The mechanism of two reversible phase transitions and dielectric anomalies in a singlecomponent organic molecule 1-sulfo-4-dimethylamino-pyridinium

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Figure S1. The powder X-ray diffraction pattern of compound 1 with the simulated

one in red and the experimental one in black.



Figure S2. Infrared diagram of 1.



Figure S3. Thermogravimetric (TGA) curve of 1.



Figure S4. ¹H NMR diagram of 1.



Figure S5. Synthesis and structural diagram of 1.

S1-O1	1.419(2)	N2-C7	1.458(3)
S1-O3	1.4278(18)	N2-C6	1.459(3)
S1-O2	1.429(2)	C1-C2	1.349(3)
S1-N1	1.7697(19)	C2-C3	1.427(3)
N1-C1	1.362(3)	C3-C4	1.425(3)
N1-C5	1.361(3)	C4-C5	1.350(3)
N2-C3	1.334(3)		
O1-S1-O3	116.22(13)	C3-N2-C6	121.2(2)
O1-S1-O2	117.08(14)	C7-N2-C6	117.28(19)
O3-S1-O2	113.15(12)	C2-C1-N1	121.9(2)
01-S1-N1	102.05(10)	C1-C2-C3	120.5(2)
O3-S1-N1	102.72(10)	N2-C3-C4	122.3(2)
02-S1-N1	102.58(10)	N2-C3-C2	121.6(2)
C1-N1-C5	119.16(19)	C4-C3-C2	116.1(2)
C1-N1-S1	121.63(15)	C5-C4-C3	120.3(2)
C5-N1-S1	118.86(15)	C4-C5-N1	122.1(2)
C3-N2-C7	121.46(19)		

 Table S1. Bond lengths [Å] and angles [°] for 1 (243 K).

Table S2. Bond lengths [Å] and angles [°] for 1(293 K)

S1-N1	1.756(4)	N1-C1	1.363(3)
S1-O2 ¹	1.449(5)	N1-C1 ¹	1.363(3)
S1-O2	1.449(5)	N2-C3	1.328(5)
S1-O1 ¹	1.477(5)	N2-C4	1.464(4)
S1-O1	1.477(5)	N2-C4 ¹	1.464(4)
S1-O2A ¹	1.457(8)	C3-C2 ¹	1.420(3)
S1-O2A	1.457(8)	C3-C2	1.420(3)
S1-01A	1.409(7)	C1-C2	1.344(5)
S1-O1A ¹	1.409(7)		
O21-S1-N1	102.0(3)	01A ¹ -S1-N1	109.2(7)
02-S1-N1	102.0(3)	O1A-S1-O2A	59.1(9)
O2-S1-O2 ¹	108.2(6)	O1A ¹ -S1-O2A	122.6(11)
O2 ¹ -S1-O1 ¹	89.4(4)	O1A-S1-O1A ¹	140.9(13)
O2-S1-O1 ¹	147.4(4)	C1-N1-S1	120.90(18)
O2 ¹ -S1-O1	147.4(4)	C11-N1-S1	120.90(18)
02-S1-O1	89.4(4)	C1-N1-C1 ¹	117.9(4)
O21-S1-O2A1	22.0(8)	C3-N2-C4	121.5(2)
O2-S1-O2A ¹	88.7(9)	C3-N2-C4 ¹	121.5(2)
01 ¹ -S1-N1	100.7(2)	C4-N2-C4 ¹	116.9(4)
01-S1-N1	100.7(2)	N2-C3-C2	122.16(18)

01-S1-O1 ¹	63.5(6)	N2-C3-C2 ¹	122.15(18)
O2A ¹ -S1-N1	97.0(9)	C21-C3-C2	115.7(4)
O2A-S1-N1	97.0(9)	C2-C1-N1	122.5(3)
01A-S1-N1	109.2(7)	C1-C2-C3	120.7(3)

Table S3. Bond lengths [Å] and angles [°] for 1 (328 K)

S1-N1	1.776(3)	N1-C1	1.349(3)
S1-O2	1.478(6)	N1-C1 ¹	1.349(3)
S1-O2 ¹	1.478(6)	N2-C3	1.326(5)
S1-O1 ¹	1.480(5)	N2-C4 ¹	1.455(3)
S1-O1	1.480(5)	N2-C4	1.455(3)
S1-O2A	1.466(12)	C3-C2	1.419(3)
S1-O2A ¹	1.466(12)	C3-C2 ¹	1.419(3)
S1-O1A	1.311(9)	C1-C2	1.347(4)
S1-O1A ¹	1.311(9)	02A-01A	1.382(18)
S1-N1	1.776(3)	N1-C1	1.349(3)
O2 ¹ -S1-N1	101.4(3)	01A1-S1-O2A	121.7(9)
O2-S1-N1	101.4(3)	O1A-S1-O1A ¹	143.8(11)
O2-S1-O2 ¹	107.1(5)	C1-N1-S1	120.19(16)
O2-S1-O1 ¹	149.2(4)	C1 ¹ -N1-S1	120.19(16)
O2-S1-O1	89.4(4)	C1-N1-C1 ¹	119.4(3)
O2 ¹ -S1-O1	149.2(4)	C3-N2-C4 ¹	121.44(19)
O2 ¹ -S1-O1 ¹	89.4(4)	C3-N2-C4	121.44(19)
01-S1-N1	100.5(2)	C41-N2-C4	117.0(4)
O1 ¹ -S1-N1	100.5(2)	N2-C3-C2 ¹	122.14(16)
01-S1-O1 ¹	65.5(5)	N2-C3-C2	122.15(16)
O2A-S1-N1	98.1(8)	C2-C3-C2 ¹	115.7(3)
O2A ¹ -S1-N1	98.1(8)	C2-C1-N1	121.8(2)
O1A ¹⁻ S1-N1	107.7(5)	C1-C2-C3	120.6(2)
01A-S1-N1	107.7(5)	01A-02A-S1	54.7(6)
01A-S1-O2A	59.4(7)	S1-O1A-O2A	65.9(7)