Supplementary Materials

Acheving High Quantum Capacitance Graphdiyne through Doping and Adsorption

Guangmin Yang, Zhuo Li, Siqi Wang, Jianyan Lin*

College of Physics, Changchun Normal University, Changchun 130032, China

Corresponding author:

*E-mail: linjy994@nenu.edu.cn



Fig. S1 Structure of B/N/P/S doped GDY optimized at the lowest formation energy



Fig. S2 Density of states maps of N atoms with doping concentrations of 1.39%, 2.78%, 4.17% and 5.56% at positions I, II, and III, respectively.



Fig. S3 Density of states maps of P atoms with doping concentrations of 1.39%, 2.78%, 4.17% and 5.56% at positions I, II, and III, respectively.



Fig. S4 Density of states maps of S atoms with doping concentrations of 1.39%, 2.78%, 4.17% and 5.56% at positions I, II, and III, respectively.



Fig. S5 Optimized structure diagram of metal atom adsorbed at V positions



Fig. S6 (a) Density of electronic states of Au, Ag, Cu, Ti, and Al atoms adsorbed at the IV 4

position; C_Q as a function of (b) potential (-0.6~0.6V) and (c) temperature (0~400 K).



Fig. S7 C_Q of Cu adsorbed GDY with different concentrations as a function of (a) potential and (b) temperature.



Fig. S8 Density of electronic states of 4BCu, 4NCu and 3NCu co-doped at I site (a) and II site (e); C_Q as a function of potential at I site (b) and II site (f), the potential range is -0.6~0.6V. Relationship between surface charge density and potential (-0.6~0.6V) at I site (c-d) and II site (g-h).



Fig. S9 Comparison of the C_Q of pristine and B/N/P/S-doped GDY calculated by HSE hybrid functional.



Fig. S10 Comparison of the C_Q of metal-adsorbed GDY calculated by HSE hybrid functional.



Fig. S11 Comparison of the C_Q of transition metal-adsorbed GDY after adding effective U value. The selected U value is 2.0 eV for Au¹, 2.0 eV for Ag², 2.5 eV for Ti², 6.52 eV for Cu³ (Physical Review B, 2013, 87, 115111).

References

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