

Supporting Information

Interplay of conformational relaxation and hydrogen bond dynamics
in the excited states of fluorescent Schiff base anions

Souradip Dasgupta, Arkaprava Chowdhury, Dipak Kumar Sahoo,
Anindya Datta

Department of Chemistry, Indian Institute of Technology Bombay,
Powai, Mumbai 400 076, India

e-mail: anindya@chem.iitb.ac.in (AD)

Phone: +91 22 2576 7149, Fax: +91 22 2576 7152

Table of Contents

Sr. No.	Content	Description	Page
1.	Figure S1	^1NMR Spectra of the salbn	S1
2.	Figure S2	Steady state spectra of Schiff base anions	S2
3.	Table ST1	Photophysical parameters of Schiff base anions	S3-S4
4.	Figure S3	Quantum yield of Schiff base anions	S4
5.	Table ST2	Temporal parameters of Schiff base anions	S5-S8
6.	Figure S4	Fluorescence decays (in TCSPC) of Schiff base anions	S9
7.	Figure S5	Wavelength dependent fluorescence decays of Schiff base anions	S10
8.	Figure S6	Fluorescence decays in TFE	S11
9.	Figure S7	Time Resolved Emission Spectra of Schiff base anions	S11
10.	Figure S8	Apparent solvent correlation function of Schiff base anions	S12
10.	Table ST3	Time evolution of apparent solvent correlation function	S12
11.	Figure S9	Wavelength dependent decays in polar aprotic solvents	S13
12.	Figure S10	Fluorescence decays of C153 in TFE	S13
13.	Table ST4	. Temporal parameters of C153 in 2,2,2-Trifluoroethanol	S13
13.	Figure S11	Time resolved area normalized emission spectrum of Schiff base anions	S14
14.	Table ST5	Solvent Parameters	S14
14	Table ST6	Time dependent fluorescent Stokes shift of the anions in polar protic solvents	S15
15	Table ST7	Coordinates of ground and excited states of salampy $^-$ obtained from quantum chemical calculations	S16-S17

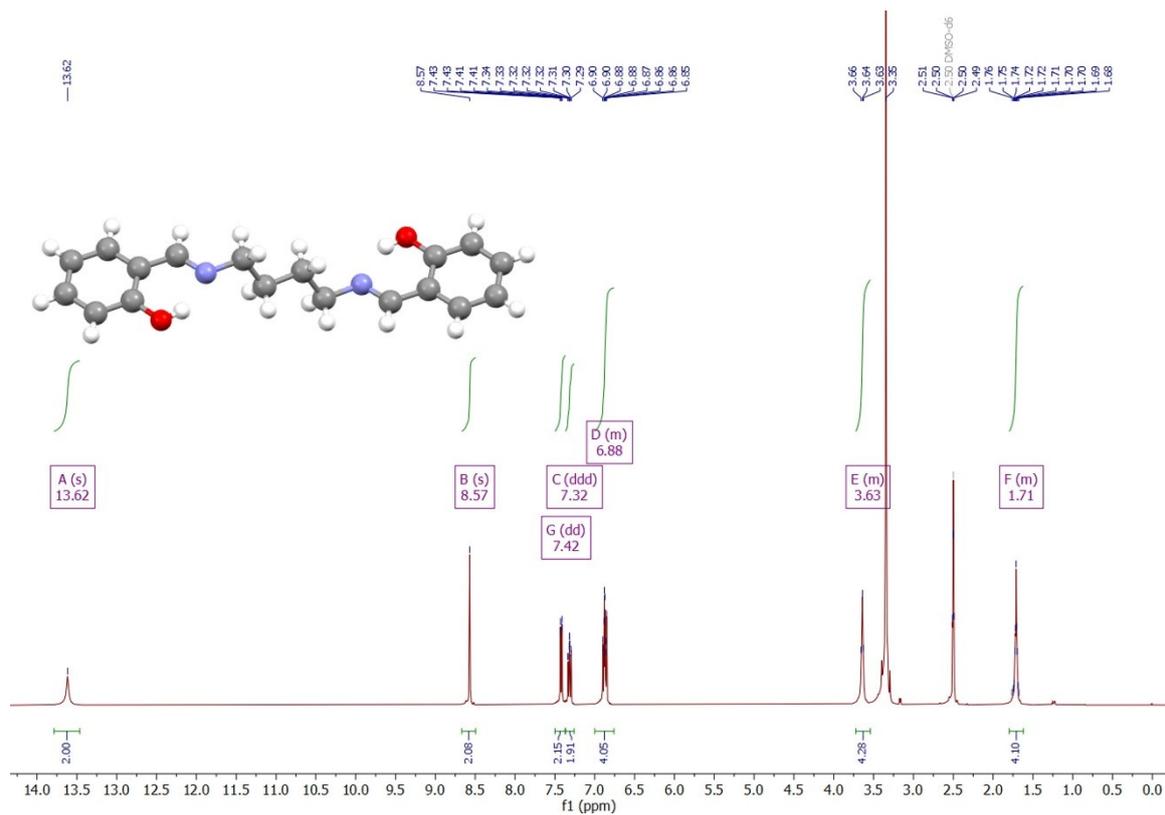


Figure S1. ^1H NMR spectra of salbn in $\text{DMSO-}d_6$. Inset- crystal structure.

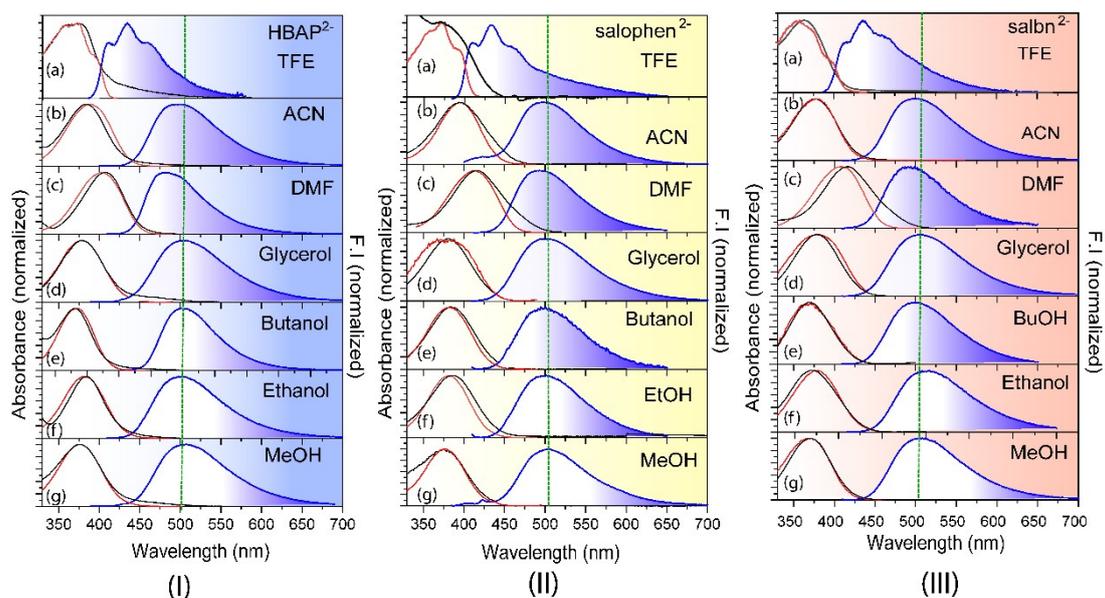


Figure S2. Absorption (black), Fluorescence Emission (blue) and excitation (red) of (I) HBAP²⁻ (II) salophen²⁻ (III) salbn²⁻ in (a) TFE (b) ACN (c) DMF (d) Glycerol (e) BuOH (f) EtOH (g) MeOH. Steady state spectra of salophen²⁻, HBAP²⁻ in MeOH was found to be same as that reported in our earlier reports. (references 33-35 of the main manuscript)

Table ST1. Photophysical parameters of (a) HBAP²⁻ (b) salophen²⁻ (c) salbn²⁻ in different solvents

(a) HBAP²⁻						
Solvent	Δf	$\tilde{\nu}_{abs}(\text{cm}^{-1})$	$\tilde{\nu}_{em}(\text{cm}^{-1})$	$\Delta\tilde{\nu}(\text{cm}^{-1})$	ϕ_f	$\tau_f(\text{ns})$
Glycerol	0.27	26316	19685	6631	0.13	1.5
ACN	0.31	25189	20619	4570	0.08	3.3
MeOH	0.31	26316	19685	6631	0.03	0.9
Acetone	0.28	24691	20619	4072	0.07	2.76
DMF	0.27	24570	20790	3780	0.07	4.2
EtOH	0.29	25974	20000	5974	0.03	1.2
IPA	0.27	25840	20080	5760	0.04	1.9
BuOH	0.24	26110	20040	6070	0.05	2.0
TFE	0.57	26667	23041	3626	0.005	-
(b) salophen²⁻						
Glycerol	0.27	26042	19493	6549	0.15	1.4
ACN	0.31	25381	20833	4548	0.07	3.2
MeOH	0.31	25316	19493	5823	0.01	0.9
Acetone	0.28	24691	20534	4157	0.073	3.0
DMF	0.27	24631	20790	3841	0.077	4.1
EtOH	0.29	25381	19763	5618	0.02	1.4
IPA	0.27	25575	20040	5535	0.05	2.4
BuOH	0.24	26110	19881	6229	0.05	2.1
TFE	0.57	26738	22989	3749	0.003	-
(c) salbn²⁻						
Glycerol	0.27	26247	19608	6639	0.15	2.1
ACN	0.31	25126	20704	4422	0.10	3.8
MeOH	0.31	25510	19569	5941	0.06	0.9
Acetone	0.28	24814	20492	4322	0.09	3.5
DMF	0.27	24691	20747	3944	0.09	3.9
EtOH	0.29	25840	20000	5840	0.06	1.8
IPA	0.27	25707	19920	5787	0.07	2.3
BuOH	0.24	26316	20202	6114	0.08	2.4
TFE	0.57	27778	23095	4683	0.001	-

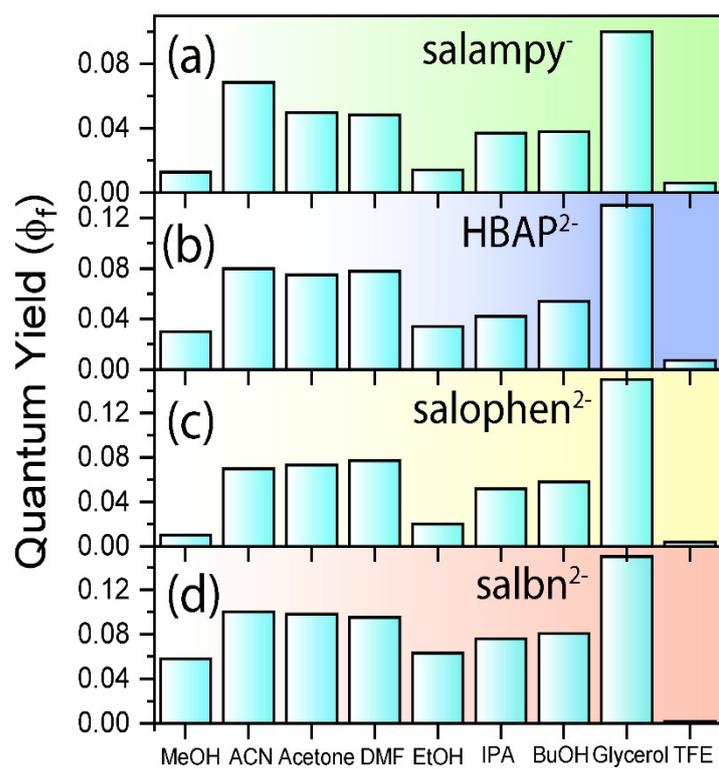


Figure S3. Quantum yields (ϕ_f) of (a) salampy⁻ (b) HBAP²⁻ (c) salophen²⁻ (d) salbn²⁻ in different solvents.

Table ST2. Temporal parameters of salampy⁻, HBAP²⁻, salophen²⁻, salbn²⁻ in different solvents at different wavelengths

(a) salampy⁻		
Solvent	$\lambda_{ex} / \lambda_{em}$ (nm)	τ (ns) ^a
EtOH	400/500	1.0 (1.0)
Glycerol	400/500	1.3 (1.0)
ACN	400/490	3.3 (1.0)
	$\lambda_{ex} / \lambda_{em}$ (nm)	τ (ns) ^a
Glycerol	400/450	0.40 (0.33) 1.23 (0.67)
	400/600	0.80 (-0.2) 1.38 (1.2)
	$\lambda_{ex} / \lambda_{em}$ (nm)	τ (ps) ^a
EtOH	400/450	11.33 (0.41) 76.8 (0.34) 1250 (0.235)
	400/600	27.96 (-0.69) 1350 (1.69)
BuOH	400/450	8.18 (0.35) 60.18(0.38) 678.17 (0.27)
	400/600	128 (-2.03) 270 (0.81) 1800 (2.21)
ACN	400/450	5 (0.70) 1200 (0.3)
	400/600	5.1 (0.67) 1200 (0.33)
TFE	400/450	1.0(0.41) 8.3(0.13) 97.81(0.46)
	400/600	1.3(0.34) 7.7(0.12)

			121.7(0.64)
(b) HBAP²⁻			
Solvent	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ns)^a
EtOH	400/500		1.2 (1.0)
Glycerol	400/500		1.5 (1.0)
ACN	400/490		3.3 (1.0)
	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ns)^a
Glycerol	400/450		0.60 (0.4) 1.1 (0.6)
	400/600		0.70 (-0.18) 1.4 (1.18)
	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ps)^a
EtOH	400/450		8.25 (0.33) 50.75 (0.41) 980 (0.36)
	400/600		50.51 (-0.35) 1300 (1.35)
BuOH	400/450		5.04 (0.35) 37.65 (0.46) 971.46 (0.17)
	400/600		135 (-1.25) 264.19 (1.17) 1870 (1.07)
ACN	400/450		12.3 (0.2) 1400 (0.8)
	400/600		12.3 (0.3) 1400 (0.7)
TFE	400/450		1.0(0.42) 12.2(0.36) 212.7(0.22)
	400/600		3.24(0.25) 20.50(0.59) 159.24(0.16)

(c) salophen²⁻

Solvent	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ns)a
EtOH	400/500		1.4
Glycerol	400/500		1.4
ACN	400/490		3.2
	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ns)a
Glycerol	400/450		0.3 (0.6) 800 (0.3)
	400/600		0.4(-0.2) 1450 (1.2)
	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ps)a
EtOH	400/450		1.22 (0.63) 9.21 (0.10) 303.14 (0.27)
	400/600		23.61 (-0.94) 7.95 (0.01) 1200 (1.92)
BuOH	400/450		3.52 (0.35) 34.23 (0.43) 1118.1 (0.204)
	400/600		42 (-0.42) 1482.2 (1.42)
TFE	400/450		6.06(0.32) 87.19(0.38) 1200(0.28)
	400/600		12.8(0.12) 218.36(0.42) 333.93(0.46)

(d) salbn²⁻

Solvent	$\lambda_{ex} / \lambda_{em}$	(nm)	τ (ns)a
EtOH	400/500		1.8
Glycerol	400/500		2.1
ACN	400/490		3.8

	$\lambda_{ex} / \lambda_{em}$ (nm)	τ (ns) ^a
Glycerol	400/450	0.4 (0.4)
		1.7 (0.6)
	400/600	0.6 (-0.3)
		2340 (1.3)
	$\lambda_{ex} / \lambda_{em}$ (nm)	τ (ps) ^a
EtOH	400/450	3.5 (0.62)
		15.4 (0.23)
		730 (0.15)
	400/600	30.33 (-0.71)
1200 (1.71)		
BuOH	400/450	5.04 (0.35)
		37.65 (0.46)
		971.47 (0.17)
	400/600	135 (-1.25)
264.19 (1.17)		
1870 (1.07)		
ACN	400/450	50 (0.27)
		1100 (0.73)
	400/600	52 (0.30)
		1100 (0.70)
TFE	400/450	4.55(0.38)
		15.9(0.43)
		125.2(0.19)
	400/600	7.57(0.30)
15.07(0.41)		
222.82(0.29)		

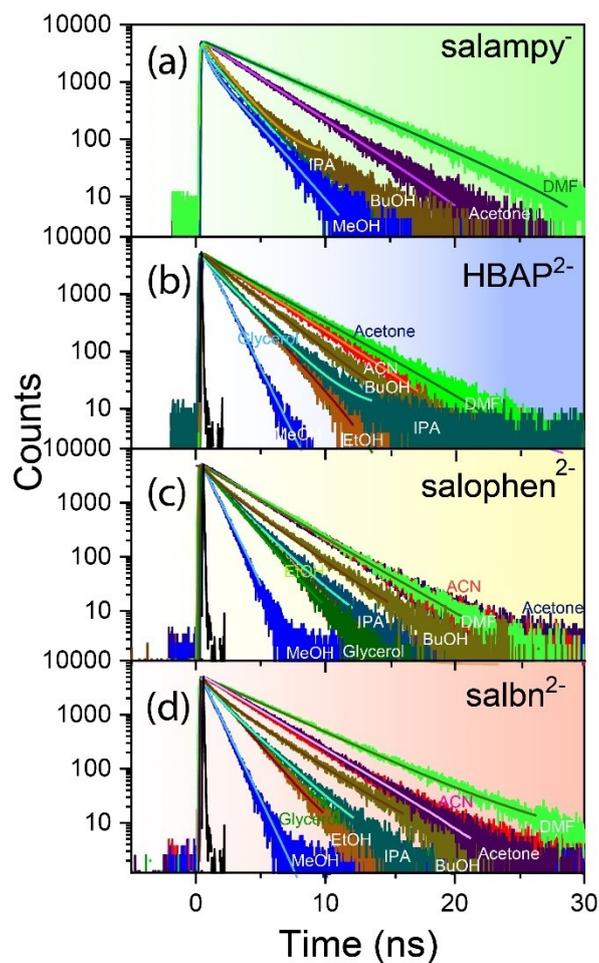


Figure S4. Fluorescence decays of (a) salampy⁻ (b) HBAP²⁻ (c) salophen²⁻ (d) salbn²⁻ recorded at their corresponding emission maxima with $\lambda_{ex} = 400 \text{ nm}$.

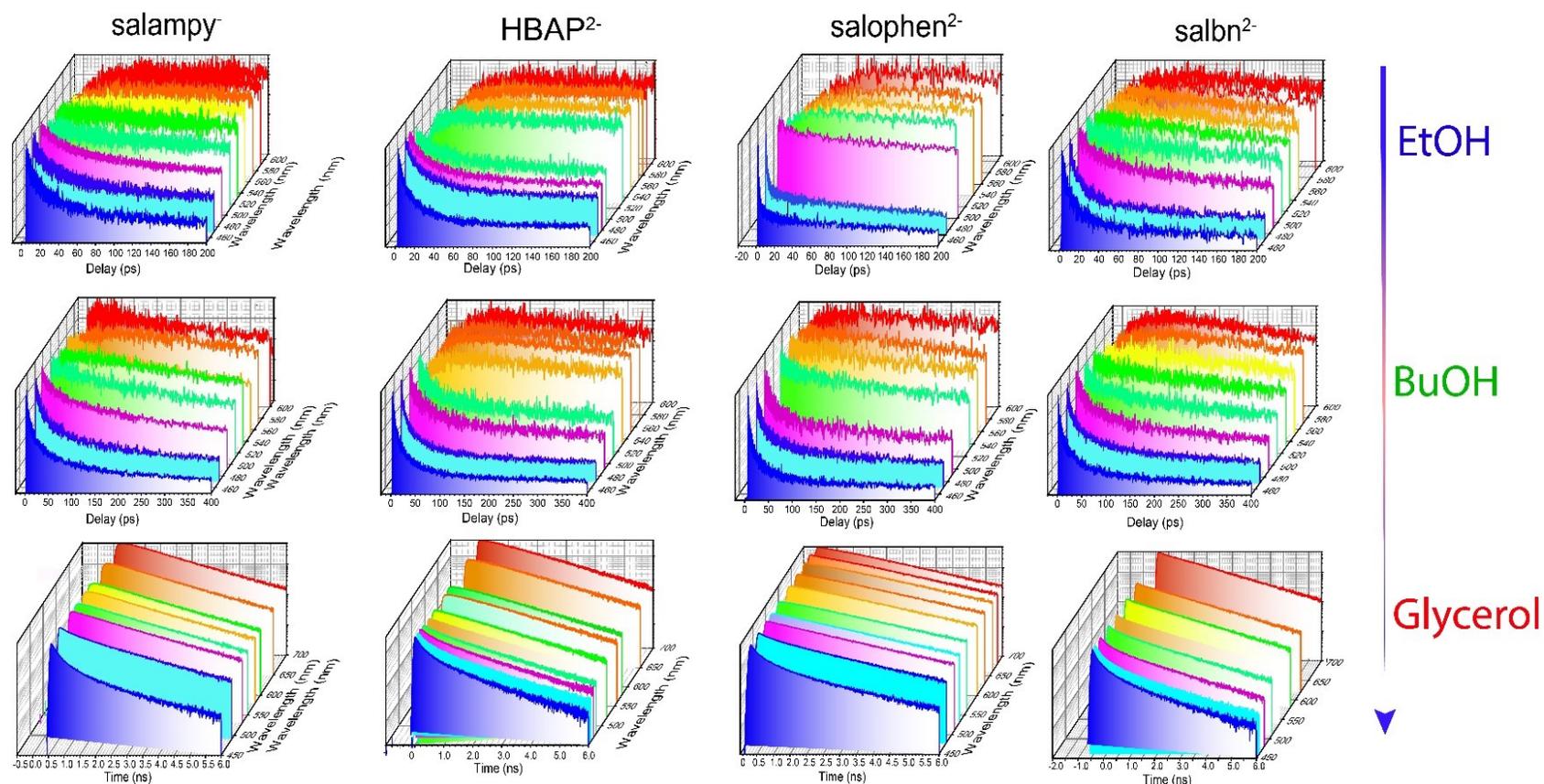


Figure S5. Wavelength dependent fluorescence decays of salampy⁻, HBAP, salophen²⁻, and salbn²⁻ recorded at their corresponding emission maxima with $\lambda_{ex} = 400 \text{ nm}$.

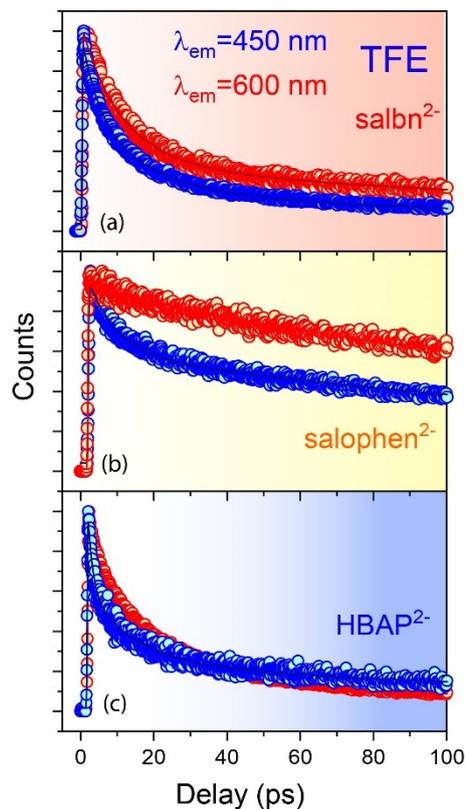


Figure S6. Fluorescence decays of (a) salbn²⁻ (b) salophen²⁻ (c) HBAP²⁻ in TFE

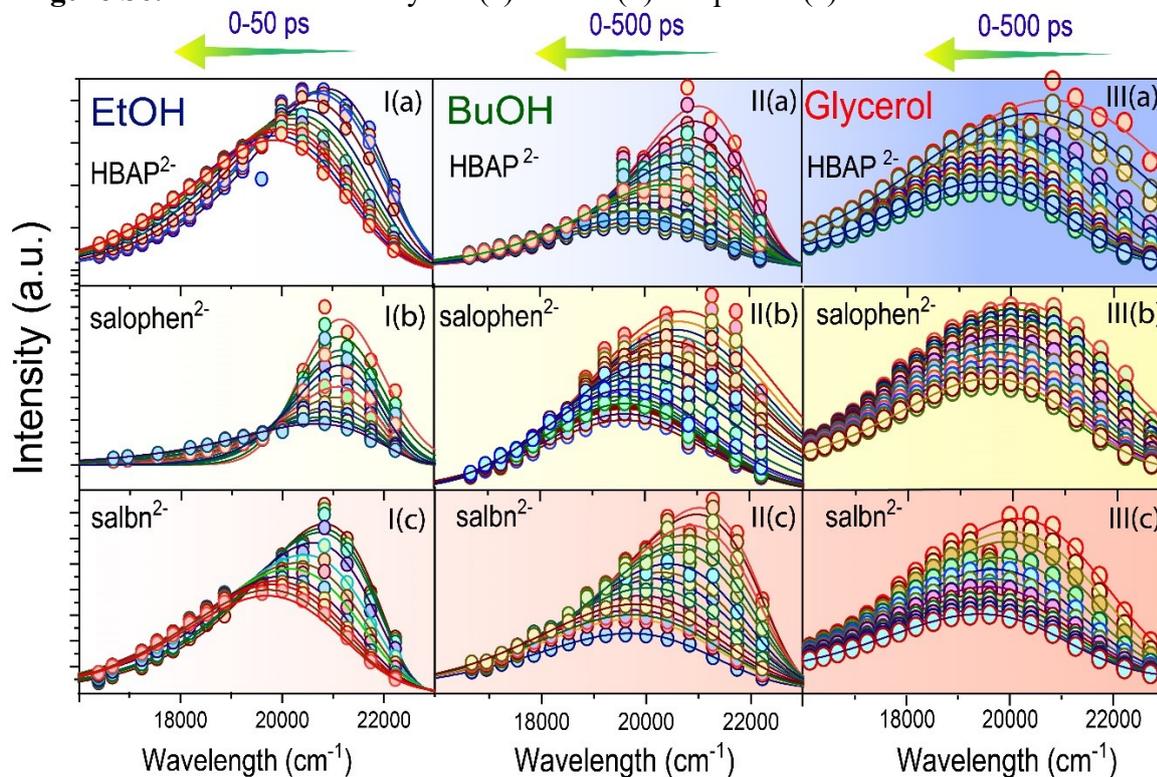


Figure S7. Time Resolved Emission Spectra of (a) HBAP²⁻ (b) salophen²⁻ (c) salbn²⁻ in (I) EtOH (II) BuOH and (III) Glycerol

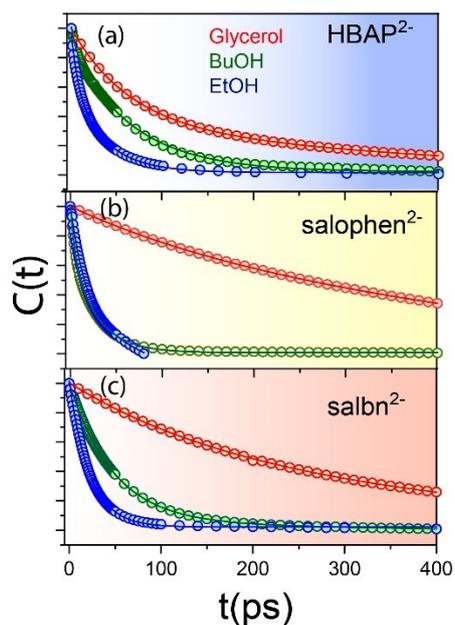


Figure S8. Apparent solvent correlation function $[\tilde{\nu}(t) - \tilde{\nu}(\infty)]$ of (a) HBAP²⁻ (b) salophen²⁻ (c) salbn²⁻

Table ST3. Decay parameters of $[\tilde{\nu}(t) - \tilde{\nu}(\infty)]$ vs $t(\text{ps})$					
	Solvent	$\tau_1(\text{ps})$	a_1	$\tau_2(\text{ps})$	a_2
salampy⁻	EtOH	14.8	0.4	79.8	0.6
	BuOH	15.3	0.4	137.2	0.6
	Glycerol	103	0.3	680	0.7
HBAP²⁻	EtOH	7.5	0.3	36.1	0.7
	BuOH	16.8	0.1	66.5	0.8
	Glycerol	62	0.5	510.0	0.4
salophen²⁻	EtOH	5	0.3	28.0	0.7
	BuOH	4.8	0.3	37.6	0.7
	Glycerol	148.6	0.2	639.6	0.8
salbn²⁻	EtOH	22.3	0.5	55.3	0.5
	BuOH	31.6	0.3	70.3	0.7
	Glycerol	165	0.2	650.1	0.8

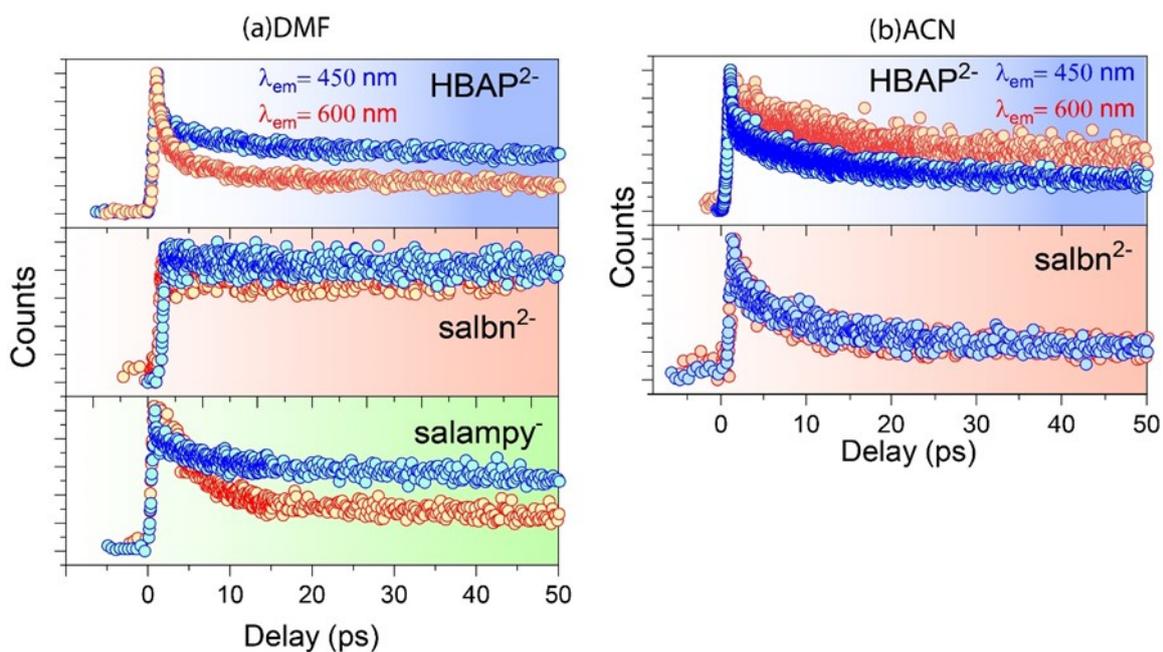


Figure S9. Wavelength dependent fluorescence decays of salampy⁻, salbn²⁻, HBAP²⁻ in (a) DMF and (b) ACN

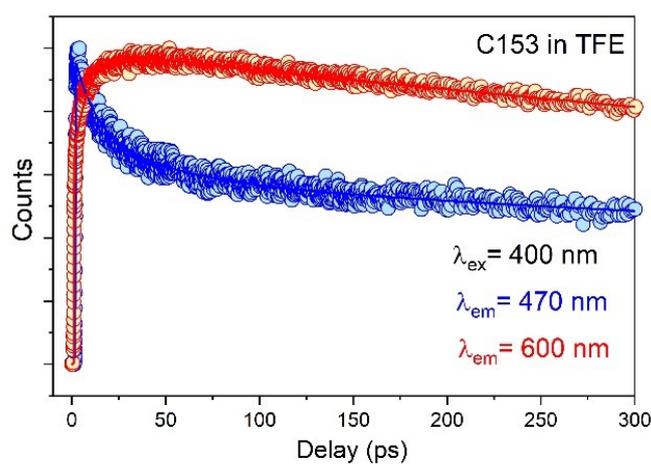


Figure S10. Fluorescence decays of C153 in TFE recorded at $\lambda_{em} = 470 \text{ nm}$ and 600 nm with $\lambda_{ex} = 400 \text{ nm}$

Table ST4. Temporal parameters of C153 in 2,2,2-Trifluoroethanol (TFE)

Wavelength (ex/em) nm	$\tau_1(\text{ps})$	a_1	$\tau_2(\text{ps})$	a_2	$\tau_3(\text{ps})$	a_3
400/470	10.84	0.21	44.7	0.18	1594.3	0.61
400/600	10.0	-0.82	13.5	0.36	1610.7	1.45

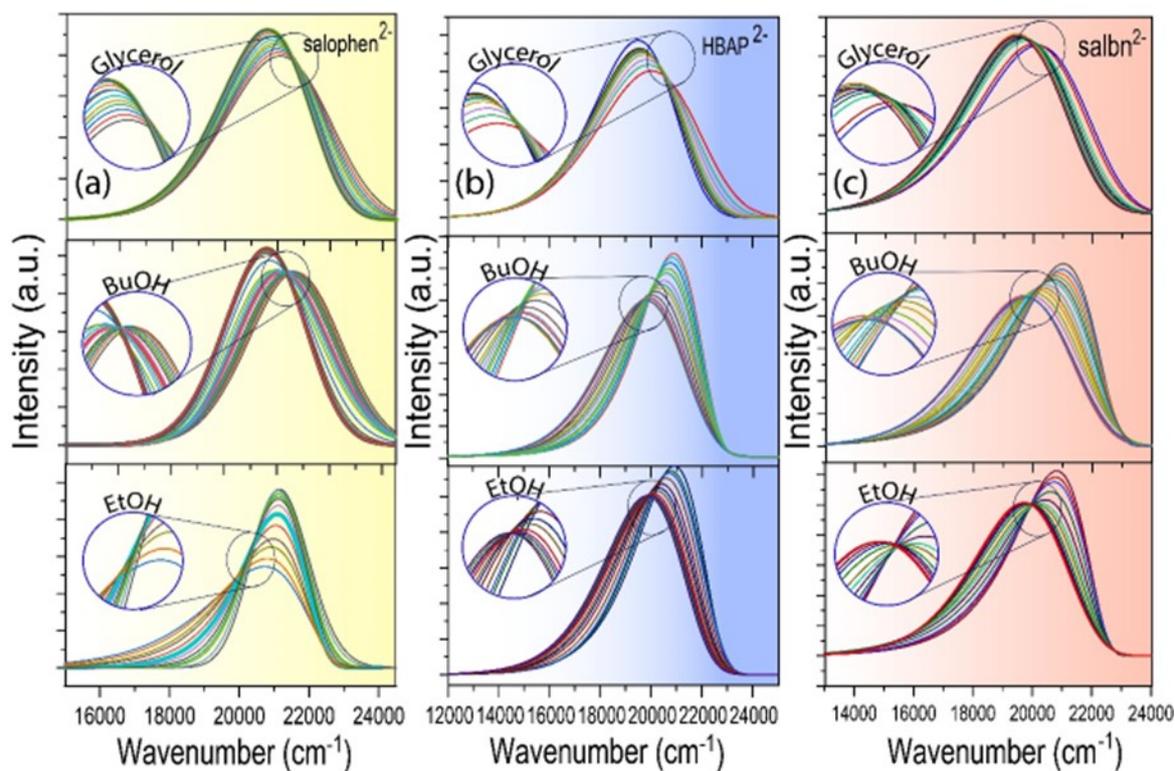


Figure S11. Time Resolved Area Normalized Emission Spectrum (TRANES) of (a) salophen²⁻ (b) HBAP²⁻ (c) salbn²⁻ in EtOH, BuOH and Glycerol (inset shows zoomed in view of the diffused isoemissive point)

Table ST5. Parameters of different solvents

Solvent	Δf	$E_T(30)$	Viscosity (η) cP	Hydrogen bond donating ability (α)
Glycerol	0.265	57	934	1.13
ACN	0.305	45.6	0.37	0.19
MeOH	0.308	55.5	0.54	0.98
Acetone	0.284	42.2	0.31	0.08
DMF	0.272	43.8	0.79	0.00
EtOH	0.288	51.9	1.07	0.86
IPA	0.272	48.6	2.04	0.76
BuOH	0.239	50.2	2.54	0.84
TFE	0.57	59.8	0.90	1.51

Table ST6. Time Dependent Fluorescent Stokes Shift of the anions in polar protic solvents

Sample	Solvent	$\tilde{\nu}(t)$ (cm ⁻¹)	$\tilde{\nu}(\infty)$ (cm ⁻¹)	$[\tilde{\nu}(t) - \tilde{\nu}(\infty)]$ (cm ⁻¹)
salampy⁻	EtOH	20850	19850	1000
	BuOH	21600	20050	1550
	Glycerol	20760	19600	1160
HBAP²⁻	EtOH	20900	20000	900
	BuOH	21000	19900	1100
	Glycerol	20850	19600	1250
salophen²⁻	EtOH	21200	20500	700
	BuOH	20801	19700	1100
	Glycerol	20200	19600	600
salbn²⁻	EtOH	20850	19800	1050
	BuOH	21100	19900	1200
	Glycerol	20300	19550	750

Table ST7. Optimized coordinates of Ground State and excited states of salampy

Ground State			
Atom	x	y	z
N	-0.2018	0.86143	0.19426
N	-3.78807	0.5568	-0.49995
C	2.7883	0.60937	-0.36864
C	2.00622	1.82467	-0.27968
C	-2.45414	0.37257	-0.44797
C	-1.57817	1.18618	0.45357
H	-1.8431	2.25247	0.33867
H	-1.84107	0.91821	1.48897
C	4.74972	2.04039	-0.81091
H	5.81594	2.12068	-1.01782
C	4.19299	0.79988	-0.64403
H	4.79071	-0.10376	-0.71722
C	-2.04996	-0.6309	-1.30029
C	-4.10475	-1.19392	-2.15923
H	-4.78648	-1.75458	-2.78947
C	3.97301	3.21316	-0.72143
H	4.42132	4.19261	-0.8544
C	0.58929	1.84982	-0.0055
H	0.17431	2.8762	0.03097
C	-2.72889	-1.34388	-2.0662
C	-4.56301	-0.17874	-1.30497
H	-5.62424	0.05573	-1.27145
C	2.6287	3.07517	-0.45941
H	2.00496	3.96592	-0.38401
O	2.29635	-0.54675	-0.2194
C	4.34497	-2.54975	1.1904
H	3.49215	-2.97032	1.75307
H	4.54231	-1.55865	1.63201
O	4.07355	-2.46006	-0.17688
H	3.35391	-1.76603	-0.27894
C	-4.73334	2.43131	2.32989
H	-3.86389	1.85291	2.68101
H	-5.61197	1.77224	2.4371
O	-4.56697	2.85973	1.00213
H	-4.37471	2.07369	0.4507
C	-4.91207	3.65454	3.20815
H	-5.78192	4.23005	2.87986
H	-5.05352	3.37229	4.25576
H	-4.03333	4.3002	3.13492
C	5.55903	-3.44063	1.39395
H	5.80284	-3.55325	2.456
H	5.37202	-4.43259	0.97174
H	6.42625	-3.01739	0.87867
Excited State			
N	-0.8403	-0.59463	-0.27617
N	-3.37919	1.12486	-0.74681

C	2.21376	0.03532	0.14914
C	1.00632	0.83528	0.26776
C	-3.18284	-0.27604	-0.88999
C	-2.22902	-0.97543	0.00264
H	-2.45793	-0.78289	1.07047
H	-2.3147	-2.05249	-0.16577
C	3.47683	2.07664	-0.32836
H	4.40852	2.58039	-0.56454
C	3.44827	0.71779	-0.17568
H	4.33369	0.09824	-0.2676
C	-3.95852	-0.75133	-1.95338
C	-4.98392	1.25303	-2.60872
H	-5.63788	1.8657	-3.21258
C	2.29773	2.82791	-0.17254
H	2.32946	3.90515	-0.28968
C	-0.31348	0.20416	0.54299
H	-0.82512	0.54545	1.45318
C	-4.74741	-0.15686	-2.70059
C	-4.20803	1.77284	-1.55758
H	-4.27603	2.84425	-1.3639
C	1.08175	2.20385	0.12834
H	0.17051	2.78273	0.2485
O	2.203	-1.20591	0.35245
C	5.09361	-2.27702	1.65079
H	4.54918	-3.09031	2.15634
H	4.81802	-1.34511	2.16961
O	4.76921	-2.21789	0.28315
H	3.81721	-1.99867	0.21836
C	-2.6674	2.91591	1.9986
H	-2.03077	3.48036	2.69363
H	-2.75537	1.89673	2.4111
O	-2.03656	2.91113	0.74814
H	-2.49069	2.2092	0.18009
C	-4.05362	3.54819	1.94893
H	-4.70088	2.9711	1.28442
H	-4.51541	3.58189	2.94258
H	-3.98782	4.56873	1.56005
C	6.58643	-2.5095	1.78017
H	6.88701	-2.57318	2.83022
H	6.86696	-3.43959	1.27887
H	7.13887	-1.69294	1.30732