**Supplementary Materials** 

## Structural Transformations in Cobalt(II) Coordination Polymers Constructed from Flexible *N*,*N*'-Bis(3-pyridylmethyl)sebacoamide and benzene-1,3,5-tricarboxylic acid

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Fig. S1. (a) Simulated and (b) experimental PXRD patterns of 1.



Fig. S2. (a) Simulated and (b) experimental PXRD patterns of 2.



Fig. S3. (a) Simulated and (b) experimental PXRD patterns of 3.



Fig. S4. (a) Simulated and (b) experimental PXRD patterns of 4.



Complex	1	2	3	4
<sup>a</sup> N-HO			2.04 (176), 2.10 (177)	
<sup>b</sup> N-HO	2.46 (134)	2.07 - 2.42 (132 - 166)		
°N-HO	2.03 (169)			
dO-HO	1.82 (161)	1.75 (172), 1.89 (167)		
°O-HO	2.14 (160), 2.24 (145)	1.87 – 2.23 (120 – 169)		2.25 (136)
<sup>f</sup> O-HO		2.31 (179), 2.46 (179)		
<sup>g</sup> O-HO		2.28 (155), 2.53 (162)		2.60 (118)
<sup>h</sup> O-HO		1.91 - 2.27 (134 - 178)		
<sup>i</sup> O-HO		1.79 – 2.51 (111 – 165)		1.87 – 2.64 (118 - 173)
jO-HO		1.82 - 2.04 (171 - 177)		2.11 (129)
<sup>k</sup> O-HO			1.77 - 2.33 (113 - 165)	
<sup>l</sup> O-HO			1.75 (155), 1.95 (130)	
<sup>m</sup> O-HO			1.97 (139)	
<sup>n</sup> O-HO			2.26 (117)	
°O-HO			2.44 (111)	
<sup>p</sup> O-HO			2.11 (136)	
qO-HO				1.88 (156)
<sup>r</sup> N-HO				2.40 (146)

**Table S1.** H---O distances (Å) of hydrogen bonds with angles (°) in parenthesis for 1 - 4.

a from the amine hydrogen atoms of L to the carboxylate oxygen atoms.

b from the amine hydrogen atoms of L to the cocrystallized water oxygen atoms.

c from the amine hydrogen atoms of L to the amide oxygen atoms.

d from the carboxylate hydrogen atoms of 1,3,5-HBTC<sup>2-</sup> to the cocrystallized water molecules.

e from the cocrystallized water molecules to the carboxylate oxygen atoms.

f from the cocrystallized water molecules to the amide oxygen atoms.

g from the cocrystallized water molecules to the coordinated water molecules.

h from the cocrystallized water molecules to the cocrystallized water molecules

i from the coordinated water molecules to the carboxylate oxygen atoms.

j from the coordinated water molecules to the amide oxygen atoms.

k from the coordinated methanol molecules to the cocrystallized methanol molecules.

l from the coordinated methanol molecules to the carboxylate oxygen molecules.

m from the cocrystallized methanol molecules to the carboxylate oxygen molecules.

n from the cocrystallized methanol molecules to the amide oxygen molecules.

o from the cocrystallized methanol molecules to coordinated methanol oxygen atoms.

p from the cocrystallized methanol molecules to cocrystallized methanol oxygen atoms.

q from the coordinated water molecules to the cocrystallized water molecules.

r from the amine hydrogen atoms of L to the coordinated methanol molecules.

Fig. S5. Packing diagram for 1.



Fig. S6. Packing diagram for 2.



Fig. S7. Packing diagram for 3.



Fig. S8. Packing diagram for 4.



Fig. S9. Drawings showing the (a) GAAAAAG *trans anti-anti*, (b) AAAAAAA *trans syn-syn*, (c) AGAAAAA *cis anti-syn* conformations and (d) GAAAAAG *trans anti-anti* for the L ligands in 1 - 4, respectively.



(d)

Fig. S10. The TGA curve for complex 1.



Fig. S11. The TGA curve for complex 2.



Fig. S12. The TGA curve for complex 3'.



Fig. S13. The TGA curve for complex 4.



Fig. S14. PXRD patterns for transformation from 3 to 3'. (a) simulation of 3, (b) 3 kept in  $CH_3OH$  then exposed to air for (c) 8 min, (d) 16 min, (e) 24 min and (f) 3 h.



Figure S15. PXRD patterns of (a) as-synthesized of 3', (b) complex 3' under vacuum for 20 minutes and (c) then immersed in MeOH for 1 h and (d) simulation of 3.

