

## Carbon-doped anatase titania nanoparticles: similarities and differences with respect to bulk and extended surface models

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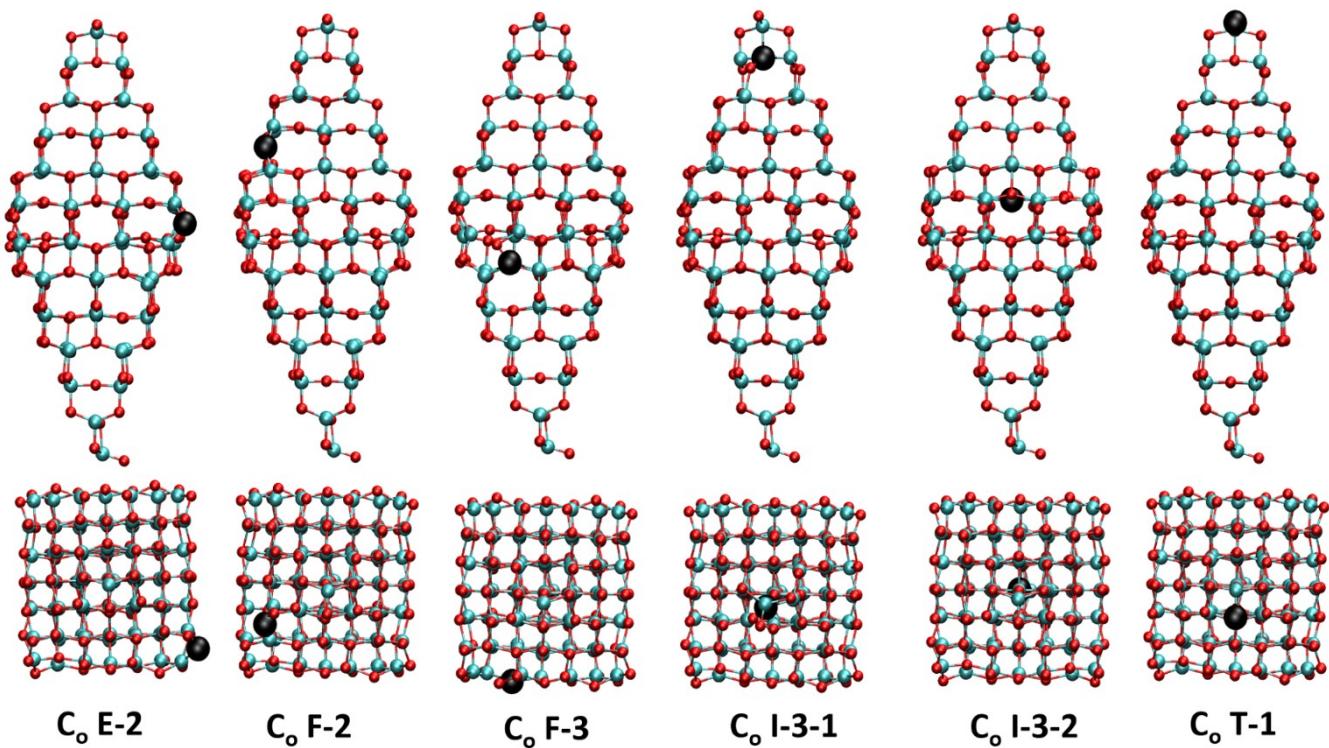
Figures S1 to S3 shows to the sites where take place the doping at oxygen, titanium and interstitial sites, respectively.

Table S1 compiles the formation energy of C-doping at oxygen and titanium sites in the anatase bulk phase.

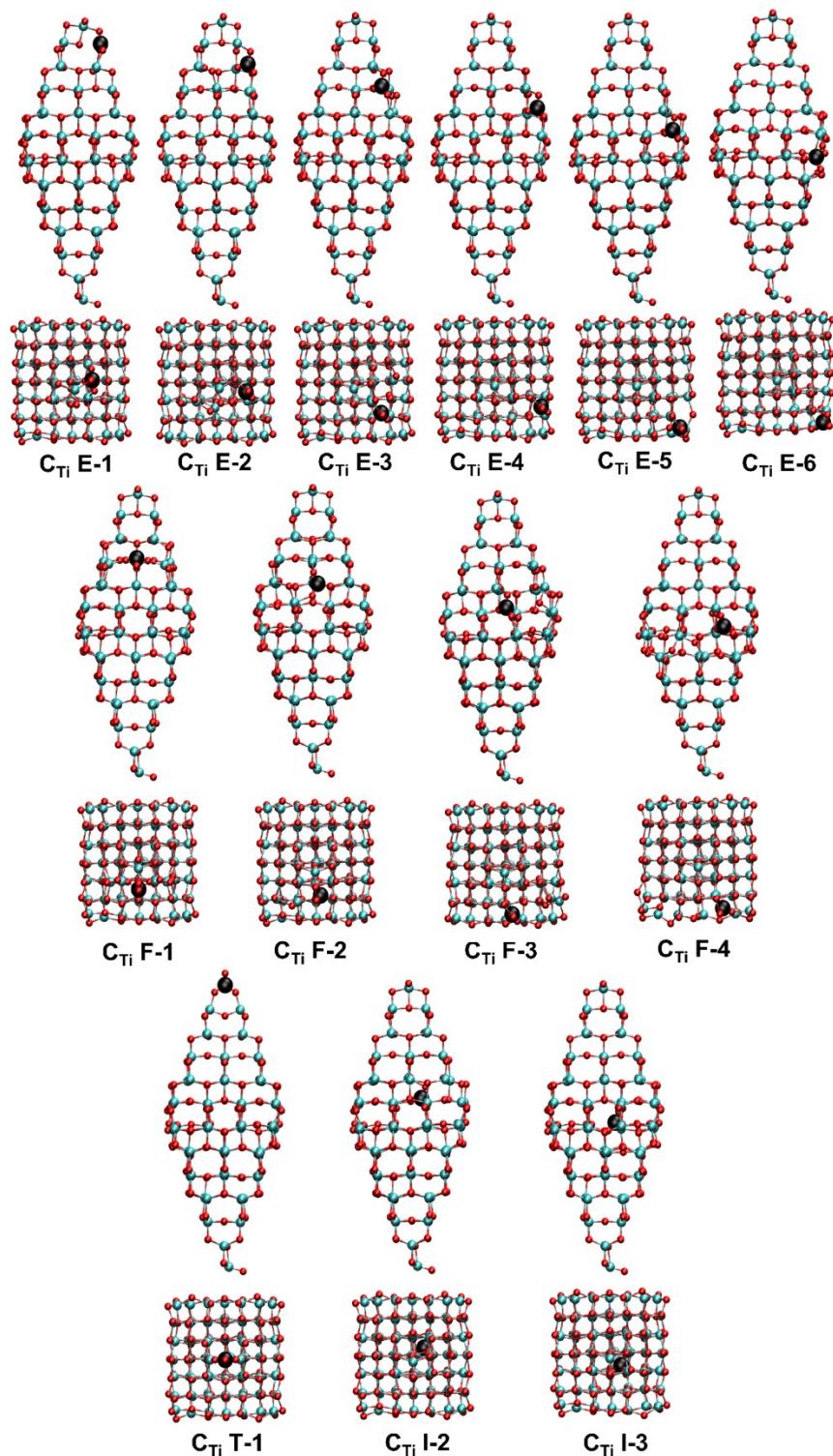
Table S2 compiles the formation energy of C-doping titania at oxygen, titanium and interstitial sites in the  $(\text{TiO}_2)_{84}$  NP.

Table S3 compiles the energy gap, and the energy orbitals LUMO (eV) and HOMO (eV) of carbon doping titania using PBE and PBEx density functionals.

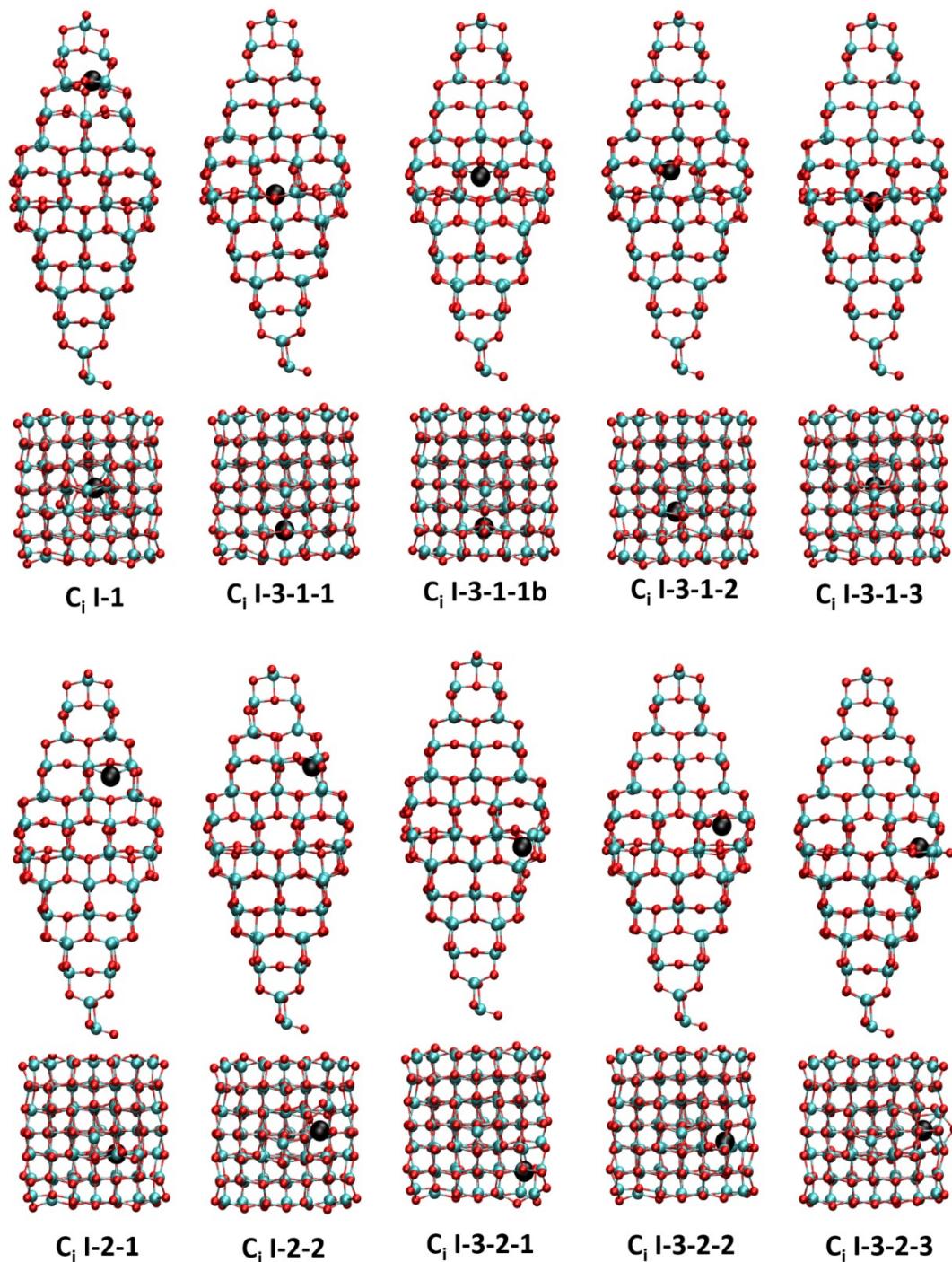
**Fig. S1** Side and top views of the optimized  $(\text{TiO}_2)_{84}$  NPs C-doping at O sites ( $\text{C}_\text{O}$ ). Blue, red, and black spheres denote titanium, oxygen, and carbon, respectively. Note that C-atom have been represented with double radius to show clearly doping site.



**Fig. S2** Side and top views of the optimized  $(\text{TiO}_2)_{84}$  NPs C-doping at Ti site ( $\text{C}_{\text{Ti}}$ ). Blue, red, and black spheres denote titanium, oxygen, and carbon, respectively. Note that C-atom have been represented with double radius to show clearly doping site.



**Fig. S3** Side and top views of the different  $C_i$  doping ( $TiO_2$ )<sub>84</sub> NPs. Blue, red, and black spheres denote titanium, oxygen, and carbon, respectively. Note that C-atom have been represented with double radius to show clearly doping site.



**Table S1.** Formation energy (in eV) of C-doped anatase bulk system as a function of the oxygen chemical potential (in eV). Formation of  $C_O$  and  $C_{Ti}$  are included considering the (1x1x1) and the (2x2x2) anatase supercells.

	$\mu_O$				
	-4	-3	-2	-1	0
1x1x1 $C_O$ - TiO <sub>2</sub>	8.38	9.38	10.38	11.38	12.38
1x1x1 $C_{Ti}$ - TiO <sub>2</sub>	16.09	14.09	12.09	10.09	8.09
2x2x2 $C_O$ - TiO <sub>2</sub>	8.50	9.50	10.50	11.50	12.50
2x2x2 $C_{Ti}$ - TiO <sub>2</sub>	17.17	15.17	13.17	11.17	9.17

**Table S2.** Formation energy (in eV) of C-doped TiO<sub>2</sub> NPs as a function of the oxygen chemical potential (in eV).

NP	$\mu_O'$				
	-4	-3	-2	-1	0
C <sub>O</sub> E-2	7.41	8.41	9.41	10.41	11.41
C <sub>O</sub> F-2	7.68	8.68	9.68	10.68	11.68
C <sub>O</sub> F-3	7.70	8.70	9.70	10.70	11.70
C <sub>O</sub> I-3-1	6.60	7.60	8.60	9.60	10.60
C <sub>O</sub> I-3-2	8.42	9.42	10.42	11.42	12.42
C <sub>O</sub> T-1	9.13	10.13	11.13	12.13	13.13
C <sub>Ti</sub> T-1	12.40	10.40	8.40	6.40	4.40
C <sub>Ti</sub> E-1	14.00	12.00	10.00	8.00	6.00
C <sub>Ti</sub> E-2	12.73	10.73	8.73	6.73	4.73
C <sub>Ti</sub> E-3	12.97	10.97	8.97	6.97	4.97
C <sub>Ti</sub> E-4	13.29	11.29	9.29	7.29	5.29
C <sub>Ti</sub> E-5	13.34	11.34	9.34	7.34	5.34
C <sub>Ti</sub> E-6	14.61	12.61	10.61	8.61	6.61
C <sub>Ti</sub> F-1	15.79	13.79	11.79	9.79	7.79
C <sub>Ti</sub> F-2	14.78	12.78	10.78	8.78	6.78
C <sub>Ti</sub> F-3	13.90	11.90	9.90	7.90	5.90
C <sub>Ti</sub> F-4	13.69	11.69	9.69	7.69	5.69
C <sub>Ti</sub> I-2	13.69	11.69	9.69	7.69	5.69
C <sub>Ti</sub> I-3	14.85	12.85	10.85	8.85	6.85
C <sub>i</sub> I-1	-	-	-	-	8.97
C <sub>i</sub> I-2-1	-	-	-	-	9.31
C <sub>i</sub> I-2-2	-	-	-	-	8.68
C <sub>i</sub> I-3-1-1	-	-	-	-	8.75
C <sub>i</sub> I-3-1-1B	-	-	-	-	8.96
C <sub>i</sub> I-3-1-2	-	-	-	-	8.25
C <sub>i</sub> I-3-1-3	-	-	-	-	9.03
C <sub>i</sub> I-3-2-1	-	-	-	-	8.37
C <sub>i</sub> I-3-2-2	-	-	-	-	8.36
C <sub>i</sub> I-3-2-3	-	-	-	-	8.34

**Table S3.** Energy gap, LUMO (eV) and HOMO (eV) of C-doped  $(\text{TiO}_2)_{84}$  NPs described with PBE and PBEx density functionals.

NP	PBE			PBEx		
	Egap	LUMO	HOMO	Egap	LUMO	HOMO
<b>C<sub>O</sub> E-2</b>	0.47	-4.89	-5.36	1.14	-4.62	-5.76
<b>C<sub>O</sub> F-2</b>	0.27	-4.91	-5.18	0.88	-4.63	-5.51
<b>C<sub>O</sub> F-3</b>	0.74	-4.90	-5.65	1.53	-4.62	-6.15
<b>C<sub>O</sub> I-3-1</b>	0.94	-4.95	-5.89	1.70	-4.67	-6.37
<b>C<sub>O</sub> I-3-2</b>	0.67	-5.58	-6.25	1.13	-5.78	-6.92
<b>C<sub>O</sub> T-1</b>	0.68	-4.91	-5.59	0.83	-4.64	-5.47
<b>C<sub>Ti</sub> T-1</b>	2.07	-4.94	-7.01	3.16	-4.65	-7.81
<b>C<sub>Ti</sub> E-1</b>	2.20	-4.95	-7.15	-	-	-
<b>C<sub>Ti</sub> E-2</b>	2.94	-4.50	-7.44	3.58	-4.65	-8.23
<b>C<sub>Ti</sub> E-3</b>	2.53	-4.90	-7.42	3.61	-4.60	-8.21
<b>C<sub>Ti</sub> E-4</b>	2.52	-4.89	-7.40	-	-	-
<b>C<sub>Ti</sub> E-5</b>	2.54	-4.90	-7.44	-	-	-
<b>C<sub>Ti</sub> E-6</b>	2.32	-4.93	-7.26	-	-	-
<b>C<sub>Ti</sub> F-1</b>	2.46	-4.90	-7.36	-	-	-
<b>C<sub>Ti</sub> F-2</b>	2.45	-4.93	-7.38	-	-	-
<b>C<sub>Ti</sub> F-3</b>	2.49	-4.91	-7.40	-	-	-
<b>C<sub>Ti</sub> F-4</b>	2.37	-4.95	-7.32	-	-	-
<b>C<sub>Ti</sub> I-2</b>	2.51	-4.86	-7.37	-	-	-
<b>C<sub>Ti</sub> I-3</b>	2.56	-4.86	-7.42	-	-	-
<b>C<sub>i</sub> I-1</b>	1.42	-4.95	-6.37	-	-	-
<b>C<sub>i</sub> I-2-1</b>	1.38	-4.90	-6.29	-	-	-
<b>C<sub>i</sub> I-2-2</b>	0.49	-4.94	-5.42	-	-	-
<b>C<sub>i</sub> I-3-1-1</b>	0.04	-4.82	-4.86	-	-	-
<b>C<sub>i</sub> I-3-1-1b</b>	1.03	-4.90	-5.94	-	-	-
<b>C<sub>i</sub> I-3-1-2</b>	0.11	-4.97	-5.08	0.22	-4.58	-4.79
<b>C<sub>i</sub> I-3-1-3</b>	0.05	-4.81	-4.85	-	-	-
<b>C<sub>i</sub> I-3-2-1</b>	1.33	-4.91	-6.24	-	-	-
<b>C<sub>i</sub> I-3-2-2</b>	0.08	-4.81	-4.89	-	-	-
<b>C<sub>i</sub> I-3-2-3</b>	0.25	-4.91	-5.16	0.92	-4.63	-5.55