

## Supplementary Information

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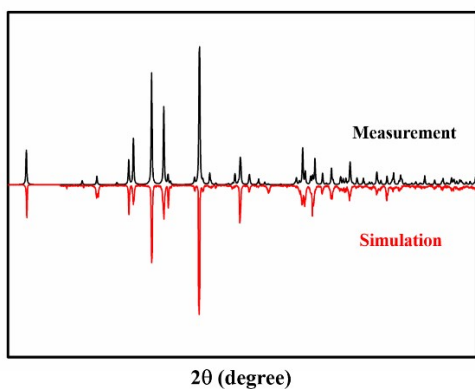
### Experimental Measurement Methods

#### Hirshfeld Surfaces and Intermolecular Interaction Analysis.

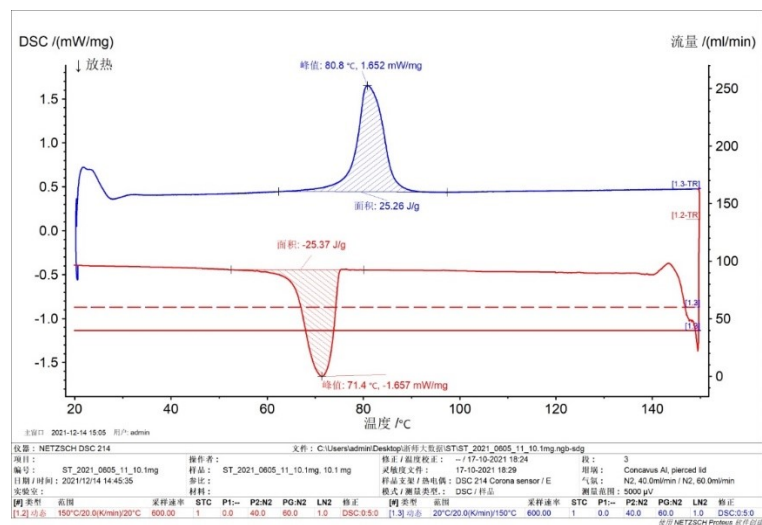
For compounds **1**, the CrystalExplorer program was used to input the structure file in CIF format at 273 K, respectively, to calculate the Hirshfeld surface and 2D fingerprint<sup>1</sup>. The intensity of molecular interaction is expressed by mapping to the Hirshfeld surface using the corresponding red-blue-white scheme: where red, white, and blue represent contacts shorter than, equal to, and longer than the van der Waals distance, respectively. In the 2D fingerprint map, each point represents a pair ( $d_i$ ,  $d_e$ ), reflecting the distance between the inner ( $d_i$ ) and outer ( $d_e$ ) nearest atoms of the Hirshfeld  $d_{\text{norm}}$  surface. The normalized contact distance  $d_{\text{norm}}$  is based on  $d_e$ ,  $d_i$  and the van der Waals (vdW) radii of the two atoms external ( $r_e^{\text{vdW}}$ ) and internal ( $r_i^{\text{vdW}}$ ) to the surface:

$$d_{\text{norm}} = \frac{d_i - r_i^{\text{vdW}}}{r_i^{\text{vdW}}} + \frac{d_e - r_e^{\text{vdW}}}{r_e^{\text{vdW}}}$$

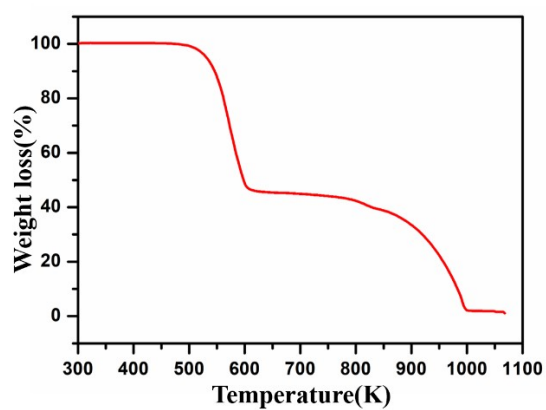
$d_{\text{norm}}$  surface represents close intermolecular interactions.



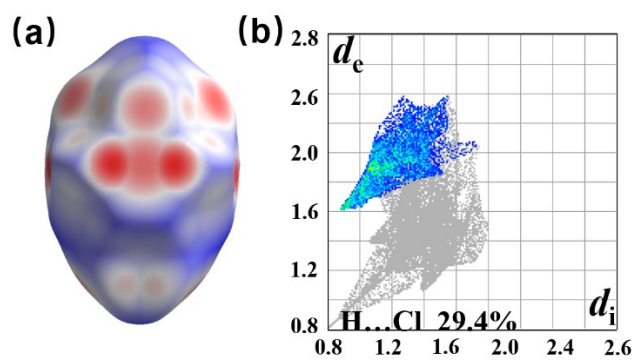
**Figure S1** Measured and simulated powder X-ray diffraction patterns of compound **1**.



**Figure S2** The original integration diagram of DSC curves.



**Figure S3** TGA curve of compounds **1** in the temperature range of 300 K-1050 K.



**Figure S4** (a) Hirshfeld surface of compound **1** at 373 K. (b) 2D fingerprint plot of compound **1** at 373 K.

**Table S1.** Crystal data and structure refinements for compound at 273 K and 373 K.

	LTP (273 K)	HTP (373 K)
Empirical formula	C <sub>8</sub> H <sub>16</sub> CdCl <sub>4</sub> F <sub>4</sub> N <sub>2</sub>	C <sub>8</sub> H <sub>16</sub> CdCl <sub>4</sub> F <sub>4</sub> N <sub>2</sub>
Formula weight	470.43	470.43
Space group	<i>C</i> <sub>2</sub> / <i>c</i>	<i>P</i> 4 <sub>2</sub> / <i>ncm</i>
Crystal system	Monoclinic	Tetragonal
<i>a</i> / Å	26.252	7.747
<i>b</i> / Å	7.5959	7.747
<i>c</i> / Å	7.6397	26.484
<i>β</i> /°	95.399	90
Volume/ Å <sup>3</sup>	1516.6	1589.4
<i>Z</i>	4	4
F(000)	920	920
GOF	1.177	1.030
<i>R</i> <sub>1</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.1220	0.0723
<i>wR</i> <sub>2</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.3419	0.2869

**Table S2.** Selected bond lengths [Å] and bond angles at 273 K.

bond lengths [Å]		bond angles [°]	
Cd1—C11 <sup>i</sup>	2.482 (2)	C11 <sup>i</sup> —Cd1—C11	180.0
Cd1—C11	2.482 (2)	C11 <sup>i</sup> —Cd1—C12	88.79 (7)
Cd1—C12	2.722 (2)	C11—Cd1—C12	91.21 (7)
Cd1—C12 <sup>i</sup>	2.722 (2)	C11 <sup>i</sup> —Cd1—C12 <sup>i</sup>	91.21 (7)
Cd1—C12 <sup>ii</sup>	2.748 (2)	C11—Cd1—C12 <sup>i</sup>	88.79 (7)
Cd1—C12 <sup>iii</sup>	2.748 (2)	C12—Cd1—C12 <sup>i</sup>	180.0
C12—Cd1 <sup>iv</sup>	2.748 (2)	C11 <sup>i</sup> —Cd1—C12 <sup>ii</sup>	87.21 (7)
		C11—Cd1—C12 <sup>ii</sup>	92.79 (7)
		C12—Cd1—C12 <sup>ii</sup>	90.816 (18)
		C12 <sup>i</sup> —Cd1—C12 <sup>ii</sup>	89.184 (18)
		C11 <sup>i</sup> —Cd1—C12 <sup>iii</sup>	92.79 (7)
		C11—Cd1—C12 <sup>iii</sup>	87.21 (7)
		C12—Cd1—C12 <sup>iii</sup>	89.184 (18)
		C12 <sup>i</sup> —Cd1—C12 <sup>iii</sup>	90.816 (18)
		C12 <sup>ii</sup> —Cd1—C12 <sup>iii</sup>	180.0
		Cd1—C12—Cd1 <sup>iv</sup>	159.97 (9)

**Table S3.** Selected bond lengths [ $\text{\AA}$ ] and bond angles at 373 K.

bond lengths [ $\text{\AA}$ ]		bond angles [ $^\circ$ ]	
Cd1—Cl3	2.462 (3)	Cl3 <sup>i</sup> —Cd1—Cl3	180.0
Cd1—Cl3 <sup>i</sup>	2.462 (3)	Cl3 <sup>i</sup> —Cd1—Cl2 <sup>i</sup>	90.0
Cd1—Cl2 <sup>i</sup>	2.7389	Cl3—Cd1—Cl2 <sup>i</sup>	90.0
Cd1—Cl2	2.7389	Cl3 <sup>i</sup> —Cd1—Cl2	90.0
Cd1—Cl1	2.7784 (8)	Cl3—Cd1—Cl2	90.0
Cd1—Cl1 <sup>i</sup>	2.7784(9)	Cl2 <sup>i</sup> —Cd1—Cl2	180.0
Cl1—Cd1 <sup>ii</sup>	2.7784 (8)	Cl3 <sup>i</sup> —Cd1—Cl1	89.53 (14)
Cl2—Cd1 <sup>iii</sup>	2.7389	Cl3—Cd1—Cl1	90.47 (14)
		Cl2 <sup>i</sup> —Cd1—Cl1	90.0
		Cl2—Cd1—Cl1	90.0
		Cl3 <sup>i</sup> —Cd1—Cl1 <sup>i</sup>	90.47 (14)
		Cl3—Cd1—Cl1 <sup>i</sup>	89.53 (14)
		Cl2 <sup>i</sup> —Cd1—Cl1 <sup>i</sup>	90.0
		Cl2—Cd1—Cl1 <sup>i</sup>	90.0
		Cl1—Cd1—Cl1 <sup>i</sup>	180.0
		Cd1 <sup>ii</sup> —Cl1—Cd1	160.67 (19)
		Cd1 <sup>iii</sup> —Cl2—Cd1	180.0

**Table S4.** H<sub>inside</sub>-Cl<sub>outside</sub> surface area, mean  $d_i$  and mean  $d_e$  in compound **1**.

Compound	Temperature	H <sub>inside</sub> -Cl <sub>outside</sub> surface area	Mean $d_i$	Mean $d_e$
<b>1</b>	273 K	30.5%	1.5085	1.7491
	373 K	29.4%	1.3185	1.6284

**Table S5.** Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ) for compound **1** at 273 K.

D—H	D(D—H)	D(H $\cdots$ A)	<DHA	D(D $\cdots$ A)	A
N1-H1A	0.89	2.617	161.5	3.472	Cl2
N1-H1B	0.89	2.816	113.7	3.277	Cl2
N1-H1C	0.89	2.842	121.62	3.394	Cl2
N1-H1C	0.89	2.696	117.69	3.208	Cl1
N1-H1C	0.89	2.901	121.05	3.445	Cl1
C4-H4A	0.97	2.587	110.24	3.061	F2
C4-H4A	0.97	2.406	177.68	3.376	F1

Symmetry codes: x, y+1, z; x, -y+1, z+1/2; -x+1/2, y+1/2, -z+1/2; x, -y+1, z-1/2; -x+1, -y+2, -z+1; x, -y+2, z+1/2.

**Table S6.** Hydrogen-bond geometry (Å, °) for compound **1** at 373 K.

<b>D—H</b>	<b>D(D—H)</b>	<b>D(H···A)</b>	<b>&lt;DHA</b>	<b>D(D···A)</b>	<b>A</b>
N1-H1B_a	0.89	2.95	112.68	3.394	C13
N1-H1B_a	0.89	2.651	157.03	3.488	C12
N1-H1C_a	0.89	2.599	149.09	3.394	C13
N1-H1D_a	0.89	2.807	129.24	3.44	C11
N1-H1D_a	0.89	2.759	139.99	3.488	C12
C6_a-H6B_a	0.96	2.542	131.08	3.254	F1_a

Symmetry codes: x-1, y, z; -x, -y+1, -z+1; -x+1/2, -y+3/2, z; -y+1, x+1/2, -z+3/2.

## Reference

1. Turner, M. J.; McKinnon, J. J.; Wolff, S. K.; Grimwood, D. J.; Spackman, P. R.; Jayatilaka, D.; Spackman, M. A. *CrystalExplorer17*; University of Western Australia, 2017, <http://hirshfeldsurface.net>