

# Static discrete disorder in the crystal structure of Iododiflunisal: On the Importance of Hydrogen Bond, Halogen Bond and $\pi$ -Stacking Interactions

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## Electronic Supplementary Information

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# 1. Crystal data and structure refinement

## 1.1 Iododiflunisal (IDIF) (mo\_023WB108\_0m\_a)

Table S1. Crystal data and structure refinement for mo\_023WB108\_0m\_a.

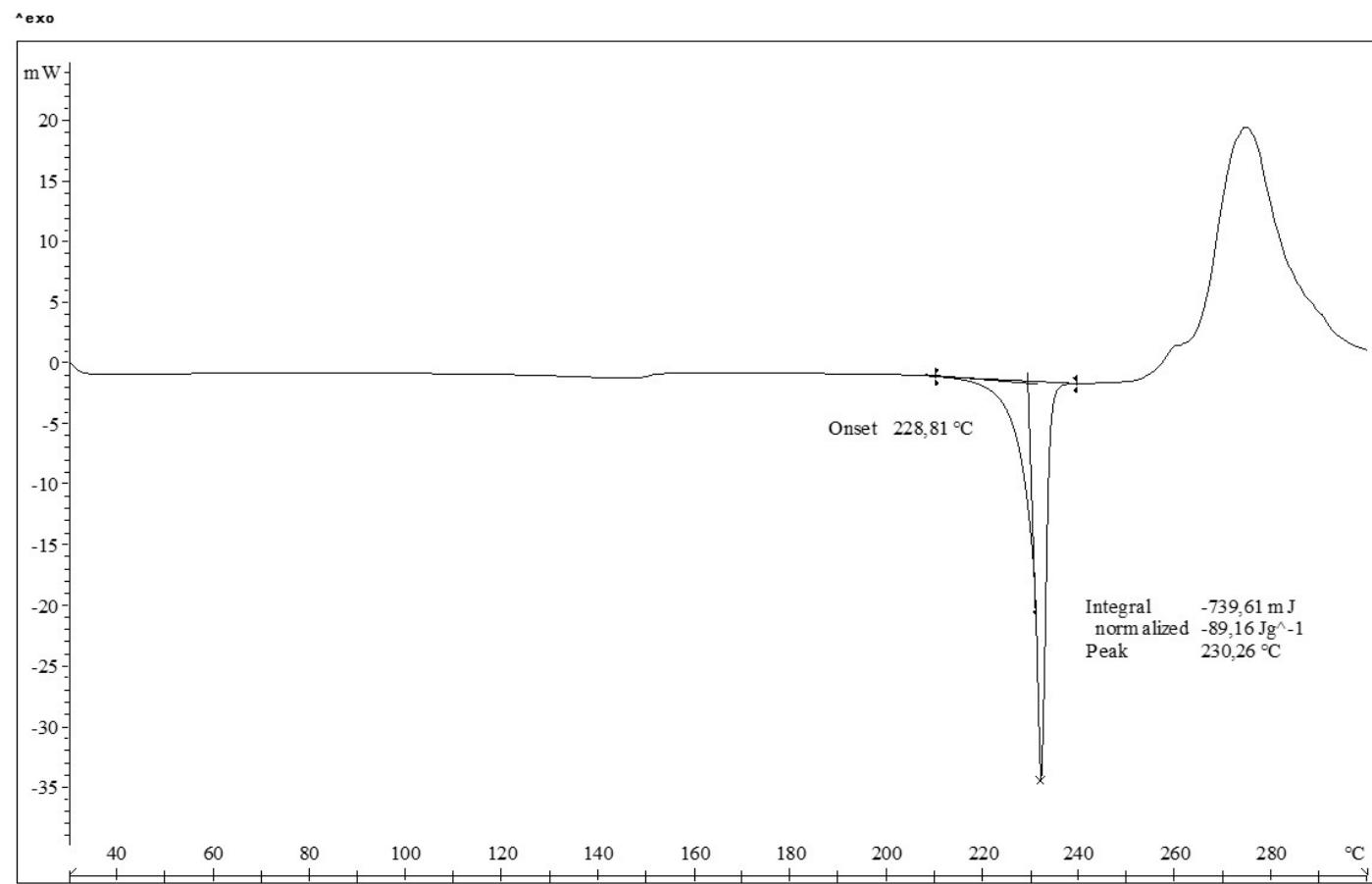
Identification code	mo_023WB108_0m_a		
Empirical formula	$C_{13}H_7F_2IO_3$		
Formula weight	376.09		
Temperature	296(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	$P\bar{2}_1/n$		
Unit cell dimensions	$a = 16.9777(10)$ Å	$\alpha = 90^\circ$ .	
	$b = 4.0424(2)$ Å	$\beta = 93.412(2)^\circ$ .	
	$c = 18.0441(11)$ Å	$\gamma = 90^\circ$ .	
Volume	$1236.18(12)$ Å <sup>3</sup>		
Z	4		
Density (calculated)	2.021 Mg/m <sup>3</sup>		
Absorption coefficient	2.616 mm <sup>-1</sup>		
F(000)	720		
Crystal size	$0.348 \times 0.196 \times 0.078$ mm <sup>3</sup>		
Theta range for data collection	2.404 to 30.592°.		
Index ranges	$-24 \leq h \leq 24, -5 \leq k \leq 5, -25 \leq l \leq 25$		
Reflections collected	26273		
Independent reflections	3717 [R(int) = 0.0638]		
Completeness to theta = 25.242°	96.6 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7461 and 0.5695		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data / restraints / parameters	3717 / 0 / 189		
Goodness-of-fit on F <sup>2</sup>	1.080		
Final R indices [I>2sigma(I)]	R1 = 0.0352, wR2 = 0.0908		
R indices (all data)	R1 = 0.0394, wR2 = 0.0956		
Extinction coefficient	n/a		
Largest diff. peak and hole	1.014 and -1.418 e.Å <sup>-3</sup>		

Table S2. Hydrogen bonds for mo\_023WB108\_0m\_a [Å and °].

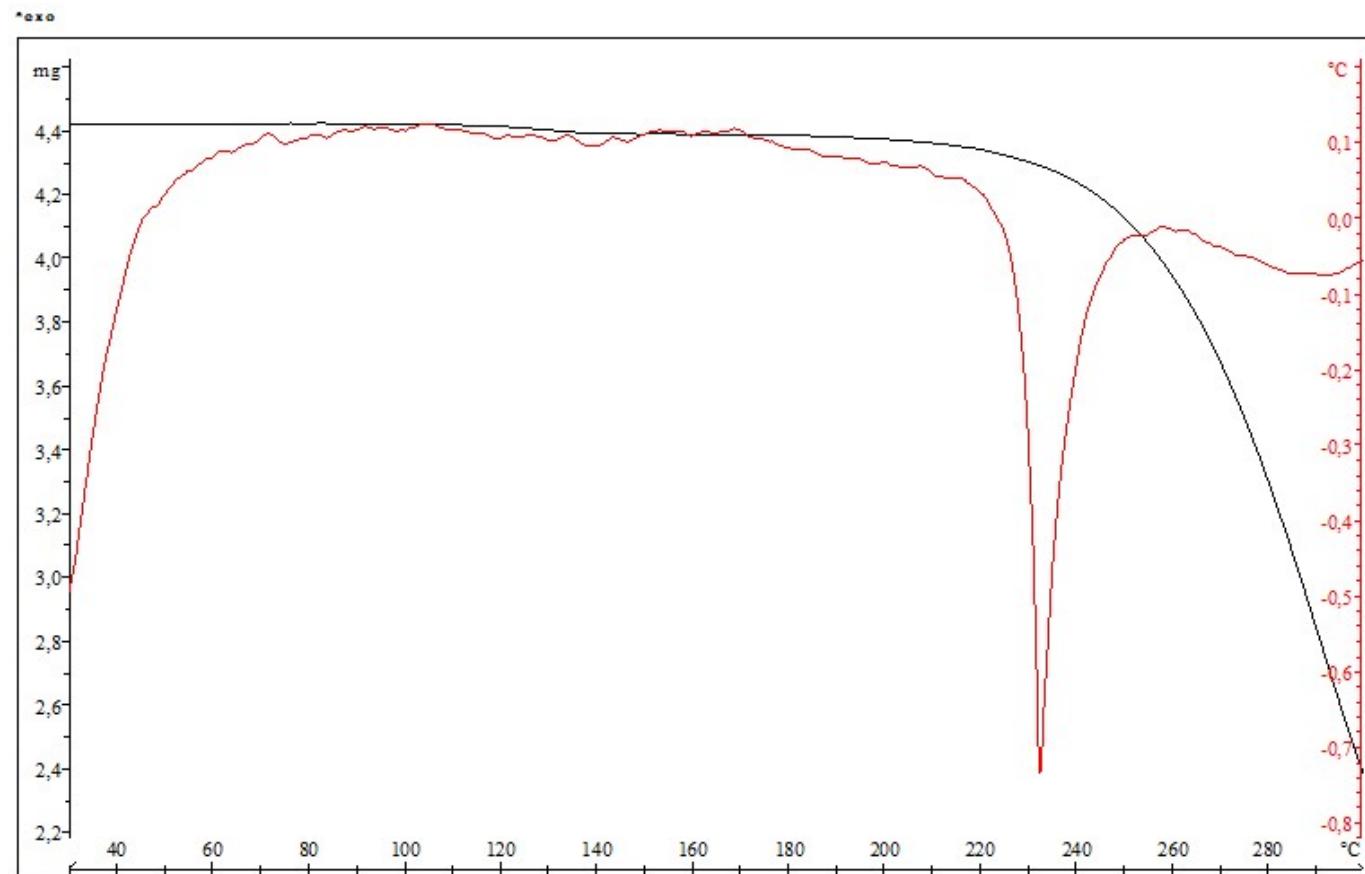
Donor --- H....Acceptor	[ARU]	d( D - H)	d( H...A)	d(D...A)	$\angle(D - H...A)$
O(2) ---H(2O) ..O(1)	[1-x,1-y,-z]	0.68(6)	1.99(5)	2.660(3)	175(8)
Intra O(3) ---H(3O) ..O(1)	[x,y,z]	0.76(4)	1.91(4)	2.619(3)	154(4)

## 2.- Characterization of iododiflunisal bulk powder

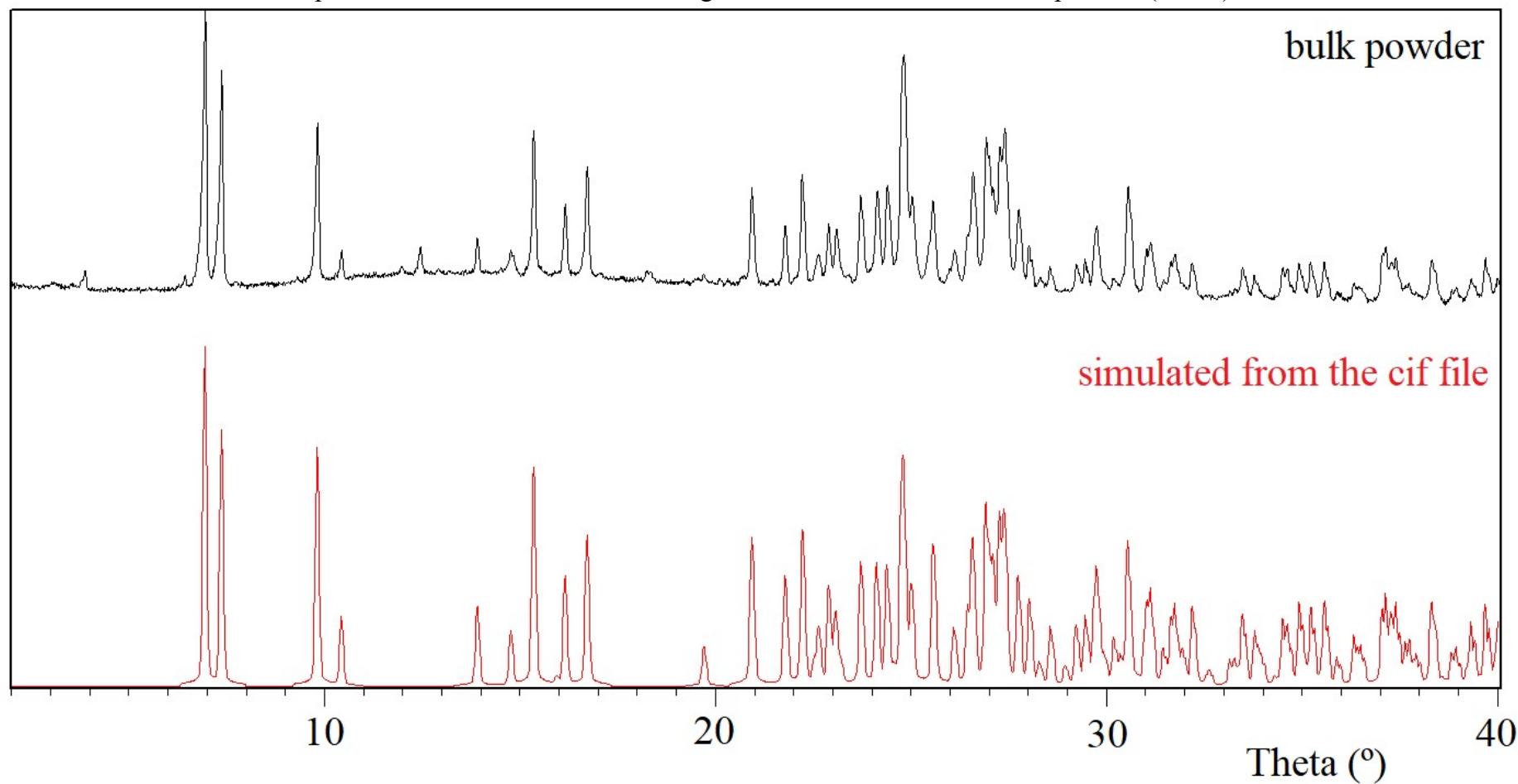
**Figure S1: Differential Scanning Calorimetry (DSC) of IDIF analysis was carried out by means of a Mettler-Toledo DSC-822e calorimeter. Experimental conditions: aluminum crucibles of 40 µL volume, atmosphere of dry nitrogen with 50 mL/min flow rate, heating rate of 10 °C/min. The calorimeter was calibrated with indium of 99.99% purity (m.p.: 156.4 °C, ΔH: 28.67 J/g).**



**Figure S2: Thermogravimetric Analysis (TGA) of IDIF** was performed on a Mettler-Toledo TGA-851e thermobalance. Experimental conditions: alumina crucibles of 70  $\mu\text{L}$  volume, atmosphere of dry nitrogen with 50 mL/min flow rate, heating rate of 10  $^{\circ}\text{C}/\text{min}$ . SDTA signal is represented in red.

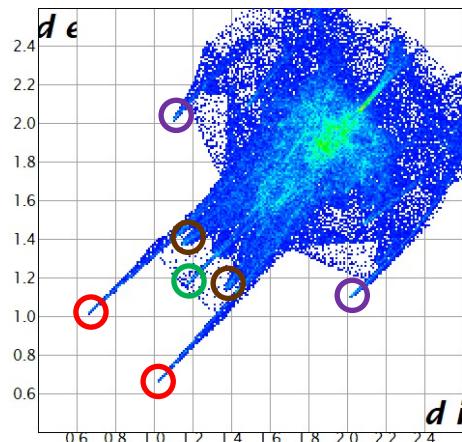


**Figure S3: Comparative PXRD diffractograms between bulk IDIF (black) and simulated from the cif file (red).** Powder X-ray diffraction (PXRD) pattern was obtained on a PANalytical X’Pert PRO MPD diffractometer in transmission configuration using Cu K $\alpha$ 1+2 radiation ( $\lambda = 1.5406 \text{ \AA}$ ) with a focusing elliptic mirror and a PIXcel detector working at a maximum detector’s active length of 3.347°. Configuration of convergent beam with a focalizing mirror and a transmission geometry with flat sample sandwiched between low absorbing films measuring from 2 to 40° in  $2\theta$ , with a step size of 0.026° and a total measuring time of 30 minutes at room temperature (298 K).

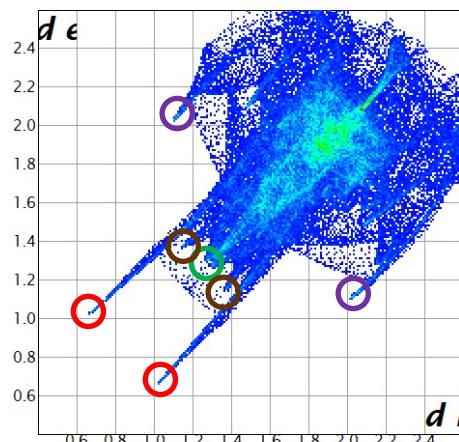


### 3.- Hirshfeld analysis

Hirshfeld surfaces fingerprint plots of two different asymmetric units of IDIF: comparative footprint and contribution (%) of intermolecular contacts



IDIF\_Syn



IDIF\_Anti

Table S3. Contribution (%) of intermolecular contacts of two different asymmetric units of IDIF at 296 K

Contacts	IDIF_Syn	IDIF_Anti
<b>H.....H</b>	<b>16.3</b>	<b>19.5</b>
<b>F.....H</b>	<b>15.9</b>	<b>12.1</b>
<b>O.....H</b>	<b>12.2</b>	<b>12.2</b>
<b>C.....H</b>	<b>10.1</b>	<b>7.9</b>
<b>I.....H</b>	<b>8.5</b>	<b>8.5</b>
<b>F.....F</b>	<b>2.7</b>	<b>3.4</b>
<b>Residual</b>	<b>34.3</b>	<b>36.4</b>

#### 4.- Crystal structures of diflunisal reported in the CCDC

Table S4.

CCDC refcode	Space group	Solid form	Temperature (K)	Disorder (in F ortho position)
FAFWIS	$C\ 2/c$	anhydrous (polymorph V)	283-303	Yes
FAFWIS01	$P\ -1$	anhydrous (polymorph I)	283-303	Yes
FAFWIS02	$P\ 2_1\ 2_1\ 2_1$	anhydrous (polymorph III)	298	No <sup>1</sup>
NUZGUM	$C\ 2/c$	pTHF inclusion complex	100	Yes
NUZHAT	$P\ 2_1/n$	caprolactone inclusion complex	100	Yes
OPOGAD	$P\ 2_1/c$	theophylline cocrystal	150	No
QOQXAV	$C\ 2/c$	monohydrate clathrate	283-303	Yes
RUXRUX	$C\ 2/c$	chloroform solvate	153	Yes
RUXSAE	$P\ 2_1/n$	acetic acid solvate	150	Yes
UWOKEY	$P\ -1$	1,2-bis(Pyridinium-4-yl)ethene salt	150	No
UWOKIC	$P\ -1$	acetonitrile solvate	283-303	No
UWOKOI	$P\ 2_1/c$	pyrazine cocrystal	150	No
YEJWEP	$C\ 2/c$	1,3-bis(Pyridinium-4-yl)propane salt	283-303	Yes
		hexane solvate		

<sup>1</sup> Crystal structure solved from PXRD data, no F atom disorder was considered