## Organic and inorganic Gas Sensing Performance of BC3N2 monolayer:a

## first-principles study

Guogang Liu<sup>1</sup>, Tong Chen <sup>2\*</sup>, Xiansheng Dong<sup>2</sup>, Lin Huang<sup>2</sup>, Zhonghui Xu<sup>1</sup>, Xianbo Xiao<sup>3\*</sup>

1 School of Software Engineering, Jiangxi University of Science and Technology, Nanchang 330013, China.

2 School of Energy and Mechanical Engineering, Energy materials computing center, Jiangxi University of Science and Technology, Nanchang 330013, China.

3 School of Computer Science, Jiangxi University of Traditional Chinese Medicine, Nanchang 330004, China

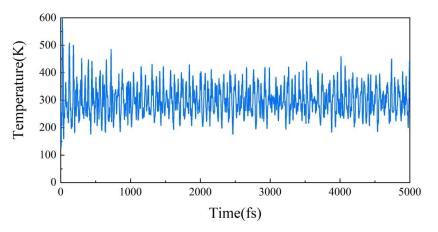


Fig. S1: The molecular dynamics simulation of 2D  $BC_3N_2$  monolayer.

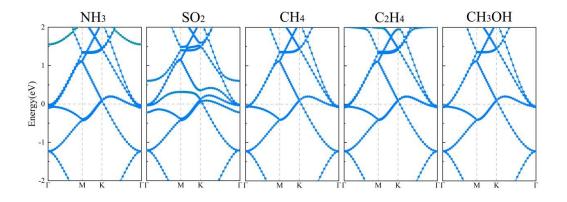


Fig. S2: The band structures of physical adsorption molecules of  $NH_3$  (a),  $SO_2$  (b),  $CH_4$  (c),  $C_2H_4$  (d) and  $CH_3OH$  (e) on  $BC_3N_2$  monolayer.

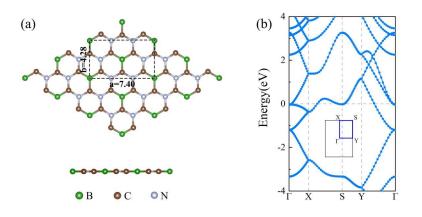


Fig. S3: The rectangled supercell of  $BC_3N_2$  monolayer outlined with black dashed lines (a) and its corresponding electronic band structure (b).

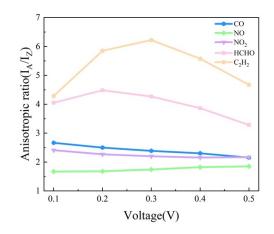


Fig. S4: The anisotropic ratio of the sensitivity of CO NO, NO<sub>2</sub>, HCHO and  $C_2H_2$  along the A direction to that along the Z direction (Ia/Iz)

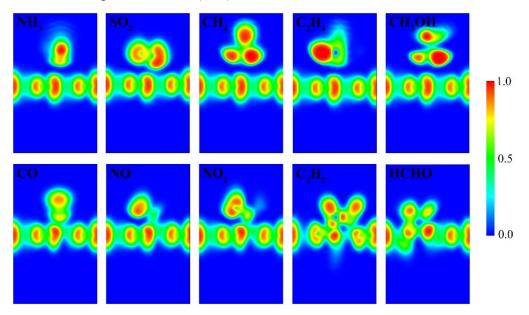


Fig. S5: The ELF maps for gas molecules adsorbed on  $BC_3N_2$  monolayer.