

**Surface modification with lithium silicate to enhance electrochemical performance
for high-valence metal oxide compound MnTeMoO₆**

Xiangyan Zhao ^a, Bo Li ^{a*}, Ji Zhuang ^a, Chao Liu ^a, He Zhu ^a, Ni Xue ^a, Mei Xue ^b, Lei Wang ^a, Xutang Tao ^{a*}

^a State Key Laboratory of Crystal Materials, School of Crystal Materials, Shandong University, Jinan, 250100, P. R. China.

^b College of Chemistry, Chemical Engineering and Materials Science, Shandong Normal University, Jinan, 250014, P.R. China

*Corresponding author:

Tel/Fax: 86-531-88364963/86-531-88574518

E-mail address: boli@sdu.edu.cn (Bo Li)

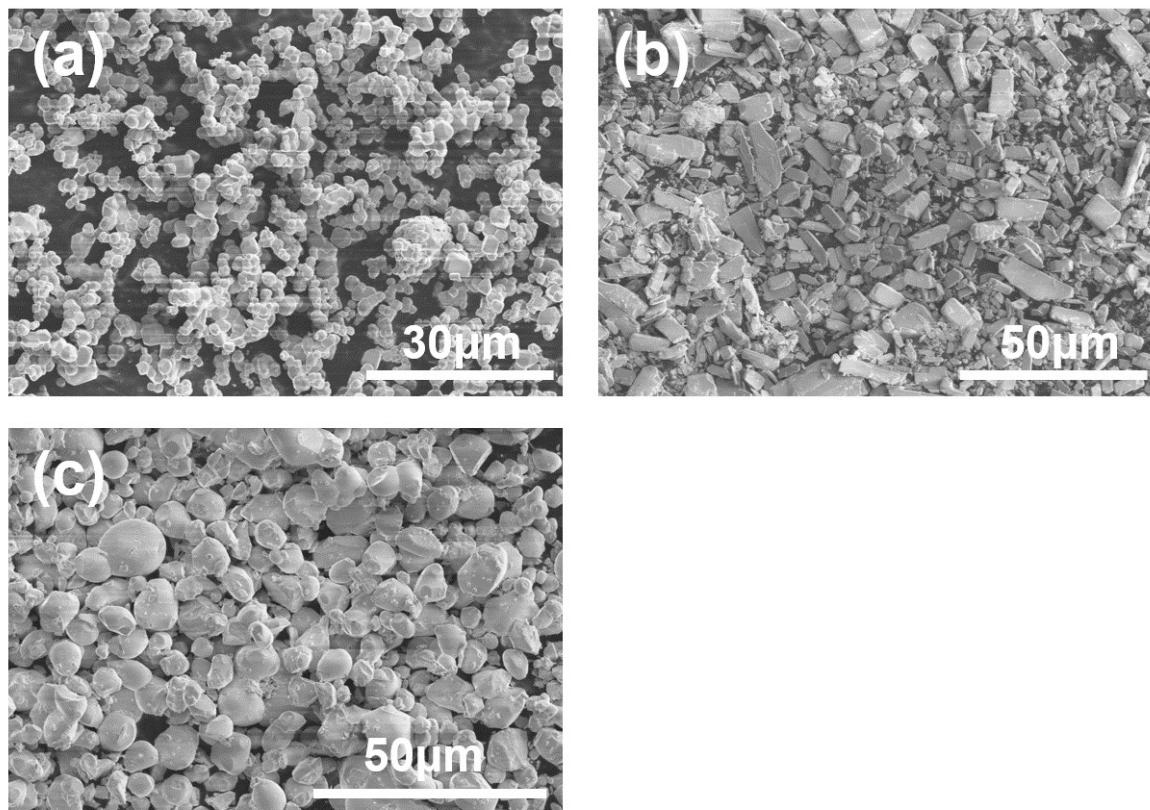


Fig. S1. SEM images of (a) MnO, (b) MoO₃ and (c) TeO₂. All raw materials are bought from Shanghai Macklin Biochemical Co., Ltd. The Purity is 99.5% for MnO, 99.95% for MoO₃ and for 99.99% TeO₂.

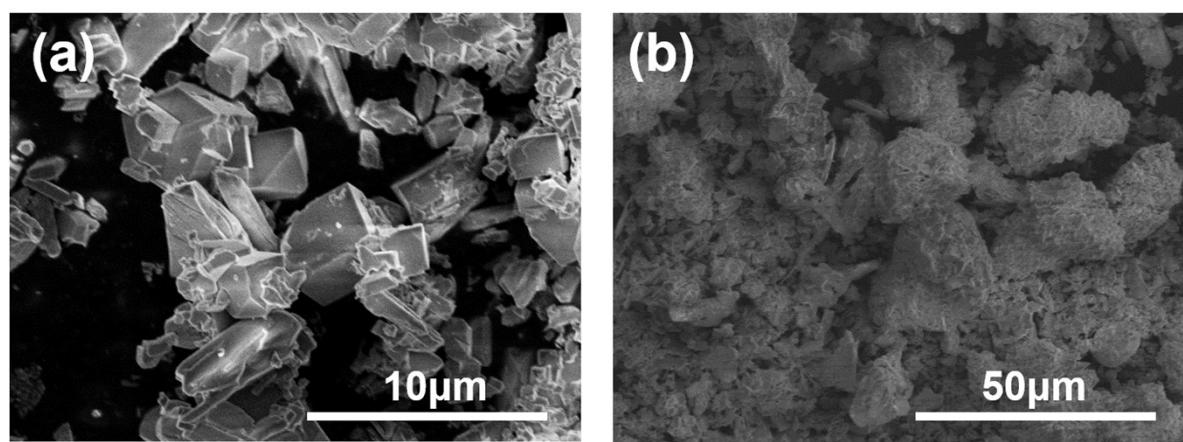


Fig. S2. SEM images of MnTeMoO₆ compound synthesized (a) in supercritical water system, and (b) by commonly-used solid state method.

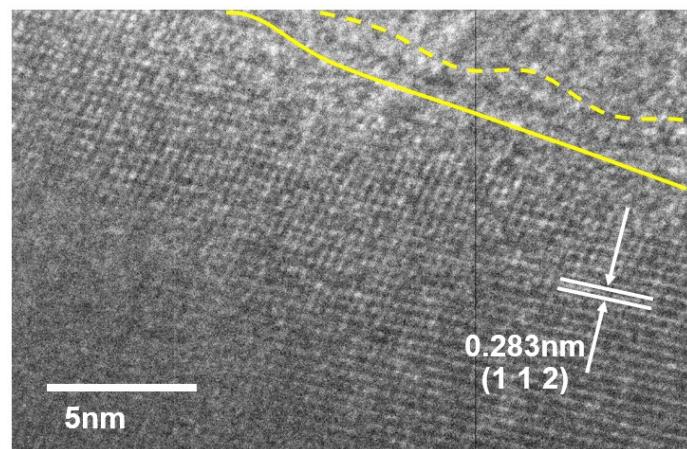


Fig. S3. HRTEM image of S-MnTM@LSO

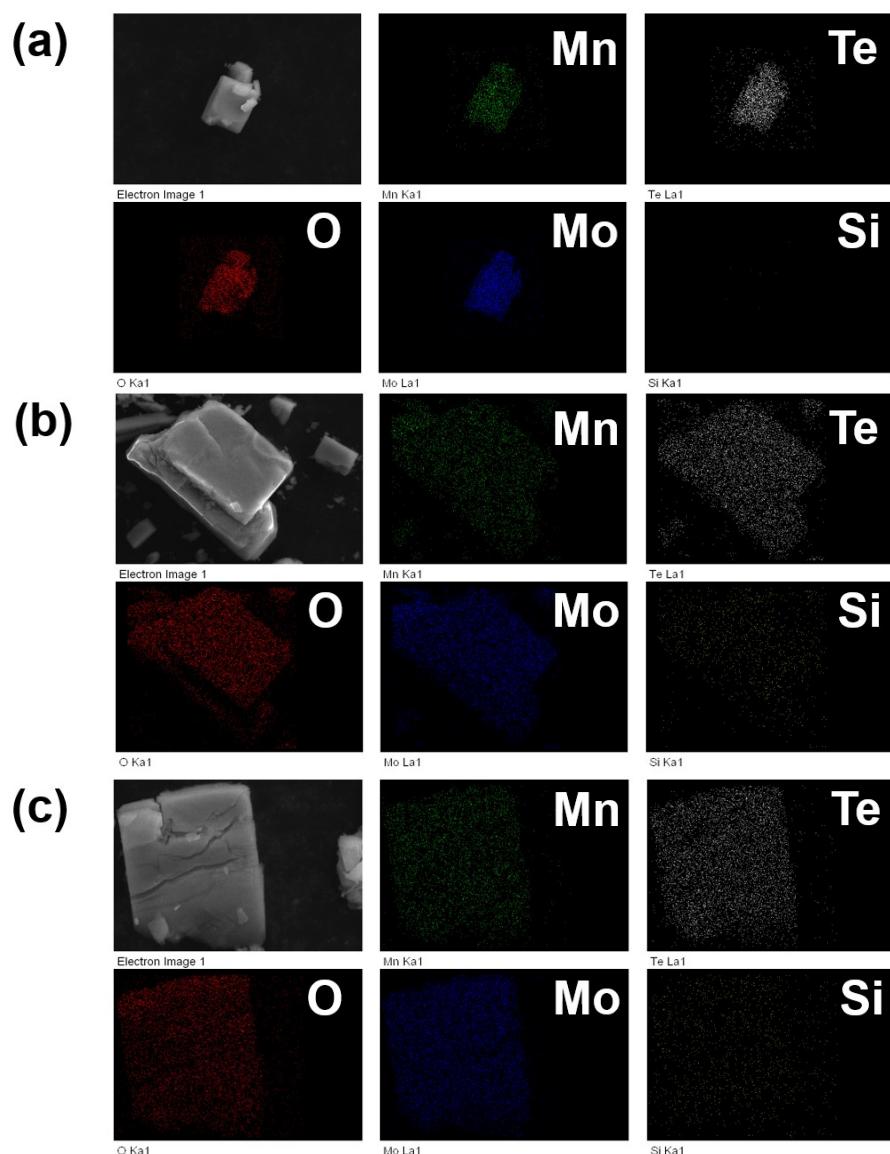


Fig. S4. SEM mapping images of (a) bare MnTM, (b) MnTM@LSO-1 and (c) MnTM@LSO-3.

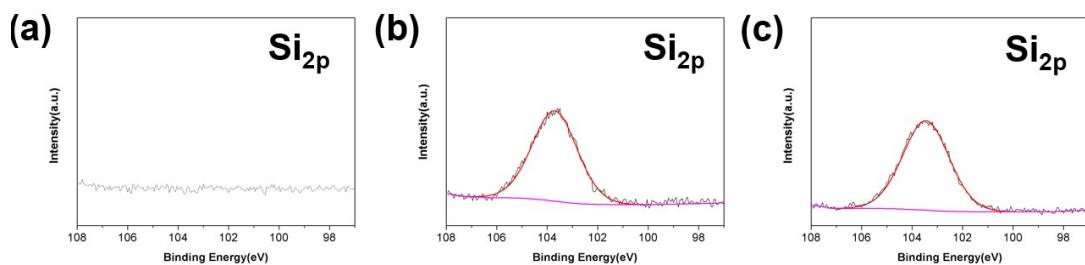


Fig. S5. XPS spectra of Si element of (a) bare MnTM, (b) MnTM@LSO-1 and (c) MnTM@LSO-3.

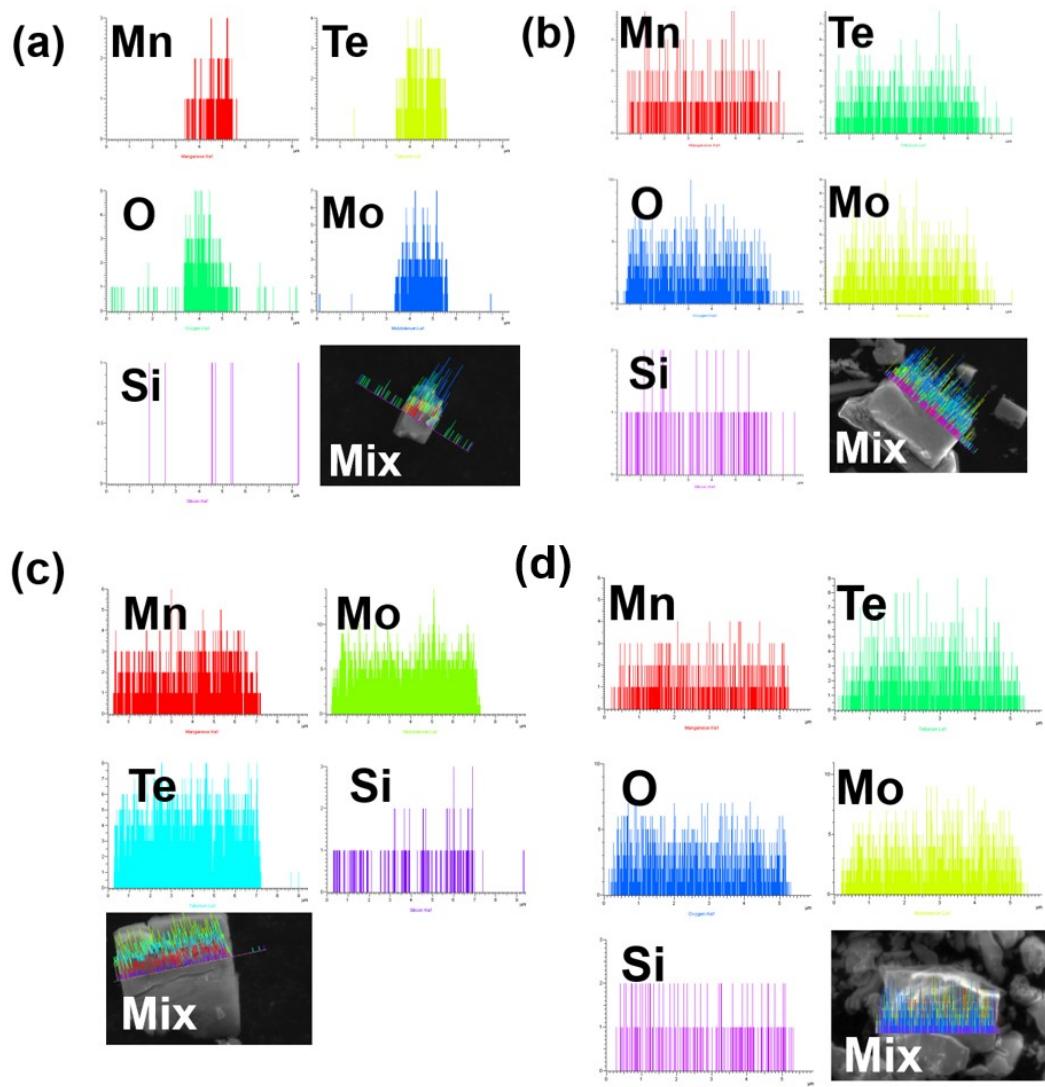


Fig. S6. SEM linear scanning mapping results of (a) bare MnTM, (b) MnTM@LSO-1, (c) MnTM@LSO-3 and (d) S-MnTM@LSO.

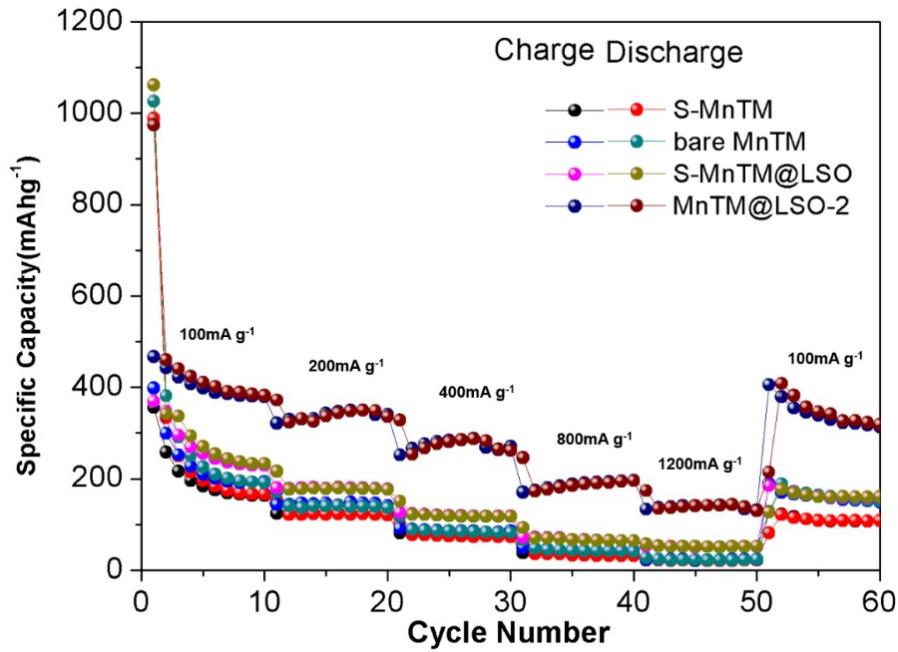


Fig. S7. Rate performance of MnTM composites and S-MnTM composites.

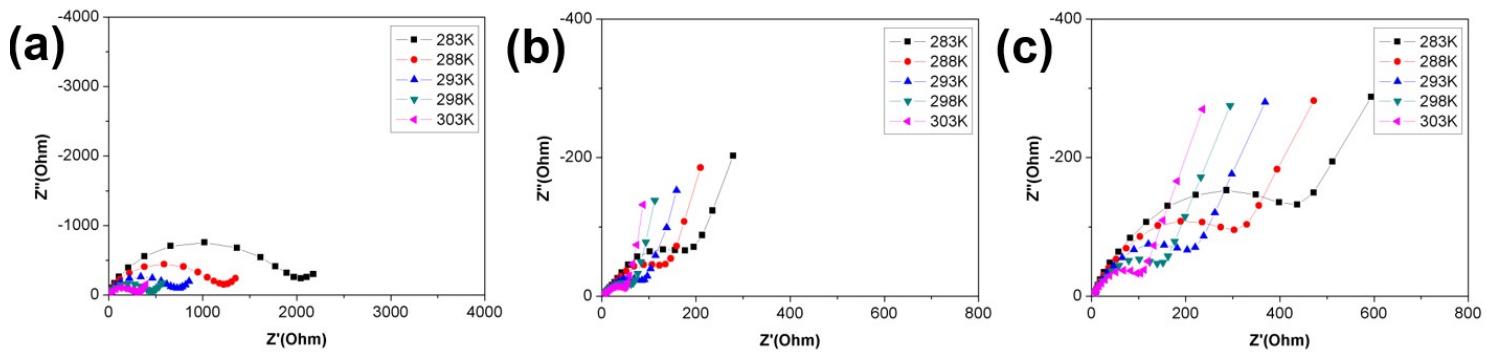


Fig. S8. The results of A.C. impedance test under differing temperatures. (a) Bare MnTM, (b) MnTM@LSO-1 and (c) MnTM@LSO-3.