## Supporting Information

# Co-Crystals of Zwitterionic GABA API's Pregabalin and Phenibut: Properties and Application 

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## 1 Structural Details

All structures are deposited to CCDC. Deposition Number: 2170100-2170108. Bijvoet analysis was conducted on all received structures with PLATON software v. 1.19 to confirm molecular chirality, shown in chapter 5. ${ }^{[1]}$

### 1.1 Pregabalin (1)

Pure Pregabalin is examined as its (S)- and (R)-enantiomer in the context of this work. Single crystals of the enantiomerically pure forms were obtained by slow evaporation of solvent from aqueous solution. Block shaped diffraction quality crystals were obtained. The lattice and measurement parameters are shown in Table S1. Interaction distances and angles are shown in Table S2.

Table S1. Lattice and measurement parameters of (R)- and (S)-Pregabalin.

| Parameters | (R)-Pregabalin | (S)-Pregabalin |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{NO}_{2}$ | $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{~N} \mathrm{O}_{2}$ |
| $\mathrm{Mr}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 159.22 | 159.22 |
| Temperature [K] | 100(2) | 100(2) |
| System/space group | Orthorhombic, $P 2_{1} 2_{1} 2_{1}$ | Orthorhombic, $P 2_{1} 2_{1} 2_{1}$ |
| a (Å) | 6.368(4) | 6.344(4) |
| b (A) | 7.817(6) | 7.809(4) |
| $c$ ( $\AA$ ) | 18.504(4) | 18.493(3) |
| $\beta$ ( ${ }^{\circ}$ ) | 90 | 90 |
| $\mathrm{V}\left(\mathbf{A}^{3}\right)$ | 921.25(5) | 916.27(5) |
| Z/Z' | 4/1 | 4/1 |
| Density [ $\mathrm{g} / \mathrm{cm}^{3}$ ] | 1.148 | 1.154 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.656 | 0.660 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.514/1.000 | 0.622/1.000 |
| F (000) | 352 | 352 |
| Crystal size [mm] | $0.36 \cdot 0.25 \cdot 0.10$ | $0.32 \cdot 0.20 \cdot 0.03$ |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 4.8-77.5 | 4.8-77.5 |
| Completeness [\%] | 99.7 | 99.1 |
| Recorded refl. | 4569 | 6940 |
| Independent refl. | 1779 | 1757 |
| Flack x | -0.04(19) | 0.03(11) |
| Goodness-of-fit $\mathrm{F}^{2}$ | 1.053 | 1.043 |
| X-Ray Source | Cu Ka ( $\lambda=1.54184$ ) | Cu Ka ( $\lambda=1.54184$ ) |
| $\mathbf{R}_{1}$ [\%] /wR ${ }^{\text {2 }}$ [\%] / S | 4.05/ 9.95/ 1.053 | 3.01/7.86/1.043 |

Table S2. Hydrogen bond distances and angles in (R)- as well as (S)-Pregabalin.

|  | D - H [Å] | H...A [Å] | D...A [Å] | D - H...A [ ${ }^{\circ}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| (S)-Pregabalin |  |  |  |  |
| N1-H6...O1 | 0.90(3) | 1.86(3) | 2.755(8) | 171(2) |
| N1-H7...O1 | 0.94(2) | 1.79(2) | 2.728(0) | 171(2) |
| N1-H8...O2 | 0.91(2) | 1.87(2) | 2.771(1) | 172(2) |
| (R)-Pregabalin |  |  |  |  |
| N1-H6...O1 | 0.91(3) | 1.86(3) | 2.757(2) | 170(3) |
| N1-H7...O1 | 0.93(3) | 1.80 (3) | $2.731(2)$ | 174(2) |
| N1-H8...O2 | 0.92(3) | 1.86(3) | 2.767(2) | 172(3) |

A comparison of the asymmetric unit and lattice packing of both compounds is shown in Figure S1. Three distinct HB are formed in both systems. From each donor nitrogen N1 two interactions are formed with acceptor oxygen O1 with a length of $2.755(8) \AA(S) / 2.757(2) \AA$ $(R)$ and $2.728(0) \AA(S) / 2.731(2) \AA(R)$ and one interaction with acceptor oxygen O 2 at $2.771(1)$ $\AA(\mathrm{S}) / 2.767(2) \AA(\mathrm{R})$. The donor/acceptor distance in each case is short enough to be considered a strong hydrogen bond and the D-H...A angle is around 172. ${ }^{[2]}$ These values suggest that the ionicity on the ammonium and carboxylate groups may enforce the HB , leading to strong, charge assisted HB. To further elucidate on possible binding interactions, Hirshfeld- as well as electron density surface analysis was conducted with Crystal Explorer 21.5 (Figure S2). ${ }^{[3]}$ In electron density surfaces, red spots suggest high electron density which indicates high probability for binding interactions while white spots suggest surface electron density as expected from not interacting atoms and blue spots indicate low surface electron density most. In Hirshfeld surfaces red spots indicate close contacts between surface atoms shorter than the sum of their respective van der Waals radii, white spots indicate surface atoms close contacts equal to the sum of their van der Waals radii and blue spots indicate that surface atoms are further away from other surface atoms than the sum of their van der Waals radii. Both surface analyses additionally confirm that bonding interactions take place via the ammonium- and carboxylate subunits and molecular vicinity as well as electron density is increased exactly around the HB interaction spots. In congruence with the packing motif depicted in Figure S1 it can be concluded that rows of symmetrically equivalent Pregabalin molecules connect via charge assisted HB-interactions. As no conspicuous electron density features or close vicinity is present around the isobutyl group and furthermore no $\pi$-systems are present in Pregabalin, dispersive forces probably best explain the structural motif with regard to the alkylic residues.


Figure S1. The asymmetric unit of (S)-Pregabalin ( a), top) and (R)-Pregabalin ( a), bottom) is shown: from left to right a-axis, $b$-axis, $c$-axis. In b) the packing of a $2 \cdot 2 \cdot 2$ unit cell from a-axis view is shown, (S)-Pregabalin top, (R)-Pregabalin bottom. In c) the packing of a $2 \cdot 2 \cdot 2$ unit cell from $b$-axis view is shown, (S)-Pregabalin right, (R)-Pregabalin left. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red.


Figure S2. Electron density and Hirshfeld surfaces of (S)- and (R)-Pregabalin. a) shows the electron density and Hirshfeld surface of the sole symmetrically inequivalent Pregabalin molecule in the (S)-Pregabalin lattice from two sides. b) shows the electron density and Hirshfeld surface of the sole symmetrically inequivalent Pregabalin molecule in the (R)-Pregabalin lattice from two sides.


Figure S3. Powder pattern of (S)-Pregabalin recorded (red) compared to a simulated pattern by single crystal data (blue) and powder pattern of (R)-Pregabalin (purple) compared to a simulated pattern by single crystal data (green) in a range of $5^{\circ}-40^{\circ}$. (S)-Pregabalin was recrystallized from water leading to preferential directions while (R)Pregabalin was measured after stirring in acetone/water mixture for 40 h as by the racemic separation process described in this work leading to more uniform sized particles and distribution of signal intensities. The simulated pattern for (S)-Pregabalin was customized regarding hkl distribution to account for preferential directions (hkl = 4, 4, 0, March-Dollase parameter = 4).

### 1.2 Phenibut (2)

The structural properties of (rac)-Phenibut are revaluated regarding the herein discussed items based on our published structure in ZAAC 2021. ${ }^{[4]}$ Needle shaped single crystals were obtained from aqueous solution by slow evaporation of the solvent. The lattice and measurement parameters are shown in Table S3. Interaction distances and angles, including those for $\pi$-interactions, are shown in Table S4. The asymmetric unit consisting of a single Phenibut molecule as well as the crystal packing from axis $a-c$ is shown in Figure S4. (rac)-Phenibut behaves very similar in terms of its packing compared to the Pregabalin enantiomers. Two axes, in the case of Phenibut the $a$ - and $c$-axis, are relatively short at below $10 \AA$ considering the large c-axis at $27.505(3) A ̊$. While in Pregabalin the longest axis is 18.504(4) $\AA$ the difference is easily explained by the number of molecules in the unit cell, Z . Contrary to enantiomerically pure Pregabalin forms (rac)-Phenibut contains 8 molecules in its unit cell.

Table S3. Lattice and measurement parameters of (rac)-Phenibut.

| Parameters | (rac)-Phenibut |
| :---: | :---: |
| Formula | $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{~N} \mathrm{O}_{2}$ |
| $\mathrm{Mr}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 179.21 |
| Temperature [K] | 140(2) |
| System/space group | Orthorhombic, Pbca |
| a (A) | 9.384(7) |
| b (A) | 6.978(10) |
| c (A) | 27.505(3) |
| $\beta\left({ }^{\circ}\right)$ | 90 |
| $\mathrm{V}\left(\mathrm{A}^{3}\right)$ | 1801.4(4) |
| ZIZ' | 8/1 |
| Density [g/cm ${ }^{3}$ ] | 1.322 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.092 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.7990/ 1.0000 |
| F (000) | 768 |
| Crystal size [mm] | 0.06-0.10 - 0.60 |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 2.63-25.17 |
| Completeness [\%] | 99.8 |
| Recorded refl. | 6997 |
| Independent refl. | 1609 |
| Goodness-of-fit $\mathbf{F}^{2}$ | 1.013 |
| X-Ray Source | Mo Ka ( $\lambda=0.71073$ ) |
| $\mathrm{R}_{1}$ [\%] /wR $\mathrm{R}_{2}$ [\%] /S | 4.81/ 11.97/ 1.013 |

Table S4. Hydrogen bond and edge-to-face interaction distances and angles in (rac)-Phenibut.

|  | D - H (Å) | H...A [ ${ }^{\text {A }}$ ] | D...A [Å] | D - H...A [ ${ }^{\circ}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| HB |  |  |  |  |
| N1-H6...O2 | 0.95(3) | 1.84(3) | 2.775(3) | 172(3) |
| N1-H7...O1 | 1.13(3) | 1.64(3) | 2.732(3) | 160(3) |
| N1-H8...O2 | 0.96(4) | 1.84(4) | 2.795(3) | 173(2) |
| C-H... ${ }^{\text {C }}$ |  | H...Cg [ ${ }^{\text {d }}$ ] | C...Cg [Å] | C-H...Cg [ ${ }^{\text {] }}$ |
| C8-H11...Cg1 |  | 2.85(3) | 3.845(3) | 170(2) |

Contrary to Pregabalin, Phenibut has a further phenyl residue as opposed to Pregabalins isobutyl chain. Therefore, next to the HB interactions, $\pi$-interactions can be performed possibly strengthening the overall molecular attractions. The HB-network in (rac)-Phenibut is very similar to that in the Pregabalin entities. Three distinct HB with the different ammonium subunit hydrogens towards the two carboxylate oxygens are formed. HB lengths are nearly nondistinguishable from those in Pregabalin at about $2.7 \AA$ between nitrogen and oxygen atoms. The angles in Phenibut are also very close apart from N1-H7...O1 at $160^{\circ}$ slightly further from the "ideal" $180^{\circ}$ as compared to all other angles in Phenibut and Pregabalin at about $170^{\circ}$. The edge-to-face $\pi$-interaction between $\mathrm{C} 8-\mathrm{H} 11 \ldots \mathrm{Cg} 1$ is the distinctive feature of (rac)-Phenibut
interactions compared to Pregabalin. Electron density- and Hirshfeld surface confirm the discussed intermolecular bonding motif (Figure S5). Higher-than-average electron density and close atom proximity can be observed right around the relevant molecular subunits. The additional close range or high electron density spots indicate a closer packing as compared to the Pregabalin enantiomers. In the latter, some close ranges could be observed along the GABA chain that do not contribute to attractive interactions but none along the isobutyl subunit. In Phenibut however, more of these spots can be observed on both surface types which also indicates a closer packing as discussed in regard to the lattice properties.


Figure S4. The asymmetric unit of (rac)-Phenibut is shown in a): from top to bottom a-axis, b-axis, c-axis. In b) the packing of a $2 \cdot 2 \cdot 2$ unit cell from a-axis view is shown. In c) the packing of a $2 \cdot 2 \cdot 2$ unit cell from $b$-axis view is shown. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red


Figure S5. Electron density and Hirshfeld surfaces of (rac)-Phenibut. a) shows the electron density surface of the sole symmetrically inequivalent Phenibut molecule in the (rac)-Phenibut lattice from two sides. b) shows the Hirshfeld surface of the same entity.


Figure S6. Powder pattern of (rac)-Phenibut as recorded (red) compared to a simulated pattern by single crystal data (green) in a range of $5^{\circ}-40^{\circ}$.

### 1.3 Homo- and heterochiral Pregabalin:mandelic acid (1:3), ratio (1:1)

Four Pregabalin:mandelic acid multicomponent systems were obtained. The homochiral (R, R) and $(S, S)$ as well as the heterochiral $(R, S)$ and $(S, R)$ are isostructural. In each case single crystals were obtained by dissolving equimolar amounts of enantiomerically purified Pregabalin and mandelic acid in water and subsequent slow evaporation of the solvent. Thereby, colourless elongated plates were obtained. The lattice and measurement parameters for ( $R, R$ ) and ( $\mathrm{S}, \mathrm{S}$ ) are given in Table $\mathbf{S 5}$ and intermolecular interactions for them are shown in Table S6 while lattice and measurement parameters for ( $\mathrm{S}, \mathrm{R}$ ) and ( $\mathrm{R}, \mathrm{S}$ ) are presented in Table S7 and their intermolecular interactions are depicted in Table S8.

The asymmetric unit and packing of homochiral compound (S, S)-1:3 is shown in Figure S7, heterochiral compound ( $\mathbf{R}, \mathbf{S}$ )-1:3 is depicted in Figure $\mathbf{S 8}$. The homochiral systems crystallize in the same space group with the same unit cell parameters, as do the heterochiral forms respectively. In homochiral forms, Pregabalin and mandelic acid keep their default protonation status, while in heterochiral forms both molecules become formally charged. Notable is the occurrence of an uncommon packing phenomenon in $Z^{\prime}=2$. $Z^{\prime}$ is defined as the number of formula units in the unit cell divided by the number of independent general positions and a
value larger than 1 could indicate a non-optimal structure in terms of stability or symmetry, as defined by Steed and Desiraju respectively. ${ }^{[5,6]}$

Table S5. Lattice and measurement parameters of (R, R)- and (S, S)-Pregabalin:mandelic acid systems.

| Parameters | (R, R)-1:3 | (S, S)-1:3 |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{NO}_{2}, \mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{3}$ | $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{~N} \mathrm{O}_{2}, \mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{3}$ |
| $\mathrm{Mr}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 311.37 | 311.37 |
| Temperature [K] | 100(2) | 100(2) |
| System/space group | Monoclinic, P2 ${ }_{1}$ | Monoclinic, $P 2_{1}$ |
| a (A) | 6.252(7) | 6.245(5) |
| b (A) | 27.384(8) | 27.388(0) |
| c (A) | 9.960(3) | 9.962(4) |
| $\beta\left({ }^{\circ}\right)$ | 90.483(0) | 90.462(2) |
| $\mathrm{V}\left(\mathbf{A}^{3}\right)$ | 1705.43(4) | 1704.03(5) |
| Z/Z' | 4/2 | 4/2 |
| Density [ $\mathrm{g} / \mathrm{cm}^{3}$ ] | 1.221 | 1.214 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.738 | 0.739 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.458/1.000 | 0.560/1.000 |
| F (000) | 672 | 672 |
| Crystal size [mm] | $0.40 \cdot 0.11 \cdot 0.09$ | $0.59 \cdot 0.22 \cdot 0.07$ |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 3.2-77.7 | 3.2-77.8 |
| Completeness [\%] | 99.9 | 99.9 |
| Recorded refl. | 47481 | 19149 |
| Independent refl. | 6627 | 6561 |
| Flack x | -0.02(5) | 0.15(11) |
| Goodness-of-fit $\mathrm{F}^{2}$ | 1.035 | 1.084 |
| X-Ray Source | $\mathrm{Cu} \mathrm{Ka}(\lambda=1.54184)$ | Cu Ka ( $\lambda=1.54184$ ) |
| $\mathrm{R}_{1}$ [\%] /wR ${ }_{2}$ [\%] /S | 2.78/6.89/1.035 | 4.58/ 11.86/ 1.084 |

There are two symmetrically distinct Pregabalin and mandelic acid molecules. HB are the dominating intermolecular attraction force. Twelve distinct HB interactions occur in the (S, S) variant and fourteen in the ( $R, R$ ) system. The discrepancy in the number of interactions between (S, S)- and (R, R)-Pregabalin:mandelic acid might have two reasons. In (R, R) an additional interaction is added for N1A-H6A...O3C and N1B-H8B...O3D. Both interaction distances are larger than $3 \AA$ in ( $R, R$ ) and thus may not have been recognized as interactions by PLATON in (S, S). Mercury however, measures distances of $3.003 \AA$ and $3.033 \AA$ for the former and the latter interaction. It should further be noted that the melting points of both systems differ, which could be accounted to by these differences. It is notable that the shortest interactions occur from the non-ionic mandelic acid oxygens O1C and O3D and to Pregabalin carboxylate Oxygen O1A or B. The distances range from 2.487(3) $\AA$ - $2.690(3) \AA$ in (S, S) and from 2.489(7) $\AA$ - 2.695(2) $\AA$ in (R, R). When these distance values are compared to HB formed between two oppositely charged molecular fragments like N1A-H7A...O2B at 2.770(4) $\AA$ in ( $\mathrm{S}, \mathrm{S}$ ) or the corresponding N1A-H7A...O2B at $2.767(2) \AA$ in $(R, R)$ it is highlighted that assumed charge assistance between two charged sub molecular entities does not result in the
shortest interaction distances in both systems. In the heterochiral (S, R) and (R, S)-forms HB are also the main occurring bonding interaction and the same 14 types are observable in both compounds.

Table S6. Hydrogen bond distances and angles in (S, S) as well as (R, R) Pregabalin:mandelic acid cosystems. Bold written entities are charged at their oxygen or nitrogen atoms. $A$ and $B$ indices mark the different Pregabalin molecules, C and D the different mandelic acid molecules.

|  | D - H [Å] | H...A [Å] | D...A [Å] | D - H...A [ ${ }^{\circ}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| (S, S)-1:3 |  |  |  |  |
| N1A-H6A...O2C | 0.87(4) | 1.93(4) | 2.787(4) | 165(3) |
| N1A-H7A...O2B | 0.88(5) | 1.89(5) | 2.770(4) | 175(4) |
| N1A-H8A...01B | 0.88(4) | 2.24(4) | 3.036(4) | 151(4) |
| N1A-H8A...O2A | 0.88(4) | 2.40(4) | 2.940(4) | 120(3) |
| N1B-H6B...O2A | 1.07(6) | 1.72(6) | 2.780(4) | 171(5) |
| N1B-H7B...01A | 0.90(5) | 2.26(4) | 3.033(4) | 144(4) |
| N1B-H7B...O2B | 0.90(5) | 2.30(5) | 2.906(4) | 125(4) |
| N1B-H8B...O2D | 0.86(4) | 1.96(4) | 2.802(4) | 168(4) |
| O1C-H1C...01A | 0.97(8) | 1.52(8) | 2.487(3) | 174(2) |
| O3C-H2C...01B | 0.92(6) | 1.80(6) | 2.690(3) | 161(6) |
| O1D-H1D...01B | 0.93(6) | 1.56(6) | 2.492 (3) | 178(9) |
| O3D-H2D...01A | 0.87(6) | 1.82(6) | 2.682(4) | 169(5) |
| (R, R)-1:3 |  |  |  |  |
| N1A-H6A...O2C | 0.91(3) | 1.88(3) | 2.786(2) | 171(3) |
| N1A-H6A...O3C | 0.91 (3) | 2.55(3) | 3.006(2) | 111(2) |
| N1A-H7A... O2B | 0.98(3) | 1.79(3) | 2.767(2) | 171(3) |
| N1A-H8A...01B | 0.87(3) | 2.26(3) | 3.038(2) | 148(3) |
| N1A-H8A... 024 | 0.87(3) | 2.37(3) | 2.934(2) | 123(2) |
| N1B-H6B...O2A | 0.96(3) | 1.83(3) | 2.783(2) | 176(2) |
| N1B-H7B...01A | 0.91(3) | 2.23(3) | 3.035(2) | 148(3) |
| N1B-H7B...O2B | 0.91(3) | 2.33(3) | 2.906(2) | 121(2) |
| N1B-H8B...O2D | 0.88(3) | 192(3) | 2.798(2) | 172(3) |
| N1B-H8B...O3D | 0.88(3) | 2.56(3) | 3.031(2) | 114(2) |
| O1C-H1C...01A | 0.97(4) | 1.53(4) | 2.489(7) | 172(4) |
| O3C-H3C...01B | 0.84(3) | 1.88(3) | $2.695(2)$ | 166(3) |
| O1D-H1D...01B | 0.92(3) | 1.58(3) | 2.497(9) | 177(3) |
| O3D-H2D...01A | 0.89(4) | 1.81(4) | 2.682(2) | 166(4) |

Table S7. Lattice and measurement parameters of $(S, R)$ and $(R, S)$ Pregabalin:mandelic acid cosystems.

| Parameters | (S, R)-1:3 | (R, S)-1:3 |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{8} \mathrm{H}_{18} \mathrm{NO}_{2}, \mathrm{C}_{8} \mathrm{H}_{7} \mathrm{O}_{3}$ | $\mathrm{C}_{8} \mathrm{H}_{18} \mathrm{~N} \mathrm{O}_{2}, \mathrm{C}_{8} \mathrm{H}_{7} \mathrm{O}_{3}$ |
| $\mathrm{M}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 311.37 | 311.37 |
| Temperature [K] | 100(2) | 100(2) |
| System/space group | Monoclinic, $P 2_{1}$ | Monoclinic, $P 2_{1}$ |
| a (Å) | 6.077(1) | 6.051(7) |
| b (A) | 29.926(5) | 29.898(9) |
| c (Å) | 9.306(6) | 9.306(9) |
| $\beta$ ( ${ }^{\circ}$ | 92.662(0) | 92.641(0) |
| $\mathrm{V}\left(\mathrm{A}^{3}\right)$ | 1690.73(4) | 1682.19(4) |
| Z/Z' | 4/2 | 4/2 |
| Density [ $\mathrm{g} / \mathrm{cm}^{3}$ ] | 1.223 | 1.229 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.745 | 0.748 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.351/1.000 | 0.580/1.000 |
| F (000) | 672 | 672 |
| Crystal size [mm] | $0.54 \cdot 0.38 \cdot 0.15$ | $0.31 \cdot 0.12 \cdot 0.09$ |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 3.0-77.8 | 3.0-77.1 |
| Completeness [\%] | 100 | 99.8 |
| Recorded refl. | 21324 | 25468 |
| Independent refl. | 6745 | 6336 |
| Flack x | -0.05(12) | -0.15(7) |
| Goodness-of-fit $\mathrm{F}^{\mathbf{2}}$ | 1.050 | 1.048 |
| X-Ray Source | $\mathrm{CuKa}(\lambda=1.54184)$ | Cu Ka ( $\lambda=1.54184$ ) |
| $\mathrm{R}_{1}$ [\%]/wR2 [\%] /S | 4.97/ 12.05/ 1.05 | 3.00/7.33/1.05 |

Consideration of HB interaction distances shows no especially noticeable deviations from ( $\mathrm{S}, \mathrm{S}$ ) and (R, R) systems HB network. The shortest distances occur between carboxyl Pregabalin donors and carboxylate mandelic acid acceptor oxygens at 2.570(3) $\AA$ for O1B-H1B...O1D (S, R) and (R, S). This highlights that the shortest contacts occur between a charged and a neutral subunit as is the case in the ( $\mathrm{S}, \mathrm{S}$ ) and ( $\mathrm{R}, \mathrm{R}$ ) systems. Interactions where both subunits are charged like N1A-H7A...O1D, N1A-H9A...O2D, N1B-H7B...O2C and N1B-H8B...O1C with values around $2.7-2.8 \AA$ A lie in a median range compared to all shown interactions in Tables S2, S4, S6 and S8. To further elucidate on the binding interactions, Hirshfeld surfaces and electron density surfaces of homo- and heterochiral systems were calculated (Figures S9 \& S10). Again, the closest interactions as derived from the Hirshfeld surface occur around the HB interactions sites, where the highest electron density can also be located. In homochiral forms, neither close distances nor electron density accumulations can be observed around the mandelic acid phenyl subunits. However, C7D-H7D in (S, S) and (R, R) mandelic acid phenyl rings are directed towards the phenyl ring centre of gravity of their respective symmetry unrelated mandelic acid molecule C. While this possible interaction is not recognized by PLATON analysis, Mercury measures a distance C7D-H7D...Cgc of 3.672 Å or

C7D...Cgc of $4.664 \AA$ for ( $\mathrm{S}, \mathrm{S}$ ) and $3.707 \AA$ as well as $4.658 \AA$ in ( $\mathrm{R}, \mathrm{R}$ ). This could suggest possible weak edge-to-face C-H...T influences in both systems.

Table S8. Hydrogen bond distances and angles in $(S, R)$ as well as ( $R, S$ ) Pregabalin:mandelic acid cosystems. Bold written entities are charged at their oxygen or nitrogen atoms. A and B indices mark the different Pregabalin molecules, $C$ and $D$ the different mandelic acid molecules.

|  | D - H [ ] | H...A [ ${ }^{\text {d }}$ ] | D...A [Å] | D - H...A [ ${ }^{\circ}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| (S, R)-1:3 |  |  |  |  |
| N1A-H7A...O1D | 0.85(4) | 1.91(4) | 2.736(3) | 163(4) |
| N1A-H8A...O1B | 0.84(4) | 2.45(3) | 2.916(3) | 116(3) |
| N1A-H8A...O2A | 0.84(4) | 2.19(4) | 2.876(3) | 139(3) |
| N1A-H9A... O2D | 0.92(4) | 1.92(5) | 2.786 (3) | 158(4) |
| N1B-H7B... $02 C$ | 0.94(4) | 1.82(4) | 2.757(3) | 173(4) |
| N1B-H8B...01C | 0.96(6) | 1.87(5) | 2.812(3) | 167(5) |
| N1B-H9B...O2A | 0.91(4) | 2.35(4) | 3.004(3) | 129(3) |
| N1B-H9B...O3D | 0.91(4) | 2.44(4) | 3.012(3) | 121(3) |
| O1A-H1A...01C | 0.89(6) | 1.72(6) | 2.578(3) | 161(5) |
| O1B-H1B...01D | 0.97(5) | 1.60(5) | 2.570(3) | 179(6) |
| O3C-H2C...O2B | 0.77(4) | 2.23(4) | 2.880(3) | 143(4) |
| O3C-H2C...O2C | 0.77(4) | 2.19(4) | 2.645 (3) | 118(4) |
| O3D-H2D...O1A | 0.84(4) | 2.19(4) | 2.963 (3) | 152(4) |
| O3D-H2D...O2D | 0.84(4) | 2.19(4) | 2.645 (3) | 114(3) |
| (R, S)-1:3 |  |  |  |  |
| N1A-H7A...O1D | 0.93(3) | 1.83(3) | 2.734(2) | 163(3) |
| N1A-H8A...O1B | 0.87(3) | 2.44(2) | 2.905 (3) | 115(2) |
| N1A-H8A...O2A | 0.87(3) | 2.19(3) | 2.881(2) | 136(2) |
| N1A-H9A... O2D | 0.94(3) | 1.92(3) | 2.782(2) | 153(3) |
| N1B-H7B... 02 C | 0.95(3) | 1.81(3) | 2.758(2) | 169(3) |
| N1B-H8B...01C | 0.98(4) | 1.84(4) | 2.807(2) | 166(3) |
| N1B-H9B...O2A | 0.92(3) | 2.35(3) | 2.993(2) | 127(2) |
| N1B-H9B...O3D | 0.92(3) | 2.34(3) | 3.006(2) | 129(2) |
| O1A-H1A...01C | 0.92(4) | 1.70(4) | 2.588(2) | 163(3) |
| O1B-H1B...01D | 0.92(4) | 1.66(4) | 2.570(2) | 174(4) |
| O3C-H2C...O2B | 0.87(4) | 2.18 (3) | 2.881 (3) | 138(2) |
| O3C-H2C...O2C | 0.87(4) | 2.10(3) | 2.642(2) | 120(3) |
| O3D-H2D...O1A | 0.86(3) | 2.17(3) | 2.958(2) | 152(3) |
| O3D-H2D...O2D | 0.86(3) | 2.16(3) | 2.648(2) | 115(2) |

The heterochiral systems contrast the ( $\mathrm{S}, \mathrm{S}$ ) and ( $\mathrm{R}, \mathrm{R}$ ) forms, the phenyl subunits are not tilted in a favourable angle to engage into $\mathrm{C}-\mathrm{H} . . . \pi$ interactions with each other. As again no such interaction is identified by PLATON analysis, the measured distances between $\mathrm{C}_{\text {phenyl}}-\mathrm{H}$ and close $\mathrm{C}_{\text {phenyl }}$ atoms lie between $3.060 \AA$ and $3.204 \AA$ with Mercury software calculations. Centroid distances are not taken into consideration this time as the angles between possible $\mathrm{C}_{\text {phenyl}}-\mathrm{H}$ and close centroids are too unfavourable. While the distances of these possible edge-to-edge interactions seem reasonably short the charge distribution, which is alike on all phenyl rings, makes them unlikely. ${ }^{[7]}$

b)

c)


Figure S7. The asymmetric unit of $(S, S)$-Pregabalin:mandelic acid species is shown in a): from top to bottom aaxis, $b$-axis, c-axis. In b) the packing of a $2 \cdot 2 \cdot 2$ unit cell from a-axis view is shown. In $\boldsymbol{c}$ ) the packing of a $2 \cdot 2 \cdot 2$ unit cell from c-axis view is shown. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red.


Figure S8. The asymmetric unit of (R, S)-Pregabalin:mandelic acid species is shown: from top to bottom a-axis, $b$ axis, $c$-axis. In b) the packing of a $2 \cdot 2 \cdot 2$ unit cell from a-axis view is shown. In c) the packing of a $2 \cdot 2 \cdot 2$ unit cell from c-axis view is shown. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red.


Figure S9. Electron density and Hirshfeld surfaces of (S, S)- and (R, R) Pregabalin:mandelic acid species. a) and b) show the electron density surface of the two symmetrically inequivalent Pregabalin molecules in the $(S, S)$ and $(R, R)$ lattice respectively, from two sides. c) and d) show the corresponding Hirshfeld surfaces. e) and f) show the electron density surface of the two symmetrically inequivalent mandelic acid molecules in the $(S, S)$ and ( $R, R$ ) lattice respectively, from two sides. $\boldsymbol{g}$ ) and $\boldsymbol{h}$ ) show the corresponding Hirshfeld surfaces.


Figure S10. Electron density and Hirshfeld surfaces of (S, R)- and (R, S) Pregabalin:mandelic acid species. a) and b) show the electron density surface of the two symmetrically inequivalent Pregabalin molecules in the $(S, R)$ and $(R, S)$ lattice respectively, from two sides. c) and d) show the corresponding Hirshfeld surfaces. e) and f) show the electron density surface of the two symmetrically inequivalent mandelic acid molecules in the $(S, R)$ and ( $R, S$ ) lattice respectively, from two sides. $\mathbf{g}$ ) and $\boldsymbol{h}$ ) show the corresponding Hirshfeld surfaces.


Figure S11. Powder pattern of (S, S) Pregabalin:mandelic acid as recorded (green) compared to simulated patterns of $(S, S)(r e d)$ and $(R, R)$ (blue) systems single crystal data in a range of $5^{\circ}-40^{\circ}$. Only $(S, S)_{\text {rec }}$ is shown for clarity.


Figure S12. Powder pattern of (S, R) Pregabalin:mandelic acid as recorded (green) compared to simulated patterns of $(S, R)(r e d)$ and $(R, S)$ (blue) systems single crystal data in a range of $5^{\circ}-40^{\circ}$. Only $(S, R)$ rec is shown for clarity.

### 1.4 Phenibut:mandelic acid (2:3), ratio (1:1)

A single crystal of (R)-Phenibut:(S)-mandelic acid was obtained by dissolving equimolar amounts of (rac)-Phenibut and (S)-mandelic acid in aqueous solution. Elongated, diffraction quality crystalline plates were obtained by slow evaporation of the solvent. Table $\mathbf{S} 9$ shows lattice and measurement parameters. Supramolecular interactions involving HB- and $\pi$-interactions are shown in Table S10.

Table S9. Lattice and measurement parameters of (R)-Phenibut:(S)-mandelic acid.

| Parameters | (R, S)-2:3 |
| :---: | :---: |
| Formula | $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{2}, \mathrm{C}_{8} \mathrm{H}_{8} \mathrm{O}_{3}$ |
| $\mathrm{Mr}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 331.36 |
| Temperature [K] | 100(2) |
| System/space group | Monoclinic, $P 2_{1}$ |
| a (A) | 10.130(7) |
| b (A) | 6.353(9) |
| $c$ ( $A$ ) | 13.063(3) |
| $\beta$ ( ${ }^{\circ}$ ) | 95.469(2) |
| $\mathrm{V}\left(\mathrm{A}^{3}\right)$ | 837.06(2) |
| ZIZ' | 2/1 |
| Density [g/cm ${ }^{3}$ ] | 1.315 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.794 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.239/1.000 |
| F (000) | 352 |
| Crystal size [mm] | $0.52 \cdot 0.37 \cdot 0.12$ |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 3.4-77.9 |
| Completeness [\%] | 99.9 |
| Recorded refl. | 19556 |
| Independent refl. | 3246 |
| Flack x | 0.00(10) |
| Goodness-of-fit $\mathrm{F}^{\mathbf{2}}$ | 1.068 |
| X-Ray Source | Cu Ka ( $\lambda=1.54184$ ) |
| $\mathrm{R}_{1}$ [\%]/wR2 [\%] /S | 3.15/8.24/1.070 |

Table S10. Hydrogen bond and edge-to-face interaction distances and angles in (R)-Phenibut:(S)-mandelic acid. Bold written entities are charged at their oxygen or nitrogen atoms. A indices mark Phenibut molecules, B mandelic acid molecules.

|  | D - H (Å) | H...A [Å] | D...A [Å] | D - H...A [ ${ }^{\circ}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| HB |  |  |  |  |
| N1A-H6A...O2A | 0.90(3) | 2.36(3) | 2.929(2) | 121(2) |
| N1A-H6A...O1A | 0.90(3) | 2.21 (3) | 2.998(3) | 146(3) |
| N1A-H7A...O2B | 0.91(3) | 1.92(3) | 2.808(2) | 163(3) |
| N1A-H7A...O3B | 0.91(3) | 2.50(3) | 3.020(2) | 117(2) |
| N1A-H8A...O2A | 0.97(3) | 1.83(3) | 2.801 (3) | 177(3) |
| O1B-H1B...01A | 0.99(4) | 1.52(4) | 2.504(2) | 175(4) |
| O3B-H3B...01A | 0.95(4) | 1.77(3) | 2.689(2) | 163(3) |
| C-H...t |  | H...Cg [ ${ }^{\text {] }}$ ] | C...Cg [Ȧ] | C-H...Cg [ ${ }^{\circ}$ ] |
| C5B-H5B...Cg1* |  | 2.57(3) | 3.354(2) | 136(2) |
| C7A-H10A...Cg2* |  | 2.88(3) | 3.582(2) | 133(2) |
| C9A-H12A...Cg1* |  | 2.98(3) | 3.657(2) | 129(2) |

Asymmetric unit and packing are shown in Figure S13. The protonation status of Phenibut and mandelic acid stays in its default state, zwitterionic and neutral respectively. However, contrary to $\mathbf{1 : 3}$ species, a heterochiral species with these characteristics is received. The Phenibut:mandelic acid system could not be accessed as readily as its Pregabalin analogues. PXRD data shows that disregarding the mandelic acid chirality in the crystallization attempt if a multicomponent structure is formed at all, all possible chiral permutations lead to an isostructural form (Figure S15). The number of HB with 7 distinct types is halved compared to Pregabalin:mandelic acid systems. The shortest interaction distance derives from the carboxyl mandelic acid donor to the carboxylate Phenibut acceptor in the O1B-H1B...O1A interaction with 2.504(2) $\AA$. In this case the corresponding angle at $175(4)^{\circ}$ is also close to $180^{\circ}$. It can be noted that this HB is not between two subunits of opposing charges but rather between a charged and a neutral subunit. Interactions formed with the positively charged ammonium subunit lead to median ranged distances of about $2.8 \AA-3 \AA$ which includes charged-charged HB. Furthermore, the angles exhibited for HB in this structure show a distribution of values close to $180^{\circ}$. In contrast to other multicomponent species, strong evidence for $\pi$-interactions is shown in the lattice. While each Phenibut phenyl Cg 1 is in interaction twice being connected to another Phenibut phenyl and additionally a mandelic acid phenyl Cg 2 , the latter solely interacts once with Phenibut. The C5B-H5B...Cg1 edge-to-face interaction is also the shortest in $\mathrm{H} . . . \mathrm{Cg}$ as well as $\mathrm{C} . . \mathrm{Cg}$ distance at $2.57(3) \AA$ and $3.354(2) \AA$ respectively, between all described compounds. Clear evidence for edge-to-face interactions is visible upon examination of Hirshfeld and electron density surface (Figure S14). In the former, close contacts occur around Phenibut and Mandelic Acid phenyl subunits and in the latter, higher electron density can be observed on the same spots.


Figure S13. The asymmetric unit of (R)-Phenibut:(S)-mandelic acid is shown in a): from top to bottom a-axis, b-axis, c-axis. In b) the packing of a $2 \cdot 2 \cdot 2$ unit cell from b-axis view is shown. In c) the packing of a $2 \cdot 2 \cdot 2$ unit cell from c-axis view is shown. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red.


Figure S14. Electron density and Hirshfeld surfaces of (R)-Phenibut:(S)-mandelic acid. a) shows the electron density and Hirshfeld surface of the sole symmetrically inequivalent Phenibut molecule in the lattice from two sides. b) shows the electron density and Hirshfeld surface of the sole symmetrically inequivalent mandelic acid molecule in the lattice from two sides.


Figure S15. Powder pattern of (R)-Phenibut:(S)-mandelic acid simulated from single crystal data (green) compared to a recorded pattern of a (rac)-Phenibut:(R)-mandelic acid co-crystallization attempt (red) in a range of $5^{\circ}-40^{\circ}$. The comparison to a pattern received by an attempted crystallization with ( $R$ )-mandelic acid to the simulated pattern with (S)-mandelic acid while (rac)-Phenibut was used in both cases highlights isostructural crystallization. The presence of less intense Phenibut and mandelic acid signals at low angles shows the problematic crystallization of the multicomponent system. The multicomponent system seems to be not favourable and cannot always be obtained.

### 1.5 Homo- and heterochiral Pregabalin:malic acid (1:4), ratio (1:1)

Single crystals of (S)-Pregabalin and (R)- as well as (S)-malic acid were grown from aqueous solution. Diffraction quality, plate shaped crystals were obtained by slow evaporation of the solvent. During the conducted experiments the Pregabalin:malic acid entities exhibited a similarly problematic crystallization behaviour as the Phenibut:mandelic acid system. While the latter could sometimes not be obtained and always showed impurities of its precursors in the powder pattern, the former can take weeks or even months to crystallize. Lattice and measurement parameters for ( $\mathrm{S}, \mathrm{S}$ ) and ( $\mathrm{S}, \mathrm{R}$ )-Pregabalin:malic acid systems are shown in Table S11, HB properties are shown in Table S12.

The asymmetric unit shows one symmetry independent and charged Pregabalin and malic acid molecule in both homo- and heterochiral 1:4 (Figure S16). Pregabalin units are intertwined with malic acid in closely HB-connected rows, with the alkylic residues of Pregabalin being shifted towards each other and away from the HB interacting subunits. There are 8 distinct HB present in each compound. The shortest interaction in the presented
structures is between two malic acid molecules, connected via carboxyl donor to a carboxylate of an identical molecule via O1B-H1B...O5B in (S, S) and O1B-H1B...O4B in (S, R) at $2.446(4) \AA$ and $2.475(2) \AA$ respectively. Furthermore, as is the case in heterochiral Pregabalin:mandelic acid, uncharged HB interactions are present in both discussed entities even though all participating molecules are formally charged. The comparison between (S, S) and $(S, R)$ reveals differences in the interaction motif that are larger than those present in homochiral Pregabalin:mandelic acid forms and more akin to differences between homo- and heterochiral Pregabalin:mandelic acid.

Table S11. Lattice and measurement parameters of (S, S) and (S, R) Pregabalin:malic acid co-systems.

| Parameters | (S, S)-1:4 | (S, R)-1:4 |
| :---: | :---: | :---: |
| Formula | $\mathrm{C}_{8} \mathrm{H}_{18} \mathrm{~N} \mathrm{O}_{2}, \mathrm{C}_{4} \mathrm{H}_{5} \mathrm{O}_{5}$ | $\mathrm{C}_{8} \mathrm{H}_{18} \mathrm{~N} \mathrm{O}_{2}, \mathrm{C}_{4} \mathrm{H}_{5} \mathrm{O}_{5}$ |
| $\mathrm{M}_{\mathrm{r}}\left[\mathrm{g} \mathrm{mol}{ }^{-1}\right]$ | 293.31 | 293.31 |
| Temperature [K] | 100(2) | 100(2) |
| System/space group | Orthorhombic, $P 2_{1} 2_{1} 2_{1}$ | Orthorhombic, $\mathrm{P}_{2} 2_{1} 2_{1}$ |
| a (Å) | 7.500(9) | 7.440(4) |
| b (A) | 7.558(1) | 7.703(9) |
| c (A) | 26.180(5) | 25.956(6) |
| $\beta$ ( ${ }^{\circ}$ ) | 90 | 90 |
| $\mathrm{V}\left(\mathrm{A}^{3}\right)$ | 1484.24(3) | 1487.83(3) |
| Z/Z' | 4/1 | 4/1 |
| Density [ $\mathrm{g} / \mathrm{cm}^{3}$ ] | 1.313 | 1.309 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.915 | 0.913 |
| $\mathrm{T}_{\text {min }} / \mathrm{T}_{\text {max }}$ | 0.535/1.000 | 0.500/1.000 |
| F (000) | 632 | 632 |
| Crystal size [mm] | $0.26 \cdot 0.22 \cdot 0.05$ | $0.36 \cdot 0.16 \cdot 0.08$ |
| $2 \theta$ range [ ${ }^{\circ}$ ] | 3.4-77.7 | 6.0-77.8 |
| Completeness [\%] | 99.4 | 99.6 |
| Recorded refl. | 8059 | 22743 |
| Independent refl. | 2826 | 3009 |
| Flack x | 0.01(6) | 0.01(5) |
| Goodness-of-fit $\mathrm{F}^{2}$ | 1.053 | 1.109 |
| X-Ray Source | Cu Ka ( $\lambda=1.54184$ ) | $\mathrm{Cu} \mathrm{Ka}(\lambda=1.54184)$ |
| $\mathbf{R}_{1}$ [\%] /wR ${ }^{\text {[\%] }} / \mathrm{S}$ | 2.49/5.94/1.05 | 3.01/ 7.78/ 1.11 |

Table S12. Hydrogen bond distances and angles in (S, S) and (S, R) Pregabalin:malic acid cosystems. Bold written entities are charged at their oxygen or nitrogen atoms. A indices mark Pregabalin molecules, B malic acid molecules.

|  | D - H (Å) | H...A [Å] | D...A [Å] | D - H...A [ ${ }^{\text {] }}$ ] |
| :---: | :---: | :---: | :---: | :---: |
| (S, S)-1:4 |  |  |  |  |
| N1A-H7A...O3B | 0.90(2) | 2.46(3) | 3.065(2) | 125(2) |
| N1A-H7A...O4B | 0.90(2) | 1.94(2) | 2.792(6) | 157(2) |
| N1A-H8A...O3B | 0.93(2) | 2.11(2) | 2.976(1) | 155(2) |
| N1A-H9A... $04 B$ | 0.93(2) | 1.87(2) | 2.761 (0) | 161(2) |
| O1A-H1A...O2B | 0.90(4) | 1.85(4) | 2.733(4) | 168(3) |
| O1B-H1B...05B | 1.07(4) | 1.39(4) | $2.446(4)$ | 172(3) |
| O1B-H1B...04B | 1.07(4) | 2.59(3) | 3.154(3) | 112(2) |
| O3B-H5B...O2A | 0.81(3) | 1.88(3) | 2.688(9) | 175(3) |
| (S, R)-1:4 |  |  |  |  |
| N1A-H7A...O2B | 0.86(3) | 2.15(3) | 2.880(2) | 142(3) |
| N1A-H7A...O3B | 0.86(3) | 2.52(3) | 3.103(2) | 127(3) |
| N1A-H8A...O3B | 0.93(3) | 1.94(3) | 2.850(2) | 166(3) |
| N1A-H9A...O4B | 0.87(3) | 2.09(3) | 2.920(2) | 161(3) |
| N1A-H9A...O4B | 0.87(3) | 2.33(3) | 2.854(2) | 119(2) |
| O1A-H1A...05B | 0.97(4) | 1.64(4) | 2.609(2) | 174(4) |
| O1B-H1B...04B | 0.90(4) | 1.58(4) | 2.475(2) | 175(3) |
| O3B-H5B...O2A | 0.85(4) | 1.86(4) | 2.693(2) | 166(4) |

A Hirshfeld and electron density surface analysis was conducted (Figure S17). Close contacts and electron density maxima occur around the HB interaction sites. The close packing of the Pregabalin:malic acid systems leads to closer contacts around alkylic Pregabalin residues, which can be derived from both surface types. However, actual binding interactions with $\mathrm{C}_{\text {akyl }}-\mathrm{H}$ are highly unlikely. The high connectivity stemming from the large sum of possible HB donors and acceptors leads to close distances for large parts of both molecules which results in these short contacts.


Figure S16. The asymmetric unit of (S, S)-Pregabalin:malic acid is shown in a): top to bottom a-axis, b-axis, $c$-axis. In e) the same for (S, R)-Pregabalin:malic acid is presented. In b) and e) the packing of a $2 \cdot 2 \cdot 2$ unit cell from aaxis view is shown, for $(S, S)$ and $(S, R)$ respectively. In $c$ ) and $f$ ) the packing of a $2 \cdot 2 \cdot 2$ unit cell from $b$-axis view is shown, for $(S, S)$ and $(S, R)$ respectively. Carbon atoms are depicted in grey, hydrogen atoms in white, nitrogen atoms in blue and oxygen atoms in red.


Figure S17. Electron density and Hirshfeld surfaces of (S, S)- and (S, R) Pregabalin:malic acid species. a) and b) show the electron density surface of the Pregabalin and malic acid molecules in ( $S, S$ ) from two sides. c) and d) show the corresponding Hirshfeld surfaces. e) and f) show the electron density surface of Pregabalin and malic acid molecules in the ( $S, R$ ) from two sides. g) and h) show the corresponding Hirshfeld surfaces.


Figure S18. Powder pattern of (S)-Pregabalin:(S)-malic acid (red) and (S)-Pregabalin:(R)-malic acid (purple) as recorded compared simulated from single crystal data of (S)-Pregabalin:(S)-malic acid (green) and (S)-Pregabalin:(R)-malic acid (blue) in a range of $5^{\circ}-40^{\circ}$.

## 2 Physical properties

### 2.1 Determination of melting points with differential scanning calorimetry

Table S13 shows all melting points that could be determined via DSC.

Table S13. Melting points of all examined substances as determined by DSC.

| Substance name | Melting point [ $\left.{ }^{\circ} \mathrm{C}\right]$ |
| :--- | ---: |
| (S)-Pregabalin | 185 |
| (R)-Pregabalin | 187 |
| (rac)-Phenibut ${ }^{\star}$ | - |
| (S)-mandelic acid | 133 |
| (R)-mandelic acid | 133 |
| (S)-malic acid | 106 |
| (R)-malic acid | 106 |
| (S)-Pregabalin:(S)-mandelic acid | 138 |
| (R)-Pregabalin:(R)-mandelic acid | 132 |
| (S)-Pregabalin:(R)-mandelic acid | 111 |
| (R)-Pregabalin:(S)-mandelic acid | 105 |
| (S)-Pregabalin:(S)-malic acid | 85 |
| (S)-Pregabalin:(R)-malic acid | 95 |
| (R)-Phenibut:(S)-mandelic acid | 150 |

[^0]Corresponding DSC-curves are shown in Figures S19-S21.


Figure S19. Differential Scanning Calorimetry of a) (S)-Pregabalin b) (R)-Pregabalin c) (S)-mandelic acid, d) $(R)$-mandelic acid, e) (S)-malic acid and $f$ ) ( $R$ )-malic acid. The samples were heated with $5^{\circ} \mathrm{C} / \mathrm{min}$, specific ranges including the measured melting signal are depicted. Background has been subtracted.


Figure S20. Differential Scanning Calorimetry of a) (S, S)-Pregabalin:mandelic acid b) (R, R)-Pregabalin:mandelic acid c) (S, R)-Pregabalin:mandelic acid, d) (R, S)-Pregabalin:mandelic acid, e) (S, S)-Pregabalin:malic acid and f) $(S, R)$-Pregabalin:malic acid. The samples were heated with $5^{\circ} \mathrm{C} / \mathrm{min}$, specific ranges including the measured melting signal are depicted. Background has been subtracted.


Figure S21. Differential Scanning Calorimetry of (R)-Phenibut:(S)-mandelic acid. The sample was heated with $5^{\circ} \mathrm{C} / \mathrm{min}$, the range from $120^{\circ} \mathrm{C}-170^{\circ} \mathrm{C}$ is depicted. Melting starts at $150^{\circ} \mathrm{C}$. Background has been subtracted.

### 2.2 Determination of solubility with ${ }^{1} \mathrm{H}$-NMR-spectroscopy

Solubilities of the discussed compounds were determined by ${ }^{1} \mathrm{H}$-NMR. Powdery samples of each system were layered with phosphate buffer at pH 6.8 in such a way that the solid did not dissolve completely after multiple days. The dispersion was left in an incubator at $37^{\circ} \mathrm{C}$ and shaken at $60 \mathrm{~min}^{-1}$. After 1, 2 and in some cases 3 and 4 days $50 \mu \mathrm{~L}$ of solution were taken from the sample and added to $450 \mu \mathrm{~L}$ of $\mathrm{D}_{2} \mathrm{O}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ was measured from these samples. The processes were repeated twice for each system. Solubility was determined by comparing the integrals of water to that of the investigated substance. For this, a prevalent signal was integrated and the corresponding value of the water signal was taken from the spectrum. The following Equation 1 was used to determine the solubility in $\mathrm{gL}^{-1}$.

$$
\begin{equation*}
S_{p}=\left(\frac{M_{p}}{\frac{I_{w}}{2} M_{w}} \rho_{w}\right) \tag{1}
\end{equation*}
$$

Where $\mathrm{S}_{\mathrm{p}}$ is the products solubility in the buffer at pH 6.8 and $37^{\circ} \mathrm{C}$ in $\mathrm{gL}^{-1}, \mathrm{M}_{\mathrm{p}}$ is the molar mass of the investigated substance in $\mathrm{gmol}^{-1}, \mathrm{I}_{\mathrm{w}}$ is the value of the water integral taken from the ${ }^{1} \mathrm{H}$-NMR spectrum, $\mathrm{M}_{\mathrm{w}}$ the molar mass of water and $\rho_{\mathrm{w}}$ the density of water at $37^{\circ} \mathrm{C}$ of $993.33 \mathrm{gL}^{-1}$. The used model does not consider trace impurities of water in the $\mathrm{D}_{2} \mathrm{O}$ as well as the error of the micropipette. For each system 4 values were calculated based on 2 spectra recorded after 1 and 2 days or 3 and 4 days, based on when stability of the received values was reached. The average of these values is given as the final solubility. A standard deviation on these 4 values was calculated by Equation 2.

$$
\begin{equation*}
V=\sqrt{\frac{\sum(x-\bar{x})^{2}}{(n-1)}} \tag{2}
\end{equation*}
$$

With V as the standard deviation, x the average of the 4 values, $\bar{x}$ the singular values and n the number of all values, four. Table S14 shows solubilities and their standard deviations as well as the borders set for the integral of the Pregabalin or Phenibut signal. The water signal was always integrated between 4.830 and 4.750 ppm. A ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum for each compound is shown in Chapter 5.

Table S14. Solubilities and their standard deviations for the examined compounds and integral borders set for the ${ }^{1} \mathrm{H}$-NMR signal integrals of the investigated substance in the spectra.

| Substance name | Solubility/Standard <br> deviation [gL-1] | Integral borders [ppm] |
| :--- | ---: | ---: |
| (S)-1 | $35 \pm 0.4$ | $0.865 / 0.870-0.910$ |
| (R)-1 | $33 \pm 0.3$ | $0.865-0.915$ |
| (rac)-2*1 | $18 \pm 1$ | $2.520-2.665$ |
| (S)/(R)-3 | $203 \pm 3$ | $7.370-7.500$ |
| (S)/(R)-4 | $2061 \pm 76$ | $2.760-2.900$ |
| (S, S)-1:3 | $37 \pm 1$ | $0.910 / 0.915$ |
| (R, R)-1:3 | $40 \pm 4$ | $0.865-0.915$ |
| (S, R)-1:3 | $316 \pm 18$ | $0.850 / 0.860 / 0.890-0.910 / 0.920 / 0.950$ |
| (R, S)-1:3 | $307 \pm 6$ | $0.855 / 0.860-0.905 / 0.910$ |
| (S, S)-1:4*2 | $>800$ | $0.825 / 0.845-0.890 / 0.910$ |
| (S, R)-1:4*2 | $>800$ | $0.835 / 0.845 / 0.850-0.900 / 0.910 / 0.915$ |
| (R, S)-2:3 | $71 \pm 3$ | $2.650-2.800$ |

${ }^{* 1}=$ solubility equilibrium was reached after 3 and 4 days respectively. *2 $=$ a maximum solubility could not be determined. Complete dissolution of the substance always occurred. At higher concentrations a highly viscous substance was received with which reliable measurement was impossible.

## 3 Lattice energies

### 3.1 Model description and verification

To calculate lattice energies, an adapted method based on Marchese Robinson et al. ${ }^{[8]}$ and Voronin et al. ${ }^{[9]}$ was used. Step by step, the following actions were performed:

- Recording of crystal structure in .cif format and subsequent conversion to QE readable .qein input format.
- Self-consistent field (SCF) calculations were performed on the files for (S)-Pregabalin, (rac)-Phenibut, (R, R)/(R, S)-Pregabalin:mandelic acid, (S, S)-Pregabalin:malic acid and ( $\mathrm{R}, \mathrm{S}$ )-Phenibut:mandelic acid to determine a uniform energy convergence threshold (ecutrho) and force convergence threshold (ecutwfc) (Table S15).
- K-points were set to form a grid of about 20-25 $\AA^{3}$ around the first Brillouin zone for each compound, see Table S15.
- Geometric optimization using the 'variable cell relax' (vc-relax) command in QE of the examined structures as well as some literature known validation examples was
performed with the determined threshold and K-point grids to receive the ideal static solid energy $\mathrm{E}_{\text {iss }}$.
- Geometry parameters for each singular geometrically independent molecule in each structure were gathered from the previously received outputs. For each of these molecules, a further optimization was performed in a fixed $20 \AA^{3}$ otherwise empty cell with the 'relax' command in QE to calculate the energy of the ideal static gas for this molecule, Eisg.
- Equation 3, $\mathbf{5}$ or $\mathbf{6}$ were then used to calculate the lattice energy $\mathrm{E}_{\text {lat }}$ from the received values. These equations shall be explained in detail in the upcoming remarks.

SCF calculations were performed on the described systems with an energy convergence threshold starting at 30 Ry and a force convergence threshold starting at 3 Ry subsequently adding 10 Ry or 1 Ry up until 100 Ry and 10 Ry were reached. Based on the received energy values for these calculations it was determined that 60 Ry and 6 Ry were acceptable values to reach convergence for all examined systems.

Table S15. Variables ecutrho, ecutwfc and chosen K-point grid for each calculated sample. All values were used in SCF calculations and geometry optimizations. For the simulated gas phase calculations K-points were neglected.

| Sample | ecutroh [Ry] | ecutwfc [Ry] | K-point grid |
| :---: | :---: | :---: | :---: |
| Aspirin ${ }^{[10]}$ |  |  | 2.4.2 |
| Glycine- ${ }^{[11]}$ |  |  | 5-2.5 |
| Glycine- $\beta^{[12]}$ |  |  | 5-4.5 |
| Glycine- ${ }^{[13]}$ |  |  | 4-4.4 |
| Benzoic acid ${ }^{[14]}$ |  |  | 4-5.1 |
| Naphthalene ${ }^{[15]}$ |  |  | 3-4.3 |
| Carbendazim Maleate ${ }^{[9]}$ |  |  | 3-3.1 |
| Fenamic acid ${ }^{[16]}$ |  |  | 3-3.2 |
| (S)-Pregabalin/(1) | 60 | 6 | 4-3.2 |
| (R)-Pregabalin/(1) |  |  | 4-3.2 |
| (rac)-Phenibut/(2) |  |  | 3-4 1 |
| (S, S)-Pregabalin:malic acid/(1:4) |  |  | 3-3.1 |
| (S, R)-Pregabalin:malic acid/(1:4) |  |  | 3-3.1 |
| (S, S)-Pregabalin:mandelic acid/(1:3) |  |  | 4-1.3 |
| (R, R)-Pregabalin:mandelic acid/(1:3) |  |  | 4-1.3 |
| (S, R)-Pregabalin:mandelic acid/(1:3) |  |  | 4-1.3 |
| (R, S)-Pregabalin:mandelic acid/(1:3) |  |  | 4-1.3 |
| (R, S)-Phenibut:mandelic acid/(2:3) |  |  | 3-4.2 |

Naphthalene, aspirin and benzoic acid were chosen as model compounds that are commonly used to verify energy models, they show similar capabilities for intermolecular interactions as the investigated compounds. Glycine polymorphs were chosen because they are well investigated amino acids that exhibit zwitterionicity and as polymorphs should be energetically close regarding each other. Carbendazim maleate was examined by Voronin et al. and serves
to verify that the model works for multicomponent entities. Fenamic acid exhibits $Z^{\prime}=2$ and was energetically investigated by Yang et al. in the past, ${ }^{[17]}$ thus enabling its use as a model compound for systems with this characteristic. The necessary crystallographic information files (.cif) for these compounds were gathered from the crystal structure database via Mercury 2020.2.0. ${ }^{[9-16]}$ A geometric optimization and subsequent simulated gas phase of the geometrically optimized singular molecules was performed using the determined parameters to receive values for $\mathrm{E}_{\text {iss }}$ and $\mathrm{E}_{\text {isg, }}$, sufficient to calculate $\mathrm{E}_{\text {lat }}$ as shown in Equation 3 for naphthalene, aspirin, benzoic acid and the glycine polymorphs.

$$
\begin{equation*}
E_{\text {lat }}=\frac{E_{i s s}}{Z}-E_{i s g} \tag{3}
\end{equation*}
$$

Where $Z$ is the crystallographic $Z$, describing the formular units in the unit cell. For further validation purposes, it was assumed that systems that are highly comparable by means of the participating molecules could be used to obtain $\Delta \mathrm{E}_{\text {lat }}$ values without the need to calculate $\mathrm{E}_{\text {isg }}$ by variation of Equation 3 under the assumptions that $\mathrm{E}_{\text {isg }}$ values for the same molecules would be negligible as shown in Equation 4.

$$
\begin{gather*}
\Delta E_{\text {lat }}=E_{\text {lat } 1}-E_{\text {lat } 2} \\
\Delta E_{\text {lat }}=\left(\frac{E_{\text {iss } 1}}{Z_{1}}-E_{\text {isg }}\right)-\left(\frac{E_{i s s 2}}{Z_{2}}-E_{\text {isg }}\right) \\
\Delta E_{\text {lat }}=\frac{E_{\text {iss } 1}}{Z_{1}}-\frac{E_{\text {iss2 }}}{Z_{2}} \tag{4}
\end{gather*}
$$

These values could then be used to verify lattice energy differences calculated by Elat values obtained via Equation 3. It was further attempted to consider the influence of $Z^{\prime}$. In systems where Z' > 1 Eiss has to be adjusted by Z' via division to obtain correct values for $E_{\text {lat }}$ in the presented model (Equation 5).

$$
\begin{equation*}
E_{\text {lat }}=\frac{E_{i s s}}{Z}-\frac{E_{i s g}}{Z^{\prime}} \tag{5}
\end{equation*}
$$

Finally, for multicomponent species each crystallographcally distinct molecule has to be considered by their $\mathrm{E}_{\text {isg }}$ values to obtain $\mathrm{E}_{\text {lat }}$, which can be achieved by summation of the different contributions. Equation 6 considers all discussed influences.

$$
\begin{equation*}
E_{l a t}=\frac{E_{i s s}}{Z}-\frac{\sum E_{i s g n}}{Z^{\prime}} \tag{6}
\end{equation*}
$$

Tables $\mathbf{S 1 6}$ shows $\mathrm{E}_{\text {iss }}$ and Eisg values on the model compounds obtained by the described model, Table S17 shows corresponding lattice energies.

Table S16. Resulting values of $E_{i s s}$ and $E_{i s g}$ for the validation compounds. In carbendazim maleate, carbendazim entities are marked with ' $C$ ' and maleate entities are marked with ' $M$ '. In fenamic acid, the crystallographically independent molecules are marked as $A$ and $B$.

| Sample | Z | E $_{\text {iss }}$ [Ry] | E $_{\text {isg }}$ (neutral) [Ry] | $\mathbf{E}_{\text {isg }}$ (zwitt./charge) <br> [Ry] |
| :--- | :--- | ---: | ---: | ---: |
| Aspirin | 4 | -1348.01234 | -336.89571 | - |
| Benzoic acid | 4 | -864.49243 | -216.03921 | - |
| Naphthalene | 2 | -380.81223 | -190.34592 | - |
| Carbendazim | 4 | -2238.98103 | $-317.81712^{\mathrm{C}}$ | $-318.65171^{\mathrm{C}}$ |
| maleate |  | $-241.69462^{\mathrm{M}}$ | $-240.74749^{\mathrm{M}}$ |  |
| Fenamic acid | 4 | -1405.17232 | $-351.18464^{\mathrm{A}}$ | - |
| Glycine- $\alpha$ | 4 | -580.77370 | $-351.18407^{\mathrm{B}}$ | - |
| Glycine- $\boldsymbol{\beta}$ | 2 | -145.06565 | -145.06563 |  |
| Glycine-ץ | 3 | -435.58351 | -145.06570 | -145.06570 |

Table S17. Resulting values of Elat for the validation compounds. Literature values were obtained from the cited sources.

| Sample |  | $\begin{gathered} \hline \text { Eharge) } \\ \text { (zwitt./ } \\ \text { [Ry] } \\ \hline \end{gathered}$ | $\mathrm{E}_{\text {lat }}$ (neutral) <br> [ $\mathrm{kJmol}^{-1}$ ] | $\begin{gathered} \hline \mathrm{E}_{\text {lat }} \text { (zwitt./ } \\ \text { charge) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \end{gathered}$ | $\begin{gathered} \mathrm{E}_{\text {lat }} \text { (lit.) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Aspirin | -0.10737 |  | -140.96 |  | $-131.27{ }^{[18]]^{*}}$ |
| Benzoic acid | -0.08389 |  | -110.14 |  | $-96.0{ }^{[19]}$ |
| Naphthalene | -0.06020 | - | -79.03 | - | $-79.4{ }^{[19]}$ |
| Carbendazim maleate | -0.23351 | -0.34606 | -306.54 | -454.29 | $\begin{aligned} & -278.3^{[9] *} \\ & -625.0^{[9] *} \end{aligned}$ |
| Fenamic acid | -0.10873 | - | -142.73 | - | $-136.5^{[17]}$ |
| Glycine-a | -0.12777 | -0.12779 | -167.73 | -167.76 | $-144.93{ }^{[20]}$ |
| Glycine- $\beta$ | -0.12606 | -0.12605 | -165.48 | -165.48 | $-143.13^{[20]}$ |
| Glycine-Y | -0.14329 | -0.12829 | -188.11 | -168.41 | $-144.1{ }^{[20]}$ |

${ }^{*}=$ the cited sources give a plethora of values obtained through differing models and equations. The closest one to the calculated results was picked for this presentation. For Carbendazim maleate the lattice energy values based on neutral molecules and charged molecules are given.

The model's accuracy for polymorphic substances was tested on glycine polymorphs assuming zwitterionic conditions for $\mathrm{E}_{\text {isg }}$ using Equations 3 \& 4 (Table S18).

Table S18. $\Delta$ Elat as calculated by Equation 3 \& Equation 4 for permutations of the Glycine polymorphs, the zwitterionic state energy values were used.

| Sample | $\Delta \mathbf{E}_{\text {lat }}$ (Eq. 3) $\left[\mathrm{kJmol}^{-1}\right]$ | $\Delta \mathrm{E}_{\text {lat }}$ (Eq. 4) $\left[\mathrm{kJmol}^{-1}\right]$ | Deviation [kJmol ${ }^{-1}$ ] |
| :--- | ---: | ---: | ---: |
| Glycine- - <br> Glycine- $\beta$ | -2.28 | -2.19 | 0.09 |
| Glycine- - <br> Glycine- | 0.58 | 0.65 | 0.17 |
| Glycine- $\beta-$ <br> Glycine- | 2.93 | 2.77 | 0.16 |

### 3.2 Compound categorization and model application

After verification of the used model its principles were applied to the central Pregabalin and Phenibut based compounds. First, the compounds were categorized based on some of their properties which could influence the calculation results. Category A contains (S)-Pregabalin, (R)-Pregabalin and (rac)-Phenibut. These are the simplest examined compounds as they exhibit only one type of molecule in the unit cell, $Z^{\prime}=1$ and each molecule is in its default zwitterionic state. In Category B only (R, S)-Phenibut:mandelic acid is placed. Here, two different kinds of molecules occur in the unit cell but $Z$ ' remains 1 and each molecule keeps its default state with zwitterionic Phenibut and neutral mandelic acid. (S, S)- and (S, R)-Pregabalin:malic acid are placed in Category C. Here, two types of molecules occur in the cell, $Z^{\prime}=1$ but each species contains a formal charge. Category D is composed of (S, S)and (R, R)-Pregabalin:mandelic acid. These systems are formed from two different molecular species and four crystallographically inequivalent molecules but $Z^{\prime}=2$ and default charge status on the molecules is retained. Lastly, Category E consists of (S, R)- and (R, S)-Pregabalin:mandelic acid. All previously established complications are applied to this category, as there are two different kinds of molecules and four crystallographically inequivalent ones, $Z^{\prime}=2$ and formal charges are present on each molecule. A comprehensive overview on the categories, the involved samples and properties is shown in Table S19.

Table S19. Characteristics of the determined categories: contained samples, number of different molecular species, Z' value as determined by PLATON software and information on whether the molecular species are charged (neither zwitterionic nor neutral).

|  | Samples | Molecule types | Z' | Charge |
| :--- | :--- | ---: | ---: | ---: |
| Category A | (S)-, (R)-Pregabalin, (rac)-Phenibut | 1 | 1 | No |
| Category B | (R)-Phenibut:(S)-mandelic acid | 2 | 1 | No |
| Category C | (S, S)-, (S, R)-Pregabalin:malic acid | 2 | 1 | Yes |
| Category D | (S, S)-, (R, R)-Pregabalin:mandelic acid | 2 | 2 | No |
| Category E | (S, R)-, (R, S)-Pregabalin:mandelic acid | 2 | 2 | Yes |

It was determined that Category A would not require any additional effort to apply the used model on, as its characteristics are the same as those of the model compounds. For Category B - E it was assumed that $\mathrm{E}_{\text {isg }}$ would be determined by summation of $\mathrm{E}_{\mathrm{isg} A}+\mathrm{E}_{\mathrm{isg} B}+\ldots+\mathrm{E}_{\mathrm{isgn}}$ for all geometrically independent molecules. For each category it was furthermore examined whether neutral, zwitterionic or, if applicable, charged states on isolated molecules would lead to the most consistent results. Additionally, for Category D and E it was attempted to consider the influence of $Z^{\prime}$.

## Category A

Table S20. Resulting values of Eiss and Eisg for the Category A compounds.

| Sample | Z | $\mathbf{E}_{\text {iss }}$ [Ry] | $\mathbf{E}_{\text {isg }}$ (neutral) [Ry] | $\mathbf{E}_{\text {isg }}$ (zwitt.) [Ry] |
| :---: | :--- | :--- | ---: | ---: |
| (S)-Pregabalin (1) | 4 | -1071.82432 | -267.79802 | -267.80748 |
| (R)-Pregabalin (1) | 4 | -1071.82525 | -267.79161 | -267.80764 |
| (rac)-Phenibut (2) | 8 | -2396.20653 | -299.23844 | -299.12074 |

Table S21. Resulting values of Elat as calculated by Equation 6 for the Category A compounds.

| Sample | $\mathbf{E}_{\text {lat }}$ (neutral) <br> [Ry] | $\mathbf{E}_{\text {lat }}$ (zwitt.) <br> [Ry] | $\mathbf{E}_{\text {lat }}$ (neutral) <br> [kJmol ${ }^{-1}$ ] | $\mathbf{E}_{\text {lat }}$ (zwitt.) <br> [kJmol <br> [ |
| :--- | :---: | :---: | :---: | ---: |
| (S)-Pregabalin (1) | -0.15806 | -0.14860 | -207.50 | -195.08 |
| (R)-Pregabalin (1) | -0.16470 | -0.14867 | -216.21 | -195.17 |
| (rac)-Phenibut (2) | -0.16238 | -0.28008 | -213.16 | -367.68 |

Table S22. Comparison of $\Delta E_{l a t}$ values obtained by Equation 6 and Equation 4 for Category A compounds (S)and (R)-Pregabalin.

|  | $\begin{gathered} \hline \mathrm{E}_{\text {lat }} \text { (Eq. 3) } \\ \text { [kJmol } \end{gathered}$ | $\begin{gathered} \Delta \mathrm{E}_{\text {lat }} \text { (Eq. 4) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \end{gathered}$ | Deviation [ $\mathrm{kJmol}^{-1}$ ] |
| :---: | :---: | :---: | :---: |
| (S)-1 - (R)-1 (neutral) | -8.71 |  | 8.40 |
| (S)-1 - (R)-1 (zwitt.) | -0.09 |  | 0.22 |

## Category B \& C

Table S23. Resulting values of $E_{i s s}$ and $E_{i s g}$ for the Category B \& Compounds crystallographically independent occurring molecular species.

| Sample | $\mathrm{E}_{\text {iss }}$ [Ry] | $\mathrm{E}_{\text {isg }}$ (neutral) [Ry] | $\begin{gathered} \hline \text { Eisg (zwitt.) } \\ \text { [Ry] } \\ \hline \end{gathered}$ | $\mathrm{E}_{\text {isg }}$ (charged) [Ry] |
| :---: | :---: | :---: | :---: | :---: |
| (S, S)-1:4 (Z = 4) |  |  |  |  |
| (S)-Pregabalin | -2213.33805 | -267.79195 | -267.80910 | -268.61130 |
| (S)-malic acid |  | -285.25372 | - | -284.27132 |
| (S, R)-1:4 (Z = 4) |  |  |  |  |
| (S)-Pregabalin | -2213.34508 | -267.79369 | -267.80880 | -268.61136 |
| (R)-malic acid |  | -285.25091 | - | -284.22893 |
| (R, S)-2:3 (Z = 2) |  |  |  |  |
| (R)-Phenibut | -1154.28106 | -299.24119 | -299.24904 |  |
| (S)-mandelic acid |  | -277.63003 | - |  |

Table S24. Resulting values of Elat as calculated by Equation 6 for the Category B \& C compounds using the Eisg values of singular molecules depicted in Table S23.

| Sample | $\mathrm{E}_{\text {lat }}$ (neutral) [Ry] | $\mathrm{E}_{\text {lat }}$ (zwitt.) [Ry] | $\mathrm{E}_{\text {lat }}$ (charged) [Ry] |
| :---: | :---: | :---: | :---: |
| (S, S)-1:4 | -0.28884 | -0.27169 | -0.45190 |
| (S, R)-1:4 | -0.29166 | -0.27656 | -0.49598 |
| (R, S)-2:3 | -0.26931 | -0.26146 |  |
|  | $\mathrm{E}_{\text {lat }}$ (neutral) [ $\mathrm{kJmol}^{-1}$ ] | $E_{\text {lat }}$ (zwitt.) <br> [kJmol ${ }^{-1}$ ] | $\mathrm{E}_{\text {lat }}$ (charged) <br> [ $\mathrm{kJmol}^{-1}$ ] |
| (S, S)-1:4 | -379.17 | -356.66 | -593.23 |
| (S, R)-1:4 | -382.88 | -363.05 | -651.10 |
| (R, S)-2:3 | -353.54 | -343.23 |  |

Table S25. Comparison of $\Delta E_{\text {lat }}$ values received by Equation 6 and Equation 4 for Category Compounds (S, S)and (S, R)-Pregabalin:malic acid.

| Sample | $\begin{gathered} \hline \Delta \mathrm{E}_{\text {lat }} \text { (Eq. 3) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \\ \hline \end{gathered}$ | $\begin{gathered} \hline \Delta \mathrm{E}_{\text {lat }} \text { (Eq. 4) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \end{gathered}$ | Deviation [ $\mathrm{kJmol}^{-1}$ ] |
| :---: | :---: | :---: | :---: |
| (S, S)-1:4-(S, R)-1:4 (neutral) | 3.71 |  | 1.40 |
| (S, S)-1:4-(S, R)-1:4 (zwitt.) | 6.39 | 2.31 | 4.08 |
| (S, S)-1:4-(S, R)-1:4 (charged) | 57.87 |  | 55.66 |

## Category D \& E

Table S26. Resulting values of $E_{i s s}$ and $E_{i s g}$ for the Category D \& E compounds crystallographically independent occurring molecular species.

| Sample | $\mathrm{E}_{\text {iss }}$ [Ry] | $\mathrm{E}_{\text {isg }} \text { (neutral) }$ [Ry] | $\begin{gathered} \hline \mathrm{E}_{\text {isg }} \text { (zwitt.) } \\ \text { [Ry] } \end{gathered}$ | $\begin{gathered} \hline \mathrm{E}_{\text {isg }} \text { (charged) } \\ \text { [Ry] } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: |
| (S, S)-1:3 (Z = 4) |  |  |  |  |
| (S)-Pregabalin 1 |  | -267.79208 | -267.80897 |  |
| (S)-Pregabalin 2 | -2182 72832 | -267.79387 | -267.80898 |  |
| (S)-mandelic acid 1 | -2182.72832 | -277.62920 |  |  |
| (S)-mandelic acid 2 |  | -277.62893 | - |  |
| (R, R)-1:3 ( $\mathrm{Z}=4$ ) |  |  |  |  |
| (R)-Pregabalin 1 |  | -267.79375 | -267.80898 |  |
| (R)-Pregabalin 2 | -2182 73030 | -267.79208 | -267.80900 |  |
| (R)-mandelic acid 1 | -2182.73030 | -277.62918 |  |  |
| (R)-mandelic acid 2 |  | -277.82695 | - |  |
| (S, R)-1:3 (Z = 4) |  |  |  |  |
| (S)-Pregabalin 1 |  | -267.79976 | -267.80809 | -268.64999 |
| (S)-Pregabalin 2 | 218268327 | -267.81182 | -267.80928 | -268.59870 |
| (R)-mandelic acid 1 | -2182.68327 | -277.62918 | - | -276.65999 |
| (R)-mandelic acid 2 |  | -277.63009 | - | -276.63286 |
| (R, S)-1:3 (Z = 4) |  |  |  |  |
| (R)-Pregabalin 1 |  | -267.79397 | -267.80932 | -268.59859 |
| (R)-Pregabalin 2 | -218268421 | -267.79977 | -267.80897 | -268.65003 |
| (S)-mandelic acid 1 | -2182.68421 | -277.62918 | - | -276.64477 |
| (S)-mandelic acid 2 |  | -277.62575 | - | -276.63280 |

Table S27. Resulting values of $E_{l a t}$ as calculated by Equation 6 for the Category $\mathbf{D}$ \& $\mathbf{E}$ compounds using the $E_{i s g}$ values of singular molecules depicted in Table S26.

| Sample | $\mathrm{E}_{\text {lat }}$ (neutral) [Ry] | Elat (zwitt.) [Ry] | Elat (charged) [Ry] |
| :---: | :---: | :---: | :---: |
| (S, S)-1:3 | -0.26004 | -0.24404 |  |
| (R, R)-1:3 | -0.26095 | -0.24452 |  |
| (S, R)-1:3 | -0.23540 | -0.23208 | -0.40005 |
| (R, S)-1:3 | -0.24672 | -0.23445 | -0.40796 |
|  | $\mathrm{E}_{\text {lat }}$ (neutral) [ $\mathrm{kJmol}^{-1}$ ] | $\mathrm{E}_{\text {lat }} \text { (zwitt.) }$ $\left[\mathrm{kJmol}^{-1}\right]$ | $\mathrm{E}_{\text {lat }}$ (charged) <br> [ $\mathrm{kJmol}^{-1}$ ] |
| (S, S)-1:3 | -341.37 | -320.36 | - |
| (R, R)-1:3 | -342.09 | -320.99 | - |
| (S, R)-1:3 | -309.02 | -305.66 | -525.17 |
| (R, S)-1:3 | -323.88 | -307.77 | -535.55 |

Table S28. Comparison of $\Delta E_{l a t}$ values received by Equation 6 and Equation 4 for Category $\boldsymbol{D}$ \& $\boldsymbol{E}$ compounds inside the respective categories and between both categories in varying permutations.

| Sample | $\begin{gathered} \hline \Delta \mathrm{E}_{\text {lat }} \text { (Eq. 5) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \\ \hline \end{gathered}$ | $\begin{gathered} \hline \Delta \mathrm{E}_{\text {lat }} \text { (Eq. 4) } \\ {\left[\mathrm{kJmol}^{-1}\right]} \\ \hline \end{gathered}$ | Deviation [ $\mathrm{kJmol}^{-1}$ ] |
| :---: | :---: | :---: | :---: |
| $\begin{aligned} & \hline \text { (S, S)-1:3 - (R, R)-1:3 (neutral) } \\ & \text { (S, S)-1:3 - (R, R)-1:3 (zwitt.) } \end{aligned}$ | $\begin{aligned} & \hline 0.72 \\ & 0.63 \end{aligned}$ | 0.65 | 0.07 0.02 |
| (S, R)-1:3-(R, S)-1:3 (neutral) (S, R)-1:3-(R, S)-1:3 (zwitt.) (S, R)-1:3-(R, S)-1:3 (charged) | $\begin{array}{r} \hline \hline 14.86 \\ 2.11 \\ 10.38 \end{array}$ | 0.31 | $\begin{array}{r} 14.55 \\ 1.80 \\ 10.07 \\ \hline \end{array}$ |
| (S, S)-1:3-(S, R)-1:3 (neutral) <br> (S, S)-1:3 - (S, R)-1:3 (zwitt.) | $\begin{aligned} & \hline-32.35 \\ & -15.71 \end{aligned}$ | -14.79 | $\begin{array}{r} \hline 17.56 \\ 0.92 \end{array}$ |
| (S, S)-1:3-(R, S)-1:3 (neutral) <br> (S, S)-1:3 - (R, S)-1:3 (zwitt.) | $\begin{aligned} & \hline-17.49 \\ & -12.59 \end{aligned}$ | -14.48 | 3.01 1.89 |
| $\begin{aligned} & \hline \text { (R, R)-1:3-(S, R)-1:3 (neutral) } \\ & \text { (R, R)-1:3-(S, R)-1:3 (zwitt.) } \end{aligned}$ | $\begin{aligned} & \hline-33.07 \\ & -16.33 \end{aligned}$ | -15.44 | $\begin{array}{r} \hline 17.63 \\ 0.89 \end{array}$ |
| $\begin{aligned} & \hline(\mathrm{R}, \mathrm{R})-1: 3 \text { - (R, S)-1:3 (neutral) } \\ & \text { (R, R)-1:3 - (R, S)-1:3 (zwitt.) } \end{aligned}$ | $\begin{aligned} & -18.21 \\ & -13.22 \end{aligned}$ | -15.13 | $\begin{aligned} & \hline \hline 3.08 \\ & 1.91 \end{aligned}$ |

It was determined that the most overall consistent results can be obtained by application of zwitterionicity on Pregablin/Phenibut molecules and neutral charge status on mandelic- or malic acid.

## 4 Applications: enantiopurification of (rac)-Pregabalin hydrate

Detailed PXRD pattern comparisons of products received during enantiopurification steps IIII are presented in Figures S22 - S24.


Figure S22. Powder pattern of the milling product of racemic Pregabalin hydrate and (S)-mandelic acid (green). A comparison to the two co-crystalline compounds is shown for (S, R)-Pregabalin:mandelic acid (red) as well as (S, S)-Pregabalin:mandelic acid (blue). Thus, the milling product is identified as a mixture. The depicted angle range is $5^{\circ}-40^{\circ}$.


Figure S23. Powder pattern of the milling product of racemic Pregabalin hydrate and (R)-mandelic acid after subsequent washing and drying procedure from $5^{\circ}-40^{\circ}$. The signals of the more soluble heterochiral co-crystalline entity vanish from red - blue pattern, easily visible in the lowest ${ }^{\circ}$ signal.


Figure S24. Powder pattern of the final (R)-Pregabalin (green) after removal of (R)-mandelic acid) compared to a simulated pattern from single crystal data (blue) and the recorded pattern of the racemic Pregabalin hydrate (red) in a range of $5^{\circ}-40^{\circ}$.

## 5 Addendum

Table S29. Bijvoet-analysis for confirmation of chiral purity conducted with PLATON software.

|  | (S)-1 | (R)-1 | (S, S)-1:3 | (R, R)-1:3 | (S, R)-1:3 | (R, S)-1:3 | (S, S)-1:4 | (S, R)-1:4 | (R, S)-2:3 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Flack x | 0.03(11) | -0.04(19) | 0.15(11) | -0.02(5) | -0.05(12) | -0.15(7) | 0.01(6) | 0.01(5) | 0.00(10) |
| Parsons z | 0.06(11) | 0.00(18) | 0.14(9) | -0.04(5) | 0.05(8) | -0.17(6) | -0.01(6) | -0.01(4) | -0.01(8) |
| Bijvoet Pairs | 663 | 696 | 3038 | 3150 | 3209 | 2991 | 1133 | 1210 | 1412 |
| Coverage | 84 | 88 | 86 | 89 | 91 | 85 | 86 | 91 | 87 |
| DiffCalcMax. | 6.47 | 6.58 | 18.43 | 18.07 | 28.52 | 30.71 | 38.34 | 39.17 | 10.17 |
| Outlier Crit. | 12.94 | 13.16 | 36.85 | 36.14 | 57.03 | 61.43 | 76.67 | 78.33 | 20.34 |
| Sigma Crit. | 0.25 | 0.25 | 0.25 | 0.25 | 0.25 | 0.25 | 0.25 | 0.25 | 0.25 |
| Select Pairs | 84 | 27 | 86 | 528 | 99 | 322 | 280 | 464 | 167 |
| Number Plus | 55 | 18 | 55 | 352 | 58 | 208 | 179 | 313 | 106 |
| Number Minus | 29 | 9 | 31 | 176 | 41 | 114 | 101 | 151 | 61 |
| Slope | 0.908 | 1.525 | 0.709 | 1.024 | 1.175 | 1.358 | 0.949 | 0.951 | 1.091 |
| Sample Size | 653 | 686 | 3028 | 3140 | 3199 | 2981 | 1123 | 1200 | 1402 |
| Corr. Coeff. | 0.997 | 0.999 | 0.999 | 1.000 | 0.999 | 1.000 | 0.999 | 0.998 | 0.997 |
| Intercept | 0.018 | -0.065 | 0.014 | -0.034 | -0.004 | -0.065 | -0.076 | 0.033 | -0.151 |
| Slope | 0.854 | 0.840 | 0.911 | 0.947 | 1.009 | 1.034 | 0.870 | 0.944 | 0.985 |
| Student_T Nu | 100 | 32 | 9 | 32 | 7 | 32 | 100 | 99 | 19 |
| Select Pairs | 663 | 696 | 3038 | 3150 | 3209 | 2991 | 1133 | 1210 | 1412 |
| Theta_min | 9.32 | 9.30 | 5.49 | 3.23 | 2.95 | 2.96 | 8.50 | 8.46 | 7.76 |
| Theta_Max | 76.05 | 76.13 | 77.15 | 76.90 | 77.53 | 75.38 | 76.51 | 77.79 | 76.14 |
| P2 (true) | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 |
| P3 (true) | 1.000 | 0.995 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 |
| P3 (rac-twin) | 0.3E-04 | 0.005 | 0.3E-03 | 0.1E-32 | 0.9E-06 | 0.1E-21 | 0.2E-17 | 0.9E-30 | 0.8E-08 |
| P3 (false) | 0.2E-20 | 0.3E-09 | 0.3E-21 | 0.1E-121 | 0.22E-23 | 0.4E-70 | 0.3E-70 | 0.2E-120 | 0.2E-31 |
| G | 0.8773 | 0.9606 | 0.7438 | 1.0754 | 1.0198 | 1.3180 | 1.0004 | 0.9897 | 1.0117 |
| G (su) | 0.1916 | 0.2946 | 0.1728 | 0.0871 | 0.1929 | 0.1270 | 0.1107 | 0.0840 | 0.1652 |
| Hooft y | 0.06(10) | 0.02(15) | 0.13(9) | -0.04(4) | -0.01(10) | -0.16(6) | 0.00(6) | 0.01(4) | -0.01(8) |

## ${ }^{1}$ H-NMR spectroscopy

(S)-Pregabalin


Figure S25. ${ }^{1} \mathrm{H}$-NMR spectrum of (S)-Pregabalin recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 0.25 ppm and 3.75 ppm is depicted, the signals are marked by letters according to the position on the molecule.
(R)-Pregabalin


Figure S26. ${ }^{1} \mathrm{H}$-NMR spectrum of (R)-Pregabalin recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 0.25 ppm and 3.75 ppm is depicted, the signals are marked by letters according to the position on the molecule.


Figure S27. ${ }^{1} \mathrm{H}$-NMR spectrum of (rac)-Phenibut recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 2.00 ppm and 7.75 ppm is depicted, the signals are marked by letters according to the position on the molecule.
(S)-Mandelic Acid


Figure S28. ${ }^{1} \mathrm{H}$-NMR spectrum of (S)-mandelic acid recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 4.8 ppm and 7.8 ppm is depicted, the signals are marked by letters according to the position on the molecule.
(R)-Mandelic acid


Figure S29. ${ }^{1} \mathrm{H}$-NMR spectrum of (R)-mandelic acid recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 4.8 ppm and 7.8 ppm is depicted, the signals are marked by letters according to the position on the molecule.
(S)-Malic acid


Figure S30. ${ }^{1} \mathrm{H}$-NMR spectrum of (S)-Malic acid recorded in $\mathrm{D}_{2} \mathrm{O}$. The range between 2.6 ppm and 4.7 ppm is depicted, the signals are marked by letters according to the position on the molecule.
(R)-Malic acid


Figure S31. ${ }^{1} \mathrm{H}$-NMR spectrum of (R)-Malic acid recorded in $D_{2} O$. The range between 2.6 ppm and 4.7 ppm is depicted, the signals are marked by letters according to the position on the molecule.

## IR-Spectroscopy



Figure S32. IR-spectra of (S)-Pregabalin (red) and (R)-Pregabalin (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted.


Figure S33. IR-spectra of (S)-mandelic (red) and (R)-mandelic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted.


Figure S34. IR-spectra of (S)-malic (red) and (R)-malic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted.


Figure S35. IR-spectra of (rac)-Phenibut (red) and (R, S)-Phenibut:mandelic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted,


Figure S36. IR-spectra of (S, S)-Pregabalin:mandelic acid (red) and (R, R)-Pregabalin:mandelic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted.


Figure S37. IR-spectra of (S, R)-Pregabalin:mandelic acid (red) and (R, S)-Pregabalin:mandelic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted.


Figure S38. IR-spectra of (S, S)-Pregabalin:malic acid (red) and (S, R)-Pregabalin:malic acid (green). The range from $400 \mathrm{~cm}^{-1}$ to $4000 \mathrm{~cm}^{-1}$ is depicted

Table S30. Eiss in aspirin, benzoic acid, naphthalene, carbendazim and fenamic acid.

|  | Aspirin | Benzoic acid | Naphthalene | Carbendazim Maleate | Fenamic acid |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | -336.89571169 Ry 9.5E-10 Ry | $\begin{array}{r} \hline-864.48414572 \mathrm{Ry} \\ 6.3 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ | -380.80902318 Ry 4.0E-09 Ry | $\begin{array}{r} \hline-2238.93849076 \mathrm{Ry} \\ 3.6 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-1405.14750341 \mathrm{Ry} \\ 2.5 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ |
| One-electron contribution <br> Hartree contribution <br> Xc contribution <br> Ewald contribution <br> DFT-D3 Dispersion | $\begin{array}{r} \hline-1312.88831935 \mathrm{Ry} \\ \text { 666.38808897 Ry } \\ -75.71501968 \mathrm{Ry} \\ \text { 474.77906097 Ry } \\ -0.01546025 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-431.66148218 \mathrm{Ry} \\ \text { 288.25452830 Ry } \\ -202.93841066 \mathrm{Ry} \\ -280.15301891 \mathrm{Ry} \\ -0.19853529 \mathrm{Ry} \\ \hline \hline \end{array}$ | -151.00679470 Ry <br> 111.67278034 Ry <br> -102.34427854 Ry <br> -121.17790015 Ry <br> -0.12203006 Ry | $\begin{array}{r} \hline-1356.79696542 \mathrm{Ry} \\ \text { 865.02235307 Ry } \\ -522.41418926 \mathrm{Ry} \\ -699.48265014 \mathrm{Ry} \\ -0.46184980 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-729.97165977 \mathrm{Ry} \\ \text { 491.21003984 Ry } \\ -350.97309805 \mathrm{Ry} \\ -435.89581744 \mathrm{Ry} \\ -0.36727809 \mathrm{Ry} \\ \hline \hline \end{array}$ |
| One-center paw contribution | -89.44406235 Ry | -237.78722698 Ry | -117.83080007 Ry | -524.80518921 Ry | -379.14968990 Ry |
| PAW Hartree energy AE PAW Hartree energy PS PAW xc energy AE PAW xc energy PS <br> Total E_H with PAW <br> Total E XC with PAW | 27.95344349 Ry <br> -27.92749991 Ry <br> -30.96576143 Ry <br> 13.26792153 Ry 666.41403255 Ry -93.41285959 Ry | $\begin{array}{r} \hline 112.86367996 \mathrm{Ry} \\ -112.54564220 \mathrm{Ry} \\ -95.09279017 \mathrm{Ry} \\ 41.22003547 \mathrm{Ry} \\ 288.57256606 \mathrm{Ry} \\ -256.81116536 \mathrm{Ry} \end{array}$ | 46.89218991 Ry -46.84845552 Ry <br> -51.68945232 Ry 22.18801792 Ry 111.71651473 Ry -131.84571294 Ry | $\begin{array}{r} \hline 140.64296658 \mathrm{Ry} \\ -140.30422592 \mathrm{Ry} \\ -126.04609233 \mathrm{Ry} \\ 54.48770059 \mathrm{Ry} \\ 865.36109373 \mathrm{Ry} \\ -593.97258100 \mathrm{Ry} \end{array}$ | 84.07972051 Ry <br> -84.00187446 Ry <br> -92.95386241 Ry <br> 39.85979623 Ry 491.28788589 Ry -404.06716423 Ry |

Table S31. Eiss in glycine polymorphs $\alpha-\gamma$.

|  | Glycine a | Glycine $\beta$ | Glycine Y |
| :---: | :---: | :---: | :---: |
| Total energy | -580.75928600 Ry | -290.37728560 Ry | -435.57100873 Ry |
| Estimated scf accuracy | 2.5E-09 Ry | 1.4E-09 Ry | 6.9E-10 Ry |
| One-electron contribution | -214.06562382 Ry | -110.70421627 Ry | -173.43139417 Ry |
| Hartree contribution | 161.38931511 Ry | 82.32377227 Ry | 126.72740778 Ry |
| Xc contribution | -133.72136785 Ry | -66.86096752 Ry | -100.28869051 Ry |
| Ewald contribution | -274.25350224 Ry | -135.08390726 Ry | -198.50249159 Ry |
| DFT-D3 Dispersion | -0.15132600 Ry | -0.07472014 Ry | -0.11108147 Ry |
|  |  |  |  |
| One-center paw contribution | -119.95678120 Ry | 59.97724668 Ry | -89.96475877 Ry |
| PAW Hartree energy AE | 18.36286775 Ry | 18.33019638 Ry | 65.24929324 Ry |
| PAW Hartree energy PS | -18.34599330 Ry | -18.31334187 Ry | -64.97916769 Ry |
| PAW xc energy AE | -20.57600967 Ry | -20.56792933 Ry | -43.28660728 Ry |
| PAW xc energy PS | 8.78096894 Ry | 8.77300925 Ry | 18.91587444 Ry |
| Total E_H with PAW | 161.40618955 Ry | 82.34062678 Ry | 126.99753334 Ry |
| Total E_XC with PAW | -145.51640859 Ry | -78.65588760 Ry | -124.65942334 Ry |

Table S32. Eiss in (S)- and (R)-Pregabalin and (rac)-Phenibut

|  | (S)-1 | (R)-1 | (rac)-2 |
| :---: | :---: | :---: | :---: |
| Total energy | -1071.80165524 Ry | -1071.80170922 Ry | -2395.20650159 Ry |
| Estimated scf accuracy | 5.1E-09 Ry | 5.5E-09 Ry | 0.00000016 Ry |
| One-electron contribution | -577.36224332 Ry | -577.93818768 Ry | -1174.63541450 Ry |
| Hartree contribution | 387.44493557 Ry | 387.65747211 Ry | 811.05092165 Ry |
| Xc contribution | -277.66481361 Ry | -277.65938926 Ry | -604.74515244 Ry |
| Ewald contribution | -342.65463124 Ry | -342.29766682 Ry | -809.28613616 Ry |
| DFT-D3 Dispersion | -0.32003300 Ry | -0.31943849 Ry | -0.68856853 Ry |
| One-center paw contribution | -261.24486964 Ry | -261.24449907 Ry | -616.90215161 Ry |
| PAW Hartree energy AE | 129.27305249 Ry | 129.26729739 Ry | 148.84527047 Ry |
| PAW Hartree energy PS | -128.94468573 Ry | -128.93889306 Ry | -148.49765287 Ry |
| PAW xc energy AE | -115.21439958 Ry | -115.21241556 Ry | -136.12237355 Ry |
| PAW xc energy PS | 49.57481541 Ry | 49.57288647 Ry | 58.66198700 Ry |
| Total E_H with PAW | 387.77330234 Ry | 387.98587643 Ry | 811.39853925 Ry |
| Total E_XC with PAW | -343.30439778 Ry | -343.29891835 Ry | -682.20553899 Ry |

Table S33. Eiss in (S, S)-, (R, R)-, (S, R)- and (R, S)-Pregabalin:mandelic acid.

|  | (S, S)-1:3 | (R, R)-1:3 | (S, R)-1:3 | (R, S)-1:3 |
| :---: | :---: | :---: | :---: | :---: |
| Total energy | -2182.72816251 Ry | -2182.72828907 Ry | -2182.68324007 Ry | -2182.68437385 Ry |
| Estimated scf accuracy | 1.3E-09 Ry | 1.8E-09 Ry | 9.5E-09 Ry | 8.0E-09 Ry |
| One-electron contribution | -1533.51665515 Ry | -1549.49021097 Ry | -1814.50402550 Ry | -1813.82535919 Ry |
| Hartree contribution | 956.64051386 Ry | 964.50310881 Ry | 1097.70041506 Ry | 1097.23254769 Ry |
| Xc contribution | -534.05440736 Ry | -534.04862113 Ry | -534.00508650 Ry | -533.99359867 Ry |
| Ewald contribution | -512.26754807 Ry | -504.16306644 Ry | -372.32105602 Ry | -372.54636041 Ry |
| DFT-D3 Dispersion | -0.55923671 Ry | -0.55880178 Ry | -0.56773385 Ry | -0.56710848 Ry |
|  |  |  |  |  |
| One-center paw contribution | -558.97082908 Ry | -558.97069756 Ry | -558.98575325 Ry | -558.98449480 Ry |
| PAW Hartree energy AE | 262.04303170 Ry | 129.73847311 Ry | 270.35132574 Ry | 242.94333815 Ry |
| PAW Hartree energy PS | -261.15762810 Ry | -129.40728577 Ry | -269.46383582 Ry | -242.18723498 Ry |
| PAW xc energy AE | -202.21748403 Ry | -115.29581401 Ry | -212.40131788 Ry | -196.00997753 Ry |
| PAW xc energy PS | 87.81410231 Ry | 49.65568953 Ry | 92.09786596 Ry | 84.94470266 Ry |
| Total E_H with PAW | 957.52591746 Ry | 964.83429615 Ry | 1098.58790498 Ry | 1097.98865086 Ry |
| Total E_XC with PAW | -648.45778908 Ry | -599.68874561 Ry | -654.30853843 Ry | -645.05887354 Ry |

Table S34. $E_{i s s}$ in (S, S)- and (S, R)-Pregabalin:malic acid and (R, S)-Phenibut:mandelic acid.

|  | (S, S)-1:4 | (S, R)-1:4 | (R, S)-2:3 |
| :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-2213.33656663 R y \\ 1.3 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-2213.34434886 \mathrm{Ry} \\ \text { 1.0E-08 Ry } \end{array}$ | $\begin{array}{r} \hline-1154.26774184 R y \\ 1.6 \mathrm{E}-09 \mathrm{Ry} \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1432.30160955 \mathrm{Ry} \\ \text { 901.97731634 Ry } \\ -514.37445599 \mathrm{Ry} \\ -630.60581621 \mathrm{Ry} \\ -0.51310661 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-1411.01995314 \mathrm{Ry} \\ \text { 891.31218524 Ry } \\ -514.40952926 \mathrm{Ry} \\ -641.20468387 \mathrm{Ry} \\ -0.51584702 \mathrm{Ry} \\ \hline \end{array}$ | -662.88344334 Ry <br> 431.46603161 Ry -279.31815141 Ry <br> -340.16377027 Ry <br> -0.28393580 Ry |
| One-center paw contribution | -537.51889461 Ry | -537.50652081 Ry | -303.08447264 Ry |
| PAW Hartree energy AE | 280.18921538 Ry | 280.30096324 Ry | 159.32291534 Ry |
| PAW Hartree energy PS | -279.18535867 Ry | -279.29446844 Ry | -158.85326536 Ry |
| PAW xc energy AE | -208.23496725 Ry | -208.24379479 Ry | -132.16104066 Ry |
| PAW xc energy PS | 90.51428806 Ry | 90.52373303 Ry | 57.25806117 Ry |
| Total E_H with PAW | 902.98117305 Ry | 892.31868004 Ry | 431.93568158 Ry |
| Total E_XC with PAW | -632.09513518 Ry | -632.12959102 Ry | -354.22113089 Ry |

Table S35. Eisg in aspirin, benzoic acid and naphthalene.

|  | Aspirin | Benzoic acid | Naphthalene |
| :--- | ---: | ---: | ---: |
| Total energy | -336.89571169 Ry | -216.03921007 Ry | -190.34591573 Ry |
| Estimated scf accuracy | $9.5 \mathrm{E}-10 \mathrm{Ry}$ | $4.8 \mathrm{E}-09 \mathrm{Ry}$ | $1.2 \mathrm{E}-09 \mathrm{Ry}$ |
| One-electron contribution | -1312.88831935 Ry | -739.41836998 Ry | -736.59527298 Ry |
| Hartree contribution | 666.38808897 Ry | 376.18668573 Ry | 374.44538692 Ry |
| Xc contribution | -75.71501968 Ry | -50.59087816 Ry | -51.06730146 Ry |
| Ewald contribution | 474.77906097 Ry | 257.24406025 Ry | 281.79361367 Ry |
| DFT-D3 Dispersion | -0.01546025 Ry | -0.00788014 Ry | -0.00959746 Ry |
|  |  |  |  |
| One-center paw | -89.44406235 Ry | -59.45282779 Ry | -58.91274441 Ry |
| contribution |  |  |  |
| PAW Hartree energy AE | 27.95344349 Ry | 56.37489016 Ry | 28.13589881 Ry |
| PAW Hartree energy PS | -27.92749991 Ry | -56.11525462 Ry | -28.10975625 Ry |
| PAW xc energy AE | -30.96576143 Ry | -33.04993323 Ry | -31.01244409 Ry |
| PAW xc energy PS | 13.26792153 Ry | 14.57723675 Ry | 13.31273914 Ry |
| Total E_H with PAW | 666.41403255 Ry | 376.44632128 Ry | 374.47152947 Ry |
| Total E_XC with PAW | -93.41285959 Ry | -69.06357463 Ry | -68.76700641 Ry |

Table S36. Eisg in Carbendazim maleate, Carbendazim charged, Carbendazim neutral, maleic acid charged and maleic acid neutral.

|  | Carbendazim (neutral) | Carbendazim (charge) | Maleic acid (neutral) | Maleic acid (charge) |
| :---: | :---: | :---: | :---: | :---: |
| Total energy | -317.81712031 Ry | -318.65170796 Ry | -241.69462491 Ry | -240.74748977 Ry |
| Estimated scf accuracy | 5.4E-09 Ry | 1.4E-09 Ry | 8.6E-09 Ry | 2.6E-09 Ry |
| One-electron contribution | -1331.80840675 Ry | -1349.82911878 Ry | -750.38874570 Ry | -741.67106561 Ry |
| Hartree contribution | 675.48840558 Ry | 677.97198443 Ry | 381.58547061 Ry | 383.02285946 Ry |
| Xc contribution | -80.09207190 Ry | -80.33880782 Ry | -50.11394916 Ry | -49.86629580 Ry |
| Ewald contribution | 489.82645736 Ry | 504.77755561 Ry | 237.21931250 Ry | 227.76142495 Ry |
| DFT-D3 Dispersion | -0.01432587 Ry | -0.01529829 Ry | -0.00631915 Ry | -0.00561997 Ry |
|  |  |  |  |  |
| One-center paw contribution | -71.21717873 Ry | -71.21802311 Ry | -59.99039401 Ry | -59.98879279 Ry |
| PAW Hartree energy AE | 0.00000000 Ry | 0.00000000 Ry | 56.55146170 Ry | 55.97546094 Ry |
| PAW Hartree energy PS | 0.00000000 Ry | 0.00000000 Ry | -56.29063300 Ry | -55.71544737 Ry |
| PAW xc energy AE | 0.00000000 Ry | 0.00000000 Ry | -33.07681606 Ry | -32.98567477 Ry |
| PAW xc energy PS | 0.00000000 Ry | 0.00000000 Ry | 14.60409227 Ry | 14.51384719 Ry |
| Total E_H with PAW | 675.48840558 Ry | 677.97198443 Ry | 381.84629932 Ry | 383.28287303 Ry |
| Total E_XC with PAW | -80.09207190 Ry | -80.33880782 Ry | -68.58667296 Ry | -68.33812339 Ry |

Table S37. Eisg in fenamic acid, molecule A and molecule B as well as Glycine $\alpha$, neutral and zwitterionic.

|  | Fenamic acid A | Fenamic acid B | Glycine a (neutral) | Glycine a (zwitt.) |
| :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-351.18463954 \mathrm{Ry} \\ 6.6 \mathrm{E}-09 \mathrm{Rv} \end{array}$ | $\begin{array}{r} \hline-351.18406565 \mathrm{Ry} \\ 8.5 \mathrm{E}-10 \mathrm{Rv} \end{array}$ | $\begin{array}{r} \hline-145.06565496 \mathrm{Ry} \\ \text { 2.0E-09 Ry } \end{array}$ | $\begin{array}{r} \hline-145.06563313 \mathrm{Ry} \\ \text { 5.3E-09 Ry } \end{array}$ |
| One-electron contribution | -1464.34559523 Ry | -1464.03566170 Ry | -444.69888481 Ry | -444.67960905 Ry |
| Hartree contribution | 743.17809909 Ry | 743.04041847 Ry | 226.93765648 Ry | 226.92649777 Ry |
| Xc contribution | -87.54431237 Ry | -87.54001696 Ry | -33.25141345 Ry | -33.25068021 Ry |
| Ewald contribution | 552.33753090 Ry | 552.16178021 Ry | 135.94233456 Ry | 135.93350594 Ry |
| DFT-D3 Dispersion | -0.01986038 Ry | -0.01987773 Ry | -0.00368100 Ry | -0.00368579 Ry |
| One-center paw contribution | -94.79050154 Ry | -94.79070794 Ry | -29.99166675 Ry | -29.99166178 Ry |
| PAW Hartree energy AE | 37.19247007 Ry | 27.85123707 Ry | 18.22767627 Ry | 18.22421611 Ry |
| PAW Hartree energy PS | -37.15820627 Ry | -27.82543755 Ry | -18.21084902 Ry | -18.20739007 Ry |
| PAW xc energy AE | -41.26733921 Ry | -30.94234266 Ry | -20.54299767 Ry | -20.54215868 Ry |
| PAW xc energy PS | 17.67303721 Ry | 13.24517069 Ry | 8.74823976 Ry | 8.74740701 Ry |
| Total E_H with PAW | 743.21236288 Ry | 743.06621799 Ry | 226.95448372 Ry | 226.94332381 Ry |
| Total E_XC with PAW | -111.13861438 Ry | -105.23718894 Ry | -45.04617137 Ry | -45.04543188 Ry |

Table S38. Eisg in Glycine $\beta$, neutral and zwitterionic as well as Glycine $\gamma$, neutral and zwitterionic.

|  | Glycine $\beta$ (neutral) | Glycine $\beta$ (zwitt.) | Glycine Y (neutral) | Glycine Y (zwitt.) |
| :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-145.06569801 \mathrm{Ry} \\ 6.9 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-145.06570159 \mathrm{Ry} \\ 1.8 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-145.05056883 \mathrm{Ry} \\ 4.9 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-145.06557827 \mathrm{Ry} \\ 6.8 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-444.75995154 \mathrm{Ry} \\ \text { 226.97201711 Ry } \\ -33.25476841 \mathrm{Ry} \\ \text { 135.97232156 Ry } \\ -0.00367606 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline \hline-444.62073980 \mathrm{Ry} \\ \text { 226.90162080 Ry } \\ -33.25109138 \mathrm{Ry} \\ \text { 135.89987598 Ry } \\ -0.00367672 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-441.73616736 \mathrm{Ry} \\ 225.42566450 \mathrm{Ry} \\ -33.22831266 \mathrm{Ry} \\ \text { 134.483477770 Ry } \\ -0.00395088 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-444.81722053 \mathrm{Ry} \\ \text { 226.99643361 Ry } \\ -33.25077079 \mathrm{Ry} \\ \text { 136.00136033 Ry } \\ -0.00368777 \mathrm{Ry} \\ \hline \hline \end{array}$ |
| One-center paw contribution | -29.99164066 Ry | -29.99169048 Ry | -29.99128013 Ry | -29.99169313 Ry |
| PAW Hartree energy AE PAW Hartree energy PS PAW xc energy AE PAW xc energy PS Total E_H with PAW Total E_XC with PAW | $\begin{array}{r} \hline 18.23912880 \mathrm{Ry} \\ -18.22229626 \mathrm{Ry} \\ -20.54581868 \mathrm{Ry} \\ \text { 8.75102687 Ry } \\ \text { 226.98884964 Ry } \\ -45.04956022 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline 18.22323513 \mathrm{Ry} \\ -18.20640903 \mathrm{Ry} \\ -20.54192591 \mathrm{Ry} \\ 8.74717092 \mathrm{Ry} \\ 226.91844690 \mathrm{Ry} \\ -45.04584637 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline 56.42606530 \mathrm{Ry} \\ -56.16667109 \mathrm{Ry} \\ -33.05806132 \mathrm{Ry} \\ \text { 14.58523343 Ry } \\ \text { 225.68505871 Ry } \\ -51.70114056 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline 56.36291519 R y \\ -56.10418032 R y \\ -33.04778665 R y \\ \text { 14.57528354 Ry } \\ \text { 227.25516848 Ry } \\ -51.72327390 \text { Ry } \end{array}$ |

Table S39. Eisg in (S)-Pregabalin, neutral and zwitterionic, (R)-Pregabalin, neutral and zwitterionic as well as (rac)-Phenibut, neutral and zwitterionic.

|  | (S)-1 (neutral) | (S)-1 (zwitt.) | (R)-1 (neutral) | (R)-1 (zwitt.) | (rac)-2 (neutral) | (rac)-2 (zwitt.) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-267.79801705 \mathrm{Ry} \\ 1.0 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ | -267.80747791 Ry <br> 8.1E-09 Ry | $\begin{array}{r} \hline-267.79160852 \mathrm{Ry} \\ 1.6 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-267.80763781 \mathrm{Ry} \\ 1.1 \mathrm{E}-09 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-299.23843736 \mathrm{Ry} \\ 3.8 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-299.12073572 \mathrm{Ry} \\ 2.5 \mathrm{E}-09 \mathrm{Ry} \\ \hline \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1132.97801838 \mathrm{Ry} \\ \text { 576.18367680 Ry } \\ -69.16008416 \mathrm{Ry} \\ \text { 423.49226678 Ry } \\ -0.02235537 \mathrm{Ry} \end{array}$ | -1137.10596586 Ry 577.94914245 Ry -69.18545259 Ry 425.86835966 Ry -0.02261513 Ry | -1139.37899930 Ry 579.17479627 Ry -69.15366478 Ry 426.90277629 Ry -0.02260823 Ry | $\begin{array}{r} \hline-1137.15690002 \mathrm{Ry} \\ \text { 577.96962803 Ry } \\ -69.18727646 \mathrm{Ry} \\ \text { 425.90049677 Ry } \\ -0.02262865 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-1233.30321001 \mathrm{Ry} \\ \text { 626.59819637 Ry } \\ \text {-75.30456653 Ry } \\ \text { 459.90584669 Ry } \\ -0.02054518 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-1243.07956735 \mathrm{Ry} \\ 631.59450463 \mathrm{Ry} \\ -75.28054105 \mathrm{Ry} \\ 464.78434800 \mathrm{Ry} \\ -0.02103308 \mathrm{Ry} \end{array}$ |
|  |  |  |  |  |  |  |
| One-center paw contribution | -65.31350272 Ry | -65.31094644 Ry | -65.31390877 Ry | -65.31095748 Ry | -77.11415871 Ry | -77.11844687 Ry |
| PAW Hartree energy AE | 65.42093235 Ry | 65.47325474 Ry | 65.40966588 Ry | 65.46862311 Ry | 37.40491936 Ry | 36.88010451 Ry |
| PAW Hartree energy PS | -65.15326146 Ry | -65.20405897 Ry | -65.14206060 Ry | -65.19944719 Ry | -37.26504078 Ry | -36.74389647 Ry |
| PAW xc energy AE | -43.30887949 Ry | -43.31644828 Ry | -43.30718024 Ry | -43.31556576 Ry | -26.80943871 Ry | -26.72862750 Ry |
| PAW xc energy PS | 18.93592371 Ry | 18.94489265 Ry | 18.93411522 Ry | 18.94402132 Ry | 11.67352971 Ry | 11.59424725 Ry |
| Total E_H with PAW | 576.45134769 Ry | 578.21833823 Ry | 579.44240155 Ry | 578.23880394 Ry | 626.73807495 Ry | 631.73071267 Ry |
| Total E_XC with PAW | -93.53303994 Ry | -93.55700822 Ry | -93.52672979 Ry | -93.55882090 Ry | -90.44047553 Ry | -90.41492130 Ry |


|  | (S)-1_A (neutral) | (S)-1_A (zwitt.) | (S)-1_B (neutral) | (S)-1_B (zwitt.) | (S)-3_A (neutral) | (S)-3_B (neutral) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} -267.79207687 \mathrm{Ry} \\ 8.4 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} -267.80896808 \mathrm{Ry} \\ 9.6 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | -267.79386670 Ry <br> 1.4E-09 Ry | $\begin{array}{r} -267.80897994 \mathrm{Ry} \\ \text { 7.2E-09 Ry } \end{array}$ | $\begin{array}{r} -277.62920472 \mathrm{Ry} \\ \text { 8.4E-09 Ry } \end{array}$ | $-277.62892955 \mathrm{Ry}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1146.57467144 R y \\ \text { 582.84266957 Ry } \\ -69.15187735 R \mathrm{Ry} \\ \text { 430.42887044 Ry } \\ -0.02309849 \mathrm{Ry} \\ \hline \end{array}$ | -1149.12964591 Ry 584.04851409 Ry -69.18326169 Ry 431.78954630 Ry -0.02287750 Ry | $\begin{array}{r} \hline-1150.27288277 \mathrm{Ry} \\ \text { 584.64607854 Ry } \\ -69.16093982 \mathrm{Ry} \\ \text { 432.33012274 Ry } \\ -0.02321006 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-1149.76744088 \mathrm{Ry} \\ \text { 584.36864185 Ry } \\ -69.18745151 \mathrm{Ry} \\ \text { 432.11151848 Ry } \\ -0.02290010 \mathrm{Ry} \\ \hline \end{array}$ | -1029.52101511 Ry 523.17133645 Ry -63.90041076 Ry 367.07359829 Ry -0.01230029 Ry | $\begin{array}{r} \hline-1029.30778810 \mathrm{Ry} \\ \text { 523.06281742 Ry } \\ -63.89877731 \mathrm{Ry} \\ \text { 366.96757988 Ry } \\ -0.01227256 \mathrm{Ry} \\ \hline \end{array}$ |
| One-center paw contribution | -65.31396958 Ry | -65.31124336 Ry | -65.31303533 Ry | -65.31134778 Ry | -74.44041329 Ry | -74.44048887 Ry |
| PAW Hartree energy AE PAW Hartree energy PS PAW xc energy AE PAW xc energy PS <br> Total E_H with PAW <br> Total E XC with PAW | 65.39274545 Ry -65.12532377 Ry -43.30298523 Ry 18.93003382 Ry 583.11009124 Ry -93.52482877 Ry | $\begin{array}{\|c\|} \hline \hline 65.45833287 \text { Ry } \\ -65.18933924 R y \\ -43.31464608 ~ R y \\ \text { 18.94297919 Ry } \\ \text { 584.31750773 Ry } \\ -93.55492858 \text { Ry } \end{array}$ | $65.43766939 R y$ $-65.16942296 R y$ $-43.30937372 R y$ $18.93652996 R y$ 584.91432497 Ry -93.53378358 Ry | 65.46017095 Ry -65.19119186 Ry -43.31535966 Ry <br> 18.94368674 Ry <br> 584.63762094 Ry $-93.55912444 \mathrm{Ry}$ | 84.58268227 Ry <br> -84.19016010 Ry <br> -49.57454296 Ry <br> 21.86736942 Ry <br> 523.56385862 Ry <br> -91.60758429 Ry | 84.58009975 Ry -84.18763805 Ry -49.57420263 Ry 21.86699399 Ry 523.45527912 Ry -91.60598595 Ry |



|  | (R)-1_A (neutral) | (R)-1_A (zwitt.) | (R)-1_B (neutral) | (R)-1_B (zwitt.) | (R)-3_A (neutral) | (R)-3_B (neutral) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} -267.79375084 \mathrm{Ry} \\ 4.5 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-267.80898302 \mathrm{Ry} \\ \text { 1.2E-09 Ry } \end{array}$ | $\begin{array}{r} =-267.79208315 \mathrm{Ry} \\ 8.4 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-267.80900234 \mathrm{Ry} \\ 1.7 \mathrm{E}-09 \mathrm{Rv} \end{array}$ | $\begin{array}{r} \hline-277.62918420 \mathrm{Ry} \\ 4.2 \mathrm{E}-09 \mathrm{Rv} \end{array}$ | $\begin{array}{r} \hline-277.62894959 \mathrm{Ry} \\ 2.3 \mathrm{E}-09 \mathrm{Ry} \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1150.18357840 \mathrm{Ry} \\ 584.60436837 \mathrm{Ry} \\ -69.16129377 \mathrm{Ry} \\ 432.28297207 \mathrm{Ry} \\ -0.02319512 \mathrm{Ry} \\ \hline \end{array}$ | -1149.37655749 Ry 584.17203235 Ry -69.18656757 Ry 431.91632775 Ry -0.02287986 Ry | $\begin{array}{r} \hline-1146.61835649 \mathrm{Ry} \\ 582.86745891 \mathrm{Ry} \\ -69.15378472 \mathrm{Ry} \\ 430.44963452 \mathrm{Ry} \\ -0.02309843 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-1149.27662069 \mathrm{Ry} \\ 584.12180134 \mathrm{Ry} \\ -69.18739558 \mathrm{Ry} \\ 431.86736104 \mathrm{Ry} \\ -0.02287231 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-1029.64141539 \mathrm{Ry} \\ 523.23017098 \mathrm{Ry} \\ -63.90315362 \mathrm{Ry} \\ \text { 367.13794787 Ry } \\ -0.01230760 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-1029.31076807 \mathrm{Ry} \\ \text { 523.06429268 Ry } \\ -63.89762768 \mathrm{Ry} \\ \text { 366.96786671 Ry } \\ -0.01227643 \mathrm{Ry} \\ \hline \end{array}$ |
| One-center paw contribution | -65.31302397 Ry | -65.31133819 Ry | -65.31393693 Ry | -65.31127615 Ry | -74.44042645 Ry | -74.44043679 Ry |
| PAW Hartree energy AE | 65.43436647 Ry | 65.45507206 Ry | 65.39334619 Ry | 65.46939876 Ry | 84.59206754 Ry | 84.57999025 Ry |
| PAW Hartree energy PS | -65.16609918 Ry | -65.18611142 Ry | -65.12588796 Ry | -65.20034783 Ry | -84.19946124 Ry | -84.18751782 Ry |
| PAW xc energy AE | -43.30861886 Ry | -43.31433639 Ry | -43.30285552 Ry | -43.31665117 Ry | -49.57601963 Ry | -49.57417644 Ry |
| PAW xc energy PS | 18.93578963 Ry | 18.94268044 Ry | 18.92991494 Ry | 18.94496975 Ry | 21.86883673 Ry | 21.86697238 Ry |
| Total E_H with PAW | 584.87263565 Ry | 584.44099299 Ry | 583.13491713 Ry | 584.39085227 Ry | 523.62277728 Ry | 523.45676512 Ry |
| Total E_XC with PAW | -93.53412301 Ry | -93.55822352 Ry | -93.52672531 Ry | -93.55907699 Ry | -91.61033652 Ry | -91.60483174 Ry |

Table S42. Eisg in (S, R)-Pregabalin:mandelic acid, Pregabalin molecule A neutral, zwitterionic and charged as well as Pregabalin molecule B neutral, zwitterionic and charged.

|  | (S)-1_A (neutral) | (S)-1_A (zwitt.) | (S)-1_A (charge) | (S)-1_B (neutral) | (S)-1_B (zwitt.) | (S)-1_B (charge) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy | -267.79975583 Ry | -267.80894456 Ry | -268.64998668 Ry | -267.79703425 Ry | -267.80927522 Ry | -268.59869837 Ry |
| Estimated scf accuracy | 2.6E-09 Ry | 3.7E-09 Ry | 4.5E-09 Ry | 6.3E-09 Ry | 1.2E-09 Ry | 1.4E-09 Ry |
| One-electron contribution | -1150.32791801 Ry | -1144.52122571 Ry | -1157.74153844 Ry | -1061.81071755 Ry | -1170.79281370 Ry | -1175.44653807 Ry |
| Hartree contribution | 584.88706105 Ry | 581.71822027 Ry | 582.42093762 Ry | 540.85882190 Ry | 594.89073189 Ry | 591.44772100 Ry |
| Xc contribution | -69.16647533 Ry | -69.18155183 Ry | -69.41358892 Ry | -69.16717044 Ry | -69.19667954 Ry | -69.36887551 Ry |
| Ewald contribution | 432.14302255 Ry | 429.50919174 Ry | 441.41688917 Ry | 387.65948308 Ry | 442.62496333 Ry | 450.10991996 Ry |
| DFT-D3 Dispersion | -0.02306737 Ry | -0.02269933 Ry | -0.02362598 Ry | -0.02360571 Ry | -0.02410924 Ry | -0.02498421 Ry |
|  |  |  |  |  |  |  |
| One-center paw contribution | -65.31237873 Ry | -65.31087970 Ry | -65.30906012 Ry | -65.31384553 Ry | -65.31136797 Ry | -65.31594154 Ry |
| PAW Hartree energy AE | 65.49651598 Ry | 65.35817895 Ry | 65.73962312 Ry | 65.38430547 Ry | 65.41420775 Ry | 65.45657948 Ry |
| PAW Hartree energy PS | -65.22864191 Ry | -65.08926967 Ry | -65.46853461 Ry | -65.11687067 Ry | -65.14561066 Ry | -65.18980703 Ry |
| PAW xc energy AE | -43.31230587 Ry | -43.27938939 Ry | -43.33943230 Ry | -43.30085731 Ry | -43.30791188 Ry | -43.31980347 Ry |
| PAW xc energy PS | 18.94509686 Ry | 18.91257740 Ry | 18.97314775 Ry | 18.92766341 Ry | 18.93618130 Ry | 18.94662604 Ry |
| Total E_H with PAW | 585.15493511 Ry | 581.98712956 Ry | 582.69202613 Ry | 541.12625670 Ry | 595.15932898 Ry | 591.71449345 Ry |
| Total E_XC with PAW | -93.53368433 Ry | -93.54836382 Ry | -93.77987347 Ry | -93.54036434 Ry | -93.56841013 Ry | -93.74205294 Ry |

Table S43. Eisg in (S, R)-Pregabalin:mandelic acid, mandelic acid molecule $A$. neutral and charged, as well as mandelic acid molecule B, neutral and charged.

|  | (R)-3_A (neutral) | (R)-3_A (charge) | (R)-3_B (neutral) | (R)-3_B (charge) |
| :---: | :---: | :---: | :---: | :---: |
| Total energy | -277.62952062 Ry | -276.65200500 Ry | -277.63008521 Ry | -276.63285774 Ry |
| Estimated scf accuracy | 9.8E-09 Ry | 1.8E-09 Ry | 5.7E-09 Ry | 0.00000095 Ry |
| One-electron contribution | -984.02556825 Ry | -836.65595354 Ry | -1033.27892563 Ry | -1015.73101764 Ry |
| Hartree contribution | 500.53925577 Ry | 431.98760346 Ry | 525.05776382 Ry | 522.32497785 Ry |
| Xc contribution | -63.91233143 Ry | -63.65826059 Ry | -63.91244102 Ry | -63.62993845 Ry |
| Ewald contribution | 344.22059760 Ry | 266.12644157 Ry | 368.95578786 Ry | 354.85796838 Ry |
| DFT-D3 Dispersion | -0.01254664 Ry | -0.01182641 Ry | -0.01256058 Ry | -0.01144589 Ry |
|  |  |  |  |  |
| One-center paw contribution | -74.43892767 Ry | -74.44000949 Ry | -74.43970966 Ry | -74.44340199 Ry |
| PAW Hartree energy AE | 84.68111468 Ry | 83.61206042 Ry | 84.64220632 Ry | 83.33311012 Ry |
| PAW Hartree energy PS | -84.28761471 Ry | -83.22235287 Ry | -84.24910181 Ry | -82.94553301 Ry |
| PAW xc energy AE | -49.58968685 Ry | -49.41921197 Ry | -49.58361938 Ry | -49.37621997 Ry |
| PAW xc energy PS | 21.88259150 Ry | 21.71404709 Ry | 21.87654439 Ry | 21.67079171 Ry |
| Total E_H with PAW | 500.93275574 Ry | 432.37731101 Ry | 525.45086832 Ry | 522.71255496 Ry |
| Total E_XC with PAW | -91.61942678 Ry | -91.36342547 Ry | -91.61951601 Ry | -91.33536671 Ry |

Table S44. Eisg in (R, S)-Pregabalin:mandelic acid, Pregabalin molecule A neutral, zwitterionic and charged as well as Pregabalin molecule B neutral, zwitterionic and charged.

|  | (R)-1_A (neutral) | (R)-1_A (zwitt.) | (R)-1_A (charge) | (R)-1_B (neutral) | (R)-1_B (zwitt.) | (R)-1_B (charge) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} -267.79396641 \mathrm{Ry} \\ 4 \mathrm{OF}-0.9 \mathrm{Rv} \end{array}$ | $-267.80931677 \mathrm{Ry}$ 7.7E-10 Ry | $\begin{array}{r} -268.59858887 \mathrm{Ry} \\ 8.0 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} -267.79977164 \mathrm{Ry} \\ \text { 6.1E-09 Ry } \end{array}$ | $\begin{array}{r} -267.80896960 \mathrm{Ry} \\ 2.4 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} -268.65003203 \mathrm{Ry} \\ 1.4 \mathrm{E}-09 \mathrm{Ry} \end{array}$ |
| One-electron contribution | -1153.10618141 Ry | -1170.83568908 Ry | -1175.35713834 Ry | -1151.10825527 Ry | -1144.69063057 Ry | -1158.22261487 Ry |
| Hartree contribution | 586.31973900 Ry | 594.91086176 Ry | 591.41405146 Ry | 585.27837240 Ry | 581.80257085 Ry | 582.65951124 Ry |
| Xc contribution | -69.15451484 Ry | -69.19512398 Ry | -69.36695264 Ry | -69.16943185 Ry | -69.18286439 Ry | -69.41795409 Ry |
| Ewald contribution | 433.48355758 Ry | 442.64608092 Ry | 450.05223978 Ry | 432.53498839 Ry | 429.59556094 Ry | 441.66382314 Ry |
| DFT-D3 Dispersion | -0.02323917 Ry | -0.02411122 Ry | -0.02498764 Ry | -0.02309794 Ry | -0.02270447 Ry | -0.02364287 Ry |
|  |  |  |  |  |  |  |
| One-center paw contribution | -65.31332757 Ry | -65.31133516 Ry | -65.31580150 Ry | -65.31234738 Ry | -65.31090194 Ry | -65.30915459 Ry |
| PAW Hartree energy AE | 65.38648975 Ry | 65.42178049 Ry | 65.46347958 Ry | 65.42252282 Ry | 65.47250700 Ry | 65.75107390 Ry |
| PAW Hartree energy PS | -65.11894629 Ry | -65.15316949 Ry | -65.19662209 Ry | -65.15426466 Ry | -65.20327838 Ry | -65.47967284 Ry |
| PAW xc energy AE | -43.30046858 Ry | -43.30947470 Ry | -43.32057164 Ry | -43.30306322 Ry | -43.31621869 Ry | -43.35300760 Ry |
| PAW xc energy PS | 18.92728254 Ry | 18.93772130 Ry | 18.94743884 Ry | 18.93057606 Ry | 18.94465890 Ry | 18.98099465 Ry |
| Total E_H with PAW | 586.58728247 Ry | 595.17947276 Ry | 591.68090895 Ry | 585.54663056 Ry | 582.07179947 Ry | 582.93091231 Ry |
| Total E_XC with PAW | -93.52770087 Ry | -93.56687738 Ry | -93.74008544 Ry | -93.54191901 Ry | -93.55442418 Ry | -93.78996704 Ry |

Table S45. Eisg in (R, S)-Pregabalin:mandelic acid, mandelic acid molecule $A$. neutral and charged, as well as mandelic acid molecule B, neutral and charged.

|  | (S)-3_A (neutral) | (S)-3_A (charge) | (S)-3_B (neutral) | (S)-3_B (charge) |
| :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-277.62917802 \mathrm{Ry} \\ 1.8 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-276.64477180 \mathrm{Ry} \\ 2.3 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-277.62574657 \mathrm{Ry} \\ 1.2 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-276.63279628 \mathrm{Ry} \\ 8.7 \mathrm{E}-09 \mathrm{Ry} \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1032.66358245 \mathrm{Ry} \\ \text { 524.74114097 Ry } \\ -63.90590518 \mathrm{Ry} \\ \text { 368.65199764 Ry } \\ -0.01242153 \mathrm{Ry} \\ \hline \hline \end{array}$ | $\begin{array}{r} \hline-1015.49434460 \mathrm{Ry} \\ 522.28819002 \mathrm{Ry} \\ -63.64845335 \mathrm{Ry} \\ 354.66134689 \mathrm{Ry} \\ -0.01169329 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-1029.31666511 \mathrm{Ry} \\ \text { 523.03937919 Ry } \\ -63.90083528 \mathrm{Ry} \\ \text { 367.00633024 Ry } \\ -0.01226152 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-1015.84614954 \mathrm{Ry} \\ \text { 522.38115878 Ry } \\ -63.62862274 \mathrm{Ry} \\ \text { 354.91564817 Ry } \\ -0.01145176 \mathrm{Ry} \\ \hline \end{array}$ |
| One-center paw contribution | -74.44040748 Ry | -74.43981747 Ry | -74.44169408 Ry | -74.44337920 Ry |
| PAW Hartree energy AE | 84.60083134 Ry | 83.58818023 Ry | 84.51268735 Ry | 83.32830041 Ry |
| PAW Hartree energy PS | -84.20818012 Ry | -83.19857792 Ry | -84.12089990 Ry | -82.94079809 Ry |
| PAW xc energy AE | -49.57735772 Ry | -49.41527993 Ry | -49.56417359 Ry | -49.37532787 Ry |
| PAW xc energy PS | 21.87018418 Ry | 21.71022287 Ry | 21.85672289 Ry | 21.66997428 Ry |
| Total E_H with PAW | 525.13379219 Ry | 522.67779234 Ry | 523.43116664 Ry | 522.76866110 Ry |
| Total E_XC with PAW | -91.61307873 Ry | -91.35351042 Ry | -91.60828599 Ry | -91.33397632 Ry |

Table S46. Eisg in (S, S)-Pregabalin:malic acid, Pregabalin neutral, zwitterionic and charged, as well as malic acid neutral and charged.

|  | (S)-1 (neutral) | (S)-1 (zwitt.) | (S)-1 (charge) | (S)-4 (neutral) | (S)-4 (charge) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy | -267.79195190 Ry | -267.80910065 Ry | -268.61129733 Ry | -285.25372379 Ry | -284.27132001 Ry |
| Estimated scf accuracy | 6.9E-10 Ry | 8.4E-09 Ry | 5.7E-09 Ry | 1.7E-09 Ry | 4.4E-09 Ry |
| One-electron contribution | -1136.31452856 Ry | -1145.23419082 Ry | -1149.91381033 Ry | -950.21680717 Ry | -935.83302748 Ry |
| Hartree contribution | 577.57489783 Ry | 582.07656352 Ry | 578.72969549 Ry | 483.26247560 Ry | 482.11646018 Ry |
| Xc contribution | -69.15816236 Ry | -69.18910101 Ry | -69.37979494 Ry | -58.94285641 Ry | -58.66689854 Ry |
| Ewald contribution | 425.44242786 Ry | 429.87147012 Ry | 437.28987967 Ry | 309.73863934 Ry | 297.20839694 Ry |
| DFT-D3 Dispersion | -0.02283875 Ry | -0.02270041 Ry | -0.02381648 Ry | -0.00965105 Ry | -0.00883028 Ry |
|  |  |  |  |  |  |
| One-center paw contribution | -65.31374792 Ry | -65.31114206 Ry | -65.31345074 Ry | -69.08552410 Ry | -69.08742084 Ry |
| PAW Hartree energy AE | 65.41975252 Ry | 65.47797193 Ry | 65.59295620 Ry | 56.20194601 Ry | 56.03295602 Ry |
| PAW Hartree energy PS | -65.15208371 Ry | -65.20880712 Ry | -65.32452334 Ry | -55.94103201 Ry | -55.77213855 Ry |
| PAW xc energy AE | -43.30944981 Ry | -43.31787678 Ry | -43.33918912 Ry | -33.02222717 Ry | -32.99354773 Ry |
| PAW xc energy PS | 18.93635720 Ry | 18.94625060 Ry | 18.96680997 Ry | 14.55058100 Ry | 14.52276720 Ry |
| Total E_H with PAW | 577.84256664 Ry | 582.34572833 Ry | 578.99812834 Ry | 483.52338960 Ry | 482.37727766 Ry |
| Total E XC with PAW | -93.53125497 Ry | -93.56072719 Ry | -93.75217409 Ry | -77.41450259 Ry | -77.13767908 Ry |

Table S47. Eisg in (S, R)-Pregabalin:malic acid, Pregabalin neutral, zwitterionic and charged, as well as malic acid neutral and charged.

|  | (S)-1 (neutral) | (S)-1 (zwitt.) | (S)-1 (charge) | (R)-4 (neutral) | (R)-4 (charge) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Total energy Estimated scf accuracy | $\begin{array}{r} \hline-267.79369461 \mathrm{Ry} \\ 1.6 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-267.80879947 \mathrm{Ry} \\ \text { 3.6E-09 Ry } \end{array}$ | $\begin{array}{r} \hline-268.61135562 \mathrm{Ry} \\ 1.0 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-285.25091193 \mathrm{Ry} \\ 1.9 \mathrm{E}-09 \mathrm{Ry} \end{array}$ | $\begin{array}{r} \hline-284.22893057 \mathrm{Ry} \\ 9.8 \mathrm{E}-10 \mathrm{Ry} \\ \hline \end{array}$ |
| One-electron contribution Hartree contribution Xc contribution Ewald contribution DFT-D3 Dispersion | $\begin{array}{r} \hline-1131.28671043 R y \\ \text { 575.19633096 Ry } \\ -69.15608675 \mathrm{Ry} \\ \text { 422.78905988 Ry } \\ -0.02257561 \mathrm{Ry} \\ \hline \end{array}$ | -1143.86090533 Ry 581.38466378 Ry -69.18925831 Ry 429.19015031 Ry -0.02266884 Ry | $\begin{array}{r} \hline-1149.82367760 \mathrm{Ry} \\ 578.68652083 \mathrm{Ry} \\ -69.37972215 \mathrm{Ry} \\ 437.24272236 \mathrm{Ry} \\ -0.02379310 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-952.95477737 \mathrm{Ry} \\ \text { 484.68449087 Ry } \\ -58.94229589 \mathrm{Ry} \\ 311.05700150 \mathrm{Ry} \\ -0.00974501 \mathrm{Ry} \\ \hline \end{array}$ | $\begin{array}{r} \hline-939.85881162 \mathrm{Ry} \\ \text { 484.06054389 Ry } \\ -58.63695052 \mathrm{Ry} \\ 299.30453850 \mathrm{Ry} \\ -0.00955013 \mathrm{Ry} \\ \hline \end{array}$ |
| One-center paw contribution | -65.31371266 Ry | -65.31078108 Ry | -65.31340595 Ry | -69.08558604 Ry | -69.08870068 Ry |
| PAW Hartree energy AE | 65.43919007 Ry | 65.48267393 Ry | 65.59452810 Ry | 56.82543535 Ry | 56.02910396 Ry |
| PAW Hartree energy PS | -65.17159110 Ry | -65.21334090 Ry | -65.32605466 Ry | -56.56252940 Ry | -55.77055731 Ry |
| PAW xc energy AE | -43.31264115 Ry | -43.31764738 Ry | -43.33959060 Ry | -33.11632309 Ry | -32.99343133 Ry |
| PAW xc energy PS | 18.93954975 Ry | 18.94609891 Ry | 18.96723529 Ry | 14.64453392 Ry | 14.52167757 Ry |
| Total E_H with PAW | 575.46392994 Ry | 581.65399680 Ry | 578.95499427 Ry | 484.94739682 Ry | 484.31909053 Ry |
| Total E_XC with PAW | -93.52917815 Ry | -93.56080677 Ry | -93.75207746 Ry | -77.41408506 Ry | -77.10870427 Ry |

Table S48. Eisg in (R, S)-Phenibut:mandelic acid, Phenibut neutral and zwitterionic as well as mandelic acid neutral.

|  | (R)-2 (neutral) | (R)-2 (zwitt.) | (S)-3 (neutral) |
| :---: | :---: | :---: | :---: |
| Total energy | -299.24118523 Ry | -299.24903734 Ry | -277.63003077 Ry |
| Estimated scf accuracy | 2.0E-09 Ry | 5.2E-09 Ry | 4.4E-09 Ry |
| One-electron contribution | -1241.39331178 Ry | -1237.67805463 Ry | -1030.68394340 Ry |
| Hartree contribution | 630.84461220 Ry | 628.70733242 Ry | 523.76518044 Ry |
| Xc contribution | -75.31692750 Ry | -75.33402510 Ry | -63.90941862 Ry |
| Ewald contribution | 463.75757796 Ry | 462.18718074 Ry | 367.65037561 Ry |
| DFT-D3 Dispersion | -0.02048529 Ry | -0.02027812 Ry | -0.01241815 Ry |
| One-center paw contribution | -77.11265082 Ry | -77.11119265 Ry | -74.43980665 Ry |
| PAW Hartree energy AE | 65.42696385 Ry | 65.47298401 Ry | 84.63610579 Ry |
| PAW Hartree energy PS | -65.15878090 Ry | -65.20386654 Ry | -84.24306855 Ry |
| PAW xc energy AE | -43.30416156 Ry | -43.31677390 Ry | -49.58274610 Ry |
| PAW xc energy PS | 18.93168269 Ry | 18.94518322 Ry | 21.87563244 Ry |
| Total E_H with PAW | 631.11279515 Ry | 628.97644989 Ry | 524.15821768 Ry |
| Total E_XC with PAW | -99.68940637 Ry | -99.70561579 Ry | -91.61653228 Ry |

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[^0]:    * $=(\mathrm{rac})$-Phenibut decomposes prior to melting at about $200^{\circ} \mathrm{C}$, as such no melting point could be determined.

