Conformational Flexibility and Substitution Pattern Lead to Polymorphism of Methyl-2-

(phenylamino)benzoic acid

Synthesis of MPABAs

1. Synthesis of 3-methyl-2-phenylamino-benzoic acid (1)



A mixture of 2-chloro-*m*-toluic acid (1 g, 5.86 mmol), *o*-toluidine (640 mg, 5.97 mmol), K_2CO_3 (0.809 g, 5.9 mmol), Cu powder (34 mg, 0.54 mmol), and Cu₂O (38 mg, 0.3 mmol) in 3 mL of 2-ethoxyethanol was refluxed at 130°C under inert atmosphere for 24 h. The cooled reaction mixture was poured into 30 mL of water. The crude product was obtained by precipitation upon acidification of the filtrate with dilute HCl. Then the crude product was purified on column chromatography and recrystallization. The product was obtained as a green solid (0.504 g, yield%: 37.9).

¹H NMR (400 MHz, DMSO-*d*₆) δ*ppm* 8.53 (s, 1H), 7.77 (d, *J* = 7.5 Hz, 1H), 7.43 (d, *J* = 7.0 Hz, 1H), 7.16 (s, 2H), 7.11 (s, 1H), 6.80 (s, 1H), 6.62 (d, *J* = 7.5 Hz, 2H), 2.03 (s, 3H); ¹³C NMR (151 MHz, DMSO-*d*₆) δ*ppm* 169.4, 144.9, 142.5, 136.0, 135.9, 133.2, 129.5, 129.4, 128.4, 123.4, 123.4, 116.9, 115.8, 19.3; mp: 188.9°C.

2. Synthesis of 3-methyl-2-o-tolylamino-benzoic acid (2)



The procedure is similar to the preparation of 1 with 2-chloro-*m*-toluic acid (1 g, 5.86 mmol), *o*-toluidine (640 mg, 5.97 mmol), K_2CO_3 (1.38 g, 9.98 mmol), Cu (100 mg, 1.57 mmol), Cu₂O (100 mg, 0.7 mmol), 2-ethoxyethanol (3 mL) to yield green solid (0.3 g, yield%: 21.4).

¹H NMR (500 MHz, DMSO- d_6) δppm 8.47 (s, 1H), 7.81 (d, J = 7.7 Hz, 1H), 7.43 (d, J = 7.3 Hz, 1H), 7.16 (d, J = 7.3 Hz, 1H), 7.08 (t, J = 7.6 Hz, 1H), 6.99 (t, J = 7.5 Hz, 1H), 6.80 (t, J = 7.3 Hz, 1H), 6.24 (d, J = 7.9 Hz, 1H), 2.30 (s, 3H), 1.96 (d, J = 12.3 Hz, 3H); ¹³C NMR (126 MHz, DMSO- d_6) δppm 169.8, 144.1, 142.7, 135.8, 132.4, 130.3, 129.2, 126.4, 126.3, 122.2, 121.7, 120.5, 115.4, 18.9, 17.6; mp: 191.8 °C.

3. Synthesis of 3-methyl-2-m-tolylamino-benzoic acid (3)



The procedure is similar to the preparation of 1 with 2-chloro-*m*-toluic acid (1.5 g, 8.79 mmol), *m*-toluidine (1.9 g, 17.73 mmol), K_2CO_3 (2.43 g, 17.58 mmol), Cu (100 mg, 1.57 mmol), Cu₂O (100 mg, 0.7 mmol), 2-ethoxyethanol (5 mL) to yield green solid (0.36 g, yield%: 17.1).

¹H NMR (500 MHz, DMSO- d_6) δppm 8.48 (s, 1H), 7.76 (d, J = 7.6 Hz, 1H), 7.43 (d, J = 7.3 Hz, 1H), 7.07 (dt, J = 27.7, 7.7 Hz, 2H), 6.63 (d, J = 7.4 Hz, 1H), 6.50 – 6.35 (m, 2H), 2.19 (s, 3H), 2.01 (d, J = 14.2 Hz, 3H); ¹³C NMR (126 MHz, DMSO- d_6) δppm 169.5, 144.8, 142.7, 138.2, 135.5, 133.2, 129.0, 128.8, 123.2, 122.7, 120.7, 117.1, 113.7, 21.2, 18.9; mp: 172.8 °C.

4. Synthesis of 3-methyl-2-p-tolylamino-benzoic acid (4)



The procedure is similar to the preparation of 1 with 2-chloro-*m*-toluic acid (2 g, 11.72 mmol), *p*-toluidine (2.2 g, 20.53 mmol), K_2CO_3 (3.24 g, 23.41 mmol), Cu (100 mg, 1.57 mmol), Cu₂O (100 mg, 0.7 mmol), 2-ethoxyethanol (5 mL) to yield green solid (0.32 g, yield%: 11.4).

¹H NMR (500 MHz, DMSO- d_6) δppm 8.55 (s, 1H), 7.77 (d, J = 7.6 Hz, 1H), 7.41 (d, J = 7.3 Hz, 1H), 7.06 (t, J = 7.6 Hz, 1H), 6.99 (d, J = 8.0 Hz, 2H), 6.56 (d, J = 8.1 Hz, 2H), 4.03 (dd, J = 14.2, 7.1 Hz, 6H); ¹³C NMR (126 MHz, CDCl₃) δppm 171.6, 166.1, 144.7, 141.9, 137.1, 133.3, 131.2, 130.2, 129.7, 128.4, 122.7, 120.9, 118.6, 20.8, 19.6; mp: 195.4 °C.

5. Synthesis of 2-(2,3-mimethyl-phenylamino)-3-methyl-benzoic acid (5)



The procedure is similar to the preparation of 1 with 2-chloro-*m*-toluic acid (2 g, 11.72 mmol), 2,3-dimethylphenylamine (2.2 g, 18.15 mmol), K_2CO_3 (2.8 g, 20.26 mmol), Cu (100 mg, 1.57 mmol), Cu₂O (100 mg, 0.7 mmol), 2-ethoxyethanol (5 mL) to yield green solid (0.7 g, yield%: 24.1).

¹H NMR (500 MHz, DMSO- d_6) δppm 8.52 (s, 1H), 7.80 (t, J = 12.2 Hz, 1H), 7.40 (d, J = 6.8 Hz, 1H), 7.04 (t, J = 7.3 Hz, 1H), 6.87 (t, J = 7.3 Hz, 1H), 6.74 (d, J = 6.9 Hz, 1H), 6.14 (d, J = 7.5 Hz, 1H), 2.28 (t, J = 34.4 Hz, 6H), 1.90 (s, 3H); ¹³C NMR (101 MHz, DMSO- d_6) δppm 169.84, 144.85, 142.54, 136.80, 135.82, 131.81, 129.13, 125.50, 125.30, 122.76, 121.62, 120.95, 114.36, 20.22, 19.02, 13.10; ESI-MS m/z (M+18) 274.3; mp: 194.1 °C.

Crystal Growth

Example: 250 mg of **1** was suspended in 5 mL methanol. The mixture was stirred overnight and the remaining solid was removed by pipette filtration. A vial containing the clear solution was covered with perforated parafilm. Slow evaporation led to single crystals in about a week.



Crystal Structures

Figure S1 The acid-acid dimer motif in 1-4 (for clarity, H atoms not involved in hydrogen bonding are omitted)



Figure S2 The acid-acid heterodimer motif (A: red; B: blue) in R1, and homodimer in R2-

R4 (for clarity, H atoms not involved in hydrogen bonding are omitted)



Figure S3 The acid-acid heterodimer motif (A: red; B: blue) in **5-I**, and homodimer in **5-II** and **5-III** (for clarity, H atoms not involved in hydrogen bonding are omitted)



Figure S4 The acid-acid homodimer in R5-I, R5-II and R5-III (for clarity, H's not involved in hydrogen bonding are omitted)

PXRD



Figure S5 PXRD patterns of compounds 1-4.



Figure S6 DSC thermograms of compounds 1-4.

IR



Figure S7 IR spectra of compound 1-4.



Figure S8 IR spectra of the three polymorphs and solvate of compound 5.

Bond composition and bond order



 $\begin{array}{ll} \text{Bond composition: } \sigma_{CBAN} = 0.627(sp^{2.59})_{C} + 0.778(sp^{1.87})_{N} ; \sigma_{CbenzeneN} = 0.622(sp^{2.72})_{C} + \\ & 0.783(sp^{1.95})_{N} \\ \text{Bond order: } C_{BA}N & 0.981 ; C_{benzene}N & 0.976 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{CBAN} = 0.629(sp^{2.62})_C + 0.778(sp^{1.90})_N; \sigma_{CbenzezeN} = 0.623(sp^{2.71})_C + \\ 0.782(sp^{1.91})_N \\ \text{Bond order: } C_{BA}N \quad 0.980; \ C_{benzene}N \quad 0.975 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{CBAN} = 0.629(sp^{2.61})_{C} + 0.777(sp^{1.91})_{N} ; \sigma_{CbenzeneN} = 0.624(sp^{2.62})_{C} + \\ 0.782(sp^{1.90})_{N} \\ \text{Bond order: } C_{BA}N \quad 0.980 ; C_{benzene}N \quad 0.976 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{\text{CBAN}} = 0.628(sp^{2.62})_{\text{C}} + 0.778(sp^{1.93})_{\text{N}}; \; \sigma_{\text{CbenzeneN}} = 0.622(sp^{2.73})_{\text{C}} + \\ 0.783(sp^{1.88})_{\text{N}} \\ \text{Bond order: } C_{\text{BA}}\text{N} \quad 0.979; \; C_{\text{benzene}}\text{N} \quad 0.975 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{CBAN} = 0.628(sp^{2.55})_{C} + 0.778(sp^{1.85})_{N} ; \sigma_{CbenzeneN} = 0.621(sp^{2.75})_{C} + \\ 0.784(sp^{1.92})_{N} \\ \text{Bond order: } C_{BA}N \quad 0.981 ; C_{benzene}N \quad 0.976 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{CBAN} = 0.628(sp^{2.58})_{C} + 0.778(sp^{1.88})_{N} ; \sigma_{CbenzeneN} = 0.621(sp^{2.77})_{C} + \\ 0.784(sp^{1.91})_{N} \\ \text{Bond order: } C_{BA}N \quad 0.981 ; C_{benzene}N \quad 0.975 \end{array}$



 $\begin{array}{l} \text{Bond composition: } \sigma_{\text{CBAN}} = 0.629(sp^{2.62})_{\text{C}} + 0.777(sp^{1.94})_{\text{N}} ; \sigma_{\text{CbenzeneN}} = 0.622(sp^{2.72})_{\text{C}} + \\ 0.783(sp^{1.89})_{\text{N}} \\ \text{Bond order: } C_{\text{BA}} N \quad 0.980 ; C_{\text{benzene}} N \quad 0.975 \end{array}$



 $\begin{array}{c} \text{Bond composition: } \sigma_{\text{CBAN}} = 0.629(\text{sp}^{2.64})_{\text{C}} + 0.778(\text{sp}^{1.92})_{\text{N}}, \\ \sigma_{\text{CbenzeneN}} = 0.621(\text{sp}^{2.75})_{\text{C}} + \\ 0.784(\text{sp}^{1.84})_{\text{N}} \\ \text{Bond order: } C_{\text{BA}}\text{N} \quad 0.980; \\ C_{\text{benzene}}\text{N} \quad 0.976 \\ \text{Figure S9 Bond composition and bond order of the two N-C bonds of compounds 2-5} \end{array}$