

Supporting Information

The quest for unligated oxaphosphiranes with phosphorus in different coordination numbers

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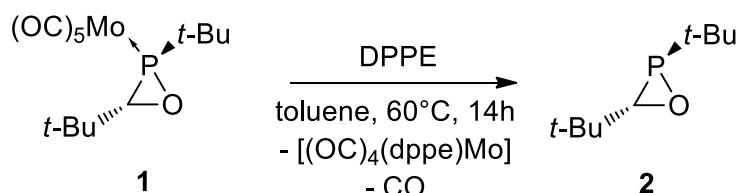
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General working techniques

The syntheses of all compounds were performed under an argon atmosphere, using common Schlenk techniques and dry solvents. Toluene, tetrahydrofuran, diethyl ether and *n*-pentane were dried over sodium wire, CH₂Cl₂ over CaH₂ and further purified by subsequent distillation. All NMR spectra were recorded on a Bruker AVI-300 or a Bruker AV III HD Prodigy 500 spectrometer. The ¹H and ¹³C NMR spectra were referenced to the residual proton resonances and the ¹³C NMR signals of the deuterated solvents and ³¹P to 85% H₃PO₄ as external standards, respectively. Mass spectrometric data were collected either on a MAT 95 XL Finnigan spectrometer using EI, 70 eV, on a Bruker Dalton micrOTOF-Q using ESI(+/−) or on a Orbitrap XL spectrometer for APCI-measurements.

Complex **1** was synthesized according to the literature protocol.¹

Synthesis of racemic *trans*-2,3-bis(2,2-dimethylethyl)oxaphosphirane (**2**)



Synthesis of **2:** 0.286 g (0.697 mmol) of the oxaphosphirane complex **1** and 0.398 g (0.769 mmol) of DPPE were dissolved in 10 ml of toluene in a 25 ml-Schlenk tube with Teflon valve. The mixture was heated at 60 °C for 14h. After cooling down to room temperature all volatiles were recondensed *in vacuo* (2·10⁻² mbar) in another Schlenk tube with Teflon valve. The obtained solution solely contained the *trans*-isomer of **2** as detected by the ¹H-¹H-NOESY-NMR spectrum; the displayed stereochemistry is just arbitrary chosen but should not display the absolute stereochemistry!

The concentration of **2** was determined by NMR spectroscopy using an aliquot, to which 0.150 ml (0.697 mmol) of cyclohexane (for ¹H-NMR, reference signal at δ 1.40) and 0.075 ml (0.697 mmol) of diphenyl(ethoxy)phosphane (for ³¹P-NMR, reference signal at δ 111.1) was added as references. The NMR measurements were performed using different relaxation times (see figures S1-S6), and the obtained signals were integrated using the reference signal.

¹H-NMR (400 MHz, 298.0 K, toluene): δ /ppm = 0.81 (9H, d, |³J_{P,H}| = 11.79 Hz, P(C(CH₃)₃)), 0.96 (9H, s, C(C(CH₃)₃)), 1.41 (Cyclohexane CH₂, internal reference) 2.74 (1H, s, P(O)CH).

¹ J. Fassbender, G. Schnakenburg, A. Espinosa Ferao and R. Streubel, *Dalton Trans* **2018**, 47, 9347.

³¹P-NMR (400 MHz, 298.0 K, toluene): δ /ppm = -23.7 (m), compound **2**), 111.1 (m), diphenylethoxy-phosphane-internal reference)

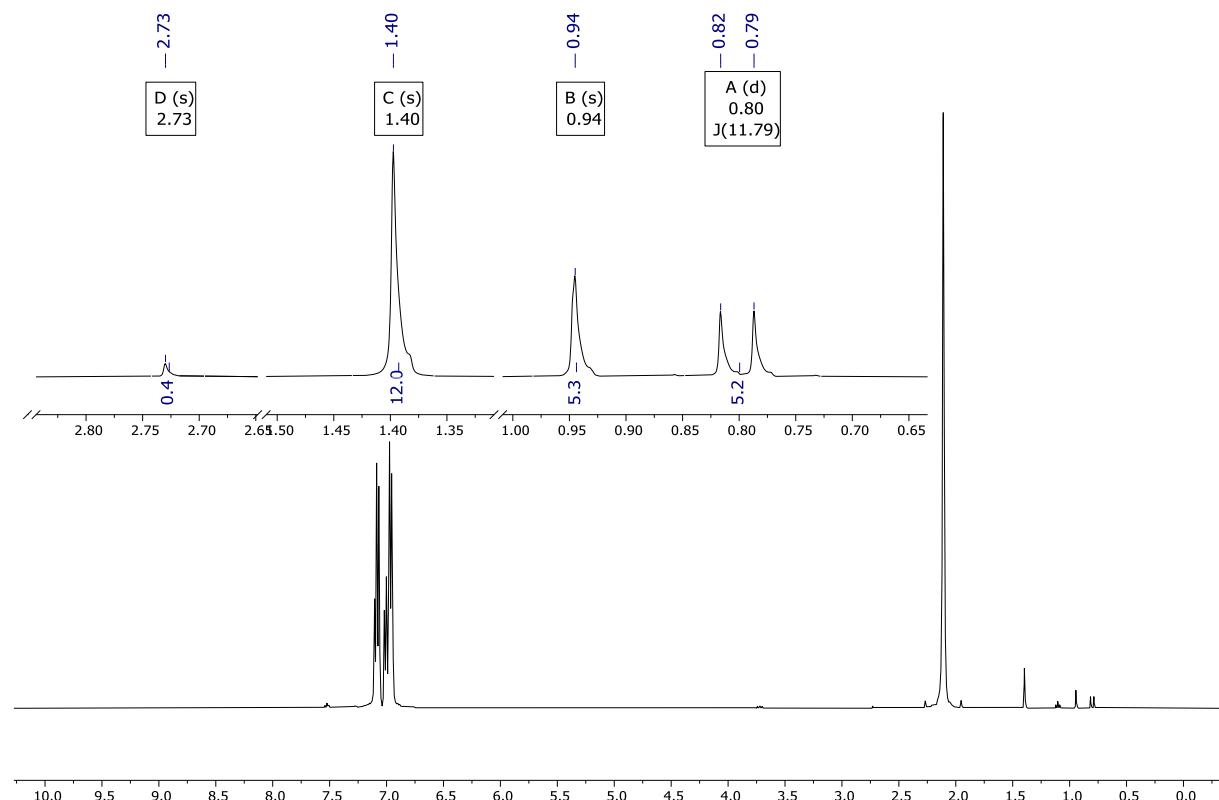


Figure S1: ¹H-NMR spectrum of **2** in toluene (Relaxation Delay = 1.000 s)

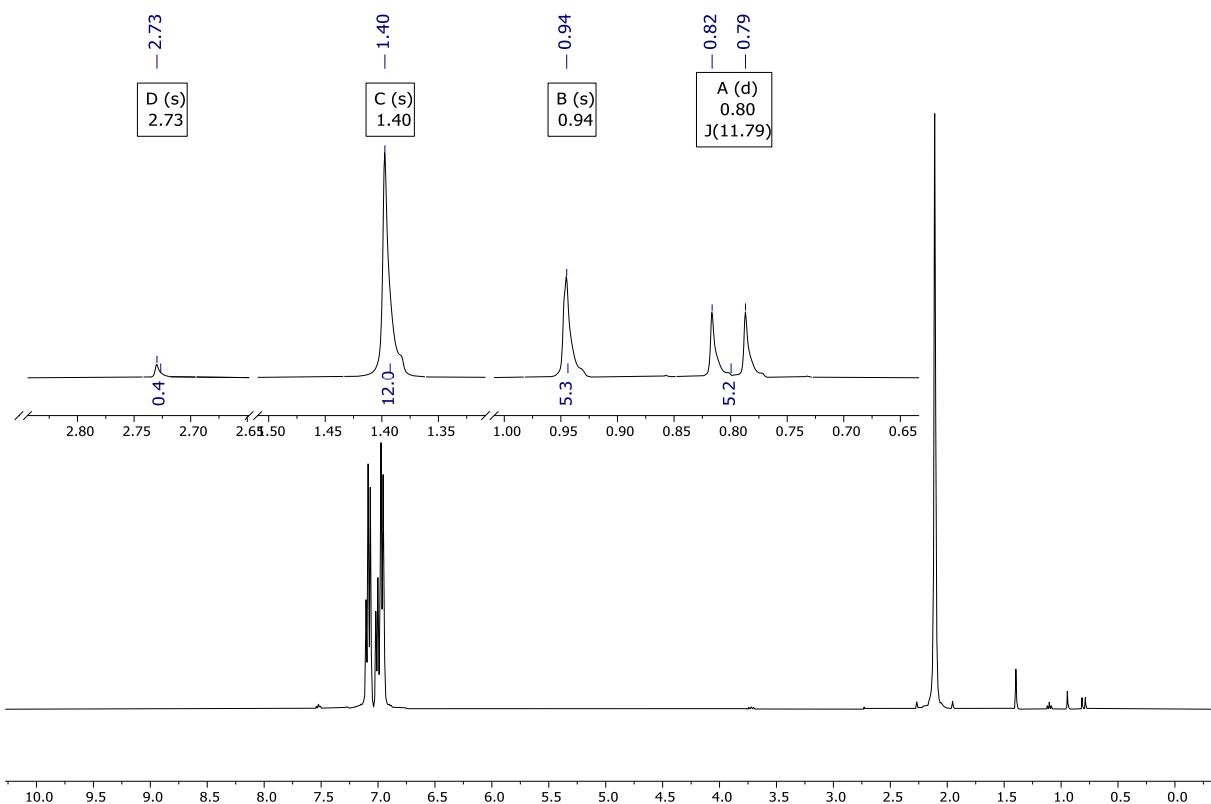


Figure S2: ¹H-NMR spectrum of **2** in toluene (Relaxation Delay = 1.500 s)

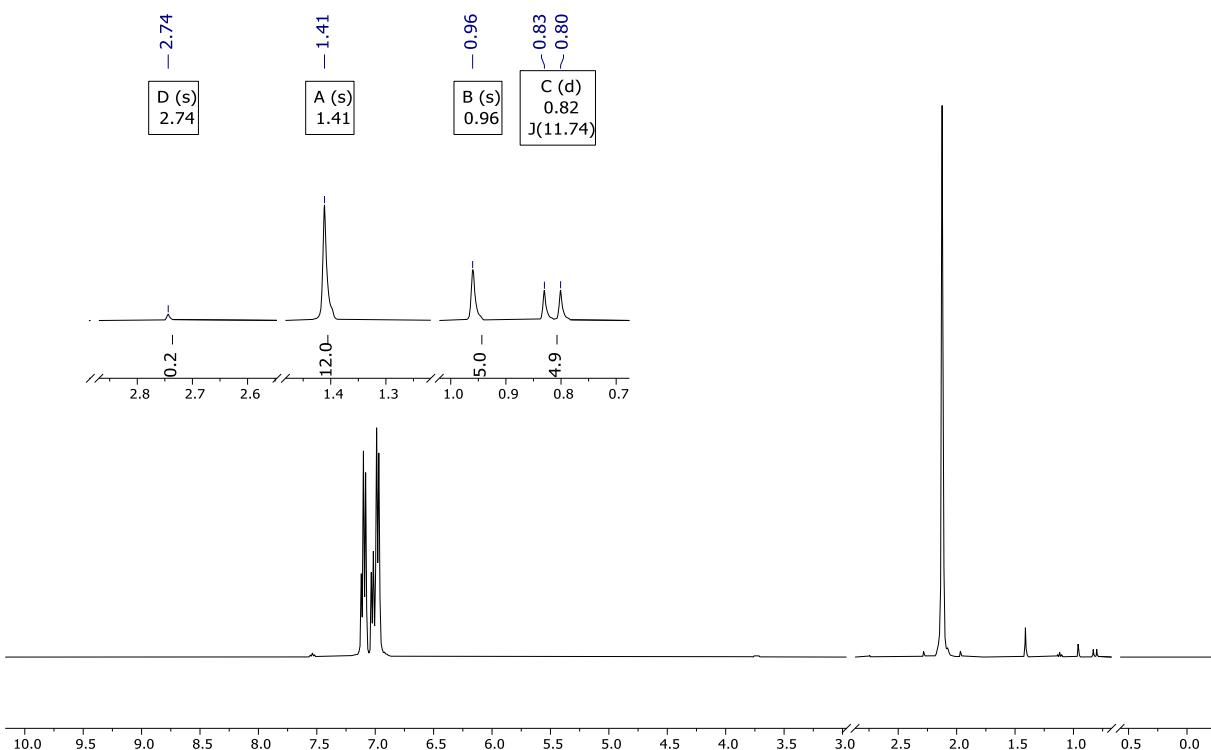


Figure S3: ¹H-NMR spectrum of **2** in toluene (Relaxation Delay = 2.000 s)

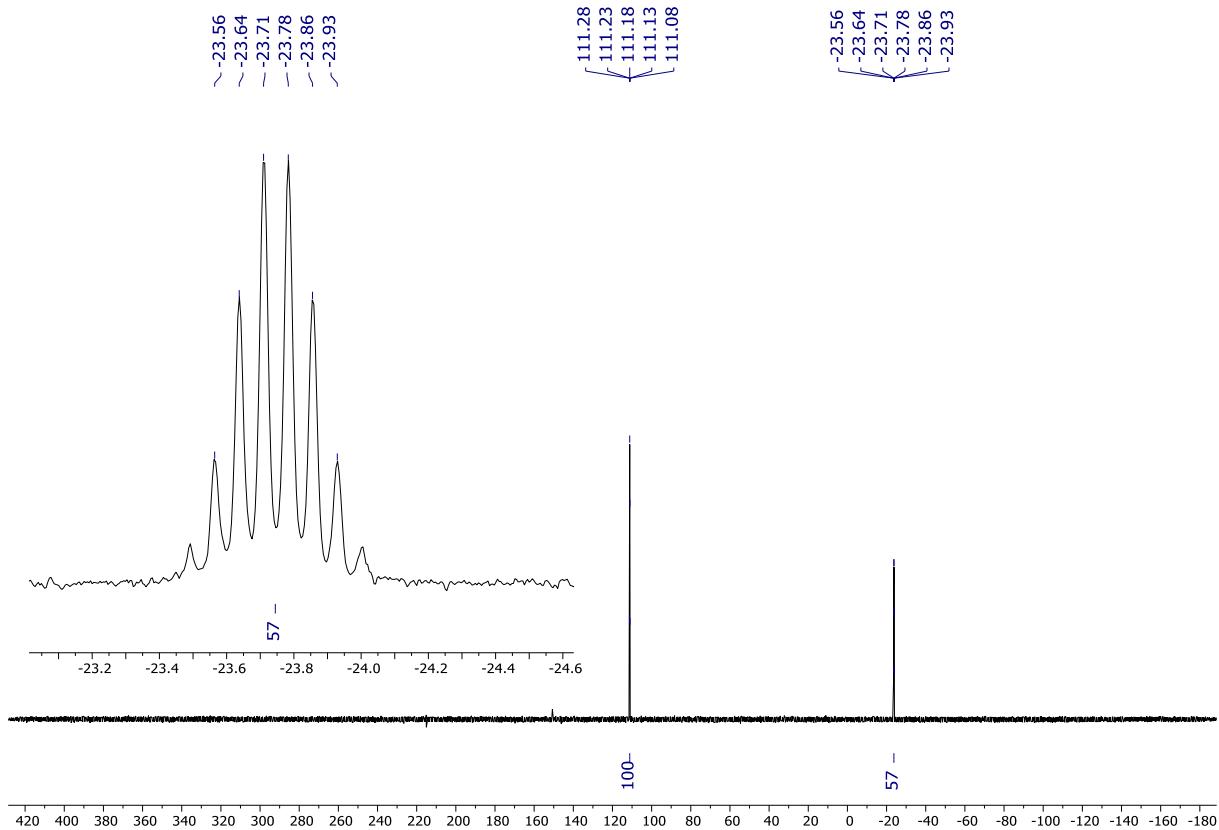


Figure S4: ^{31}P -NMR spectrum of **2** in toluene (Relaxation Delay = 1.000 s)

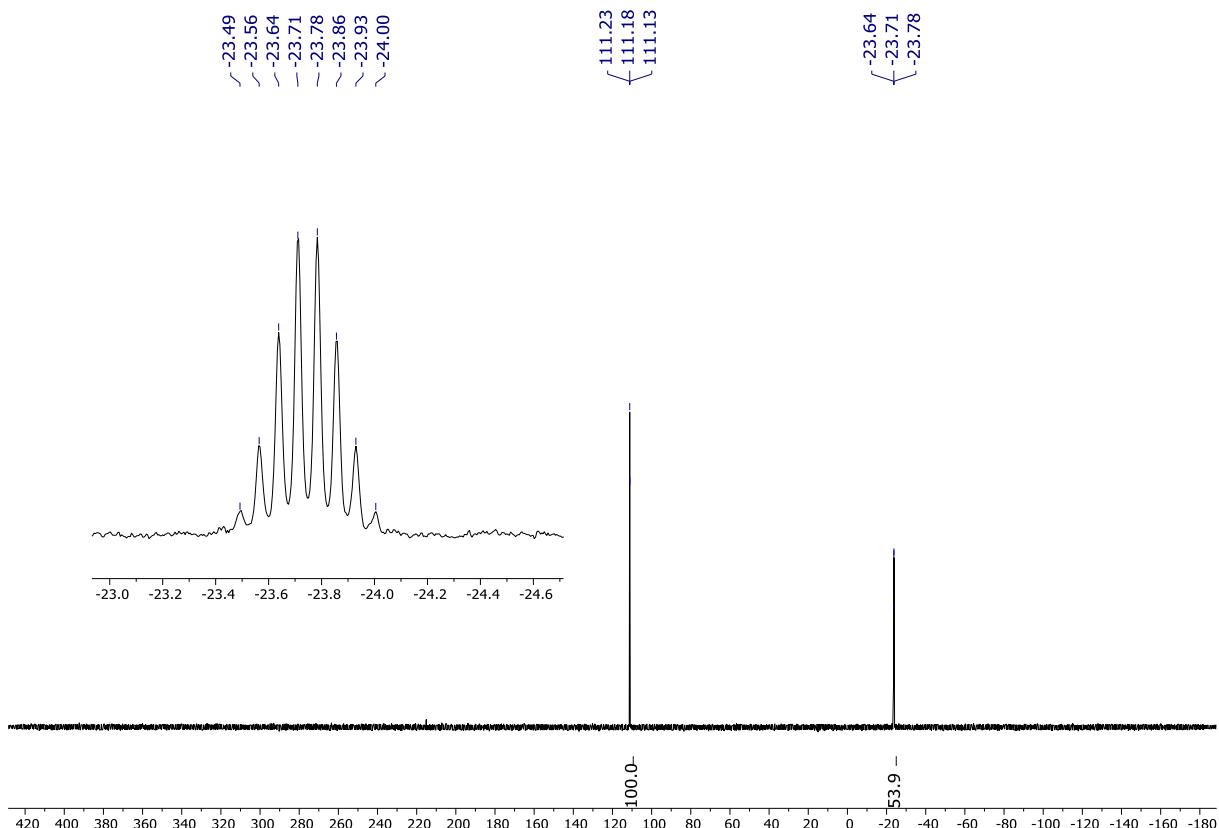


Figure S5: ^{31}P -NMR spectrum of **2** in toluene (Relaxation Delay = 1.500 s)

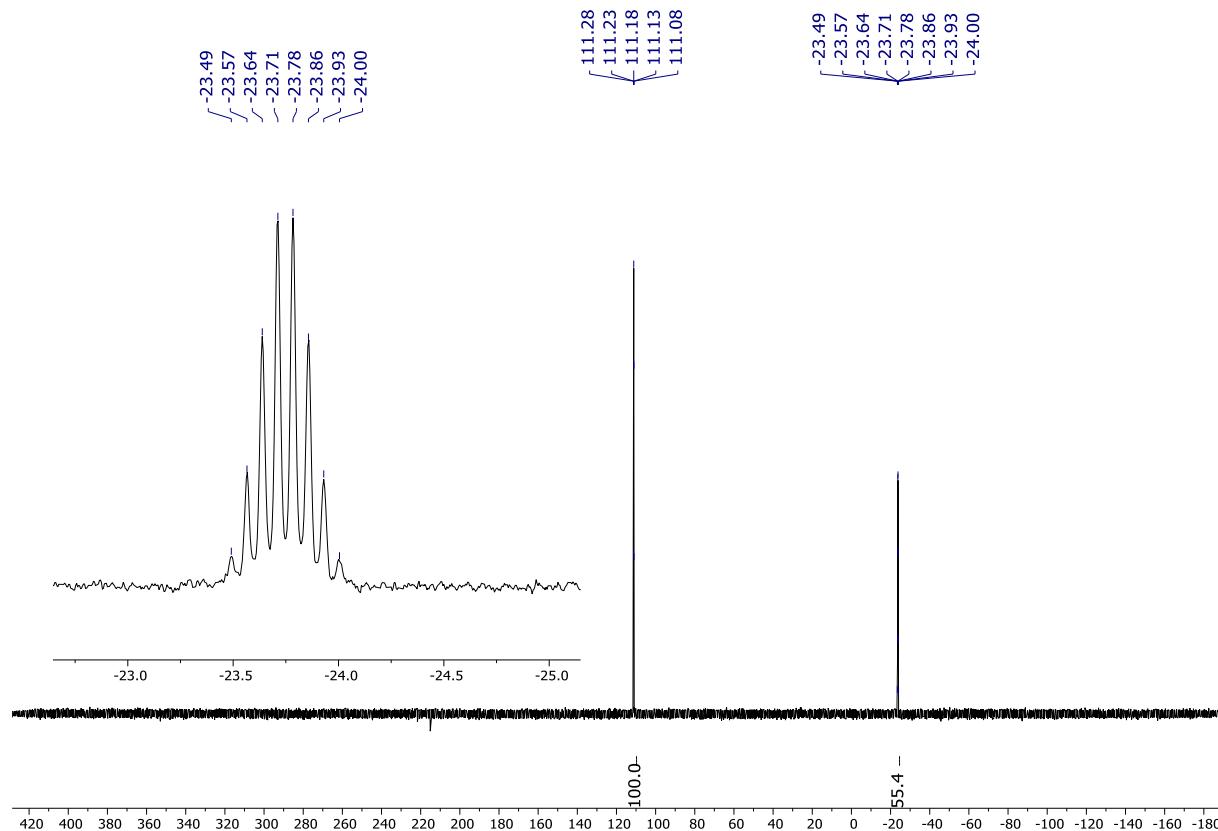


Figure S6: ^{31}P -NMR spectrum of **2** in toluene (Relaxation Delay = 2.000 s)

Content (%) calculated based on the respective NMR spectrum

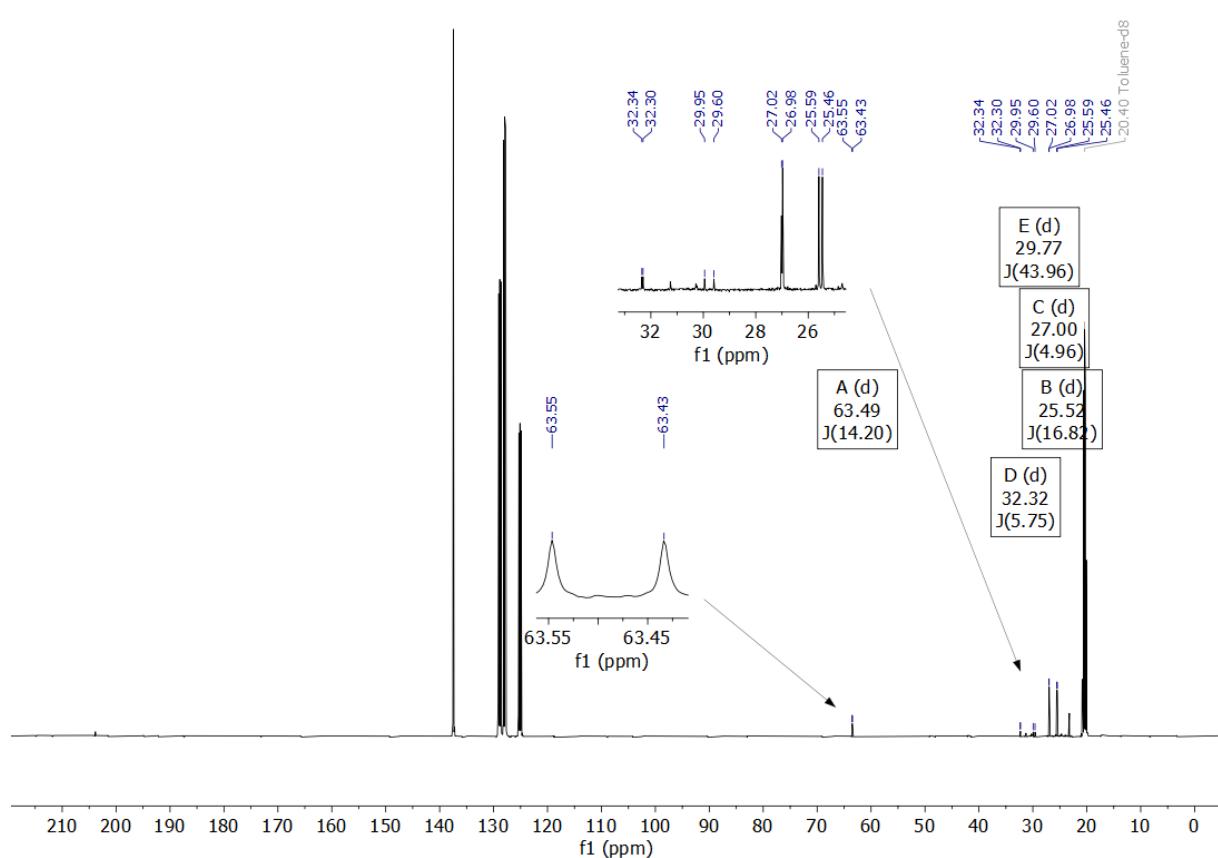
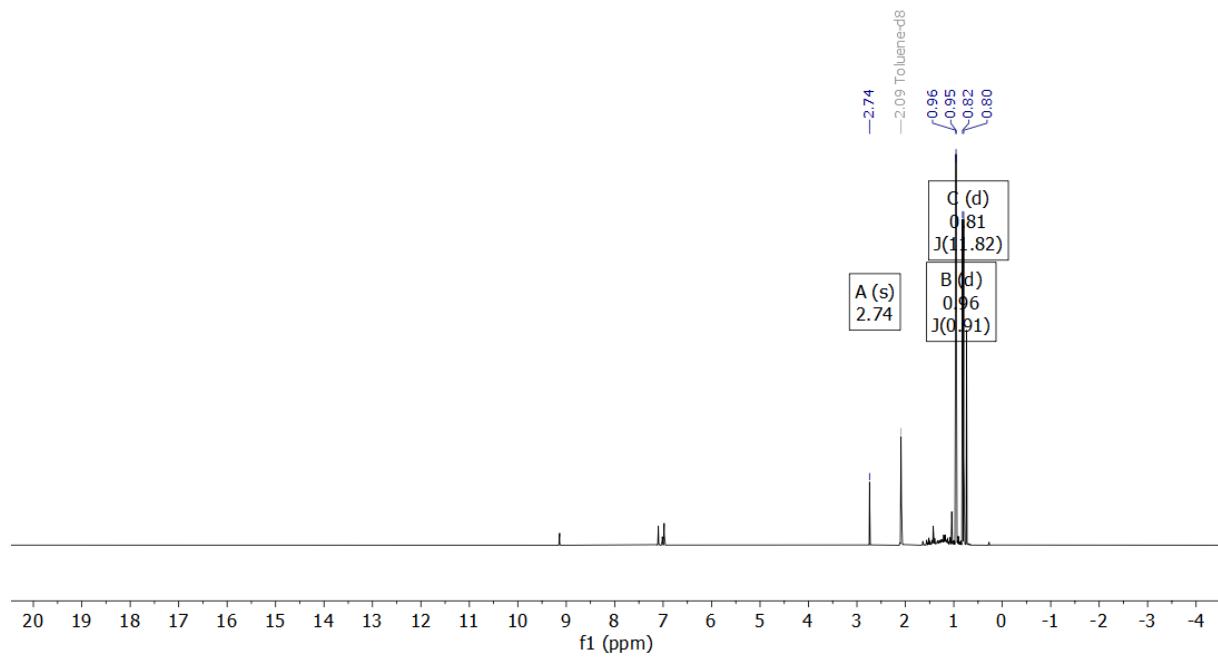
Relaxation Delay in sec.	1.000	1.500	2.000
^1H NMR	55.55	54.55	56.94
^{31}P NMR	57.0	53.9	55.4

MS (LIFDI): m/z (%): 174.1 (2, $[\text{M}]^+$), 92.1 (100, $[\text{toluene}]^+$).

$^1\text{H-NMR}$ (500.1 MHz, 298.0 K, toluene-d₈): δ /ppm = 0.81 (9H, d, $|^3J_{\text{P},\text{H}}|$ = 11.8 Hz, P(C(CH₃)₃)), 0.96 (9H, d, $|^{4/5}J_{\text{P},\text{H}}|$ = 0.9 Hz, C(C(CH₃)₃)), 2.74 (1H, s, P(O)CH).

$^{13}\text{C}\{^1\text{H}\}$ -NMR (125.8 MHz, 298.0 K, toluene-d₈): δ /ppm = 25.5 (d, $|^2J_{\text{P},\text{C}}|$ = 16.8 Hz P(C(CH₃)₃)), 27.0 (d, $|^{3/4}J_{\text{P},\text{C}}|$ = 5.0 Hz, C(C(CH₃)₃)), 29.8 (d, $|^1J_{\text{P},\text{C}}|$ = 44.0 Hz, P(C(CH₃)₃)), 32.3 (d, $|^{2/3}J_{\text{P},\text{C}}|$ = 5.8 Hz, C(C(CH₃)₃)), 63.5 (d, $|^{1/2}J_{\text{P},\text{C}}|$ = 14.2 Hz, P(O)CH).

$^{31}\text{P}\{^1\text{H}\}$ -NMR (202.5 MHz, 298.0 K, toluene-d₈): δ /ppm = -23.7 (s).



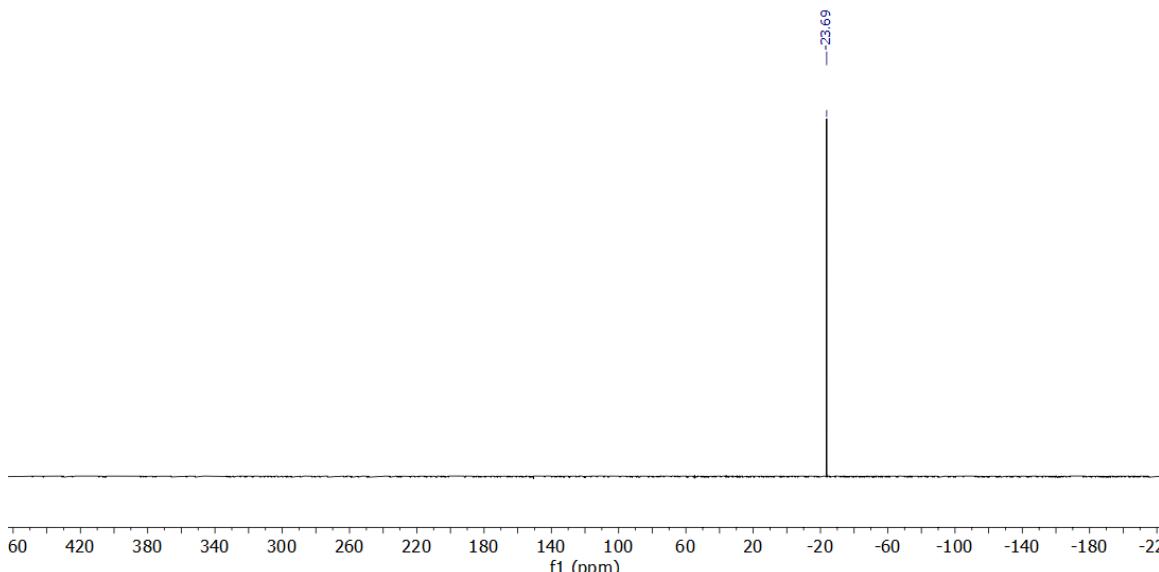
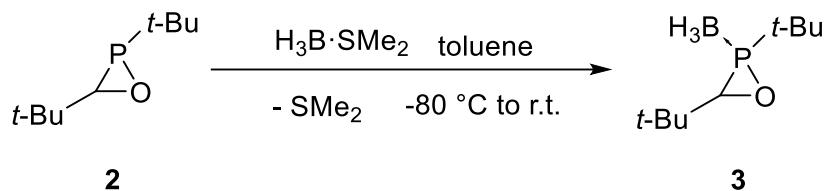


Figure S9: $^{31}\text{P}\{\text{H}\}$ -NMR spectrum of **2** in toluene- d_8 .

Generation of [{2,3-Bis-(2,2-dimethylethyl)oxaphosphirane- κP }borane] (**3**)



Synthesis of **2:** To a precooled solution (-80 °C) of the oxaphosphirane **2** (approx. 1.5 ml), 0.01 ml of $\text{H}_3\text{B}\cdot\text{SMe}_2$ (10M in SMe_2) were added. The solution was warmed to room temperature and was investigated via ^{11}B - and ^{31}P -NMR spectroscopy.

$^{11}\text{B-NMR}$ (96.29 MHz, 299.0 K, toluene): δ /ppm = -41.1 (qd, $^1J_{\text{B},\text{H}} = 105.3$ Hz, $^2J_{\text{P},\text{H}} = 29.6$ Hz).

$^{31}\text{P}\{\text{H}\}$ -NMR (121.51 MHz, 299.0 K, toluene): δ /ppm = -23.5 (**2**, 5 %), 38.3 (br m, **3**, 95 %).

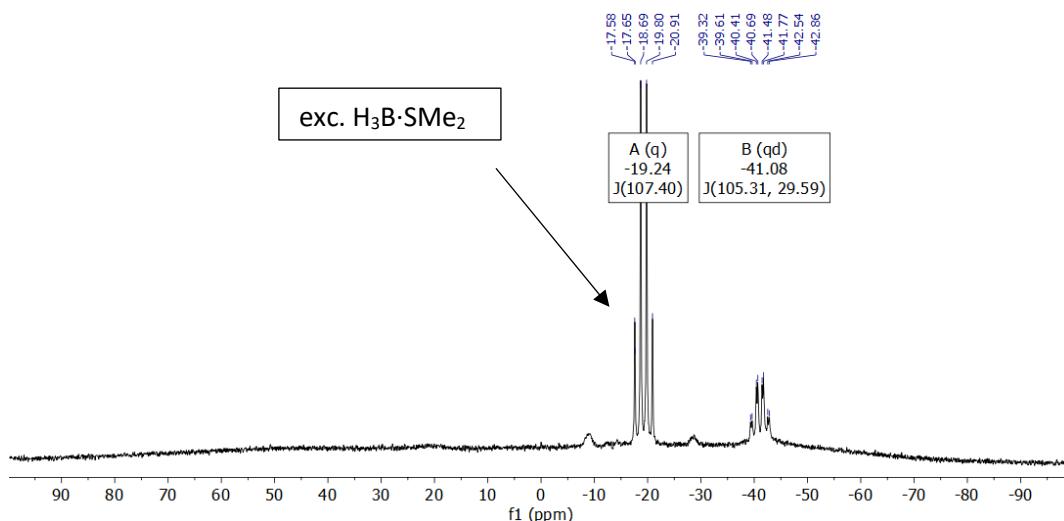


Figure S10: ^{11}B -NMR spectrum of the reaction solution containing **3**.

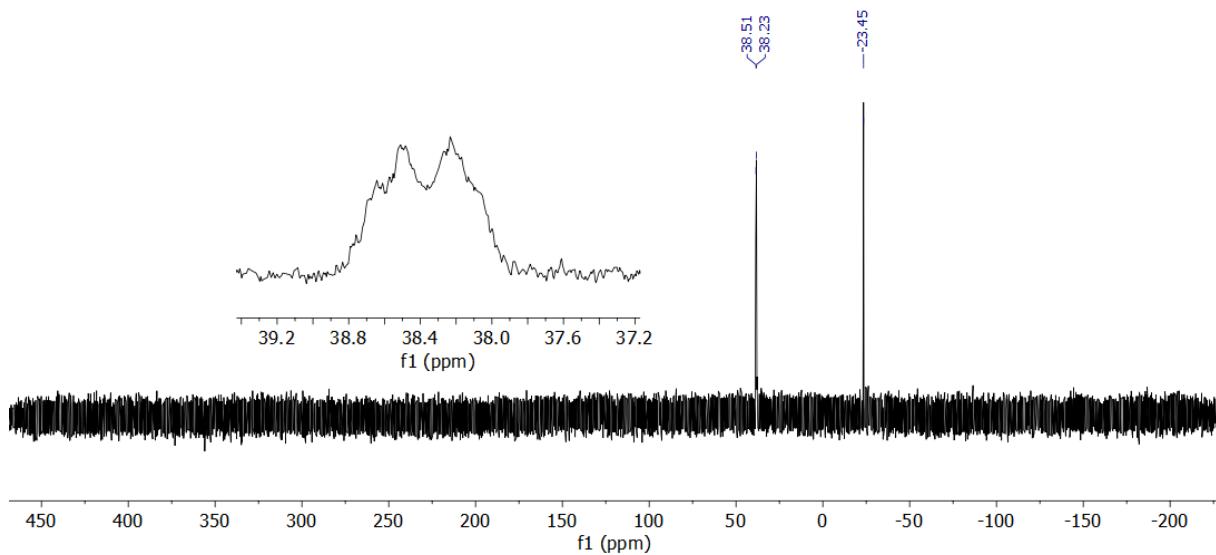
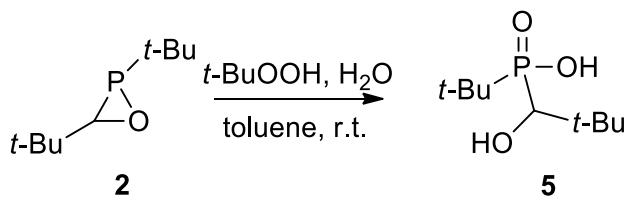


Figure S11: $^{31}\text{P}\{\text{H}\}$ -NMR spectrum of the reaction solution containing **3**.

Synthesis of (1,1-dimethylethyl)(1-hydroxy-2,2-dimethylpropyl)phosphonic acid (**5**)



Synthesis of **5:** To the solution of the oxaphosphirane **2** (approx. 8 ml), 0.1 ml of *t*-BuOOH (5-6 M in decane) were added. After stirring for 1h at room temperature all volatiles were removed *in vacuo* ($2 \cdot 10^{-2}$ mbar). and the remaining solid was washed with *n*-pentane (5.1 ml). After drying *in vacuo* ($2 \cdot 10^{-2}$ mbar) **5** was obtained as a colorless solid.

$^1\text{H-NMR}$ (500.1 MHz, 298.0 K, C_6D_6): δ /ppm = 1.12 (9H, d, $|^3J_{\text{P},\text{H}}|$ = 15.1 Hz, $\text{P}(\text{C}(\text{CH}_3)_3)$), 1.21 (9H, s, $\text{C}(\text{C}(\text{CH}_3)_3)$), 3.55 (1H, d, $|^2J_{\text{P},\text{H}}|$ = 0.6 Hz, $\text{PC}(\text{H})\text{OH}$).

$^{13}\text{C}\{\text{H}\}$ -NMR (125.8 MHz, 298.0 K, C_6D_6): δ /ppm = 24.0 (s, $\text{P}(\text{C}(\text{CH}_3)_3)$), 26.8 (d, $|^3J_{\text{P},\text{C}}|$ = 4.9 Hz, $\text{C}(\text{C}(\text{CH}_3)_3)$), 33.1 (d, $|^1J_{\text{P},\text{C}}|$ = 88.8 Hz, $\text{P}(\text{C}(\text{CH}_3)_3)$), 36.3 (s, $\text{C}(\text{C}(\text{CH}_3)_3)$), 73.1 (d, $|^1J_{\text{P},\text{C}}|$ = 87.6 Hz, $\text{PC}(\text{H})\text{OH}$).

$^{31}\text{P}\{\text{H}\}$ -NMR (202.5 MHz, 298.0 K, C_6D_6): δ /ppm = 62.0 (s).

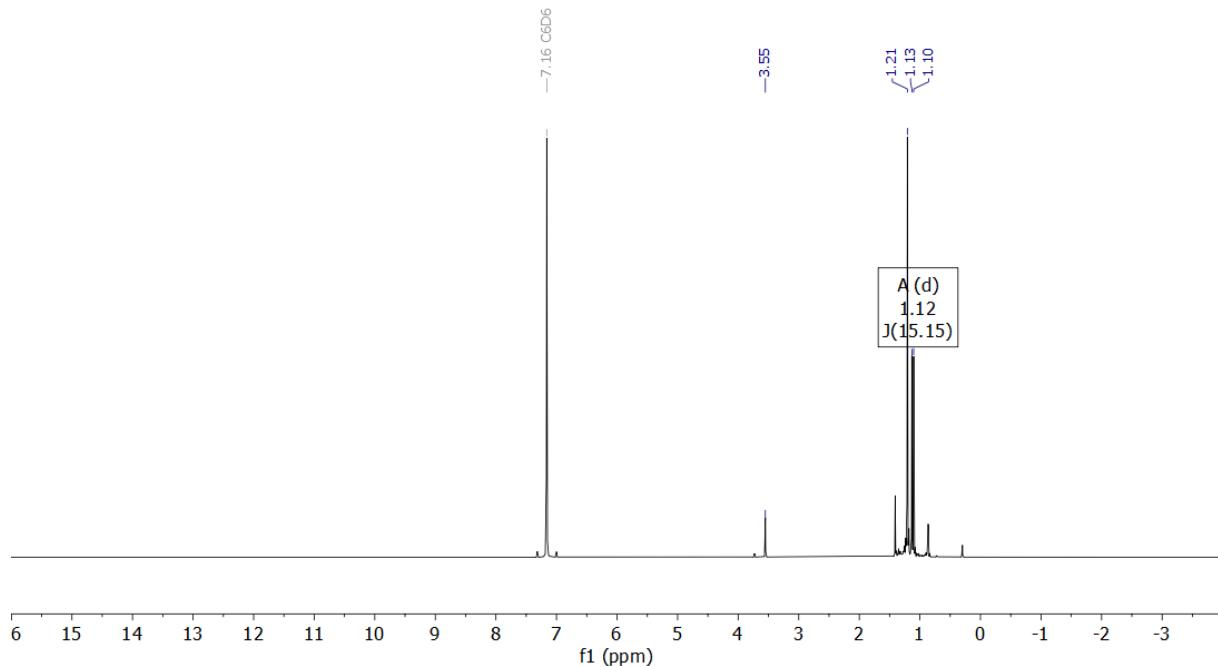


Figure S12: ^1H -NMR spectrum of **5** in C₆D₆.

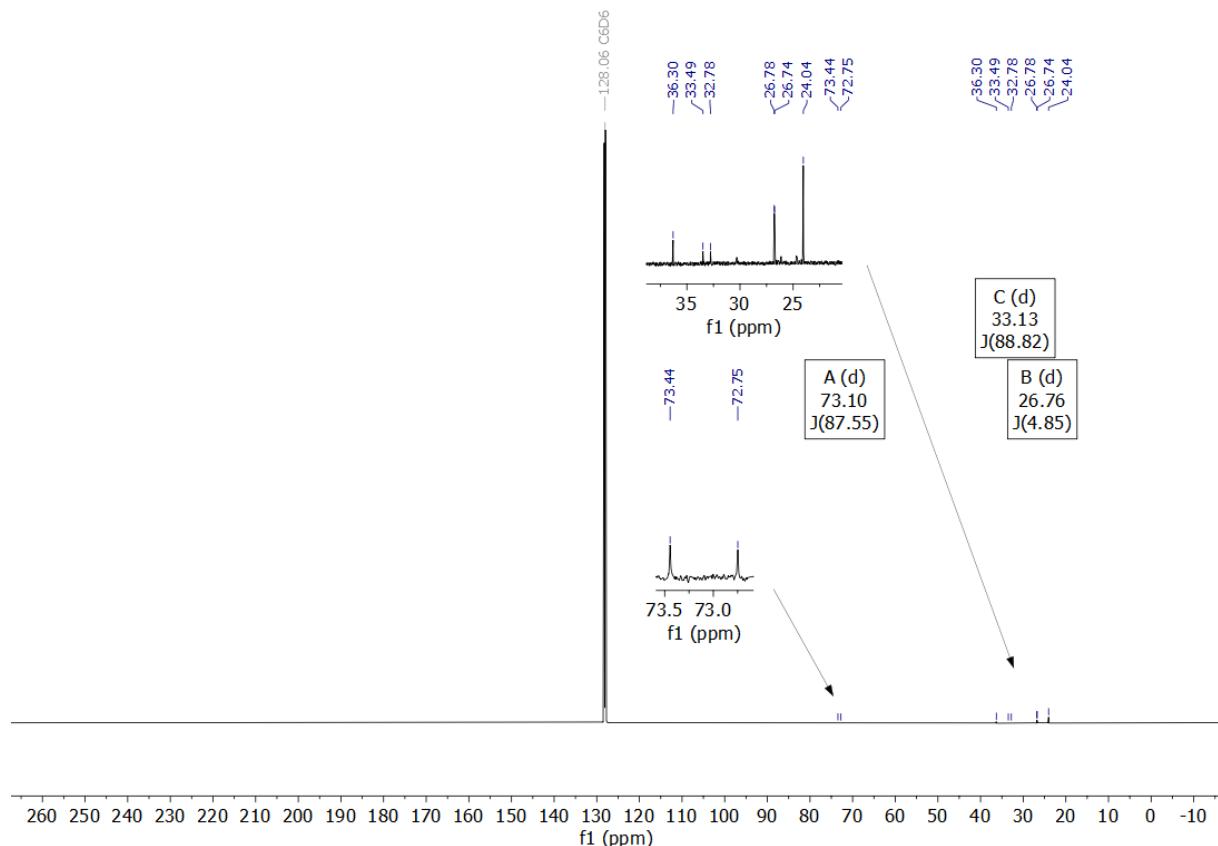


Figure S13: $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of **5** in C₆D₆.

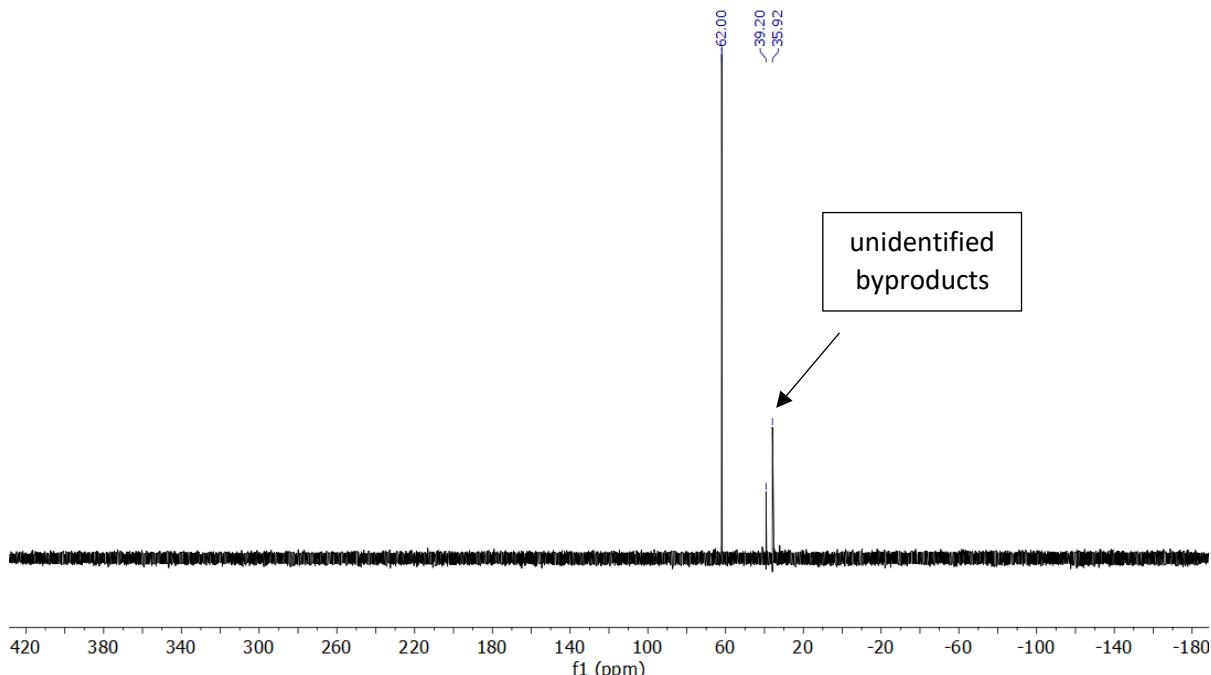
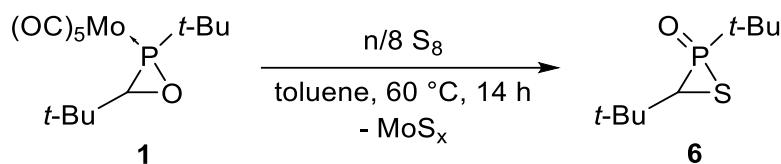
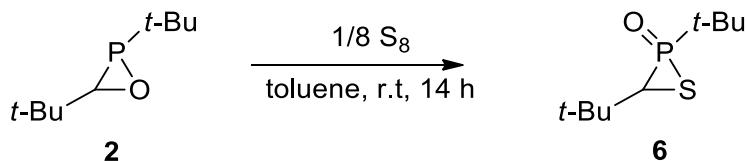


Figure S14: $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of **5** in C_6D_6 .

Synthesis of 2,3-bis(2,2-dimethylethyl)thiaphosphirane-*P*-oxide (**6**)



Method A: 180.3 mg of the oxaphosphirane complex **1** (0.44 mmol, 1.00 eq.) and 49.1 mg of sulfur (1.50 mmol, 3.41 eq.) were dissolved in toluene (5 ml) in a Schlenk tube with Teflon valve. The mixture was stirred at 60 °C for 14 h. After cooling down to room temperature all volatiles were removed *in vacuo* ($2 \cdot 10^{-2}$ mbar) and the remaining solid was washed with *n*-pentane (3·4 ml). After drying *in vacuo* **6** was obtained as a yellow solid.



Method B: To the solution of the oxaphosphirane **2** (approx. 3 ml), 1 ml THF and one spatula tip of sulfur (exc.) were added. After stirring for 24h at room temperature all volatiles were removed *in vacuo* ($2 \cdot 10^{-2}$ mbar). and the remaining solid was washed with *n*-pentane (3·2 ml). After drying *in vacuo* ($2 \cdot 10^{-2}$ mbar) **6** was obtained as a yellow solid.

$^1\text{H-NMR}$ (300.1 MHz, 298.0 K, C_6D_6): δ /ppm = 1.06 (9H, d, $|^3J_{\text{P},\text{H}}|$ = 21.4 Hz, P(C(CH₃)₃)), 1.18 (9H, d, $|^{4/5}J_{\text{P},\text{H}}|$ = 0.9 Hz, C(C(CH₃)₃)), 2.67 (1H, d, $|^{2/3}J_{\text{P},\text{H}}|$ = 1.8 Hz, P(O)CH)).

$^{13}\text{C}\{\text{H}\}$ -NMR (75.5 MHz, 298.0 K, C_6D_6): δ /ppm = 26.2 (d, $|^{3/4}J_{\text{P},\text{C}}|$ = 2.6 Hz, C(C(CH₃)₃)), 28.9 (d, $|^2J_{\text{P},\text{C}}|$ = 5.8 Hz P(C(CH₃)₃)), 33.8 (d, $|^{2/3}J_{\text{P},\text{C}}|$ = 3.6 Hz, C(C(CH₃)₃)), 40.9 (d, $|^1J_{\text{P},\text{C}}|$ = 47.5 Hz, P(C(CH₃)₃)), 42.8 (d, $|^{1/2}J_{\text{P},\text{C}}|$ = 20.0 Hz, P(O)CH)).

$^{31}\text{P}\{\text{H}\}$ -NMR (121.5 MHz, 298.0 K, C_6D_6): δ /ppm = 21.5 (s).

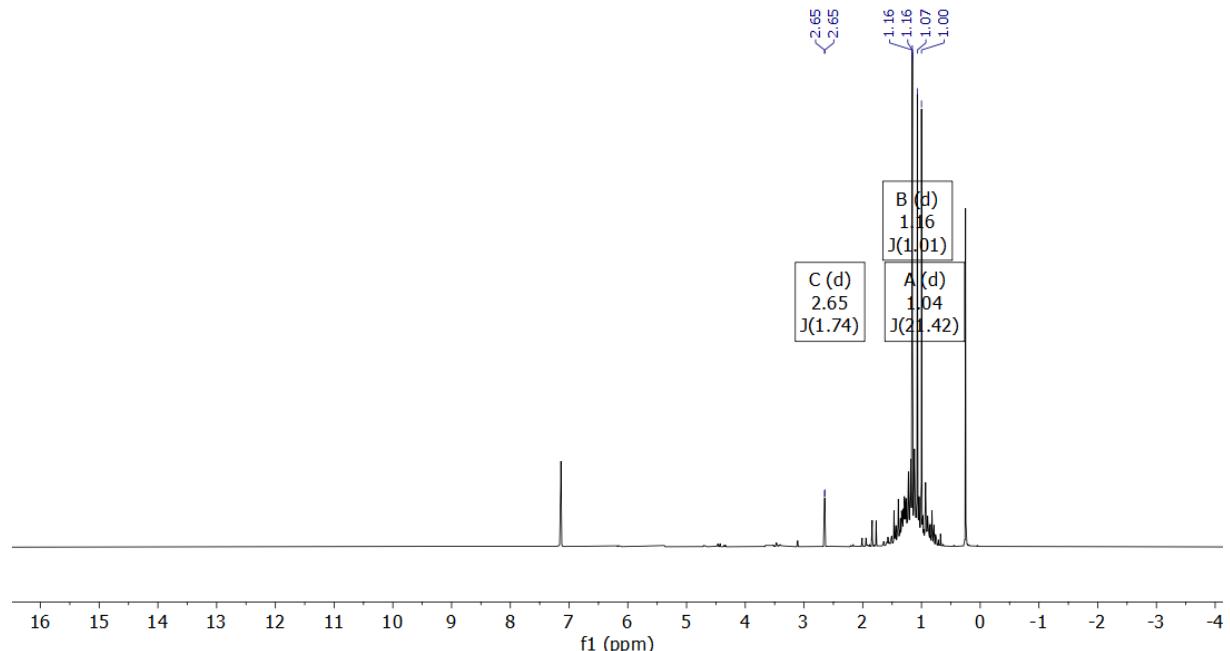
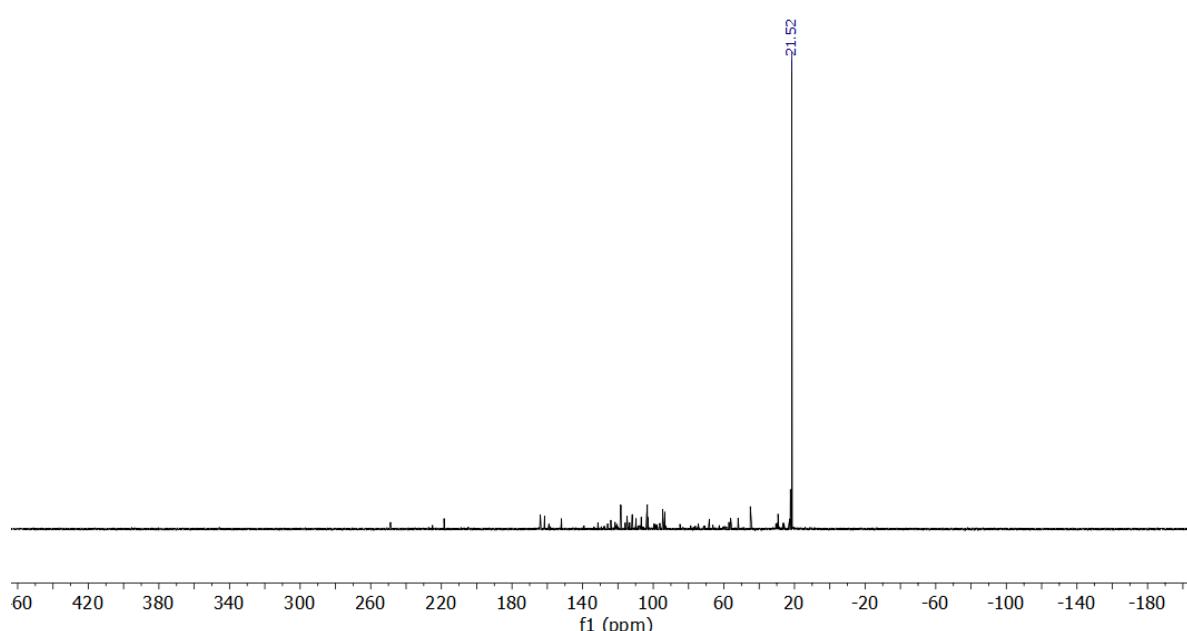
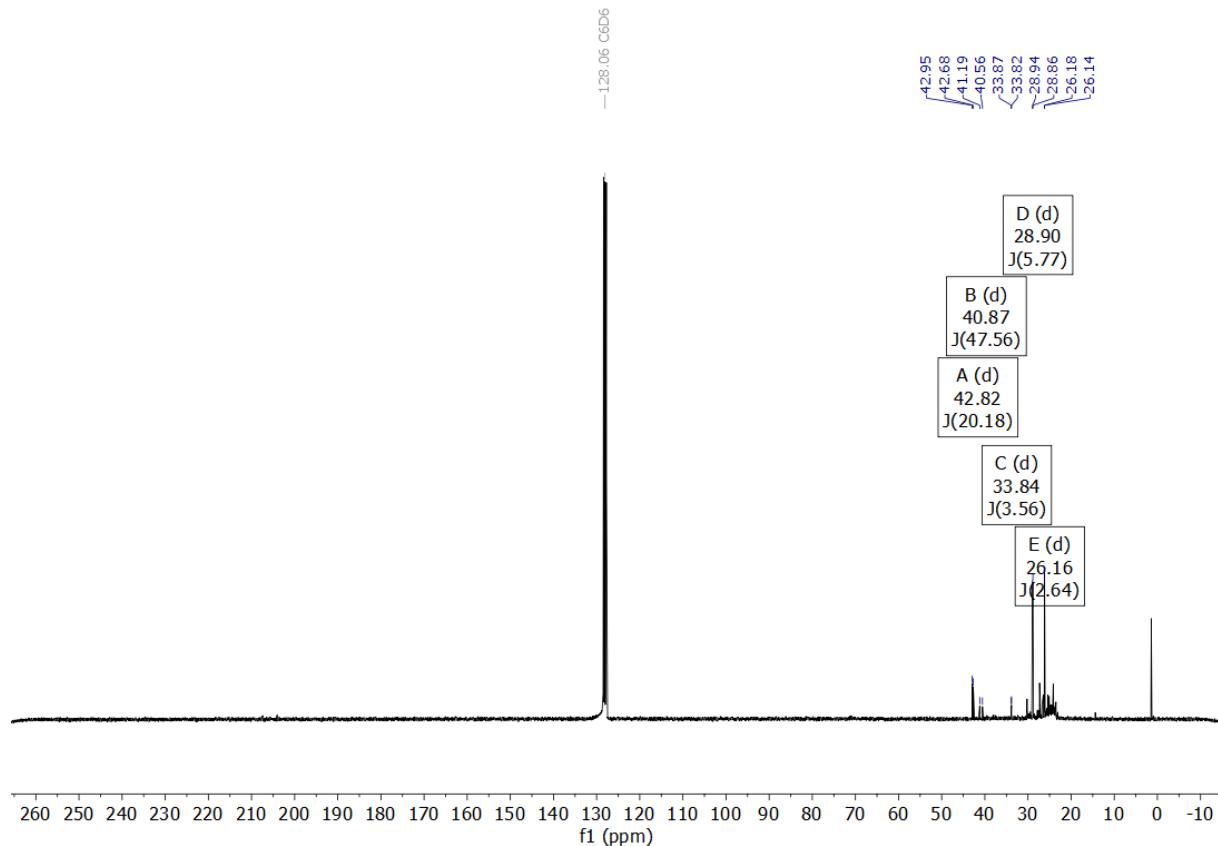
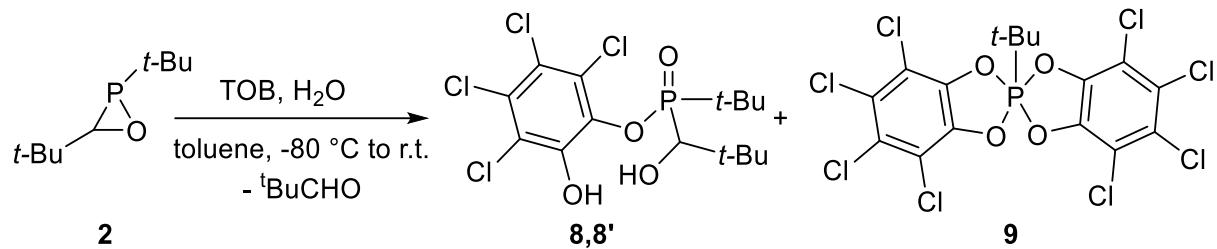


Figure S15: $^1\text{H-NMR}$ spectrum of **6** in C_6D_6 .



Synthesis of a mixture of (1,1-dimethylethyl)-(1-hydroxy-2,2-dimethylpropyl)-(2,3,4,5-tetrachloro-6-hydroxyphenoxy)phosphaneoxide (8,8') and 4,4',5,5',6,6',7,7'-octachloro-2-(1,1-dimethylethyl)-2,2'-spirobis[1,3,2-benzodioxaphosphole] (9)



VT-NMR investigation

6.5 ml of a cooled (-80 °C) oxaphosphirane solution **2** were transferred with a transfer cannula to a -80 °C precooled Schlenk tube containing 49.2 mg of TOB (0.200 mmol). After stirring for 1 minute 0.5 ml of the solution were transferred to a -90 °C precooled Young NMR tube and VT-NMR measurements were conducted from -70 °C to 25 °C.

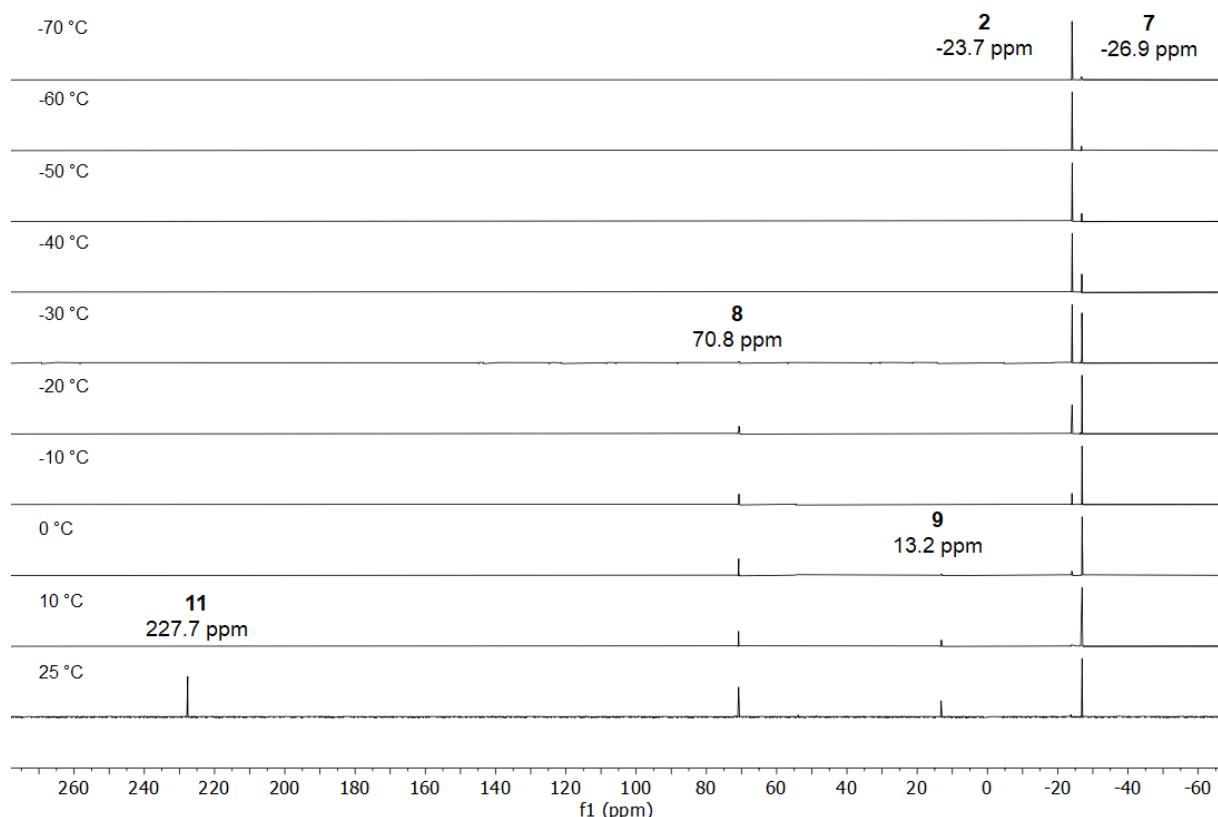


Figure 18: $^{31}\text{P}\{\text{H}\}$ -VT-NMR investigation from -70 °C to 25 °C in toluene.

Reaction solution

The reaction solution was warmed up to -10 °C and the solvent was removed *in vacuo* ($2 \cdot 10^{-2}$ mbar) at -10 °C. The obtained bright yellow residue was washed with *n*-pentane (3·1.5 ml) at -30 °C and dried *in*

vacuo ($2 \cdot 10^{-2}$ mbar). A mixture of **8,8'** and **9** was obtained in a 1:0.05:0.25 ratio as a bright yellow solid. The following data refer to the mixture.

9: MS (EI) m/z (%): calculated: 575.77 found: 575.7 (10, $[M]^{+•}$), 519.6 (14, $[M-C_4H_8]^{+•}$), 293.8 (10, $[M-C_6Cl_4O]^{+•}$), 276.8 (60, $[M-C_6Cl_4HO_2]^{+•}$), 57.1 (100, $[M-C_{12}Cl_8O_4P]^{+•}$).

1H -NMR (300.13 MHz, 298.0 K, CDCl₃): δ /ppm = 1.27 (d, $|^3J_{P,H}|$ = 22.5 Hz, 9H, P(C(CH₃)₃)).

$^{13}C\{^1H\}$ -NMR (75.48 MHz, 298.0 K, CDCl₃): δ /ppm = 26.9 (d, $|^2J_{P,C}|$ = 1.9 Hz, C(C(CH₃)₃), 43.1 (d, $|^1J_{P,C}|$ = 144.6 Hz, C(C(CH₃)₃), 126.2 (s, Ar^{OP}), 128.4 (s, Ar^{Cl}), 129.2 (s, Ar^{Cl}), 141.0 (d, $|^xJ_{P,C}|$ = 4.9 Hz, Ar^{Cl}), 144.6 (d, $|^xJ_{P,C}|$ = 2.3 Hz, Ar^{Cl}).

^{31}P -NMR (121.51 MHz, 298.0 K, CDCl₃): δ /ppm = 14.0 (dez, $|^3J_{P,H}|$ = 22.5 Hz).

8: **1H -NMR** (300.13 MHz, 298.0 K, CDCl₃): δ /ppm = 1.11 (s, 9H, C(C(CH₃)₃), 1.35 (d, $|^3J_{P,H}|$ = 16.4 Hz, 9H, P(C(CH₃)₃)), 3.97 (d, $^2J_{P,H}$ = 5.4 Hz, 1H, PC(H)OH), 11.3 (s, 1H, Ar-OH).

$^{13}C\{^1H\}$ -NMR (75.48 MHz, 298.0 K, CDCl₃): δ /ppm = 24.4 (s, C(CH₃)₃), 26.5 (d, $|^2J_{P,C}|$ = 5.0 Hz, P(C(CH₃)₃)), 35.5 (d, $|^1J_{P,C}|$ = 78.1 Hz, P(C(CH₃)₃)), 36.4 (s, C(C(CH₃)₃)), 74.5 (d, $|^1J_{P,C}|$ = 81.2 Hz, PC(H)OH), 126.2 (s, Ar^{OP}/Ar^{OH}), 128.4 (s, Ar^{Cl}), 129.2 (s, Ar^{Cl}), 141.0 (d, $|^xJ_{P,C}|$ = 4.9 Hz, Ar^{Cl}), 144.6 (d, $|^xJ_{P,C}|$ = 2.3 Hz, Ar^{Cl}).

^{31}P -NMR (121.51 MHz, 298.0 K, CDCl₃): δ /ppm = 71.8 ppm (dez, $|^3J_{P,H}|$ = 16.4 Hz, $|^2J_{P,H}|$ = 5.4 Hz).

8': **$^{31}P\{^1H\}$ -NMR** (121.51 MHz, 298.0 K, CDCl₃): δ /ppm = 72.0 ppm.

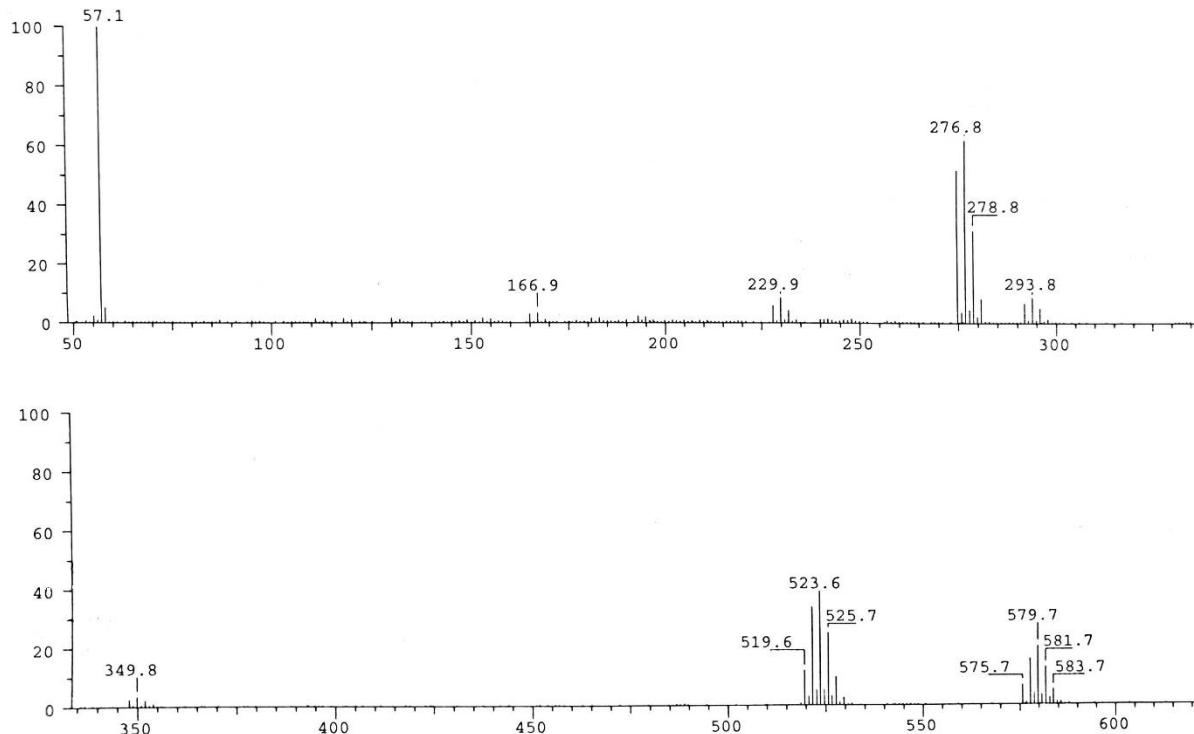


Figure S19: MS spectrum (EI) of the product mixture of **8,8'** and **9**.

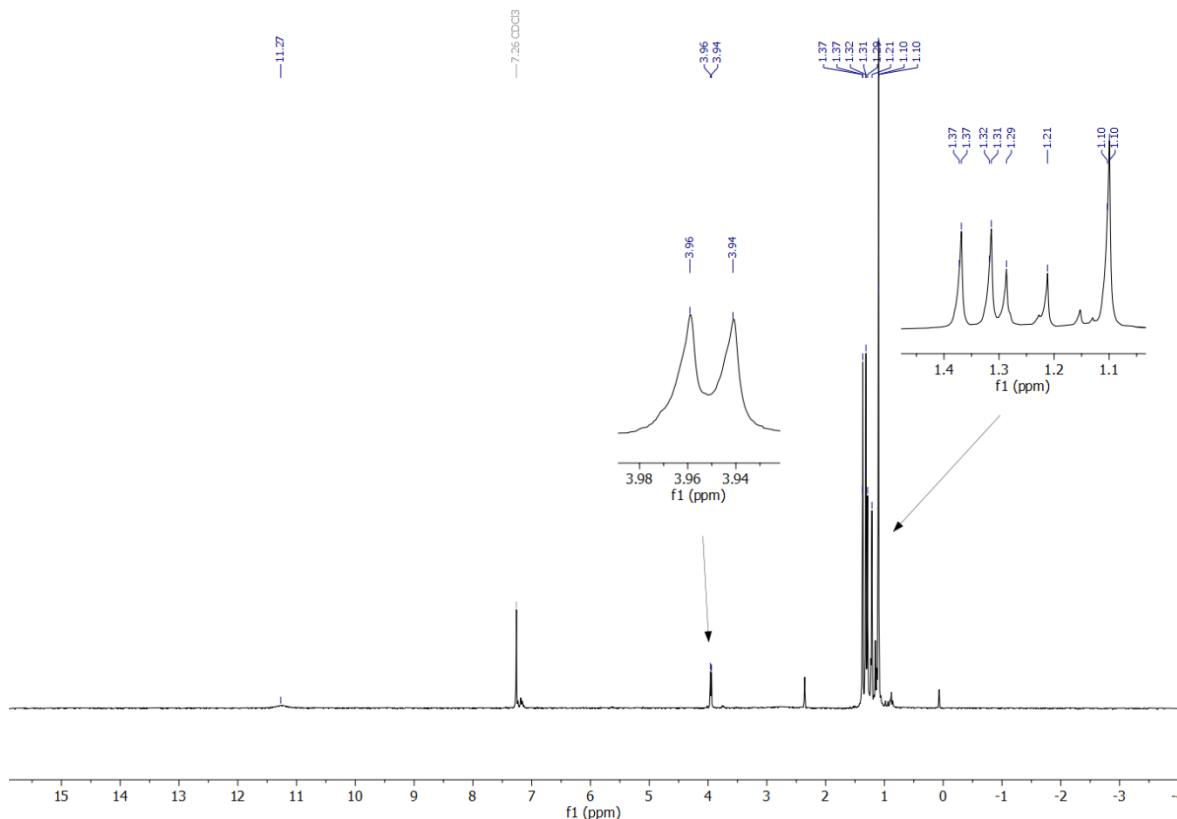


Figure S20: ^1H -NMR spectrum of the product mixture **8,8'** and **9** in CDCl_3 .

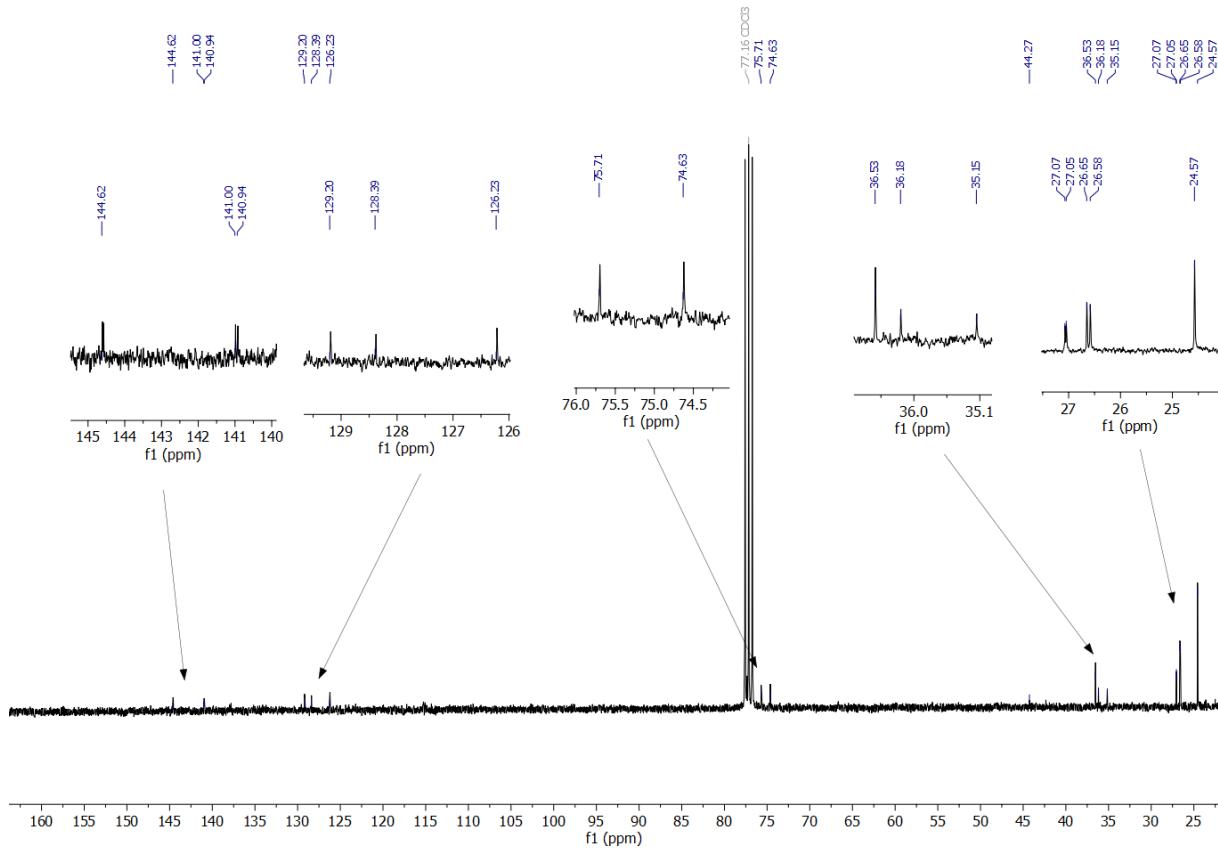


Figure S21: $^{13}\text{C}\{\text{H}\}$ -NMR spectrum of the product mixture **8,8'** and **9** in CDCl_3 .

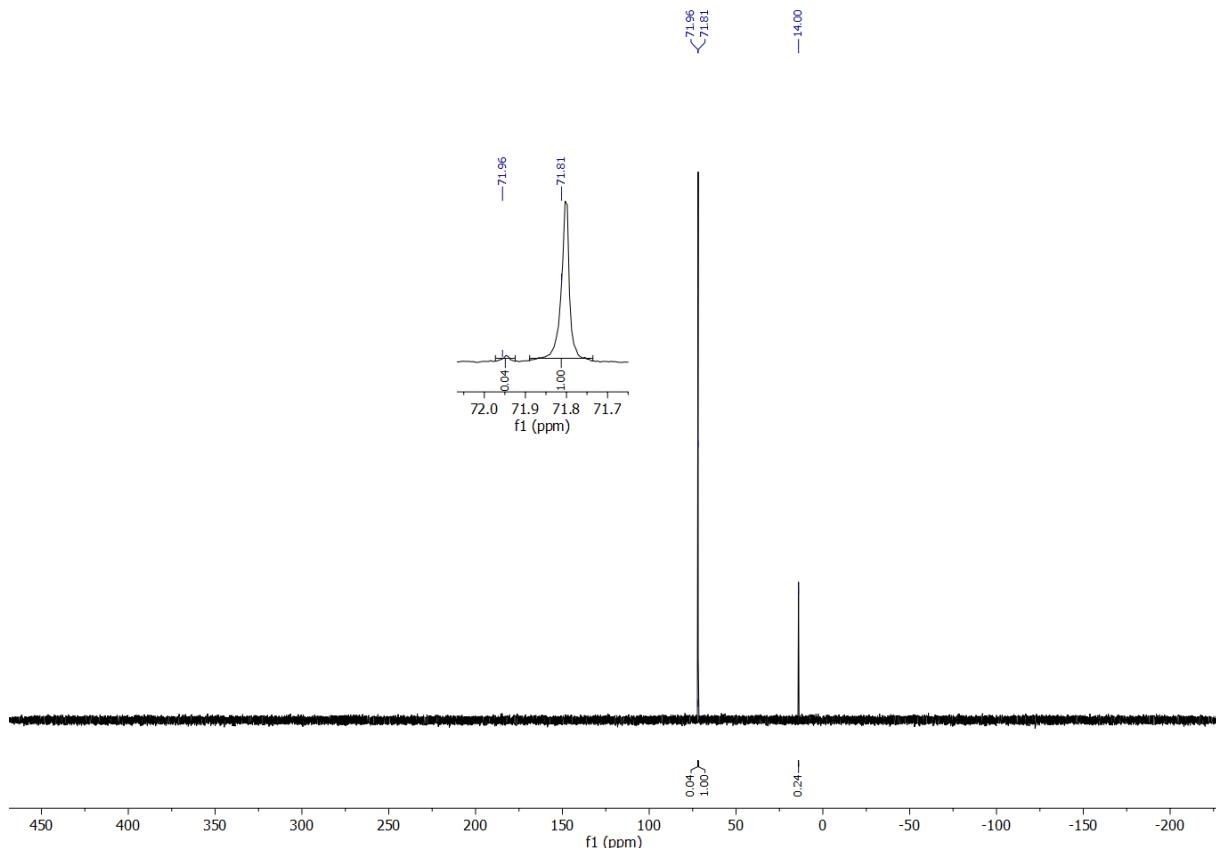
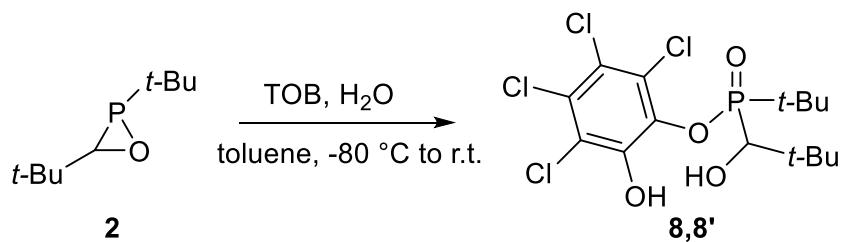


Figure S22: $^{31}\text{P}\{\text{H}\}$ -NMR spectrum of the product mixture **8,8'** and **9** in CDCl_3 .

Synthesis of (1,1-dimethylethyl)-(1-hydroxy-2,2-dimethylpropyl)-(2,3,4,5-tetrachloro-6-hydroxyphenoxy)phosphaneoxide (8,8')



To a solution of 23.2 mg TOB in 1 ml toluene, 1 ml of the oxaphosphirane solution **2** was added at -80 °C. The solution was slowly warmed up to room temperature over five hours and stirred at room temperature for 2 days. The solvent was removed *in vacuo* ($2 \cdot 10^{-2}$ mbar) and a red solid was obtained. The solid was washed with *n*-pentane (4x1 ml) and Et₂O (1x1 ml) at -30 °C and dried *in vacuo* ($2 \cdot 10^{-2}$ mbar). The product was obtained as a beige solid and as a mixture of both diastereomers **8,8'** (1 : 0.02).

8: MS (APCI) m/z (%): calculated.: 435.99 found: 437.000 (10, [M+H]⁺), 350.927 (76, [M-C₅H₉O]⁺).

¹H-NMR (500.04 MHz, 298.0 K, CD₂Cl₂): δ /ppm = 1.01 (s, 9H, C(C(CH₃)₃)), 1.25 (d, |³J_{P,H}|= 16.4 Hz, 9H, P(C(CH₃)₃)), 2.45 (s, 1H, OH), 3.98 (s, 1H, PC(H)OH), 11.3 (s, 1H, Ar-OH).

¹³C{¹H}-NMR (125.75 MHz, 298.0 K, CD₂Cl₂): δ /ppm = 24.5 (s, C(C(CH₃)₃)), 26.6 (d, |²J_{P,C}|= 5.0 Hz, P(C(CH₃)₃)), 35.8 (d, |¹J_{P,C}|= 78.1 Hz, P(C(CH₃)₃)), 36.6 (s, C(C(CH₃)₃)), 75.3 (d, |¹J_{P,C}|= 81.8 Hz, PC(H)OH), 123.3 (s, Ar^{OP}/Ar^{OH}), 123.4 (s, Ar^{OP}/Ar^{OH}), 129.5 (s, Ar^{Cl}), 138.5 (d, |^xJ_{P,C}|= 12.4 Hz, Ar^{Cl}), 145.0 (d, |^xJ_{P,C}|= 2.5 Hz, Ar^{Cl}).

³¹P-NMR (202.44 MHz, 298.0 K, CD₂Cl₂): δ /ppm = 72.1 ppm (dez, |³J_{P,H}|= 16.4 Hz).

8': ³¹P{¹H}-NMR (202.44 MHz, 298.0 K, CD₂Cl₂): δ /ppm = 72.3 ppm.

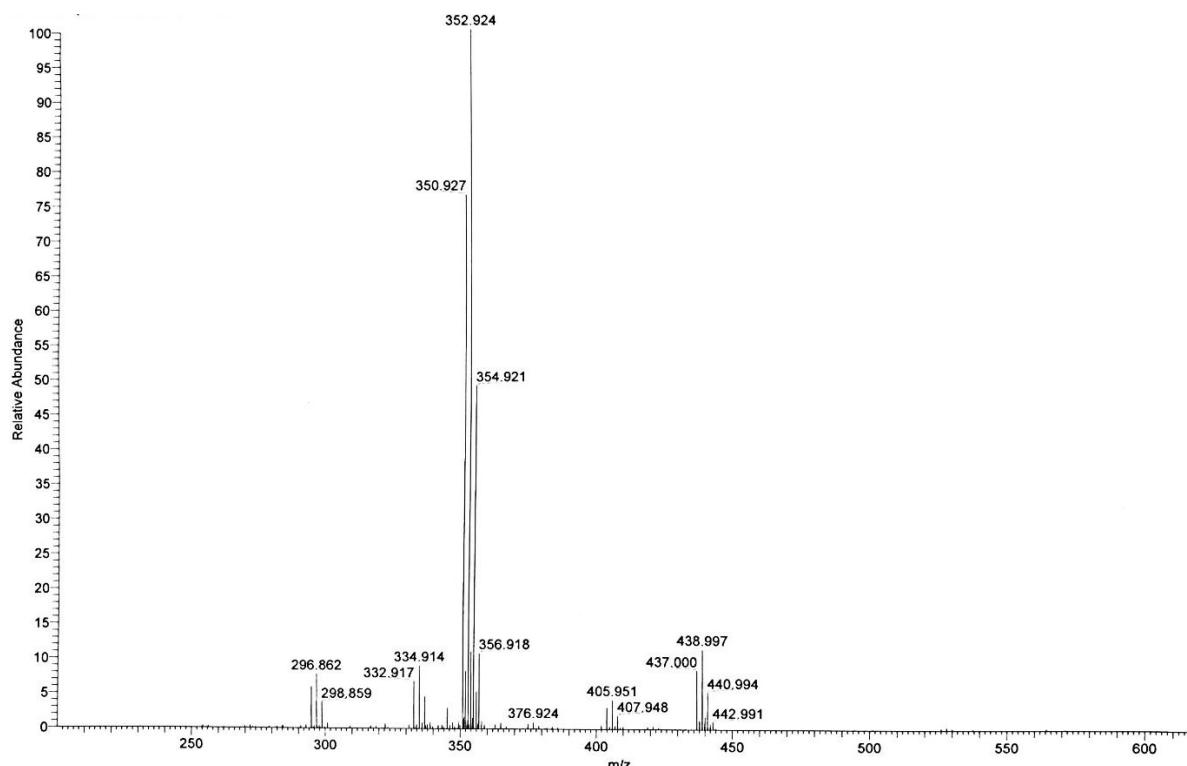
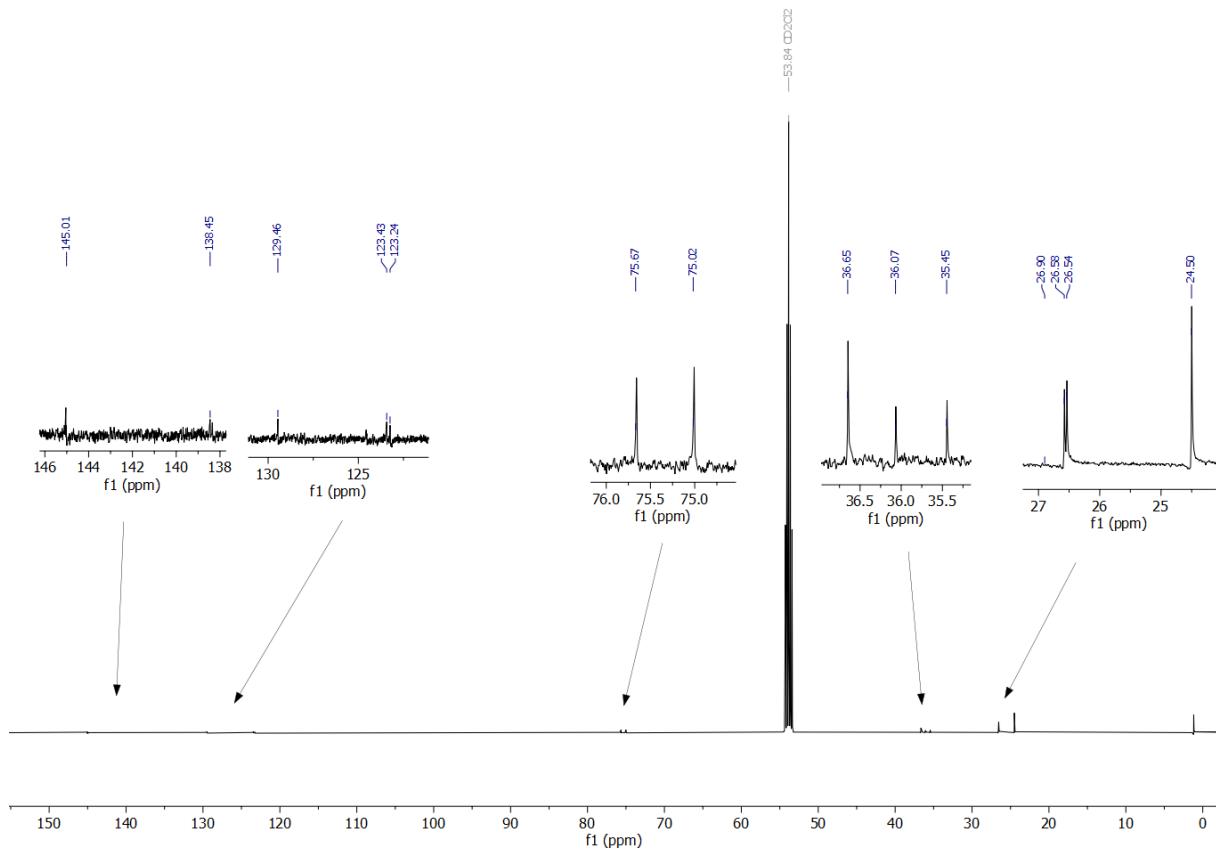
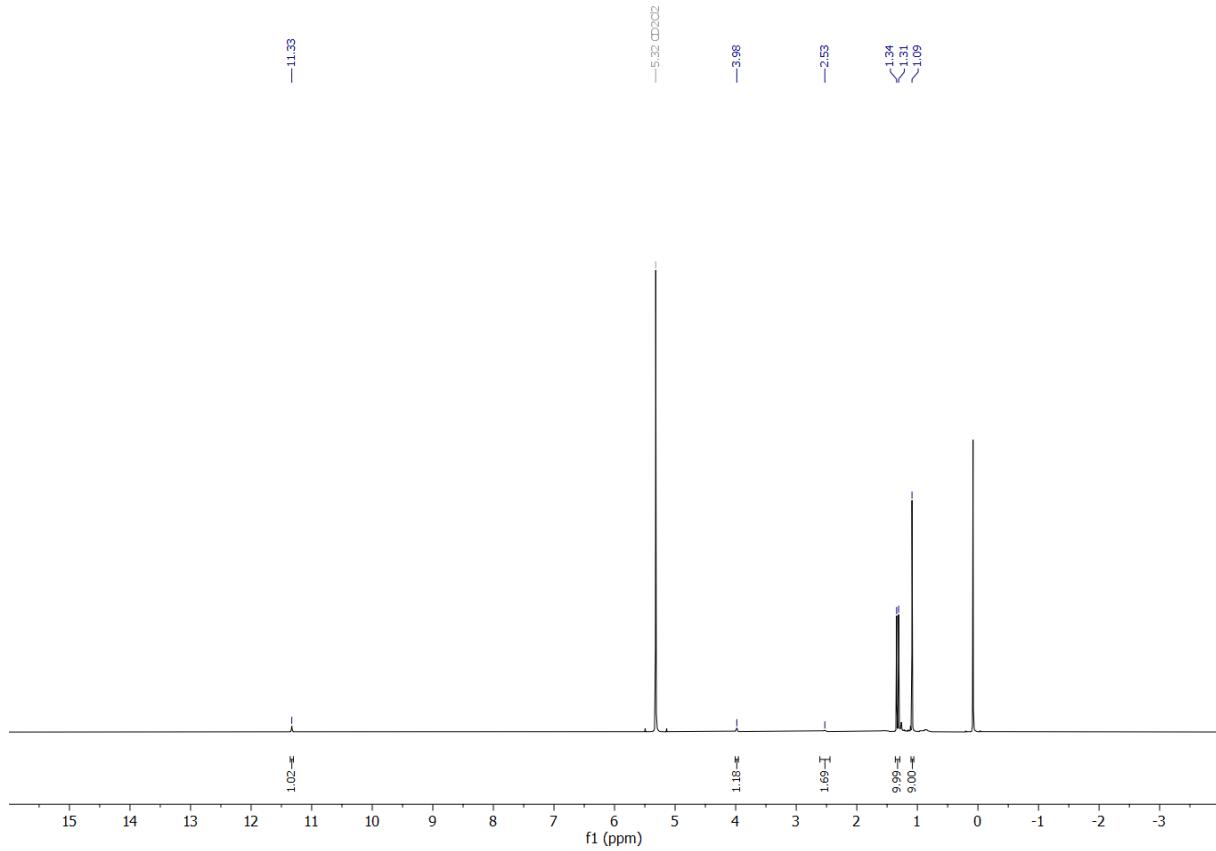


Figure S23: MS spectrum (APCI) of **8,8'**.



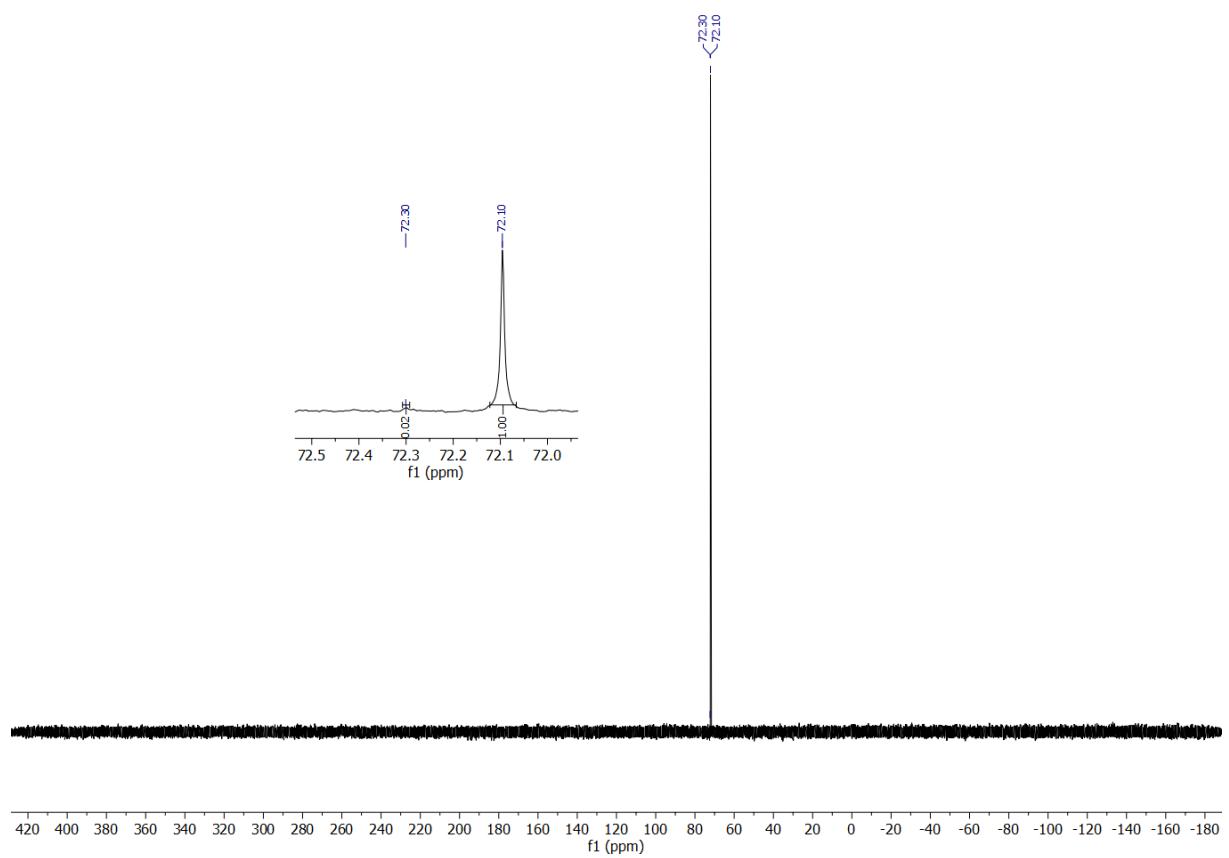


Figure S26: $^{31}\text{P}\{\text{H}\}$ -NMR spectrum of **8,8'** in CD_2Cl_2 .

Computational details

DFT calculations were performed with the ORCA electronic structure program package (version 4.2.1).² All geometry optimizations were run in redundant internal coordinates with tight convergence criteria, in the gas phase and using Grimme's dispersion-corrected composite PBEh-3c level.³ For the mechanistic study, solvent (toluene) effects were taken into consideration with the CPCM solvation method⁴ as implemented in ORCA. Harmonic frequency calculations verified the nature of ground or transition states (TS) having all positive frequencies or only one imaginary frequency, respectively. TS structures were confirmed by following the intrinsic reaction path in both directions of the negative eigenvector. From these optimized geometries, all reported data were obtained by means of single-point (SP) calculations using the higher quality def2-QZVP basis set.⁵ Reported energies include the zero point-energy (ZPE) correction term at the optimization level. In case of mechanistic aspects, final energies were obtained by means of double-hybrid-meta-GGA functional PW6B95⁶ using the RI-JK⁷ approximation for Coulomb and exchange integrals. Additionally, the latest Grimme's semiempirical atom-pair-wise London dispersion correction D4 was included.⁸ The *P*-thionation (and isomerization) process **2**→**10**→**6** was re-evaluated by optimization (and frequency calculation) using the B3LYP functional,⁹ the D4 correction and the def2-TZVP basis set, and final single-point energy calculation with the double-hybrid-meta-GGA functional PW-PB95,¹⁰ the Grimme's semiempirical D3 correction¹¹ and the def2-QZVPP functional (Figure S22). The RSEs of the parent oxaphosphirane *P*-sulfide and of its thiaphosphirane *P*-oxide isomer were obtained as average of the energies (including zero-point correction) of the three homodesmotic reactions for the cleavage of the three endocyclic calculations, as in previous reports,¹² the final energies being

² F. Neese, The ORCA program system. *Wiley Interdiscip. Rev. Comput. Mol. Sci.*, 2012, **2**, 73–78.

³ S. Grimme, J. G. Brandenburg, C. Bannwarth and A. Hansen, Consistent structures and interactions by density functional theory with small atomic orbital basis sets, *J. Chem. Phys.*, 2015, **143**, 54107.

⁴ V. Barone and M. Cossi, Quantum Calculation of Molecular Energies and Energy Gradients in Solution by a Conductor Solvent Model, *J. Phys. Chem. A*, 1998, **102**, 1995–2001.

⁵ F. Weigend and R. Ahlrichs, Balanced basis sets of split valence, triple zeta valence and quadruple zeta valence quality for H to Rn: Design and assessment of accuracy, *Phys. Chem. Chem. Phys.* 2005, **7**, 3297–3305

⁶ Y. Zhao and G. Truhlar, Design of Density Functionals That Are Broadly Accurate for Thermochemistry, Thermochemical Kinetics, and Nonbonded Interactions, *J. Phys. Chem. A*, 2005, **109**, 5656–5657.

⁷ (a) F. Weigend, M. Häser, H. Patzelt and R. Ahlrichs, RI-MP2: optimized auxiliary basis sets and demonstration of efficiency, *Chem. Phys. Lett.*, 1998, **294**, 143–152. (b) F. Weigend and M. Häser, RI-MP2: first derivatives and global consistency, *Theor. Chem. Acc.*, 1997, **97**, 331–340. (c) D.E. Bernholdt and R.J. Harrison, Large-scale correlated electronic structure calculations: the RI-MP2 method on parallel computers, *Chem. Phys. Lett.*, 1996, **250**, 477–484.

⁸ E. Caldeweyher, C. Bannwarth and S. Grimme, Extension of the D3 dispersion coefficient model. *J. Chem. Phys.*, 2017, **147**, 34112.

⁹ (a) A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648–5652. (b) C. T. Lee, W. Yang and R. G. Parr, *Phys. Rev. B*, 1988, **37**, 785–789.

¹⁰ (a) L. Goerigk and S. Grimme, Efficient and accurate double-hybrid-meta-GGA density functionals - Evaluation with the extended GMTKN30 database for general main group thermochemistry, kinetics, and noncovalent interactions, *J. Chem. Theory Comput.* 2011, **7**, 291–309. (b) L. Goerigk and S. Grimme, A thorough benchmark of density functional methods for general main group thermochemistry, kinetics, and noncovalent interactions, *Phys. Chem. Chem. Phys.* 2011, **13**, 6670–6688.

¹¹ (a) S. Grimme, J. Antony, S. Ehrlich and H. Krieg, A consistent and accurate *ab initio* parametrization of density functional dispersion correction (DFT-D) for the 94 elements H–Pu, *J. Chem. Phys.*, 2010, **132**, 154104–19. (b) S. Grimme, S. Ehrlich and L. Goerigk, Effect of the damping function in dispersion corrected density functional theory *J. Comput. Chem.*, 2011, **32**, 1456–1465.

¹² A. Espinosa Ferao, A. Rey Planells and R. Streubel, Between oxirane and phosphirane: the spring loaded oxaphosphirane ring, *Eur. J. Inorg. Chem.*, 2021, **4**, 348–353.

computed with the near linear scaling domain-based local pair natural orbital (DLPNO)¹³ method to achieve coupled cluster theory with single-double and perturbative triple excitations (CCSD(T))¹⁴ and using the def2-TZVPP basis set.

In case of compounds **2** and **7**, their RSEs were evaluated by means of only one homodesmotic reaction corresponding to the endocyclic C–O bond cleavage (Scheme S1), at the working CPCM_{tol}/PW6B95-D4/def2-QZVP//CPCM_{tol}/PBEh-3c level.

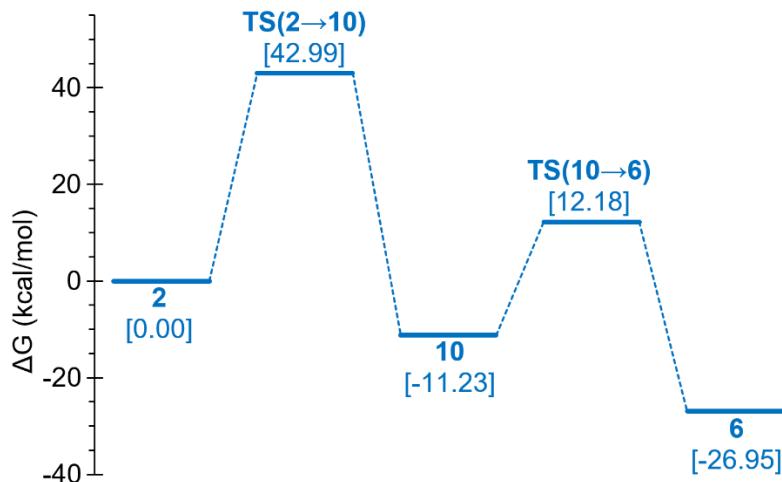
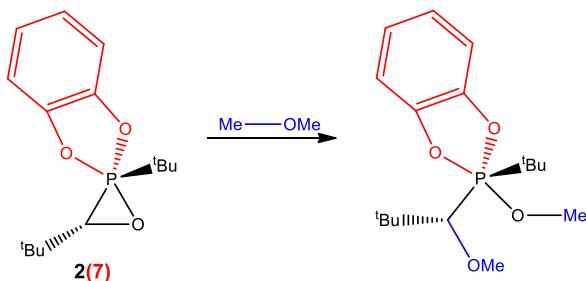


Figure S27: Computed (CPCM_{tol}/PW6B95-D4/def2-QZVP//CPCM_{tol}/B3LYP-D4/def2-TZVP) Gibbs free energy profile for the *P*-thionation of oxaphosphirane **2** followed by isomerization.



Scheme S1: Homodesmotic reactions used for RSE evaluation of compounds **2** and **7**.

¹³ C. Ripplinger, B. Sandhoefer, A. Hansen and F. Neese, Natural triple excitations in local coupled cluster calculations with pair natural orbitals. *J. Chem. Phys.*, 2013, **139**, 134101.

¹⁴ J. A. Pople, M. Head-Gordon and K. Raghavachari, Quadratic configuration interaction. A general technique for determining electron correlation energies, *J. Chem. Phys.*, 1987, **87**, 5968–5975.

Calculated structures

Cartesian coordinates (in Å), G correction (G-E) and zero-point energy (ZPE) correction (in hartrees) for minima and transition states were computed, unless otherwise stated, at the CPCM_{tol}/PBEh-3c whereas electronic energies (in hartrees) were quoted at the CPCM_{tol}/PW6B95-D4/def2-QZVP level. Imaginary frequencies were obtained upon frequency calculation at the optimization level.

2^{cis}:

E = -772.025356736206 au

ZPE = 0.27405774 au

G_{corr} = 0.23621866 au

C	-0.189079	-0.120914	0.016626	H	-4.197103	0.966071	2.262392
O	-0.244630	-0.135579	1.430808	H	-3.184219	-0.342699	1.668046
H	0.836465	0.013150	-0.349862	C	-0.841362	-1.232877	-0.792599
P	-0.676812	1.414528	0.853790	C	-2.155169	-1.766224	-0.237639
C	-2.524328	1.673319	1.112768	H	-2.961550	-1.039393	-0.294025
C	-2.564876	3.059065	1.779753	H	-2.468549	-2.634167	-0.820540
H	-3.601852	3.348317	1.965708	H	-2.054229	-2.086320	0.799263
H	-2.041525	3.066619	2.737782	C	0.165342	-2.395766	-0.805912
H	-2.114742	3.831529	1.152762	H	-0.216265	-3.224107	-1.405822
C	-3.264166	1.771707	-0.220400	H	1.123761	-2.092973	-1.231991
H	-3.338448	0.815031	-0.732383	H	0.350532	-2.771831	0.201661
H	-4.281347	2.137045	-0.055925	C	-1.029684	-0.720341	-2.220627
H	-2.776542	2.472452	-0.902116	H	-0.092153	-0.345455	-2.636822
C	-3.164276	0.667910	2.063232	H	-1.381430	-1.520551	-2.874296
H	-2.643863	0.636802	3.021256	H	-1.757551	0.090895	-2.261789

2^{cis}:

E = -769.762141170222 au [CPCM_{tol}/DLNPO-CCSD(T)/def2-TZVPP]

ZPE = 0.26467317 au [CPCM_{tol}/B3LYP-D4/def2-TZVP]

G_{corr} = 0.22620513 au [CPCM_{tol}/B3LYP-D4/def2-TZVP]

C	0.278473	0.105502	0.141911	H	-0.299747	-0.052487	3.021142
O	1.702417	0.087437	0.367696	H	-0.818780	-1.725296	2.909671
P	1.147428	1.696771	0.272500	C	1.198277	2.470034	2.000039
H	0.051892	-0.060957	-0.915194	C	2.064652	1.685697	2.986138
C	-0.631960	-0.784210	0.984947	H	1.658246	0.706101	3.216369
C	-0.607561	-2.173203	0.311430	H	3.073420	1.544033	2.593705
H	0.403210	-2.586583	0.308494	H	2.146999	2.249146	3.921006
H	-1.257193	-2.866522	0.850104	C	-0.211008	2.739937	2.538790
H	-0.955619	-2.116057	-0.722146	H	-0.831151	3.246360	1.794423
C	-2.055636	-0.216718	0.909203	H	-0.724702	1.829982	2.837763
H	-2.370487	-0.085259	-0.128981	H	-0.151529	3.394092	3.414326
H	-2.761814	-0.898869	1.388218	C	1.875822	3.829499	1.726680
H	-2.124549	0.751012	1.406714	H	2.882919	3.703304	1.323131
H	0.842627	-1.293560	2.496704	H	1.304481	4.431324	1.015968
C	-0.194534	-0.963306	2.438310	H	1.952309	4.392957	2.660335

2:

E = -772.043444222908 au
ZPE = 0.27353954 au
G_{corr} = 0.23520200 au

C	1.067595	0.912873	0.183950	H	-1.645258	-1.522377	-2.839707
O	0.625812	2.130624	-0.392179	H	-0.831066	-1.772557	-1.296772
H	1.756279	0.360807	-0.457994	C	1.516408	0.936373	1.633914
P	-0.657807	1.012041	-0.325551	C	0.539198	1.685667	2.536617
C	-0.745334	0.279020	-2.045327	H	-0.443677	1.211101	2.552572
C	-1.641625	1.221783	-2.855057	H	0.913776	1.697537	3.561629
H	-1.834534	0.799974	-3.844706	H	0.408745	2.721437	2.221761
H	-2.608457	1.383207	-2.373504	C	2.888024	1.617446	1.691631
H	-1.172819	2.197009	-2.997595	H	3.271748	1.626355	2.713735
C	0.585984	0.095413	-2.761279	H	3.618472	1.096674	1.068792
H	1.205804	-0.665938	-2.286763	H	2.831522	2.651558	1.348539
H	0.408706	-0.229452	-3.789908	C	1.638171	-0.516041	2.093433
H	1.156657	1.023857	-2.806267	H	2.329798	-1.080132	1.463666
C	-1.445063	-1.070891	-1.865022	H	2.011045	-0.567905	3.117787
H	-2.404258	-0.973101	-1.350821	H	0.670862	-1.021353	2.064082

2:

E = -769.779931293223 au [CPCM_{tol}/DLNPO-CCSD(T)/def2-TZVPP]
ZPE = 0.26416819 au [CPCM_{tol}/B3LYP-D4/def2-TZVP]
G_{corr} = 0.22536264 au [CPCM_{tol}/B3LYP-D4/def2-TZVP]

C	-0.093847	0.115261	0.191694	H	-2.056320	1.245704	-1.350690
O	1.267786	-0.336947	0.313156	H	-2.955740	-0.243659	-1.661780
P	1.231807	1.281774	-0.187050	C	1.533093	2.261394	1.394637
H	-0.580115	0.204441	1.160280	C	3.061871	2.320786	1.546904
C	-0.969458	-0.573061	-0.845505	H	3.478837	1.328382	1.731245
C	-1.286657	-1.984535	-0.323519	H	3.542524	2.729198	0.654620
H	-1.801558	-1.942268	0.639452	H	3.329741	2.960686	2.392814
H	-1.930781	-2.518638	-1.025893	C	0.915964	1.669793	2.659864
H	-0.370196	-2.562331	-0.193196	H	-0.171776	1.752156	2.658059
C	-0.278902	-0.678307	-2.208770	H	1.184132	0.618580	2.778608
H	0.669900	-1.209816	-2.127938	H	1.282805	2.209436	3.538341
H	-0.915817	-1.224115	-2.907942	C	0.977986	3.666395	1.122924
H	-0.080768	0.307438	-2.633461	H	1.398042	4.099060	0.210888
H	-2.766584	0.340333	-0.007439	H	-0.109758	3.654448	1.021167
C	-2.262091	0.242088	-0.972065	H	1.228725	4.336862	1.949982

S₈:

E = -3188.42287609029 au
ZPE = 0.01286354 au
G_{corr} = -0.02474872 au

S	0.096285	0.023139	-0.015430	S	3.675032	-2.523372	1.613054
S	-0.024125	-0.096651	2.035593	S	2.489958	-3.696945	0.407372
S	1.896446	0.023106	2.765254	S	1.873990	-2.524020	-1.167452
S	2.507908	-1.895997	3.187733	S	-0.040023	-1.895929	-0.746123

TS(2+S₈→10): E = -3960.41964625693 au

ZPE = 0.2873280847 au
 G_{corr} = 0.2338461562 au
 v = -228.66 cm⁻¹

C	0.930745	-0.975857	-0.245944	H	-1.653643	-0.869387	-1.572848
O	0.458841	-0.641640	1.094126	H	-1.834504	-2.615294	-1.558457
H	2.005435	-1.166635	-0.259315	H	-1.769018	-1.728824	-0.042840
P	0.497308	0.700040	0.161418	C	0.480739	-3.361896	-0.460320
C	1.858805	1.831610	0.653409	H	0.002315	-4.144797	-1.050110
C	1.391652	2.621009	1.878562	H	1.554532	-3.559372	-0.456514
H	2.188799	3.306515	2.171826	H	0.115165	-3.448108	0.563576
H	0.506386	3.220823	1.667504	C	0.688981	-1.899329	-2.489793
H	1.178154	1.974812	2.729988	H	1.767535	-2.064525	-2.533644
C	3.118618	1.030923	0.976070	H	0.215288	-2.658274	-3.113125
H	3.506431	0.499314	0.106996	H	0.475536	-0.925632	-2.934133
H	3.894209	1.727072	1.299126	S	-1.269797	1.596870	-0.252118
H	2.960894	0.317624	1.784624	S	-2.041349	1.529545	1.977790
C	2.095647	2.770032	-0.532713	S	-2.995670	-0.124401	2.406704
H	1.205609	3.347440	-0.783769	S	-5.068504	0.227508	2.651365
H	2.882789	3.477478	-0.266618	S	-5.867976	-1.174022	1.417418
H	2.421996	2.232707	-1.424368	S	-6.189823	-0.186408	-0.401671
C	0.153761	-1.987082	-1.059509	S	-4.317346	0.518425	-0.906418
C	-1.356555	-1.776633	-1.050645	S	-4.191986	2.378369	-0.145629

S₇: E = -2789.86046580544 au
 ZPE = 0.0110838 au
 G_{corr} = -0.0245969 au

S	-0.234923	-0.336486	0.064648	S	3.023824	-2.919252	1.531573
S	0.101038	0.328233	2.122330	S	1.593122	-3.038949	0.064795
S	2.017874	-0.029836	2.554105	S	1.500575	-1.054077	-0.615228
S	2.095484	-2.045067	3.140125				

10: E = -1170.62391275481 au
 ZPE = 0.2768399 au
 G_{corr} = 0.23684445 au

C	1.094318	0.784601	0.223760	H	-1.915975	-1.359281	-2.914639
O	0.800298	2.010409	-0.530474	H	-1.285696	-1.689602	-1.302652
H	1.744007	0.110736	-0.338268	C	1.545480	0.916257	1.666374
C	-0.787758	0.282064	-2.103796	C	0.818578	2.008678	2.446247
C	-1.447829	1.353539	-2.977098	H	-0.246209	1.806994	2.545312
H	-1.659428	0.930175	-3.960995	H	1.242048	2.073364	3.449863
H	-2.389148	1.699281	-2.550856	H	0.932997	2.985495	1.976514
H	-0.798032	2.217852	-3.120847	C	3.046091	1.231500	1.644883
C	0.533882	-0.162671	-2.726410	H	3.433812	1.299123	2.662762
H	1.004157	-0.969687	-2.163873	H	3.613452	0.456706	1.124961
H	0.336273	-0.543689	-3.730121	H	3.244540	2.183777	1.150602
H	1.244521	0.657971	-2.821526	C	1.308091	-0.445305	2.322190
C	-1.723619	-0.915899	-1.935666	H	1.810085	-1.249723	1.779957
H	-2.680171	-0.624455	-1.503877	H	1.693302	-0.448242	3.342963

H	0.243204	-0.680019	2.365423	S	-2.136044	1.567993	0.480542
P	-0.526172	1.037786	-0.440397				

TS(10→6):

E = -1170.58552933398 au
ZPE = 0.27398764 au
G_{corr} = 0.23429785 au
v = -225.67 cm⁻¹

C	0.554873	-0.059960	0.496065	C	1.102167	0.108867	1.833175
O	-1.535762	-1.340534	0.656083	C	1.682373	1.562390	1.673690
H	1.311648	-0.286178	-0.257501	H	0.880232	2.290302	1.577942
C	-1.149400	-0.203320	-1.849490	H	2.248684	1.759352	2.585991
C	-2.610831	-0.214869	-2.308010	H	2.364530	1.664411	0.829138
H	-2.650588	-0.337380	-3.392412	C	2.267954	-0.861463	2.065264
H	-3.168233	-1.038496	-1.859559	H	2.754972	-0.622866	3.010934
H	-3.118056	0.716186	-2.054726	H	1.908390	-1.888503	2.120053
C	-0.412598	0.969937	-2.491623	H	3.016185	-0.799495	1.274018
H	0.636686	1.022627	-2.190032	C	0.115637	0.068072	2.991855
H	-0.425654	0.859933	-3.578438	H	-0.316145	-0.926060	3.091402
H	-0.879926	1.921520	-2.241313	H	0.641476	0.309934	3.915650
C	-0.486200	-1.528336	-2.231565	H	-0.697316	0.779186	2.862077
H	-0.986008	-2.379099	-1.771062	P	-1.174339	-0.029311	-0.009901
H	-0.533351	-1.657304	-3.315021	S	-1.911253	1.709109	0.522888
H	0.569854	-1.568653	-1.955570				

6:

E = -1170.64885667933 au
ZPE = 0.27643125 au
G_{corr} = 0.23664363 au

C	1.527669	2.161770	0.300012	C	1.209126	1.285668	1.506183
O	-1.040793	2.104819	-1.047771	C	2.248079	1.583998	2.588647
H	2.562322	2.041723	-0.019026	H	2.179773	2.620325	2.924574
C	1.141912	2.770837	-2.656865	H	2.090325	0.942689	3.457498
C	0.365030	3.881563	-3.368468	H	3.264780	1.412928	2.228773
H	0.735470	3.983110	-4.390405	C	1.351126	-0.164592	1.032968
H	-0.701648	3.664169	-3.420397	H	1.204083	-0.853210	1.866701
H	0.495361	4.843909	-2.871217	H	0.609612	-0.404695	0.268744
C	2.621887	3.127987	-2.577604	H	2.342132	-0.357889	0.616701
H	3.215499	2.334156	-2.122967	C	-0.189086	1.507239	2.076707
H	3.003342	3.273404	-3.590193	H	-0.970772	1.279367	1.354367
H	2.792369	4.049483	-2.021901	H	-0.331797	0.857918	2.941902
C	0.954774	1.436335	-3.387406	H	-0.327407	2.536583	2.408134
H	-0.095319	1.159276	-3.468212	P	0.371961	2.569261	-0.997877
H	1.358587	1.521933	-4.398150	S	1.067555	3.988153	0.339499
H	1.485397	0.622020	-2.890180				

TOB:

E = -2222.01913774255 au
ZPE = 0.04920142 au
G_{corr} = 0.01163246 au

00.277907-0.929869 -3.792104 02.6238110.081715 -2.875890

C2.418104	0.048633	-4.055498	C1.824567	-0.069012	-6.863906
C3.383185	0.531386	-5.053241	Cl4.8438851	1.152056	-4.450911
C1.088273-0.522223		-4.574396	Cl4.2181821	0.036155	-7.514079
C3.099318	0.474727	-6.367164	Cl1.576198	-0.089471	-8.543271
CO.878892-0.536778		-6.029011	Cl-0.599882	-1.169487	-6.571774

TS(2+TOB→7):

$$E = -2994.05722169897 \text{ au}$$

$$\text{ZPE} = 0.32406719 \text{ au}$$

$$G_{\text{corr}} = 0.27246393 \text{ au}$$

$$v = -222.28 \text{ cm}^{-1}$$

C	-0.194473	-0.129717	-0.072095	H	3.521967	-0.641243	2.333302
O	0.253547	0.314650	1.218856	C	3.350020	-2.212013	0.064002
P	1.565791	-0.069037	0.235930	H	3.736422	-3.129467	0.514241
H	-0.707223	-1.091699	-0.036145	H	3.024074	-2.453822	-0.947163
C	-0.892253	0.886541	-0.955889	H	4.172943	-1.503310	-0.002686
C	-0.918870	0.312795	-2.372658	C	1.093707	-2.733857	1.006552
H	0.089844	0.194223	-2.771405	H	0.716431	-3.004399	0.019992
H	-1.409133	-0.662682	-2.400917	H	1.496343	-3.641766	1.463001
H	-1.467961	0.975143	-3.043851	H	0.257151	-2.405611	1.623858
C	-2.321934	1.065544	-0.434134	O	2.814042	0.186463	-1.778810
H	-2.872986	1.770407	-1.059363	O	3.692988	1.217112	0.431223
H	-2.868649	0.120522	-0.440453	C	3.811071	1.899681	-0.590678
H	-2.327997	1.450717	0.586625	C	4.215622	3.281754	-0.595510
C	-0.183239	2.239793	-0.954173	C	3.328638	1.308869	-1.847060
H	-0.120759	2.670315	0.045098	C	4.283009	3.973474	-1.764910
H	0.829748	2.174804	-1.360277	C	3.294724	2.132933	-3.024711
H	-0.726253	2.944852	-1.585233	C	3.812549	3.391161	-2.997060
C	2.205342	-1.690369	0.931788	Cl	4.697238	3.957991	0.891967
C	2.717130	-1.375239	2.340895	Cl	4.899116	5.563077	-1.794469
H	1.922883	-1.007686	2.990875	Cl	3.884976	4.323586	-4.421537
H	3.110688	-2.291387	2.787366	Cl	2.650628	1.441084	-4.443753

7:

$$E = -2994.1696173707 \text{ au}$$

$$\text{ZPE} = 0.3279652 \text{ au}$$

$$G_{\text{corr}} = 0.27714634 \text{ au}$$

C	-0.068076	0.074076	0.055255	H	-0.559449	2.196142	-2.437836
O	1.401459	0.033129	0.077351	H	0.798615	2.302138	-3.553105
P	0.811068	1.587156	0.093985	H	0.457114	0.780002	-2.745411
H	-0.482263	-0.260983	-0.900361	O	-0.342908	2.819904	0.575017
C	1.415623	2.300796	-1.494709	O	1.822882	2.091967	1.333162
C	2.827488	1.774244	-1.767802	C	1.373347	3.173588	1.995845
H	2.852139	0.692828	-1.883163	C	2.028203	3.814024	3.019003
H	3.189753	2.218009	-2.696822	C	0.127042	3.588135	1.551919
H	3.528256	2.050751	-0.979212	C	1.407480	4.924004	3.605277
C	1.456942	3.830671	-1.429947	C	-0.494431	4.676933	2.121910
H	1.877819	4.200330	-2.366548	C	0.159153	5.348641	3.161083
H	0.470159	4.276861	-1.320672	Cl	3.557997	3.252499	3.530376
H	2.099135	4.201503	-0.629575	Cl	2.201250	5.753923	4.874064
C	0.464208	1.861971	-2.611861	Cl	-0.592273	6.707517	3.880022

Cl	-2.028202	5.162100	1.544432	H	-1.162926	-2.584801	1.916238
C	-0.806468	-0.561256	1.221116	H	-0.967811	-2.465658	0.168681
C	-2.286078	-0.211621	1.053230	H	0.441659	-2.348116	1.225503
H	-2.448665	0.866291	1.099924	C	-0.313096	-0.074705	2.582892
H	-2.678263	-0.569960	0.099130	H	0.753571	-0.251165	2.720902
H	-2.877197	-0.673740	1.845346	H	-0.514168	0.987073	2.740979
C	-0.609894	-2.076836	1.124209	H	-0.837067	-0.610717	3.375441

TS(7→11): E = -2994.12927936905 au
 ZPE = 0.32545893 au
 G_{corr} = 0.27468801 au
 v = -264.32 cm⁻¹

C	0.379720	-0.171431	0.174359	C	-0.026198	3.796592	1.467124
O	1.655856	-0.230356	0.296882	C	1.172504	4.934333	3.672467
P	0.549704	1.798665	0.055271	C	-0.625992	4.890556	2.047086
H	-0.071163	-0.415283	-0.815452	C	-0.011480	5.463246	3.166008
C	1.328111	2.337294	-1.535145	Cl	3.213589	3.130612	3.658921
C	2.440244	1.399778	-2.005172	Cl	1.914940	5.640985	5.040616
H	2.091570	0.384634	-2.178361	Cl	-0.731248	6.825485	3.906907
H	2.823522	1.789467	-2.950889	Cl	-2.077339	5.501931	1.389303
H	3.265900	1.356551	-1.298133	C	-0.570582	-0.658429	1.275088
C	1.892948	3.751023	-1.368982	C	-2.004500	-0.288788	0.912413
H	2.309284	4.064912	-2.327907	H	-2.157113	0.792704	0.898825
H	1.137391	4.487228	-1.096482	H	-2.283404	-0.678344	-0.069311
H	2.703424	3.788551	-0.641185	H	-2.704338	-0.704236	1.638752
C	0.182261	2.338492	-2.557927	C	-0.413471	-2.184419	1.280238
H	-0.622087	3.024015	-2.290099	H	-1.091322	-2.629643	2.010436
H	0.576359	2.659458	-3.523656	H	-0.649323	-2.615563	0.304816
H	-0.244370	1.343699	-2.697387	H	0.603722	-2.476532	1.541224
O	-0.478363	3.121000	0.395502	C	-0.199944	-0.105196	2.646783
O	1.598822	2.199915	1.285318	H	0.860914	-0.235818	2.857681
C	1.152312	3.274331	1.970404	H	-0.441041	0.957027	2.740545
C	1.767769	3.818746	3.072338	H	-0.764461	-0.619544	3.426046

^tBu-CHO: E = -272.243208915917 au
 ZPE = 0.14485613 au
 G_{corr} = 0.11447791 au

C	-0.595919	-0.355377	0.735427	C	-1.291264	-2.623836	0.140745
O	0.431020	0.269134	0.733351	H	-1.518165	-3.625669	0.508348
H	-1.521486	0.078903	0.301965	H	-2.204027	-2.222684	-0.304549
C	-0.787262	-1.745292	1.292199	H	-0.544201	-2.718339	-0.648789
C	-1.873370	-1.649736	2.370764	C	0.506304	-2.299109	1.871995
H	-1.551111	-1.033195	3.211229	H	1.290481	-2.363997	1.117310
H	-2.799168	-1.225704	1.976473	H	0.880057	-1.679237	2.687304
H	-2.105169	-2.644274	2.754782	H	0.341084	-3.302829	2.266532

11: E = -2721.91069673137 au
 ZPE = 0.17842989 au

$G_{corr} = 0.13477619$ au

P1.238320-0.822633	-1.143904	H0.050189-2.913417	0.346687
C1.740746-2.604152	-1.018033	O0.322143-0.818588	-2.576987
C2.795068-2.642347	0.094204	O2.582646-0.135203	-1.927078
H2.436583-2.190548	1.022004	C2.392354-0.140925	-3.260667
H3.050729-3.680152	0.317573	C3.332886 0.197276	-4.205076
H3.715043-2.133182	-0.196701	C1.112174-0.528029	-3.628732
C2.291818-3.231479	-2.293895	C2.965805 0.150962	-5.555344
H2.551064-4.274269	-2.094754	C0.738996-0.587235	-4.950896
H1.564170-3.236057	-3.105948	C1.682332-0.237072	-5.924357
H3.200898-2.740744	-2.642732	Cl4.9015410.651006	-3.704672
C0.476150-3.341973	-0.563398	Cl4.1157350.572549	-6.749338
H-0.297632-3.340260	-1.332533	Cl1.241008-0.296343	-7.575850
H0.721002-4.383769	-0.346592	Cl-0.843238-1.086252	-5.356466

TS(11→9):

$E = -4943.92502864806$ au

$ZPE = 0.22885534$ au

$G_{corr} = 0.17240629$ au

$v = -233.14 \text{ cm}^{-1}$

C0.275432-0.430784	-0.737790	C-0.8095000.923435	-3.866261
C0.530101-1.785507	0.612499	C-0.1093591.486421	-4.888546
C2.027895-1.935623	0.906305	Cl3.7388001.062543	-3.757110
H2.353892-1.315050	1.740099	Cl2.1437332.266872	-6.174731
H2.217388-2.974908	1.182767	Cl-0.9273512.191681	-6.202076
H2.650393-1.709684	0.043274	Cl-2.5109750.918667	-3.815491
C0.123255-2.745146	-0.506301	P0.236940 0.014569	0.129705
H0.325215-3.763727	-0.166930	O-1.3641870.294018	0.613676
H-0.938886-2.685239	-0.739755	O0.8314240.775957	1.552494
H0.691110-2.587147	-1.420251	C-0.1733861.199464	2.331476
C-0.267992-2.124154	1.874685	C-1.4230620.911178	1.804526
H-1.344826-2.067971	1.719902	C-1.2072152.221906	4.219523
H-0.037366-3.155771	2.149957	C-2.5771901.259200	2.467192
H-0.001223-1.508367	2.733699	C-2.4609331.925376	3.692435
O-0.682164-0.253245	-1.830572	C-0.0431491.858148	3.533536
O1.914049-0.200674	-1.804111	Cl1.5158862.189982	4.145771
C1.358613 0.277046	-2.789663	Cl-1.0743913.035549	5.717991
C2.041030 0.995010	-3.839834	Cl-3.8784612.371349	4.538793
C-0.1127530.240430	-2.815972	Cl-4.0917420.870931	1.781533
C1.336386 1.522483	-4.874236		

9:

$E = -4944.03937673687$ au

$ZPE = 0.23334925$ au

$G_{corr} = 0.17859671$ au

C1.774844-2.460749	-0.881760	C0.650804-0.465218	-5.026411
C3.192810-2.537566	-0.300095	C1.556896-0.182256	-6.055234
H3.278332-2.045530	0.667531	Cl4.9624920.494961	-4.041419
H3.439315-3.590048	-0.151811	Cl3.9848340.454683	-7.033302
H3.945293-2.117997	-0.966167	Cl1.014199-0.215915	-7.676189
C1.727023-3.259805	-2.190075	Cl-0.985399-0.843627	-5.327166
H2.000122-4.292152	-1.966550	P1.348039-0.691007	-1.217992
H0.733753-3.278157	-2.636595	O-0.151200-0.436233	-0.459805
H2.440111-2.902825	-2.934074	O2.0863150.186336	0.023406
C0.786658-3.074055	0.115894	C1.264553 0.442471	1.051295
H-0.238218-3.073356	-0.252169	C-0.0366080.081181	0.767953
H1.074299-4.113245	0.282885	C0.595993 1.204874	3.197860
H0.806660-2.581368	1.088784	C-1.0479740.264741	1.680449
O0.409593-0.666889	-2.622802	C-0.7175210.838446	2.913656
O2.677593-0.164753	-2.133608	C1.611234 1.007640	2.254654
C2.437511-0.142770	-3.449686	Cl3.2357221.432784	2.556738
C3.342609 0.141236	-4.444306	Cl0.9875511.897863	4.710989
C1.119999-0.435510	-3.735704	Cl-1.9471461.077795	4.077132
C2.886481 0.116768	-5.767443	Cl-2.640053-0.206819	1.288096