## Supplementary Table Captions

ST1: Binding energy $\left(\mathrm{E}_{\mathrm{b}}\right)$ and related bond lengths of $\mathrm{H}_{2}, \mathrm{H}, 2 \mathrm{H}, \mathrm{CO}_{2}, \mathrm{HOCO}, \mathrm{HCOO}$ and HCOOH on $\mathrm{Pd}_{4} \mathrm{Cu}_{4}$ cluster.

ST2: Binding energy $\left(\mathrm{E}_{\mathrm{b}}\right)$ and related bond lengths of $\mathrm{H}_{2}, \mathrm{H}, 2 \mathrm{H}, \mathrm{CO}_{2}, \mathrm{HOCO}, \mathrm{HCOO}$ and HCOOH on $\mathrm{Pd}_{12} \mathrm{Cu}$ cluster.

ST3: Geometric isomers of $\mathrm{Pd}_{4}, \mathrm{Pd}_{\mathrm{m}} \mathrm{Cu}_{\mathrm{n}}(\mathrm{m}+\mathrm{n}=4)$ and $\mathrm{Cu}_{4}$ clusters (Atomic colours: Green $=$ Pd, Brown= Cu )

ST4: Geometric isomers of $\mathrm{Pd}_{8}, \mathrm{Pd}_{\mathrm{m}} \mathrm{Cu}_{\mathrm{n}}(\mathrm{m}+\mathrm{n}=8)$ and $\mathrm{Cu}_{8}$ clusters (Atomic colours: Green $=$ Pd, Brown $=\mathrm{Cu}$ )

ST5: Geometric isomers of $\mathrm{CuPd}_{12}$ cluster (Atomic colours: Green $=\mathrm{Pd}$, Brown $=\mathrm{Cu}$ )
ST6: Total Charge distribution (e/ $/ \AA^{3}$ ) on atoms of $\mathrm{Pd}_{2} \mathrm{Cu}_{2}$ and $\mathrm{CO}_{2}$ adsorbed $\mathrm{Pd}_{2} \mathrm{Cu}_{2}$ cluster $\left(\mathrm{Pd}_{2} \mathrm{Cu}_{2}-\mathrm{CO}_{2}\right)$.

ST7: Total Charge distribution (e/ $/ \AA^{3}$ ) on atoms of $\mathrm{Pd}_{4} \mathrm{Cu}_{4}$ and $\mathrm{CO}_{2}$ adsorbed $\mathrm{Pd}_{4} \mathrm{Cu}_{4}$ cluster $\left(\mathrm{Pd}_{4} \mathrm{Cu}_{4}-\mathrm{CO}_{2}\right)$.

ST8: Total Charge distribution (e/ $/ \AA^{3}$ ) on atoms of $\mathrm{Pd}_{12} \mathrm{Cu}$ and $\mathrm{CO}_{2}$ adsorbed $\mathrm{Pd}_{12} \mathrm{Cu}$ cluster $\left(\mathrm{Pd}_{12} \mathrm{Cu}-\mathrm{CO}_{2}\right)$.

ST9: Lowest energy $\mathrm{Pd}_{4}, \mathrm{Cu}_{4}, \mathrm{Pd}_{8}, \mathrm{Cu}_{8}$ and $\mathrm{Pd}_{\mathrm{m}} \mathrm{Cu}_{\mathrm{n}}(\mathrm{m}+\mathrm{n}=4,8$ and 13) clusters along with their position coordinates (in cartesian). (Atomic colours: Green: Pd , Brown: Cu )

## Supplementary Figure Captions

SF1: Lowest energy structure of $\mathrm{Pd}_{4} \mathrm{Cu}_{4}$ clusters along with their (a) $\mathrm{H}_{2}$ (b) H (c) 2 H (d) $\mathrm{CO}_{2}$ (e) $\mathrm{CO}_{2}+\mathrm{H}$ (f) Carboxyl (g) HCOO (h) $\mathrm{HCOO}+\mathrm{H}$ and (i) HCOOH adsorbed species.

SF2: Lowest energy structure of $\mathrm{Pd}_{12} \mathrm{Cu}$ clusters along with their a) $\mathrm{H}_{2}$ (b) H (c) 2 H (d) $\mathrm{CO}_{2}$ (e) $\mathrm{CO}_{2}+\mathrm{H}$ (f) Carboxyl (g) HCOO (h) $\mathrm{HCOO}+\mathrm{H}$ and (i) HCOOH adsorbed species.

SF3: The potential energy profile for the formation of (a) bidentate formate by the reaction of H -atom and $\mathrm{CO}_{2}$ (b) formic acid by the reaction of H -atom and formate on $\mathrm{Pd}_{4} \mathrm{Cu}_{4}$ cluster.

SF4: The potential energy profile for the formation of (a) bidentate formate by the reaction of H -atom and $\mathrm{CO}_{2}$ (b) formic acid by the reaction of H -atom and formate on $\mathrm{Pd}_{12} \mathrm{Cu}$ cluster.

SF5(a): Total density of states of $\mathrm{Pd}_{2} \mathrm{Cu}_{2}$ cluster along with Pd and Cu in that cluster.
SF5(b): The orbital projected density of states of $\mathrm{Pd}_{2} \mathrm{Cu}_{2}$ cluster.
SF6(a): Total density of states of $\mathrm{Pd}_{2} \mathrm{Cu}_{2} \mathrm{CO}_{2}$ cluster along with its constituent elements.
SF6(b): The orbital projected density of states of $\mathrm{Pd}_{2} \mathrm{Cu}_{2} \mathrm{CO}_{2}$ cluster.

