# A reversible pressure-induced bond rearrangement of flexible lanthanide 2,5-bis(allyloxy)terephthalate coordination polymer networks.

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### **Supplementary Information**

### Synthesis of 2,5-bis(allyloxy)terephthalic acid



### Synthesis of diethyl 2,5-dihydroxyterephthalate

To a solution of 2,5-dihydroxyterephalic acid (1 g, 5 mmol) in ethanol (20 mL) 5-10 drops of c.  $H_2SO_4$  were added carefully. The solution was stirred and heated under reflux for 12h. When cooling the solution, a yellow solid precipitated, that was collected by filtration and washed with ethanol to afford the title compound as a bright yellow solid (1.1 g, 88 %);  $\delta_H$  (300 MHz, CDCl<sub>3</sub>) 10.14 (2H, s, Ar-OH), 7.51 (2H, s, Ar-H), 4.43 (4H, q, J 7, Ar-CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); all data agrees with that given in the literature.<sup>1</sup>

### Synthesis of diethyl 2,5-bis(allyloxy)terephthalate

Allyl bromide (3 mL, 18 mmol), diethyl 2,5-dihydroxyterephthalate (0.5 g, 2 mmol) and potassium carbonate (1 g, 8 mmol) were added to acetonitrile (10 mL) and refluxed for 12h under nitrogen. Excess solvent was removed under vacuum and the remaining residue diluted with water before extraction into ethyl acetate. Organics were dried with MgSO<sub>4</sub> and concentrated to afford a crude product as pale orange waxy residue. The crude product was passed through a silica plug with *n*-hexane:ethyl acetate [3:1] before being concentrated under vacuum to afford the title compound as a yellow waxy solid (420 mg, 57 %);  $\delta_{\rm H}$  (300 MHz, CDCl<sub>3</sub>) 7.36 (2H, s, Ar-*H*), 6.12-6.00 (2H, m, Ar-OCH<sub>2</sub>CH), 5.52-5.44 (2H, m, Ar-OCH<sub>2</sub>CH=CHH), 5.32-5.27 (2H, m, Ar-OCH<sub>2</sub>CH=CHH), 4.60-4.58 (4H, m, ArOCH<sub>2</sub>), 4.41-4.34 (4H, q, J 7, Ar-CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.41-1.36 (6H, t, J 7, Ar-CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); all data agrees with that given in the literature.<sup>1</sup>

### Synthesis of 2,5-bis(allyloxy)terephthalic acid

Three equivalents of LiOH were added to diethyl 2,5-bis(allyloxy)terephthalate (0.4 g, 1.2 mmol) in equal parts THF:H<sub>2</sub>O (10 mL) and stirred overnight at room temperature. Solvents were removed by rotary evaporation and the remaining residue diluted with H<sub>2</sub>O. The solution was neutralised using 1M HCl to a pH of ~ 2 to precipitate the product. The precipitate was filtered to afford the title compound as a white solid (0.32 g, 97 %);  $\delta_{\rm H}$  (300 MHz, DMSO-d<sub>6</sub>) 13.00 (2H, bs, CO<sub>2</sub>H), 7.30 (2H, s, Ar-*H*), 6.08-5.95 (2H, m, Ar-OCH<sub>2</sub>CH), 5.47-5.40 (2H, m, Ar-OCH<sub>2</sub>CH=CHH), 5.27-5.22 (2H, m, Ar-OCH<sub>2</sub>CH=CHH), 4.61-4.59 (4H, m, ArOCH<sub>2</sub>); all data agrees with that given in the literature.<sup>1</sup>

### <sup>1</sup>H NMR Spectra of synthesised compounds









### **Crystal synthesis**

20 mg of 2,5-bis(allyloxy)terephthalic acid and 1.2 equivalent of lanthanide nitrate was added to 2 mL of diethylformamide (DEF) and heated in a sealed vial for 3 days at 80 °C, crystals were subsequently washed with fresh DEF and stored in solvent.

 $Ce_2(L)_2(DEF)_2(NO_3)_2$  (1): Complex 1 was synthesised following the general procedure above using  $Ce(NO_3)_3$ .xH<sub>2</sub>O yielding colourless plate crystals after 3 days.  $Ce_2C_{44}H_{46}O_{20}N_4$  Mw = 1159.02 g/mol. Selected FTIR: 2976 (w) 1602 (m) 1418 (m) 1285 (m) 1208 (m).

 $Pr_2(L)_4(DEF)_2(NO_3)_2$  (2): Complex 2 was synthesised following the general procedure above using  $Pr(NO_3)_3.xH_2O$  yielding pale yellow plate crystals after 3 days.  $Pr_2C_{44}H_{46}O_{20}N_4$  Mw = 1160.60 g/mol. Selected FTIR: 2975 (w) 1600 (m) 1418 (s) 1287 (m) 1208 (m).

 $Nd_2(L)_4(DEF)_2(NO_3)_2$  (3): Complex 3 was synthesised following the general procedure above using  $Nd(NO_3)_3.xH_2O$  yielding pale pink plate crystals after 3 days.  $Nd_2C_{44}H_{46}O_{20}N_4$  Mw = 1167.26 g/mol. Selected FTIR: 2981 (w) 1603 (m) 1418 (m) 1292 (m) 1211 (m).

 $Eu_2(L)_4(DEF)_2(NO_3)_2$  (4): Complex 4 was synthesised following the general procedure above using  $Eu(NO_3)_3.xH_2O$  yielding colourless plate crystals after 3 days.  $Eu_2C_{44}H_{46}O_{20}N_4$  Mw = 1182.71 g/mol. Selected FTIR: 2986 (w) 1634 (m) 1412 (m) 1281 (m) 1208 (m).

 $Gd_2(L)_4(DEF)_2(NO_3)_2$  (5): Complex 5 was synthesised following the general procedure above using  $Gd(NO_3)_3.xH_2O$  white colourless plate crystals after 3 days.  $Gd_2C_{44}H_{46}O_{20}N_4$  Mw = 1193.28 g/mol. Selected FTIR: 2985 (w) 1609 (m) 1411 (m) 1285 (m) 1213 (m).

 $Tb_2(L)_4(DEF)_2(NO_3)_2$  (6): Complex 6 was synthesised following the general procedure above using  $Tb(NO_3)_3.xH_2O$  yielding white plate crystals after 3 days.  $Tb_2C_{44}H_{46}O_{20}N_4$  Mw = 1196.64 g/mol. Selected FTIR: 2984 (w) 1631 (m) 1417 (m) 1286 (m) 1214 (m).

 $Dy_2(L)_4(DEF)_2(NO_3)_2$  (7): Complex 7 was synthesised following the general procedure above using  $Dy(NO_3)_3.xH_2O$  yielding pale yellow plate crystals after 3 days.  $Dy_2C_{44}H_{46}O_{20}N_4 = 1203.78$  g/mol.

### Single crystal X-ray Diffraction

Suitable single crystals of frameworks **1-7** were analysed using a Bruker D8 Venture diffractometer using a CuK $\alpha$  IµS X-radiation source ( $\lambda$  = 1.5406 Å), and a Photon II detector. The single crystals were maintained at 150 K during data collections using an Oxford Cryosystems cryostream. Data collection and reduction were carried out using APEX III software interface, before final post processing and absorption scaling with SADABS.<sup>2</sup> Structures were solved and refined using the Olex2 interface<sup>3</sup> to the ShelX suite of programs, XT for structure solution and XL for structure refinement.<sup>4,5</sup>

Complex	a (σ)	b	C	α	β	γ	Volume (ų)
1 (Ce)	9.9080(16)	11.1776(18)	13.939(2)	109.534(4)	95.570(4)	96.701(5)	1429.6(4)
2 (Pr)	9.8826(9)	11.1554(10)	13.9347(12)	109.747(3)	95.520(2)	96.847(2)	1420.2(2)
3 (Nd)	9.9056(10)	11.1381(11)	13.9162(14)	109.803(3)	95.425(3)	96.535(4)	1429.6(2)
4 (Eu)	10.381(7)	10.4095(7)	13.861(10)	81.397(4)	70.936(4)	89.265(4)	1399.02(17)
5 (Gd)	10.3633(8)	10.4061(8)	13.8433(10)	81.375(3)	70.977(3)	89.188(3)	1394.43(18)
6 (Tb)	10.3607(8)	10.4017(8)	13.8104(10)	81.551(3)	71.010(4)	89.160(3)	1391.18(19)
7 (Dy)	10.3582(8)	10.4047(8)	13.7990(11)	91.692(3)	71.113(3)	89.156(3)	1391.51(19)

Table 1: Summary of unit cell parameters for complexes 1-7 at 150 K



Figure S1: Contact surface void space plots of complexes 1-3 (a and c) and complexes 4-7 (b and d). For complexes 1-3 a probe radius of 1.0 Å is shown, for complexes 4-7 a smaller probe radius of 0.8 Å is shown. Analysis shows that the largest accessible void for complexes 1-3 sits in a pocket of the edge of the network sheet. Whereas for complex 4-7 the largest void space penetrates the framework down the c axis. C-grey, O- red, N-blue, Ln- cream/green. Hydrogen atoms omitted for clarity.

#### **High pressure**

A crystal was placed in a diamond anvil cell containing a 1:1 pentane/isopentane mixture with diamond culets of 0.8 mm. The crystal occupies a sample chamber created by a steel gasket of 0.25 mm thickness, pre-indented to 0.15 mm with a precision drilled hole of 300  $\mu$ m. A ruby sphere was introduced to the sample chamber for pressure determination. To begin the pressure was increased from ambient in 5 kbar increments until there was a notable change in unit cell parameters. Then the diamond anvil cell (DAC) was mounted directly onto the goniometer of XIPHOS II, a four circle Huber diffractometer with Ag–K $\alpha$  I $\mu$ S generator.<sup>6</sup> High pressure data were handled using the Bruker APEX2 software suite, which incorporates SAINT and SADABS for integration, cell refinement, and scaling <sup>2,7,8</sup> The SHELX program suite was used for structure solution and refinement of all structures within the OLEX2 interface.<sup>3</sup>



Figure S2: A scatter graph showing the unit cell volume deviation from the ambient pressure structure solution for complex 2 over the pressure range of 0-35 kbar. The complex exhibits a first order phase transition between 30 and 35 kbar.



Figure S3: A scatter graph showing the unit cell volume deviation from the ambient pressure structure solution for complex 6 over the pressure range of 0-25 kbar. The complex exhibits an initial compression at 5 kbar resulting in a volume decrease of approximately 185 Å<sup>3</sup>.



Figure S4: Depicts individual representations of complex 2 at ambient pressure (orange) and at 35 kbar (purple). a) shows a closer inspection of the secondary building unit, highlighting the breaking of a bond. b) highlights the change in the organic allyloxy groups from staggered to eclipsed.



Identification code	022 R34
Empirical formula	C24H34CeN3O11
Formula weight	680.66
Temperature/K	150.0
Crystal system	triclinic
Space group	P-1
a/Å	9.9080(16)
b/Å	11.1776(18)
c/Å	13.939(2)
a/°	109.534(4)
β/°	95.570(4)
$\gamma/^{\circ}$	96.701(5)
Volume/Å3	1429.6(4)
Ζ	2
pcalcg/cm3	1.581
μ/mm 1	12.835
F(000)	690.0
Crystal size/mm3	0.2  imes 0.07  imes 0.04
Radiation	$CuK\alpha (\lambda = 1.54178)$
20 range for data collection/°	6.8 to 135.666
Index ranges	$-11 \le h \le 11, -13 \le k \le 13, -16 \le l \le 16$
Reflections collected	42338
Independent reflections	5030 [Rint = 0.0680, Rsigma = 0.0332]
Data/restraints/parameters	5030/318/375
Goodness-of-fit on F2	1.039
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0391, $wR2 = 0.1092$
Final R indexes [all data]	R1 = 0.0398, $wR2 = 0.1099$
Largest diff. peak/hole / e Å-3	1.53/-0.78



Identification code	lrh 022 r35
Empirical formula	C24H34N3O11Pr
Formula weight	681.45
Temperature/K	150.0
Crystal system	triclinic
Space group	P-1
a/Å	9.8826(9)
b/Å	11.1554(10)
c/Å	13.9347(12)
α/°	109.747(3)
β/°	95.520(2)
$\gamma/^{\circ}$	96.847(2)
Volume/Å3	1420.2(2)
Ζ	2
pcalcg/cm3	1.594
μ/mm 1	13.694
F(000)	692.0
Crystal size/mm3	0.1  imes 0.06  imes 0.03
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
2@ range for data collection/°	10.554 to 133.188
Index ranges	$-11 \le h \le 11, -13 \le k \le 13, -16 \le l \le 16$
Reflections collected	58828
Independent reflections	4970 [Rint = 0.0548, Rsigma = 0.0232]
Data/restraints/parameters	4970/296/376
Goodness-of-fit on F2	1.044
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0306, wR2 = 0.0818
Final R indexes [all data]	R1 = 0.0307, wR2 = 0.0819
Largest diff. peak/hole / e Å-3	1.70/-1.47



Identification code	lrh 022 r36
Empirical formula	C24H34N3NdO11
Formula weight	684.78
Temperature/K	150.0
Crystal system	triclinic
Space group	P-1
a/Å	9.9056(10)
b/Å	11.1381(11)
c/Å	13.9162(14)
α/°	109.803(3)
β/°	95.425(3)
γ/°	96.535(4)
Volume/Å3	1420.6(2)
Ζ	2
pcalcg/cm3	1.601
μ/mm 1	14.479
F(000)	694.0
Crystal size/mm3	0.08  imes 0.03  imes 0.01
Radiation	$CuK\alpha (\lambda = 1.54178)$
2@ range for data collection/°	6.82 to 134.222
Index ranges	$-11 \le h \le 11, -13 \le k \le 13, -16 \le l \le 16$
Reflections collected	54599
Independent reflections	4998 [Rint = 0.0623, Rsigma = 0.0271]
Data/restraints/parameters	4998/294/376
Goodness-of-fit on F2	1.107
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0376, wR2 = 0.1043
Final R indexes [all data]	R1 = 0.0385, wR2 = 0.1054
Largest diff. peak/hole / e Å-3	1.77/-0.77



Identification code	022 R37
Empirical formula	C24H34EuN3O11
Formula weight	692.50
Temperature/K	150.0
Crystal system	triclinic
Space group	P-1
a/Å	10.3817(7)
b/Å	10.4095(7)
c/Å	13.8631(10)
α/°	81.397(4)
β/°	70.936(4)
$\gamma/^{\circ}$	89.265(4)
Volume/Å3	1399.02(17)
Z	2
pcalcg/cm3	1.644
μ/mm 1	16.589
F(000)	700.0
Crystal size/mm3	0.08  imes 0.02  imes 0.01
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
20 range for data collection/°	8.598 to 145.534
Index ranges	$-12 \le h \le 12, -12 \le k \le 12, -16 \le l \le 16$
Reflections collected	35592
Independent reflections	5429 [Rint = 0.0999, Rsigma = 0.0524]
Data/restraints/parameters	5429/265/356
Goodness-of-fit on F2	1.090
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0467, wR2 = 0.1251
Final R indexes [all data]	R1 = 0.0516, $wR2 = 0.1289$
Largest diff. peak/hole / e Å-3	1.88/-0.74



Identification code	lrh 022 r38
Empirical formula	C24H34GdN3O11
Formula weight	697.79
Temperature/K	150.0
Crystal system	triclinic
Space group	P-1
a/Å	10.3633(8)
b/Å	10.4061(8)
c/Å	13.8433(10)
α/°	81.375(3)
β/°	70.977(3)
γ/°	89.188(3)
Volume/Å3	1394.43(18)
Ζ	2
pcalcg/cm3	1.662
μ/mm 1	15.927
F(000)	702.0
Crystal size/mm3	0.06  imes 0.02  imes 0.01
Radiation	$CuK\alpha (\lambda = 1.54178)$
20 range for data collection/°	6.836 to 134.368
Index ranges	$-12 \le h \le 11, -12 \le k \le 12, -16 \le l \le 16$
Reflections collected	53274
Independent reflections	4904 [Rint = 0.0629, Rsigma = 0.0271]
Data/restraints/parameters	4904/265/357
Goodness-of-fit on F2	1.081
Final R indexes [I>= $2\sigma$ (I)]	R1 = 0.0222, wR2 = 0.0593
Final R indexes [all data]	R1 = 0.0231, $wR2 = 0.0598$
Largest diff. peak/hole / e Å-3	0.42/-0.55



Identification code	LRH 022 R39 0m
Empirical formula	C24H34N3O11Tb
Formula weight	699.46
Temperature/K	150.00
Crystal system	triclinic
Space group	P-1
a/Å	10.3607(8)
b/Å	10.4017(8)
c/Å	13.8104(10)
a/°	81.551(3)
β/°	71.010(4)
$\gamma/^{\circ}$	89.160(3)
Volume/Å3	1391.18(19)
Ζ	2
pcalcg/cm3	1.670
μ/mm 1	13.045
F(000)	704.0
Crystal size/mm3	0.07  imes 0.03  imes 0.02
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
2@ range for data collection/°	6.846 to 134.008
Index ranges	$-12 \le h \le 12, -12 \le k \le 12, -16 \le l \le 16$
Reflections collected	61503
Independent reflections	4892 [Rint = 0.0662, Rsigma = 0.0259]
Data/restraints/parameters	4892/265/356
Goodness-of-fit on F2	1.116
Final R indexes [I>= $2\sigma$ (I)]	R1 = 0.0265, wR2 = 0.0719
Final R indexes [all data]	R1 = 0.0281, wR2 = 0.0728
Largest diff. peak/hole / e Å-3	0.66/-0.47



Identification code	LRH 022 R40
Empirical formula	C24H34DyN3O11
Formula weight	703.04
Temperature/K	150.00
Crystal system	triclinic
Space group	P-1
a/Å	10.3582(8)
b/Å	10.4047(8)
c/Å	13.7990(11)
α/°	81.692(3)
β/°	71.113(3)
$\gamma/^{\circ}$	89.156(3)
Volume/Å3	1391.52(19)
Ζ	2
pcalcg/cm3	1.678
μ/mm 1	14.911
F(000)	706.0
Crystal size/mm3	0.07  imes 0.04  imes 0.02
Radiation	$CuK\alpha (\lambda = 1.54178)$
2O range for data collection/°	8.594 to 144.876
Index ranges	$-12 \le h \le 12, -12 \le k \le 12, -16 \le l \le 15$
Reflections collected	20691
Independent reflections	5362 [Rint = 0.0598, Rsigma = 0.0509]
Data/restraints/parameters	5362/0/357
Goodness-of-fit on F2	1.072
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0401, wR2 = 0.1080
Final R indexes [all data]	R1 = 0.0415, $wR2 = 0.1091$
Largest diff. peak/hole / e Å-3	1.38/-1.24



Cr	vstal	data	and	structure	refinement	for	022	<b>R35</b>	35kbar

Identification code	022 R35 35kbar
Empirical formula	C24H34N3O11Pr
Formula weight	681.45
Temperature/K	293.0
Crystal system	triclinic
Space group	P-1
a/Å	9.8824(4)
b/Å	10.0039(4)
c/Å	13.3049(9)
a/°	98.203(4)
β/°	111.376(4)
$\gamma^{\prime \circ}$	91.257(2)
Volume/Å3	1208.43(11)
Ζ	2
pcalcg/cm3	1.873
μ/mm 1	1.107
F(000)	692.0
Crystal size/mm3	0.06  imes 0.03  imes 0.02
Radiation	AgKa ( $\lambda = 0.56086$ )
20 range for data collection/°	3.256 to 39.304
Index ranges	$-11 \le h \le 11, -11 \le k \le 11, -9 \le l \le 10$
Reflections collected	21137
Independent reflections	1688 [Rint = 0.0449, Rsigma = 0.0225]
Data/restraints/parameters	1688/268/338
Goodness-of-fit on F2	1.074
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0216, $wR2 = 0.0502$
Final R indexes [all data]	R1 = 0.0244, WR2 = 0.0513
Largest diff. peak/hole / e Å-3	0.46/-0.32



Cry	vstal da	ta and	structure	refinemen	t for 022	R35	25kbar
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	022 125 2511
Identification code	022_R35_25kbar
Empirical formula	C24H34N3O11Pr
Formula weight	681.45
Temperature/K	296.15
Crystal system	triclinic
Space group	P-1
a/Å	10.0371(3)
b/Å	11.1440(3)
c/Å	14.0248(8)
α/°	109.464(3)
β/°	94.951(3)
$\gamma/^{\circ}$	96.1010(10)
Volume/Å3	1458.25(11)
Z	2
pcalcg/cm3	1.552
μ/mm 1	0.917
F(000)	692.0
Crystal size/mm3	0.06  imes 0.03  imes 0.02
Radiation	AgKa ( $\lambda = 0.56086$ )
2@ range for data collection/°	3.09 to 39.232
Index ranges	$-11 \le h \le 12, -13 \le k \le 13, -10 \le l \le 10$
Reflections collected	26704
Independent reflections	2021 [Rint = 0.0466, Rsigma = 0.0233]
Data/restraints/parameters	2021/292/332
Goodness-of-fit on F2	1.019
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0324, $wR2 = 0.0804$
Final R indexes [all data]	R1 = 0.0363, wR2 = 0.0833
Largest diff. peak/hole / e Å-3	0.62/-0.74



## Crystal data and structure refinement for 022\_R39\_HP

Identification code	022_R39_25kbar
Empirical formula	C24H34N3O11Tb
Formula weight	699.46
Temperature/K	298.0
Crystal system	triclinic
Space group	P-1
a/Å	9.8113(6)
b/Å	9.9140(10)
c/Å	13.2568(8)
α/°	81.104(7)
β/°	68.313(3)
$\gamma/^{\circ}$	88.342(7)
Volume/Å3	1183.23(16)
Ζ	2
pcalcg/cm3	1.963
μ/mm 1	1.630
F(000)	704.0
Crystal size/mm3	0.07  imes 0.03  imes 0.02
Radiation	$AgK\alpha \ (\lambda = 0.56086)$
2@ range for data collection/°	2.64 to 39.17
Index ranges	$-11 \le h \le 11, -6 \le k \le 5, -15 \le l \le 15$
Reflections collected	22945
Independent reflections	1625 [Rint = 0.0862, Rsigma = 0.0438]
Data/restraints/parameters	1625/192/300
Goodness-of-fit on F2	1.041
Final R indexes $[I \ge 2\sigma(I)]$	R1 = 0.0303, wR2 = 0.0589
Final R indexes [all data]	R1 = 0.0396, $wR2 = 0.0620$
Largest diff. peak/hole / e Å-3	0.55/-0.38



Cr	ystal data an	d structure	refinement	for 022	R39 1	rt (Com	olex 6)
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Identification code	022_R39_rt
Empirical formula	C <sub>24</sub> H <sub>34</sub> N <sub>3</sub> O <sub>11</sub> Tb
Formula weight	699.46
Temperature/K	290.0
Crystal system	triclinic
Space group	P-1
a/Å	10.4991(13)
b/Å	10.5703(13)
c/Å	13.8816(17)
α/°	83.999(5)
β/°	71.592(6)
γ/°	89.746(5)
Volume/Å <sup>3</sup>	1453.1(3)
Z	2
$\rho_{calc} g/cm^3$	1.599
µ/mm <sup>-1</sup>	12.490
F(000)	704.0
Crystal size/mm <sup>3</sup>	$0.145\times0.075\times0.052$
Radiation	$CuK\alpha \ (\lambda = 1.54178)$
2Θ range for data collection/°	9.298 to 134.434
Index ranges	$-12 \le h \le 12, -12 \le k \le 12, -16 \le l \le 16$
Reflections collected	15951
Independent reflections	5063 [ $R_{int} = 0.0637, R_{sigma} = 0.0569$ ]
Data/restraints/parameters	5063/273/355
Goodness-of-fit on F <sup>2</sup>	1.111
Final R indexes [I>=2σ (I)]	$R_1 = 0.0535, wR_2 = 0.1506$
Final R indexes [all data]	$R_1 = 0.0567, wR_2 = 0.1533$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.44/-0.87



## Crystal data and structure refinement for 022\_R35\_rt (Complex 2)

Identification code	022_R35_rt
Empirical formula	C <sub>24</sub> H <sub>34</sub> N <sub>3</sub> O <sub>11</sub> Pr
Formula weight	681.45
Temperature/K	290.0
Crystal system	triclinic
Space group	P-1
a/Å	10.0678(11)
b/Å	11.1512(12)
c/Å	14.0297(15)
α/°	109.416(3)
β/°	95.138(3)
γ/°	95.947(3)
Volume/Å <sup>3</sup>	1464.6(3)
Z	2
$\rho_{calc}g/cm^3$	1.545
µ/mm <sup>-1</sup>	13.280
F(000)	692.0
Crystal size/mm <sup>3</sup>	0.16 × 0.16 × 0.13
Radiation	$CuK\alpha$ ( $\lambda = 1.54178$ )
$2\Theta$ range for data collection/°	10.442 to 141.198
Index ranges	$-12 \le h \le 12, -13 \le k \le 13, -16 \le l \le 17$
Reflections collected	21987
Independent reflections	5455 [R <sub>int</sub> = 0.0571, R <sub>sigma</sub> = 0.0479]
Data/restraints/parameters	5455/272/355
Goodness-of-fit on F <sup>2</sup>	1.084
Final R indexes [I>=2σ (I)]	$R_1 = 0.0411, wR_2 = 0.1136$
Final R indexes [all data]	$R_1 = 0.0420, wR_2 = 0.1146$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.17/-0.79

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