Electronic Supporting Information for

Climbing up Conical Intersections: A Direct Dynamics Study of the Exotic

Photochemistry of Criegee Intermediates

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Figure S1: Molecular orbitals that make up the 10 electrons in 8 orbitals active space of CH₂OO.



Figure S2: Molecular orbitals that make up the 6 electrons in 5 orbitals active space of formaldehyde.



Figure S3: Energy vs. Time for a representative trajectory that undergoes hopping at long R_{OO}

Figure S4: Comparison between the Relaxed PE profiles along *A*_{HCO} of formaldehyde, calculated at the SS-CASPT2/aug-cc-pVTZ//B3LYP/cc-pVTZ and SS-SR-CASPT2/aug-cc-pVTZ//B3LYP/cc-pVTZ levels of theory.



Table S1: Comparison between the S₀ and T₁ optimized A_{HCO} and R_{CO} values, optimized at the B3LYP/cc-pVTZ and SS-SR-CASPT2/aug-cc-pVTZ levels of theory.

Parameter	S ₀ opt	S ₀ opt	T ₁ opt	T_1 opt
	B3LYP/VTZ	SS-SR-CASPT2	B3LYP/VTZ	SS-SR-CASPT2
		/AVTZ		/AVTZ
$R_{\rm CO}$ / Å	1.19904	1.21006	1.30240	1.31584
$A_{\rm HCO}$ / deg.	122.12062	121.71083	114.02247	113.94501