

Electronic Supplementary Information

Temperature-dependent behavior in the local structure of BaTiO₃ nanocrystals

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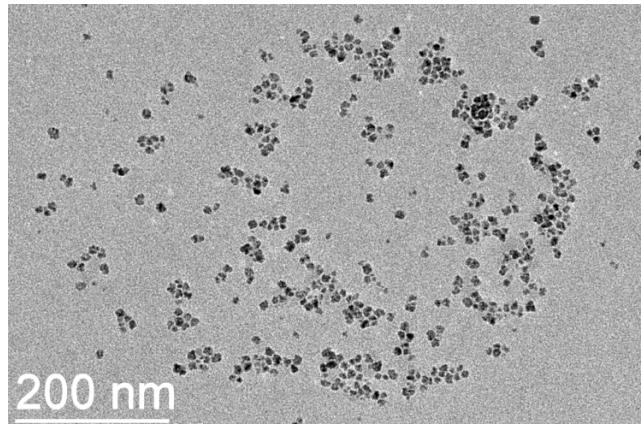


Fig. S1 TEM micrograph of *ca.* 12 nm BaTiO₃ nanocrystals.

Table S1. Parameters from Rietveld refinements of the $Pm\bar{3}m$ cubic structure of BaTiO₃ to synchrotron X-ray diffraction data collected on 12 nm BaTiO₃ nanocrystals at $T = 253$ K and $T = 413$ K.

$T = 253$ K; $Rw = 3.97\%$				
$a = 4.03646(22)$ Å; $V = 65.766(11)$ Å ³				
atom	x	y	z	U_{iso}
Ba	0.0	0.0	0.0	0.00996(27)
Ti	0.5	0.5	0.5	0.0156(5)
O	0.5	0.5	0.0	0.0023(7)
$T = 413$ K; $Rw = 3.93\%$				
$a = 4.04086(24)$ Å; $V = 65.981(12)$ Å ³				
atom	x	y	z	U_{iso}
Ba	0.0	0.0	0.0	0.01255(30)
Ti	0.5	0.5	0.5	0.0166(5)
O	0.5	0.5	0.0	0.0040(8)

Table S2. Parameters from fits of the $Pm\bar{3}m$ cubic, $P4mm$ tetragonal, and $Amm2$ orthorhombic structures of BaTiO₃ to PDFs from synchrotron X-ray total scattering data collected on 12 nm BaTiO₃ nanocrystals at $T = 253$ K and $T = 413$ K.

$Pm\bar{3}m$ cubic BaTiO ₃ ; $T = 253$ K; $Rw = 14.9\%$				
$a = 4.029 \text{ \AA}; V = 65.421 \text{ \AA}^3$				
atom	x	y	z	U_{iso}
Ba	0.000	0.000	0.000	0.006
Ti	0.500	0.500	0.500	0.017
O	0.500	0.500	0.000	0.027
$P4mm$ tetragonal BaTiO ₃ ; $T = 253$ K; $Rw = 13.4\%$				
$a = 4.007 \text{ \AA}; c = 4.077 \text{ \AA}; V = 65.469 \text{ \AA}^3$				
Atom	x	y	z	U_{iso}
Ba	0.000	0.000	0.000	0.005
Ti	0.500	0.500	0.523	0.012
O	0.500	0.500	0.000	0.025
$Amm2$ orthorhombic BaTiO ₃ ; $T = 253$ K; $Rw = 13.9\%$				
$a = 3.994 \text{ \AA}; b = 5.689 \text{ \AA}; c = 5.763 \text{ \AA}; V = 130.951 \text{ \AA}^3$				
Atom	x	y	z	U_{iso}
Ba	0.000	0.000	0.000	0.005
Ti	0.500	0.000	0.519	0.011
O	0.000	0.000	0.500	0.025
$Pm\bar{3}m$ cubic BaTiO ₃ ; $T = 413$ K; $Rw = 14.6\%$				
$a = 4.035 \text{ \AA}; V = 65.718 \text{ \AA}^3$				
Atom	x	y	z	U_{iso}
Ba	0.000	0.000	0.000	0.008
Ti	0.500	0.500	0.500	0.019

O	0.500	0.500	0.000	0.030
<i>P4mm</i> tetragonal BaTiO ₃ ; <i>T</i> = 413 K; <i>Rw</i> = 13.3%				
<i>a</i> = 4.012 Å; <i>c</i> = 4.086 Å; <i>V</i> = 65.771 Å ³				
Atom	x	y	z	<i>U</i> _{iso}
Ba	0.000	0.000	0.000	0.007
Ti	0.500	0.500	0.525	0.013
O	0.500	0.500	0.000	0.028
<i>Amm2</i> orthorhombic BaTiO ₃ ; <i>T</i> = 413 K; <i>Rw</i> = 13.1%				
<i>a</i> = 3.998 Å; <i>b</i> = 5.692 Å; <i>c</i> = 5.783 Å; <i>V</i> = 131.604 Å ³				
atom	x	y	z	<i>U</i> _{iso}
Ba	0.000	0.000	0.000	0.007
Ti	0.500	0.000	0.521	0.012
O	0.000	0.000	0.500	0.028

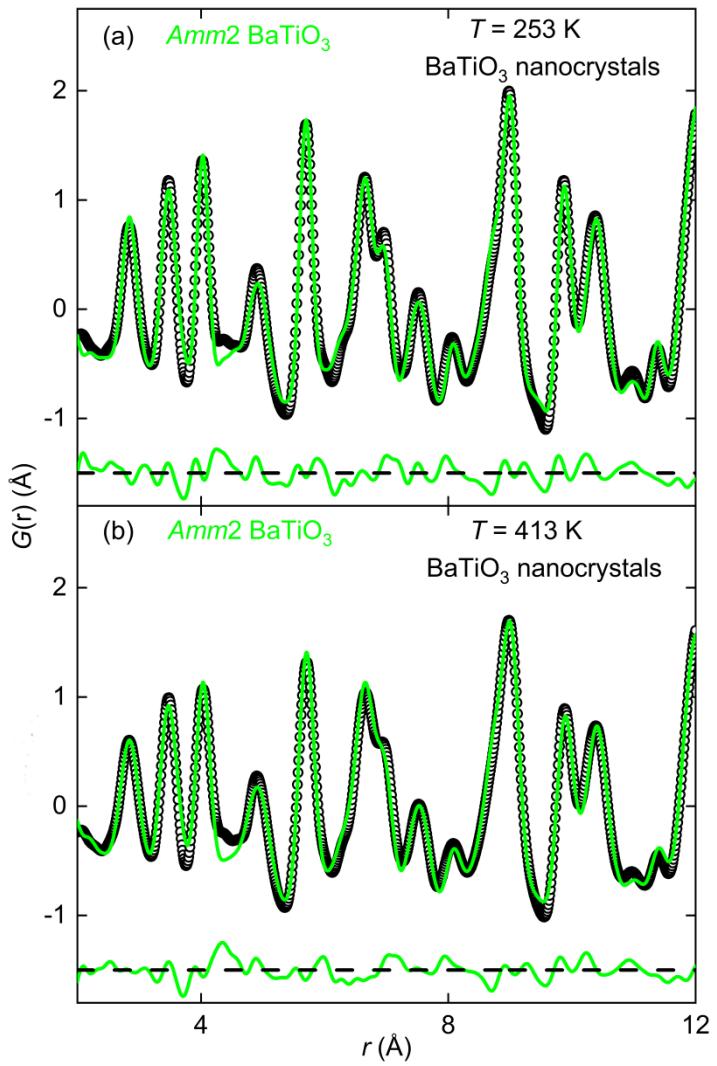


Fig. S2 Orthorhombic fits to PDFs of X-ray total scattering data collected on BaTiO_3 nanocrystals collected at (a) $T = 253$ K and (b) $T = 413$ K. Upper green lines represent the fit to the data. Lower green lines represent the difference between the fit and the data, with a dashed line to indicate a reference for no difference.

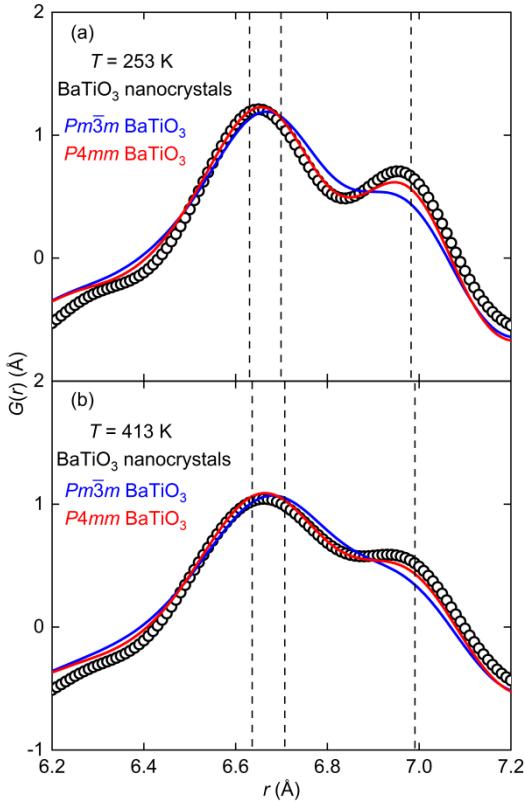


Fig. S3 Zoom-in on features of the PDFs shown in Fig. 3. The feature in the PDF centered at $r \approx 6.8 \text{ \AA}$ contains contributions from three atomic pairs in the tetragonal model of BaTiO_3 ; that is, two Ba-Ti pairs and one Ba-Ba pair. The dashed vertical lines indicate the radii of those pairs, with $r_{\text{Ba-Ti}} = 6.63, 6.67$ and $r_{\text{Ba-Ba}} = 6.98 \text{ \AA}$ at $T = 253 \text{ K}$ and $r_{\text{Ba-Ti}} = 6.63, 6.70$ and $r_{\text{Ba-Ba}} = 6.99 \text{ \AA}$ at $T = 413 \text{ K}$. The tetragonal $P4mm$ model provides better resolution of these peaks at both temperatures.

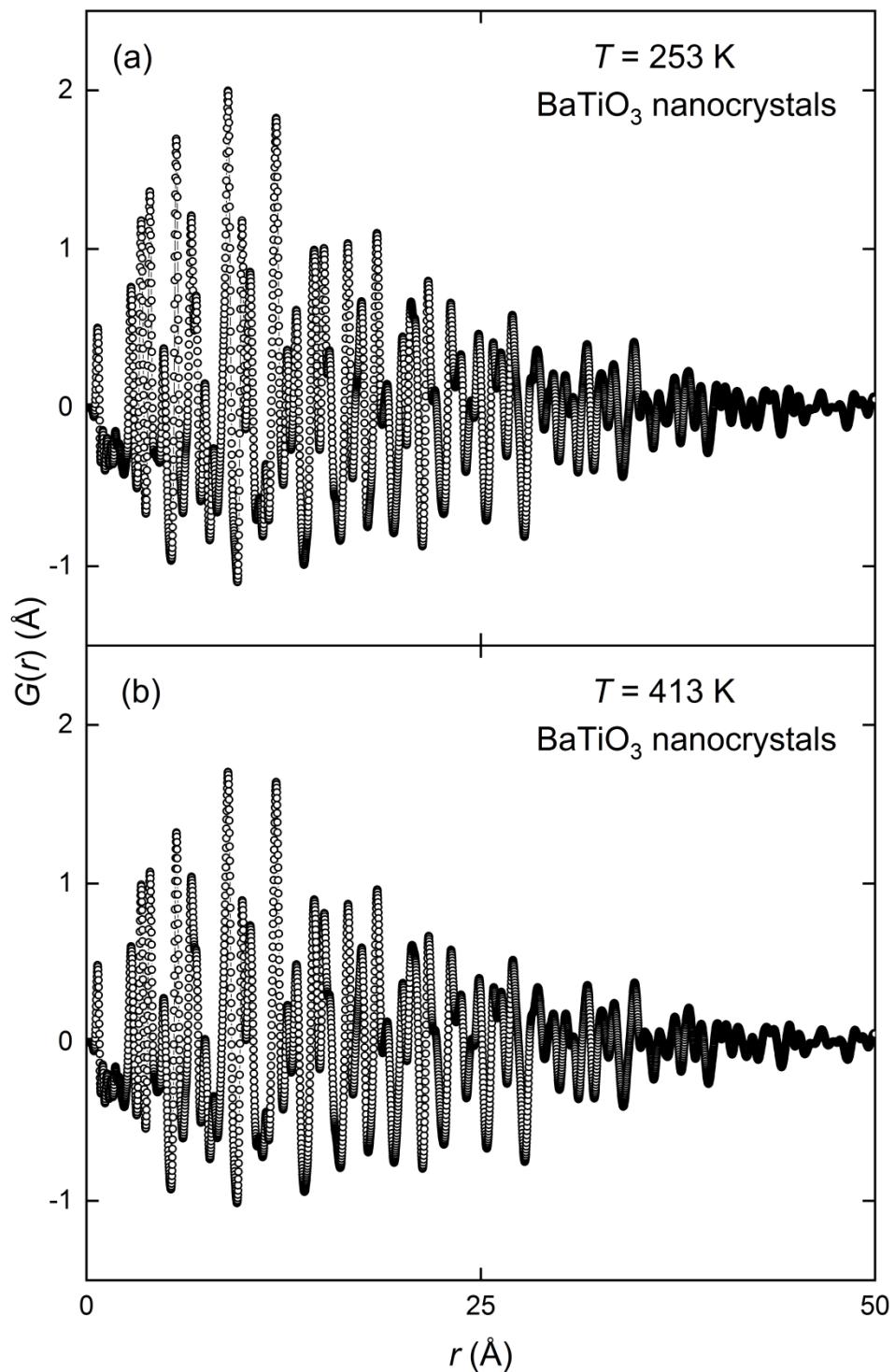


Fig. S4 PDFs, $G(r)$, of BaTiO_3 nanocrystals collected at (a) $T = 253 \text{ K}$ and (b) $T = 413 \text{ K}$. The rapid attenuation of $G(r)$ at higher values of r is due to the loss of coherence of the overall structure.