

# Influence of the Lewis basicity hardness of recrystallization solvents on the coordination sphere of the complex [Co(3,5-dinitrobenzoate-O,O')<sub>2</sub>]: Crystallographic and Theoretical Analysis.

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Table S1. Dielectric constants and dipole moments for water (H<sub>2</sub>O), methanol (CH<sub>3</sub>OH), ethanol (EtOH), dimethyl sulfoxide (DMSO), acetone (Act), and acetonitrile (ACN).

	<b>Dielectric constant*</b>	<b>Dipole moment (D)*</b>
<b>H<sub>2</sub>O</b>	80	1.85
<b>DMSO</b>	47	3.96
<b>ACN</b>	37	3.92
<b>CH<sub>3</sub>OH</b>	33	1.70
<b>EtOH</b>	30	1.69
<b>Act</b>	21	2.88

\* Li, C.-P.; Du, M. Role of solvents in coordination supramolecular systems. *Chem. Commun.* 2011, 47, 5958–5972. DOI: 10.1039/c1cc10935a

Table S2. Crystallographic data of complex [Co(DNB)<sub>2</sub>] crystallized in acetone (Act), acetonitrile (ACN), dimethyl sulfoxide (DMSO), and water H<sub>2</sub>O.

Crystal Data	Act	ACN	DMSO	H <sub>2</sub> O
Chemical Formula	C <sub>66</sub> H <sub>74</sub> Co <sub>3</sub> N <sub>12</sub> O <sub>48</sub>	C <sub>54</sub> H <sub>38</sub> Co <sub>3</sub> N <sub>18</sub> O <sub>37</sub>	C <sub>18</sub> H <sub>22</sub> CoN <sub>4</sub> O <sub>16</sub> S <sub>2</sub>	C <sub>14</sub> H <sub>22</sub> CoN <sub>4</sub> O <sub>20</sub>
M <sub>r</sub>	1980.16	1707.81	673.44	625.28
Solvent for Crystallization	<i>Acetone</i>	<i>Acetonitrile</i>	<i>dimethylsulfoxide</i>	<i>water</i>
Crystalline system, space group	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Monoclinic, <i>P</i> 2 <sub>1</sub> / <i>n</i>	Triclinic, <i>P</i> -1
<i>a</i> , <i>b</i> , <i>c</i> (Å)	12.7882 (10), 13.2551 (10), 14.4372 (11)	11.2634 (8), 12.6777 (9), 25.2927 (18)	10.3871 (9), 5.2909 (4), 22.6353 (17)	7.1199 (3), 11.6544 (5), 15.0334 (7)
<i>α</i> , <i>β</i> , <i>γ</i> (°)	67.940 (2), 79.147 (3), 69.028 (3)	99.217 (3), 102.517 (3), 100.460 (3)	90, 91.788 (3)	103.409 (2), 98.950 (2), 92.642 (2)
Volume, (Å <sup>3</sup> )	2113.7 (3)	3391.5 (4)	1243.36 (17)	1194.27 (9)
Z	1	2	2	2
Temperature, (K)	100(2)	100(2)	100(2)	100(2)
Radiation type	Mo K <sub>α</sub>	Mo K <sub>α</sub>	Mo K <sub>α</sub>	Mo K <sub>α</sub>
μ (mm <sup>-1</sup> )	0.69	0.84	0.95	0.82
<b>Data collection</b>				
Diffractometer	Bruker D8 Venture	Bruker D8 Venture	Bruker D8 Venture	Bruker D8 Venture
Absorption correction	Multi-Scan method (SADABS)	Multi-Scan method (SADABS)	Multi-Scan method (SADABS)	Multi-Scan method (SADABS)
No. of measured, independent and observed reflections [I>2σ(I)]	92160, 12892, 9316	168659, 20677, 16206	20489, 2845, 2536	42405, 5469, 5113
R <sub>int</sub>	0.060	0.049	0.048	0.037
(sin θ/λ) <sub>max</sub> (Å <sup>-1</sup> )	0.717	0.715	0.650	0.650
<b>Refinement</b>				
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )], wR(F <sup>2</sup> ), S	0.057, 0.164, 1.14	0.036, 0.102, 1.16	0.025, 0.066, 1.10	0.022, 0.058, 1.06
No. of reflections	12892	20677	2845	5469
Refined parameters	607	1026	189	352
H-atoms treatment	H atoms treated by a mixture of independent and constrained refinement	H atoms treated by a mixture of independent and constrained refinement	H-atom parameters constrained	H-atom parameters constrained
Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )	1.77, -0.94	0.54, -0.63	0.37, -0.42	0.38, -0.35

Table S3. Suggested names considering the nomenclature for compounds crystallized in water (H<sub>2</sub>O), methanol (CH<sub>3</sub>OH), ethanol: acetone (EtOH: Act), dimethyl sulfoxide (DMSO), acetone (Act), and acetonitrile (ACN)

<b>H<sub>2</sub>O</b>	Tetraaquabis(3,5-dinitrobenzoato- $\kappa^1O$ )cobalt(II) tetrahydrate.
<b>CH<sub>3</sub>OH</b>	Tetrakis(methanol)bis(3,5-dinitrobenzoato- $\kappa^1O$ )cobalt(II).
<b>EtOH: Act</b>	Tetrakis(ethanol)bis(3,5-dinitrobenzoato- $\kappa^1O$ )cobalt(II).
<b>DMSO</b>	Diaquabis(dimethylsulfoxide- $\kappa O$ )bis(3,5-dinitrobenzoato- $\kappa^1O$ )cobalt(II).
<b>ACN</b>	<p>1. Acetonitrileaqua(<math>\mu</math>-3,5-dinitrobenzoato-<math>1\kappa^2O, O':2\kappa^1O'</math>)bis(<math>\mu</math>-3,5-dinitrobenzoato-<math>1\kappa^1O:2\kappa^1O'</math>)-bis(<math>\mu</math>-3,5-dinitrobenzoato-<math>2\kappa^1O:3\kappa^1O'</math>)-acetonitrileaqua(<math>\mu</math>-3,5-dinitrobenzoato-<math>3\kappa^2O, O':2\kappa^1O'</math>)tricobalt(II)</p> <p>2. Diacetonitrile(<math>\mu</math>-3,5-dinitrobenzoato-<math>1\kappa^2O, O':2\kappa^1O'</math>)bis(<math>\mu</math>-3,5-dinitrobenzoato-<math>1\kappa^1O:2\kappa^1O'</math>)-bis(<math>\mu</math>-3,5-dinitrobenzoato-<math>2\kappa^1O:3\kappa^1O'</math>)-diacetonitrile (<math>\mu</math>-3,5-dinitrobenzoato-<math>3\kappa^2O, O':2\kappa^1O'</math>)tricobalt(II) acetonitrile trisolvate</p>

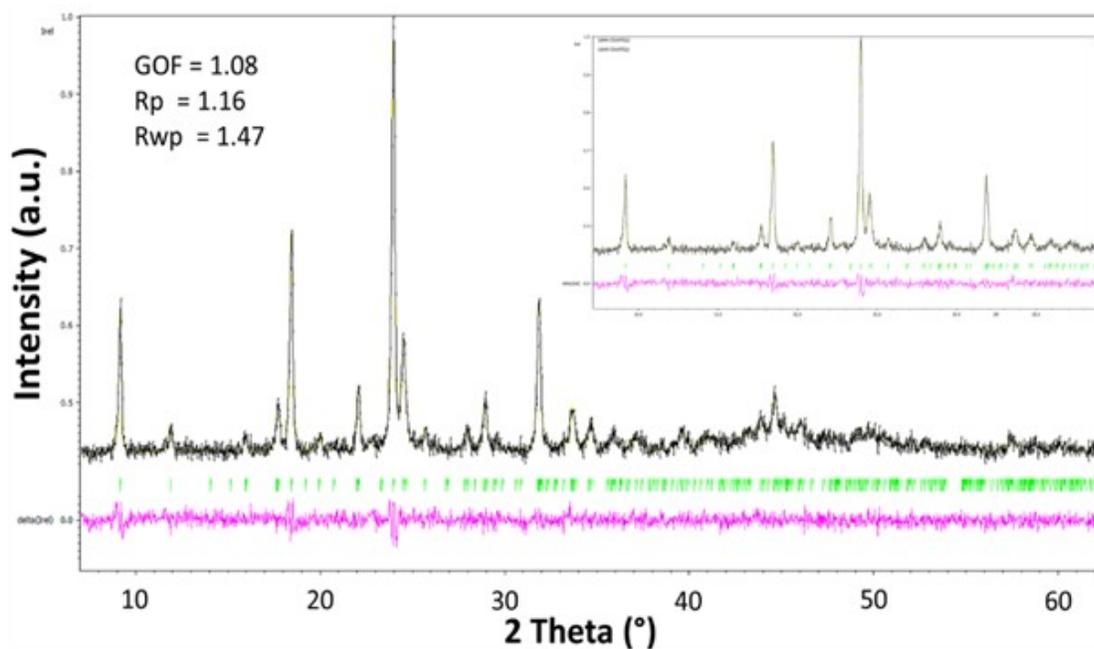


Figure S1. Graphical result of LeBail analysis of complex  $[\text{Co}(\text{DNB})_2]$ . Unit cell  $a = 9.6586$  (17) Å,  $b = 11.6717$  (14) Å,  $c = 6.3204$  (9) Å and  $\beta = 96.286^\circ$ , with most probably space group  $P2_1/m$ .

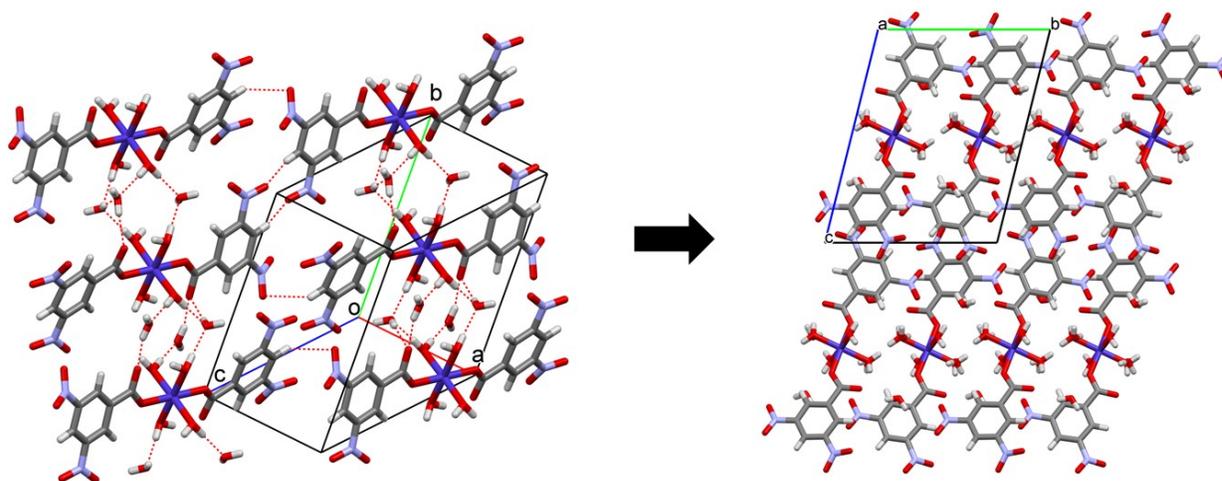


Figure S2. Packing of the complex  $[\text{Co}(\text{DNB})_2]$  recrystallized in  $\text{H}_2\text{O}$ .

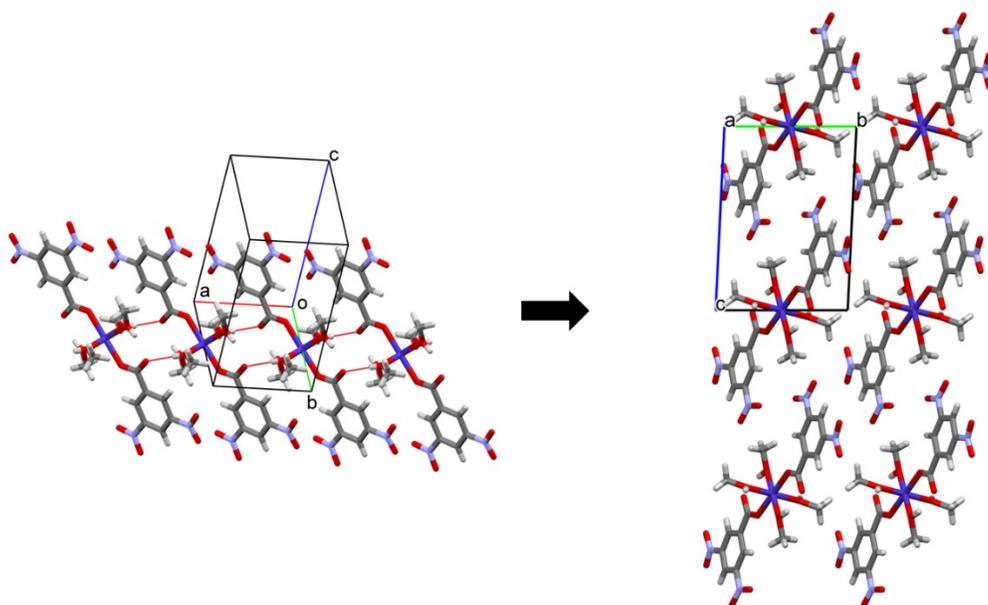


Figure S3. Packing of the complex  $[\text{Co}(\text{DNB})_2]$  recrystallized in MeOH.

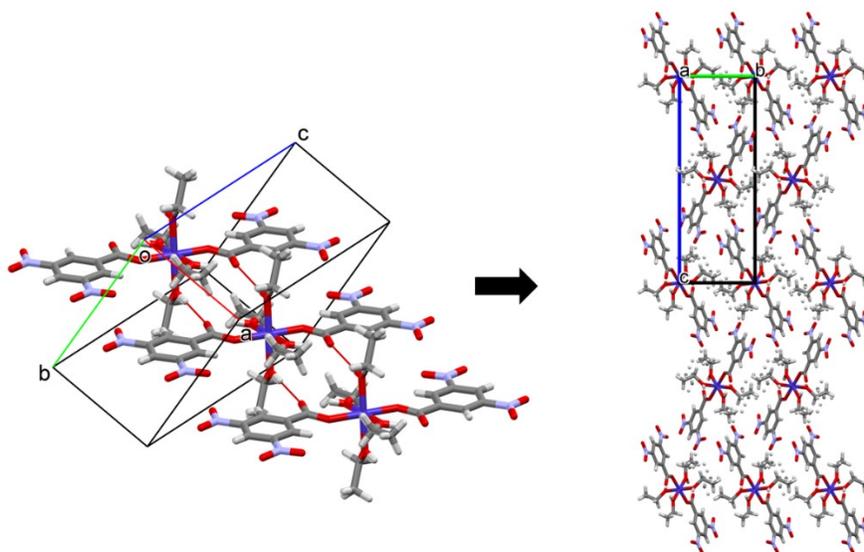


Figure S4. Packing of the complex  $[\text{Co}(\text{DNB})_2]$  recrystallized in EtOH.

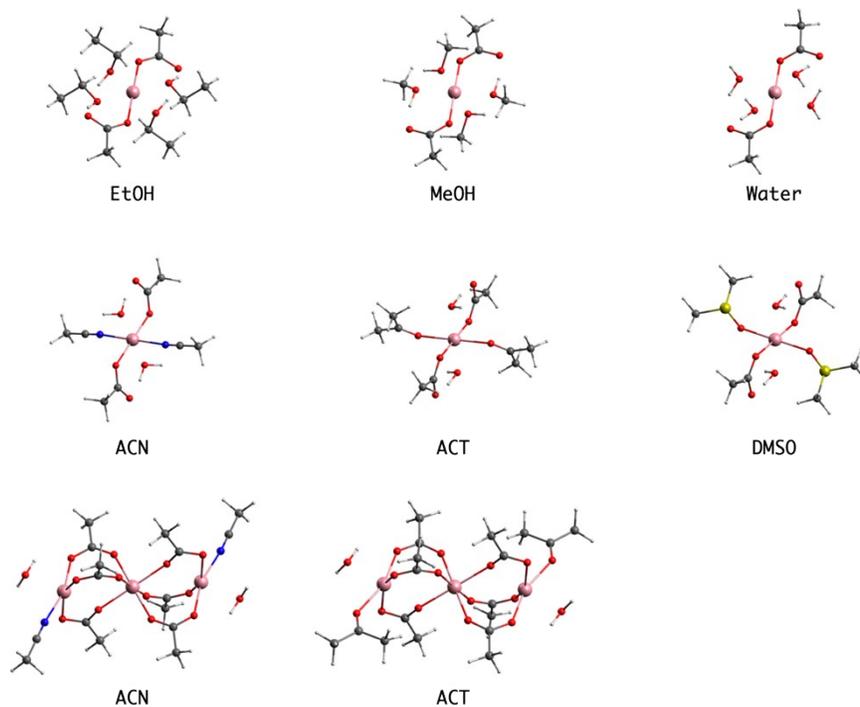


Figure S5. Optimized geometries of models of the  $[\text{Co}(\text{DBN})_2]$  complex recrystallized in the different solvents.

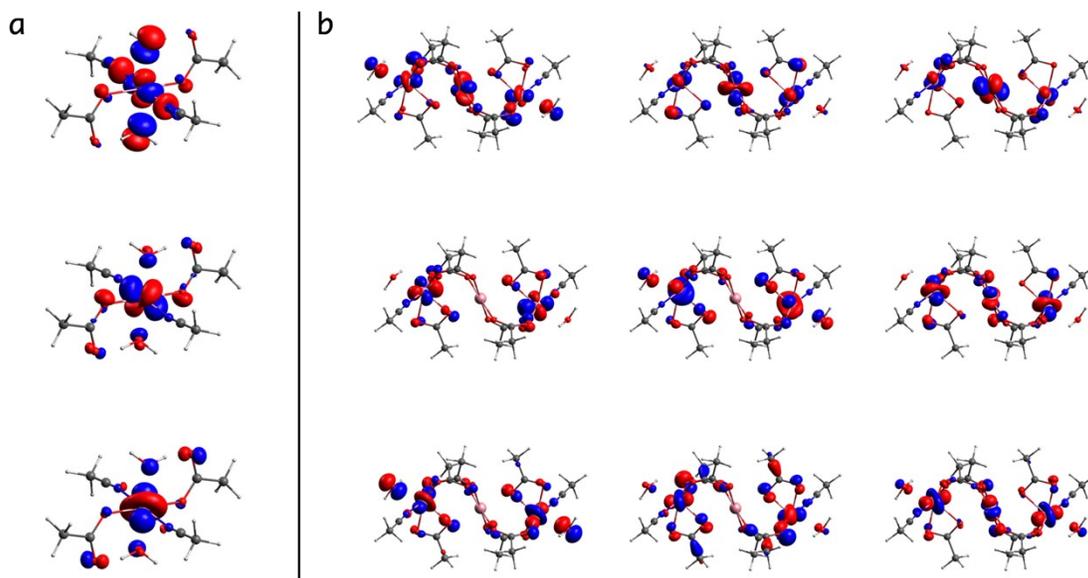


Figure S6. SOMOs of models of the  $[\text{Co}(\text{DBN})_2]$  complex recrystallized in ACN. a) Mononuclear with  $S=3/2$  ground state and b) Trinuclear with  $S=9/2$  ground state.

Table S4. Analysis of Potential Hydrogen Bonds for complex crystallized in H<sub>2</sub>O

=====  
 Analysis of Potential Hydrogen Bonds and Schemes with d(D...A) < R(D)+R(A)+0.50, d(H...A) < R(H)+R(A)-0.12 Ang., D-H...A > 100.0 Deg  
 Note: - ARU codes in [ ] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)  
 =====

Nr	Typ	Res	Donor	--- H...A	Acceptor [ ARU ]	D - H	H...A	D...A	D - H...A	A..H..A* A'..H..A"	Sum (XY, YZ)	Sum (XZ)
1		1	O3	--H3A	..O10 [ 1555.05]	0.82	1.88 2.6939(13)		179			
2		1	O3	--H3B	..07 [ 2766.02]	0.85	1.92 2.7703(12)		175			
3		1	O4	--H4A	..07 [ 1555.02]	0.82	1.92 2.7360(12)		179			
4	Intra	1	O4	--H4B	..011 [ ]	0.83	1.99 2.7535(13)		153			
5		1	O5	--H5A	..08 [ 1555.03]	0.84	1.91 2.7552(12)		176			
6		1	O5	--H5B	..09 [ 2676.04]	0.84	1.99 2.8201(13)		173			
7		1	O6	--H6A	..08 [ 2676.03]	0.82	1.96 2.7760(12)		169			
8	Intra	1	O6	--H6B	..021 [ ]	0.83	1.91 2.6915(12)		157			
9		2	O7	--H7A	..04 [ 2666.01]	0.80	2.58 3.1679(12)		131			
10		2	O7	--H7A	..05 [ 2666.01]	0.80	2.21 2.9241(12)		149'	75'	355	
11		2	O7	--H7B	..011 [ 2766.01]	0.82	2.06 2.8297(13)		157			
12		3	O8	--H8A	..09 [ 1555.04]	0.82	2.07 2.8784(13)		166			
13		3	O8	--H8B	..03 [ 1455.01]	0.81	2.20 2.9427(13)		151			
14		3	O8	--H8B	..06 [ 1455.01]	0.81	2.50 3.0954(12)		131'	75'	357	
15		4	O9	--H9A	..021 [ 1555.01]	0.81	1.93 2.7062(13)		160			
16		4	O9	--H9B	..01AA [ 2677.01]	0.82	2.55 3.2506(12)		144			
17		5	O10	--H10A	..013 [ 1556.01]	0.79	2.58 3.1001(13)		124			
18		5	O10	--H10A	..026 [ 2776.01]	0.79	2.46 2.9235(14)		119'	117'	360	
19		5	O10	--H10B	..011 [ 2766.01]	0.79	2.04 2.7648(13)		152			
20		1	C14	--H14	..025 [ 2775.01]	0.95	2.45 3.3714(15)		164			
21		1	C24	--H24	..014 [ 1456.01]	0.95	2.54 3.3232(15)		139			

Translation of ARU-Code to CIF and Equivalent Position Code

=====  
 [ 1456.] = [ 1\_456] = -1+x, y, 1+z  
 [ 1556.] = [ 1\_556] = x, y, 1+z  
 [ 2677.] = [ 2\_677] = 1-x, 2-y, 2-z  
 [ 2766.] = [ 2\_766] = 2-x, 1-y, 1-z  
 [ 2666.] = [ 2\_666] = 1-x, 1-y, 1-z  
 [ 2676.] = [ 2\_676] = 1-x, 2-y, 1-z  
 [ 2776.] = [ 2\_776] = 2-x, 2-y, 1-z  
 [ 2775.] = [ 2\_775] = 2-x, 2-y, -z  
 [ 1455.] = [ 1\_455] = -1+x, y, z

For C--H...Acceptor Interactions See: Th. Steiner, Cryst. Rev, (1996), 6, 1-57

Table S5. Analysis of Potential Hydrogen Bonds for complex crystallized in MeOH

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=====
Analysis of Potential Hydrogen Bonds and Schemes with d(D...A) < R(D)+R(A)+0.50, d(H...A) < R(H)+R(A)-0.12 Ang., D-H...A > 100.0 Deg
=====
Note: - ARU codes in [] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)
=====

```

Nr	Typ	Res	Donor	---	H....Acceptor	[ ARU ]	D - H	H...A	D...A	D - H...A	A..H...A* A'..H...A"	Sum (XY, YZ)	Sum (XZ)
1		1	O(7)	--H(7D)	..O(2)	[ 2665.01]	0.77(3)	1.90(3)	2.663(2)	178(4)			
2	Intra	1	O(8)	--H(8D)	..O(2)	[ ]	0.80(3)	1.89(2)	2.6498(19)	158(2)			

Translation of ARU-Code to CIF and Equivalent Position Code

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=====
[ 2665.] = [ 2_665] =1-x,1-y,-z

```

Table S6. Analysis of Potential Hydrogen Bonds for complex crystallized in EtOH

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=====
Analysis of Potential Hydrogen Bonds and Schemes with d(D...A) < R(D)+R(A)+0.50, d(H...A) < R(H)+R(A)-0.12 Ang., D-H...A > 100.0 Deg
=====
Note: - ARU codes in [] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)
=====

```

Nr	Typ	Res	Donor	---	H....Acceptor	[ ARU ]	D - H	H...A	D...A	D - H...A	A..H...A* A'..H...A"	Sum (XY, YZ)	Sum (XZ)
1		1	O7	--H7	..O2	[ 1655.01]	0.82(2)	1.90(2)	2.713(3)	169(3)			
2	Intra	1	O8	--H8	..O2	[ ]	0.81(3)	1.85(3)	2.632(3)	162(3)			

Translation of ARU-Code to CIF and Equivalent Position Code

```

=====
[ 1655.] = [ 1_655] =1+x,y,z

```

Table S7. Analysis of Potential Hydrogen Bonds for complex crystallized in DMSO

=====  
 Analysis of Potential Hydrogen Bonds and Schemes with  $d(D...A) < R(D)+R(A)+0.50$ ,  $d(H...A) < R(H)+R(A)-0.12$  Ang.,  $D-H...A > 100.0$  Deg  
 =====

Note: - ARU codes in [ ] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)  
 =====

Nr	Typ	Res	Donor	--- H...Acceptor	[ ARU ]	D - H	H...A	D...A	D - H...A	A..H...A* A'..H...A"	Sum (XY, YZ)	Sum (XZ)
1	Intra	1	O(3)	--H(3A)	..O(11)	[ 3676.01]	0.83	1.84	2.6311(16)	159		
2		1	O(3)	--H(3B)	..O(2)	[ 1545.01]	0.81	2.27	2.9455(15)	141		
3		1	O(3)	--H(3B)	..O(1)	[ 3666.01]	0.81	2.58	3.2683(15)	144'	72'	357
4		1	C(21)	--H(21C)	..O(15)	[ 4464.01]	0.98	2.41	3.215(2)	139		
5		1	C(22)	--H(22A)	..O(14)	[ 2566.01]	0.98	2.47	3.290(2)	141		
6		1	C(22)	--H(22B)	..O(15)	[ 2556.01]	0.98	2.52	3.208(2)	127		
7		1	C(22)	--H(22C)	..O(3)	[ 1565.01]	0.98	2.49	3.318(2)	142		

Translation of ARU-Code to CIF and Equivalent Position Code  
 =====

[ 3676.] = [ 3\_676] =1-x,2-y,1-z  
 [ 1565.] = [ 1\_565] =x,1+y,z  
 [ 3666.] = [ 3\_666] =1-x,1-y,1-z  
 [ 1545.] = [ 1\_545] =x,-1+y,z  
 [ 2556.] = [ 2\_556] =1/2-x,1/2+y,3/2-z  
 [ 4464.] = [ 4\_575] =-1/2+x,3/2-y,-1/2+z  
 [ 2566.] = [ 2\_566] =1/2-x,3/2+y,3/2-z

For C--H...Acceptor Interactions See: Th. Steiner, Cryst. Rev, (1996), 6, 1-57

Table S8. Analysis of Potential Hydrogen Bonds for complex crystallized in Acetone

=====  
 Analysis of Potential Hydrogen Bonds and Schemes with  $d(D...A) < R(D)+R(A)+0.50$ ,  $d(H...A) < R(H)+R(A)-0.12$  Ang.,  $D-H...A > 100.0$  Deg  
 =====

Note: - ARU codes in [ ] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)  
 =====

Nr	Typ	Res	Donor	--- H...Acceptor	[ ARU ]	D - H	H...A	D...A	D - H...A	A..H...A* A'..H...A"	Sum(XY,YZ)	Sum(XZ)
1		1	O12	--H12A	..O17	[ 2666.03]	0.87(5)	1.85(5)	2.710(3)	168(6)		
2		1	O12	--H12B	..O18	[ 2666.04]	0.82(6)	1.92(6)	2.702(4)	159(5)		
3		1	O13	--H13A	..O16	[ 2566.02]	0.80(5)	1.91(5)	2.709(4)	170(4)		
4		1	O13	--H13B	..O19	[ 1555.05]	0.85(5)	1.93(6)	2.767(3)	169(5)		
5	Intra	1	C4	--H4	..O11	[ 2666.01]	0.95	2.40	3.306(4)	159		
6		1	C7	--H7	..O1	[ 2657.01]	0.95	2.55	3.446(4)	158		
7		1	C11	--H11	..O16	[ 1555.02]	0.95	2.51	3.423(4)	162		
8		1	C16	--H16	..O5	[ 1554.01]	0.95	2.56	3.502(4)	174		
9		2	C30	--H30A	..O4	[ 2566.01]	0.98	2.51	3.386(5)	148		
10		2	C30	--H30C	..O7	[ 1555.01]	0.98	2.50	3.460(5)	165		
11		3	C33	--H33A	..O19	[ 1655.05]	0.98	2.56	3.460(5)	153		
12		4	C36	--H36B	..O6	[ 2656.01]	0.98	2.59	3.338(7)	133		

Translation of ARU-Code to CIF and Equivalent Position Code  
 =====

[ 2666.] = [ 2\_666] =1-x,1-y,1-z  
 [ 2566.] = [ 2\_566] =-x,1-y,1-z  
 [ 2657.] = [ 2\_657] =1-x,-y,2-z  
 [ 2656.] = [ 2\_656] =1-x,-y,1-z  
 [ 1554.] = [ 1\_554] =x,y,-1+z  
 [ 1655.] = [ 1\_655] =1+x,y,z

For C--H...Acceptor Interactions See: Th. Steiner, Cryst. Rev, (1996), 6, 1-57

Table S9. Analysis of Potential Hydrogen Bonds for complex crystallized in Acetonitrile

=====  
 Analysis of Potential Hydrogen Bonds and Schemes with  $d(D...A) < R(D)+R(A)+0.50$ ,  $d(H...A) < R(H)+R(A)-0.12$  Ang.,  $D-H...A > 100.0$  Deg  
 =====

Note: - ARU codes in [ ] are with reference to the Coordinates printed above (Possibly transformed, when MOVE .NE. 1.555)  
 =====

Nr	Typ	Res	Donor	--- H...Acceptor	[ ARU ]	D - H	H...A	D...A	D - H...A	A..H...A* A'..H...A"	Sum (XY, YZ)	Sum (XZ)
1		2	O7AA	--H7AA	..N15	[ 2565.04]	0.86 (4)	2.10 (4)	2.949 (2)			
2		2	O7AA	--H7AB	..O22	[ 2565.01]	0.82 (3)	2.56 (3)	2.8959 (18)			
3		2	O7AA	--H7AB	..N14	[ 2565.03]	0.82 (3)	2.29 (3)	3.067 (3)	158 (4) ' 84.0 (10) '		348 (5)
4		2	C3	--H3A	..O2	[ 2675.02]	0.98	2.47	3.302 (3)			142
5		2	C3	--H3C	..O12	[ 1665.02]	0.98	2.57	3.369 (3)			139
6	Intra	2	C20	--H20	..O1	[ 2565.02]	0.95	2.55	3.448 (2)			159
7		1	C32	--H32B	..O22	[ 1665.01]	0.98	2.40	3.263 (3)			146
8		1	C32	--H32C	..O13	[ 2786.01]	0.98	2.43	3.342 (3)			155
9	Intra	1	C35	--H35	..O24	[ 2676.01]	0.95	2.53	3.4422 (19)			161
10		4	C64	--H64B	..O20	[ 1445.01]	0.98	2.39	3.368 (2)			177
11		5	C65	--H65B	..O10	[ 2665.02]	0.98	2.43	3.378 (3)			164

Translation of ARU-Code to CIF and Equivalent Position Code  
 =====

[ 2565.] = [ 2\_565] = -x, 1-y, -z  
 [ 2676.] = [ 2\_676] = 1-x, 2-y, 1-z  
 [ 1445.] = [ 1\_445] = -1+x, -1+y, z  
 [ 2786.] = [ 2\_786] = 2-x, 3-y, 1-z  
 [ 1665.] = [ 1\_665] = 1+x, 1+y, z  
 [ 2675.] = [ 2\_675] = 1-x, 2-y, -z  
 [ 2665.] = [ 2\_665] = 1-x, 1-y, -z

For C--H...Acceptor Interactions See: Th. Steiner, Cryst. Rev, (1996), 6, 1-57