

Supplementary Information for
**Decomposition of multifunctionalized α -alkoxyalkyl-hydroperoxides
derived from the reactions of Criegee intermediates with diols in liquid
phases**

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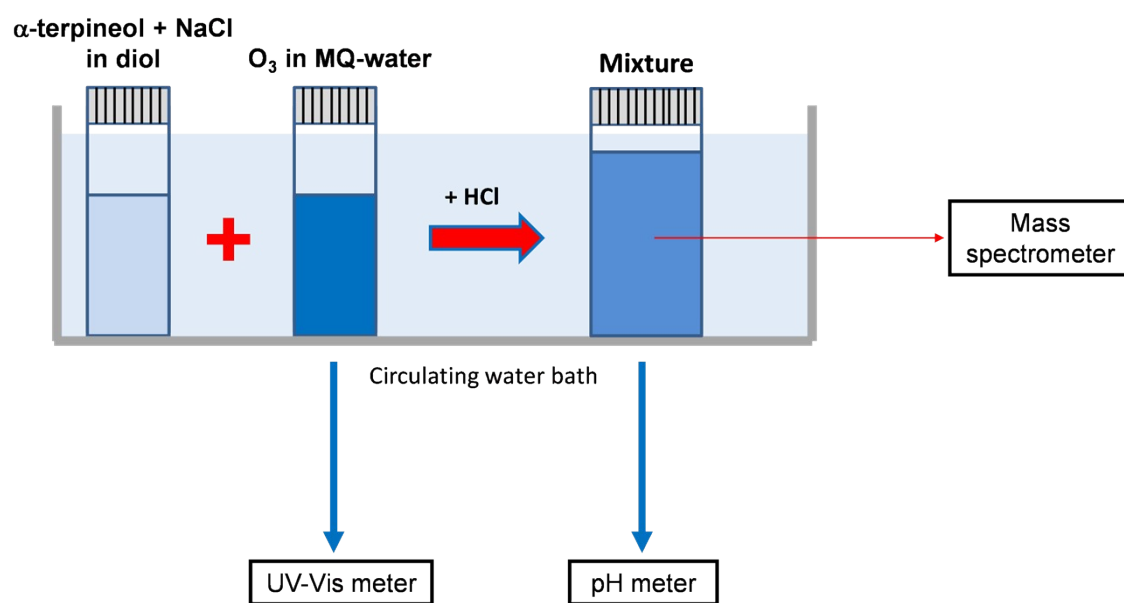


Figure S1 – Schematic setup of present experiment.

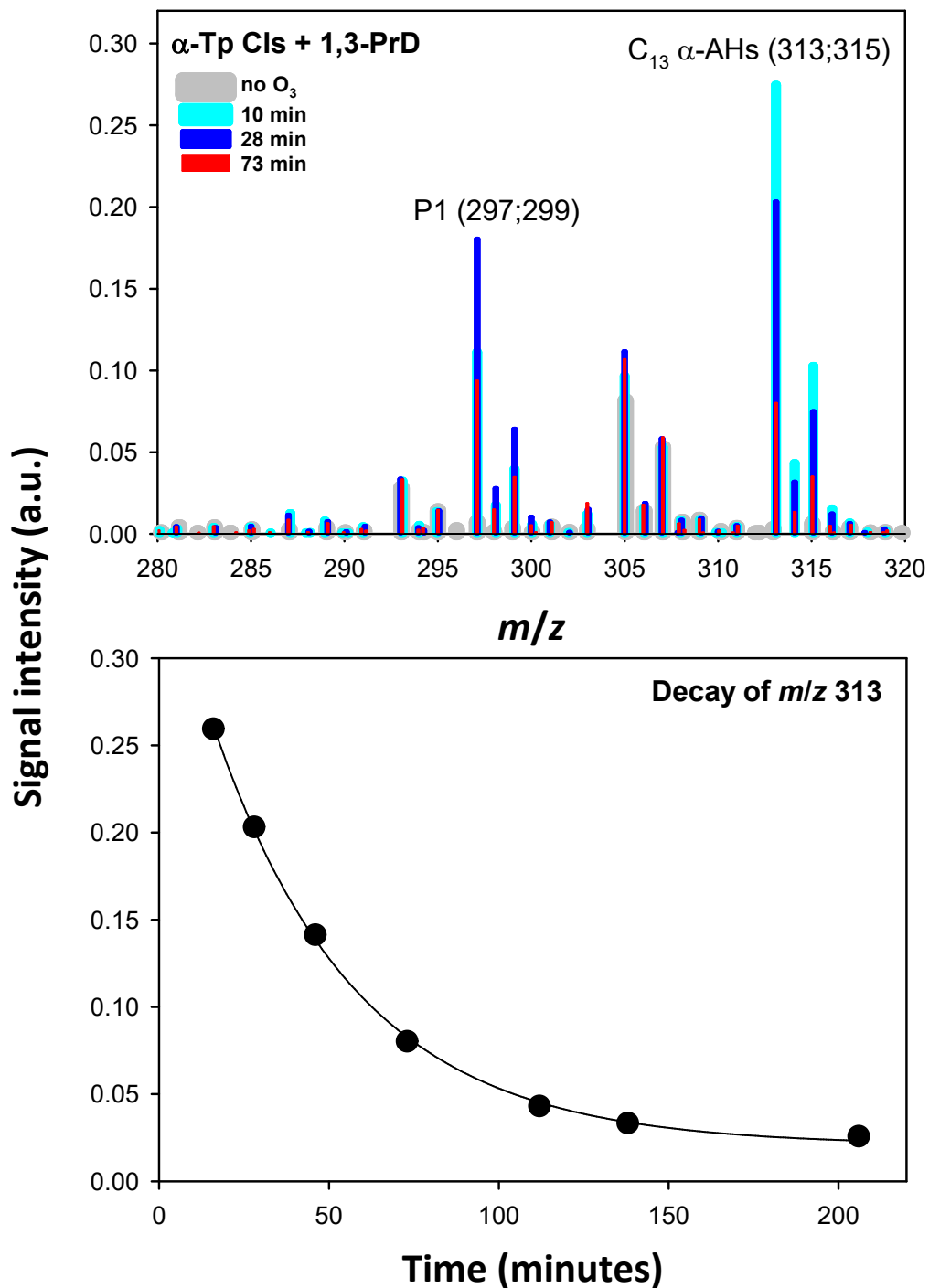


Fig. S2. Upper panel) Negative-ion mass spectra of mixtures obtained by ozonolysis ($[O_3]_0 = 0.06 \pm 0.01$ mM) of α -terpineol (1 mM)/NaCl (0.2 mM) in 1,3-propanediol:H₂O (1:1 = vol:vol) solution at pH 5.1 and $T = 299 \pm 1$ K. Lower panel) The signal intensity at m/z 313 (C_{13} α -AHs) as a function of time.

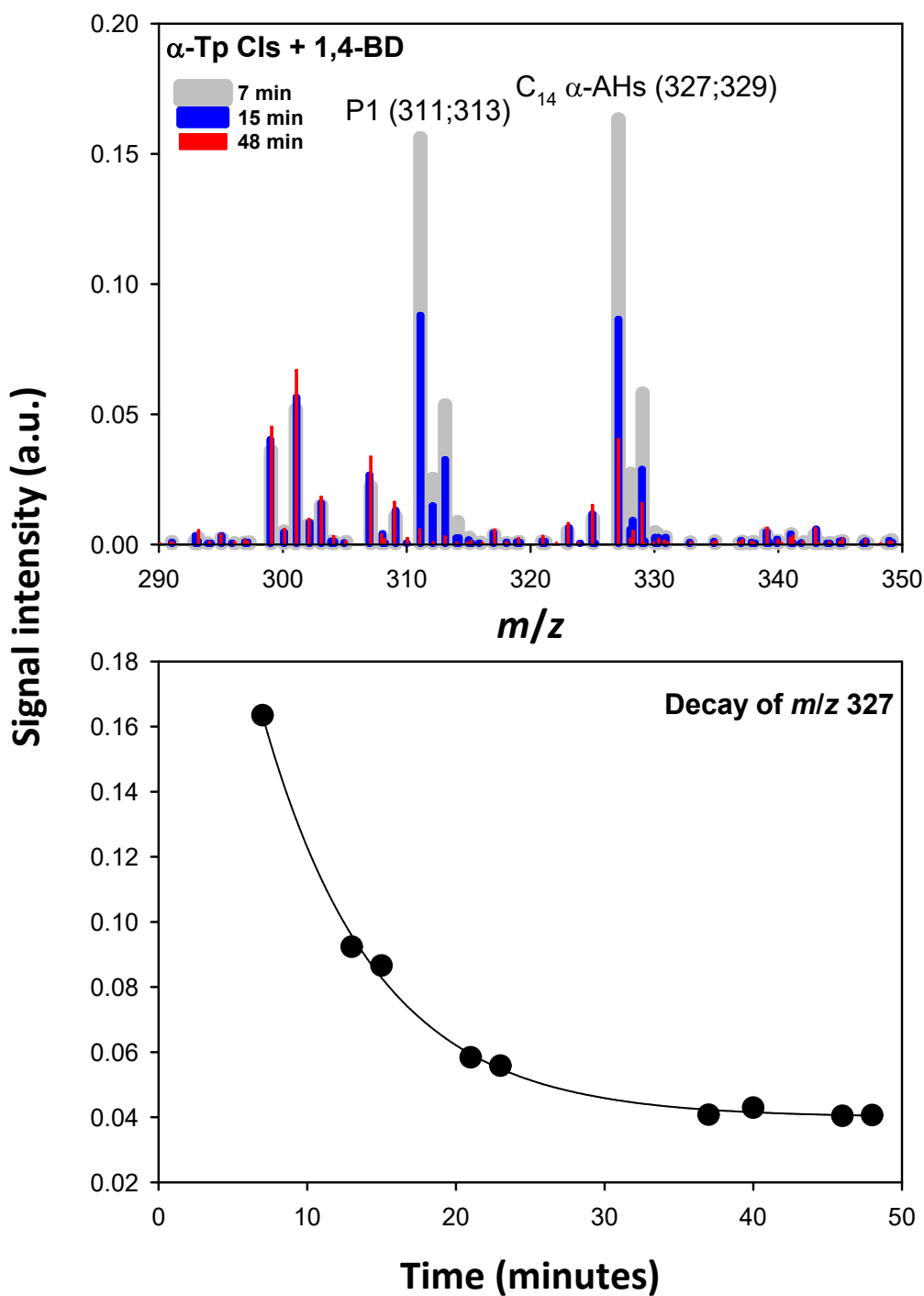


Fig. S3. Upper panel) Negative-ion mass spectra of mixtures obtained by ozonolysis ($[O_3]_0 = 0.06 \pm 0.01$ mM) of α -terpineol (1 mM)/NaCl (0.2 mM) in 1,4-butanediol:H₂O (1:1 = vol:vol) solution at pH 4.5 and $T = 299 \pm 1$ K. Lower panel) The signal intensity at m/z 327 (C₁₄ α -AHs) as a function of time.

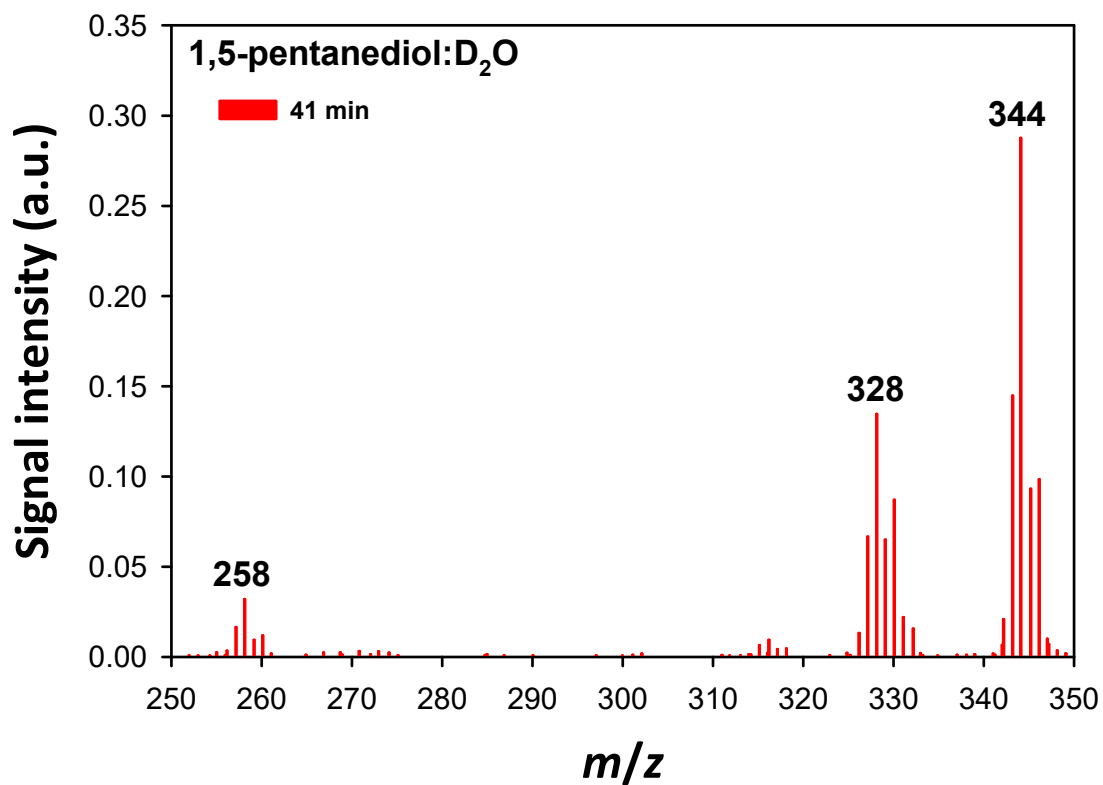


Fig. S4. Negative-ion mass spectra of mixtures obtained by ozonolysis ($[O_3]_0 = 0.06 \pm 0.01$ mM) of α -terpineol (1 mM)/NaCl (0.2 mM)/HCl (0.05 mM) in 1,5-pentanediol:D₂O (1:1 = vol:vol) solution at pD = 5.8 at $T = 299 \pm 1$ K.

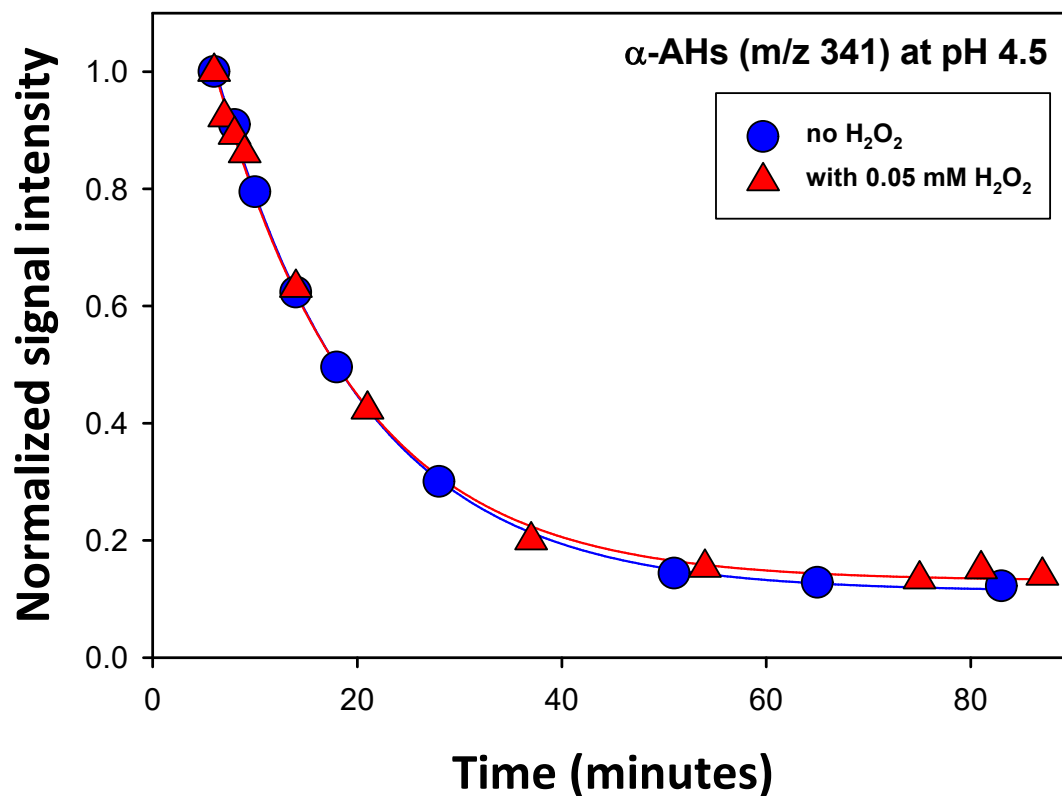


Fig. S5 Temporal profiles of the Cl⁻ adducts of the α-AHs (*m/z* 341) generated by ozonolysis of (1 mM α-terpineol and 0.2 mM NaCl) at [O₃]₀ = 0.06 ± 0.01 mM in a 1,5-pentandiol:water (1:1 = vol:vol) solution in the absence (blue circles) or presence of 0.05 mM H₂O₂ (red triangles) at *T* = 299 ± 1 K acidified by 0.05 mM to pH 4.5. Background signals obtained from mass spectra in the absence of O₃ were subtracted. Lines indicate fits of signal intensities (*S*) to single-exponential functions with baselines.

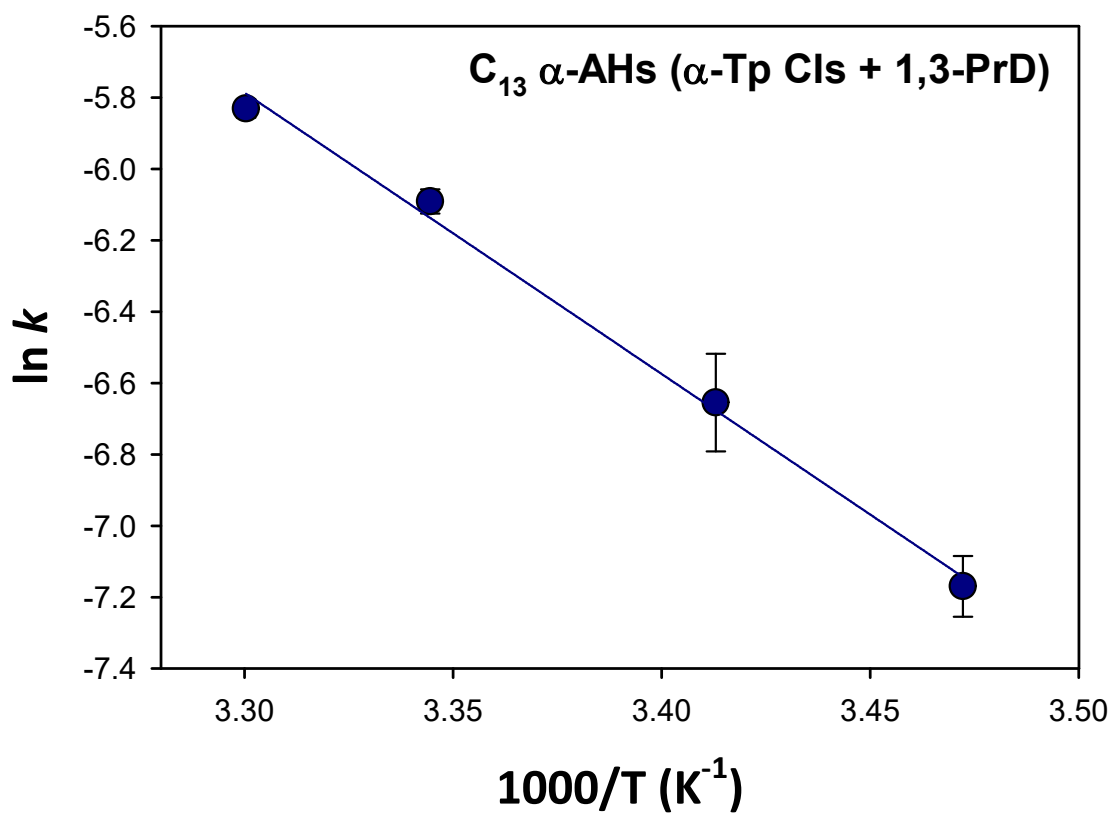


Fig. S6. Arrhenius plot of the rate coefficients for decay of the C_{13} α -AHs generated by ozonolysis of α -terpineol/NaCl in 1,3-propanediol:water (1:1) at pH 4.5. Note the error bars (= SDs) are obscured by the symbols in some cases. The linear regression yielded a preexponential factor (A) of $6.0 \times 10^8 \text{ s}^{-1}$ ($\ln A = 20.2 \pm 1.3$) and an E_a value of $15.7 \pm 0.8 \text{ kcal mol}^{-1}$.

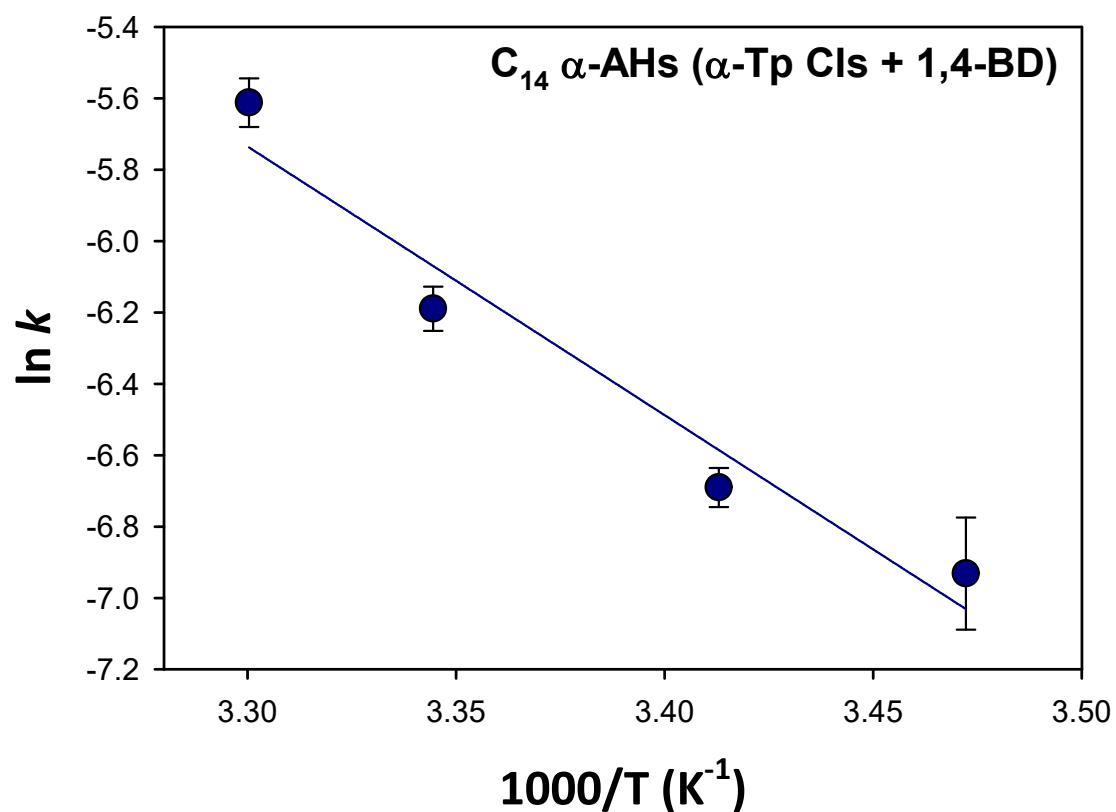


Fig. S7. Arrhenius plot of the rate coefficients for decay of the C_{14} α -AHs generated by ozonolysis of α -terpineol/NaCl in 1,4-butanediol:water (1:1) at pH 4.5. The linear regression yielded a preexponential factor (A) of $2.0 \times 10^8 \text{ s}^{-1}$ ($\ln A = 19.1 \pm 4.1$) and an E_a value of $15.0 \pm 2.4 \text{ kcal mol}^{-1}$.