Electronic Supplementary Information

Thermal stability and self-reduction of a new red phosphor

NaMg(PO₃)₃:Mn²⁺

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Table. S1 Rietveld refined crystallographic parameters of the $NaMg_{1-x}(PO_3)_3:xMn^{2+}$ samples.

Formula	x = 0.0	x = 0.1	x = 0.2	x = 0.3	x = 0.4	x = 0.5
Crystal system	Orthorho	Orthorho	Orthorho	Orthorho	Orthorho	Orthorho
	mbic	mbic	mbic	mbic	mbic	mbic
Space group	Pbca	Pbca	Pbca	Pbca	Pbca	Pbca
<i>a</i> (Å)	14.267171	14.279147	14.286875	14.304394	14.335966	14.356815
<i>b</i> (Å)	14.221594	14.237005	14.247443	14.261115	14.289226	14.313797
<i>c</i> (Å)	14.224002	14.232888	14.250390	14.266706	14.289963	14.312203
$V(Å^3)$	2886.077	2893.436	2900.687	2910.360	2927.297	2941.166
$2\theta(^{\circ})$	5-75	5-75	5-75	5-75	5-75	5-75
Ζ	16	16	16	16	16	16
$R_{\rm p}$ (%)	7.48	8.80	8.45	8.28	8.12	8.84
$R_{\rm wp}$ (%)	11.85	14.60	13.28	11.96	11.61	12.55
χ^2	1.882	3.582	2.712	2.261	1.823	2.224



Figure S1. Rietveld refinements of the XRD files of $NaMg_{1-x}(PO_3)_3:xMn^{2+}$ (x = 0, 0.1, 0.2, 0.3 and 0.4).



Figure S2. Evolution of lattice parameters and the unit cell volumes of NaMg₁₋ $_x(PO_3)_3:xMn^{2+}$ (x = 0, 0.1, 0.2, 0.3, 0.4 and 0.5).



Figure S3. Experimental UV–Vis absorption spectrum of NMP ranging from 200 to 800 nm.



Figure S4. The PLE and PL spectra of NMP:0.2Mn²⁺ phosphor.



Figure S5. Excitation line of BaSO₄ and emission spectrum of NMP:0.2Mn²⁺ phosphor collected by using an integrating sphere.



Figure S6. The temperature-dependent FWHM of NMP:0.2Mn²⁺.