

Supporting Information

Collaboratively Hydrothermal and Calcination Fabrication of ZnOS Heterostructures for Visible-Light-Driven H₂ Production

XueLi Chen^{a*}, ChenXiang Lin^b, LanHai Lai^a, MingRui Liu^c, Kai Zheng^a, SongTao Li^a, HaiTao Li^{a*}

a Jiangxi Provincial Key Laboratory of Low-Carbon Solid Waste Recycling, School of Geography and Environmental Engineering, Gannan Normal University, Ganzhou 341000, P. R. China.

b Guangxi Key Laboratory of Natural Polymer Chemistry and Physics, Nanning Normal University, Nanning 530001, China.

c Foshan Xianhu Laboratory of the Advanced Energy Science and Technology Guangdong Laboratory, Foshan 528216, PR China.

***Corresponding Author:** Xueli Chen (e-mail: xueli089@foxmail.com); Haitao Li (e-mail: lihaitao103904@126.com).

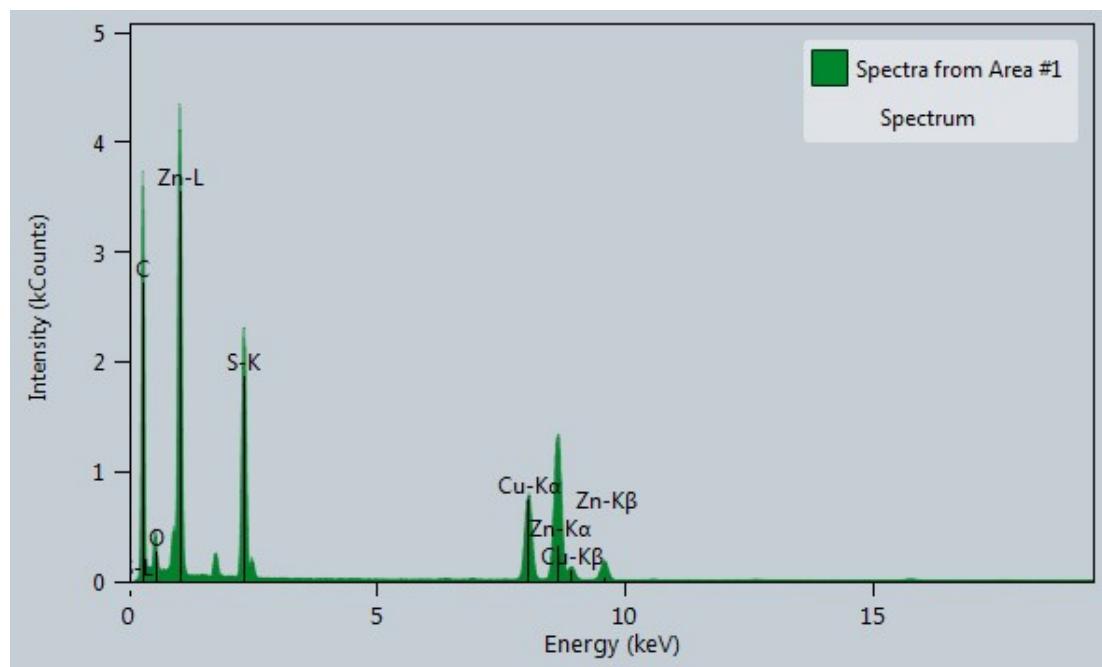


Figure S1. SEM spectra of ZnOS-3.

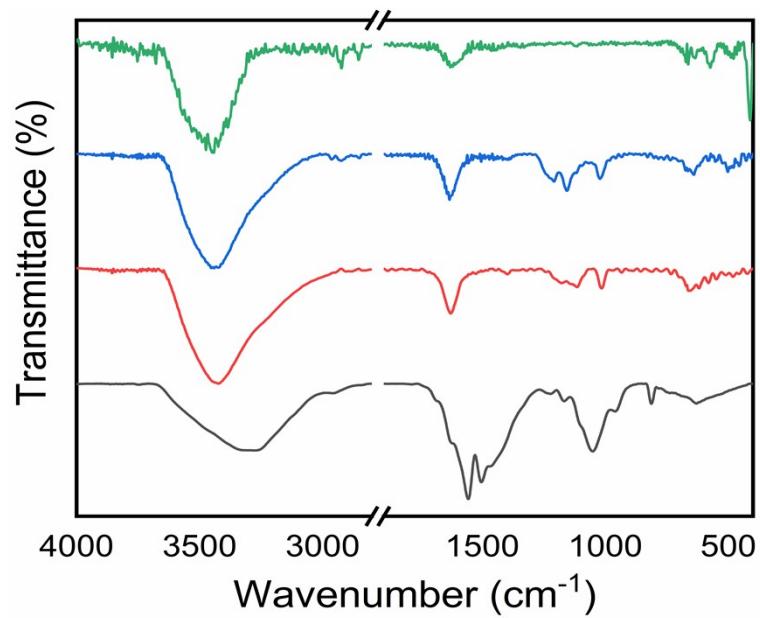


Figure S2. FT-IR spectra of ZnOS-X.

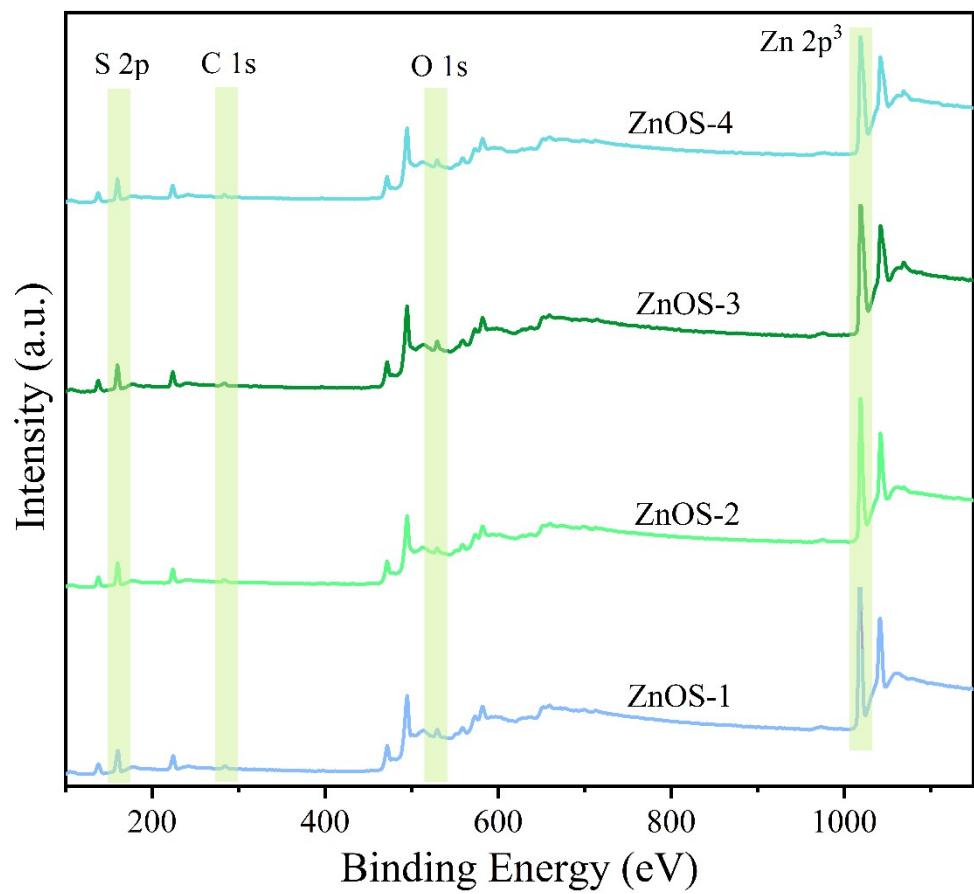


Figure S3. XPS survey spectrum of ZnOS-x.

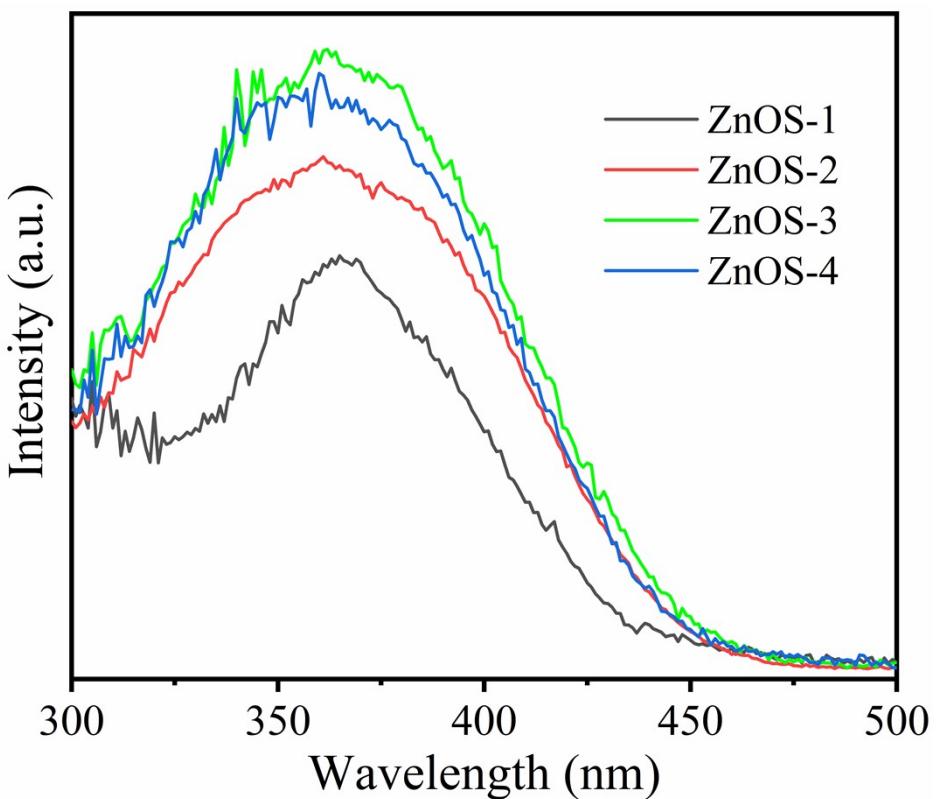


Figure S4. UV-vis diffuse-reflectance spectra of ZnOS-x.

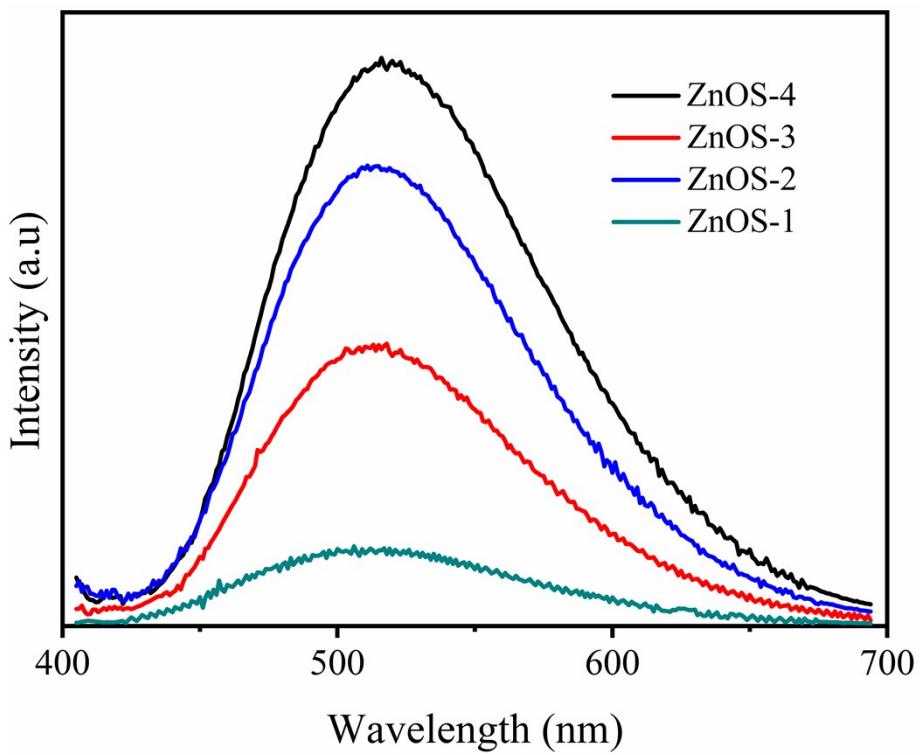


Figure S5. Photoluminescence emission spectra (excitation wavelength = 360nm) of ZnOS-x.

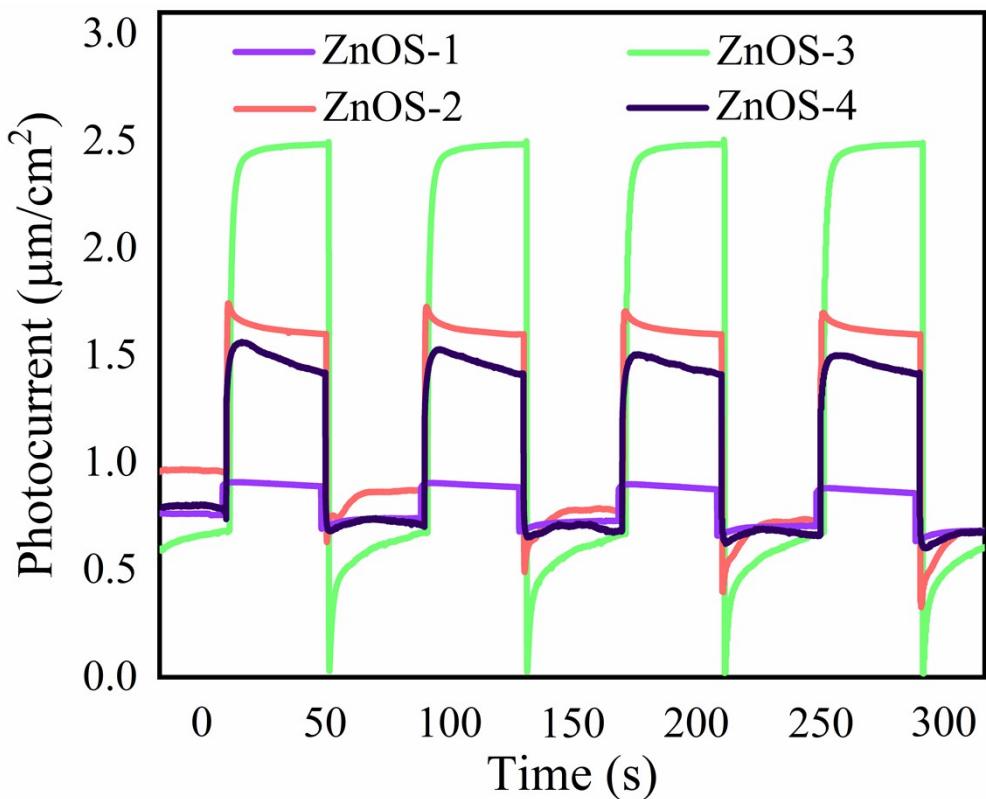


Figure S6. photocurrent-time dependence of ZnOS-x.

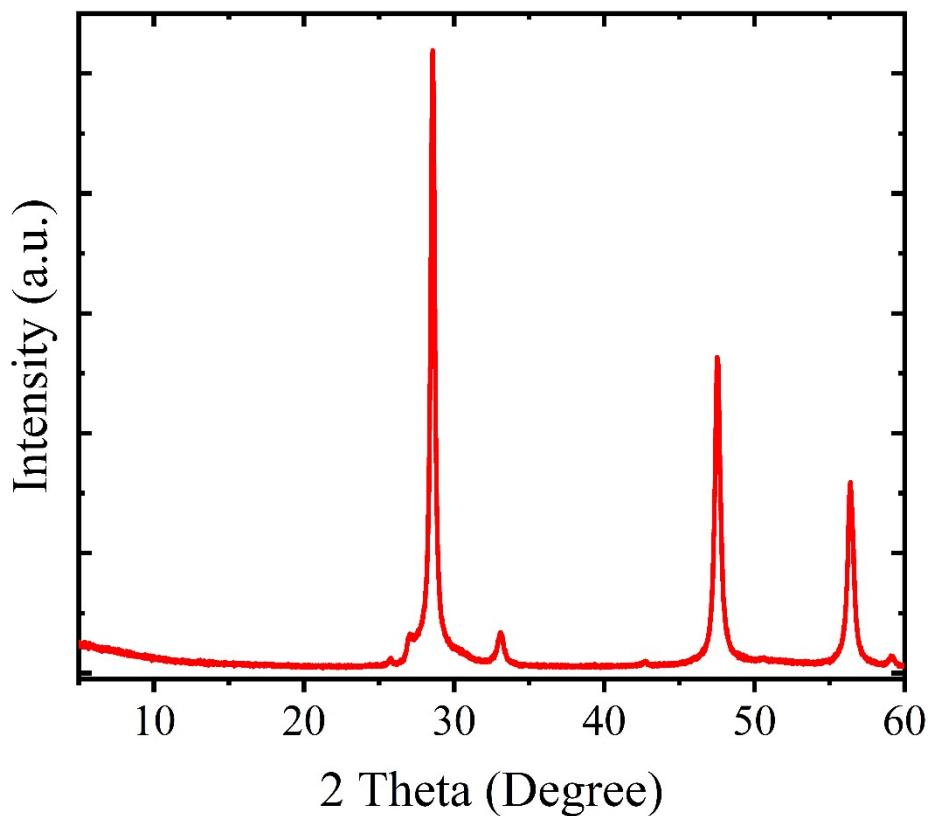


Figure S7. PXRD patterns of ZnOS-3 after photocatalytic H₂ evolution reaction.

Table S1 Specific surface area of the ZnOS-x.

	S_{BET} (m ² /g)	R ²	C
ZnOS-1	17	0.99994	87.7
ZnOS-2	19.2	0.99994	83.693
ZnOS-3	20.2	0.99995	75.192
ZnOS-4	4	0.99959	101.163

Table S2 ICP-MS results for the molar ratio of of the ZnOS-3 before and after photocatalytic reaction.

	before	after
ZnOS-3 (ppm)	2.56	2.18