

Water-induced stacking α -Fe₂O₃ hexagonal nanoplates along the [001] direction and its facet-dependent catalytic performances

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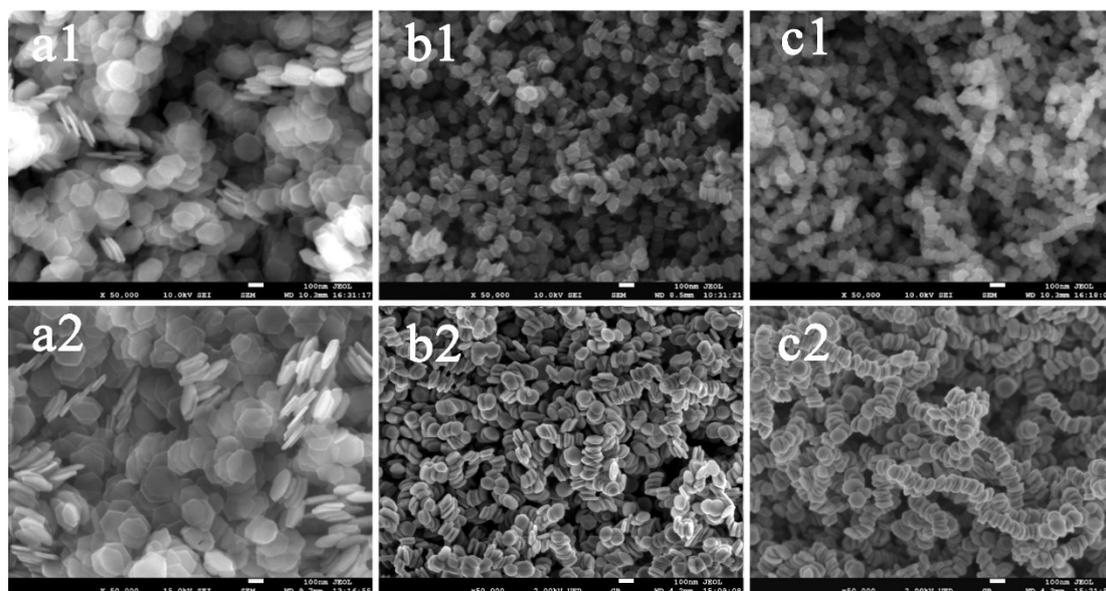


Fig. S1 SEM images of α -Fe₂O₃ nanoparticles: (a1, a2) α -Fe₂O₃ HNP, (b1, b2) α -Fe₂O₃ SSHNP and (c1, c2) α -Fe₂O₃ LSHNP.

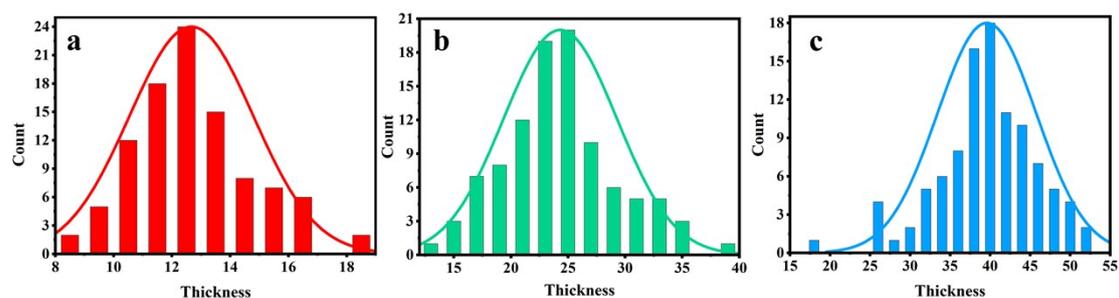


Fig. S2 The statistical graph of HNP thickness for α -Fe₂O₃ (a) HNP, (b) SSHNP and (c) LSHNP.

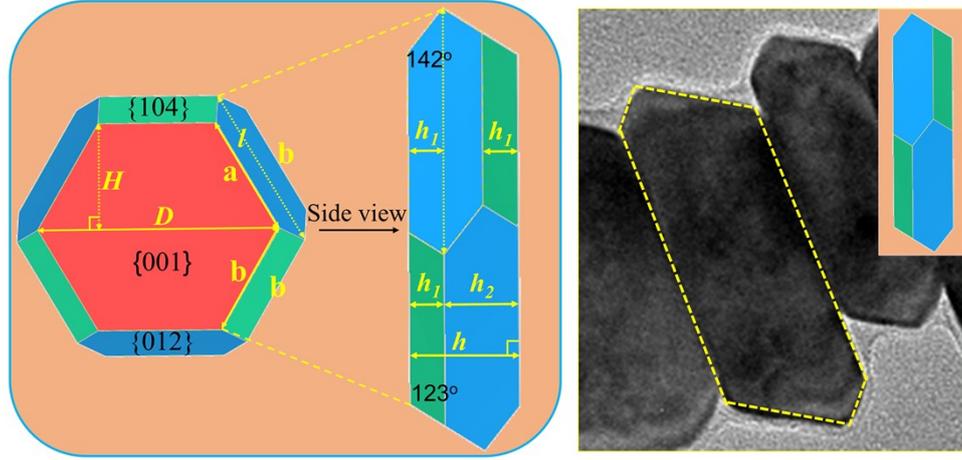


Fig. S3 The ideal crystal model of HNP for $\alpha\text{-Fe}_2\text{O}_3$.

In fact, the thickness and diameter of nanoplate in the same sample are uneven, especially in SSHNP and LSHNP. Moreover, the stacked structures do not completely cover the $\{001\}$ facets, therefore it is difficult to achieve an accurate quantification. Herein, we used an ideal model and average parameters to obtain the ratios of facets to reflect the increase trend of $\{012\} + \{104\}$, thus calculation method in this work is semi-quantitative, and the specific calculation process is as follows:

$$S_{001} = 2 \times \left[\frac{1}{2} \times (a + D) \times H + \frac{1}{2} \times (b + D) \times H \right]$$

$$S_{104} = 6 \times b \times \frac{h_1}{\cos(142^\circ - 90^\circ)}$$

$$S_{012} = 6 \times \left[\frac{1}{2} \times (a + l) \times \frac{h_1}{\cos(123^\circ - 90^\circ)} + \frac{1}{2} \times (b + l) \times \frac{h_2 - h_1}{\cos(123^\circ - 90^\circ)} \right]$$

$$l = \frac{\frac{h_1}{\tan(\theta)} + \frac{h_2 - h_1}{\tan(\theta)}}{\cos(\theta)}$$

Where D is the average diameter of hexagonal nanoplates; a and b are the side length of $\{012\}$ and $\{104\}$, respectively; H is the height of trapezium section in hexagonal $\{001\}$ plane; l is diagonal of $\{012\}$ which links the two $\{104\}$ facets; h is the thickness of nanoplate; h_1 and h_2 is the thickness of $\{012\}$ and $\{104\}$ in side view, respectively. θ is the angle between a and the vertical normal line.

After measured we found that the side length of hexagon was nearly the same, so we made a equals to b , therefore we can conducted the conclusions as follows:

$$D = 2a; H = a \times \cos(30^\circ); h_1 = h_2 - h_1; h = 3h_1; \theta = 30^\circ$$

$$S_{001} = 3\sqrt{3}a^2$$

$$S_{104} = \frac{2ah}{\cos(52^\circ)}$$

$$S_{012} = \frac{2h(a+l)}{\cos(33^\circ)}$$

$$l = \frac{h}{3\tan(38^\circ)\cos(30^\circ)} + \frac{h}{3\tan(57^\circ)\cos(30^\circ)} + a$$

$$R_{HNP} = \frac{S_{012} + S_{104}}{S_{001}}$$

$$R_{SSHNP} = 6 \times \frac{S_{012} + S_{104}}{S_{001}}$$

$$R_{LSHNP} = 18 \times \frac{S_{012} + S_{104}}{S_{001}}$$

Tab. S1 The side length and thickness statistics and facet ratios of three samples.

Sample	Side length (<i>a</i>) (nm)	Thickness (<i>h</i>) (nm)	Ratio of surface area {012}+{104}/ {001}
α -Fe ₂ O ₃ HNP	65	12	0.296
α -Fe ₂ O ₃ SSHNP	45	25	5.775
α -Fe ₂ O ₃ LSHNP	35	40	39.756

a and h are the average values of statistics.

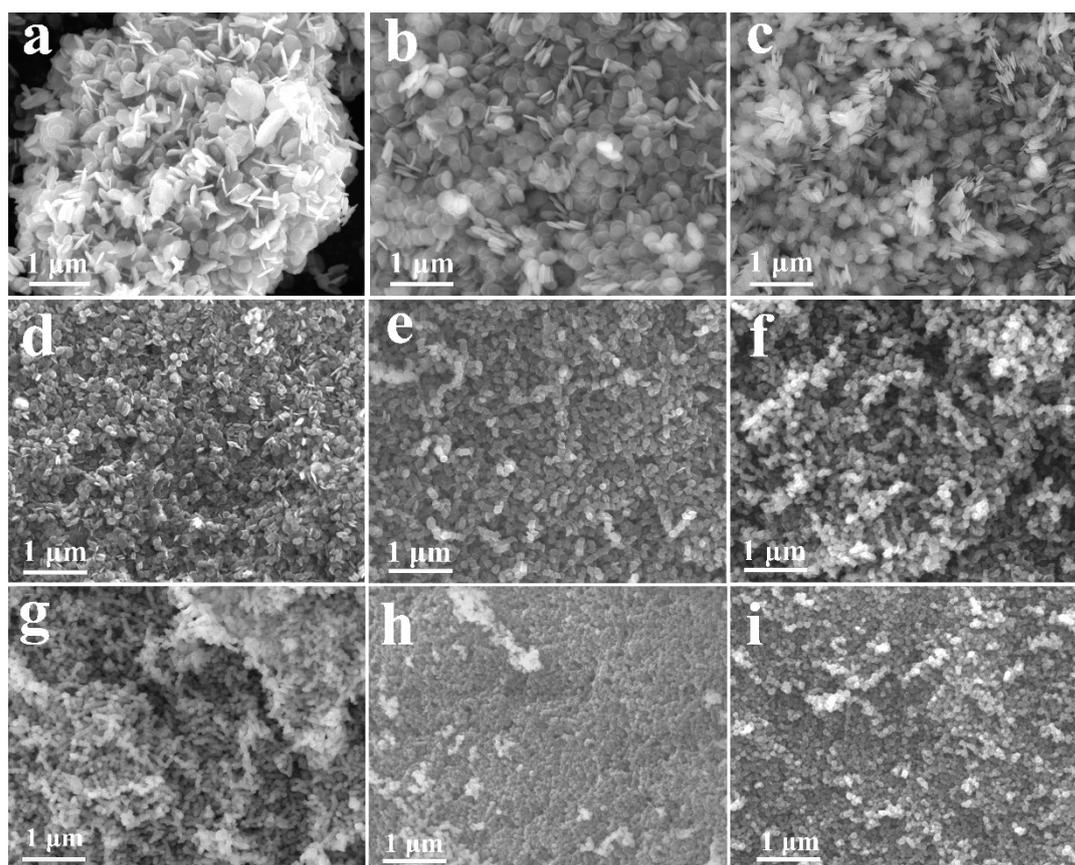


Fig. S4 SEM images in low magnification of α -Fe₂O₃ nanoplates with different H₂O content. (a) 0 mmol, (b) 28 mmol, (c) 56 mmol-HNP, (d) 111 mmol, (e) 222 mmol-SSHNP, (f) 333 mmol, (g) 444 mmol-LSHNP, (h) 555 mmol, (i) 666 mmol.

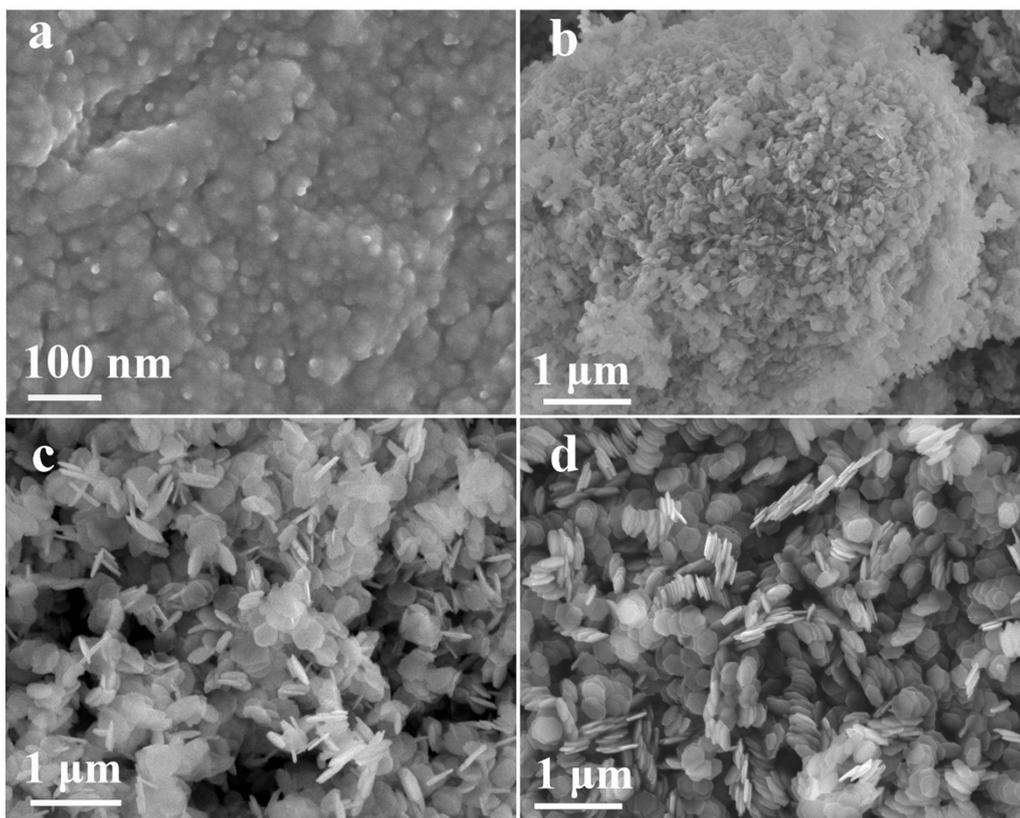


Fig. S5 SEM images of α -Fe₂O₃ HNP obtained with 56 mmol H₂O at reaction time: (a) 2 h, (b) 6 h, (c) 12 h, (d) 20 h.

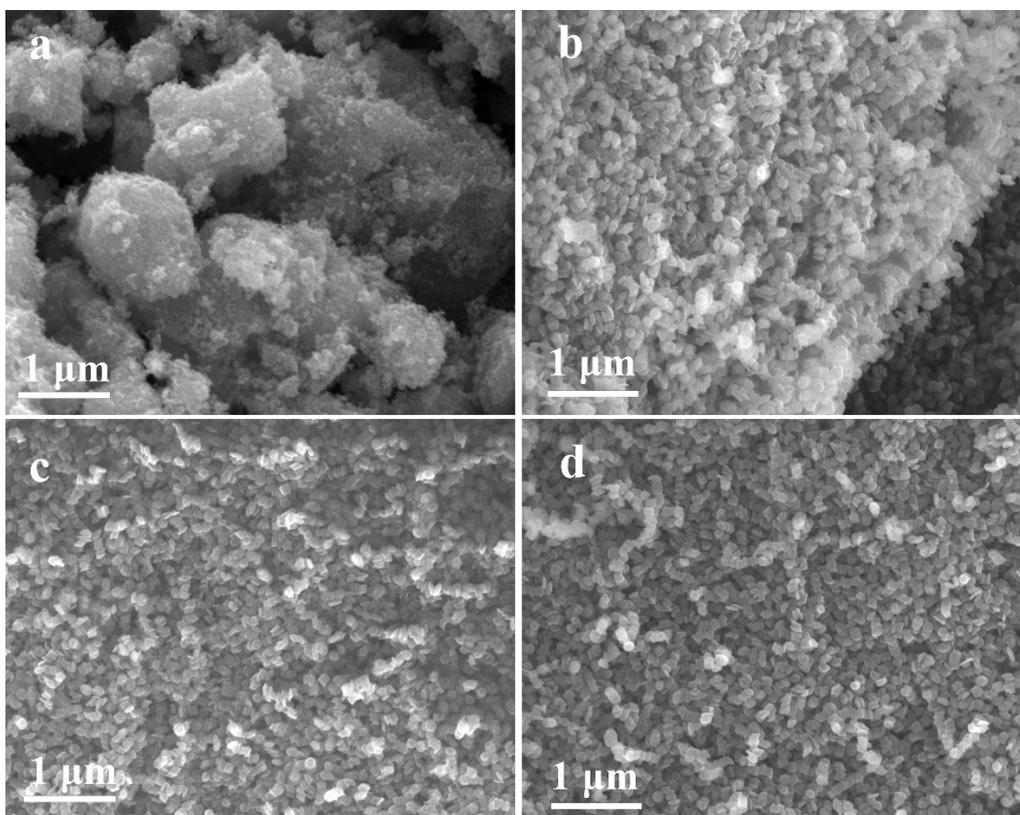


Fig. S6 SEM images of α -Fe₂O₃ SSHNP obtained with 222 mmol H₂O at reaction time (a) 3 h, (b) 6 h, (c) 12 h, (d) 21 h

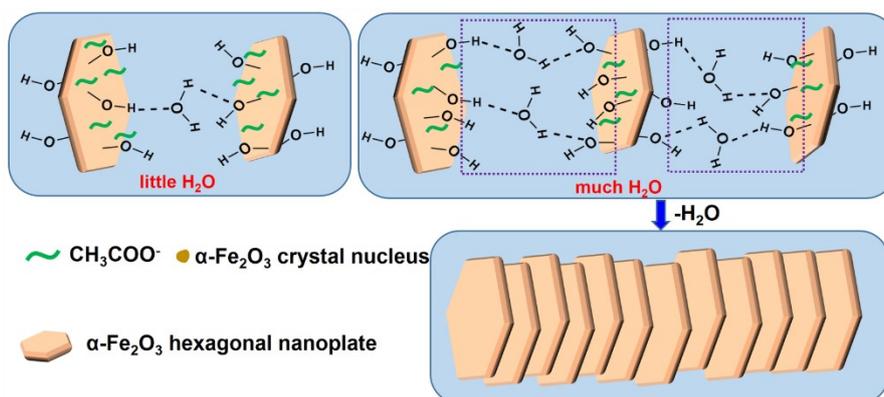


Fig. S7 Schematic diagram of hydrogen bond and dehydration process between $\alpha\text{-Fe}_2\text{O}_3$ {001} facets.