Catalyst	Space group	Lattice Parameter (Å)			Cell volume
		a	b	c	(Å <sup>3</sup> )
CoMnO <sub>x</sub> -200	I41/amd	8.097	8.097	9.266	607.5
CoMnO <sub>x</sub> -300	I41/amd	8.096	8.096	9.242	605.7
CoMnO <sub>x</sub> -400	I41/amd	8.096	8.096	9.243	605.9
CoMnO <sub>x</sub> -500	I41/amd	8.097	8.097	9.254	606.7
CuCoMnO <sub>x</sub> -200	Fd-3m	8.169			545.2
CuCoMnO <sub>x</sub> -300	Fd-3m	8.160			543.5
CuCoMnO <sub>x</sub> -400	Fd-3m	8.176			546.7
CuCoMnO <sub>x</sub> -500	Fd-3m	8.189			549.3

## Table S1. Lattice parameter and cell volume of CoMnO<sub>x</sub>-T and CuCoMnO<sub>x</sub>-T catalyst

Catalyst	θ	hkl	FWHM	D/Å
CoMnO <sub>x</sub> -200	36.23	311	0.934	89
CuCoMnO <sub>x</sub> -200	36.47	311	0.957	86
CoMnO <sub>x</sub> -300	36.43	311	1.035	80
CuCoMnO <sub>x</sub> -300	36.39	311	1.111	74
CoMnO <sub>x</sub> -400	36.36	311	0.952	87
CuCoMnO <sub>x</sub> -400	36.34	311	0.959	86
CoMnO <sub>x</sub> -500	36.39	311	0.411	201
CuCoMnO <sub>x</sub> -500	36.34	311	0.464	178

Table S2. Grain size of CoMnOx-T and CuCoMnOx-T calculated by XRD

Catalyst	$S_{BET}^{}(m^2g^{-1})$	V <sub>Pore</sub> (cm <sup>3</sup> g <sup>-1</sup> )
CoMnO <sub>x</sub> -200	75	0.246
CoMnO <sub>x</sub> -300	109	0.267
CoMnO <sub>x</sub> -400	63	0.186
CoMnO <sub>x</sub> -500	30	0.210
CuCoMnO <sub>x</sub> -200	49	0.109
CuCoMnO <sub>x</sub> -300	79	0.157
CuCoMnO <sub>x</sub> -400	53	0.157
CuCoMnO <sub>x</sub> -500	12	0.090

Table S3. Specific surface area and pore volume of  $CoMnO_x$ -T and  $CuCoMnO_x$ -T catalysts

Catalyst	Total H <sub>2</sub> cons. (mmol/g <sub>cat</sub> )		
CoMnO <sub>x</sub> -200	7.83		
CoMnO <sub>x</sub> -300	11.05		
CoMnO <sub>x</sub> -400	11.26		
CoMnO <sub>x</sub> -500	8.94		
CuCoMnO <sub>x</sub> -200	11.03		
CuCoMnO <sub>x</sub> -300	11.64		
CuCoMnO <sub>x</sub> -400	11.42		
CuCoMnO <sub>x</sub> -500	10.94		

Table S4. The  $H_2$  consumption of CoMnO<sub>x</sub>-T and CuCoMnO<sub>x</sub>-T catalyst calculated by  $H_2$ -TPR



Figure S1 TEM images of CuCoMnO<sub>x</sub>-200 (a), CuCoMnO<sub>x</sub>-300 (b), CuCoMnO<sub>x</sub>-400 (c)and

 $CuCoMnO_x$ -500 (d)



Figure S2. CO conversion (a,c) and temperature (b,d) curves of  $CoMnO_x$ -T and  $CuCoMnO_x$ -T in CO-PROX at 250 mW/cm<sup>2</sup>



Figure S3 CO conversion and  $O_2$  selectivity of CuCoMnO<sub>x</sub>-300 for photothermal CO-PROX



Figure S4. CO conversion of thermal catalysis over CoMnO<sub>x</sub>-T (a) and CuCoMnO<sub>x</sub>-T (b)



Figure S5. comparation of CO conversion of the  $CoMnO_x$  and  $CuCoMnO_x$  catalysts for thermal

and photothermal CO-PROX



Figure S6. CO conversion and temperature curve of CuCoMnO<sub>x</sub>-300 in CO-PROX at 250mW/cm<sup>2</sup> illuminated under different optical composition



Figure S7. CO conversions of pure  $Co_3O_4$  and  $Mn_2O_3$ , binery  $CoMnO_x$  and ternary  $CuCoMnO_x$  calcined at 300 °C (a) and CO conversions of  $CuCoMnO_x$ -300 with different copper amount in photothermal CO-PROX at 250mW/cm<sup>2</sup>



**Figure S8.** CO conversion of CuCoMnO<sub>x</sub>-300 in CO-PROX at different illumination power densities (a) and at different weight hourly space velocity (b) at 250 mW/cm<sup>2</sup> (b)



Figure S9. PAE values of  $CuCoMnO_x$ -300 catalyst under different illumination densities.



Figure S10. UV-Vis-IR absorption spectrums of (a) CoMnO<sub>x</sub>-T, (b) CuCoMnO<sub>x</sub>-T.



Figure S11. XPS spectra of C 1s (A) and Mn 3s (B) of  $CoMnO_x$ -300 (a),  $CuCoMnO_x$ -200 (b),  $CuCoMnO_x$ -300 (c) ,  $CuCoMnO_x$ -400 (d) and  $CuCoMnO_x$ -500 (e).



Figure S12. DRIFTS spectra (1900-1200 cm<sup>-1</sup>) of  $CoMnO_x$ -300 (a) and  $CuCoMnO_x$ -300 (b) under the reaction stream at different temperature.



Figure S13. DRIFTS spectra ( $3800-2200 \text{ cm}^{-1}$ ) of CoMnO<sub>x</sub>-300 (a) and CuCoMnO<sub>x</sub>-300 (b).