

Supporting Information for

Molecules of life: studying the interactions between water and
phosphine in rare gas matrices

Matthew H. V. Graneri, Duncan A. Wild and Allan J. McKinley*

School of Molecular Sciences, University of Western Australia, 35 Stirling Highway, Crawley,
WA, Australia, 6009

*Correspondence to: allan.mckinley@uwa.edu.au

1. Methods

1.1 Experimental

The matrix isolation Fourier transform infrared (MI-FTIR) spectroscopy apparatus used throughout this work has been described previously,^{1,2} and only the pertinent details will be described here. Infrared spectra were recorded using a Digilab Excalibur 2000 FTS, at a resolution of 0.25 cm⁻¹, averaged over 1000 scans. Matrices were deposited onto a KBr cold window with the temperature being held at 10 K. Matrices were annealed to 30 K before being cooled back down to 10 K. Spectra were recorded at each of these stages (i.e., three spectra per experiment). Spectra presented in this paper have been smoothed using a second order Savitzky-Golay algorithm³ over 15 points in Igor Pro.

Gas mixtures of argon (99.999% purity, Scientific and Technical Gases), H₂O (Milli-Q), and PH₃ at various concentrations were prepared using standard manometric techniques. The dilute PH₃ and H₂O gas mixtures were held in separate reservoirs prior to deposition. The mixtures in both reservoirs were sprayed onto the cold window simultaneously, such that the gas mixtures combined en route to the window. Isotopic substitution experiments were also conducted using D₂O (100 atom % D, Aldrich), H₂¹⁸O (97 atom % ¹⁸O, Sigma), and D₃P. The ¹H₃P sample was prepared by the reduction of phosphorus acid (H₃PO₃).⁴ The D₃P sample was synthesised by condensing deuterium chloride solution (in D₂O, 99 atom % D, Aldrich) onto Zn₃P₂ (99.999%, Alfa Aesar).

1.2 Computational

Ab initio and DFT calculations were performed to aid in the characterisation of the species investigated. In order to calculate accurate vibrational transition energies, harmonic frequency calculations were conducted using MP2,⁵ CCSD(T),⁶ ⁷ B3LYP⁸ with and without the D3 correction,⁹ and DSD-PBEP86-D3(BJ)^{10, 11} levels of theory. Harmonic and anharmonic frequencies were obtained for the DFT methods. All calculations were conducted using the augmented correlation consistent basis sets of Dunning (aug-cc-pVXZ, X=D, T, Q, 5),¹² additional diffuse functions were added to the phosphorus atom where relevant (aug-cc-pV(X+d)Z, henceforth abbreviated to AVXZ).

Binding energies were calculated on CCSD(T)/AVTZ geometries and extrapolated to the complete basis set (CBS) limit using the equation of Peterson et al.:^{13, 14}

$$D_e(n) = D_e(\infty) + Ae^{-(n-1)} + Be^{-(n-1)^2}$$

where n corresponds to the cardinal number of the basis set ($n = 3, 4, 5$ for AVTZ, AVQZ, and AV5Z, respectively), $D_e(n)$ is the energy calculated with the basis set n , and $D_e(\infty)$ is the CBS energy. A counterpoise correction was performed in order to remove any further trace of basis set superposition error.¹⁵ All coupled cluster calculations were performed using ORCA 5.0.¹⁶⁻¹⁸ Geometry optimisations were converged using the criteria: TolE = 1e-9 a.u.; TolMaxD = 1e-6 a.u.; TolMaxG = 1e-6 a.u. Energies were calculated using the 'verytightscf' keyword. All other calculations were conducted in Gaussian 16,¹⁹ optimising geometries using the 'vtight' keyword, a superfine grid (for DFT calculations), and symmetry features, when relevant. Anharmonic frequencies were calculated using generalised second order vibrational perturbation theory (GVPT2)²⁰ as implemented in Gaussian 16. Basis sets not native to Gaussian were obtained from the Basis Set Exchange.²²

2. Tables

Table S1: PH₃ spectral analysis. Spectrum of PH₃/Ar mixture at 1:500 ratio, recorded at 10 K following deposition.

Assignment	¹ H ₃ P		D ₃ P	
	Wavenumber	Intensity	Wavenumber	Intensity
v ₂	994.1	0.0350	729.5	0.0263
v ₄	1114.1	0.0101	803.4	0.0101
v ₁	2340.5	0.0265	1697.5	0.0057
v ₃	2346.5	0.0862	1704.8	0.0784

Table S2: ¹H₂O spectral analysis. Spectrum of ¹H₂O/Ar mixture at 1:500 ratio, recorded at 10 K following deposition.

Assignment		Wavenumber	Intensity	Notes
v ₂	Band Centre	1589.1		Calculated by Perchard. ²³
	1 ₁₁ → 0 ₀₀	1556.6	0.0015	
	1 ₁₀ → 0 ₀₁	1573.1	0.0114	
	nrm	1589.5	0.0015	
	1 ₀₁ → 1 ₁₀	1607.9	0.0998	
	0 ₀₀ → 1 ₁₁	1623.9	0.1630	
	1 ₀₁ → 2 ₁₂	1636.5	0.0157	
	RTC-para	1657.8	0.0015	Assigned by Michaut et al. ²⁴
	RTC-ortho	1661.1	0.0114	Assigned by Michaut et al. ²⁴
v ₁	Band Centre	3639.2		Calculated by Perchard. ²³
	1 ₁₀ → 0 ₀₁	3622.8	0.0005	
	1 ₁₀ → 1 ₀₁	3622.8	0.0005	
	nrm	3638.9		Shoulder
	1 ₀₁ → 1 ₁₀	3653.5	0.0050	
	0 ₀₀ → 1 ₁₁	3670.2	0.0081	
v ₃	Band Centre	3732.2		Calculated by Perchard. ²³
	1 ₀₁ → 0 ₀₀	3711.4	0.0511	
	1 ₁₀ → 1 ₁₁	3724.8	0.0027	
	nrm	3736.4	0.0009	
	0 ₀₀ → 1 ₀₁	3756.6	0.1331	
	1 ₀₁ → 2 ₀₂	3776.4	0.0500	
	1 ₁₀ → 2 ₁₁	3784.0		Appears as shoulder

Table S3: H₂¹⁸O spectral analysis. Spectrum of H₂¹⁸O/Ar mixture at 1:500 ratio, recorded at 10 K following deposition.

Assignment		Wavenumber	Intensity	Notes
v ₂	Band Centre	1582.5		Calculated by Perchard. ²³
	1 ₁₁ → 0 ₀₀	1551.4	0.0017	
	1 ₁₀ → 0 ₀₁	1567.4	0.0101	
	nrm	1583.1	0.0039	
	1 ₀₁ → 1 ₁₀	1600.7	0.0726	
	0 ₀₀ → 1 ₁₁	1616.1	0.1209	
v ₁	Band Centre	3631.5		Calculated by Perchard. ²³
	nrm	3633.3	0.0014	
	1 ₀₁ → 1 ₁₀	3645.7	0.0038	
	0 ₀₀ → 1 ₁₁	3661.6	0.0059	
v ₃	Band Centre	3718.8		Calculated by Perchard. ²³
	1 ₀₁ → 0 ₀₀	3697.7		
	1 ₁₀ → 1 ₁₁	3711.3	0.0076	
	nrm	3722.2	0.0015	
	1 ₁₁ → 1 ₁₀	3724.7		Approximate position
	0 ₀₀ → 1 ₀₁	3742.4	0.1229	
	1 ₀₁ → 2 ₀₂	3760.9	0.0277	
	1 ₁₀ → 2 ₁₁	3711.3	0.0076	

Table S4: D₂O spectral analysis. Spectrum of D₂O/Ar mixture at 1:500 ratio, recorded at 10 K following deposition.

Assignment		Wavenumber	Intensity	Notes
v ₂	Band Centre	1175		Calculated by Redington and Milligan. ²⁵
	0 ₀ -1 ₀	1154.5	0.0071	
	1 ₋₁ -1 ₊₁	1164.0	0.0035	
	(0 ₀ -0 ₀)	1177.3	0.0113	
	1 ₊₁ -1 ₋₁	1185.4	0.0105	
	2 ₋₂ -1 ₀	1188.3	0.0011	
	1 ₀ -0 ₀	1195.0	0.1168	
	2 ₋₁ -1 ₋₁	1203.5	0.0052	
	2 ₊₁ -1 ₊₁	1228.2	0.0019	
v ₁	Band Centre	2657		Calculated by Redington and Milligan. ²⁵
	1 ₀ → 0 ₀	2677.4	0.0096	
v ₃	Band Centre	2770		Calculated by Redington and Milligan. ²⁵
	0 ₀ -1 ₋₁	2758.7	0.0084	
	1 ₊₁ -1 ₀	2770.9	0.0044	
	1 ₋₁ -0 ₀	2782.3	0.0958	
	2 ₋₂ -1 ₋₁	2793.5	0.0137	
	2 ₀ -1 ₊₁	2803.4	0.0006	

Table S5: Theoretical frequencies of $^1\text{H}_2\text{O}$ calculated at various levels of theory.

	MP2/AVQZ		CCSD(T)/AVTZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3965.5	78.4	3919.9	53.4
ν_1	3839.7	6.1	3810.8	2.8
ν_2	1632.4	73.2	1645.9	68.1
<hr/>				
	B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3905.9	63.5	3723.1	58.7
ν_1	3803.9	4.7	3634.4	3.5
ν_2	1629.0	76.2	1576.3	77.2
<hr/>				
	DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3939.2	59.7	3746.1	55.5
ν_1	3816.9	3.6	3635.5	2.5
ν_2	1634.5	69.5	1583.6	70.4

Table S6: Theoretical frequencies of H_2^{18}O calculated at various levels of theory.

	MP2/AVQZ		CCSD(T)/AVTZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3949.4	76.8	3904.0	52.2
ν_1	3831.6	5.8	3802.8	2.6
ν_2	1625.7	72.6	1639.1	67.5
<hr/>				
	B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3889.8	62.1	3708.8	57.5
ν_1	3796.0	4.4	3627.1	3.3
ν_2	1622.2	75.6	1569.9	76.6
<hr/>				
	DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
	Wavenumber	Intensity	Wavenumber	Intensity
ν_3	3923.2	58.4	3731.8	54.3
ν_1	3808.8	3.4	3628.1	2.3
ν_2	1627.7	69.0	1577.2	69.9

Table S7: Theoretical frequencies of D₂O calculated at various levels of theory.

	MP2/AVQZ		CCSD(T)/AVTZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v₃	2905.3	46.5	2880.5	32.4
v₁	2767.9	4.7	2756.2	2.5
v₂	1194.8	38.9	1208.3	36.5

	B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v₃	2862.7	38.0	2763.6	35.8
v₁	2741.5	3.8	2653.9	3.2
v₂	1192.5	40.6	1164.5	41.0

	DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v₃	2885.9	35.9	2781.6	33.9
v₁	2752.0	3.0	2657.8	2.5
v₂	1196.0	37.0	1169.5	37.4

Table S8: Theoretical frequencies of ${}^1\text{H}_3\text{P}$ calculated at various levels of theory.

	MP2/AVQZ		CCSD(T)/AVTZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	2492.0	47.9	2416.3	46.6
v_{3a}	2492.0	47.9	2416.3	46.6
v₁	2480.7	25.9	2409.2	24.9
v_{4s}	1162.2	14.5	1143.0	11.0
v_{4a}	1162.2	14.5	1143.0	11.0
v₂	1015.0	21.3	1012.8	17.2

	B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	2394.3	57.6	2330.9	310.1
v_{3a}	2394.3	57.6	2330.9	310.1
v₁	2386.9	34.1	2338.8	463.8
v_{4s}	1137.9	12.2	1106.5	12.1
v_{4a}	1137.9	12.2	1106.5	12.1
v₂	1018.0	21.2	977.6	18.4

	DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	2439.6	59.3	2392.6	128.4
v_{3a}	2439.6	59.3	2392.6	128.4
v₁	2425.5	33.2	2392.3	157.0
v_{4s}	1144.3	11.9	1100.8	11.7
v_{4a}	1144.3	11.9	1100.8	11.7
v₂	1013.9	18.6	974.9	16.5

Table S9: Theoretical frequencies of D₃P calculated at various levels of theory.

	MP2/AVQZ		CCSD(T)/AVTZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	1791.6	25.1	1742.8	24.4
v_{3a}	1791.6	25.1	1742.8	24.4
v₁	1775.8	14.4	1730.4	13.8
v_{4s}	829.2	7.0	818.2	5.3
v_{4a}	829.2	7.0	818.2	5.3
v₂	740.3	11.1	741.0	9.0

	B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	1721.5	30.1	1687.9	42.8
v_{3a}	1721.5	30.1	1687.9	42.8
v₁	1708.8	18.8	1685.1	39.6
v_{4s}	811.8	5.9	793.6	5.9
v_{4a}	811.8	5.9	793.6	5.9
v₂	742.5	11.0	713.7	9.7

	DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
	Wavenumber	Intensity	Wavenumber	Intensity
v_{3s}	1753.9	30.9	1727.4	36.9
v_{3a}	1753.9	30.9	1727.4	36.9
v₁	1736.6	18.2	1718.0	28.0
v_{4s}	816.4	5.7	787.8	5.5
v_{4a}	816.4	5.7	787.8	5.5
v₂	739.4	9.6	712.6	8.8

Table S10: Theoretical frequencies of $^1\text{H}_2\text{O}\cdot^1\text{H}_3\text{P}$ calculated at various levels of theory.

		MP2/AVQZ		CCSD(T)/AVTZ	
		Wavenumber	Intensity	Wavenumber	Intensity
$^1\text{H}_3\text{P}$	ν_{3s}	2508.8	32.6	2434.0	31.8
	ν_{3a}	2506.8	34.9	2432.0	34.4
	ν_1	2494.2	25.0	2423.8	24.6
	ν_{4s}	1160.2	11.9	1141.5	10.1
	ν_{4a}	1160.4	13.4	1141.2	9.0
	ν_2	1011.4	34.0	1008.2	28.2
$^1\text{H}_2\text{O}$	ν_3	3932.7	157.0	3893.8	129.4
	ν_1	3774.4	191.1	3768.1	133.4
	ν_2	1639.0	35.2	1653.1	33.6
		B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
		Wavenumber	Intensity	Wavenumber	Intensity
$^1\text{H}_3\text{P}$	ν_{3s}	2414.5	40.1	2288.5	47.2
	ν_{3a}	2412.1	43.1	2320.8	37.6
	ν_1	2403.5	34.6	2326.5	29.6
	ν_{4s}	1136.1	10.0	1114.7	6.6
	ν_{4a}	1136.4	11.4	1105.4	6.8
	ν_2	1012.8	36.7	1006.8	25.4
$^1\text{H}_2\text{O}$	ν_3	3876.6	121.1	3692.9	95.1
	ν_1	3726.9	241.5	3570.3	141.6
	ν_2	1635.4	37.7	1575.1	30.9
		DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
		Wavenumber	Intensity	Wavenumber	Intensity
$^1\text{H}_3\text{P}$	ν_{3s}	2458.2	41.5	2388.6	29.5
	ν_{3a}	2455.6	44.0	2364.8	36.4
	ν_1	2440.3	32.9	2325.5	44.2
	ν_{4s}	1142.9	9.9	1120.5	7.8
	ν_{4a}	1142.8	11.2	1123.3	7.5
	ν_2	1010.4	32.2	996.3	21.3
$^1\text{H}_2\text{O}$	ν_3	3908.5	138.2	3711.2	108.6
	ν_1	3757.4	179.3	3583.6	104.4
	ν_2	1640.8	32.7	1578.3	27.2

Table S11: Theoretical frequencies of H₂¹⁸O-¹H₃P calculated at various levels of theory.

		MP2/AVQZ		CCSD(T)/AVTZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹ H ₃ P	v _{3s}	2508.8	32.6	2434.0	31.8
	v _{3a}	2506.8	34.9	2432.0	34.4
	v ₁	2494.2	25.0	2423.8	24.6
	v _{4s}	1160.2	11.9	1141.2	9.0
	v _{4a}	1160.4	13.4	1141.5	10.1
	v ₂	1011.4	34.0	1008.2	28.2
H ₂ ¹⁸ O	v ₃	3917.7	149.6	3878.7	123.1
	v ₁	3765.4	194.2	3759.4	136.2
	v ₂	1632.4	34.9	1646.4	33.4
		B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹ H ₃ P	v _{3s}	2414.5	40.1	2297.0	47.6
	v _{3a}	2412.1	43.0	2316.2	37.7
	v ₁	2403.5	34.6	2333.7	30.0
	v _{4s}	1136.1	10.0	1132.7	7.4
	v _{4a}	1136.4	11.4	1137.0	6.8
	v ₂	1012.8	36.7	1001.2	25.8
H ₂ ¹⁸ O	v ₃	3862.1	112.7	3680.5	79.8
	v ₁	3717.5	245.4	3564.8	172.7
	v ₂	1628.8	37.4	1590.2	30.2
		DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹ H ₃ P	v _{3s}	2458.2	41.5	2390.4	29.5
	v _{3a}	2455.6	44.0	2363.4	36.1
	v ₁	2440.3	32.9	2322.3	44.1
	v _{4s}	1142.9	9.9	1119.1	7.9
	v _{4a}	1142.8	11.2	1123.3	7.2
	v ₂	1010.4	32.2	996.1	21.5
H ₂ ¹⁸ O	v ₃	3893.5	131.3	3697.8	104.7
	v ₁	3748.4	182.2	3575.5	109.3
	v ₂	1634.3	32.5	1572.4	27.1

Table S12: Theoretical frequencies of D₂O-¹H₃P calculated at various levels of theory.

		MP2/AVQZ		CCSD(T)/AVTZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹H₃P	v_{3s}	2508.8	32.4	2434.0	31.7
	v_{3a}	2506.8	34.8	2431.9	34.2
	v₁	2494.2	26.2	2423.7	25.3
	v_{4s}	1159.2	7.9	1140.8	7.4
	v_{4a}	1160.3	12.9	1141.4	9.7
	v₂	1011.4	34.4	1008.2	28.5
D₂O	v₃	2877.5	105.8	2858.8	86.7
	v₁	2725.4	86.0	2728.3	59.8
	v₂	1199.4	23.2	1212.9	19.9
		B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹H₃P	v_{3s}	2414.5	39.9	2293.9	43.0
	v_{3a}	2412.1	42.9	2319.2	36.2
	v₁	2403.4	36.0	2325.3	27.6
	v_{4s}	1135.5	7.8	1103.5	8.8
	v_{4a}	1136.3	10.9	1104.2	6.7
	v₂	1012.7	37.0	988.9	30.6
D₂O	v₃	2835.2	91.9	2737.1	75.3
	v₁	2692.9	107.0	2609.4	68.9
	v₂	1196.6	22.7	1162.0	14.0
		DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
		Wavenumber	Intensity	Wavenumber	Intensity
¹H₃P	v_{3s}	2458.2	41.3	2385.1	28.5
	v_{3a}	2455.6	43.8	2360.4	34.0
	v₁	2440.2	34.2	2325.2	43.7
	v_{4s}	1142.1	7.5	1108.0	10.6
	v_{4a}	1142.8	10.7	1121.6	6.8
	v₂	1010.4	32.5	994.2	27.0
D₂O	v₃	2859.9	94.0	2755.5	71.9
	v₁	2713.3	80.6	2621.7	56.7
	v₂	1200.3	20.4	1176.1	10.5

Table S13: Theoretical frequencies of $^1\text{H}_2\text{O}\cdot\text{D}_3\text{P}$ calculated at various levels of theory.

		MP2/AVQZ		CCSD(T)/AVTZ	
		Wavenumber	Intensity	Wavenumber	Intensity
D₃P	v_{3s}	1804.1	18.1	1755.9	18.3
	v_{3a}	1802.6	18.8	1754.4	18.5
	v₁	1784.8	13.7	1740.3	13.5
	v_{4s}	827.8	6.1	817.1	4.6
	v_{4a}	828.0	7.3	817.2	5.5
	v₂	739.0	16.8	738.9	14.0
¹H₂O	v₃	3932.7	157.3	3893.8	129.6
	v₁	3774.3	191.8	3768.0	133.9
	v₂	1638.8	34.0	1652.9	32.1
		B3LYP-D3/AVQZ		GVPT2/B3LYP-D3/AVQZ	
		Wavenumber	Intensity	Wavenumber	Intensity
D₃P	v_{3s}	1736.4	22.7	1706.8	19.2
	v_{3a}	1734.6	23.1	1681.0	21.4
	v₁	1720.1	18.8	1647.4	21.2
	v_{4s}	810.6	5.1	797.0	6.1
	v_{4a}	810.8	6.2	779.0	4.9
	v₂	740.0	18.1	725.0	13.2
¹H₂O	v₃	3876.5	121.3	3692.1	93.0
	v₁	3726.8	242.5	3567.9	145.6
	v₂	1635.2	36.0	1573.9	27.0
		DSD-PBEP86-D3(BJ)/AVDZ		GVPT2/DSD-PBEP86-D3(BJ)/AVDZ	
		Wavenumber	Intensity	Wavenumber	Intensity
D₃P	v_{3s}	1767.7	23.1	1725.7	18.4
	v_{3a}	1765.8	23.5	1718.8	20.5
	v₁	1746.5	17.9	1690.3	21.4
	v_{4s}	815.4	5.0	802.9	4.2
	v_{4a}	815.4	6.1	801.8	5.4
	v₂	738.2	15.7	727.3	23.8
¹H₂O	v₃	3908.5	138.5	3710.2	109.2
	v₁	3757.3	180.1	3583.6	102.8
	v₂	1640.6	31.3	1577.0	24.0

Table S14: Geometries optimised for H₂O at various levels of theory. Coordinates are given in angstroms (Å).

CCSD(T)/aug-cc-pV(T+d)Z	x	y	z
O	-0.00528	-0.00629	0.00000
H	0.95567	-0.04135	0.00000
H	-0.20667	0.93397	0.00000
Electronic Energy (Hartree)	-76.3423		
MP2/aug-cc-pV(Q+d)Z	x	y	z
O	0.00137	0.00000	0.00107
H	0.00003	0.00000	0.95993
H	0.93098	0.00000	-0.23398
Electronic Energy (Hartree)	-76.3519		
B3LYP-D3/aug-cc-pV(Q+d)Z	x	y	z
O	0.00050	0.00000	0.00038
H	0.00027	0.00000	0.96099
H	0.92791	0.00000	-0.24998
Electronic Energy (Hartree)	-76.4722		
DSD-PBEP86-D3BJ/aug-cc-pV(D+d)Z	x	y	z
O	-0.03991	0.00000	-0.03349
H	-0.07528	0.00000	0.93024
H	0.90303	0.00000	-0.23567
Electronic Energy (Hartree)	-76.1952		
DSD-PBEP86-D3BJ/aug-cc-pV(Q+d)Z	x	y	z
O	0.00243	0.00000	0.00189
H	-0.00159	0.00000	0.96091
H	0.93154	0.00000	-0.23578
Electronic Energy (Hartree)	-76.2202		

Table S15: Geometries optimised for PH₃ at various levels of theory. Coordinates are given in angstroms (Å).

CCSD(T)/aug-cc-pV(T+d)Z	x	y	z
P	0.00293	0.00311	0.00331
H	1.41958	0.00257	0.00274
H	-0.08193	1.41721	0.00274
H	-0.08194	-0.08697	1.41454
Electronic Energy (Hartree)	-342.6990		
MP2/aug-cc-pV(Q+d)Z	x	y	z
P	0.00087	0.00093	0.00081
H	-0.00022	-0.00024	1.40933
H	1.40651	-0.00024	-0.08911
H	-0.09492	1.40330	-0.08911
Electronic Energy (Hartree)	-342.6777		
B3LYP-D3/aug-cc-pV(Q+d)Z	x	y	z
P	0.00045	-0.00049	0.00043
H	0.00004	-0.00005	1.41757
H	1.41481	-0.00005	-0.08843
H	-0.09413	-1.41168	-0.08843
Electronic Energy (Hartree)	-343.1868		
DSD-PBEP86-D3BJ/aug-cc-pV(D+d)Z	x	y	z
P	-0.00359	-0.00384	-0.00336
H	-0.00259	-0.00277	1.42229
H	1.41925	-0.00277	-0.09295
H	-0.09906	1.41579	-0.09295
Electronic Energy (Hartree)	-342.7498		
DSD-PBEP86-D3BJ/aug-cc-pV(Q+d)Z	x	y	z
P	-0.00046	-0.00049	-0.00043
H	-0.00146	-0.00157	1.41358
H	1.41067	-0.00157	-0.09061
H	-0.09652	1.40737	-0.09061
Electronic Energy (Hartree)	-342.7661		

Table S16: Geometries optimised for H₂O·PH₃ at various levels of theory. Coordinates are given in angstroms (Å).

CCSD(T)/aug-cc-pV(T+d)Z	x	y	z
P	0.00000	0.00000	0.00000
H	1.41433	0.00000	0.00000
H	-0.10934	1.41009	0.00000
H	-0.11028	-0.11915	1.40461
H	-1.54764	-1.67487	-1.33879
O	-2.15926	-2.33771	-1.68217
H	-2.08698	-2.25465	-2.63727
Electronic Energy (Hartree)	-419.0456		
MP2/aug-cc-pV(Q+d)Z	x	y	z
P	-0.00002	-0.00001	-0.00002
H	-0.00002	-0.00002	1.40634
H	1.40155	-0.00002	-0.11607
H	-0.12748	1.39525	-0.11737
H	-1.63980	-1.35966	-1.50964
O	-2.30177	-1.70441	-2.11906
H	-2.22110	-2.65675	-2.04479
Electronic Energy (Hartree)	-419.0339		
B3LYP-D3/aug-cc-pV(Q+d)Z	x	y	z
P	-0.00098	-0.00107	0.00153
H	0.00156	0.00170	1.41557
H	1.40850	-0.00176	-0.11728
H	-0.12603	1.40285	-0.11728
H	-1.61144	-1.76050	-1.06497
O	-2.24116	-2.44847	-1.31474
H	-2.24336	-2.45087	-2.27486
Electronic Energy (Hartree)	-419.6632		
DSD-PBEP86-D3BJ/aug-cc-pV(D+d)Z	x	y	z
P	0.00164	0.00179	-0.00867
H	-0.00850	-0.00931	1.41391
H	1.42083	0.00428	-0.11454
H	-0.12377	1.41543	-0.11454
H	-1.60597	-1.75784	-1.07809
O	-2.24122	-2.45316	-1.30463

H	-2.24968	-2.46242	-2.26854
Electronic Energy (Hartree)	-418.9485		
<u>DSD-PBEP86-D3BJ/aug-cc-pV(Q+d)Z</u>	<i>x</i>	<i>y</i>	<i>z</i>
P	-0.00165	-0.00180	0.00377
H	0.00394	0.00431	1.41499
H	1.40460	-0.00321	-0.11945
H	-0.12804	1.39876	-0.11945
H	-1.60512	-1.75473	-1.05684
O	-2.23371	-2.44191	-1.30482
H	-2.23096	-2.43891	-2.26355
Electronic Energy (Hartree)	-418.9897		

References

1. V. S. Langford, A. J. McKinley and T. I. Quickenden, *J. Am. Chem. Soc.*, 2000, **122**, 12859-12863.
2. M. H. V. Graneri, D. A. Wild and A. J. McKinley, *J. Mol. Spectrosc.*, 2021, **378**, 111440.
3. A. Savitzky and M. J. E. Golay, *Anal. Chem.*, 1964, **36**, 1627-1639.
4. S. D. Gokhale, W. L. Jolly, S. Thomas and D. Britton, in *Inorg. Synth.*, ed. S. Y. Tyree, Jr., McGraw-Hill, Inc., 1967, vol. 9, pp. 56-58.
5. C. Møller and M. S. Plesset, *Phys. Rev.*, 1934, **46**, 618-622.
6. K. Raghavachari, G. W. Trucks, J. A. Pople and M. Head-Gordon, *Chem. Phys. Lett.*, 1989, **157**, 479-483.
7. F. Wennmohs and F. Neese, *Chem. Phys.*, 2008, **343**, 217-230.
8. A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648-5652.
9. S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
10. S. Kozuch and J. M. L. Martin, *Phys. Chem. Chem. Phys.*, 2011, **13**, 20104-20107.
11. S. Kozuch and J. M. L. Martin, *J. Comput. Chem.*, 2013, **34**, 2327-2344.
12. R. A. Kendall, T. H. Dunning, Jr. and R. J. Harrison, *J. Chem. Phys.*, 1992, **96**, 6796-6806.
13. K. A. Peterson, D. E. Woon and T. H. Dunning, Jr., *J. Chem. Phys.*, 1994, **100**, 7410-7415.
14. V. Vasilyev, *Comput. Theor. Chem.*, 2017, **1115**, 1-3.
15. S. F. Boys and F. Bernardi, *Mol. Phys.*, 2002, **100**, 65-73.
16. F. Neese, *WIREs Comput. Mol. Sci.*, 2018, **8**, e1327.
17. F. Neese, F. Wennmohs, U. Becker and C. Riplinger, *J. Chem. Phys.*, 2020, **152**, 224108.
18. F. Neese, *WIREs Comput. Mol. Sci.*, 2022, DOI: 10.1002/wcms.1606, e1606.
19. M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, Williams, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman and D. J. Fox, *Gaussian 16*, Rev. A.03; Gaussian, Inc.: Wallingford, CT, 2016.
20. V. Barone, *J. Chem. Phys.*, 2005, **122**, 014108.
21. J. Bloino and V. Barone, *J. Chem. Phys.*, 2012, **136**, 124108.
22. B. P. Pritchard, D. Altarawy, B. Didier, T. D. Gibson and T. L. Windus, *J. Chem. Inf. Model.*, 2019, **59**, 4814-4820.

23. J. P. Perchard, *Chem. Phys.*, 2001, **273**, 217-233.
24. X. Michaut, A.-M. Vasserot and L. Abouaf-Marguin, *Vib. Spectrosc.*, 2004, **34**, 83-93.
25. R. L. Redington and D. E. Milligan, *J. Chem. Phys.*, 1962, **37**, 2162-2166.