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Electronic Supporting Information

Restricted-magnesium-vapor-reduction of amorphous SiO/C precursors to polycrystalline Si/SiO_x/C hybrid anodes

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Materials and methods

In a typical process, 3.5 g SiO@C (SC, purchased from Btr New Material Group Co., D50 is 5.6 μm) and 1.2 g metal Mg powder (D50 is 61 μm) were separated and layered, placed in a designed graphite crucible, and the graphite crucible (20 mL) was placed in a tubular furnace, heated at 850 °C under argon atmosphere (flow rate controlled at 100 mL min⁻¹) with a heating rate of 5 °C min⁻¹, then held for 6 h. After cooling to room temperature, the mixture was washed with 2 M HCl solution for 6 h to remove redundant by-products. Finally, the samples were washed with distilled water several times, then vacuumized and dried at 60 °C for 12 hours to obtain SC-MG.

3.5 g SiO@C and 1.2 g metal Mg powder (D50 is 61 μm) were placed in an agate mortar and manually stirred for 0.5 h, placed in a designed graphite crucible, then through the same steps above to obtain SC-MP as a comparison sample.

Material Characterizations

The phase and crystallinity of the obtained samples were analyzed by powder X-Ray Diffraction (XRD Bruker D8 X-ray diffractometer, Cu K α radiation, $\lambda = 1.5406 \text{ \AA}$, operated from 20° to 80° (2 θ)) and Raman spectroscopy with 532 nm excitation laser (Raman, Thermo-Nicolet ALMECA, Holland). The hydrophilicity and contact angle of samples were analyzed by spinningdrop interface tensiometer (TX500 TM). The morphologies and element composition of the obtained samples were characterized by scanning electron microscope (SEM) (JSM-6490LV, Japan) and transmission

electron microscope (TEM, FEI Tecnai G2 F20). The sample with the section diagram was prepared by a Leica 3-x ion beam cutting system with copper foil on the substrate to protect the samples from helium beam damage. X-ray photoelectron spectroscopy (XPS, Thermo Scientific K-Alpha) was used to analyze the elements and components on the surface of the prepared samples. The specific surface area (BET, Quantachrome Autosorb-I) was used to analyze the change of the porosity of composites.

Electrochemical Measurements

To measure the electrochemical performance of SC-MG and SC-MP samples, type 2032 coin-type half cells were assembled. The active materials, super-P and sodium alginate (SA) were mixed at the weight ratio of 6:2:2 in deionized water to form homogenous slurry. Then the slurry was painted onto a copper foil, followed by drying at 100 °C for 12 h. The mass loading of active material for each electrode was about 1 mg cm⁻². The half cell was assembled by using metal lithium as the reference electrode and counter electrode in an Ar-filled glove box (O₂, H₂O < 0.1 ppm). The electrolyte is made up of 1 M LiPF₆ dissolved in EC:DEC=1:1 (volume ratio) with 5% FEC. The galvanostatic discharge-charge measurements of the coin cells were tested on a LAND instrument (CT2001A, China) with a voltage range of 0.005-1.5 V at room temperature. The cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS) were performed on an electrochemical workstation (CHI760E, China). EIS was measured over the frequency range of 100000 Hz to 0.01 Hz. CV characterizations with different scanning rate were conducted in the potential range of 0.01-2.0 V at a scan rate of 0.1 mV s⁻¹ to 0.6 mV s⁻¹.

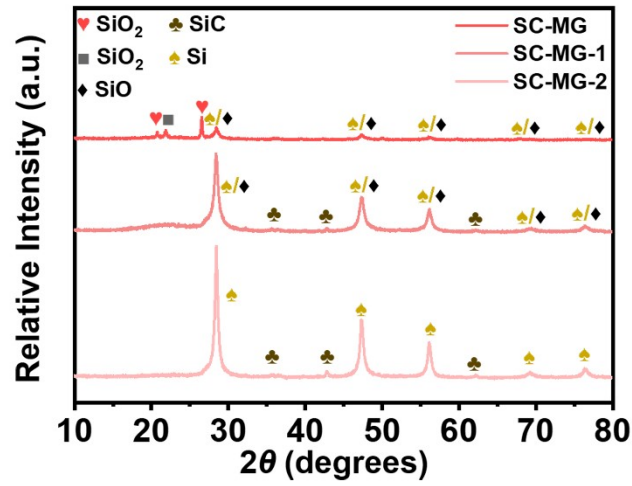


Fig. S1 XRD patterns of SC-MG samples with different proportions of SC and Mg.

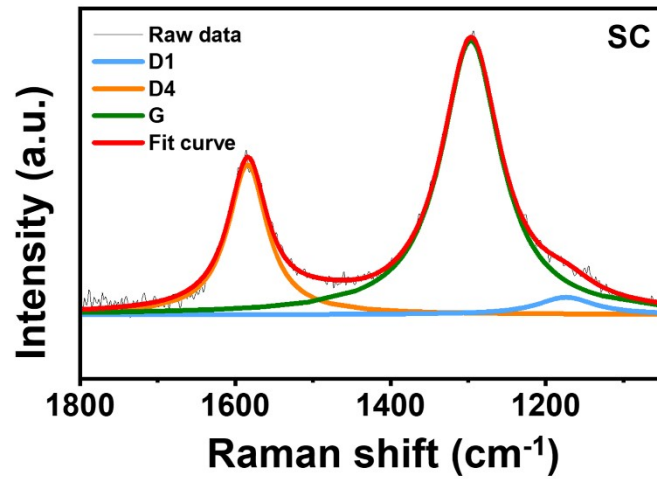


Fig. S2 Raman spectrum of SC.

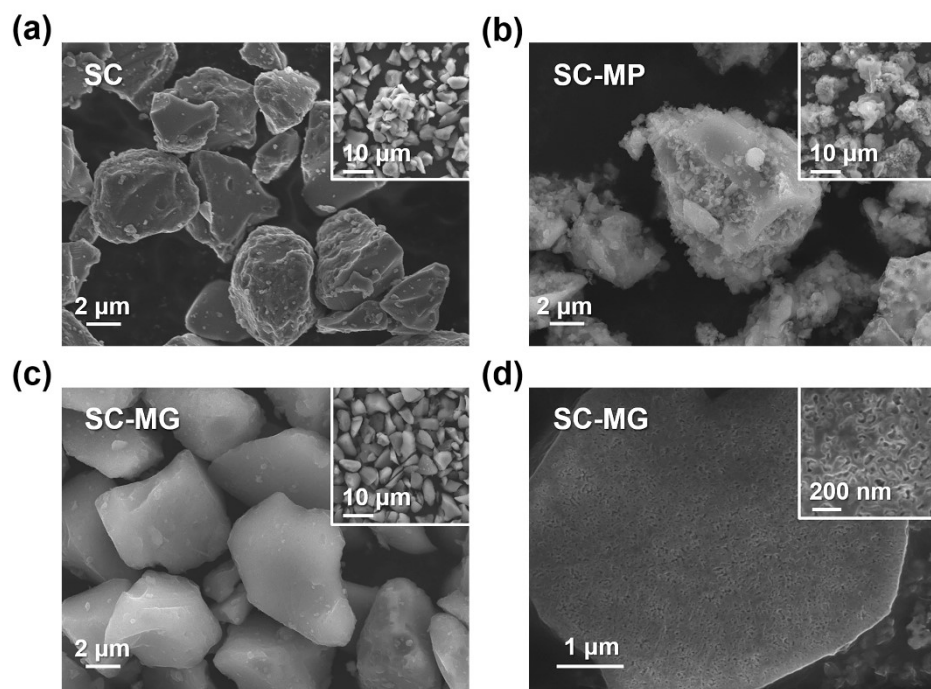


Fig. S3 SEM images of (a) SC, (b) SC-MP, and (c, d) SC-MG.

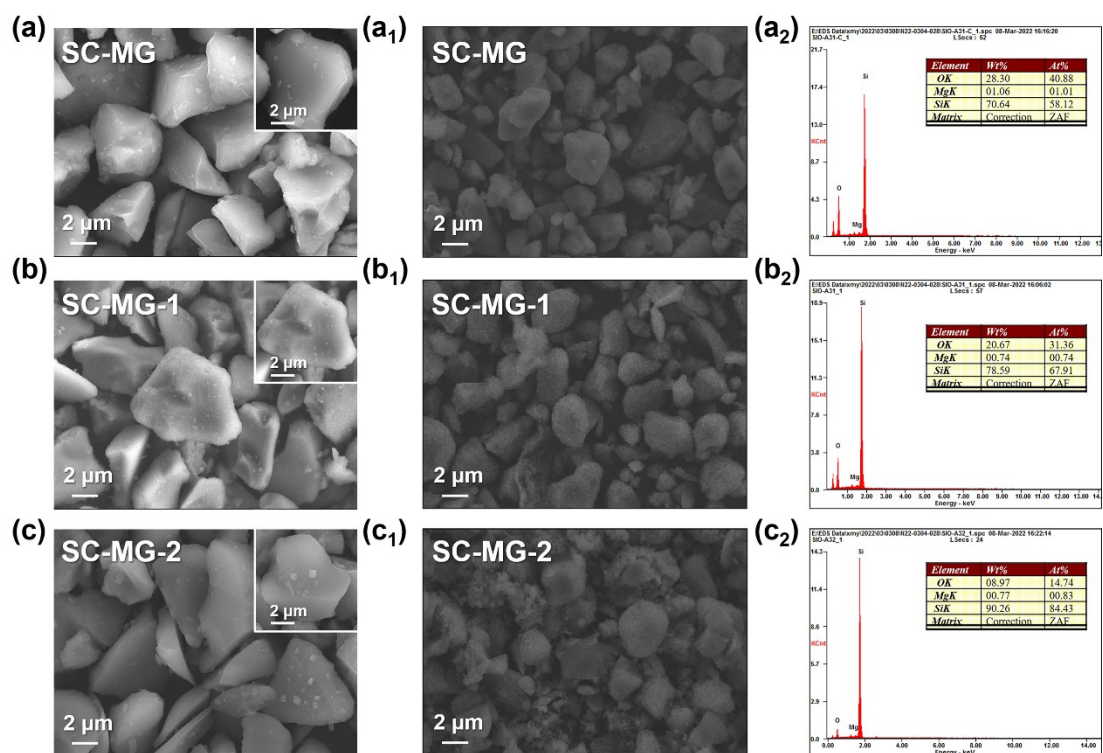


Fig. S4 SEM images of (a) SC-MG, (b) SC-MG-1, and (c) SC-MG-2.

With the increase of the ratio of Si-Mg, the magnesiothermic reduction carried out more thoroughly. It can be clearly observed that the particles surface gradually becomes rough, and the particles are obviously destroyed with the further increase of the ratio of Si-Mg. As shown in EDS mapping, the oxygen content in the products increased gradually.

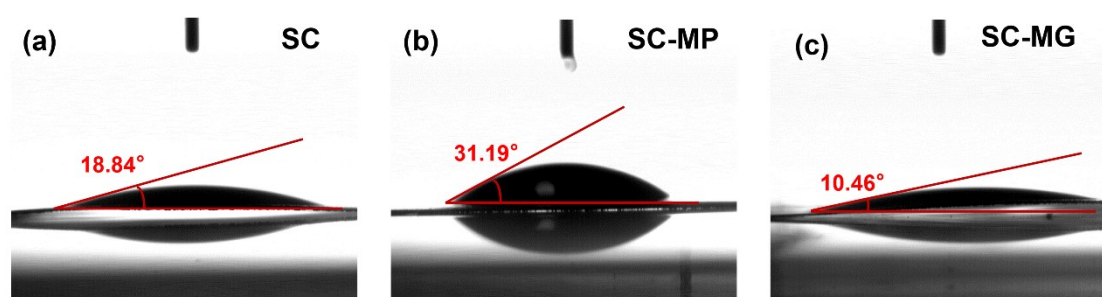


Fig. S5 The contact angle of electrodes with SC, SC-MP and SC-MG.

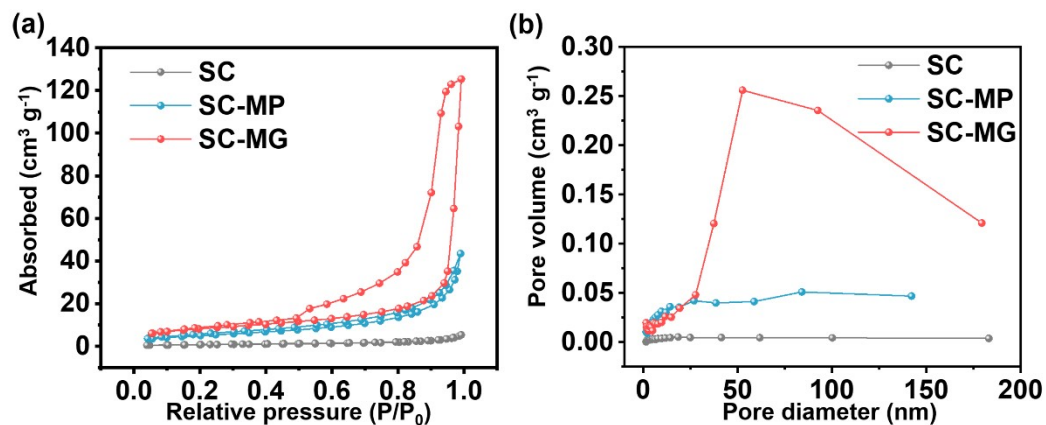


Fig. S6 (a) Nitrogen adsorption and desorption isotherms of SC, SC-MP, and SC-MG. (b) The corresponding pore size distribution of SC, SC-MP, and SC-MG.

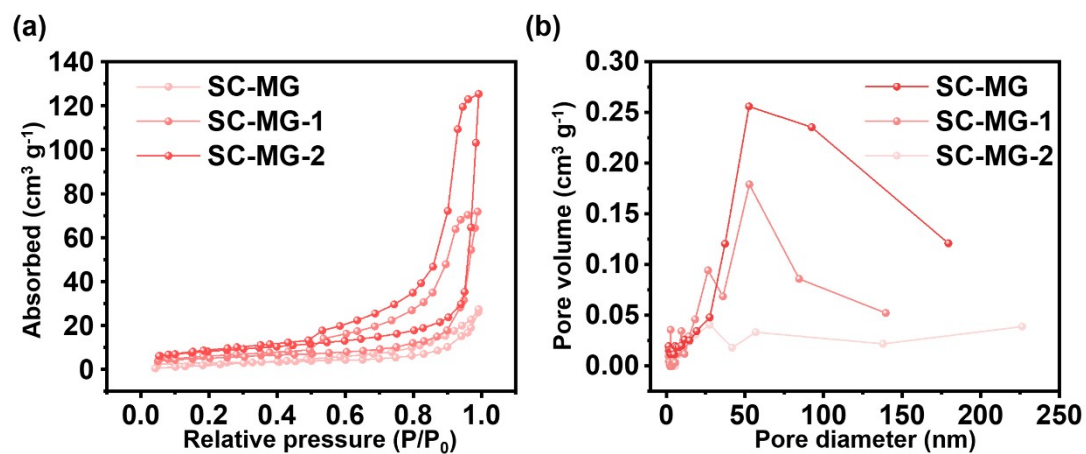


Fig. S7 (a) Nitrogen adsorption and desorption isotherms of SC-MG with different proportions of SC and Mg. (b) The corresponding pore size distribution of SC-MG with different proportions of SC and Mg.

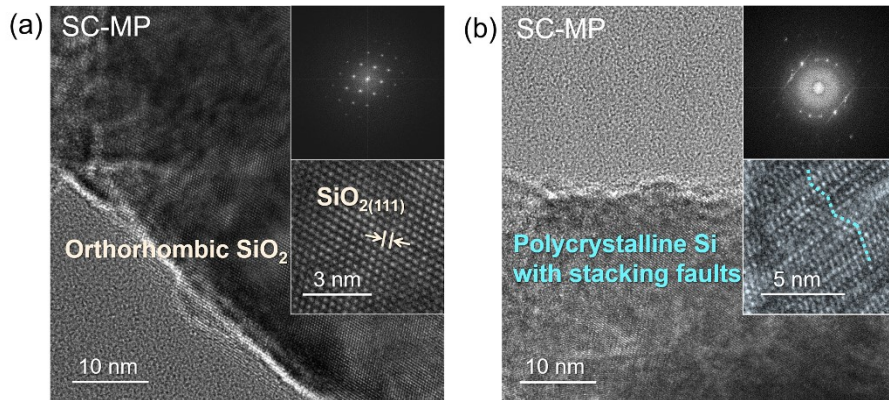


Fig. S8 HRTEM images of SC-MP and the corresponding SAED.

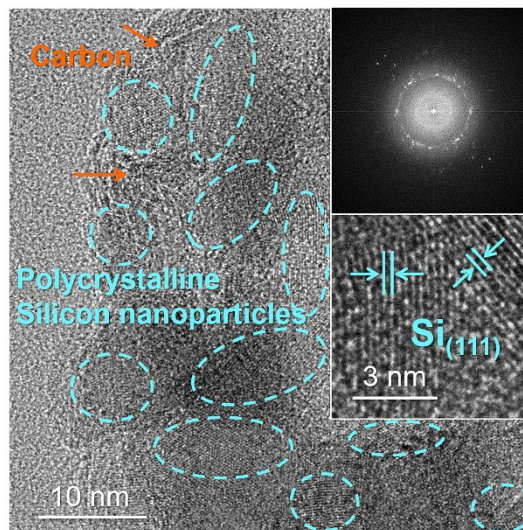


Fig. S9 HRTEM images of SC-MG and the corresponding SAED.

The presence of polycrystalline silicon nanoparticles and amorphous carbon layer observed in Fig. S9 further demonstrates that the successful synthesis of polycrystalline Si/SiO_x/C hybrid composites.

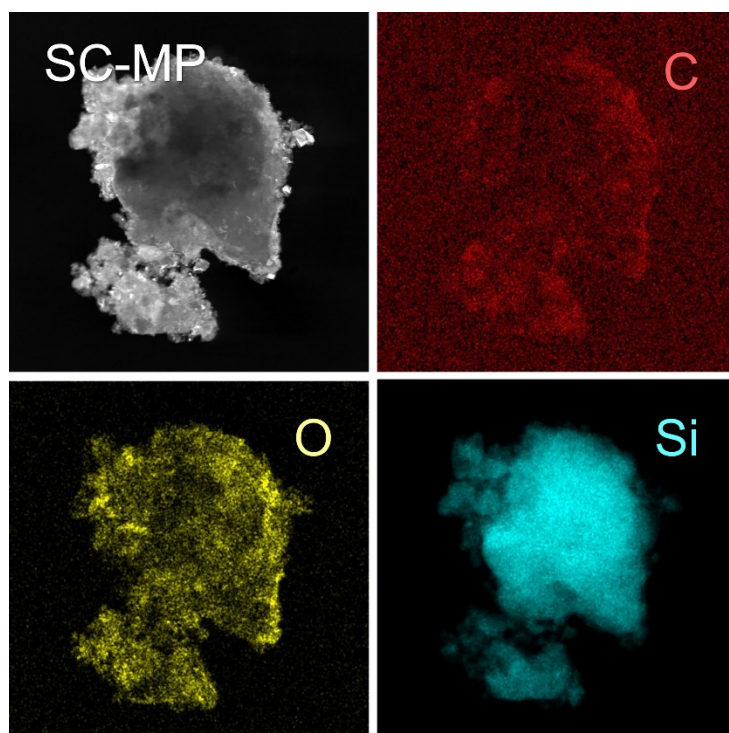


Fig. S10 EDS elemental mappings of SC-MG of Si, O and C.

The element distribution of SC-MP is shown in Fig. S10. The signal of Si significantly stronger than that of SC-MG, which indicates that the violence of Mg-powder-reduction.

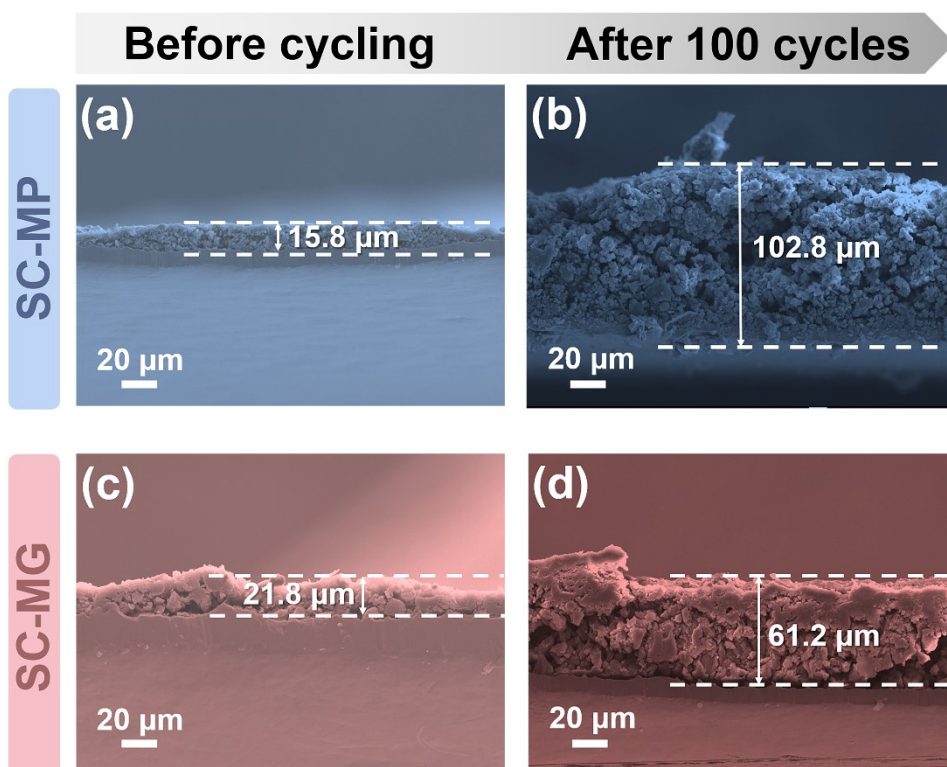


Fig. S11 Cross-sectional SEM images of the SC-MP and SC-MG electrodes at different cycles: (a) before cycling of SC-MP electrode, (b) after 100 cycles of SC-MP electrode, (c) before cycling of SC-MG electrode, and (d) after 100 cycles of SC-MG electrode.

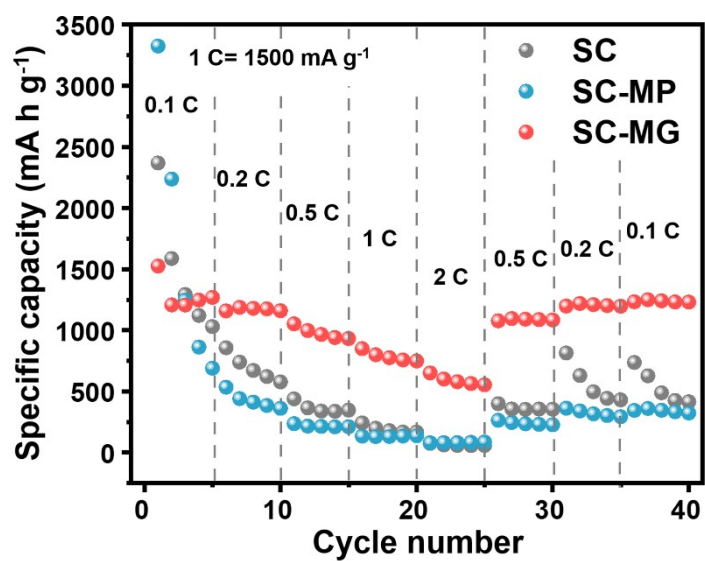


Fig. S12 Rate performance of SC, SC-MP and SC-MG anodes.

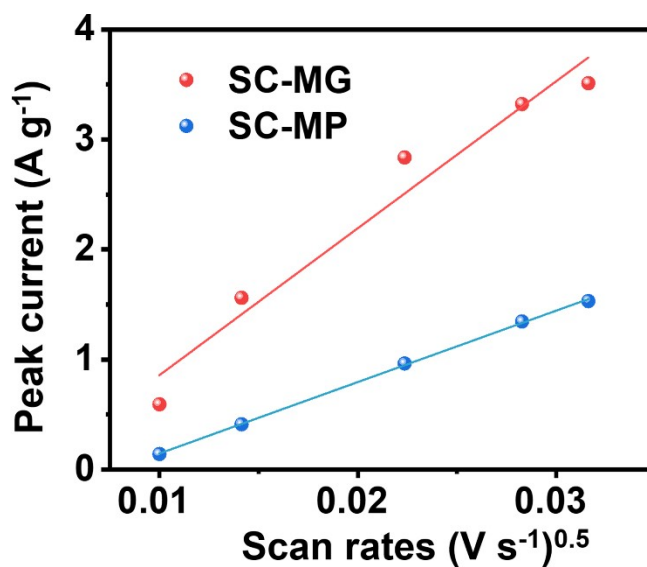


Fig. S13 Linear fits for the anodic peak current versus square root of scan rate of electrodes with SC-MP and SC-MG anodes.

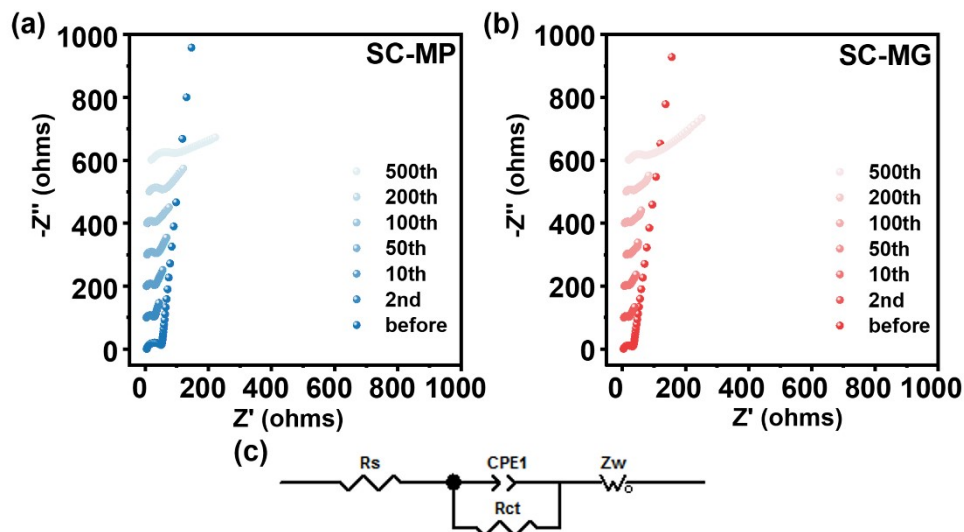


Fig. S14 EIS test with different cycles: (a) SC-MP anode, (b) SC-MG anode. (c) Analog circuit diagram corresponding to EIS results.

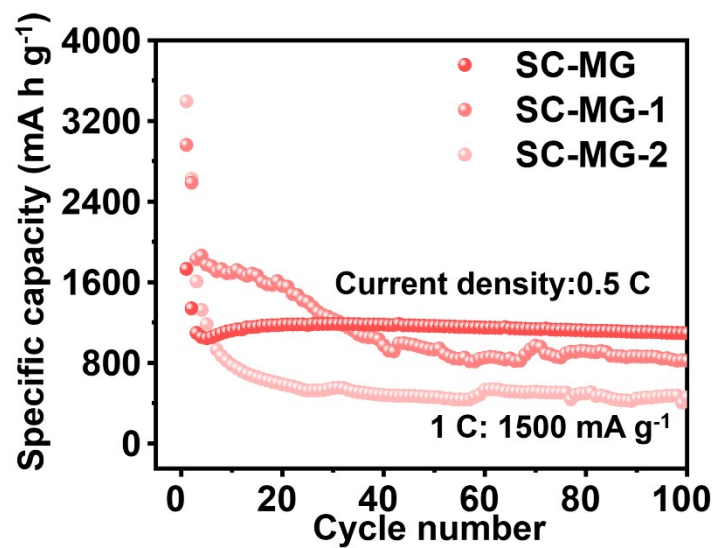


Fig. S15 Cycling performance of SC-MG anodes with different proportions of SC and Mg.

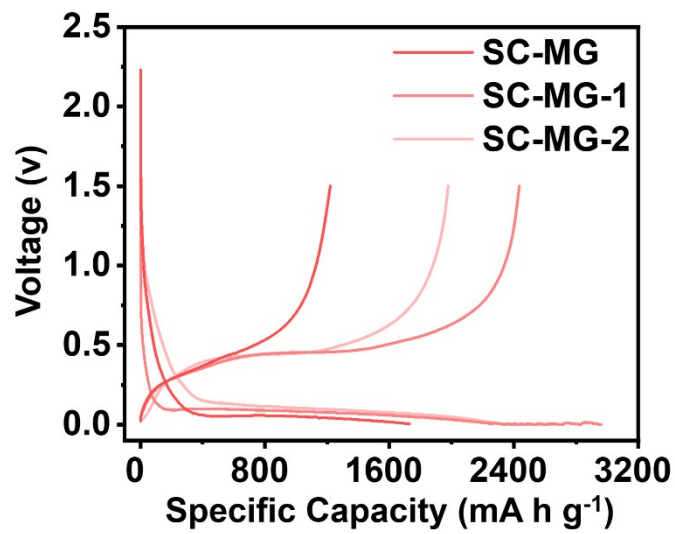


Fig. S16 Charge-discharge curves of SC-MG anodes with different proportions of SC and Mg in the first cycle.

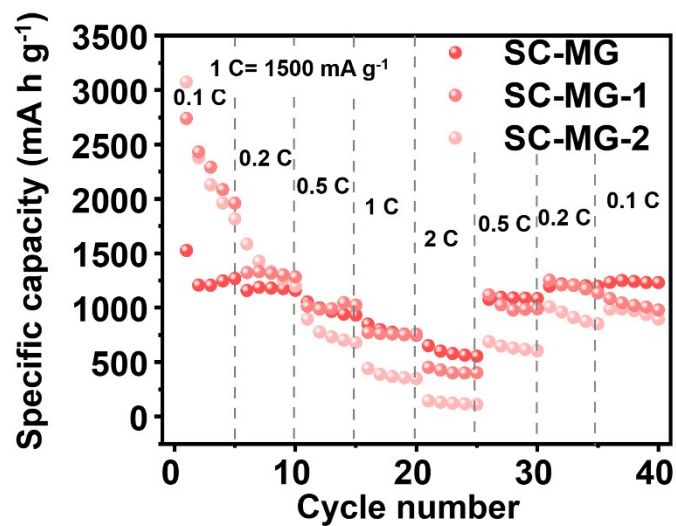


Fig. S17 Rate performance of SC-MG anodes with different proportions of SC and Mg.

Table S1 Comparison of specific surface area and total pore volume data of SC, SC-MP and SC-MG.

| Sample | S_{BET} ($\text{m}^2 \text{g}^{-1}$) | V_{G} ($\text{cm}^3 \text{g}^{-1}$) |
|--------|---|--|
| SC | 1.924 | 0.007 |
| SC-MP | 18.733 | 0.067 |
| SC-MG | 29.285 | 0.202 |

Table S2 Comparison of specific surface area and total pore volume data of SC-MG with different proportions of SC and Mg.

| Sample | S_{BET} ($\text{m}^2 \text{g}^{-1}$) | V_{G} ($\text{cm}^3 \text{g}^{-1}$) |
|---------|---|--|
| SC-MG | 29.285 | 0.202 |
| SC-MG-1 | 18.394 | 0.111 |
| SC-MG-2 | 10.188 | 0.042 |

Table S3 Comparison of EIS fitting data with different cycles of SC-MG and SC-MP anodes.

| Sample | Impedance/ Ω | 0th | 2nd | 10th | 50th | 100th | 200th | 500th |
|--------------|---------------------|-------|-------|-------|-------|-------|--------|-------|
| | R_s | 2.76 | 4.389 | 6.262 | 8.51 | 8.132 | 14.62 | 16.78 |
| SC-MG | R_{ct} | 21.38 | 12.68 | 8.185 | 9.692 | 15.86 | 19.78 | 28.63 |
| | Z_w | 20.16 | 11.78 | 23.71 | 20.03 | 31.45 | 157.6 | 240.2 |
| | R_s | 2.699 | 4.375 | 5.046 | 7.875 | 8.467 | 13.68 | 17.26 |
| SC-MP | R_{ct} | 42.46 | 24.56 | 20.42 | 26.73 | 28.48 | 34.64 | 42.08 |
| | Z_w | 25.67 | 14.72 | 27.83 | 45.82 | 70.47 | 250.18 | 960.8 |

Table S4. Comparison of the latest various routes of synthesizing SiO_x as anodes for LIBs.

| Samples | Synthetic method | Initial Capacity/ (mA h g ⁻¹) | Initial CE | Cycling performance | Ref. |
|--------------------------------|---|--|------------|---|------|
| SiO _x /graphene-CMC | liquid ball milling and ultrasonic | 1723.7 | 73.9 % | 1070.2 mA h g ⁻¹ after 100 cycles at 200 mA g ⁻¹ | S1 |
| SiO _x /SNWs@C | in-situ grown and chemical vapor deposition (CVD) | 1280 | ~ 74.5 % | 950 mA h g ⁻¹ for the following 300 mA g ⁻¹ with capacity retention >80 % over 200 cycles | S2 |
| SiO _x /C | disproportion and high-energy ball milling | 1213.9 | 65.7 % | 1292.4 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹ | S3 |
| SiO _x | disproportionation and ball milling | 2264.3 | 67.7 % | ~1250 mA h g ⁻¹ after 50 cycles at 300 mA g ⁻¹ | S4 |
| Si/SiO _x @NC | magnesiothermic reduction | 1501 | 49 % | 705 mA h g ⁻¹ after 100 cycles at 500 mA g ⁻¹ | S5 |
| Si@SiO _x /C | magnesiothermic reduction with carbonization | 2034 | 80.7 % | 1125 mA h g ⁻¹ after 100 cycles at 1 A g ⁻¹ | S6 |
| SiO _x | magnesiothermic reduction | 1862 | 83 % | 89.1 cycling retention after 200 cycles at 750 mA g ⁻¹ | S7 |
| Si/SiO _x | magnesiothermic reduction | 2224 | 89 % | 980 mA h g ⁻¹ after 100 cycles at 100 mA g ⁻¹ | S8 |
| SiO@C | milled and calcined | 1100 | 51.8 % | 699 mA h g ⁻¹ after 700 cycles at 1 A g ⁻¹ | S9 |
| D-SiO@G | in situ graphene-coated by heat treatment | 1937.6 | 78.2 % | 72.4 % capacity retention after 500 cycles at 2.0 A g ⁻¹ | S10 |
| TC-SiO | carbon coating with 3 wt% TiO ₂ | 1251 | 71.9 % | 674.5 mA h g ⁻¹ after 100 cycles at 0.14 A g ⁻¹ | S11 |
| Porous SiO | metal-assisted chemical etching | 2250 | 60 % | 54.3 % capacity retention after 50 cycles at 0.1 C (0.15 A g ⁻¹) | S12 |

| | | | | | |
|------------------------------|---|--------|--------|--|--------------|
| P-SiO@gl- C@F- doped C | milled and calcined | 1124 | 60 % | 772 mA h g ⁻¹ after 200 cycles at 0.4 A g ⁻¹ | S13 |
| Si/SiO _x /C | restricted magnesium- vapor- reduction with and CVD | 1729.6 | 70.6 % | 943 mA h g ⁻¹ after 200 cycles at 750 mA g ⁻¹ | this work |

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