

**A novel material $\text{Cs}_2\text{Rb}_x\text{Ag}_{1-x}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$ with special blue shift
and application for white light LED devices**

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Table S1 The contents of Rb for the samples

x value	Rb contents (mmol)	Ag contents (mmol)	chemical formula
0	0	1	$\text{Cs}_2\text{AgIn}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.05	0.05	0.95	$\text{Cs}_2\text{Rb}_{0.05}\text{Ag}_{0.95}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.1	0.1	0.9	$\text{Cs}_2\text{Rb}_{0.1}\text{Ag}_{0.9}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.15	0.15	0.85	$\text{Cs}_2\text{Rb}_{0.15}\text{Ag}_{0.85}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.2	0.2	0.8	$\text{Cs}_2\text{Rb}_{0.2}\text{Ag}_{0.8}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.25	0.25	0.75	$\text{Cs}_2\text{Rb}_{0.25}\text{Ag}_{0.75}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.3	0.3	0.7	$\text{Cs}_2\text{Rb}_{0.3}\text{Ag}_{0.7}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.4	0.4	0.6	$\text{Cs}_2\text{Rb}_{0.4}\text{Ag}_{0.6}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.6	0.6	0.4	$\text{Cs}_2\text{Rb}_{0.6}\text{Ag}_{0.4}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
0.8	0.8	0.2	$\text{Cs}_2\text{Rb}_{0.8}\text{Ag}_{0.2}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$
1	1	0	$\text{Cs}_2\text{RbIn}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$

*Each sample was intended to yield 1 mmol of product.

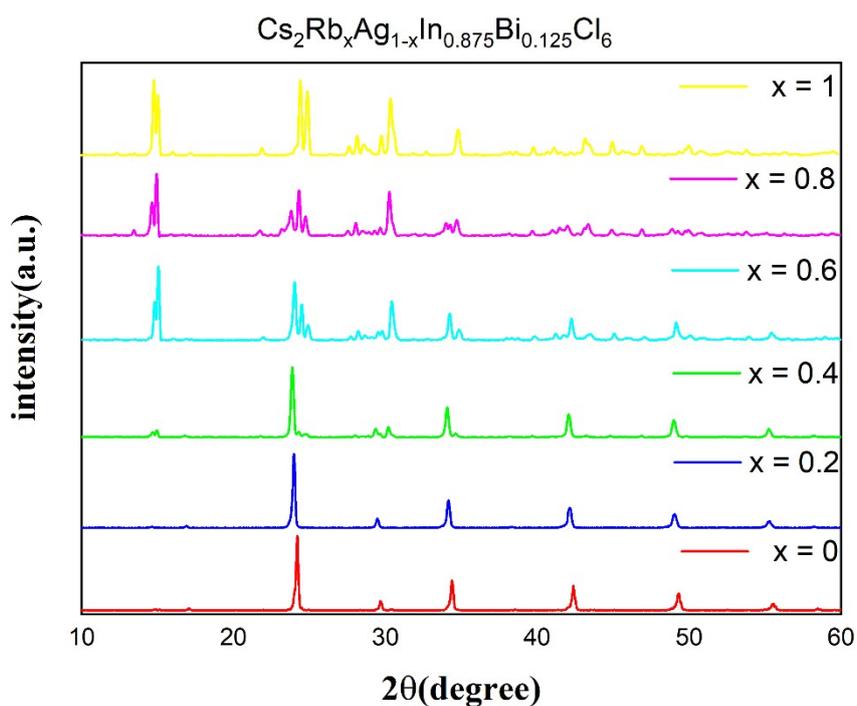


Fig S1 XRD image of $\text{Cs}_2\text{Rb}_x\text{Ag}_{1-x}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$ ($x=0, 0.2, 0.4, 0.6, 0.8, 1$).

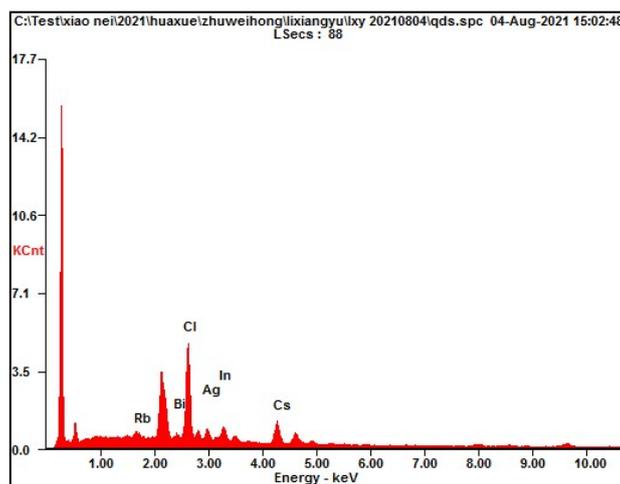


Fig S2 The EDS spectra of $\text{Cs}_2\text{Rb}_{0.2}\text{Ag}_{0.8}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$.

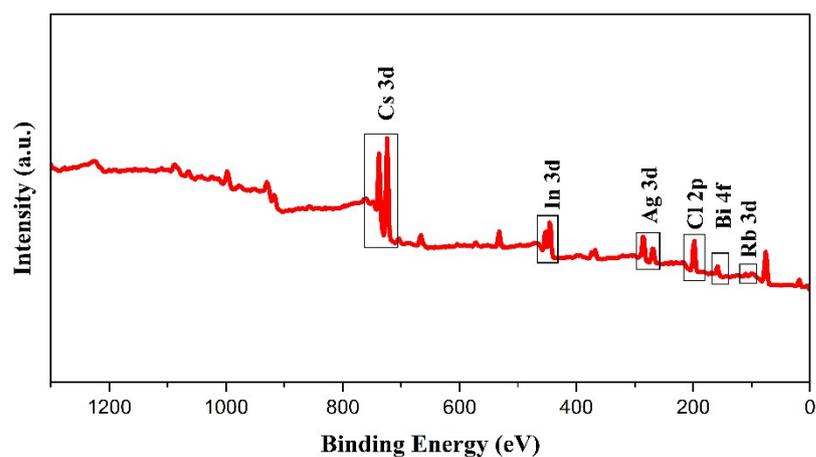


Fig S3 The XPS spectra of $\text{Cs}_2\text{Rb}_{0.2}\text{Ag}_{0.8}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$

Table S1 Summary of the EDS analysis of $\text{Cs}_2\text{Rb}_{0.2}\text{Ag}_{0.8}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$.

Element	Wt%	At%
RbL	02.74	02.34
BiM	05.04	01.76
ClK	28.27	58.26
AgL	10.27	06.96
InL	13.55	08.62
CsL	40.13	22.06
Matrix	Correction	ZAF

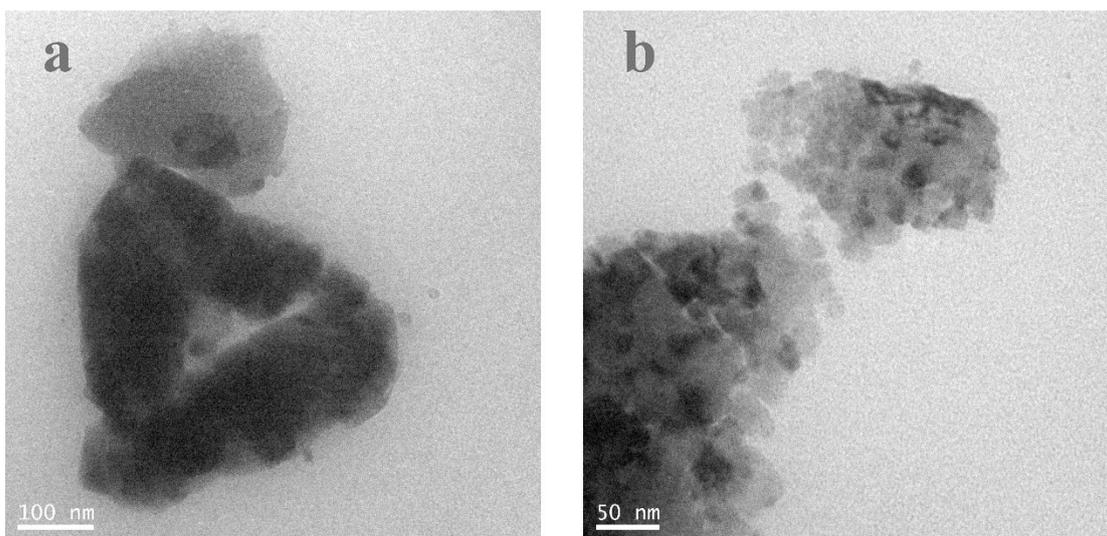


Fig S4 TEM image of $\text{Cs}_2\text{Rb}_{0.2}\text{Ag}_{0.8}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$, from which a microcrystalline structure can be analyzed.

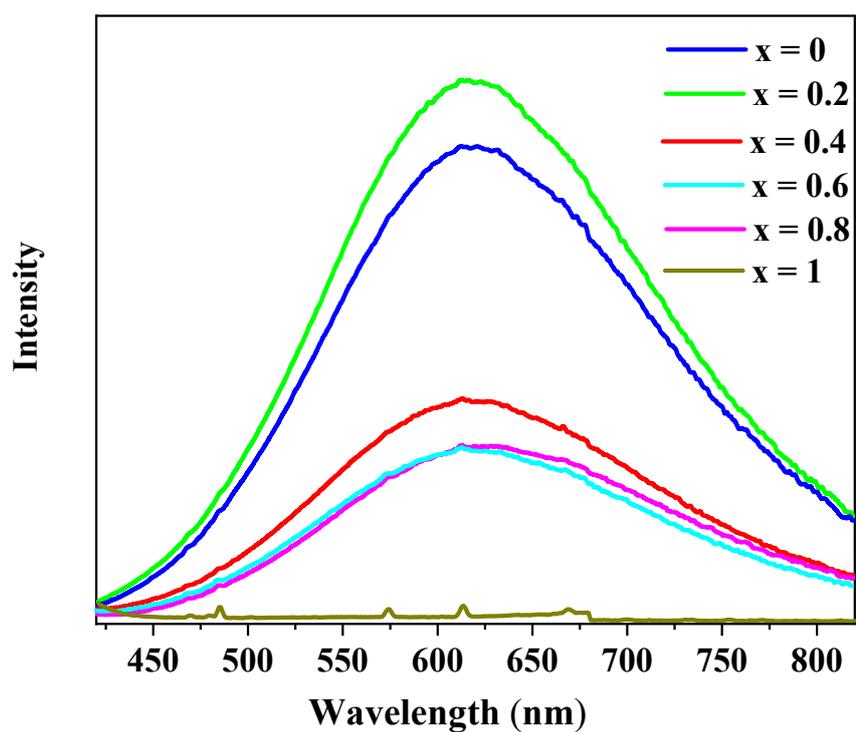


Fig S5 The emission spectra of $\text{Cs}_2\text{Rb}_x\text{Ag}_{1-x}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$ ($x=0, 0.2, 0.4, 0.6, 0.8, 1$).

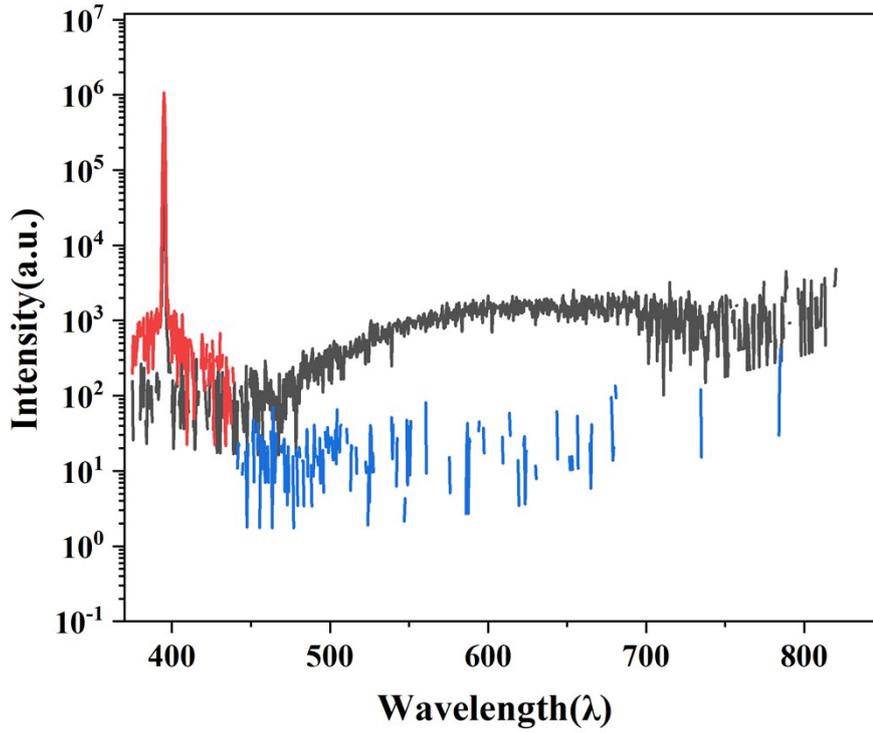


Fig S6 PL for calculation for PLQY ($\text{Cs}_2\text{Rb}_x\text{Ag}_{1-x}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$, $x = 0.15$). The red and blue curve represent for the PL background (measured without samples in), and the black curve represent PL for sample.

According to protocol,

$$\text{PLQY} = \frac{E_B - E_A}{S_A - S_B}$$

S_A is the blank spectrum when the vessel is filled Teflon instead of the sample. S_B is the is the peak signal of the remaining excitation source after the sample placed. E_A is the fluorescence background baseline and E_B is emission signal of the sample.

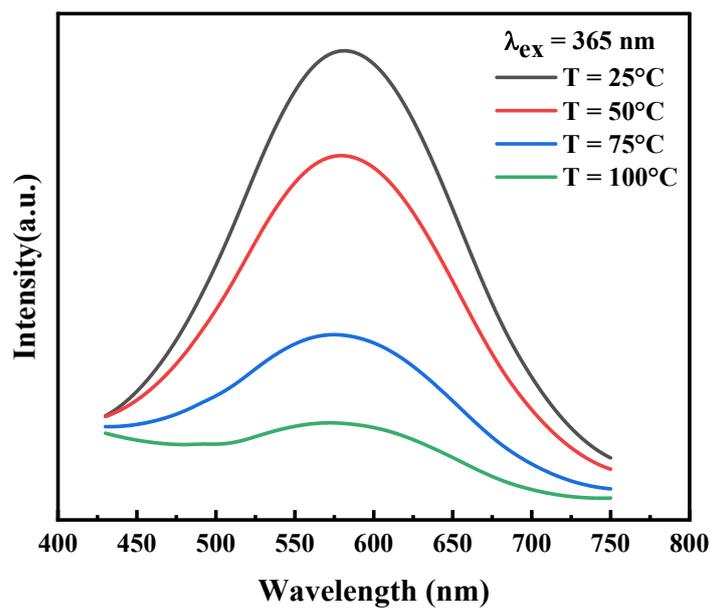


Fig S7 PL for $\text{Cs}_2\text{Rb}_{0.15}\text{Ag}_{0.85}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$ under different temperature.

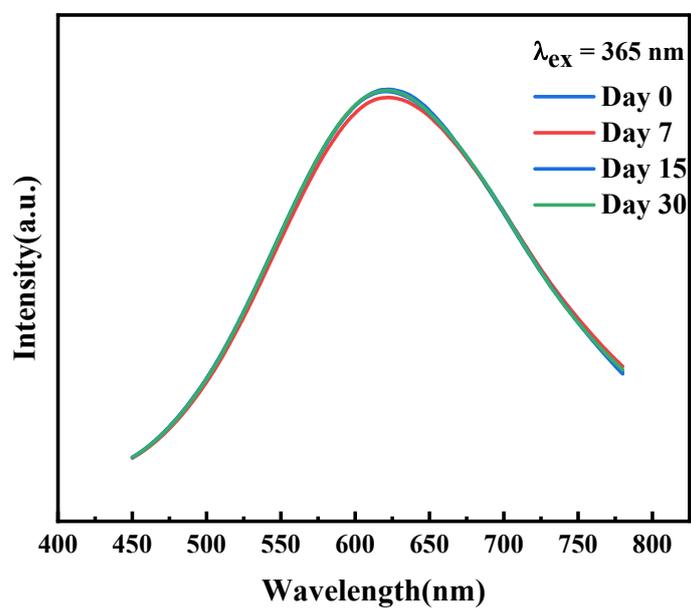


Fig S8 PL spectra for $\text{Cs}_2\text{Rb}_{0.15}\text{Ag}_{0.85}\text{In}_{0.875}\text{Bi}_{0.125}\text{Cl}_6$ after being soaked in ethanol.