room temperature

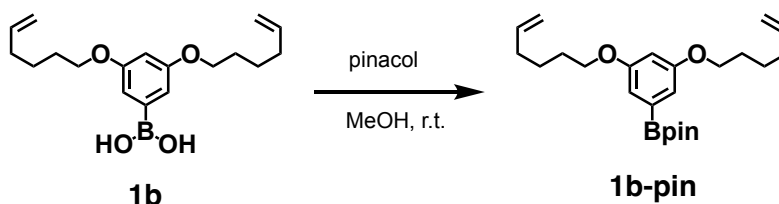
over 2 h. After warmed up, 1N HCl aq. was added and the product was extracted with *tert*-butyl methyl ether and then combined extract was washed by brine and dried over MgSO₄. After solvent was evaporated, the residue was purified by column chromatography (*n*-hexane/EtOAc = 9:1, then EtOAc only) to give **1c** (405 mg, 40%).

Physical data of **1c**

¹H NMR (400 MHz, DMSO-*d*₆): δ 7.99 (s, 2H), 6.91 (d, *J* = 2.3 Hz, 2H), 6.47 (t, *J* = 2.3 Hz, 1H), 5.81 (ddt, *J* = 17.1, 10.2, 6.7 Hz, 2H), 5.04-4.99 (m, 2H), 4.97-4.93 (m, 2H), 3.92 (t, *J* = 6.6 Hz, 4H), 2.07-2.02 (m, 4H), 1.73-1.66 (m, 4H), 1.43-1.39 (m, 8H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ 159.3, 138.7, 114.8, 112.0, 103.1, 67.2, 33.2, 28.6, 28.0, 25.1. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, DMSO-*d*₆): δ 28.3



Boronic acid **1b** (5.5 mg, 17 μmol) was treated with pinacol (6.0 mg, 51 μmol) in methanol. After the mixture was evaporated, the residue was purified by GPC to give **1b-pin** (4.29 mg, 71%).

Physical data of **1b-pin**

¹H NMR (400 MHz, CDCl₃): δ 6.92 (d, *J* = 2.4 Hz, 2H), 6.55 (t, *J* = 2.3 Hz, 1H), 5.83 (ddt, *J* = 17.0, 10.3, 6.7 Hz, 2H), 5.05-5.00 (m, 2H), 4.98-4.95 (m, 2H), 3.97 (t, *J* = 6.4 Hz, 4H), 2.15-2.09 (m, 4H), 1.78 (quin, *J* = 7.0 Hz, 4H), 1.57-1.52 (m, 4H), 1.33 (s, 12H).

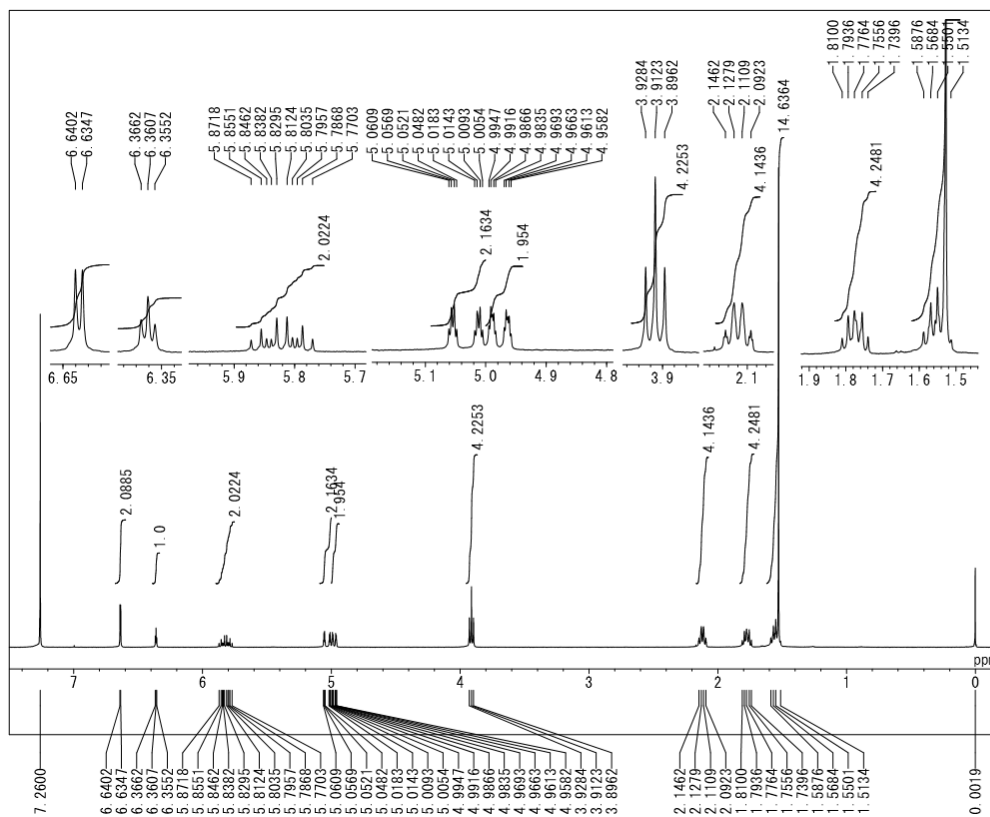
¹³C NMR (100 MHz, CDCl₃): δ 160.1, 138.8, 114.8, 112.5, 105.3, 84.0, 67.9, 33.6, 28.9, 25.5, 25.0. The boron-bound carbons were not detected due to quadrupole relaxation.

¹¹B NMR (160 MHz, CDCl₃): δ 32.2

HRMS (MALDI⁺): *m/z* calcd. for C₂₄H₃₇BO₄Na: 423.2681, found: 423.2678 [M+Na]⁺.

IR (ATR): 3075, 2977, 2934, 2868, 1641, 1586, 1429, 1356, 1308, 1164, 1138, 1052, 910, 851, 706 cm⁻¹.

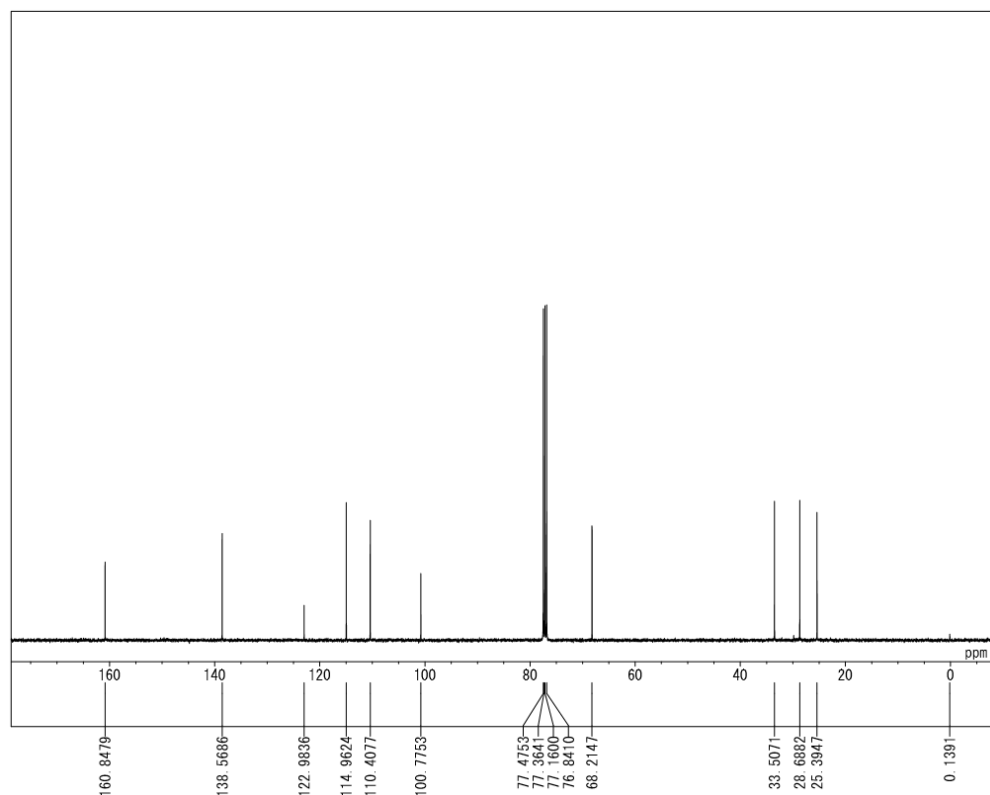
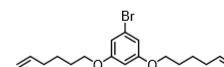
7. ¹H NMR, ¹³C NMR and ¹¹B NMR spectra



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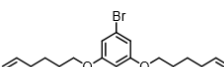
¹H NMR

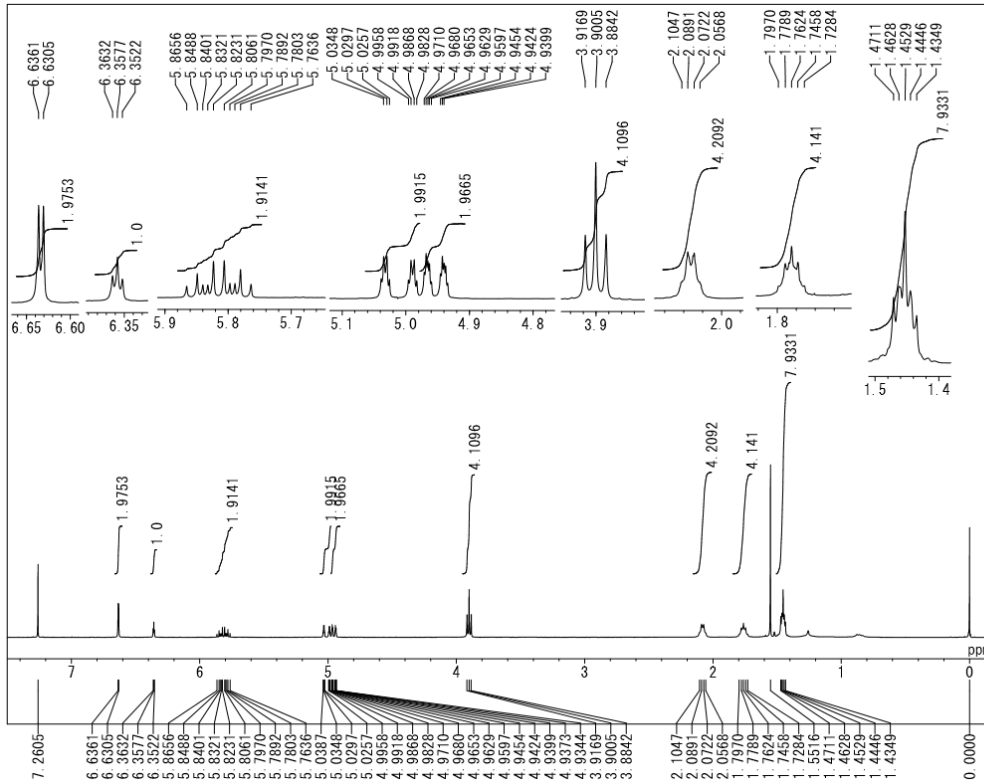


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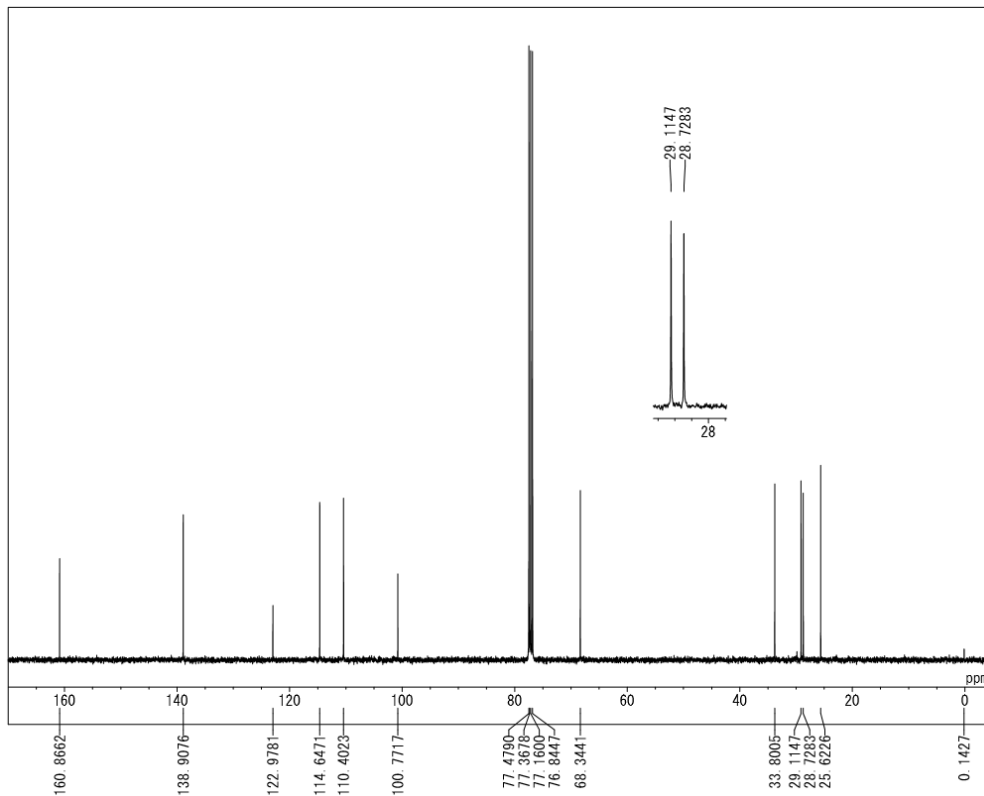
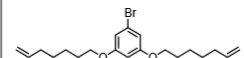
¹³C NMR





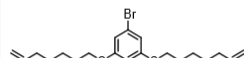
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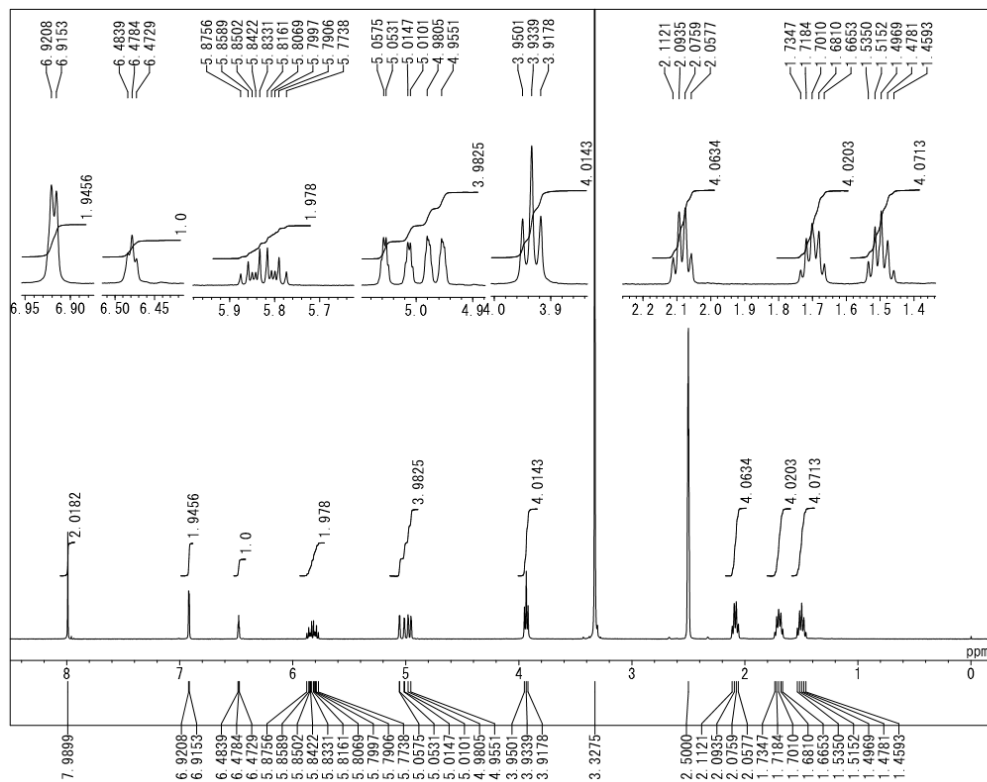
¹H NMR



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¹³C NMR

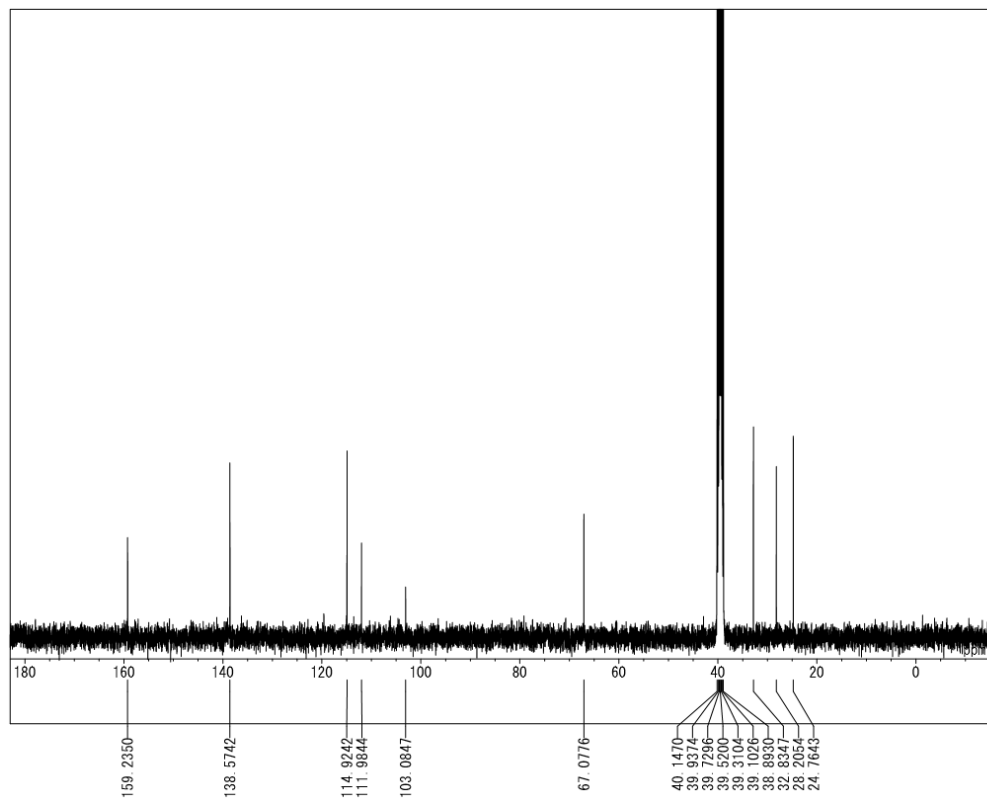
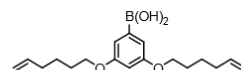




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GRDPRG
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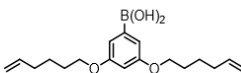
¹H NMR

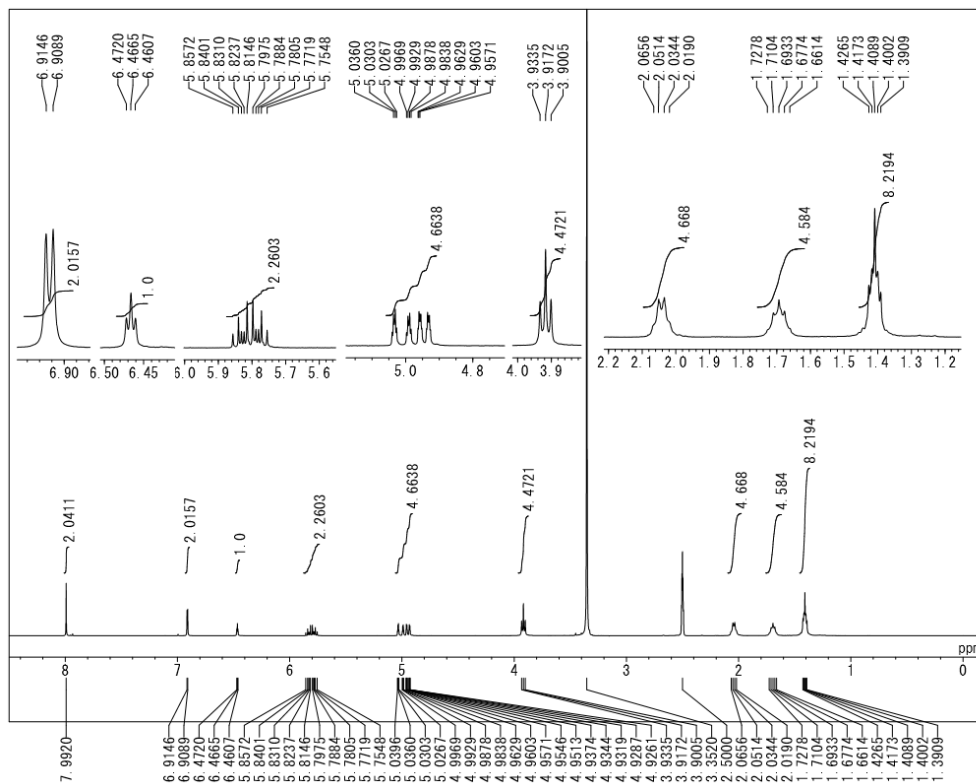
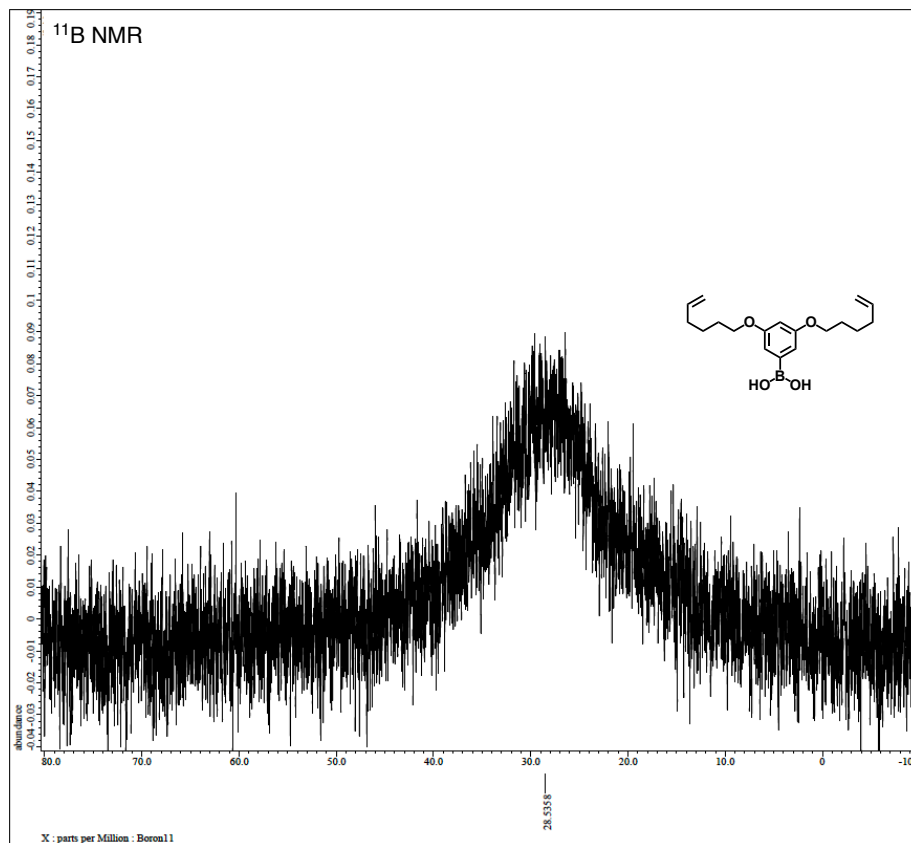


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IRNUC OFF
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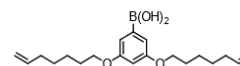
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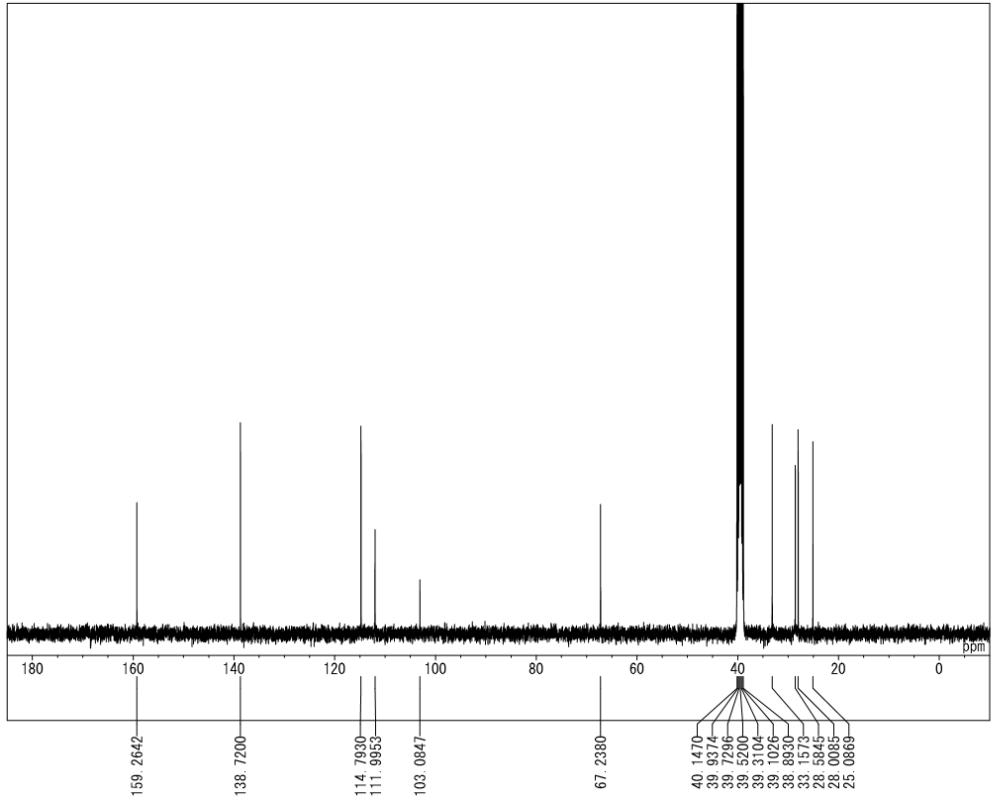




D:\Users\sator\OneDrive\修論データ\spectrol集\7 B(OH)2標\C7 B(OH)2標 1H DMSO (SO288).R
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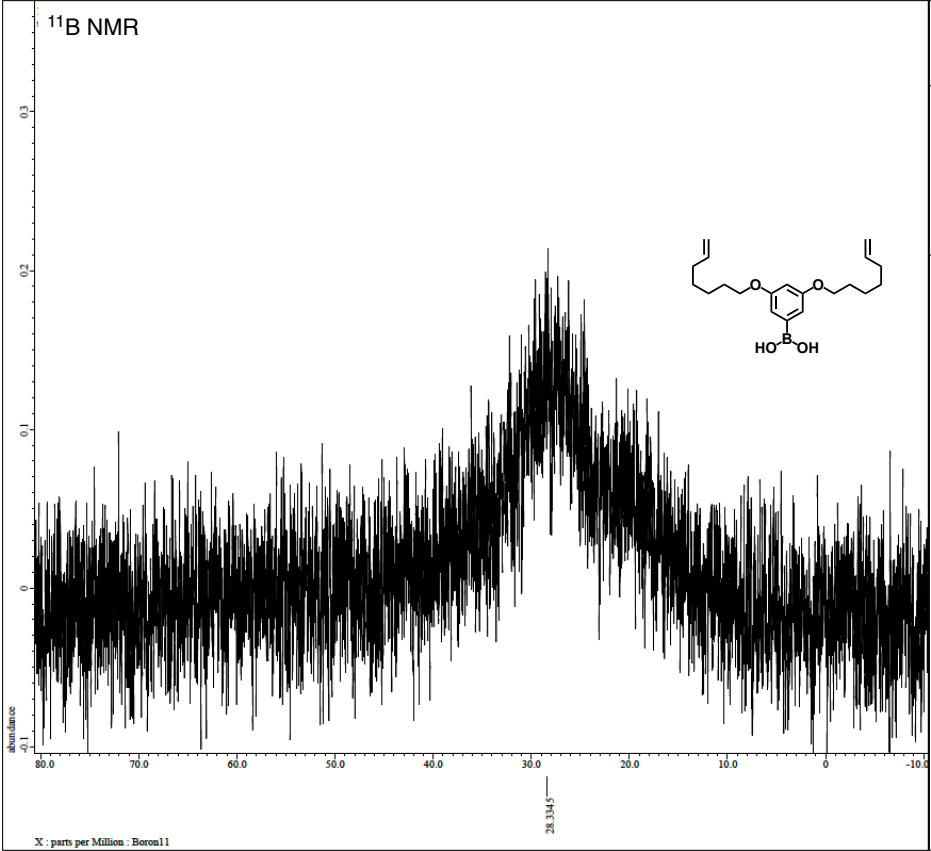
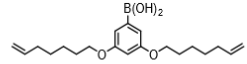
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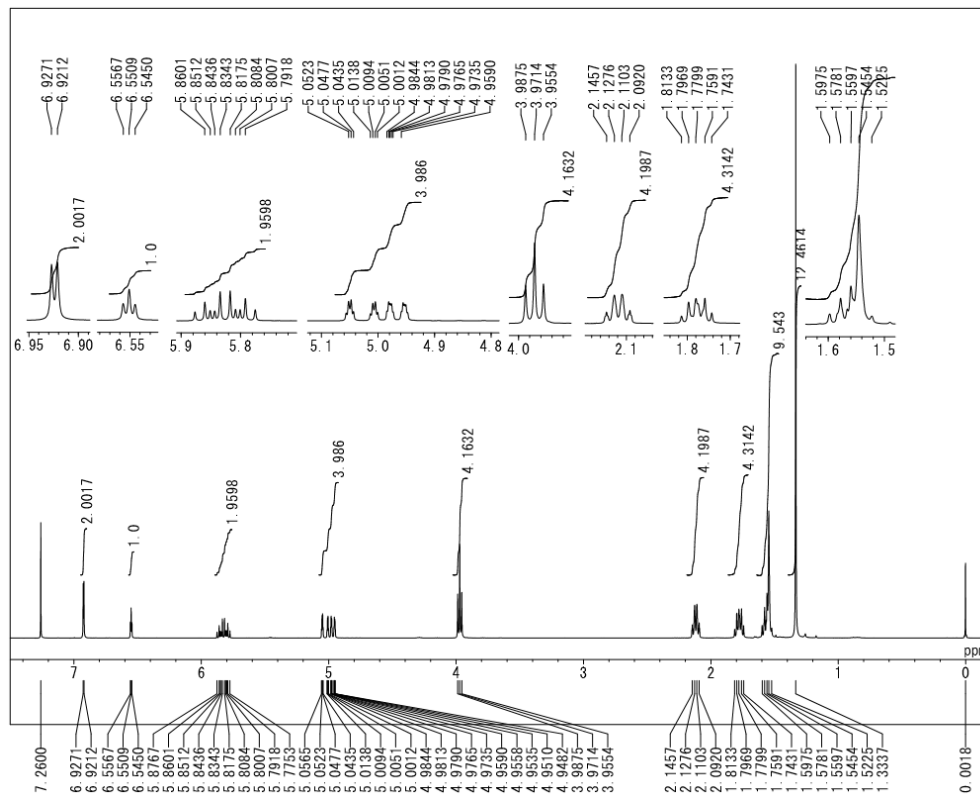




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 PROBHD 5 MM PABBO BB-1H/D Z-GRD Z104450
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 SLVNT DMSO
 BF 0.25 Hz
 RGAIN 114

¹³C NMR

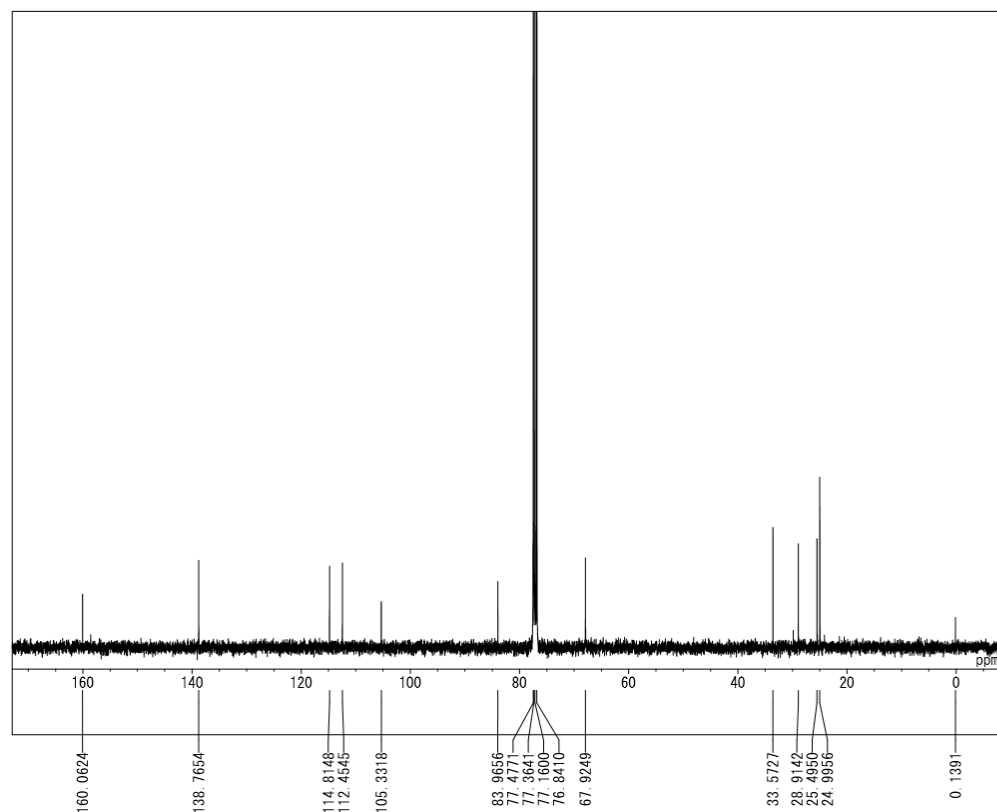
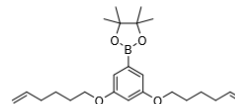




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 -S1-BBF/H
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 RGAIN 101

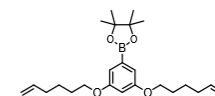
¹H NMR

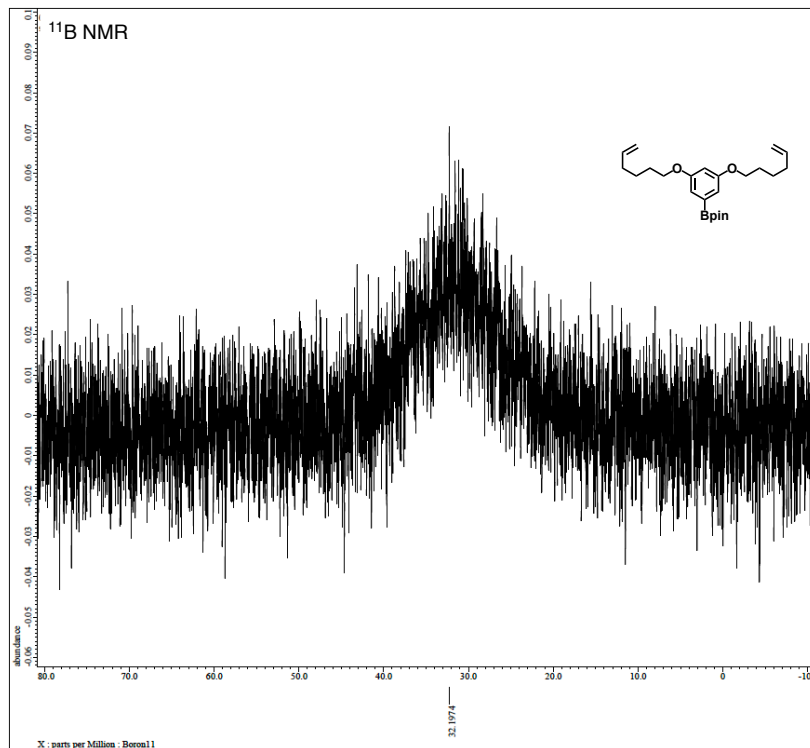


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 GRD Z104450
 INSTRUM SPECT
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 GRDPRG
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 SLVNT CDCl₃
 BF 0.25 Hz
 RGAIN 81

¹³C NMR





6. X-ray Crystallographic Analysis of **3c**

Single crystals of **3c** for X-ray diffraction analysis were obtained as colorless block by diffusion of toluene to a THF solution of the crude product of **3c**.

The single X-ray structure determination was performed on Rigaku RAPID (CuK α radiation, $\lambda = 1.54187 \text{ \AA}$). A numerical absorption correction (μ) was applied. The structure was solved by direct methods and refined by the full-matrix least-squares method on F^2 with anisotropic temperature factors for non-hydrogen atoms. All the hydrogen atoms were located at the calculated positions and refined with riding. The disordered alkylene chains of **3c** were restricted by DFIX and DELU. Crystallographic data collection and refinement information is listed in Table S1. CCDC reference number 2195814.

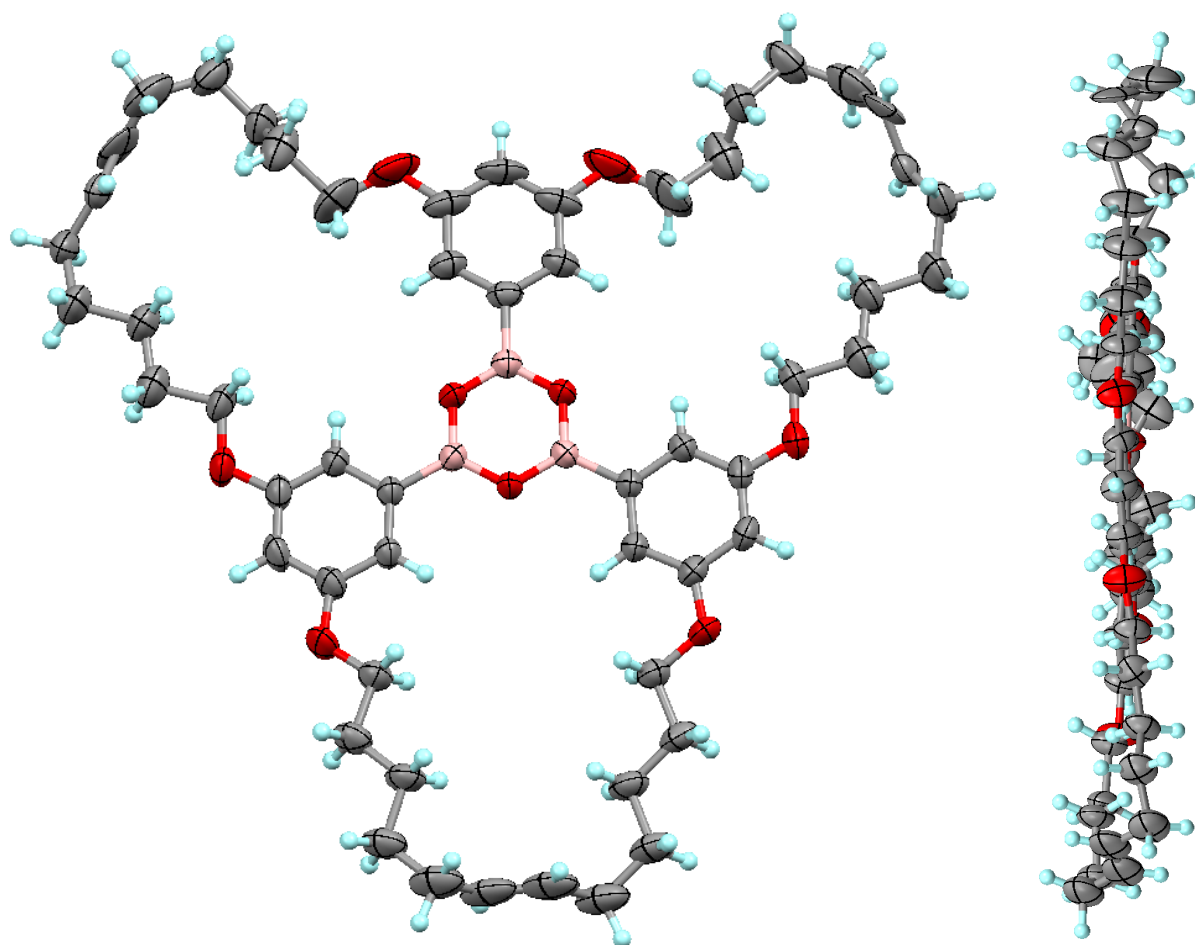


Table S1. Crystal data and structure refinement for 3c

CCDC number	2195814
Identification code	191002os
Empirical formula	C ₅₄ H ₇₅ B ₃ O ₉
Formula weight	900.61
Temperature	193 K
Wavelength	1.54187 Å
Crystal system	monoclinic
Space group	<i>C2/c</i>
Unit cell dimensions	$a = 15.6791(3) \text{ \AA}$ $\alpha = 90^\circ$. $b = 19.9517(4) \text{ \AA}$ $\beta = 110.201(8)^\circ$. $c = 18.0668(3) \text{ \AA}$ $\gamma = 90^\circ$.
Volume	5304.1(3) Å ³
Z	4
Density (calculated)	1.128 Mg/m ³
Absorption coefficient	0.587 mm ⁻¹
F(000)	1944.00
Crystal size	0.30 x 0.30 x 0.15 mm ³
Theta range for data collection	3.73 to 68.30°.
Index ranges	-18 ≤ h ≤ 18, -24 ≤ k ≤ 24, -21 ≤ l ≤ 21
Reflections collected	30762
Independent reflections	4866 [R(int) = 0.0566]
Completeness to theta = 25.24°	99.8 %
Absorption correction	multi-scan
Max. and min. transmission	0.916 and 0.632
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4866 / 36 / 381
Goodness-of-fit on F ²	1.054
Final R indices [I > 2σ(I)]	R ₁ = 0.0830, wR ₂ = 0.1690
R indices (all data)	R ₁ = 0.1492, wR ₂ = 0.1976
Largest diff. peak and hole	0.24 and -0.19 e/Å ⁻³

9. References

- S1) R. El-Haggag, K. Kamikawa, K. Machi, Z. Ye, Y. Ishino, T. Tsumuraya, I. Fujii, *Bioorg. Med. Chem. Lett.* **2010**, *20*, 1169–1172.
- S2) C. Simocko, T. C. Young, K. B. Wagener, *Macromolecules*, **2015**, *48*, 5470-5473.