

Computational investigation of α -SiO₂ surfaces as a support for Pd

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Supplementary Information

S1. Water reference structure molecular dynamics

The calculation of water adsorption energy and the surface energy of hydroxylated surfaces requires a reference state for the water molecule. In this work we compare results using a gas phase water reference and a liquid phase water reference.

For the liquid phase 64 molecules of water were placed in a cubic simulation box with a side length of 12.4 Å with the water molecules placed in a regular 4×4×4 array. This produces a system with the expected density of liquid water of 1.00 g cm⁻³. The periodic water model was first subjected to a high temperature (T=500 K) molecular dynamics (MD) simulation, using an NVE ensemble to maintain the density. For this phase of the calculation the TIP4P^{1,2} potential in the DL-POLY³ package was used with a 1 fs timestep. After 20 ps equilibration (velocity scaling) the simulation switched into the designated ensemble for a further 30 ps. This was sufficient to produce a randomised hydrogen bonded network model. The resulting structure was then used as input for a CASTEP MD simulation run using the NVT ensemble with a Nose-Hoover thermostat. The functional and basis set cut-off were the same as used for the SiO₂ slab calculations (PBE, $E_{cut} = 700$ eV) and TS dispersion corrections were included. The MD calculations were carried out at the Γ -point (k sample (1×1×1)). This DFT MD was run at a simulation temperature of 300 K for a total run time of 3.5 ps.

Figure S1 shows the temperature and potential energy of the system as a function of time over the DFT MD calculation. It can be seen that after an initial sharp drop the calculated potential energy fluctuates around a constant value and the calculated temperature remains

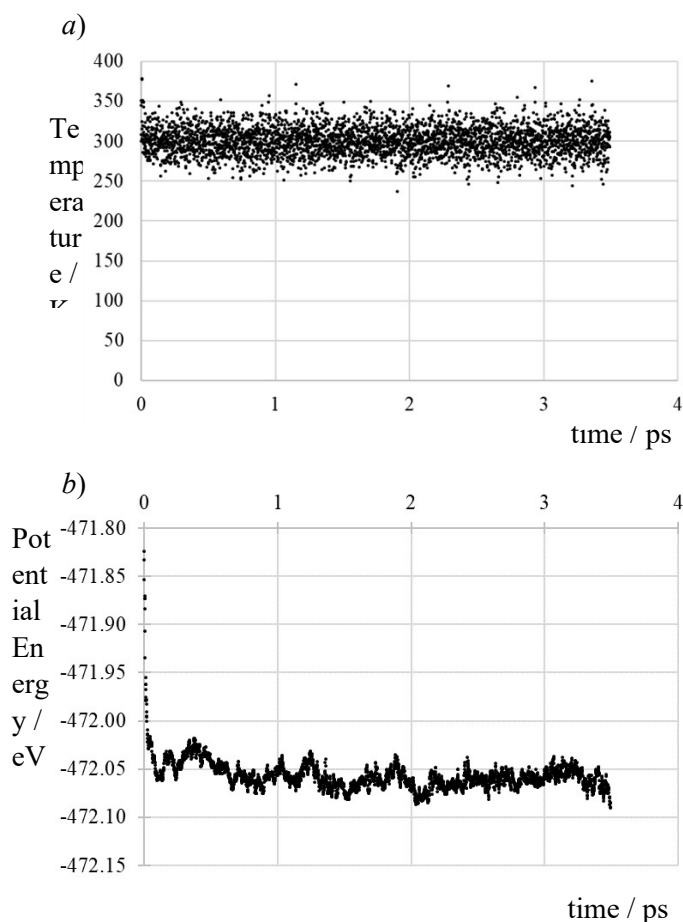


Figure S1. a) Temperature and b) potential energy as a function of time over the DFT molecular dynamics simulation for 64 water molecules at a density of 1.00 g cm^{-3} .

well within $\pm 50 \text{ K}$ of the target temperature. To obtain the water reference energy 14 structures were taken at regular intervals from the trajectory of the simulation between 1 ps and 3.5 ps. Each structure was optimised using a fixed cell procedure and the same convergence criteria as used for simulation of the silica bulk and slab models. This gave liquid water reference energies of $-472.0026 \pm 0.0074 \text{ eV}$ for the PBE functional alone and $-472.1507 \pm 0.0052 \text{ eV}$ when dispersion is included. The error estimates here are the standard deviations calculated from the 14 optimised structure energies.

References

- 1 W. L. Jorgensen, J. Chandrasekhar, J. D. Madura, R. W. Impey and M. L. Klein, Comparison of simple potential functions for simulating liquid water, *The Journal of Chemical Physics*, 1983, **79**, 926–935.
- 2 J. L. F. Abascal and C. Vega, A general purpose model for the condensed phases of water: TIP4P/2005, *The Journal of Chemical Physics*, 2005, **123**, 234505.

31. T. Todorov, W. Smith, K. Trachenko and M. T. Dove, DL_POLY_3: new dimensions in molecular dynamics simulations via massive parallelism, *J. Mater. Chem.*, 2006, **16**, 1911.