

Electronic Supplementary Material

Insights into the non-covalent interactions of hydrogen sulfide with fenchol and fenchone from a gas-phase rotational study

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Contents

Experimental setup.....	3
Theoretical calculations	4
Optimized geometries of the endo fenchol-H ₂ S complex conformers	5
Tables S1-S2. Experimental vs calculated constants for the different conformers of endo fenchol-H ₂ S	6
Differences (in %) between observed and calculated rotational constants	7
Measured rotational transitions for the observed conformer of endo fenchol-H ₂ S complex.....	8
MP2/6-311++G(d,p) cartesian coordinates of the calculated structures of endo fenchol-H ₂ S	9
Optimized geometries of the fenchone-H ₂ S complex conformers	10
Differences (in %) between observed and calculated rotational constants	12
MP2/6-311++G(d,p) cartesian coordinates of the calculated structures I _{FEN} , II _{FEN} and III _{FEN}	15
MP2/6-311++G(d,p) cartesian coordinates of the calculated structures IV _{FEN} , V _{FEN} and VI _{FEN}	16
Natural bonding orbitals representation for endo fenchol-H ₂ S	17
Comparison between the calculated structures of endo fenchol-H ₂ O and endo fenchol-H ₂ S complexes ..	19
SAPT analysis	21
References.....	23

Experimental setup

The rotational spectra were recorded using a pulsed Fabry–Pérot Fourier transform microwave (FP-FTMW) spectrometer coupled to supersonic expansion, available in the PhLAM laboratory in Lille, and operating in the frequency range 2–20 GHz^[1].

Fenchol (1,3,3-Trimethylbicyclo [2.2.1] heptan-2-ol, $C_{10}H_{18}O$, $\geq 99\%$, with endo-fenchol labeled as EF) and Fenchone (1,3,3-Trimethylbicyclo [2.2.1] heptan-2-o, $C_{10}H_{16}O$, $\geq 99\%$, labeled as FEN) were purchased from Sigma-Aldrich and used without any further purification. The H_2S (dihydrogen sulfide, $\geq 99.5\%$) was purchased in the form of a 15 bar gas cylinder from Air Liquide and used as it is without any further purification.

A premixture of H_2S and Neon as carrier gas was prepared in a 500 mL sampling cylinder (Swagelok). The mixture contained 50 mbar of H_2S and excess of Neon to reach 3.5 bar ($\approx 1.5\%$ dilution). The mixture was then seeded into an injector containing fenchol or fenchone and supersonically expanded into the vacuum chamber through a 1 mm pinhole. The injector was heated to 348 K to enhance the molecular signal-to-noise ratio. The rotational temperature of the molecules in the jet is in the order of a few Kelvins. Microwave power pulses of 2 μs duration were used to polarize the molecules in the optical cavity. The Free Induction Decay (FID) molecular signals were recorded by a 14-bit digitizer card operating at 120 MSa/s. Between 500 and 3000 FIDs were averaged, in high-resolution mode (frequency step between 0.96 and 3.68 kHz) to achieve a reasonable signal-to-noise ratio. A fast Fourier transform (FFT) was applied to the FID time-domain signals to obtain the frequency domain signals. The signals in the cavity are recorded as Doppler doublets and the experimental frequency considered is the central frequency obtained by the arithmetic averaging of the two Doppler components. Figures S1 and S2 show transitions recorded using the high-resolution mode for the EF- H_2S and FEN- H_2S complexes, respectively.

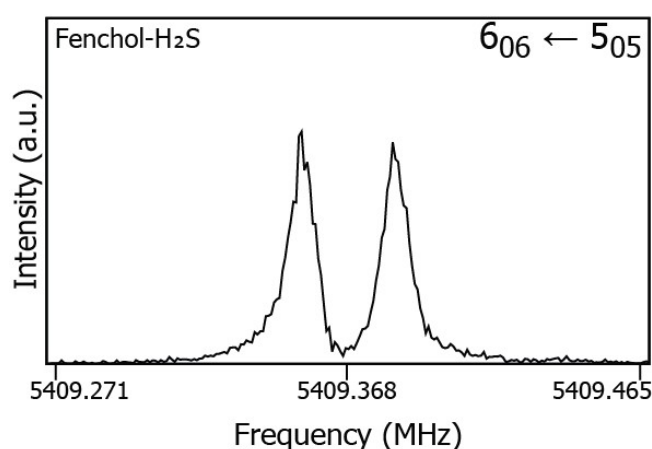


Figure S1. Observed transition at high resolution of the endo fenchol- H_2S complex. The signal is the result of 2800 averages. The transition appears as a doublet due to the coaxial alignment of the exciting radiation and molecular beam. The assignment is indicated in the top right corner.

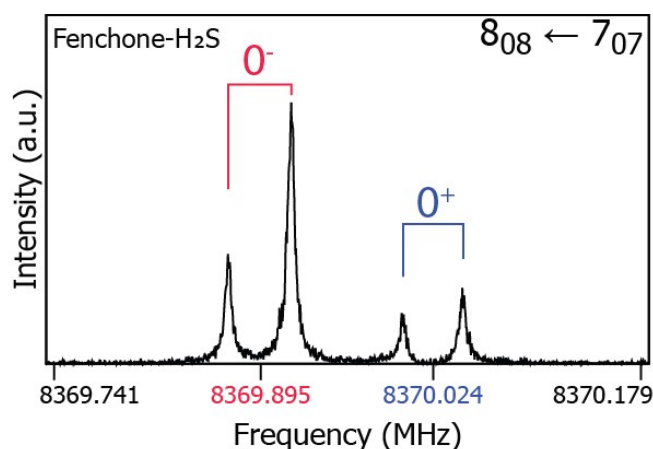


Figure S2. Observed transition at high resolution of the fenchone- H_2S complex. The signal is the result of 1000 averages. The transitions appear as doublets due to the coaxial alignment of the exciting radiation and molecular beam. The splitting due to a large amplitude motion of H_2S are labelled as 0^- and 0^+ . The assignment is indicated in the top right corner.

Theoretical calculations

Quantum chemical calculations facilitate the identification of the observed conformers and the assignment of the experimental spectrum. *Ab initio* and density-functional theory (DFT) calculations were used to predict the most stable conformers. Calculations were performed using the Gaussian 16^[2] software available on the high-performance computing cluster of PhLAM. The methods used were Møller-Plesset second-order perturbation theory (MP2)^[3], and the DFT ω B97X-D^[4] and B3LYP^[5,6] functionals. Pople split-valence triple-zeta basis set, augmented with diffuse and polarization functions on all atoms (6-311++G(d,p)) basis set^[7] was used with the MP2 and ω B97X-D methods, while Ahlrichs split-valence and triple zeta basis set with polarization (def2TZVP)^[8,9], with empirical dispersion correction (D3 version of Grimme's dispersion with Becke-Johnson damping), was used with the B3LYP method. First, geometry optimization of the conformers was done to obtain the rotational constants A, B and C. Second, harmonic frequencies calculations were performed to obtain the centrifugal distortion parameters Δ_J , Δ_{JK} , Δ_K , δ_J , and δ_K , as well as the energy ordering of the different conformers. Relative energy ordering is important to have an idea about what to predict in the spectrum and to have an idea about the relaxations taking place in the jet. Four stable conformers were predicted for the EF-H₂S complex, labeled using Roman numerals I_{EF}, II_{EF}, III_{EF} and IV_{EF}, following the energy order. Similarly, six stable conformers were predicted for the FEN-H₂S complex, and they were labeled as I_{FEN}, II_{FEN}, III_{FEN}, IV_{FEN}, V_{FEN} and VI_{FEN}, according to the increasing order of their relative energies. The calculations also gave the dipole moments values of each conformer. The calculated dipole moments were used to optimize the polarizing power, and for the identification of the observed conformers. The calculated constants for the different conformers, and at different levels of theory, are summarized in Tables S1 through S4 for the EF-H₂S complex and in the Tables S5 through S10 for the FEN-H₂S complex.

Furthermore, analysis of the natural bonding orbitals (NBOs) involved in the intermolecular charge transfer, which stabilizes the complexes with Sulfur, was carried out at the MP2/6-311++G(d,p) level of theory. The output of the MP2/6-311++G(d,p) optimization was also used to perform non-covalent interaction (NCI) calculations, using NCIPLOT^[10], that permits to plot the isosurface of the interactions occurring between the different atoms, thus helping to understand the different forces that stabilize the complex. In addition, Symmetry-adapted perturbation theory (SAPT),^[11] which calculates the total noncovalent energy, and decomposes it into electrostatic, exchange, induction and dispersion terms, was used to compare the contribution of the different interaction terms in the observed complexes. The SAPT energy was calculated at the SAPT₂₊₃ level using the PSI4 code.^[12,13]

Optimized geometries of the endo fenchol-H₂S complex conformers

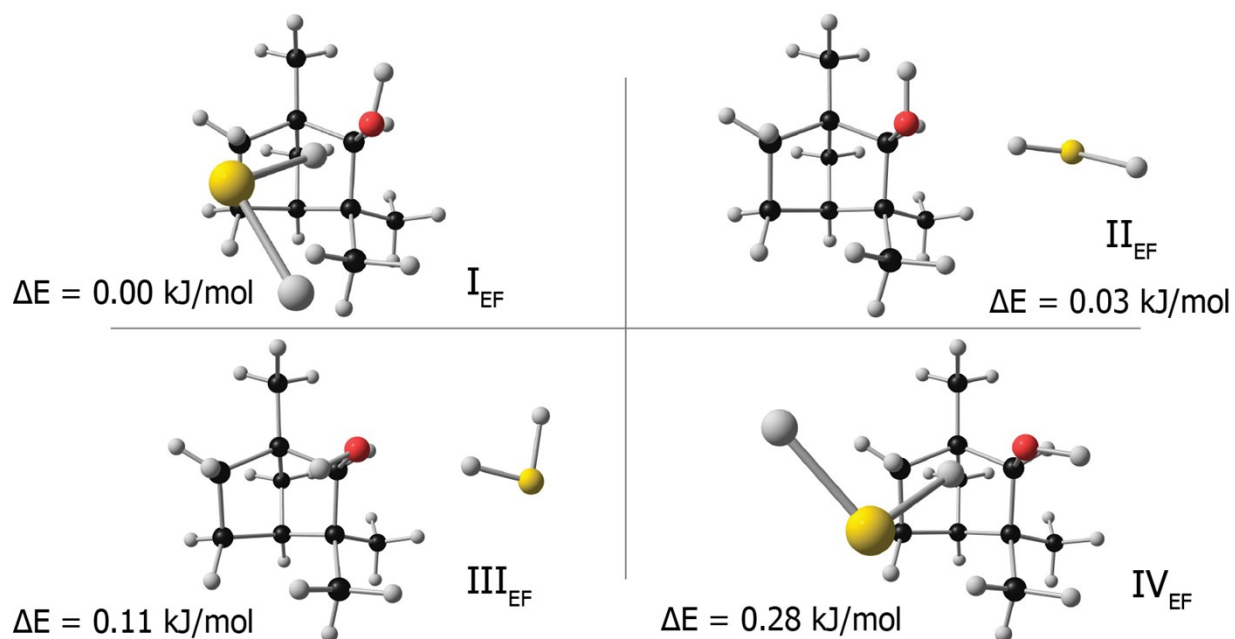


Figure S3. Calculated structures for the conformers of the fenchol-H₂S complex at the MP2/6-311++G(d,p) level of theory.

Tables S1-S2. Experimental vs calculated constants for the different conformers of endo fenchol-H₂S

Table S1

		EF-H ₂ S					
		I _{EF}			II _{EF}		
Constants	Experimental	MP2/ 6-311++G(d,p)	ωB97X-D/ 6-311++G(d,p)	B3LYP/ def2TZVP	MP2/ 6-311++G(d,p)	WB97XD/ 6-311++G(d,p)	B3LYP/ def2TZVP
A (MHz)	1161.2329(32)	992.1	996.6	994.4	1187.5	1201.7	1193.0
B (MHz)	494.21434(11)	564.6	550.3	551.2	510.2	494.7	496.8
C (MHz)	430.006772(86)	491.9	479.5	479.7	445.3	434.4	435.7
Δ _J (kHz)	0.20603(35)	0.0867	0.0911	0.1337	0.1213	0.0552	0.0728
Δ _{JK} (kHz)	0.8615(31)	0.3675	0.2997	0.4271	0.1910	0.1425	0.2224
Δ _K (kHz)	-0.4525	-0.4211	-0.3497	-0.5187	0.0580	-0.0600	-0.0828
δ _J (kHz)	0.03287(22)	0.0111	0.0110	0.0163	0.0208	0.0070	0.0101
δ _K (kHz)	0.231(12)	0.1317	0.1346	0.1797	0.1466	0.0444	0.0773
μ _a (D)	observed	2.1	2.5	2.1	1.6	2.3	2.0
μ _b (D)	Not observed	0.07	0.01	0.1	0.8	0.5	0.6
μ _c (D)	Not observed	0.9	1.0	1.0	0.2	0.1	0.1
ΔE ^a		0.14	0.96	1.47	0.00	0.61	0.49
ΔE _{ZPE} ^b		0.00	0.54	1.38	0.03	1.22	0.86
ΔG ₃₄₈ ^c		1.37	1.42	1.71	0.00	3.37	2.23
No. of lines	56						
RMS (kHz)	1.44						

^a Electronic relative energies corresponding to the global minima corresponding to the method/basis set.

^b Electronic relative energies including ZPE correction corresponding to the global minima corresponding to the method/basis set.

^c Gibbs energies corresponding to the global minima corresponding to the method/basis set.

Table S2

		EF-H ₂ S					
		III _{EF}			IV _{EF}		
Constants	Experimental	MP2/ 6-311++G(d,p)	ωB97X-D/ 6-311++G(d,p)	B3LYP/ def2TZVP	MP2/ 6-311++G(d,p)	WB97XD/ 6-311++G(d,p)	B3LYP/ def2TZVP
A (MHz)	1161.2329(32)	1182.5	1187.5	1192.7	993.9	995.7	995.5
B (MHz)	494.21434(11)	505.8	498.4	497.4	566.5	553.8	553.9
C (MHz)	430.006772(86)	439.5	434.8	434.9	492.6	481.8	481.3
Δ _J (kHz)	0.20603(35)	0.0553	0.0546	0.0605	0.0761	0.0930	0.1143
Δ _{JK} (kHz)	0.8615(31)	0.7496	0.5767	0.4782	0.5718	0.4246	0.4798
Δ _K (kHz)	-0.4525	-0.4525	-0.3646	-0.2790	-0.6174	-0.4852	-0.5599
δ _J (kHz)	0.03287(22)	0.0076	0.0071	0.0074	0.0095	0.0107	0.0135
δ _K (kHz)	0.231(12)	0.2892	0.1999	0.1653	0.2354	0.1898	0.1942
μ _a (D)	observed	2.7	3.0	2.5	2.0	2.3	1.9
μ _b (D)	Not observed	0.02	0.2	0.1	0.03	0.04	0.1
μ _c (D)	Not observed	0.1	0.1	0.1	0.7	0.9	0.9
ΔE ^a		0.29	0.00	0.00	0.53	1.64	1.48
ΔE _{ZPE} ^b		0.11	0.00	0.00	0.28	1.22	1.39
ΔG ₃₄₈ ^c		0.40	0.00	0.00	1.02	1.27	1.82
No. of lines	56						
RMS (kHz)	1.44						

^a Electronic relative energies corresponding to the global minima corresponding to the method/basis set.

^b Electronic relative energies including ZPE correction corresponding to the global minima corresponding to the method/basis set.

^c Gibbs free energies (348 K) corresponding to the method/basis set.

Differences (in %) between observed and calculated rotational constants

Table S3

Constants	Experimental	III _{EF}		
		MP2/6-311++G(d,p) ^a	ωB97X-D/6-311++G(d,p) ^a	B3LYP/def2TZVP ^a
A (MHz)	1161.2329(32)	1.8	2.3	2.7
B (MHz)	494.21434(11)	2.3	0.8	0.6
C (MHz)	430.006772(86)	2.2	1.1	1.1
Δ _J (kHz)	0.20603(35)	73.2	73.5	70.6
Δ _{JK} (kHz)	0.8615(31)	13.0	33.1	44.5
Δ _K (kHz)	-0.4525	-	-	-
δ _J (kHz)	0.03287(22)	97.7	97.8	97.8
δ _K (kHz)	0.231(12)	25.2	13.5	28.4

^a The % has been calculated using the formula: $\Delta = \frac{\text{observed} - \text{calculated}}{\text{observed}}$

Measured rotational transitions for the observed conformer of endo fenchol-H₂S complex

Table S4

J	K _a	K _c	J'	K' _a	K' _c	ν_{EXP} (MHz)	$\Delta\nu_{\text{EXP}}$ (MHz) *
3	0	3	2	0	2	2755.1185	0.0010
4	1	4	3	1	3	3560.1331	-0.0005
4	0	4	3	0	3	3653.8855	0.0004
5	2	3	4	2	2	4693.7287	-0.0012
5	1	4	4	1	3	4760.1133	-0.0021
6	1	6	5	1	5	5322.1969	0.0006
6	0	6	5	0	5	5409.3682	0.0000
6	2	5	5	2	4	5525.3124	0.0007
6	1	5	5	1	4	5696.9547	0.0041
7	1	7	6	1	6	6197.5214	-0.0005
7	0	7	6	0	6	6270.2349	0.0011
7	2	6	6	2	5	6435.1684	0.0000
7	4	4	6	4	3	6490.7416	-0.0002
7	4	3	6	4	2	6491.5472	-0.0011
7	3	5	6	3	4	6494.9286	0.0000
7	3	4	6	3	3	6517.9149	0.0006
7	1	6	6	1	5	6623.8815	0.0007
7	2	5	6	2	4	6633.3928	-0.0002
8	1	8	7	1	7	7069.3239	-0.0001
8	0	8	7	0	7	7125.4661	0.0002
8	2	7	7	2	6	7340.1844	0.0029
8	4	5	7	4	4	7423.2486	0.0018
8	3	6	7	3	5	7425.0992	-0.0016
8	4	4	7	4	3	7425.4432	-0.0010
8	3	5	7	3	4	7469.5995	-0.0039
8	1	7	7	1	6	7538.2905	-0.0018
8	2	6	7	2	5	7605.8638	-0.0007
9	1	9	8	1	8	7938.0568	0.0008
9	0	9	8	0	8	7978.8205	-0.0012
9	3	7	8	3	6	8353.8626	0.0005
9	4	6	8	4	5	8357.3959	0.0006
9	2	7	8	2	6	8573.2939	-0.0010
10	1	10	9	1	9	8804.2585	-0.0015
10	0	10	9	0	9	8832.505	0.0017
10	3	8	9	3	7	9280.1342	0.0002
10	4	7	9	4	6	9292.9756	0.0010
10	4	6	9	4	5	9304.0275	-0.0016
10	1	9	9	1	8	9321.5323	-0.0009
10	3	7	9	3	6	9404.2328	-0.0017
10	2	8	9	2	7	9531.9971	-0.0005
11	1	11	10	1	10	9668.4842	-0.0005
11	0	11	10	0	10	9687.3703	-0.0021
11	2	10	10	2	9	10022.8814	-0.0015
11	1	10	10	1	9	10190.1427	-0.0012
11	3	9	10	3	8	10202.8627	0.0000
11	4	8	10	4	7	10229.5622	0.0005
11	4	7	10	4	6	10251.0814	0.0012
11	3	8	10	3	7	10386.0615	-0.0015
11	2	9	10	2	8	10479.3008	0.0021
12	1	12	11	1	11	10531.2279	0.0023
12	0	12	11	0	11	10543.5221	-0.0008
12	2	11	11	2	10	10906.2358	0.0005
12	1	11	11	1	10	11046.9143	-0.0014
12	3	10	11	3	9	11121.0994	-0.0003
12	4	8	11	4	7	11205.3872	-0.0022
13	2	12	12	2	11	11784.6293	0.0021

* $\Delta\nu = \nu_{\text{CALC}} - \nu_{\text{EXP}}$

MP2/6-311++G(d,p) cartesian coordinates of the calculated structures of endo fenchol-H₂S

Table S5

Element	I _{EF}			II _{EF}			III _{EF}			IV _{EF}		
	E= 864.705888 Hartree			E= 864.705943 Hartree			E= 864.705834 Hartree			E= 864.705739 Hartree		
	X	Y	Z	X	Y	Z	X	Y	Z	X	Y	Z
C	1.021351	-1.222370	-0.163763	1.445694	0.981386	0.222467	1.418933	1.009223	0.218733	-0.938743	1.255171	-0.178520
C	0.020031	-1.321922	1.008062	2.192642	0.524655	-1.051134	2.199744	0.559056	-1.038594	0.091140	1.310837	0.973125
C	0.435609	-0.161940	1.956490	2.390302	-1.003587	-0.840455	2.432058	-0.963735	-0.818757	-0.361097	0.179864	1.940264
C	1.504557	0.577436	1.132650	1.572748	-1.273406	0.434200	1.610385	-1.245272	0.451109	-1.506740	-0.491625	1.161065
C	2.258374	-0.624333	0.530258	1.949453	-0.031973	1.264873	1.946749	0.016433	1.269440	-2.191192	0.751868	0.561345
C	0.867265	1.230411	-0.126194	0.057269	-1.046882	0.169133	0.091195	-1.063134	0.178560	-0.959557	-1.203352	-0.107010
C	0.630652	-0.015841	-1.039033	-0.019362	0.514272	0.117949	-0.032668	0.502297	0.114549	-0.637654	0.020213	-1.038832
C	-0.415792	2.008718	0.172665	-0.443040	-1.717723	-1.113337	-0.390846	-1.751851	-1.099961	0.260580	-2.084970	0.164102
C	1.864018	2.188986	-0.789204	-0.765876	-1.586453	1.345830	-0.725720	-1.615226	1.353192	-2.053162	-2.074178	-0.737717
C	1.209096	-2.511609	-0.942685	1.597857	2.456228	0.549713	1.520931	2.490082	0.531063	-1.059446	2.539085	-0.978008
O	-0.706645	-0.048537	-1.548274	-0.704268	0.967041	-1.054000	-0.777428	1.032325	-0.979992	0.699944	0.089320	-1.542532
H	-1.012113	-1.227286	0.661659	1.613282	0.753064	-1.951057	1.662091	0.778988	-1.970184	1.111659	1.169221	0.612533
H	0.118596	-2.296727	1.499935	3.153286	1.047159	-1.127372	3.147740	1.105547	-1.098112	0.041799	2.293044	1.457841
H	-0.404083	0.469189	2.255375	2.072601	-1.602256	-1.696903	2.142812	-1.577530	-1.675036	0.447388	-0.504232	2.207817
H	0.893354	-0.554535	2.871662	3.444619	-1.234895	-0.650404	3.490391	-1.164009	-0.617734	-0.756524	0.600025	2.871981
H	2.103717	1.286729	1.714997	1.774189	-2.247592	0.893987	1.840549	-2.207538	0.922416	-2.135399	-1.148644	1.773397
H	2.673356	-1.292379	1.294922	3.028798	0.050736	1.442606	3.022390	0.129553	1.452784	-2.531308	1.459199	1.327500
H	3.052711	-0.353613	-0.174696	1.423471	0.047839	2.222802	1.413272	0.092128	2.223217	-3.025574	0.530680	-0.113484
H	1.325525	0.028724	-1.893371	-0.562396	0.880594	1.005812	-0.583684	0.856530	0.995251	-1.331483	0.018994	-1.894869
H	-1.223072	1.368939	0.530729	-0.042787	-1.252286	-2.016129	0.057144	-1.342438	-2.011483	1.133710	-1.510758	0.477358
H	-0.210396	2.768863	0.936498	-0.156478	-2.776526	-1.104409	-0.138894	-2.818400	-1.063680	0.022952	-2.813930	0.948355
H	-0.769640	2.516059	-0.731066	-1.534996	-1.664823	-1.168532	-1.478064	-1.663339	-1.193845	0.536334	-2.648594	-0.736072
H	2.019840	3.072798	-0.159302	-0.712860	-2.681604	1.371854	-0.652015	-2.709064	1.388966	-2.277218	-2.933482	-0.094582
H	1.477956	2.524975	-1.759205	-1.817741	-1.299824	1.241904	-1.781049	-1.341549	1.248647	-1.723538	-2.458646	-1.711565
H	2.836520	1.714174	-0.956607	-0.399925	-1.206882	2.305863	-0.368570	-1.220739	2.310408	-2.981337	-1.515332	-0.894871
H	0.261441	-2.848200	-1.382949	1.215722	3.087545	-0.263781	1.069577	3.085655	-0.270209	-0.111927	2.770241	-1.476812
H	1.565541	-3.313432	-0.286458	2.653110	2.715697	0.690049	2.568155	2.795036	0.639535	-1.317822	3.379248	-0.323391
H	1.937897	-2.381408	-1.752357	1.054891	2.714342	1.467054	0.994773	2.725903	1.463665	-1.836801	2.449971	-1.746571
S	-3.746500	-0.150466	0.130995	-3.898273	0.449671	0.111308	-3.957842	0.346430	0.001828	3.742430	-0.112300	0.106759
H	-0.723548	-0.718820	-2.239893	-0.681036	1.930569	-1.026885	-0.332489	0.772126	-1.793089	0.778165	-0.572452	-2.237431
H	-2.648551	-0.000215	-0.621609	-2.757916	0.586534	-0.581209	-2.804409	0.699756	-0.584223	2.634393	0.105827	-0.613822
H	-4.048856	1.146695	0.046518	-4.627439	0.322449	-0.999615	-4.149389	1.574671	0.487391	3.979315	1.194387	0.239914

Optimized geometries of the fenchone-H₂S complex conformers

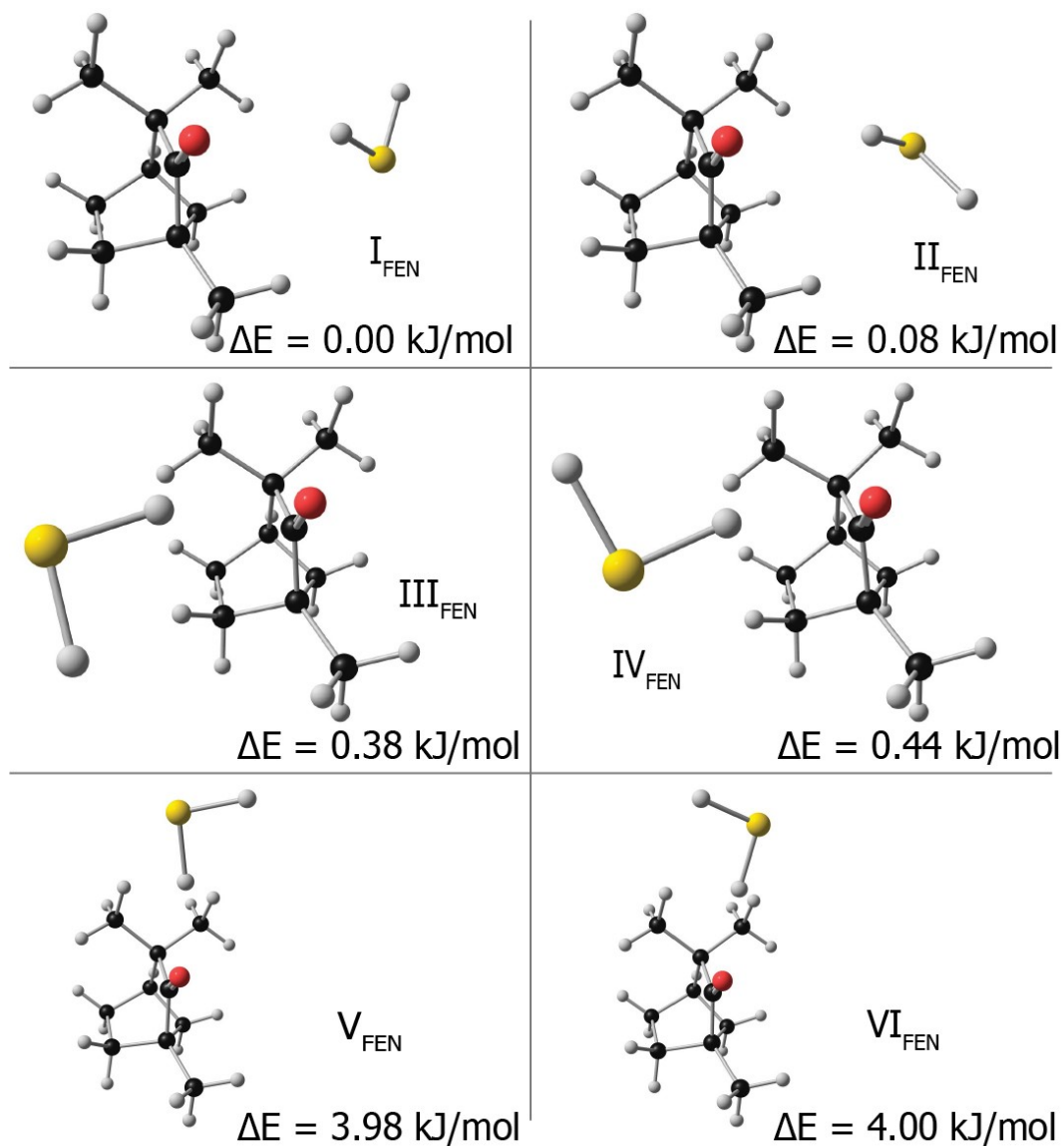


Figure S4. Calculated conformers of the FEN-H₂S complex at the MP2/6-311++G(d,p) level of theory.

Tables S6-S8 Experimental vs calculated constants for the different conformers of fenchone-H₂S

Table S6

FEN-H ₂ S								
			I _{FEN}			II _{FEN}		
Constants	Experimental		MP2/ 6-311++G(d,p)	ωB97X-D/ 6-311++G(d,p)	B3LYP/ def2TZVP	MP2/ 6-11++G(d,p)	WB97XD/ 6-311++G(d,p)	B3LYP/ def2TZVP
	0+	0-						
A (MHz)	983.41497(21)	983.40341(13)	1092.9	1103.7	1099.5	1091.5	1101.6	1095.5
B (MHz)	582.232474(64)	582.241521(34)	610.7	596.0	595.5	610.4	597.7	603.9
C (MHz)	511.016389(44)	511.007220(30)	507.2	493.8	493.5	506.6	495.5	500.3
Δ _J (kHz)	0.18499(14)		0.1136	0.0957	0.1604	0.1288	0.1354	0.1261
Δ _{JK} (kHz)	1.4868(14)		0.0192	0.0656	-0.0570	0.1082	0.0969	0.0947
Δ _K (kHz)	-1.6012(59)		-0.0363	-0.0613	0.0258	-0.1155	-0.1040	-0.1195
δ _J (kHz)	0.015727(80)		0.0146	0.0127	0.0190	0.0157	0.0165	0.0162
δ _K (kHz)	0.7361(24)		0.0758	0.0793	0.0799	0.1124	0.1330	0.1306
μ _a (D)	observed		2.0	2.9	2.6	1.9	2.5	2.1
μ _b (D)	observed		0.5	0.5	0.4	0.9	1.0	0.9
μ _c (D)	observed		1.8	1.8	1.8	1.4	1.5	1.5
ΔE ^a			0.00	0.92	0.79	0.60	1.09	0.45
ΔE _{ZPE} ^b			0.00	0.69	0.66	0.08	0.49	0.54
ΔG ₃₄₈ ^c			3.22	3.97	1.68	1.64	2.11	2.91
No. of lines	70	99						
RMS (kHz)	1.72							

^a Electronic relative energies corresponding to the global minima corresponding to the method/basis set.

^b Electronic relative energies including ZPE correction corresponding to the global minima corresponding to the method/basis set.

^c Gibbs free energies (348 K) corresponding to the method/basis set.

Table S7

FEN-H ₂ S								
			III _{FEN}			IV _{FEN}		
Constants	Experimental		MP2/ 6-311++G(d,p)	ωB97X-D/ 6-311++G(d,p)	B3LYP/ def2TZVP	MP2/ 6-311++G(d,p)	WB97XD/ 6-311++G(d,p)	B3LYP/ def2TZVP
	0+	0-						
A (MHz)	983.41497(21)	983.40341(13)	984.2	982.9	974.8	985.1	987.8	978.0
B (MHz)	582.232474(64)	582.241521(34)	597.2	594.6	602.1	599.0	592.7	595.5
C (MHz)	511.016389(44)	511.007220(30)	525.1	521.5	528.5	526.9	520.0	522.8
Δ _J (kHz)	0.18499(14)		0.1214	0.1123	0.0995	0.1091	0.0959	0.1017
Δ _{JK} (kHz)	1.4868(14)		1.4238	0.9362	0.7129	0.6980	1.1162	1.2980
Δ _K (kHz)	-1.6012(59)		-1.3961	-1.0025	-0.7733	-0.7291	-1.1260	-1.3375
δ _J (kHz)	0.015727(80)		0.0136	0.0099	0.0095	0.0101	0.0091	0.0099
δ _K (kHz)	0.7361(24)		0.6329	0.4385	0.3172	0.3034	0.4632	0.5222
μ _a (D)	observed		2.7	3.0	2.5	2.5	3.3	2.9
μ _b (D)	observed		1.1	0.9	0.8	0.4	0.4	0.4
μ _c (D)	observed		1.3	1.3	1.4	1.5	1.5	1.6
ΔE ^a			0.83	0.86	0.22	0.52	0.71	0.39
ΔE _{ZPE} ^b			0.38	0.00	0.05	0.44	0.47	0.0
ΔG ₃₄₈ ^c			0.00	0.00	1.17	2.00	1.65	0.0
No. of lines	70	99						
RMS (kHz)	1.72							

^a Electronic relative energies corresponding to the global minima corresponding to the method/basis set.

^b Electronic relative energies including ZPE correction corresponding to the global minima corresponding to the method/basis set.

^c Gibbs free energies (348 K) corresponding to the method/basis set.

Table S8

FEN-H ₂ S								
Constants	Experimental		V _{FEN}			V _{I_{FEN}}		
			MP2/ 6-311++G(d,p)	ωB97X-D/ 6-311++G(d,p)	B3LYP/ def2TZVP	MP2/ 6-311++G(d,p)	WB97XD/ 6-311++G(d,p)	B3LYP/ def2TZVP
	0+	0-						
A (MHz)	983.41497(21)	983.40341(13)	1261.9	1266.7	1269.5	1275.8	1265.7	1270.9
B (MHz)	582.232474(64)	582.241521(34)	473.1	476.4	474.4	474.4	476.8	475.0
C (MHz)	511.016389(44)	511.007220(30)	417.4	417.7	416.6	415.0	418.0	416.6
Δ _J (kHz)	0.18499(14)		0.0445	0.0397	0.0416	0.0464	0.0427	0.0421
Δ _{JK} (kHz)	1.4868(14)		0.3936	0.2611	0.2203	0.8164	0.2743	0.2319
Δ _K (kHz)	-1.6012(59)		-0.0736	-0.1377	-0.1120	-0.7087	-0.1563	-0.1281
δ _J (kHz)	0.015727(80)		0.0047	0.0043	0.0047	0.0037	0.0040	0.0048
δ _K (kHz)	0.7361(24)		-0.0249	0.0448	0.0441	0.1435	0.0652	0.0503
μ _a (D)	Observed		3.3	4.0	3.7	3.4	4.0	3.6
μ _b (D)	Observed		1.2	1.7	1.7	1.6	1.3	1.3
μ _c (D)	Observed		1.1	0.5	0.3	0.4	1.2	1.1
ΔE ^a			3.91	0.06	0.01	3.89	0.00	0.00
ΔE _{ZPE} ^b			3.98	0.22	0.42	4.00	0.12	0.36
ΔG ₃₄₈ ^c			4.89	2.94	2.51	4.27	2.84	2.25
No. of lines	70	99						
RMS (kHz)		1.72						

^a Electronic relative energies corresponding to the global minima corresponding to the method/basis set.

^b Electronic relative energies including ZPE correction corresponding to the global minima corresponding to the method/basis set.

^c Gibbs free energies (348 K) corresponding to the method/basis set.

Differences (in %) between observed and calculated rotational constants

Table S9

Constants	Experimental		IV _{FEN}		
			MP2/6-311++G(d,p) ^a	ωB97X-D/6-311++G(d,p) ^a	B3LYP/def2TZVP ^a
A (MHz)	983.41497(21)	983.40341(13)	0.2	0.4	0.6
B (MHz)	582.232474(64)	582.241521(34)	2.9	1.8	2.3
C (MHz)	511.016389(44)	511.007220(30)	3.1	1.8	2.3
Δ _J (kHz)	0.18499(14)		41.1	48.2	45.0
Δ _{JK} (kHz)	1.4868(14)		53.1	24.9	12.7
Δ _K (kHz)	-1.6012(59)		54.5	29.7	16.5
δ _J (kHz)	0.015727(80)		35.6	41.9	37.4
δ _K (kHz)	0.7361(24)		58.8	37.1	29.1

^a The % has been calculated using the formula: $\Delta = \frac{\text{observed} - \text{calculated}}{\text{observed}}$

Measured rotational transitions for the observed conformer of fenchone-H₂S complex

Table S10

0 state																	
J	Ka	Kc	J'	Ka'	Kc'	obs. (MHz)	obs - calc (MHz) *		J	Ka	Kc	J'	Ka'	Kc'	obs. (MHz)	obs - calc (MHz) *	
3	0	3	2	0	2	3245.871	-0.0002		9	1	9	8	1	8	9377.271	-0.0002	
4	1	4	3	1	3	4214.963	0.0007		9	0	9	8	0	8	9387.511	0.0000	
4	0	4	3	0	3	4293.025	-0.0002		6	4	2	5	3	2	9611.075	0.0000	
4	1	4	3	0	3	4475.436	0.0012		6	4	3	5	3	3	9627.361	0.0002	
4	1	3	3	1	2	4495.926	-0.0018		7	3	4	6	2	4	9654.054	-0.0005	
5	1	5	4	1	4	5256.432	-0.0002		9	2	8	8	2	7	9694.105	-0.0006	
5	0	5	4	0	4	5321.597	-0.0016		9	1	8	8	1	7	9812.118	0.0002	
5	3	3	4	3	2	5489.483	0.0013		9	4	6	8	4	5	9909.543	0.0002	
5	3	2	4	3	1	5503.023	0.0004		9	4	5	8	4	4	9939.417	0.0002	
5	2	3	4	2	2	5591.082	0.0006		7	3	5	6	2	5	10013.43	-0.0016	
5	1	4	4	1	3	5599.03	0.0008		9	2	7	8	2	6	10115.37	-0.0001	
6	0	6	5	1	5	6222.178	-0.0002		8	2	6	7	1	6	10278.75	-0.0028	
6	1	6	5	1	5	6292.456	0.0004		10	1	10	9	1	9	10401.15	0.0001	
6	0	6	5	0	5	6339.421	-0.0009		10	0	10	9	0	9	10406.73	-0.0003	
6	1	6	5	0	5	6409.699	0.0002		7	4	4	6	3	3	10677.43	-0.0010	
5	2	4	4	1	3	6419.479	0.0002		7	4	3	6	3	3	10684.2	0.0004	
5	1	4	4	0	4	6489.12	-0.0026		8	3	5	7	2	5	10689.12	-0.0003	
6	2	5	5	2	4	6520.677	0.0001		7	4	4	6	3	4	10730.28	-0.0001	
6	4	3	5	4	2	6588.052	0.0001		10	2	9	9	2	8	10736.35	0.0000	
6	4	2	5	4	1	6589.546	-0.0008		7	4	3	6	3	4	10737.04	0.0002	
6	3	4	5	3	3	6590.567	0.0005		10	1	9	9	1	8	10822.49	-0.0002	
6	3	3	5	3	2	6625.255	0.0003		10	3	8	9	3	7	10954.67	-0.0003	
6	1	5	5	1	4	6685.105	0.0002		10	4	7	9	4	6	11017.55	-0.0003	
5	2	3	4	1	3	6686.1	0.0007		10	4	6	9	4	5	11077.93	-0.0001	
6	2	4	5	2	3	6739.093	0.0000		10	2	8	9	2	7	11200.81	0.0000	
7	1	6	6	2	5	7093.495	0.0016		8	3	6	7	2	6	11211.43	0.0004	
5	2	4	4	1	4	7127.163	0.0004		10	3	7	9	3	6	11240.98	0.0002	
7	0	7	6	1	6	7283.909	0.0005		11	1	11	10	1	10	11424.05	0.0008	
7	1	7	6	1	6	7323.882	0.0000		11	0	11	10	0	10	11427.01	-0.0003	
7	0	7	6	0	6	7354.186	-0.0002		9	3	6	8	2	6	11760.04	-0.0006	
7	1	7	6	0	6	7394.159	-0.0001		11	2	10	10	2	9	11772.3	0.0007	
4	4	0	3	3	0	7431.381	-0.0009		8	4	5	7	3	5	11841.6	0.0006	
4	4	1	3	3	1	7432.005	-0.0005		11	3	9	10	3	8	12026.94	0.0000	
7	2	6	6	2	5	7586.683	-0.0001		11	4	8	10	4	7	12123.01	-0.0004	
5	3	2	4	2	2	7596.487	0.0007		11	4	7	10	4	6	12232.09	0.0001	
7	3	5	6	3	4	7689.677	-0.0003		11	2	9	10	2	8	12260.42	-0.0003	
7	4	4	6	4	3	7693.489	0.0004		11	3	8	10	3	7	12393.38	0.0000	
7	4	3	6	4	2	7698.377	0.0007		12	2	11	11	2	10	12803.32	0.0000	
5	3	3	4	2	3	7700.733	0.0011		12	3	10	11	3	9	13089.87	-0.0001	
7	1	6	6	1	5	7749.515	-0.0003		12	4	9	11	4	8	13223.73	-0.0004	
7	3	4	6	3	3	7762.489	-0.0006		12	2	10	11	2	9	13294.34	0.0001	
6	2	4	5	1	4	7826.162	-0.0013		12	4	8	11	4	7	13401.02	0.0005	
7	2	5	6	2	4	7880.094	-0.0002		12	3	9	11	3	8	13527.43	0.0006	
8	1	7	7	2	6	8297.39	0.0006										
8	0	8	7	1	7	8329.921	-0.0005										
8	1	8	7	1	7	8351.802	0.0002										
8	0	8	7	0	7	8369.895	-0.0001										
8	1	8	7	0	7	8391.775	0.0000										
6	3	3	5	2	3	8630.659	0.0002										
8	2	7	7	2	6	8644.449	-0.0008										
8	1	7	7	1	6	8790.579	0.0000										
8	4	5	7	4	4	8800.993	0.0004										
8	4	4	7	4	3	8814.033	0.0001										
6	3	4	5	2	4	8844.436	0.0028										
8	2	6	7	2	5	9007.115	0.0000										
7	2	5	6	1	5	9021.154	0.0011										

* $\Delta v = v_{\text{CALC}} - v_{\text{EXP}}$

Table S11

0 ⁺ state																
J	K _a	K _c	J'	K _a '	K _c '	obs (MHz)	obs -calc (MHz) *	J	K _a	K _c	J'	K _a '	K _c '	obs (MHz)	obs -calc (MHz) *	
5	1	5	4	1	4	5256.495	0.0007	11	0	11	10	0	10	11427.19	-0.0006	
5	0	5	4	0	4	5321.666	0.0008	9	3	6	8	2	6	11760.05	0.0012	
5	1	4	4	1	3	5599.012	-0.0005	11	2	10	10	2	9	11772.42	0.0011	
6	0	6	5	1	5	6222.224	0.0011	11	3	9	10	3	8	12026.97	0.0001	
6	1	6	5	1	5	6292.536	-0.0003	11	4	8	10	4	7	12122.97	0.0005	
6	0	6	5	0	5	6339.512	0.0000	11	4	7	10	4	6	12231.95	-0.0026	
6	1	6	5	0	5	6409.826	0.0005	11	2	9	10	2	8	12260.46	0.0019	
6	2	5	5	2	4	6520.698	-0.0007	11	3	8	10	3	7	12393.26	0.0003	
6	2	4	5	2	3	6739.029	0.0001	12	3	10	11	3	9	13089.93	0.0006	
7	1	6	6	2	5	7093.356	-0.0004	12	4	9	11	4	8	13223.7	-0.0017	
7	0	7	6	1	6	7283.984	-0.0001	12	2	10	11	2	9	13294.43	-0.0008	
7	1	7	6	1	6	7323.982	-0.0002	12	3	9	11	3	8	13527.33	-0.0001	
7	0	7	6	0	6	7354.297	-0.0001									
4	4	0	3	3	0	7431.471	0.0016									
4	4	1	3	3	1	7432.089	-0.0023									
7	2	6	6	2	5	7586.718	-0.0010									
5	3	2	4	2	2	7596.565	-0.0014									
7	4	4	6	4	3	7693.464	0.0000									
7	4	3	6	4	2	7698.346	-0.0006									
5	3	3	4	2	3	7700.766	-0.0002									
7	3	4	6	3	3	7762.424	0.0001									
6	2	4	5	1	4	7826.115	-0.0006									
7	2	5	6	2	4	7880.029	0.0009									
8	0	8	7	1	7	8330.026	0.0004									
8	1	8	7	1	7	8351.922	0.0000									
8	0	8	7	0	7	8370.024	0.0004									
6	3	3	5	2	3	8630.75	-0.0026									
8	2	7	7	2	6	8644.5	-0.0023									
8	1	7	7	1	6	8790.642	-0.0004									
8	4	5	7	4	4	8800.961	-0.0011									
8	4	4	7	4	3	8813.989	0.0003									
6	3	4	5	2	4	8844.442	-0.0009									
7	2	5	6	1	5	9021.04	0.0012									
9	1	9	8	1	8	9377.411	0.0000									
9	0	9	8	0	8	9387.657	-0.0001									
6	4	2	5	3	2	9611.162	-0.0004									
6	4	3	5	3	3	9627.435	0.0000									
7	3	4	6	2	4	9654.15	0.0018									
9	2	8	8	2	7	9694.179	0.0000									
9	1	8	8	1	7	9812.216	-0.0002									
9	4	6	8	4	5	9909.506	-0.0013									
7	3	5	6	2	5	10013.41	-0.0007									
9	2	7	8	2	6	10115.33	0.0001									
8	2	6	7	1	6	10278.55	0.0008									
10	1	10	9	1	9	10401.31	-0.0006									
10	0	10	9	0	9	10406.89	0.0000									
7	4	4	6	3	3	10677.54	-0.0004									
7	4	3	6	3	3	10684.3	0.0010									
8	3	5	7	2	5	10689.19	0.0019									
7	4	4	6	3	4	10730.35	0.0008									
10	2	9	9	2	8	10736.44	0.0008									
10	1	9	9	1	8	10822.62	0.0009									
10	3	8	9	3	7	10954.69	0.0021									
10	2	9	9	1	8	10965.63	-0.0001									
10	4	7	9	4	6	11017.51	0.0000									
10	4	6	9	4	5	11077.83	-0.0002									
10	2	8	9	2	7	11200.81	0.0003									
11	1	11	10	1	10	11424.23	0.0009									

* $\Delta v = v_{\text{CALC}} - v_{\text{EXP}}$

MP2/6-311++G(d,p) cartesian coordinates of the calculated structures I_{FEN}, II_{FEN} and III_{FEN}

Table S12

Element	I _{FEN}			II _{FEN}			III _{FEN}		
	E= 863.519314 Hartree			E= 863.519087 Hartree			E= 863.518998 Hartree		
	X	Y	Z	X	Y	Z	X	Y	Z
C	-0.086208	0.083348	0.762925	0.079483	-0.041715	0.758447	0.188927	-0.06139	-0.81717
C	-0.823662	1.177205	-0.024888	0.747625	-1.191090	-0.011029	1.004145	-1.14024	-0.09122
C	-0.367086	-1.244458	0.078859	0.447988	1.256352	0.059451	0.618367	1.288931	-0.26996
C	-1.498756	0.329139	-1.132524	1.489986	-0.402004	-1.118890	1.848314	-0.2643	0.869685
C	-2.637855	-0.518888	-0.536735	2.674161	0.378108	-0.517352	0.954436	0.333386	1.973004
C	-1.873451	-1.513531	0.383148	1.964782	1.434656	0.377926	0.037422	1.306734	1.17828
C	-0.443793	-0.769374	-1.386422	0.512351	0.759407	-1.399079	2.10485	0.997992	0.017944
C	0.191718	2.163536	-0.621313	-0.328227	-2.110221	-0.609759	1.892405	-1.87911	-1.10418
C	-1.769104	1.947391	0.901459	1.628336	-2.017223	0.930575	0.069627	-2.15332	0.575715
C	0.569326	-2.373164	0.450611	-0.425546	2.442113	0.405315	0.238138	2.473303	-1.13059
O	0.587118	0.266358	1.763461	-0.607814	-0.167642	1.758438	-0.66061	-0.26948	-1.66777
H	0.861216	1.686672	-1.341994	-0.954148	-1.596791	-1.344791	2.616598	-1.21597	-1.58627
H	0.800448	2.598631	0.179024	-0.973854	-2.497854	0.185871	1.269781	-2.32993	-1.88376
H	-0.344821	2.974405	-1.127876	0.155805	-2.960132	-1.104974	2.443651	-2.67645	-0.5923
H	-1.181101	2.535394	1.613345	0.994341	-2.558160	1.640248	-0.45454	-2.72748	-0.19557
H	-2.397829	2.631438	0.319071	2.214182	-2.747143	0.359326	0.647138	-2.85072	1.194231
H	-2.418959	1.283348	1.478568	2.316464	-1.394936	1.509839	-0.68683	-1.67342	1.202329
H	-1.791026	0.925662	-2.003605	1.754284	-1.026551	-1.979166	2.739078	-0.78103	1.243046
H	-3.387900	0.071231	-0.005071	3.375984	-0.252378	0.033436	0.396433	-0.41828	2.535823
H	-3.152323	-1.055586	-1.341010	3.234233	0.867844	-1.321109	1.576113	0.883737	2.687389
H	-2.099716	-2.555433	0.129114	2.259603	2.455730	0.109878	0.100533	2.330177	1.566267
H	-2.100116	-1.376218	1.446656	2.170869	1.300618	1.446013	-1.01507	1.005672	1.194759
H	-0.811710	-1.558489	-2.053497	0.939958	1.515075	-2.069323	2.574811	1.804509	0.593586
H	0.516266	-0.400658	-1.760038	-0.463878	0.449899	-1.784104	2.689336	0.821	-0.89061
H	1.599598	-2.141512	0.164302	-1.461518	2.259105	0.104360	0.679923	2.390477	-2.12906
H	0.549879	-2.545712	1.531921	-0.412013	2.624901	1.485097	-0.84926	2.522041	-1.25169
H	0.270913	-3.297784	-0.055690	-0.066903	3.342349	-0.105843	0.579958	3.407247	-0.67132
S	3.507997	0.045809	-0.301899	-3.503258	-0.140545	-0.334262	-3.58933	-0.27408	0.236739
H	2.757951	0.373035	0.754730	-2.760397	-0.188664	0.775829	-2.7315	-0.41675	-0.77902
H	3.937561	1.300561	-0.455969	-4.154004	0.946630	0.086461	-3.60181	1.054417	0.107357

MP2/6-311++G(d,p) cartesian coordinates of the calculated structures IV_{FEN}, V_{FEN} and VI_{FEN}

Table S13

Element	IV _{FEN}			V _{FEN}			VI _{FEN}		
	E= 863.519117 Hartree			E= 863.517825 Hartree			E= 863.517831 Hartree		
	X	Y	Z	X	Y	Z	X	Y	Z
C	0.192599	-0.059738	-0.820105	-0.15606	0.559208	0.072415	-0.142661	0.538750	-0.059014
C	0.993968	-1.146717	-0.089979	-0.06385	-0.95956	0.277419	-0.065399	-0.969870	0.214240
C	0.630698	1.285880	-0.269386	-1.62716	0.939539	0.078377	-1.593022	0.964695	0.096001
C	1.841304	-0.279185	0.875812	-1.56798	-1.32913	0.340336	-1.562146	-1.290440	0.458898
C	0.948138	0.327063	1.974995	-2.2151	-1.16697	-1.04876	-2.356112	-1.171490	-0.856581
C	0.042014	1.307329	1.176113	-2.15961	0.371958	-1.27326	-2.289757	0.353787	-1.158564
C	2.113152	0.980901	0.025556	-2.17406	-0.10988	1.066387	-2.052316	-0.021709	1.187789
C	1.880104	-1.894491	-1.098081	0.641673	-1.2619	1.609628	0.774232	-1.229187	1.476384
C	0.044154	-2.149635	0.570438	0.732018	-1.61575	-0.85568	0.585524	-1.702984	-0.963072
C	0.267087	2.475636	-1.129945	-1.90456	2.407282	0.317388	-1.802539	2.449790	0.293220
O	-0.654208	-0.261465	-1.675049	0.786186	1.317406	-0.0896	0.791924	1.257674	-0.373226
H	2.612093	-1.237963	-1.577397	0.107325	-0.84905	2.470257	0.355573	-0.747607	2.364770
H	1.257206	-2.340390	-1.880316	1.654164	-0.84682	1.59834	1.794016	-0.861877	1.329589
H	2.422259	-2.696152	-0.583192	0.716498	-2.34782	1.74005	0.821710	-2.308814	1.661553
H	-0.488902	-2.709927	-0.205147	1.793087	-1.37272	-0.75062	1.655882	-1.478071	-0.989822
H	0.610027	-2.860092	1.184865	0.622861	-2.70598	-0.80847	0.464613	-2.786444	-0.844379
H	-0.704267	-1.660777	1.199841	0.405415	-1.27695	-1.84331	0.155689	-1.408836	-1.925115
H	2.725575	-0.804286	1.253051	-1.74384	-2.30433	0.807037	-1.709492	-2.237595	0.988889
H	0.382239	-0.418933	2.537569	-1.70788	-1.73492	-1.8319	-1.953141	-1.788020	-1.663291
H	1.572231	0.872945	2.690724	-3.25313	-1.51414	-1.01135	-3.391402	-1.486778	-0.688718
H	0.111037	2.330170	1.564348	-3.15556	0.790171	-1.45933	-3.290244	0.792138	-1.249980
H	-1.013290	1.015945	1.187486	-1.51682	0.661469	-2.11226	-1.740555	0.587513	-2.077644
H	2.587594	1.782934	0.603703	-3.27057	-0.12836	1.076879	-3.140957	-0.006271	1.319788
H	2.700442	0.798940	-0.880252	-1.81057	0.035607	2.088464	-1.573655	0.158374	2.155312
H	0.709642	2.387603	-2.127663	-1.50724	2.726857	1.286421	-1.288329	2.799999	1.194277
H	-0.819281	2.540464	-1.251861	-1.42663	3.017318	-0.45595	-1.400087	3.008635	-0.557853
H	0.621337	3.404089	-0.668960	-2.98263	2.600962	0.299687	-2.870011	2.676765	0.388225
S	-3.571509	-0.104501	0.221774	4.141585	0.37074	-0.28869	4.151146	0.405678	-0.053655
H	-2.721434	-0.328720	-0.785728	4.558702	0.843934	0.88711	4.454349	0.156277	-1.328937
H	-3.906003	-1.394360	0.298432	2.930112	0.924885	-0.15065	2.927485	0.850318	-0.368181

Natural bonding orbitals representation for endo fenchol-H₂S

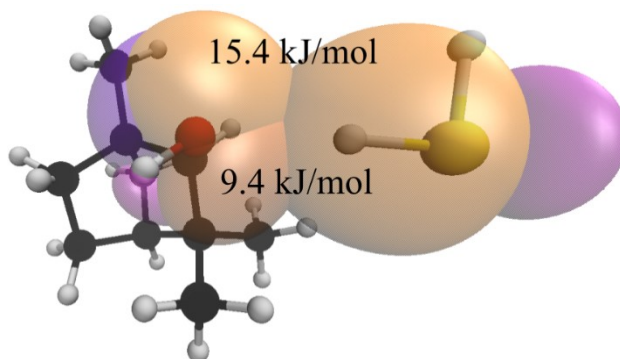


Figure S5. The iso-surface of the natural bond orbitals involved in the hydrogen bonding in the EF-H₂S complex: the two lone pairs of oxygen and SH sigma antibonding. The orange and purple colors represent the positive and negative lobes, respectively. The numbers represent the magnitude of the stabilizing energy of the interaction involved.

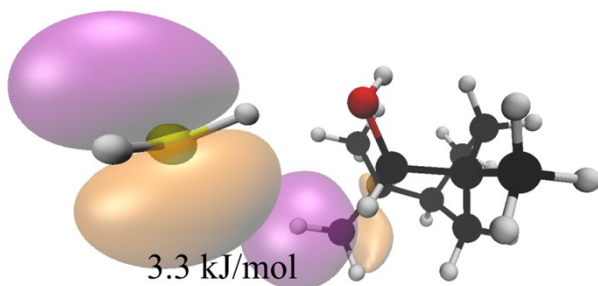


Figure S6. The iso-surface of the natural bond orbitals contributing to the dispersive forces in the EF-H₂S complex: one lone pair of sulfur and the nearest CH sigma antibonding. The orange and purple colors represent the positive and negative lobes, respectively. The numbers represent the magnitude of the stabilizing energy of the interaction involved.

Natural bonding orbitals representation for fenchone-H₂S

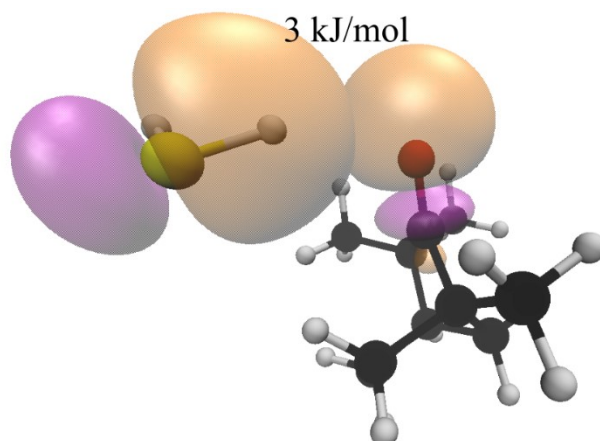


Figure S7. The iso-surface of the natural bond orbitals involved in the hydrogen bonding in the FEN-H₂S complex: one lone pair of oxygen and SH sigma antibonding. The orange and purple colors represent the positive and negative lobes, respectively. The numbers represent the magnitude of the stabilizing energy of the interaction involved.

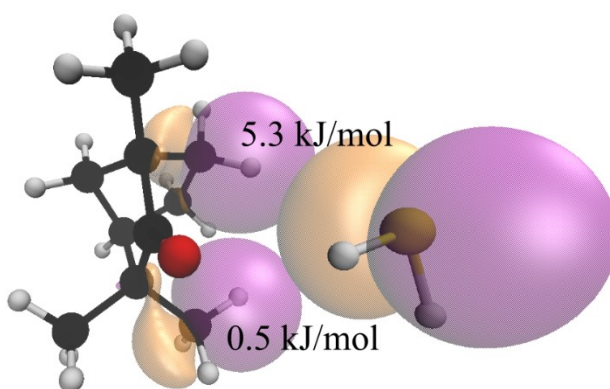


Figure S8. The iso-surface of the natural bond orbitals contributing to the dispersive forces in the FEN-H₂S complex: one lone pair of S and sigma antibonding of SH of the two nearest CH. The orange and purple colors represent the positive and negative lobes, respectively. The numbers represent the magnitude of the stabilizing energy of the interaction involved.

Comparison between the calculated structures of endo fenchol-H₂O and endo fenchol-H₂S complexes

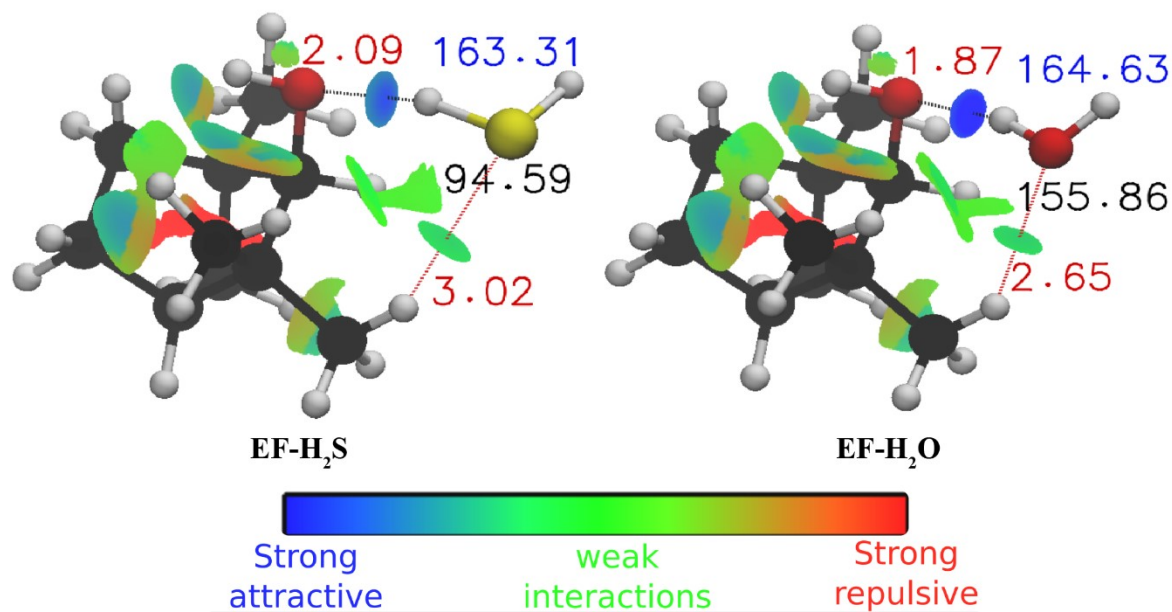


Figure S9 Comparison between the observed endo-fenchol-H₂O^[12] and the endo-fenchol-H₂S complexes. Distances (Å) are shown in red. The OHS and the OHS angles values are indicated in blue and in black, respectively.

In the EF-H₂S complex is a hydrogen bond to the fenchol in a similar way the monohydrate does. It also seems to alter the orientation of the hydroxy group in the molecule, in the same way the H₂O does.

Comparison between the calculated structures of FEN-H₂O and FEN-H₂S complexes

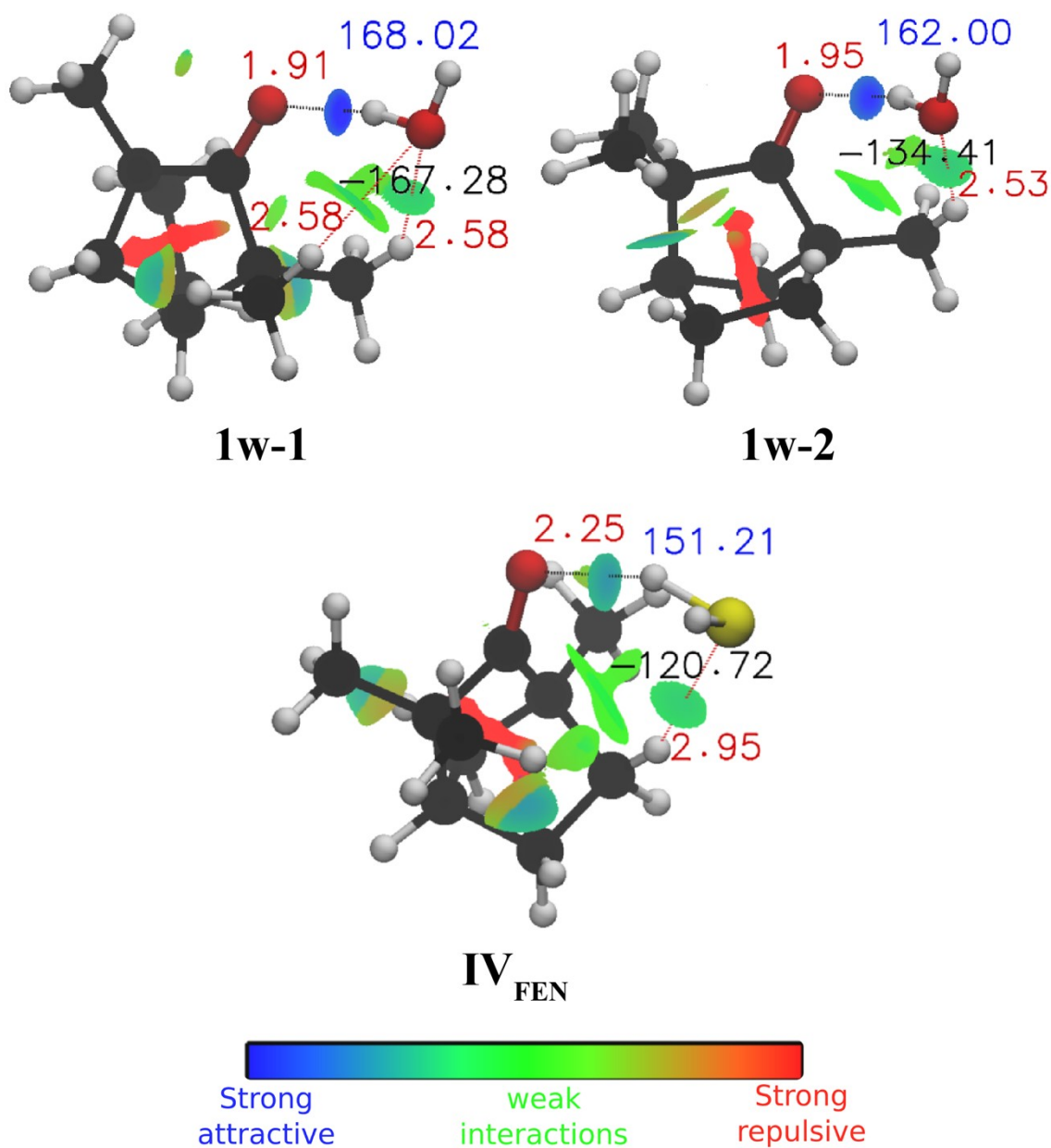


Figure S10. Comparison between the calculated structures of Fenchone-H₂O^[11] and Fenchone-H₂S complexes. Distances (Å) are shown in red. The OHS and the OHSH angles values are indicated in blue and in black, respectively.

The structure of the observed FEN-H₂S complex cannot be compared to that of the monohydrate^[14], as H₂S is hydrogen-bonded to fenchone from a different side of the ring.

SAPT analysis

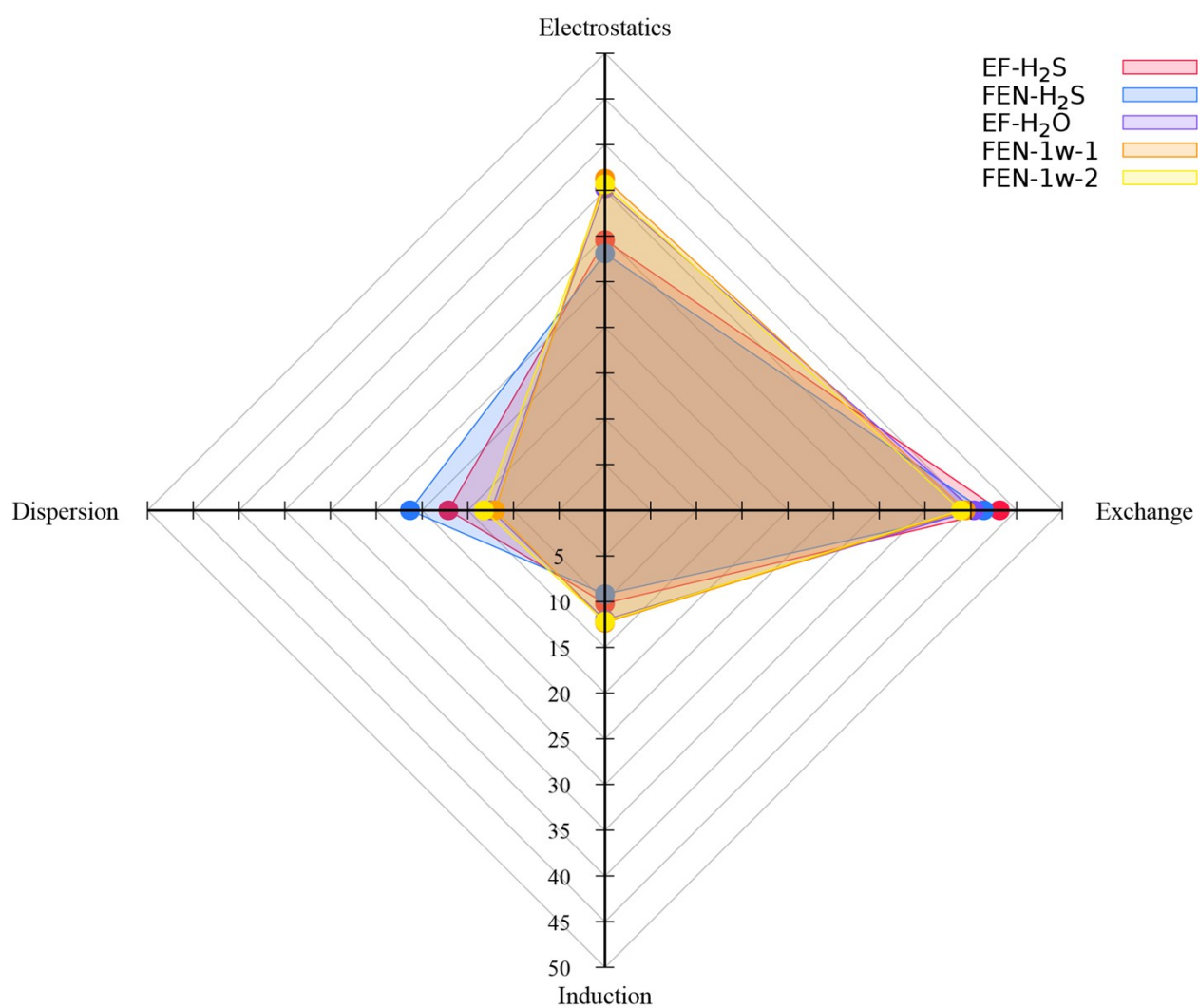


Figure S11. Contribution of the different terms (in%) to the total SAPT energy. The values are derived from SAPT₂₊₃ level. The calculations for the water complexes were performed starting from the geometries reported in the corresponding articles.

Barrier height for the rotation of H₂S around its C₂ axis in the fenchone-H₂S complex

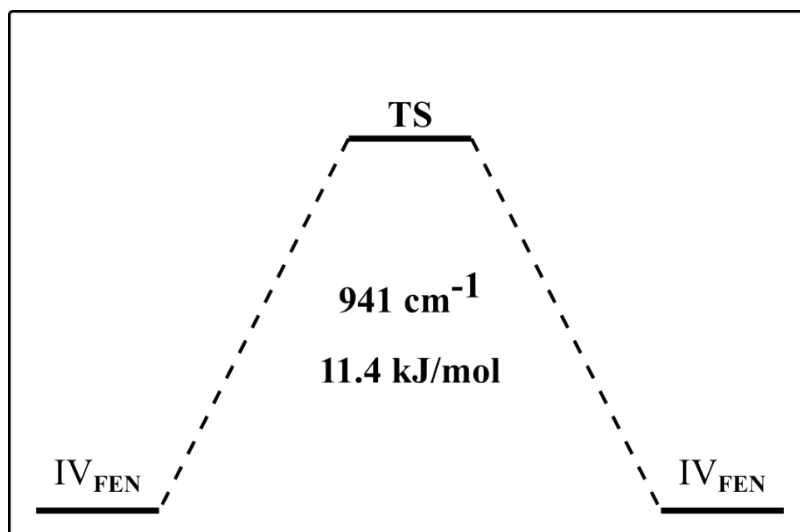


Figure S12. The transition state (TS) and the barrier height of the rotation of H₂S around its C₂ axis in the FEN-H₂S complex

Figure S7 shows the barrier height for the proton exchange in H₂S, which explains the experimental splitting observed due to tunneling. The transition state is obtained by rotating the H₂S molecule around its C₂ axis of symmetry. The barrier height extracted from the scan was calculated to be 941 cm⁻¹. The proton exchange gives two isoenergetic structures.

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