

SUPPLEMENTARY INFORMATION

Charge doping to flat AgF₂ monolayers in a chemical capacitor setup

Daniel Jezierski¹, Adam Grzelak^{1*}, Liu Xiao-Qiang², Shishir Kumar Pandey², Maria N. Gastiasoro,³ José Lorenzana³, Ji Feng^{2,4} and Wojciech Grochala^{1*}

¹Center of New Technologies, University of Warsaw, 02089 Warsaw, Poland

²International Center for Quantum Materials, School of Physics, Peking University, Beijing, 100871, China

³Institute for Complex Systems (ISC), Consiglio Nazionale delle Ricerche, Dipartimento di Fisica, Università di Roma "La Sapienza", 00185 Rome, Italy

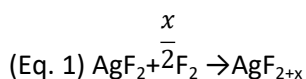
⁴Collaborative Innovation Center of Quantum Matter, Beijing, 100871, China

*a.grzelak@cent.uw.edu.pl, w.grochala@cent.uw.edu.pl

S1. Thermodynamics of fluorination/oxidation reactions

In this section we discuss the stability of hypothetical fluorinated AgF₂ monolayers, in terms of the system's tendency towards decomposition into AgF₂ and F₂. We will consider only energy-preferred (GS) AgF_{2+x} solutions: for RMF – flat and without distortion in the entire fluorination range; for LiF (LF) substrate – flat ($x = \frac{1}{2}(I)$ and 1) and slightly puckered ($x = \frac{1}{4}$ and $\frac{1}{2}(II)$) AgF_{2+x} monolayer.

The reaction (eq. 1) of fluorination of AgF₂ monolayer:



The ΔG value of fluorination reaction (eq. 1) is defined as follows:

$$\text{(Eq. 2) } \Delta G_{fluor.} = G_{\text{AgF}_{2+x}} - (G_{\text{AgF}_2} + \frac{x}{2}G_{\text{F}_2})$$

The enthalpy of reaction in eq. 1), calculated in the manner presented in eq. 2, enables us to estimate the stability of each AgF_{2+x} monolayer, formed upon fluorination. For negative values of $\Delta G_{fluor.}$, we can expect the formation of thermodynamically stable AgF_{2+x} phases. The $\frac{x}{2}G_{\text{F}_2}$ value is ground-state energy of α -F₂ crystal, obtained from DFT calculations, where ΔH value is equal 0.

To face experimental-related problems of fluorination, we also calculated temperature values (eq. 3), in which entropy term due to $n\text{F}_2(\text{gas})$ gas evolution is equal to energy effect (eq.2) of AgF₂ monolayer fluorination, considering the number of fluorine atoms involved. The standard entropy (ΔS°) for 1 mole of F₂ gas is equal ca. 202.8 J/mol*K.¹ It yields ΔG equal to ca. -1.05 meV for 0.5 molecule of F₂ at 1K, based on equation 3. Therefore, the temperature may be a factor by changing which, one can manipulate the degree of AgF₂ fluorination. Dependence of enthalpy of AgF₂ monolayer fluorination reaction (eq.1, eq.2) and temperatures evaluated from Eq. 3 on the degree of fluorination, are presented in the Fig. S1 (A) and (B).

$$\text{(Eq. 3) } \Delta G_{\frac{x}{2}\text{F}_2} = -\frac{x}{2}TS_{(\text{F}_2\text{gas})}$$

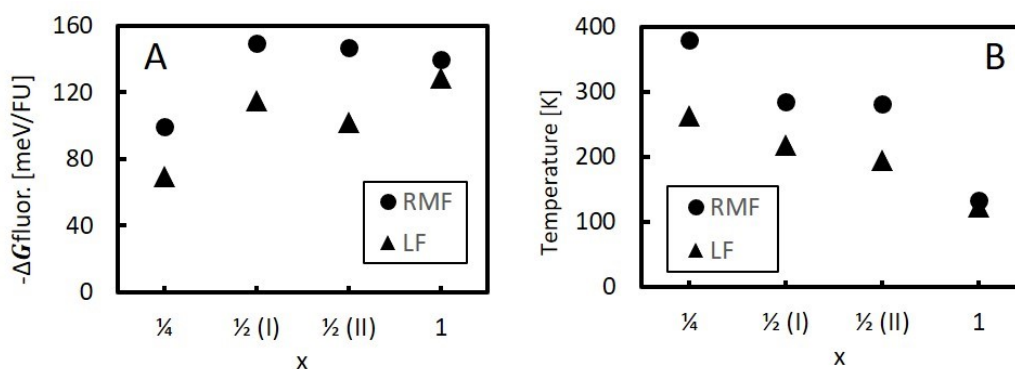


Figure S1. A – Dependence of $-\Delta G_{fluor.}$ on AgF_2 fluorination degree (x) for RMF and LF substrates, B – dependence of temperature versus AgF_2 fluorination degree of monolayer, placed on RMF and LF.

Fig. S1 (A) shows the dependence of $\Delta G_{fluor.}$ on degree of fluorination. The enthalpies of fluorination reactions (eq.1) are negative within the entire fluorination range studied. For LF values are slightly lower, compared to RMF. On the basis of these results, it may be tempted to state, that with Ag-Ag distance decreasing, the tendency toward fluorination increases. However, for full fluorination of AgF_2 monolayer, enthalpies of fluorination reaction differ only by 10.9 meV/FU between two different substrates; for RMF and LiF: ca. -139.9 meV/FU and ca. -129.0 meV/FU, respectively. This is also reflected on the values of temperatures, below which the given AgF_{2+x} phase may exist. The $x = \frac{1}{4}$ fluorination of AgF_2 monolayer should be feasible around 380K and 264K for RMF and LF, respectively. As a degree of fluorination progresses, the temperature values decrease; from 218K ($x = \frac{1}{2}$ (I)) to 123K ($x = 1$), and from 285K ($x = \frac{1}{2}$ (I)) to 133K ($x = 1$), for LF and RMF, respectively. Rather low values of temperatures marking stability of Ag(III) fluoride agree with the fact that AgF_3 prepared in the laboratory is quite thermally unstable.^{ref Zemva}

In the case of indirect fluorination, the reaction is unfavorable in the entire range of x (fig. S2).

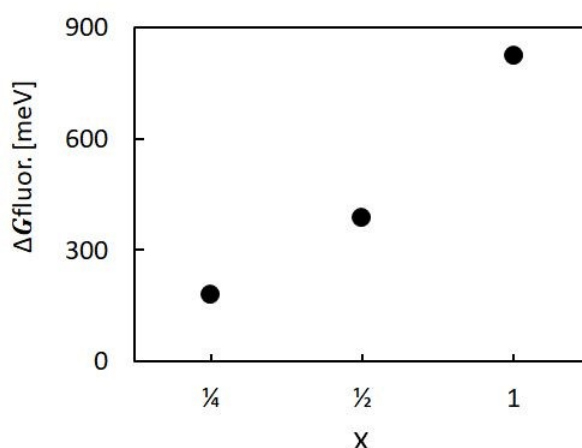


Figure S2. Dependence of $\Delta G_{fluor.}$ on AgF_2 fluorination degree (x) for indirectly fluorinated LF- AgF_2 - $\text{Li}_2\text{F}_{2+x}$ systems.

S2. Expanded eDOS graphs for hole-doped systems.

(1) Direct fluorination: RMF-AgF_{2+x} and LF-AgF_{2+x} systems

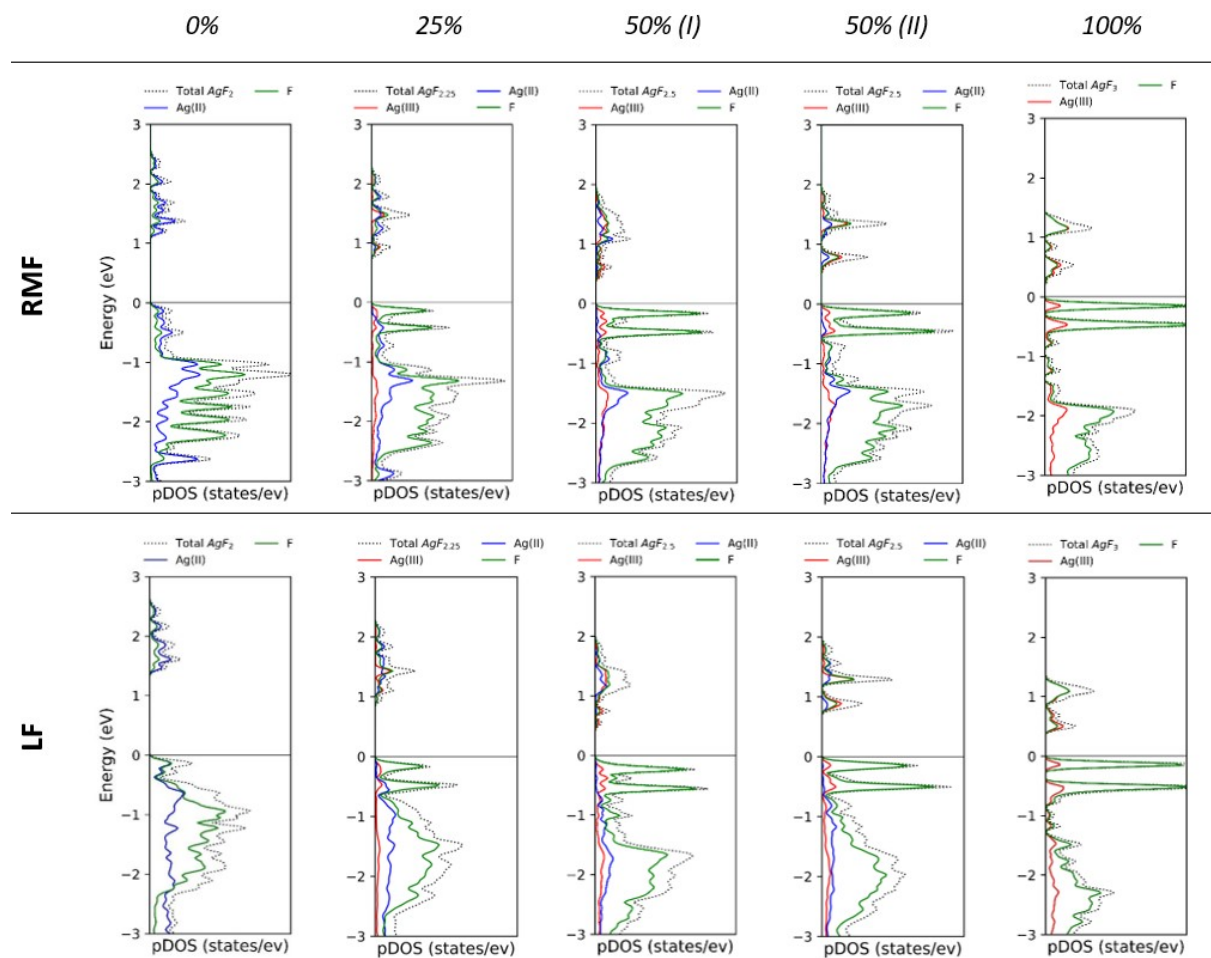


Figure S3. Atom-resolved eDOS for AgF_{2+x} monolayer placed on RMF (up) and LF (down) substrate.

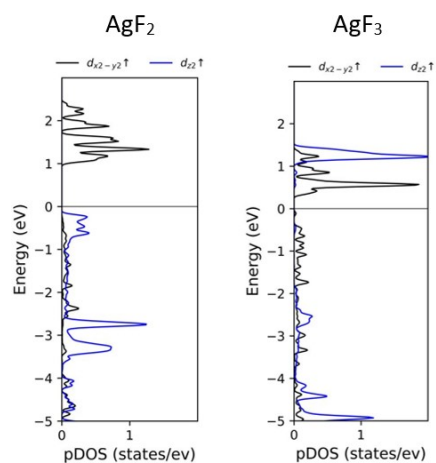


Figure S4. Orbital-resolved eDOS for $d_{x^2-y^2}$ and d_{z^2} orbitals for silver atoms from AgF₂ (right) and AgF₃ (left) monolayer placed on RMF substrate

(2) Indirect fluorination: LF-AgF₂Li₂F_{2+x} systems

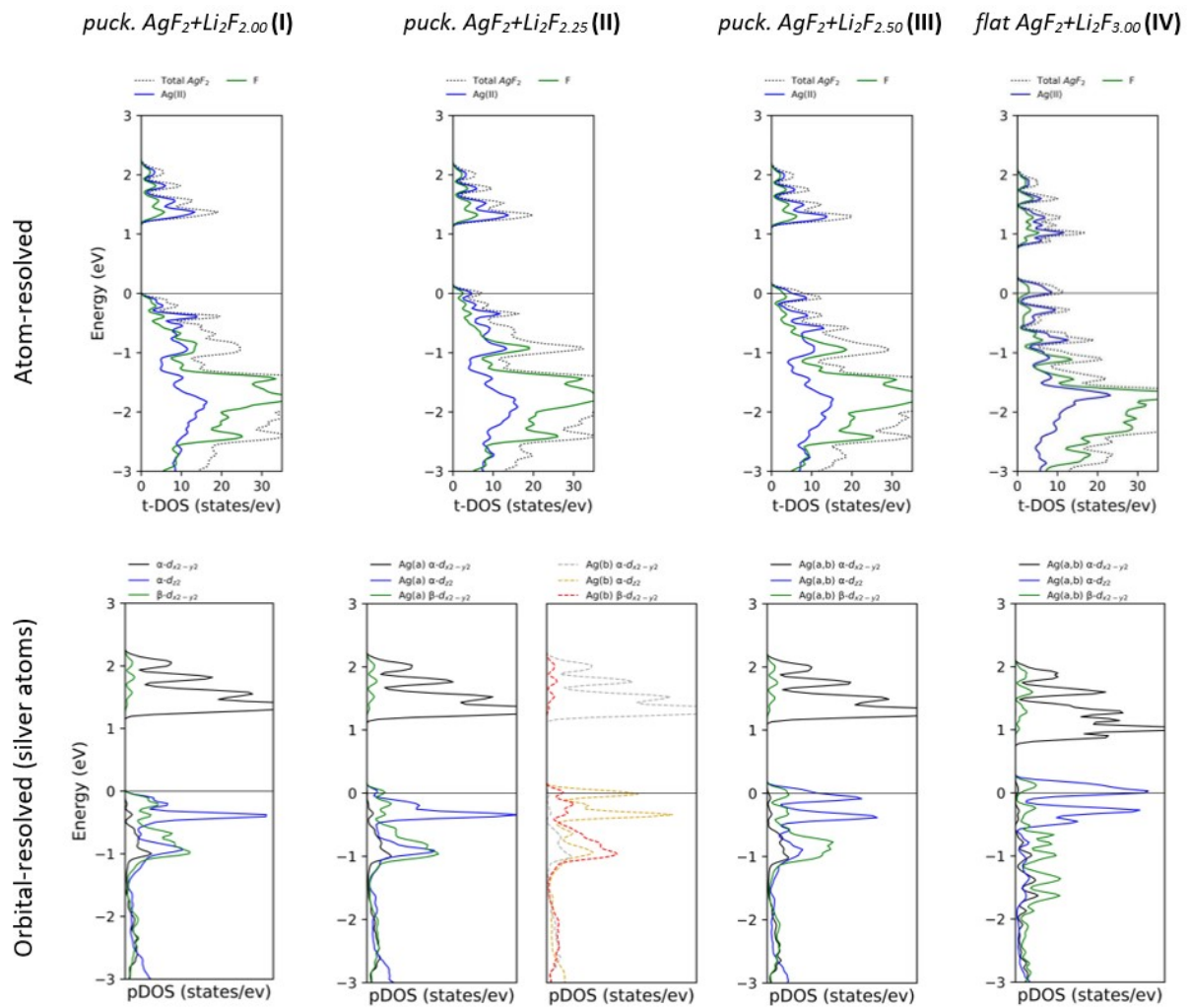


Figure S4. Atom-resolved (upper part) and orbital-resolved (lower part) eDOS for AgF₂ monolayer within the entire (0%, 25%, 50% and 100%) of Li₂F₂ atop layer fluorination.

(3) Reaction with strong oxidizer: RMF-AgF₂-PtF₆ system

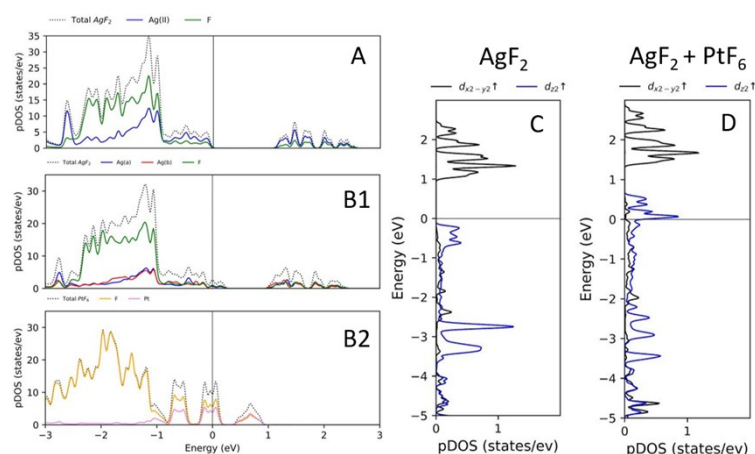


Figure S5. Atom-resolved pDOS for: **(A)** AgF₂ monolayer on RMF substrate, **(B1)** AgF₂ bonded to PtF₆ molecule, and PtF₆ **(B2)**. The right part presents orbital-resolved DOS of Ag α-d_{x²-y²} and α-d_{z²} states for: **(C)** pristine AgF₂ and **(D)** bonded to PtF₆ monolayer.

S3. Selected structures studied.

RbMgF₃-AgF₂

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RbMgF₃-AgF_{2.5} (I) ($\frac{1}{2}$ fluorination)

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RbMgF₃-AgF₃ (full fluorination)

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RbMgF₃-AgF₂-PtF₆ (strong oxidizer)

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Li20 Li 2 d 0.00000 0.50000 0.50000 1.00000 0,0,0
F1 F 4 k 0.25000 0.72172 0.75000 1.00000 0,Dy,Dz
F2 F 4 k 0.25000 0.27698 0.25000 1.00000 0,Dy,Dz
F3 F 4 g 0.00000 0.30570 0.00000 1.00000 0,Dy,0
F4 F 4 h 0.00000 0.69439 0.50000 1.00000 0,Dy,0
F5 F 4 k 0.25000 0.33333 0.75000 1.00000 0,Dy,Dz
F6 F 4 k 0.25000 0.66668 0.25000 1.00000 0,Dy,Dz
F7 F 4 h 0.00000 0.36104 0.50000 1.00000 0,Dy,0
F8 F 4 g 0.00000 0.63896 0.00000 1.00000 0,Dy,0
F9 F 4 k 0.25000 0.61119 0.25000 1.00000 0,Dy,Dz
F10 F 4 k 0.25000 0.61119 0.75000 1.00000 0,Dy,Dz
F11 F 4 h 0.00000 0.41673 0.50000 1.00000 0,Dy,0
F12 F 4 g 0.00000 0.41673 0.00000 1.00000 0,Dy,0
F13 F 4 k 0.25000 0.44449 0.25000 1.00000 0,Dy,Dz
F14 F 4 k 0.25000 0.55551 0.75000 1.00000 0,Dy,Dz
F15 F 4 h 0.00000 0.52776 0.50000 1.00000 0,Dy,0
F16 F 4 g 0.00000 0.47224 0.00000 1.00000 0,Dy,0
F17 F 2 f 0.25000 0.50000 0.75000 1.00000 0,0,Dz
F18 F 2 f 0.25000 0.50000 0.25000 1.00000 0,0,Dz
F19 F 4 k 0.25000 0.20700 0.25000 1.00000 0,Dy,Dz
F20 F 4 k 0.25000 0.79744 0.75000 1.00000 0,Dy,Dz
F21 F 4 g 0.00000 0.76476 0.00000 1.00000 0,Dy,0
F22 F 4 h 0.00000 0.24810 0.50000 1.00000 0,Dy,0
Ag1 Ag 4 k 0.25000 0.24168 0.25000 1.00000 0,Dy,Dz

end of cif

1

LiF-AgF₂-Li₂F_{2.5} (¹/₂ fluorination)

CIF file created by FINDSYM, version 7.1.3

data_findsym-output

_audit_creation_method FINDSYM

_cell_length_a 5.6492174963

_cell_length_b 71.9581220000

_cell_length_c 5.6492174963

_cell_angle_alpha 90.0000000000

_cell_angle_beta 90.0000000000

_cell_angle_gamma 90.0000000000

_cell_volume 2296.4469188569

_symmetry_space_group_name_H-M "P 1 2/m 1"

_symmetry_Int_Tables_number 10

_space_group.reference_setting '010:-P 2y'

_space_group.transform_Pp_abc a,b,c;0,0,0

loop_

_space_group_symop_id

_space_group_symop_operation_xyz

1 x,y,z

2 -x,y,-z

3 -x,-y,-z

4 x,-y,z

loop_

_atom_site_label

_atom_site_type_symbol

_atom_site_symmetry_multiplicity

_atom_site_Wyckoff_label

_atom_site_fract_x

_atom_site_fract_y

_atom_site_fract_z

_atom_site_occupancy

_atom_site_fract_symmform

Li1 Li 2 i 0.00000 0.20479 0.00000 1.00000 0,Dy,0

Li2 Li 2 l 0.50000 0.20296 0.50000 1.00000 0,Dy,0

Li3 Li 2 k 0.00000 0.20651 0.50000 1.00000 0,Dy,0

Li4 Li 2 j 0.50000 0.20234 0.00000 1.00000 0,Dy,0

Li5 Li 2 i 0.00000 0.27890 0.00000 1.00000 0,Dy,0

Li6 Li 2 l 0.50000 0.27665 0.50000 1.00000 0,Dy,0

Li7 Li 2 k 0.00000 0.27903 0.50000 1.00000 0,Dy,0

Li8 Li 2 j 0.50000 0.27693 0.00000 1.00000 0,Dy,0

Li9 Li 4 o 0.25000 0.30577 0.25000 1.00000 Dx,Dy,Dz

Li10 Li 4 o 0.75000 0.30529 0.25000 1.00000 Dx,Dy,Dz
Li11 Li 2 l 0.50000 0.33325 0.50000 1.00000 0,Dy,0
Li12 Li 2 i 0.00000 0.33349 0.00000 1.00000 0,Dy,0
Li13 Li 2 k 0.00000 0.33336 0.50000 1.00000 0,Dy,0
Li14 Li 2 j 0.50000 0.33330 0.00000 1.00000 0,Dy,0
Li15 Li 4 o 0.75000 0.36101 0.75000 1.00000 Dx,Dy,Dz
Li16 Li 4 o 0.25000 0.36098 0.75000 1.00000 Dx,Dy,Dz
Li17 Li 2 i 0.00000 0.38881 0.00000 1.00000 0,Dy,0
Li18 Li 2 l 0.50000 0.38881 0.50000 1.00000 0,Dy,0
Li19 Li 2 k 0.00000 0.38886 0.50000 1.00000 0,Dy,0
Li20 Li 2 j 0.50000 0.38863 0.00000 1.00000 0,Dy,0
Li21 Li 4 o 0.25000 0.41673 0.75000 1.00000 Dx,Dy,Dz
Li22 Li 4 o 0.25000 0.41673 0.25000 1.00000 Dx,Dy,Dz
Li23 Li 2 i 0.00000 0.44449 0.00000 1.00000 0,Dy,0
Li24 Li 2 j 0.50000 0.44449 0.00000 1.00000 0,Dy,0
Li25 Li 2 l 0.50000 0.44449 0.50000 1.00000 0,Dy,0
Li26 Li 2 k 0.00000 0.44449 0.50000 1.00000 0,Dy,0
Li27 Li 4 o 0.25000 0.47224 0.75000 1.00000 Dx,Dy,Dz
Li28 Li 4 o 0.25000 0.47224 0.25000 1.00000 Dx,Dy,Dz
Li29 Li 1 b 0.00000 0.50000 0.00000 1.00000 0,0,0
Li30 Li 1 e 0.50000 0.50000 0.00000 1.00000 0,0,0
Li31 Li 1 h 0.50000 0.50000 0.50000 1.00000 0,0,0
Li32 Li 1 f 0.00000 0.50000 0.50000 1.00000 0,0,0
F1 F 4 o 0.75000 0.27831 0.25000 1.00000 Dx,Dy,Dz
F2 F 4 o 0.75000 0.27661 0.75000 1.00000 Dx,Dy,Dz
F3 F 2 k 0.00000 0.30563 0.50000 1.00000 0,Dy,0
F4 F 2 i 0.00000 0.30570 0.00000 1.00000 0,Dy,0
F5 F 2 j 0.50000 0.30551 0.00000 1.00000 0,Dy,0
F6 F 2 l 0.50000 0.30559 0.50000 1.00000 0,Dy,0
F7 F 4 o 0.25000 0.33331 0.75000 1.00000 Dx,Dy,Dz
F8 F 4 o 0.25000 0.33332 0.25000 1.00000 Dx,Dy,Dz
F9 F 2 l 0.50000 0.36097 0.50000 1.00000 0,Dy,0
F10 F 2 j 0.50000 0.36104 0.00000 1.00000 0,Dy,0
F11 F 2 i 0.00000 0.36097 0.00000 1.00000 0,Dy,0
F12 F 2 k 0.00000 0.36105 0.50000 1.00000 0,Dy,0
F13 F 4 o 0.25000 0.38881 0.25000 1.00000 Dx,Dy,Dz
F14 F 4 o 0.75000 0.38880 0.25000 1.00000 Dx,Dy,Dz
F15 F 2 l 0.50000 0.41673 0.50000 1.00000 0,Dy,0
F16 F 2 k 0.00000 0.41673 0.50000 1.00000 0,Dy,0
F17 F 2 j 0.50000 0.41673 0.00000 1.00000 0,Dy,0
F18 F 2 i 0.00000 0.41673 0.00000 1.00000 0,Dy,0
F19 F 4 o 0.75000 0.44449 0.75000 1.00000 Dx,Dy,Dz
F20 F 4 o 0.75000 0.44449 0.25000 1.00000 Dx,Dy,Dz
F21 F 2 j 0.50000 0.47224 0.00000 1.00000 0,Dy,0
F22 F 2 k 0.00000 0.47224 0.50000 1.00000 0,Dy,0
F23 F 2 l 0.50000 0.47224 0.50000 1.00000 0,Dy,0
F24 F 2 i 0.00000 0.47224 0.00000 1.00000 0,Dy,0
F25 F 2 n 0.75000 0.50000 0.25000 1.00000 Dx,0,Dz

F26 F 2 n 0.25000 0.50000 0.25000 1.00000 Dx,0,Dz
F27 F 4 o 0.73475 0.20839 0.74501 1.00000 Dx,Dy,Dz
F28 F 4 o 0.74444 0.20328 0.25411 1.00000 Dx,Dy,Dz
F29 F 2 i 0.00000 0.82060 0.00000 1.00000 0,Dy,0
F30 F 2 i 0.00000 0.23564 0.00000 1.00000 0,Dy,0
F31 F 2 k 0.00000 0.23544 0.50000 1.00000 0,Dy,0
F32 F 2 l 0.50000 0.24813 0.50000 1.00000 0,Dy,0
F33 F 2 j 0.50000 0.24809 0.00000 1.00000 0,Dy,0
Ag1 Ag 4 o 0.25000 0.75855 0.25000 1.00000 Dx,Dy,Dz

end of cif

LiF-AgF₂-Li₂F₃ (full fluorination)

CIF file created by FINDSYM, version 7.1.3

data_findsym-output

_audit_creation_method FINDSYM

_cell_length_a 5.6492200000

_cell_length_b 71.9581220000

_cell_length_c 5.6492200000

_cell_angle_alpha 90.0000000000

_cell_angle_beta 90.0000000000

_cell_angle_gamma 90.0000000000

_cell_volume 2296.4489544370

_symmetry_space_group_name_H-M "P 1 2/m 1"

_symmetry_Int_Tables_number 10

_space_group.reference_setting '010:-P 2y'

_space_group.transform_Pp_abc a,b,c;0,0,0

loop_

_space_group_symop_id

_space_group_symop_operation_xyz

1 x,y,z

2 -x,y,-z

3 -x,-y,-z

4 x,-y,z

loop_

_atom_site_label

_atom_site_type_symbol

_atom_site_symmetry_multiplicity

_atom_site_Wyckoff_label

_atom_site_fract_x

_atom_site_fract_y

_atom_site_fract_z

_atom_site_occupancy

_atom_site_fract_symmform

Li1 Li 2 k 0.00000 0.20634 0.50000 1.00000 0,Dy,0
Li2 Li 2 j 0.50000 0.20328 0.00000 1.00000 0,Dy,0
Li3 Li 2 i 0.00000 0.20511 0.00000 1.00000 0,Dy,0
Li4 Li 2 l 0.50000 0.20023 0.50000 1.00000 0,Dy,0
Li5 Li 2 k 0.00000 0.27874 0.50000 1.00000 0,Dy,0
Li6 Li 2 j 0.50000 0.27648 0.00000 1.00000 0,Dy,0
Li7 Li 2 i 0.00000 0.27854 0.00000 1.00000 0,Dy,0
Li8 Li 2 l 0.50000 0.27642 0.50000 1.00000 0,Dy,0
Li9 Li 4 o 0.25000 0.30575 0.75000 1.00000 Dx,Dy,Dz
Li10 Li 4 o 0.75000 0.30522 0.75000 1.00000 Dx,Dy,Dz
Li11 Li 2 j 0.50000 0.33323 0.00000 1.00000 0,Dy,0
Li12 Li 2 k 0.00000 0.33339 0.50000 1.00000 0,Dy,0
Li13 Li 2 i 0.00000 0.33333 0.00000 1.00000 0,Dy,0
Li14 Li 2 l 0.50000 0.33328 0.50000 1.00000 0,Dy,0
Li15 Li 4 o 0.75000 0.36099 0.25000 1.00000 Dx,Dy,Dz
Li16 Li 4 o 0.25000 0.36095 0.25000 1.00000 Dx,Dy,Dz
Li17 Li 2 k 0.00000 0.38881 0.50000 1.00000 0,Dy,0
Li18 Li 2 j 0.50000 0.38880 0.00000 1.00000 0,Dy,0
Li19 Li 2 i 0.00000 0.38886 0.00000 1.00000 0,Dy,0
Li20 Li 2 l 0.50000 0.38863 0.50000 1.00000 0,Dy,0
Li21 Li 4 o 0.25000 0.41673 0.25000 1.00000 Dx,Dy,Dz
Li22 Li 4 o 0.25000 0.41673 0.75000 1.00000 Dx,Dy,Dz
Li23 Li 2 k 0.00000 0.44449 0.50000 1.00000 0,Dy,0
Li24 Li 2 l 0.50000 0.44449 0.50000 1.00000 0,Dy,0
Li25 Li 2 j 0.50000 0.44449 0.00000 1.00000 0,Dy,0
Li26 Li 2 i 0.00000 0.44449 0.00000 1.00000 0,Dy,0
Li27 Li 4 o 0.25000 0.47224 0.25000 1.00000 Dx,Dy,Dz
Li28 Li 4 o 0.25000 0.47224 0.75000 1.00000 Dx,Dy,Dz
Li29 Li 1 f 0.00000 0.50000 0.50000 1.00000 0,0,0
Li30 Li 1 h 0.50000 0.50000 0.50000 1.00000 0,0,0
Li31 Li 1 e 0.50000 0.50000 0.00000 1.00000 0,0,0
Li32 Li 1 b 0.00000 0.50000 0.00000 1.00000 0,0,0
F1 F 4 o 0.75000 0.27828 0.75000 1.00000 Dx,Dy,Dz
F2 F 4 o 0.75000 0.27630 0.25000 1.00000 Dx,Dy,Dz
F3 F 2 i 0.00000 0.30551 0.00000 1.00000 0,Dy,0
F4 F 2 k 0.00000 0.30562 0.50000 1.00000 0,Dy,0
F5 F 2 l 0.50000 0.30529 0.50000 1.00000 0,Dy,0
F6 F 2 j 0.50000 0.30553 0.00000 1.00000 0,Dy,0
F7 F 4 o 0.25000 0.33330 0.25000 1.00000 Dx,Dy,Dz
F8 F 4 o 0.25000 0.33328 0.75000 1.00000 Dx,Dy,Dz
F9 F 2 j 0.50000 0.36094 0.00000 1.00000 0,Dy,0
F10 F 2 l 0.50000 0.36103 0.50000 1.00000 0,Dy,0
F11 F 2 k 0.00000 0.36095 0.50000 1.00000 0,Dy,0
F12 F 2 i 0.00000 0.36104 0.00000 1.00000 0,Dy,0
F13 F 4 o 0.25000 0.38880 0.75000 1.00000 Dx,Dy,Dz
F14 F 4 o 0.75000 0.38880 0.75000 1.00000 Dx,Dy,Dz

F15 F 2 j 0.50000 0.41673 0.00000 1.00000 0,Dy,0
F16 F 2 i 0.00000 0.41673 0.00000 1.00000 0,Dy,0
F17 F 2 l 0.50000 0.41673 0.50000 1.00000 0,Dy,0
F18 F 2 k 0.00000 0.41673 0.50000 1.00000 0,Dy,0
F19 F 4 o 0.75000 0.44449 0.25000 1.00000 Dx,Dy,Dz
F20 F 4 o 0.75000 0.44449 0.75000 1.00000 Dx,Dy,Dz
F21 F 2 l 0.50000 0.47224 0.50000 1.00000 0,Dy,0
F22 F 2 i 0.00000 0.47224 0.00000 1.00000 0,Dy,0
F23 F 2 j 0.50000 0.47224 0.00000 1.00000 0,Dy,0
F24 F 2 k 0.00000 0.47224 0.50000 1.00000 0,Dy,0
F25 F 2 n 0.75000 0.50000 0.75000 1.00000 Dx,0,Dz
F26 F 2 n 0.25000 0.50000 0.75000 1.00000 Dx,0,Dz
F27 F 4 o 0.73549 0.20975 0.24959 1.00000 Dx,Dy,Dz
F28 F 4 o 0.75166 0.20534 0.74964 1.00000 Dx,Dy,Dz
F29 F 2 k 0.00000 0.81897 0.50000 1.00000 0,Dy,0
F30 F 2 j 0.50000 0.82161 0.00000 1.00000 0,Dy,0
F31 F 2 k 0.00000 0.23715 0.50000 1.00000 0,Dy,0
F32 F 2 i 0.00000 0.23687 0.00000 1.00000 0,Dy,0
F33 F 2 j 0.50000 0.24806 0.00000 1.00000 0,Dy,0
F34 F 2 l 0.50000 0.24828 0.50000 1.00000 0,Dy,0
Ag1 Ag 4 o 0.25000 0.75830 0.75000 1.00000 Dx,Dy,Dz

end of cif

References:

- 1 M. W. Chase, *NIST-JANAF Thermochemical Tables, Fourth Edition*, National Institute of Standards and Technology, Gaithersburg, MD, U.S.A, 1998.