

Supporting Information

Visible-light-driven sustainable conversion of carbon dioxide to methanol using a metal-free covalent organic framework as a recyclable photocatalyst

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Instrumentation:

Absorption spectroscopy: UV-Vis absorption spectra of the catalyst was recorded on SHIMADZU, UV-2600 UV-Vis spectrometer with a standard 1 cm x 1 cm cuvette.

Gas Chromatography: Gas Chromatography spectra of the reaction mixture was recorded on VARIAN 430-GC

IR Spectra: The FTIR spectra of the materials were recorded from a Perkin-Elmer spectrophotometer (FT-IR 783) on KBr pellets.

Fluorescence Spectroscopy: The Fluorescence Emission spectra was recorded by using Horiba Fluoro Max 4 spectrometer.

PXRD: The PXRD analysis was performed by using an X-ray diffractometer (BRUKER, Powder X-Ray eco D8 ADVANCE) equipped with Ni-filtered Cu K α ($\lambda = 0.15406$ nm) radiation.

SEM: FESEM images of the catalyst were acquired by using Scanning Electron Microscope (SEM) [JEOL JSM IT 300], was done to know about the morphological information of the sample.

TEM: Transmission Electron Microscope (TEM) [JEOL JEM 2100] was used obtain the morphological information of the sample.

BET: The N₂ adsorption-desorption analysis of TFPG-DAAQ COF sample was conducted by using a BET Surface Analyzer [QUANTACHROME ASIQC602-5].

GC/MS: The detection of methanol in the reaction mixture was conducted by Agilent 7000D Triple Quadrupole GC/MS.

Chemicals

4-aminobenzonitrile, p-toluenesulphonic acid (PTSA), and 4-4'-biphenyldicarbaldehyde (BDC) were received from Sigma Aldrich. Trifluoromethanesulfonic acid was also purchased from Sigma-Aldrich. 4,4,4-(1,3,5-Triazine-2,4,6-triyl)-trianiline (TPAT) was prepared by conducting the trimerization reaction of 4-aminobenzonitrile with the assistance of the super acid catalyst, trifluoromethanesulfonic acid. Nickel chloride hexahydrate, hydrazine monohydrate were obtained from Merck, India. Organic solvents such as acetonitrile, acetone, N, N-dimethylacetamide (DMAc) and ethylene glycol were received from Spectrochem, India, and used without further purification. All the reactions were carried out using oven-dried glassware.

Reaction set up:



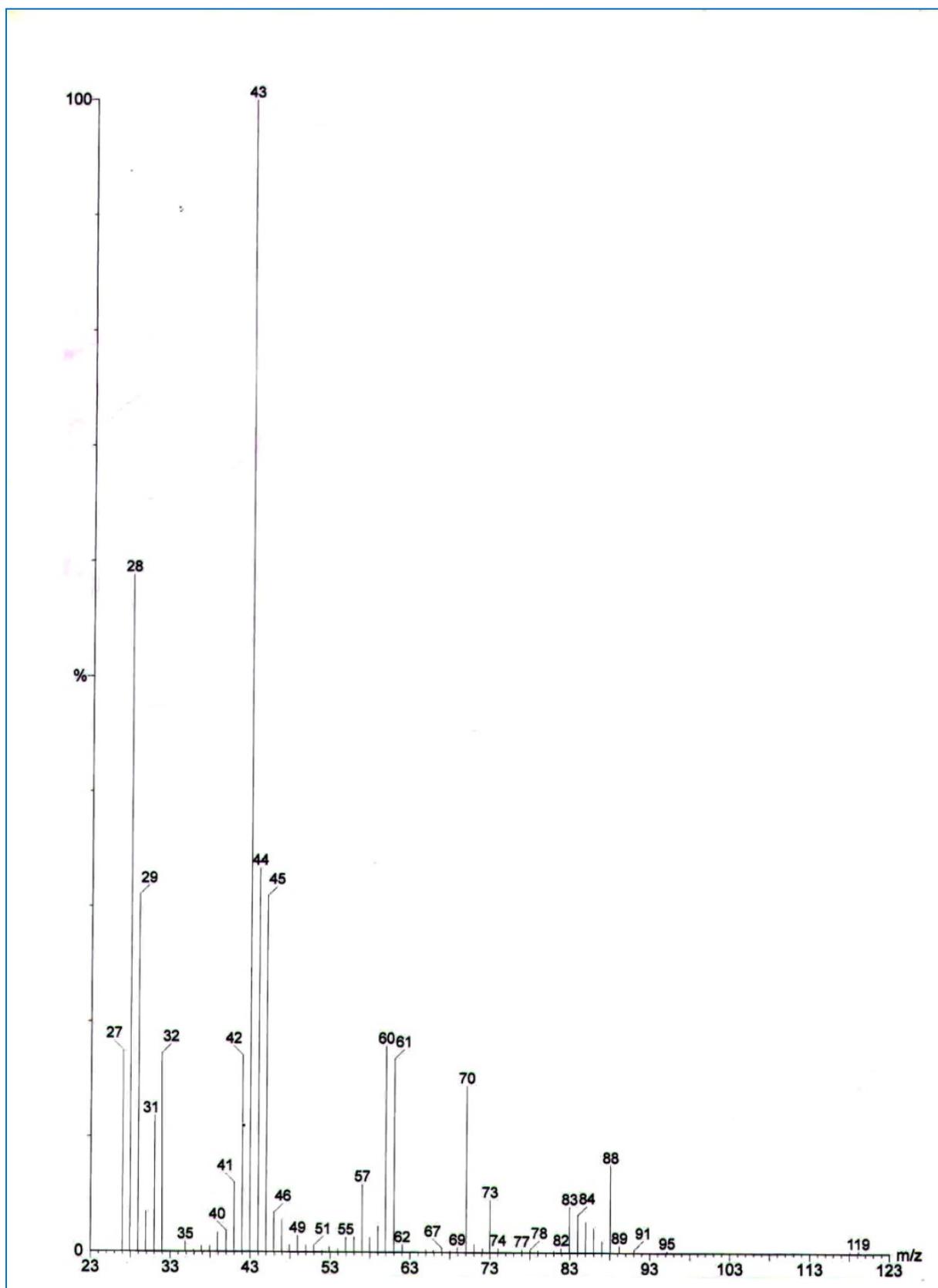
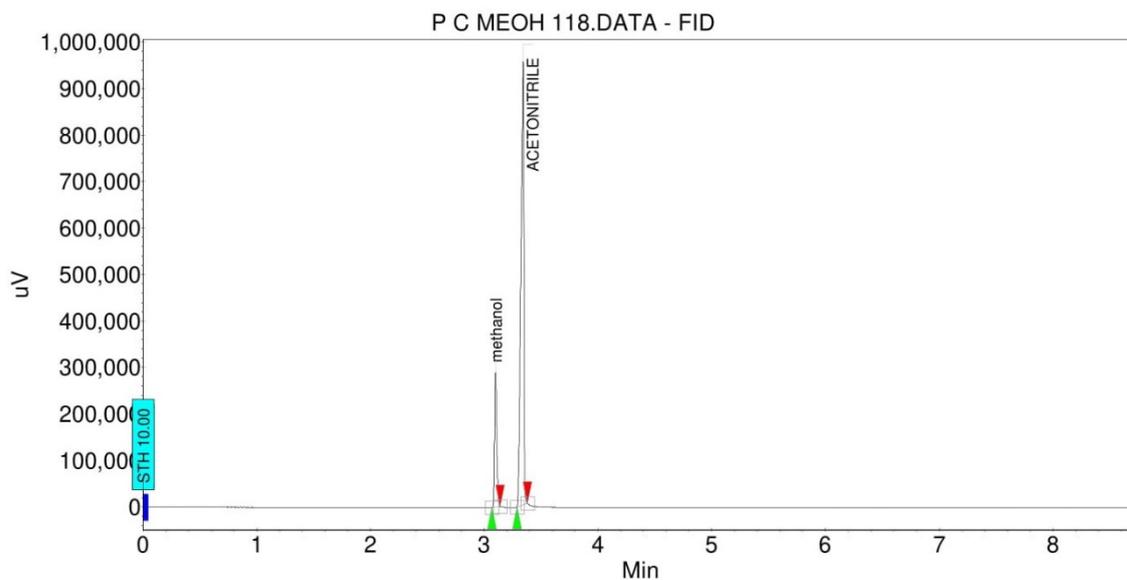


Figure S1. GC MS spectra of the reaction mixture.(methanol formation at optimised reaction condition)



Peak results :

Index	Name	Time [Min]	Quantity [M]	Height [uV]	Area [uV.Min]	Area % [%]	Area [uV.Sec]	Quantity [M]
1	methanol	3.10	0.30	289159.2	6580.8	18.727	394845.8	0.30
2	ACETONITRILE	3.34	0.48	953322.7	28558.8	81.273	1713529.7	0.48
Total			0.78	1242482.0	35139.6	100.000	2108375.5	0.78

Figure S2. GC spectrum of reaction mixture.(methanol formation at optimised reaction condition)

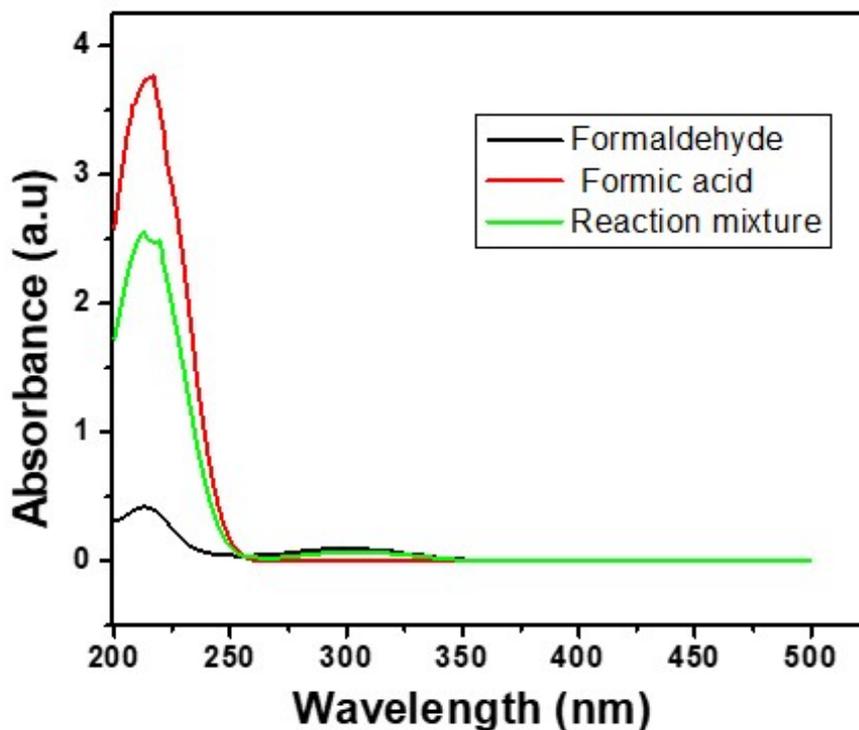


Figure S3.UV-vis spectra of reaction mixture, pure formic acid and pure formaldehyde

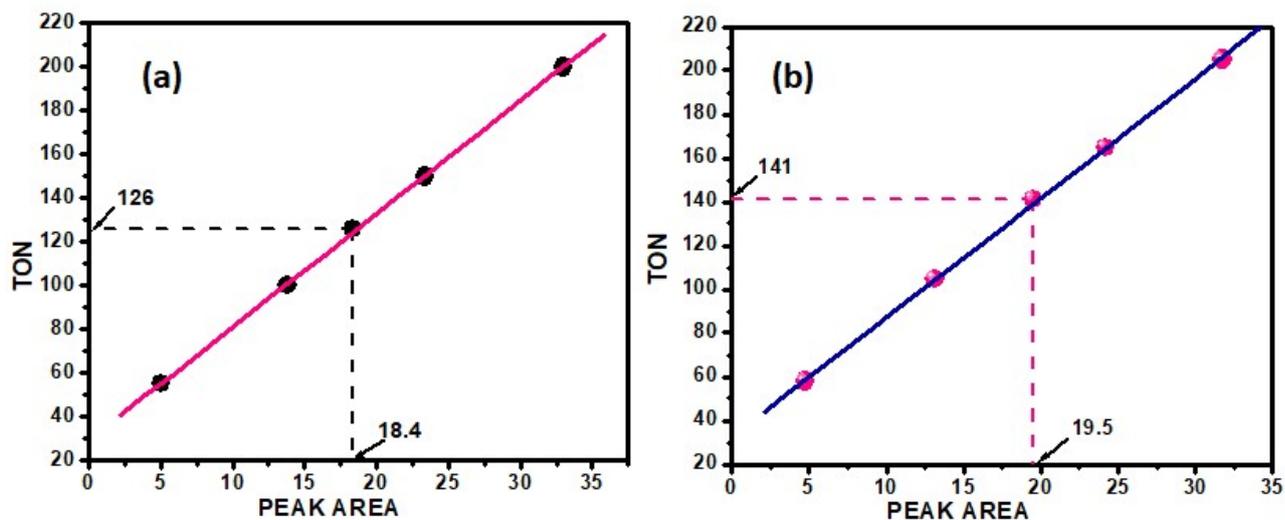


Figure S4. Calibration curve of methanol for determination of concentration of methanol produced. (a) Calibration curve of methanol in presence of COF catalyst (TRITER-2 was synthesized through hydrothermal process) (b) Calibration curve of methanol in presence of COF photocatalyst (TRITER-2 was constructed under solvothermal conditions: 6M AcOH in mesitylene:dioxane=9:1)

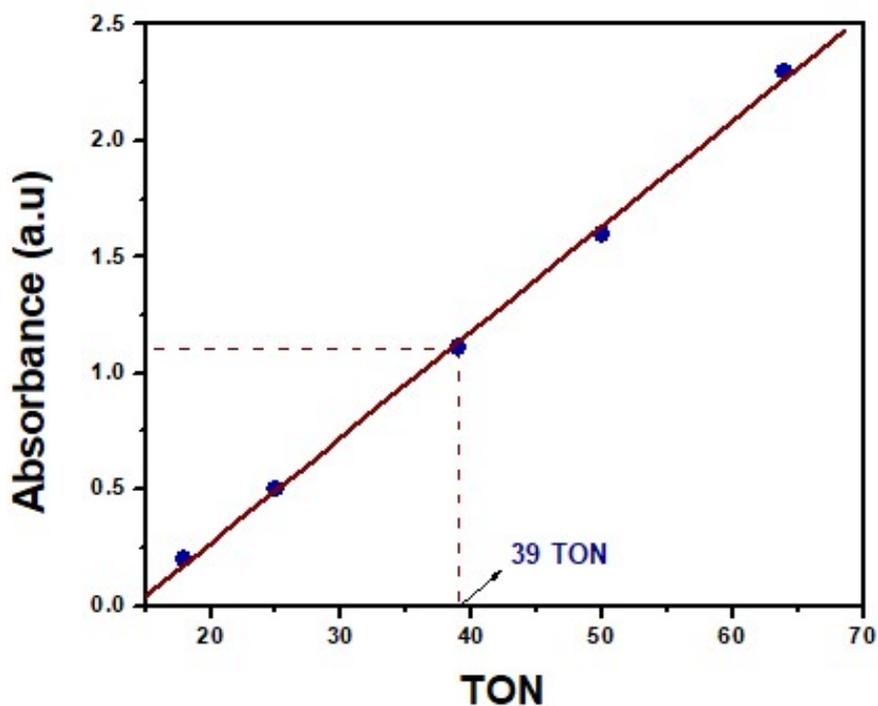


Figure S5. Calibration curve of formic acid for determination of concentration of formic acid produced.

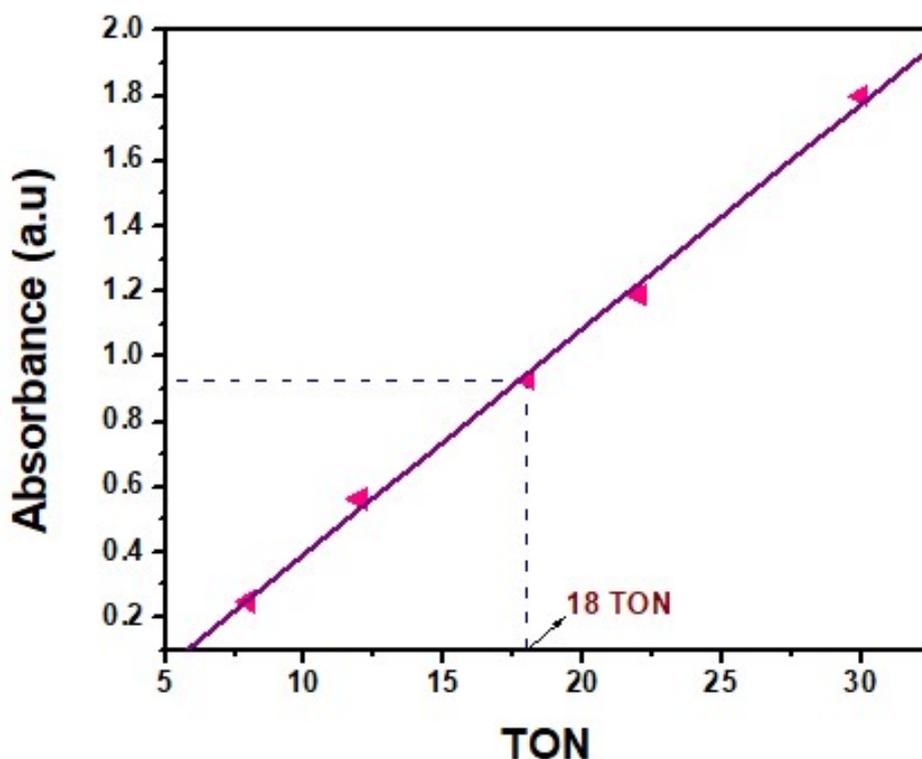


Figure S6. Calibration curve of formaldehyde for determination of concentration of formaldehyde formation.

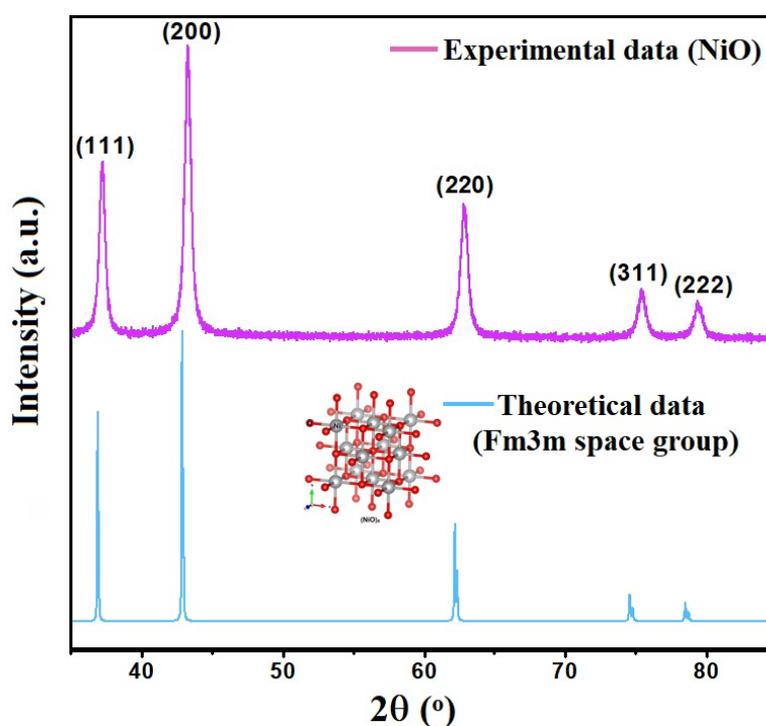


Figure S7. PXRD pattern of NiO and B3LYP/6-31G-optimized geometrical structures of $(\text{NiO})_4$. Bond lengths are given in Å and bond angles in degrees. The model of $(\text{NiO})_4$ nanocluster (space group Fm3m) has been selected to simultaneously investigate the performance of NiO and the impact of the COF on NiO.

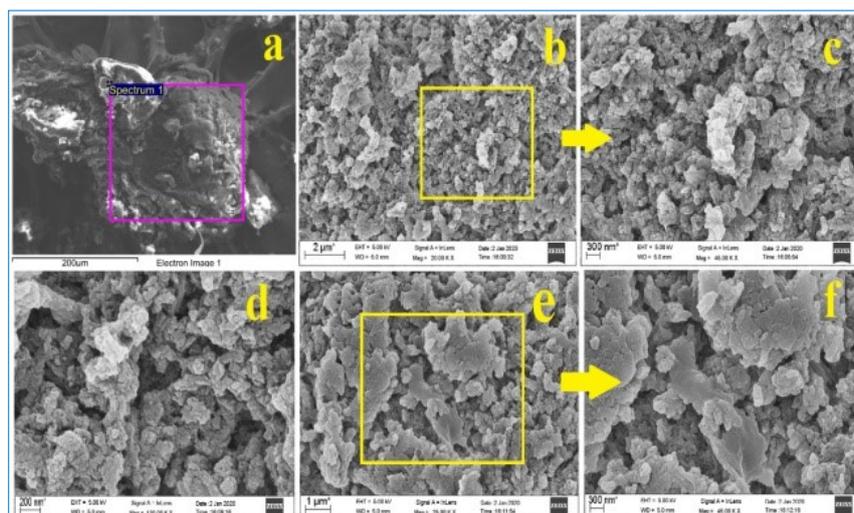


Figure S8. (a) electron image of TRITER-2, (b-f) FE-SEM images of COF (TRITER-2) at different magnifications.

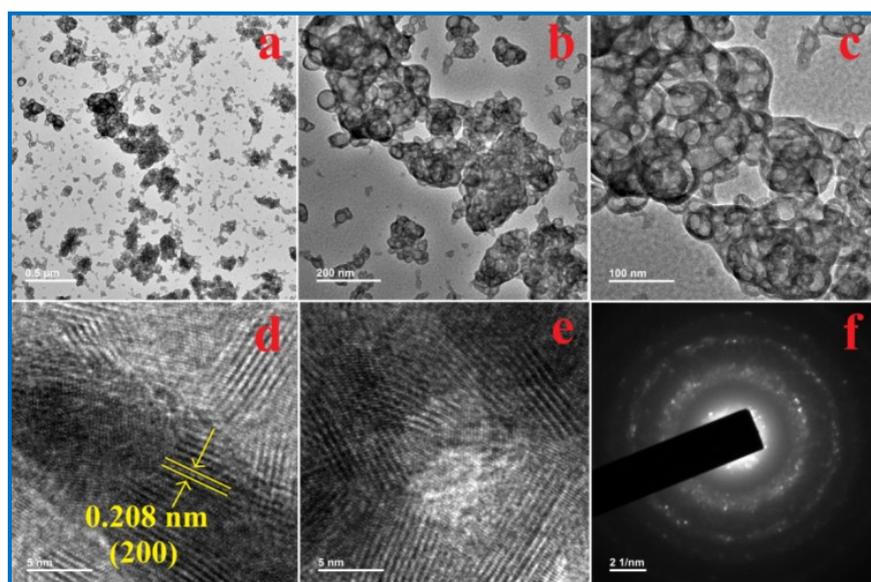


Figure S9. The TEM pictures of NiO at several scales (a) 0.5 μm , (b) 200 nm, (c) 100 nm, (d, e) HRTEM image of NiO nanoparticles showcasing the fringe pattern of NiO crystal, and (f) SAED pattern of NiO.

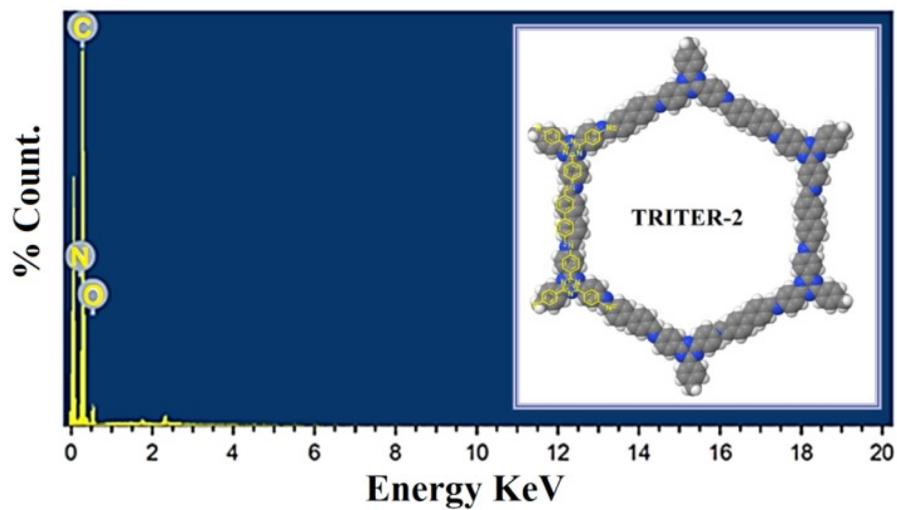


Figure S10. The EDAX pattern of the COF (TRITER-2).

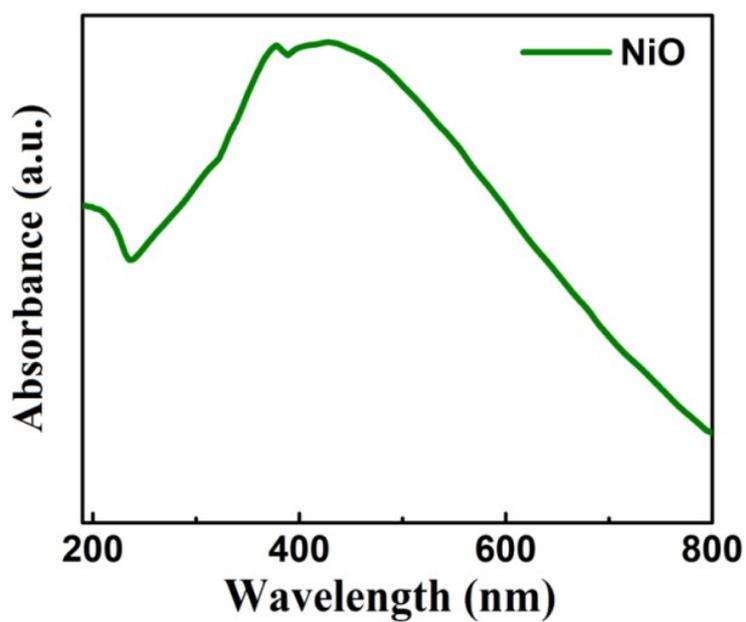


Figure S11. UV/Vis spectra of the synthesized NiO NPs.

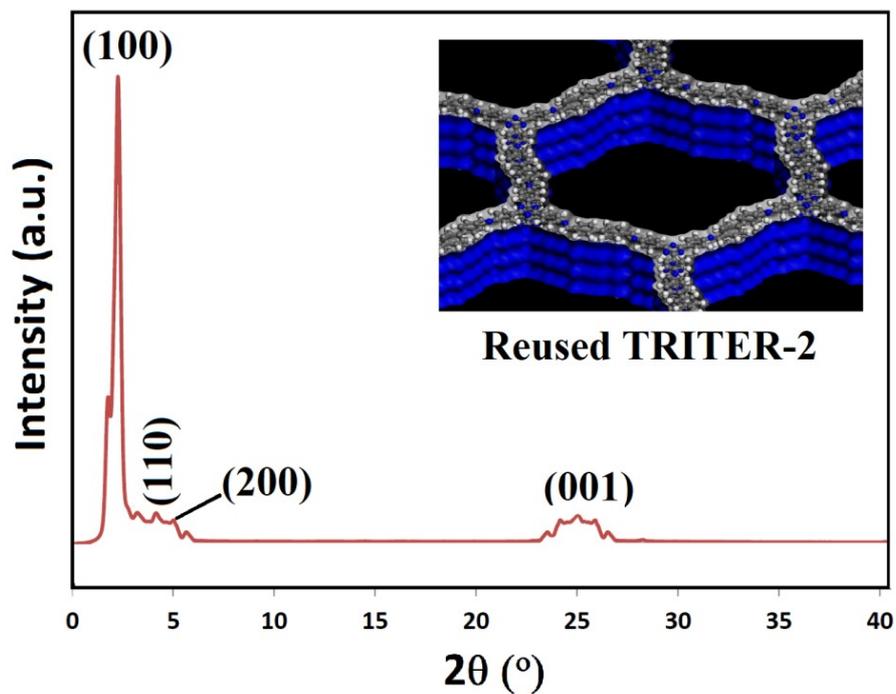


Figure S12. PXRD pattern of reused TRITER-2 catalyst after 5th run.

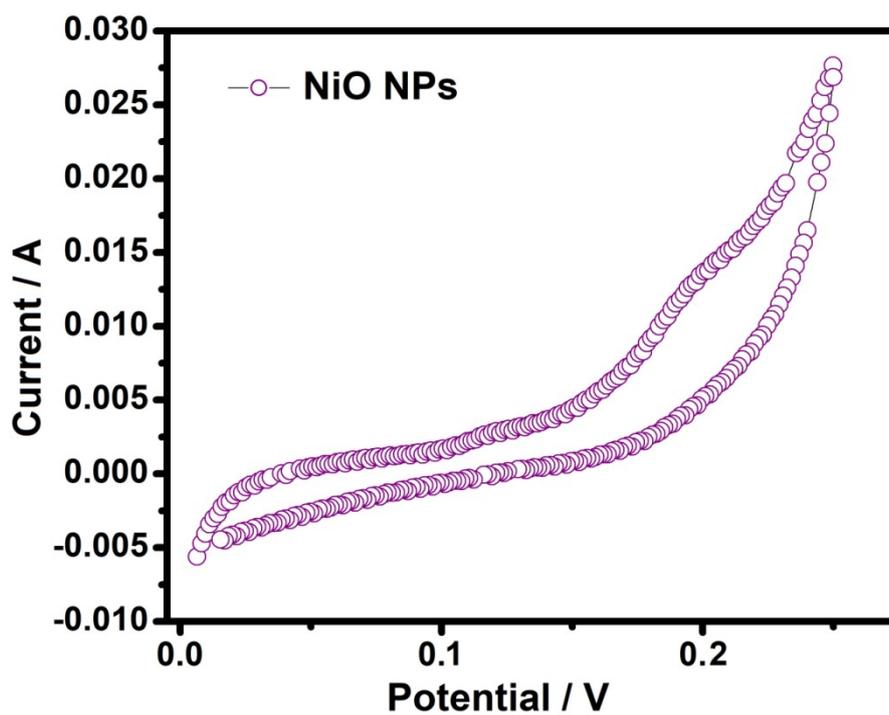


Figure S13. Cyclic voltammogram of the synthesized NiO nanoparticles.

The CV study was performed with the scan rate of 1 mV s^{-1} and the potential range of 0 to 0.3 volts. The cyclic voltammogram of the NiO nanoparticles is presented in Figure S13.

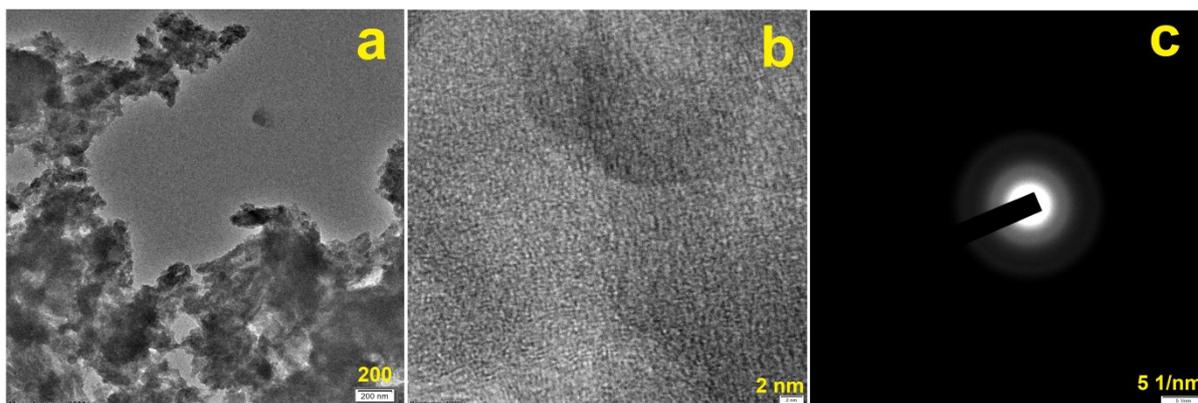


Figure S14. TEM pictures of the reused COF (TRITER-2) after 5th run at several scales (a) 200 nm, (b) 2 nm, and (c) SAED pattern of the reused COF (TRITER-2).

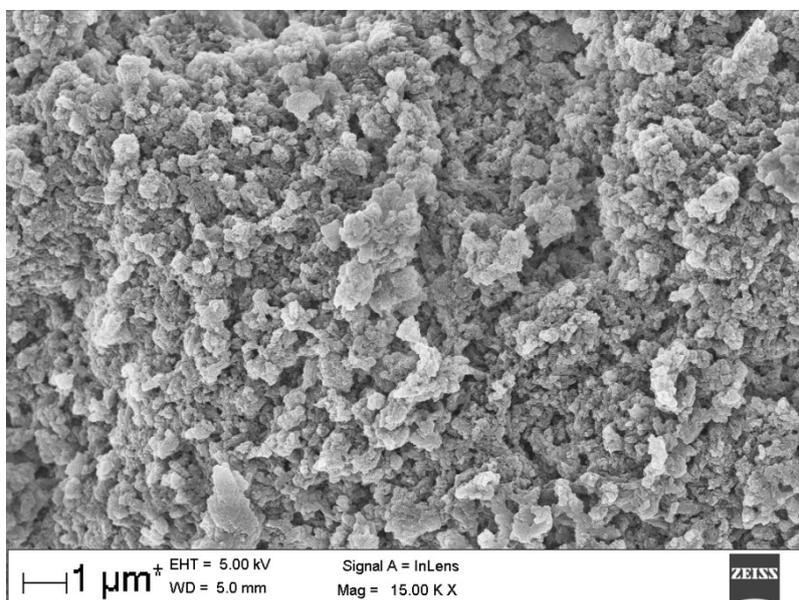


Figure S15. FE-SEM image of the reused COF (TRITER-2) after 5th cycle.

Kinetic curves (TON vs. time):

The stability of photocatalyst COF TRITER-2 was estimated via recycling and regeneration experiment. After completion of the photocatalytic reaction, COF TRITER-2 was recovered from the reaction mixture by centrifugation and washed multiple times with ethanol. After that, the recovered catalyst was kept in hot oven and dried for the next experiment. Recycling experiment was carried out up to five runs (Fig12). The methanol yield of each recycled run was consistent with the fresh catalyst. Slight reduction in yield after each run could be attributed to the crystalline destruction and pore blockage after continuous stirring during

photocatalytic experiment and centrifugation for catalyst isolation. It is worth mentioning that the catalytic efficiency dropped upon recycling, owing to pore blockage. Finally, Powder X-ray diffraction pattern of the reused photo-catalyst confirmed that there is no prominent change in the crystallinity of 2D COF (TRITER-2), as shown in Fig. S12.

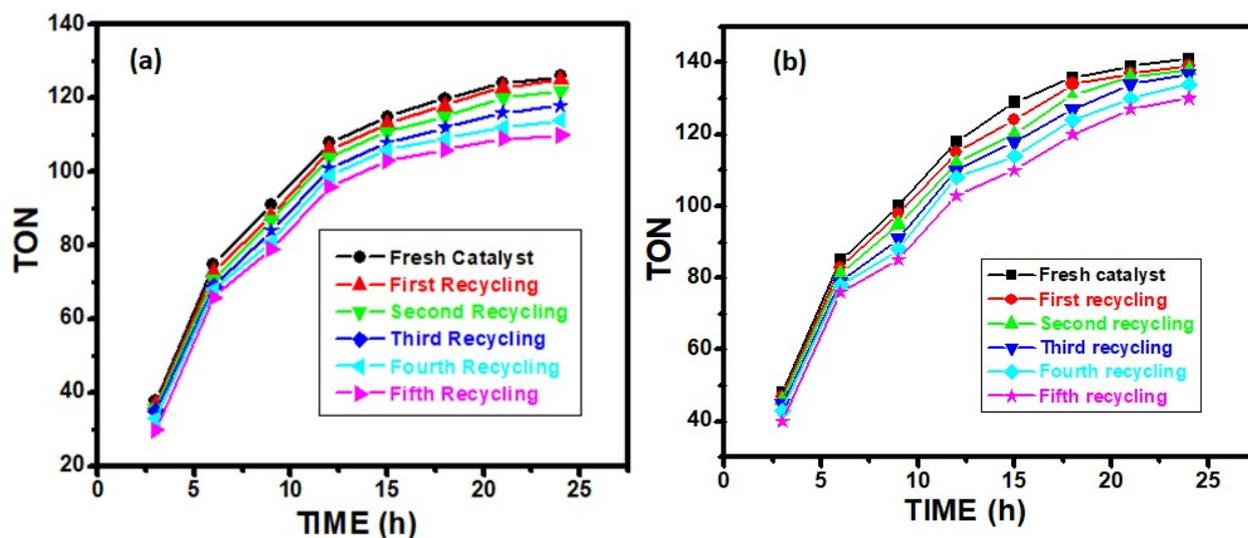
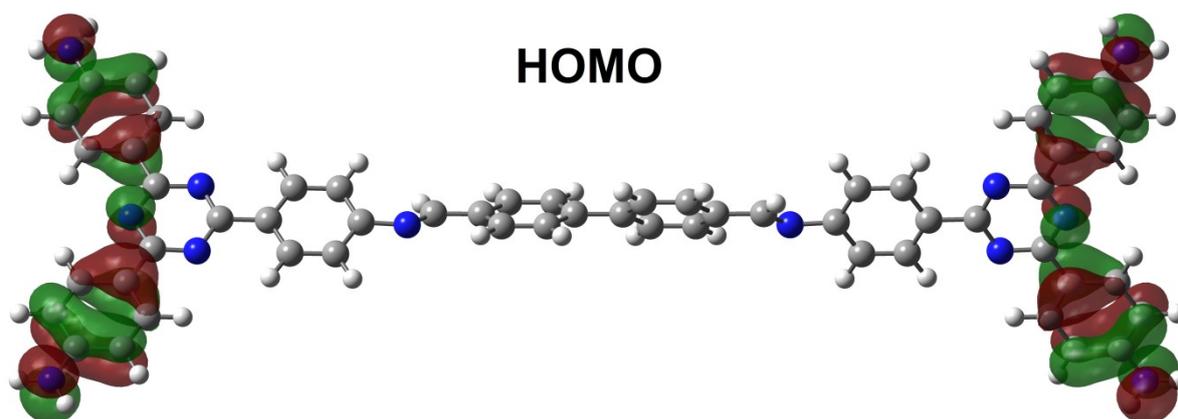


Figure S16. Kinetic curve for methanol synthesis and comparison of conversion rates of different recycling runs for methanol synthesis. (a) Kinetic curve for methanol synthesis with COF catalyst (TRITER-2 was synthesized through hydrothermal process); (b) Kinetic curve for methanol synthesis with COF photocatalyst (TRITER-2 was constructed under solvothermal conditions: 6M AcOH in mesitylene:dioxane=9:1)



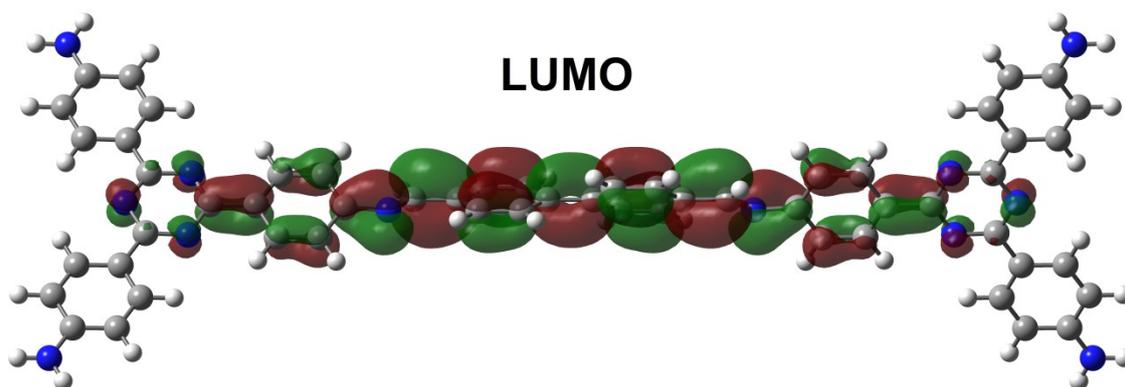


Figure S17. HOMO-LUMO of TRITER-2 fragment.

The comparison of photocatalytic activity with previously reported works and the role of COF in the catalytic process:

It is worth mentioning that the current COF for CO₂ reduction is more active than g-C₃N₄ and those coupled with semiconductors reported by others.¹⁻³ This could be ascribed to a higher adsorption affinity of the azine-based COF to CO₂, higher crystallinity (higher degree of structural order) and superior surface area compared to that of g-C₃N₄. Yield of CH₃OH over g-C₃N₄ is much lower. Thus this triazine-based COF is more competitive than g-C₃N₄ and COFs (i.e., ACOF-1 and N₃-COF)⁴ in the photocatalytic reduction of CO₂.

It's well-established that the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) are often invoked to rationalize delocalization of exciton, charge separation, and location of potential charge-transfer sites. Consequently, the density functional theory (DFT) calculations were conducted to investigate the conformational and electronic information based on the optimized geometries of TRITER-2 fragment. It is noteworthy that triazine moiety and phenyl groups in TRITER-2 are coplanar (Fig. S19), revealing a strong conjugation effect. Taking TRITER-2 fragment as an instance for better visibility, the HOMO is localized solely on the triazine linker unit, whereas the LUMO is delocalized across the conjugated π -system of the synthesized COF. The electronic distributions of TRITER-2 at the HOMO and the LUMO have a good electron-separated state and well-overlapped orbital, which is conducive to the intramolecular charge transfer transition. By calculating, the HOMO and LUMO energy levels can be predicted as -5.31 and -2.12 eV for TRITER-2, respectively. In principle, the observed HOMO-LUMO gap (3.19 eV) of TRITER-2 is large enough to enable CO₂ reduction through band gap excitation and at the same time small enough to harvest a significant portion of the visible light spectrum. The

excited electrons from the LUMO energy level can react with the adsorbed CO₂ on the catalyst surface to produce methanol, as shown in Scheme 1. Compared with other reported systems¹⁻³, the synthesized 2D COF with electron-poor character of triazine building blocks shows more efficient at stabilizing the negative charge generated on the COF, which is important for the photocatalytic process to get a better activity.⁵ This is consistent with the results from photocurrent experiments and photocatalytic reactions.

This COF was of great interest owing to the highly conjugated structure, accessible active sites, and accelerated charge transfer. Photocatalytic experiments indicated that the conjugated framework played a vital role in enhancing the photoactivity. Apart from narrowing the band gap, dialdehydes were also considered to possess higher charge carrier mobility and enable the accelerated migration of photogenerated excitons to the surface of the photocatalyst. In addition, electron donors such as tris-(4-aminophenyl)triazine (TAPT) and tris(4-aminophenyl)-benzene (TPB) were employed to construct COFs with tailored band gaps and improved charge separation and transfer.^{6,7} TRITER-2 with a narrow band gap and a negative conduction band was found to exhibit promoted visible-light harvesting efficiency and produce more charge carriers. The narrow band gap and facilitated electron transfer in the π -conjugated COF greatly enhanced the photocatalytic performance. Understandably, in the reaction of CO₂ photoreduction, the CO₂ absorption capability of the catalyst is the key point. In this study, the high surface area of TRITER-2 (1260 m² g⁻¹, see Fig. 2c) with abundant accessible nitrogen sites rendered them with high CO₂ absorption, leading to the facilitated photocatalytic reduction of CO₂ to CH₃OH. Compared with g-C₃N₄,¹⁻³ the synthesized COF (TRITER-2) with electron-poor triazine moieties was able to stabilize the negative charge generated on the COF which was important for the enhanced photocatalytic activity. Activity of the COF outperformed that of other materials such as g-C₃N₄.¹⁻³ Furthermore, the electronic properties and configuration of the COF were calculated with density functional theory (DFT). The results suggested that the potential of their LUMO was enough to drive CO₂ reduction. The band gap was suitable for the visible light response. Under visible light irradiation, the excited electrons at the LUMO energy level could reduce the adsorbed CO₂ on the catalyst surface to produce methanol (see Fig. 6b).

Photo-catalytic performance of the synthesized 2D COF can be improved following solvothermal method, profiting from the ordered nanoporous structure of the COF. The smooth flow of gas molecules, high efficient adsorption ability of CO₂, and the fast and steady transmission of hot electrons can be obtained by the high surface area and three-

dimensional transport channels of the COF. The ordered microporous COF exhibits superior material and structure, with a high surface area that offers more catalytically active sites. More importantly, the three-dimensional transport channels ensure the smooth flow of gas molecules, highly efficient CO₂ adsorption, and the fast and steady transmission of hot electrons excited, which lead to a further improvement in the photocatalytic performance. These results highlight the possibility of improving the photocatalysis for CO₂ reduction under visible light by constructing COF photocatalyst. Keeping these parameters in mind, we have developed the 2D COF (TRITER-2) for enhanced visible-light photocatalytic performance in sustainable metal-free CO₂ conversion induced by NiO.

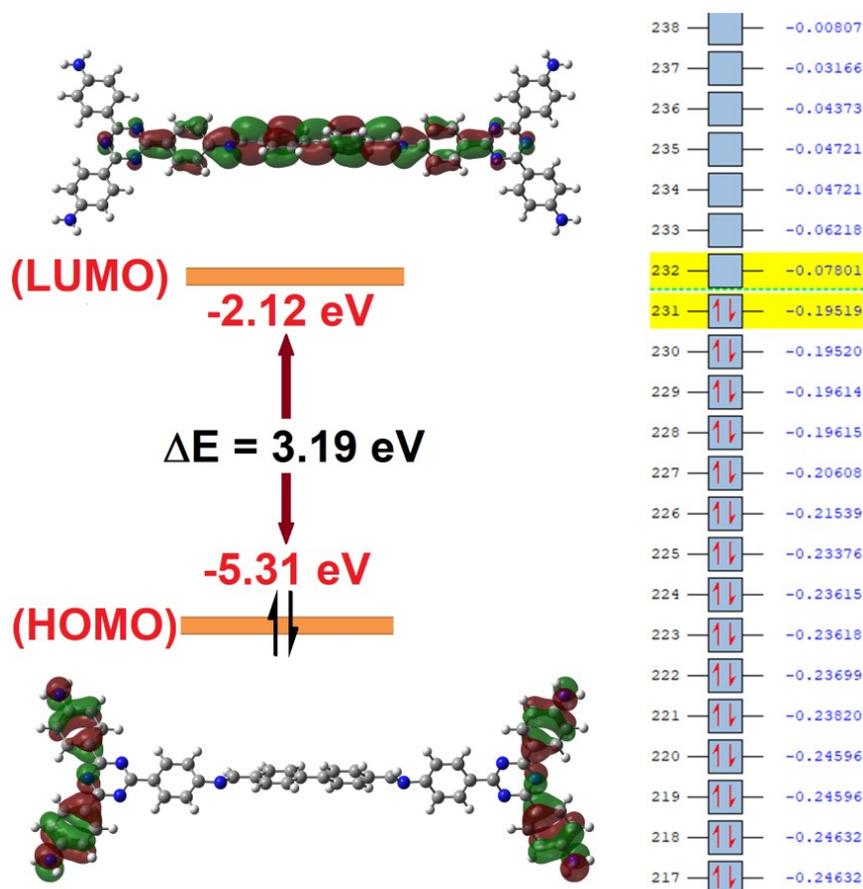


Figure S18. Diagram for computed HOMO–LUMO as achieved in the TD-DFT-B3LYP/6-31G.

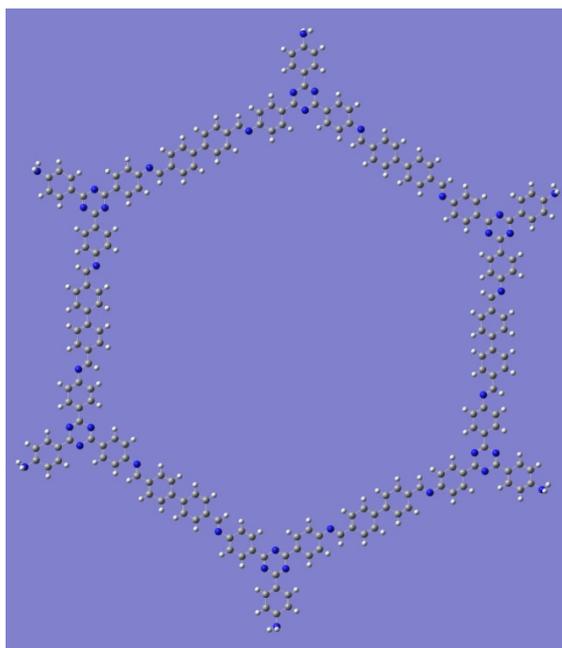


Figure S19. Optimized structure of TRITER-2 using DFT-B3LYP/6-31G, revealing that triazine moiety and phenyl groups in TRITER-2 are coplanar.

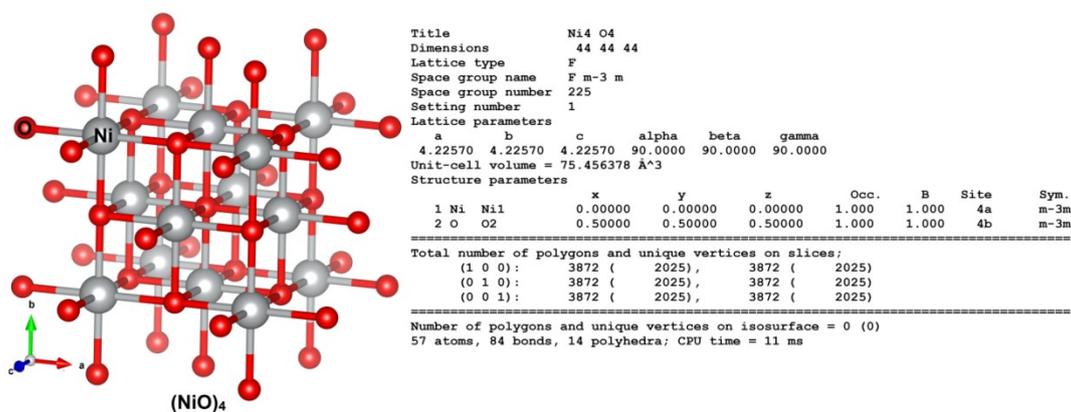


Figure S20. The model of (NiO)₄ nanocluster (space group Fm3m).

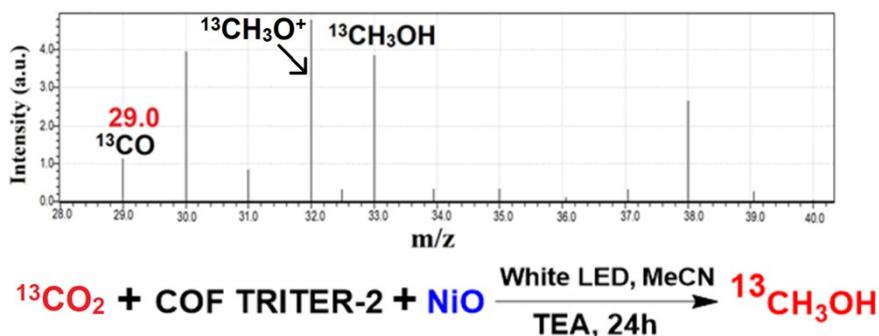
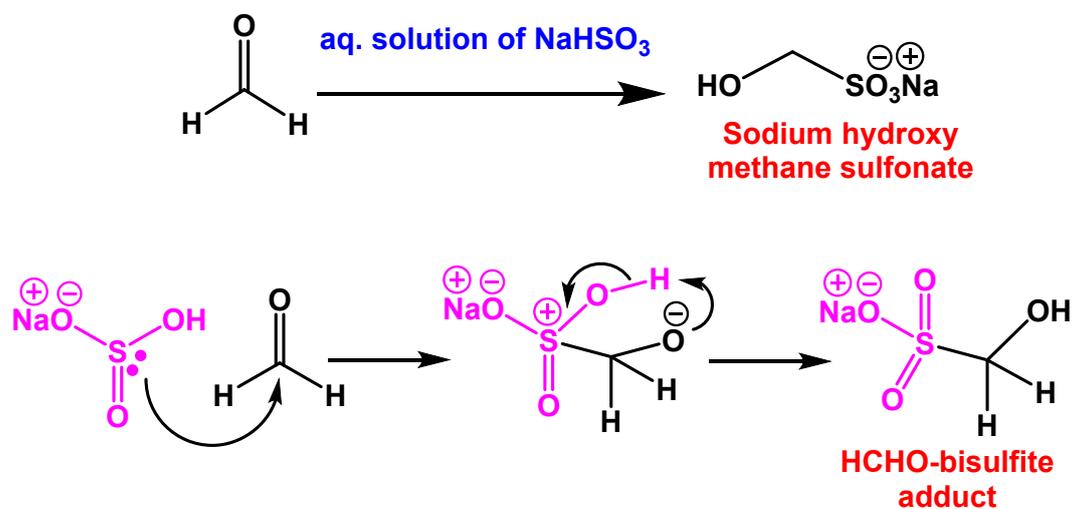


Figure S21. Mass spectra of the synthesized methanol employing ¹³CO₂ as the carbon source in the photocatalytic reaction. Reaction conditions: TRITER-2 photocatalyst (10 mg), NiO (5 mol%), time: 24 h, 1 atm ¹³CO₂ pressure (balloon), Intensity: 20W white LED light, RT. GC-

MS spectrum also showing photogenerated ^{13}CO under $^{13}\text{CO}_2$ atmosphere with TRITER-2 as the photocatalyst.



Scheme S1: Reaction between HCHO and NaHSO₃.

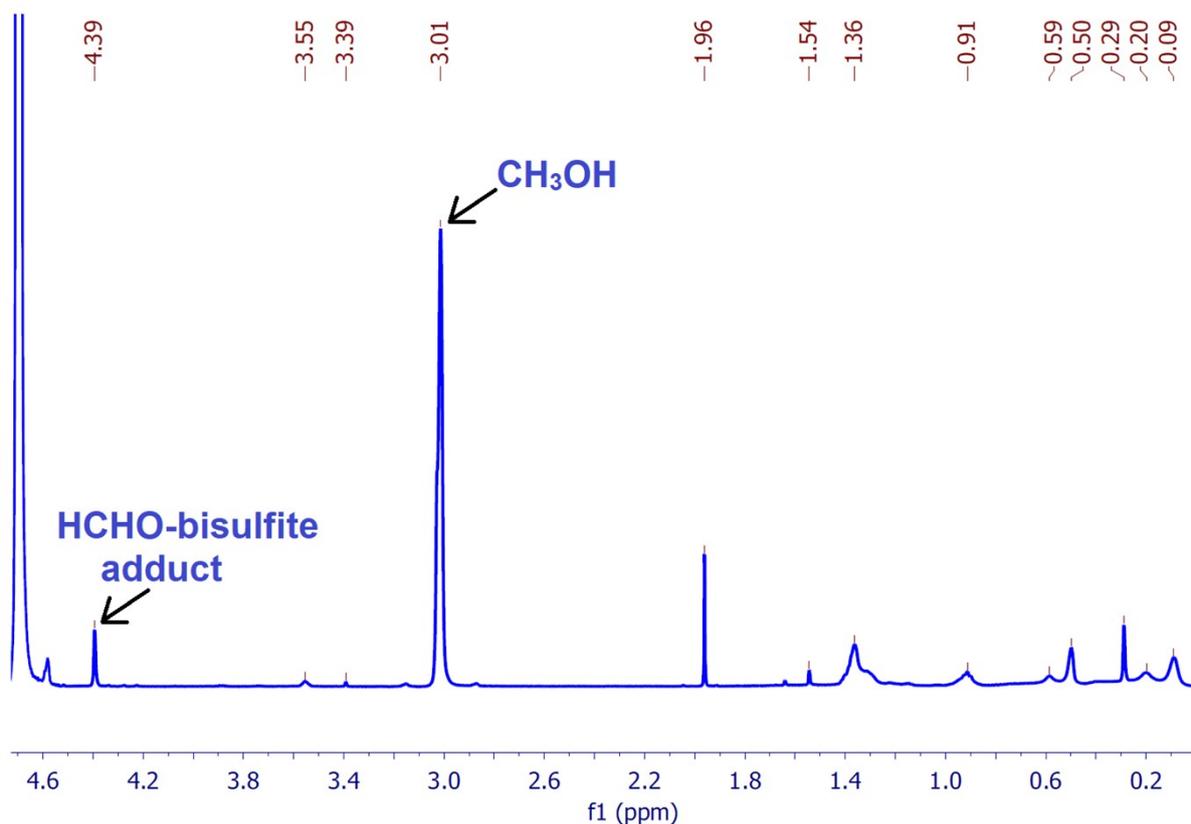


Figure S22. ^1H -NMR spectrum of the reaction mixture after photocatalytic CO_2 reduction (after treatment of the reaction mixture with NaHSO₃).

(Reaction conditions: TRITER-2 photocatalyst (10 mg), NiO (5 mol%), time: 24 h, 1 atm CO_2 pressure (balloon), Intensity: 20W white LED light, RT, solvent: acetonitrile- d_3)

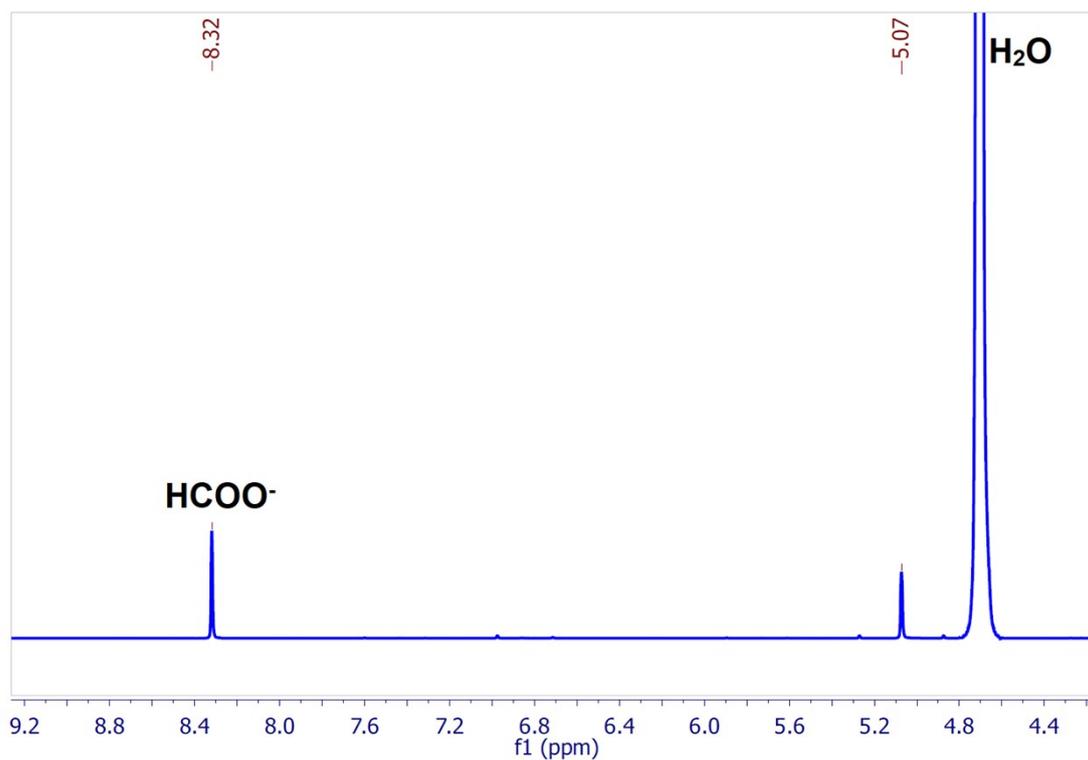
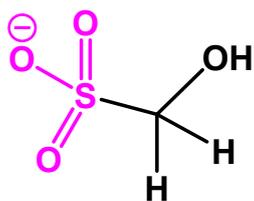


Figure S23. ^1H -NMR spectrum of the reaction mixture after CO_2 reduction (after treatment of the reaction mixture with excess water due to the presence of formic acid in the form of methyl formate).

(Reaction conditions: TRITER-2 photocatalyst (10 mg), NiO (5 mol%), time: 24 h, 1 atm CO_2 pressure (balloon), Intensity: 20W white LED light, RT, solvent: acetonitrile- d_3)

ESI-MS spectrum of the reaction mixture (after treatment with NaHSO₃):



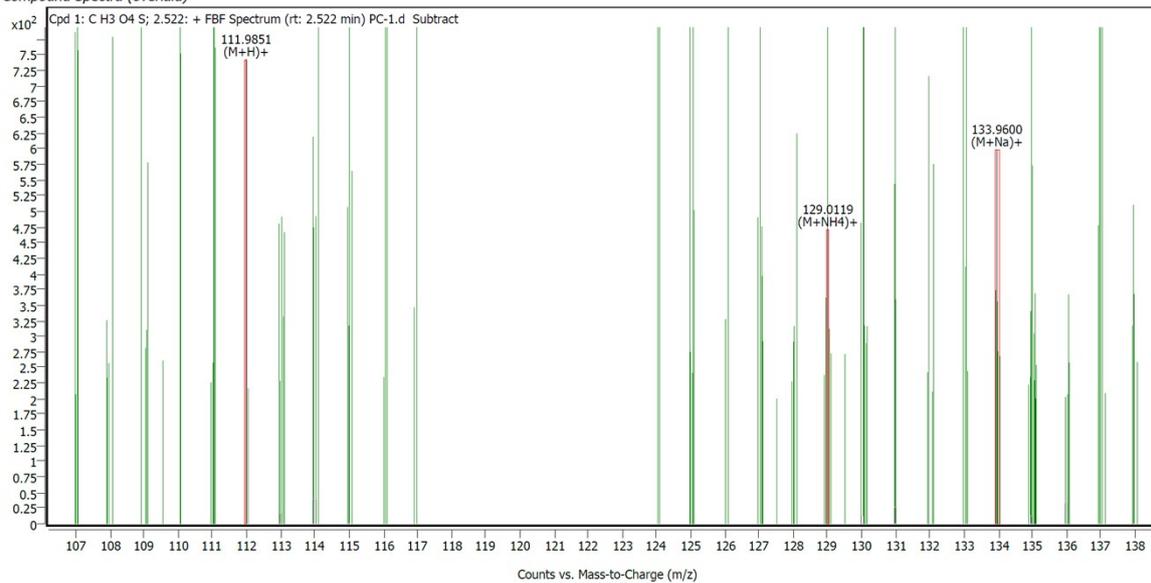
**HCHO-bisulfite
adduct**

Chemical Formula: CH₃O₄S⁻

Compound Details

Cpd. 1: C H₃ O₄ S

Compound Spectra (overlaid)



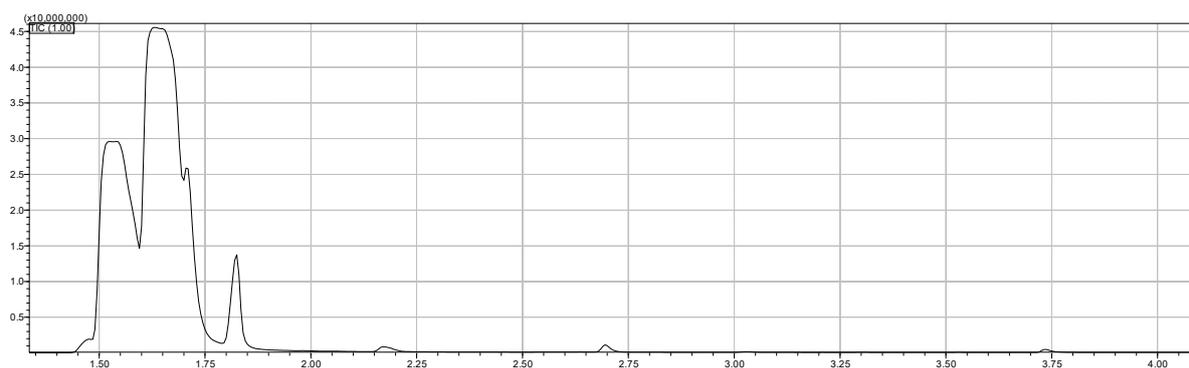
Compound ID Table

Cpd	Formula	Mass (Tgt)	Calc. Mass	Mass	Species	Diff(Tgt.ppm)	mDa
1	C H ₃ O ₄ S	110.9752	110.9756	111.9851 129.0119 133.9600	(M+H) ⁺ (M+NH ₄) ⁺ (M+Na) ⁺	3.19	0.35

GC-MS spectrum of the reaction mixture:



GC-MS spectrum of the reaction mixture (Zoom-in):



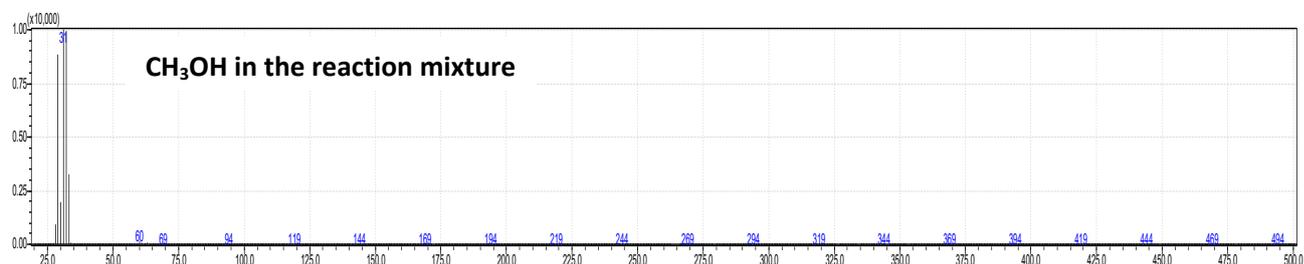
Qualitative table of the above GC-MS spectrum:

Peak#	Ret. Time	Start Tm	End Tm	m/z	Area	Area%	Height	Height%	A/H
1	1.470	1.430	1.485	TIC	3575610	0.77	1766689	1.94	2.02
2	1.540	1.485	1.595	TIC	151165454	32.70	29457164	32.42	5.13
3	1.632	1.595	1.790	TIC	285024314	61.65	45325997	49.89	6.29
4	1.823	1.790	1.885	TIC	21146404	4.57	13306512	14.65	1.59
5	2.695	2.670	2.735	TIC	1411208	0.31	991551	1.09	1.42

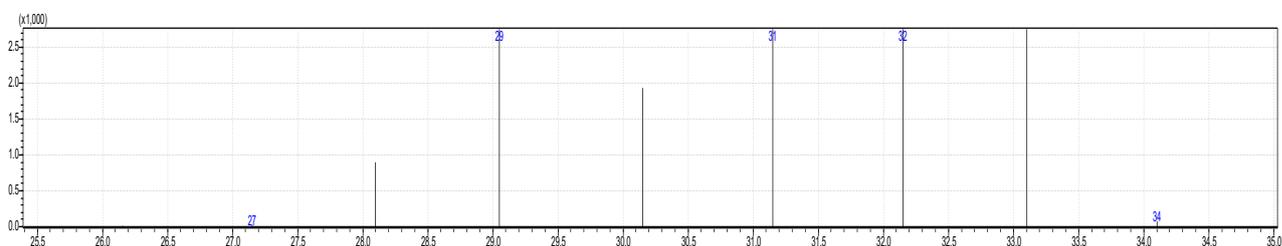
GC MS data of the reaction mixture:

1. Ret. Time: 1.540 (Methanol)

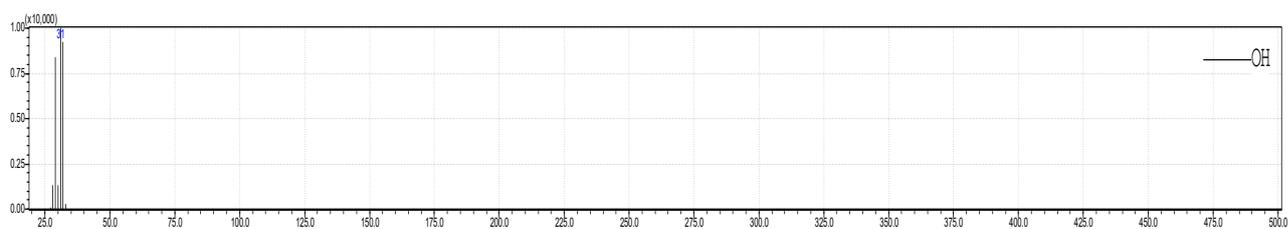
Mass spectra of the reaction mixture:



Mass spectra of the reaction mixture (Zoom-in):

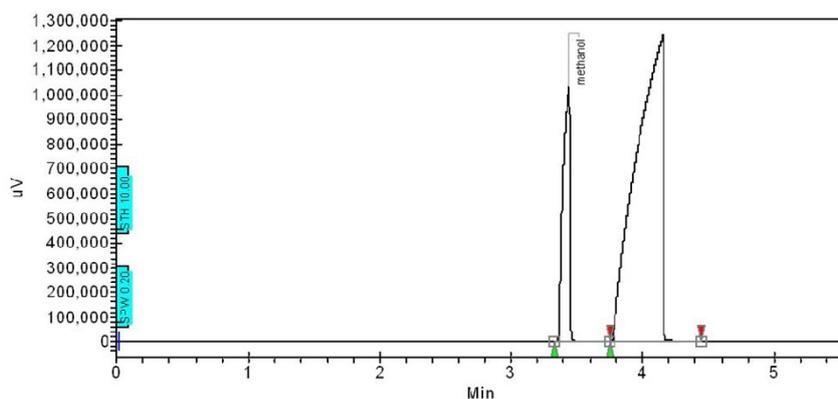
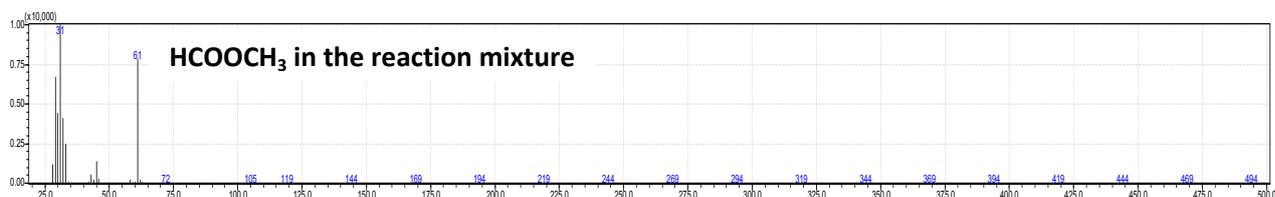


Standard sample (Pure methanol):



2. Ret. Time: 1.823 (Methyl formate)

Mass spectra of the reaction mixture:



Peak results :

Index	Name	Time [Min]	Quantity [M]	Height [uV]	Area [uV Min]	Area %	Area [uV Sec]	Quantity [M]
1	methanol	3.44	12.84	1034277.2	64324.4	18.053	3859465.7	12.84
2	UNKNOWN	4.16	0.00	1243405.4	291983.7	81.947	17519020.3	0.00
Total			12.84	2277682.6	356308.1	100.000	21378486.0	12.84

Figure S24. GC spectrum of the reaction mixture.(methanol formation catalyzed by TRITER-2 photocatalyst under optimised conditions).

(**Reaction conditions:** TRITER-2 photocatalyst (10 mg), NiO (5 mol%), time: 24 h, 1 atm CO₂ pressure (balloon), Intensity: 20W white LED light, RT.)

Preparation of g-C₃N₄:

Graphitic carbon nitride (g-C₃N₄) was synthesized by pyrolysis of urea in a muffle furnace; 20 g urea was put into an alumina crucible with a cover, then heated to 250 °C within 110 min and kept at 250 °C for 1 h. The further treatment was performed at 350 and 550 °C for 2 h, respectively. The heating rate of the whole reaction was 2 °C·min⁻¹. The yellow power (g-C₃N₄) was collected. The collected amount of the g-C₃N₄ was around 1 g.

Sample characterization

Figure S25 illustrates the XRD pattern of g-C₃N₄. Observed from Figure S25, there are two obvious diffraction peaks at around 12.9° and 27.1°, which were assigned to the (100) and (002) planes of g-C₃N₄. These two peaks are likely to be attributed to the structure of the tri-s-triazine unit with interplanar spacing and the conjugated aromatic system, respectively.⁸

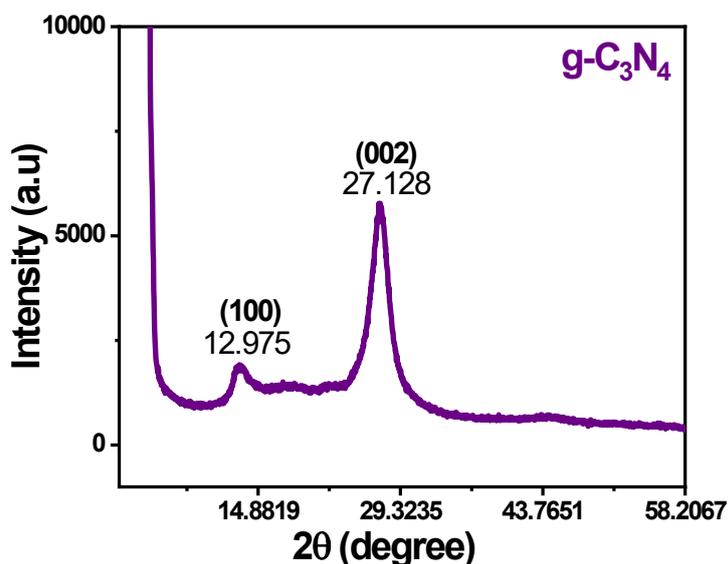
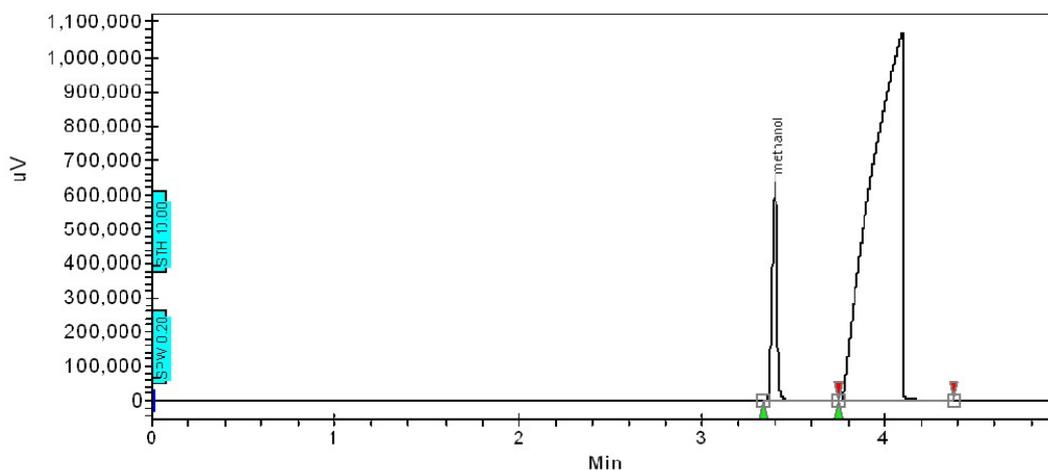


Figure S25. PXRD pattern of g-C₃N₄.



Peak results :

Index	Name	Time [Min]	Quantity [M]	Height [uV]	Area [uV.Min]	Area % [%]	Area [uV.Sec]	Quantity [M]
1	methanol	3.40	4.25	634679.5	21268.9	9.329	1276135.7	4.25
2	UNKNOWN	4.09	0.00	1068951.5	206713.7	90.671	12402822.7	0.00
Total			4.25	1703631.1	227982.6	100.000	13678958.4	4.25

Figure S26. GC spectrum of the reaction mixture.(methanol production catalyzed by g-C₃N₄ employing already established reaction conditions).

(Reaction conditions: g-C₃N₄ photocatalyst (10 mg), NiO (5 mol%), time: 24 h, 1 atm CO₂ pressure (balloon), Intensity: 20W white LED light, RT.)

The identification and quantification of the methanol production was carried out using GC-MS using headspace at elevated temperature of 90°C. For the observed GC peak areas using the observed peak area of 79.0 for 1.0 M methanol as standard the methanol concentrations were determined. Methanol was analyzed by GC-FID, and a DB-WAX 123-7033 column was used for the detection of methanol. Helium gas was introduced as the shipper gas, and the flow rate was 35 cm/s. The oven temperature was adjusted at 40 °C, and inlet temperature was fixed at 200 °C. The split ratio was 1:50 and the FID temperature was 300 °C. Nitrogen gas was used as the makeup with a flow rate of 30 mL/min at the FID detector. The yield was calculated by using the following equation.

$$Y = (C \times V/W)$$

Y = yield

C = methanol concentration, $\mu\text{mol/L}$

V = volume, L

W = mass of catalyst, g

All the relevant original spectra of HCOOH/HCHO were given in SI.

In the case of catalytic TRITER-2, methanol was attained in excellent yield of about 1,14000 $\mu\text{mol. gcat}^{-1}$. In the case of catalytic g-C₃N₄, methanol was achieved in poor yield of about 59,000 $\mu\text{mol. gcat}^{-1}$.

Table S1: The results of photochemical reduction of CO₂ to methanol using TRITER-2 and g-C₃N₄ as photocatalyst. A 5 mL and 0.01 g (i.e., 10 mg) were used as the volume and weight of the photocatalyst, respectively. GC standard peak area of 79.0 for 1.0 M Methanol (MeOH) was used.

Catalyst	Reaction time, t reaction (h)	GC area found for MeOH	Molarity of MeOH formed (M)	Moles of MeOH formed ($\mu\text{ mol.}$)
TRITER-2	24	18.053	0.228	1140
g-C ₃ N ₄	24	9.329	0.118	590

We have carried out two more experiments with blue LED (450 nm) and green LED (520-555 nm). We observed poor yield with blue LED. Although with the green LED and white LED gave us comparable reaction yield, we have selected the White LED as our reaction source. Since, we observed from the UV-Vis spectra of TRITER-2 (Figure 6a) in the manuscript, the excitation maxima is 520 nm, so we have selected

the white LED as the light source which is in between 480-590 nm, which confirms that the reaction proceeds by band gap excitation of TRITER-2.

Light source	Wave length (nm)	Yield (TON)/ (μ mol.)
White LED	480-590	141 TON (conc=0.228M), 1140 μ mol
Green LED	520-530	132 TON (conc=0.214M), 1070 μ mol
Blue	450	118 TON (conc=0.191M), 955 μ mol

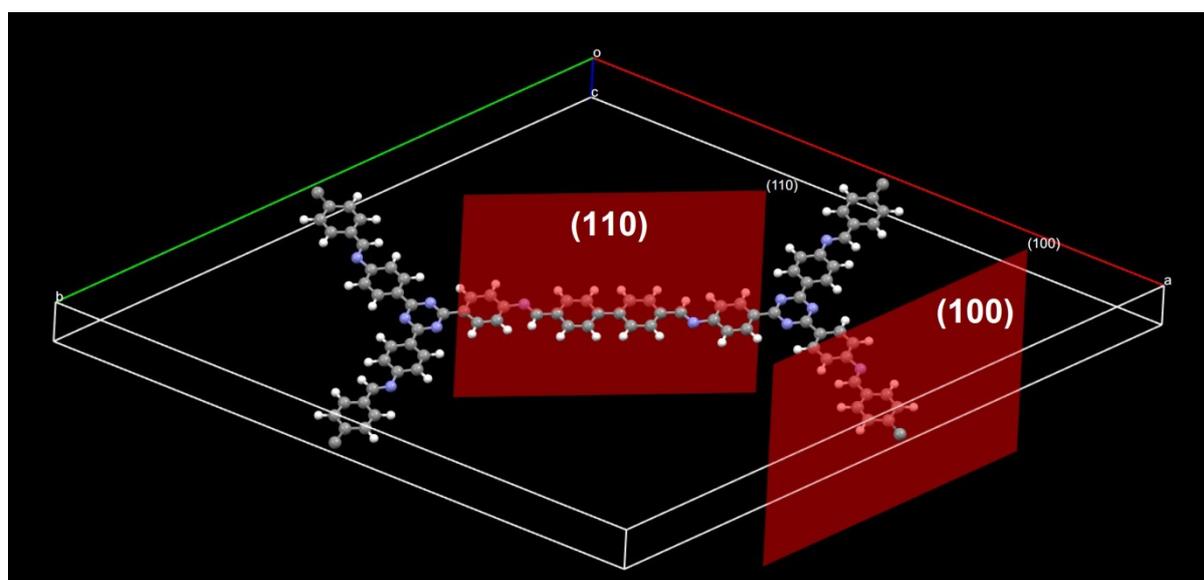


Figure S27. 1 \times 1 unit cell of the COF (TRITER-2) showing (100) and (110) plane.

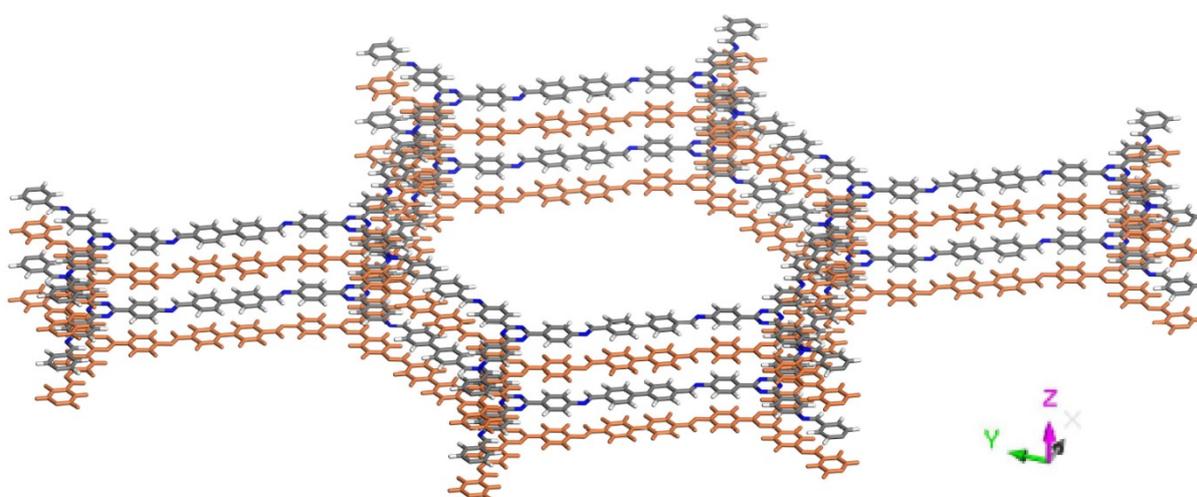


Figure S28. 2 \times 2 unit cell of the COF (TRITER-2) displaying π - π stacking between the COF layers.

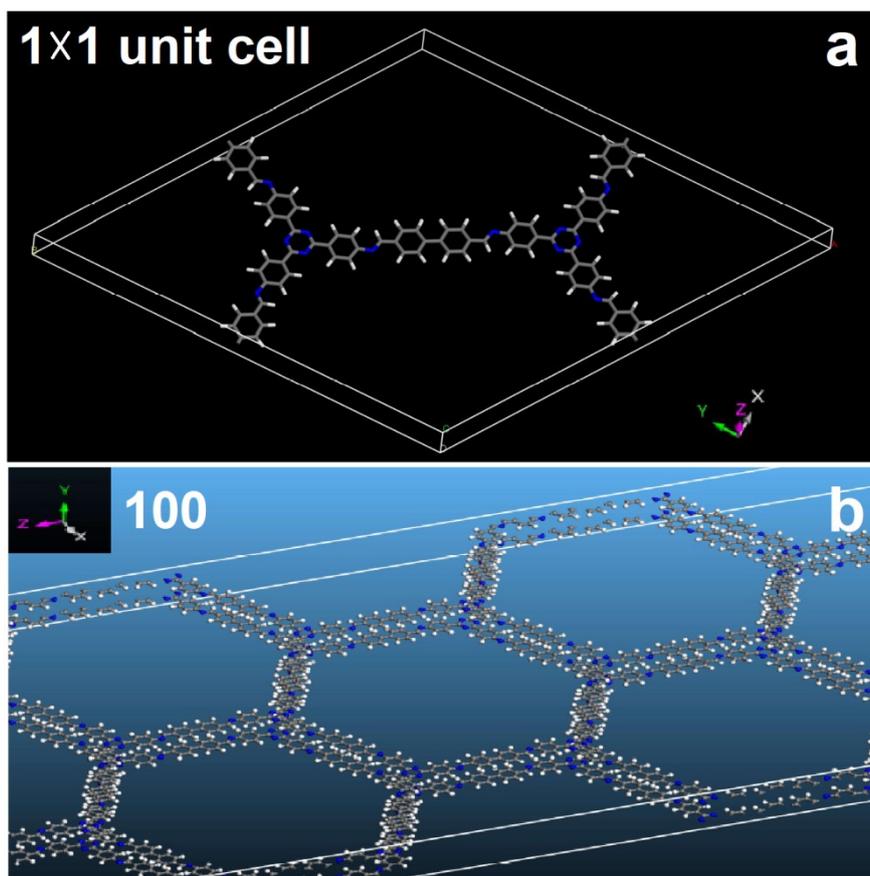


Figure S29. 1×1 unit cell of the COF (TRITER-2) (a) and TRITER-2 (100) supercell (b).

Table S2. Fractional main atomic coordinates for the unit cell of the COF (TRITER-2) after Pawley refinement.

Space group symmetry $P_{6/m}$
 $a = b = 42.3348 \text{ \AA}$, $c = 3.8828 \text{ \AA}$,
 $\alpha = \beta = 90^\circ$ and $\gamma = 120^\circ$

No	Label	Charge	SybylType	Xfrac + ESD	Yfrac + ESD	Zfrac + ESD	Symm. op.
1	N1	0	N.2	0.32022	0.68798	0.50000	x, y, z
2	C2	0	C.2	0.35481	0.70135	0.50000	x, y, z
3	C3	0	C.2	0.37825	0.73921	0.50000	x, y, z
4	C4	0	C.2	0.36496	0.76174	0.50000	x, y, z
5	C5	0	C.2	0.38706	0.79738	0.50000	x, y, z
6	C6	0	C.2	0.42290	0.81125	0.50000	x, y, z
7	C7	0	C.2	0.43624	0.78887	0.50000	x, y, z
8	C8	0	C.2	0.41417	0.75323	0.50000	x, y, z
9	N9	0	N.2	0.44667	0.84881	0.50000	x, y, z
10	C10	0	C.2	0.43488	0.87005	0.50000	x, y, z
11	C11	0	C.2	0.45454	0.90703	0.50000	x, y, z
12	C12	0	C.2	0.49064	0.92578	0.50000	x, y, z
13	C13	0	C.2	0.50825	0.96178	0.50000	x, y, z
14	C14	0	C.2	0.49047	0.98047	0.50000	x, y, z
15	C15	0	C.2	0.45404	0.96092	0.50000	x, y, z
16	C16	0	C.2	0.43663	0.92496	0.50000	x, y, z

17	H17	0	H	0.33590	0.75070	0.50000	x, y, z
18	H18	0	H	0.37602	0.81540	0.50000	x, y, z
19	H19	0	H	0.46528	0.79972	0.50000	x, y, z
20	H20	0	H	0.42524	0.73523	0.50000	x, y, z
21	H21	0	H	0.53759	0.97677	0.50000	x, y, z
22	H22	0	H	0.50563	0.91143	0.50000	x, y, z
23	H23	0	H	0.43831	0.97451	0.50000	x, y, z
24	H24	0	H	0.40730	0.90982	0.50000	x, y, z
25	H25	0	H	0.40593	0.85971	0.50000	x, y, z
26	N26	0	N.2	0.31202	0.63224	0.50000	x, y, z
27	C27	0	C.2	0.29865	0.65346	0.50000	x, y, z
28	C28	0	C.2	0.26079	0.63904	0.50000	x, y, z
29	C29	0	C.2	0.23826	0.60322	0.50000	x, y, z
30	C30	0	C.2	0.20262	0.58968	0.50000	x, y, z
31	C31	0	C.2	0.18875	0.61165	0.50000	x, y, z
32	C32	0	C.2	0.21113	0.64737	0.50000	x, y, z
33	C33	0	C.2	0.24677	0.66094	0.50000	x, y, z
34	N34	0	N.2	0.15119	0.59786	0.50000	x, y, z
35	C35	0	C.2	0.12995	0.56483	0.50000	x, y, z
36	C36	0	C.2	0.09297	0.54751	0.50000	x, y, z
37	C37	0	C.2	0.07422	0.56486	0.50000	x, y, z
38	C38	0	C.2	0.03822	0.54647	0.50000	x, y, z
39	C39	0	C.2	0.01953	0.51000	0.50000	x, y, z
40	C40	0	C.2	0.03908	0.49312	0.50000	x, y, z
41	C41	0	C.2	0.07504	0.51167	0.50000	x, y, z
42	H42	0	H	0.24930	0.58520	0.50000	x, y, z
43	H43	0	H	0.18460	0.56062	0.50000	x, y, z
44	H44	0	H	0.20028	0.66556	0.50000	x, y, z
45	H45	0	H	0.26477	0.69001	0.50000	x, y, z
46	H46	0	H	0.02323	0.56082	0.50000	x, y, z
47	H47	0	H	0.08857	0.59420	0.50000	x, y, z
48	H48	0	H	0.02549	0.46380	0.50000	x, y, z
49	H49	0	H	0.09018	0.49748	0.50000	x, y, z
50	H50	0	H	0.14029	0.54622	0.50000	x, y, z
51	N51	0	N.2	0.36776	0.67978	0.50000	x, y, z
52	C52	0	C.2	0.34654	0.64519	0.50000	x, y, z
53	C53	0	C.2	0.36096	0.62175	0.50000	x, y, z
54	C54	0	C.2	0.39678	0.63504	0.50000	x, y, z
55	C55	0	C.2	0.41032	0.61294	0.50000	x, y, z
56	C56	0	C.2	0.38835	0.57710	0.50000	x, y, z
57	C57	0	C.2	0.35263	0.56376	0.50000	x, y, z
58	C58	0	C.2	0.33906	0.58583	0.50000	x, y, z
59	N59	0	N.2	0.40214	0.55333	0.50000	x, y, z
60	C60	0	C.2	0.43517	0.56512	0.50000	x, y, z
61	C61	0	C.2	0.45249	0.54546	0.50000	x, y, z
62	C62	0	C.2	0.43514	0.50936	0.50000	x, y, z
63	C63	0	C.2	0.45353	0.49175	0.50000	x, y, z
64	C64	0	C.2	0.49000	0.50953	0.50000	x, y, z
65	C65	0	C.2	0.50688	0.54596	0.50000	x, y, z
66	C66	0	C.2	0.48833	0.56337	0.50000	x, y, z
67	H67	0	H	0.41480	0.66410	0.50000	x, y, z
68	H68	0	H	0.43938	0.62398	0.50000	x, y, z
69	H69	0	H	0.33444	0.53472	0.50000	x, y, z
70	H70	0	H	0.30999	0.57476	0.50000	x, y, z
71	H71	0	H	0.43918	0.46241	0.50000	x, y, z
72	H72	0	H	0.40580	0.49437	0.50000	x, y, z

73	H73	0	H	0.53620	0.56169	0.50000	x, y, z
74	H74	0	H	0.50252	0.59270	0.50000	x, y, z
75	H75	0	H	0.45378	0.59407	0.50000	x, y, z
76	N76	0	N.2	0.67978	0.31202	0.50000	x, y, z
77	C77	0	C.2	0.64519	0.29865	0.50000	x, y, z
78	C78	0	C.2	0.62175	0.26079	0.50000	x, y, z
79	C79	0	C.2	0.63504	0.23826	0.50000	x, y, z
80	C80	0	C.2	0.61294	0.20262	0.50000	x, y, z
81	C81	0	C.2	0.57710	0.18875	0.50000	x, y, z
82	C82	0	C.2	0.56376	0.21113	0.50000	x, y, z
83	C83	0	C.2	0.58583	0.24677	0.50000	x, y, z
84	N84	0	N.2	0.55333	0.15119	0.50000	x, y, z
85	C85	0	C.2	0.56512	0.12995	0.50000	x, y, z
86	C86	0	C.2	0.54546	0.09297	0.50000	x, y, z
87	C87	0	C.2	0.50936	0.07422	0.50000	x, y, z
88	C88	0	C.2	0.49175	0.03822	0.50000	x, y, z
89	C89	0	C.2	0.50953	0.01953	0.50000	x, y, z
90	C90	0	C.2	0.54596	0.03908	0.50000	x, y, z
91	C91	0	C.2	0.56337	0.07504	0.50000	x, y, z
92	H92	0	H	0.66410	0.24930	0.50000	x, y, z
93	H93	0	H	0.62398	0.18460	0.50000	x, y, z
94	H94	0	H	0.53472	0.20028	0.50000	x, y, z
95	H95	0	H	0.57476	0.26477	0.50000	x, y, z
96	H96	0	H	0.46241	0.02323	0.50000	x, y, z
97	H97	0	H	0.49437	0.08857	0.50000	x, y, z
98	H98	0	H	0.56169	0.02549	0.50000	x, y, z
99	H99	0	H	0.59270	0.09018	0.50000	x, y, z
100	H100	0	H	0.59407	0.14029	0.50000	x, y, z
101	N101	0	N.2	0.68798	0.36776	0.50000	x, y, z
102	C102	0	C.2	0.70135	0.34654	0.50000	x, y, z
103	C103	0	C.2	0.73921	0.36096	0.50000	x, y, z
104	C104	0	C.2	0.76174	0.39678	0.50000	x, y, z
105	C105	0	C.2	0.79738	0.41032	0.50000	x, y, z
106	C106	0	C.2	0.81125	0.38835	0.50000	x, y, z
107	C107	0	C.2	0.78887	0.35263	0.50000	x, y, z
108	C108	0	C.2	0.75323	0.33906	0.50000	x, y, z
109	N109	0	N.2	0.84881	0.40214	0.50000	x, y, z
110	C110	0	C.2	0.87005	0.43517	0.50000	x, y, z
111	C111	0	C.2	0.90703	0.45249	0.50000	x, y, z
112	C112	0	C.2	0.92578	0.43514	0.50000	x, y, z
113	C113	0	C.2	0.96178	0.45353	0.50000	x, y, z
114	C114	0	C.2	0.98047	0.49000	0.50000	x, y, z
115	C115	0	C.2	0.96092	0.50688	0.50000	x, y, z
116	C116	0	C.2	0.92496	0.48833	0.50000	x, y, z
117	H117	0	H	0.75070	0.41480	0.50000	x, y, z
118	H118	0	H	0.81540	0.43938	0.50000	x, y, z
119	H119	0	H	0.79972	0.33444	0.50000	x, y, z
120	H120	0	H	0.73523	0.30999	0.50000	x, y, z
121	H121	0	H	0.97677	0.43918	0.50000	x, y, z
122	H122	0	H	0.91143	0.40580	0.50000	x, y, z
123	H123	0	H	0.97451	0.53620	0.50000	x, y, z
124	H124	0	H	0.90982	0.50252	0.50000	x, y, z
125	H125	0	H	0.85971	0.45378	0.50000	x, y, z
126	N126	0	N.2	0.63224	0.32022	0.50000	x, y, z
127	C127	0	C.2	0.65346	0.35481	0.50000	x, y, z
128	C128	0	C.2	0.63904	0.37825	0.50000	x, y, z

129	C129	0	C.2	0.60322	0.36496	0.50000	x, y, z
130	C130	0	C.2	0.58968	0.38706	0.50000	x, y, z
131	C131	0	C.2	0.61165	0.42290	0.50000	x, y, z
132	C132	0	C.2	0.64737	0.43624	0.50000	x, y, z
133	C133	0	C.2	0.66094	0.41417	0.50000	x, y, z
134	N134	0	N.2	0.59786	0.44667	0.50000	x, y, z
135	C135	0	C.2	0.56483	0.43488	0.50000	x, y, z
136	C136	0	C.2	0.54751	0.45454	0.50000	x, y, z
137	C137	0	C.2	0.56486	0.49064	0.50000	x, y, z
138	C138	0	C.2	0.54647	0.50825	0.50000	x, y, z
139	C139	0	C.2	0.51000	0.49047	0.50000	x, y, z
140	C140	0	C.2	0.49312	0.45404	0.50000	x, y, z
141	C141	0	C.2	0.51167	0.43663	0.50000	x, y, z
142	H142	0	H	0.58520	0.33590	0.50000	x, y, z
143	H143	0	H	0.56062	0.37602	0.50000	x, y, z
144	H144	0	H	0.66556	0.46528	0.50000	x, y, z
145	H145	0	H	0.69001	0.42524	0.50000	x, y, z
146	H146	0	H	0.56082	0.53759	0.50000	x, y, z
147	H147	0	H	0.59420	0.50563	0.50000	x, y, z
148	H148	0	H	0.46380	0.43831	0.50000	x, y, z
149	H149	0	H	0.49748	0.40730	0.50000	x, y, z
150	H150	0	H	0.54622	0.40593	0.50000	x, y, z
151	C14	0	C.3	0.49047	-0.01953	0.50000	x, -1+y, z
152	C39	0	C.3	1.01953	0.51000	0.50000	1+x, y, z
153	C89	0	C.3	0.50953	1.01953	0.50000	x, 1+y, z
154	C114	0	C.3	-0.01953	0.49000	0.50000	-1+x, y, z

Table S3.List of all torsions present in the COF (TRITER-2)

No	Atom 1	Atom 2	Atom 3	Atom 4	Torsion
1	C27	N1	C2	C3	180.00
2	C27	N1	C2	N51	0.00
3	C2	N1	C27	N26	0.00
4	C2	N1	C27	C28	180.00
5	N1	C2	C3	C4	0.00
6	N1	C2	C3	C8	180.00
7	N51	C2	C3	C4	180.00
8	N51	C2	C3	C8	0.00
9	N1	C2	N51	C52	0.00
10	C3	C2	N51	C52	180.00
11	C2	C3	C4	C5	180.00
12	C2	C3	C4	H17	0.00
13	C8	C3	C4	C5	0.00
14	C8	C3	C4	H17	180.00
15	C2	C3	C8	C7	180.00
16	C2	C3	C8	H20	0.00
17	C4	C3	C8	C7	0.00
18	C4	C3	C8	H20	180.00
19	C3	C4	C5	C6	0.00
20	C3	C4	C5	H18	180.00
21	H17	C4	C5	C6	180.00
22	H17	C4	C5	H18	0.00
23	C4	C5	C6	C7	0.00
24	C4	C5	C6	N9	180.00

25	H18	C5	C6	C7	180.00
26	H18	C5	C6	N9	0.00
27	C5	C6	C7	C8	0.00
28	C5	C6	C7	H19	180.00
29	N9	C6	C7	C8	180.00
30	N9	C6	C7	H19	0.00
31	C5	C6	N9	C10	0.00
32	C7	C6	N9	C10	180.00
33	C6	C7	C8	C3	0.00
34	C6	C7	C8	H20	180.00
35	H19	C7	C8	C3	180.00
36	H19	C7	C8	H20	0.00
37	C6	N9	C10	C11	180.00
38	C6	N9	C10	H25	0.00
39	N9	C10	C11	C12	0.00
40	N9	C10	C11	C16	180.00
41	H25	C10	C11	C12	180.00
42	H25	C10	C11	C16	0.00
43	C10	C11	C12	C13	180.00
44	C10	C11	C12	H22	0.00
45	C16	C11	C12	C13	0.00
46	C16	C11	C12	H22	180.00
47	C10	C11	C16	C15	180.00
48	C10	C11	C16	H24	0.00
49	C12	C11	C16	C15	0.00
50	C12	C11	C16	H24	180.00
51	C11	C12	C13	C14	0.00
52	C11	C12	C13	H21	180.00
53	H22	C12	C13	C14	180.00
54	H22	C12	C13	H21	0.00
55	C12	C13	C14	C15	0.00
56	C12	C13	C14	C89	180.00
57	H21	C13	C14	C15	180.00
58	H21	C13	C14	C89	0.00
59	C13	C14	C15	C16	0.00
60	C13	C14	C15	H23	180.00
61	C89	C14	C15	C16	180.00
62	C89	C14	C15	H23	0.00
63	C14	C15	C16	C11	0.00
64	C14	C15	C16	H24	180.00
65	H23	C15	C16	C11	180.00
66	H23	C15	C16	H24	0.00
67	C52	N26	C27	N1	0.00
68	C52	N26	C27	C28	180.00
69	C27	N26	C52	N51	0.00
70	C27	N26	C52	C53	180.00
71	N1	C27	C28	C29	180.00
72	N1	C27	C28	C33	0.00
73	N26	C27	C28	C29	0.00
74	N26	C27	C28	C33	180.00
75	C27	C28	C29	C30	180.00
76	C27	C28	C29	H42	0.00
77	C33	C28	C29	C30	0.00
78	C33	C28	C29	H42	180.00
79	C27	C28	C33	C32	180.00
80	C27	C28	C33	H45	0.00

81	C29	C28	C33	C32	0.00
82	C29	C28	C33	H45	180.00
83	C28	C29	C30	C31	0.00
84	C28	C29	C30	H43	180.00
85	H42	C29	C30	C31	180.00
86	H42	C29	C30	H43	0.00
87	C29	C30	C31	C32	0.00
88	C29	C30	C31	N34	180.00
89	H43	C30	C31	C32	180.00
90	H43	C30	C31	N34	0.00
91	C30	C31	C32	C33	0.00
92	C30	C31	C32	H44	180.00
93	N34	C31	C32	C33	180.00
94	N34	C31	C32	H44	0.00
95	C30	C31	N34	C35	0.00
96	C32	C31	N34	C35	180.00
97	C31	C32	C33	C28	0.00
98	C31	C32	C33	H45	180.00
99	H44	C32	C33	C28	180.00
100	H44	C32	C33	H45	0.00
101	C31	N34	C35	C36	180.00
102	C31	N34	C35	H50	0.00
103	N34	C35	C36	C37	0.00
104	N34	C35	C36	C41	180.00
105	H50	C35	C36	C37	180.00
106	H50	C35	C36	C41	0.00
107	C35	C36	C37	C38	180.00
108	C35	C36	C37	H47	0.00
109	C41	C36	C37	C38	0.00
110	C41	C36	C37	H47	180.00
111	C35	C36	C41	C40	180.00
112	C35	C36	C41	H49	0.00
113	C37	C36	C41	C40	0.00
114	C37	C36	C41	H49	180.00
115	C36	C37	C38	C39	0.00
116	C36	C37	C38	H46	180.00
117	H47	C37	C38	C39	180.00
118	H47	C37	C38	H46	0.00
119	C37	C38	C39	C40	0.00
120	C37	C38	C39	C114	180.00
121	H46	C38	C39	C40	180.00
122	H46	C38	C39	C114	0.00
123	C38	C39	C40	C41	0.00
124	C38	C39	C40	H48	180.00
125	C114	C39	C40	C41	180.00
126	C114	C39	C40	H48	0.00
127	C39	C40	C41	C36	0.00
128	C39	C40	C41	H49	180.00
129	H48	C40	C41	C36	180.00
130	H48	C40	C41	H49	0.00
131	C2	N51	C52	N26	0.00
132	C2	N51	C52	C53	180.00
133	N26	C52	C53	C54	180.00
134	N26	C52	C53	C58	0.00
135	N51	C52	C53	C54	0.00
136	N51	C52	C53	C58	180.00

137	C52	C53	C54	C55	180.00
138	C52	C53	C54	H67	0.00
139	C58	C53	C54	C55	0.00
140	C58	C53	C54	H67	180.00
141	C52	C53	C58	C57	180.00
142	C52	C53	C58	H70	0.00
143	C54	C53	C58	C57	0.00
144	C54	C53	C58	H70	180.00
145	C53	C54	C55	C56	0.00
146	C53	C54	C55	H68	180.00
147	H67	C54	C55	C56	180.00
148	H67	C54	C55	H68	0.00
149	C54	C55	C56	C57	0.00
150	C54	C55	C56	N59	180.00
151	H68	C55	C56	C57	180.00
152	H68	C55	C56	N59	0.00
153	C55	C56	C57	C58	0.00
154	C55	C56	C57	H69	180.00
155	N59	C56	C57	C58	180.00
156	N59	C56	C57	H69	0.00
157	C55	C56	N59	C60	0.00
158	C57	C56	N59	C60	180.00
159	C56	C57	C58	C53	0.00
160	C56	C57	C58	H70	180.00
161	H69	C57	C58	C53	180.00
162	H69	C57	C58	H70	0.00
163	C56	N59	C60	C61	180.00
164	C56	N59	C60	H75	0.00
165	N59	C60	C61	C62	0.00
166	N59	C60	C61	C66	180.00
167	H75	C60	C61	C62	180.00
168	H75	C60	C61	C66	0.00
169	C60	C61	C62	C63	180.00
170	C60	C61	C62	H72	0.00
171	C66	C61	C62	C63	0.00
172	C66	C61	C62	H72	180.00
173	C60	C61	C66	C65	180.00
174	C60	C61	C66	H74	0.00
175	C62	C61	C66	C65	0.00
176	C62	C61	C66	H74	180.00
177	C61	C62	C63	C64	0.00
178	C61	C62	C63	H71	180.00
179	H72	C62	C63	C64	180.00
180	H72	C62	C63	H71	0.00
181	C62	C63	C64	C65	0.00
182	C62	C63	C64	C139	180.00
183	H71	C63	C64	C65	180.00
184	H71	C63	C64	C139	0.00
185	C63	C64	C65	C66	0.00
186	C63	C64	C65	H73	180.00
187	C139	C64	C65	C66	180.00
188	C139	C64	C65	H73	0.00
189	C63	C64	C139	C138	180.00
190	C63	C64	C139	C140	0.00
191	C65	C64	C139	C138	0.00
192	C65	C64	C139	C140	180.00

193	C64	C65	C66	C61	0.00
194	C64	C65	C66	H74	180.00
195	H73	C65	C66	C61	180.00
196	H73	C65	C66	H74	0.00
197	C102	N76	C77	C78	180.00
198	C102	N76	C77	N126	0.00
199	C77	N76	C102	N101	0.00
200	C77	N76	C102	C103	180.00
201	N76	C77	C78	C79	0.00
202	N76	C77	C78	C83	180.00
203	N126	C77	C78	C79	180.00
204	N126	C77	C78	C83	0.00
205	N76	C77	N126	C127	0.00
206	C78	C77	N126	C127	180.00
207	C77	C78	C79	C80	180.00
208	C77	C78	C79	H92	0.00
209	C83	C78	C79	C80	0.00
210	C83	C78	C79	H92	180.00
211	C77	C78	C83	C82	180.00
212	C77	C78	C83	H95	0.00
213	C79	C78	C83	C82	0.00
214	C79	C78	C83	H95	180.00
215	C78	C79	C80	C81	0.00
216	C78	C79	C80	H93	180.00
217	H92	C79	C80	C81	180.00
218	H92	C79	C80	H93	0.00
219	C79	C80	C81	C82	0.00
220	C79	C80	C81	N84	180.00
221	H93	C80	C81	C82	180.00
222	H93	C80	C81	N84	0.00
223	C80	C81	C82	C83	0.00
224	C80	C81	C82	H94	180.00
225	N84	C81	C82	C83	180.00
226	N84	C81	C82	H94	0.00
227	C80	C81	N84	C85	0.00
228	C82	C81	N84	C85	180.00
229	C81	C82	C83	C78	0.00
230	C81	C82	C83	H95	180.00
231	H94	C82	C83	C78	180.00
232	H94	C82	C83	H95	0.00
233	C81	N84	C85	C86	180.00
234	C81	N84	C85	H100	0.00
235	N84	C85	C86	C87	0.00
236	N84	C85	C86	C91	180.00
237	H100	C85	C86	C87	180.00
238	H100	C85	C86	C91	0.00
239	C85	C86	C87	C88	180.00
240	C85	C86	C87	H97	0.00
241	C91	C86	C87	C88	0.00
242	C91	C86	C87	H97	180.00
243	C85	C86	C91	C90	180.00
244	C85	C86	C91	H99	0.00
245	C87	C86	C91	C90	0.00
246	C87	C86	C91	H99	180.00
247	C86	C87	C88	C89	0.00
248	C86	C87	C88	H96	180.00

249	H97	C87	C88	C89	180.00
250	H97	C87	C88	H96	0.00
251	C87	C88	C89	C90	0.00
252	C87	C88	C89	C14	180.00
253	H96	C88	C89	C90	180.00
254	H96	C88	C89	C14	0.00
255	C88	C89	C90	C91	0.00
256	C88	C89	C90	H98	180.00
257	C14	C89	C90	C91	180.00
258	C14	C89	C90	H98	0.00
259	C89	C90	C91	C86	0.00
260	C89	C90	C91	H99	180.00
261	H98	C90	C91	C86	180.00
262	H98	C90	C91	H99	0.00
263	C127	N101	C102	N76	0.00
264	C127	N101	C102	C103	180.00
265	C102	N101	C127	N126	0.00
266	C102	N101	C127	C128	180.00
267	N76	C102	C103	C104	180.00
268	N76	C102	C103	C108	0.00
269	N101	C102	C103	C104	0.00
270	N101	C102	C103	C108	180.00
271	C102	C103	C104	C105	180.00
272	C102	C103	C104	H117	0.00
273	C108	C103	C104	C105	0.00
274	C108	C103	C104	H117	180.00
275	C102	C103	C108	C107	180.00
276	C102	C103	C108	H120	0.00
277	C104	C103	C108	C107	0.00
278	C104	C103	C108	H120	180.00
279	C103	C104	C105	C106	0.00
280	C103	C104	C105	H118	180.00
281	H117	C104	C105	C106	180.00
282	H117	C104	C105	H118	0.00
283	C104	C105	C106	C107	0.00
284	C104	C105	C106	N109	180.00
285	H118	C105	C106	C107	180.00
286	H118	C105	C106	N109	0.00
287	C105	C106	C107	C108	0.00
288	C105	C106	C107	H119	180.00
289	N109	C106	C107	C108	180.00
290	N109	C106	C107	H119	0.00
291	C105	C106	N109	C110	0.00
292	C107	C106	N109	C110	180.00
293	C106	C107	C108	C103	0.00
294	C106	C107	C108	H120	180.00
295	H119	C107	C108	C103	180.00
296	H119	C107	C108	H120	0.00
297	C106	N109	C110	C111	180.00
298	C106	N109	C110	H125	0.00
299	N109	C110	C111	C112	0.00
300	N109	C110	C111	C116	180.00
301	H125	C110	C111	C112	180.00
302	H125	C110	C111	C116	0.00
303	C110	C111	C112	C113	180.00
304	C110	C111	C112	H122	0.00

305	C116	C111	C112	C113	0.00
306	C116	C111	C112	H122	180.00
307	C110	C111	C116	C115	180.00
308	C110	C111	C116	H124	0.00
309	C112	C111	C116	C115	0.00
310	C112	C111	C116	H124	180.00
311	C111	C112	C113	C114	0.00
312	C111	C112	C113	H121	180.00
313	H122	C112	C113	C114	180.00
314	H122	C112	C113	H121	0.00
315	C112	C113	C114	C115	0.00
316	C112	C113	C114	C39	180.00
317	H121	C113	C114	C115	180.00
318	H121	C113	C114	C39	0.00
319	C113	C114	C115	C116	0.00
320	C113	C114	C115	H123	180.00
321	C39	C114	C115	C116	180.00
322	C39	C114	C115	H123	0.00
323	C114	C115	C116	C111	0.00
324	C114	C115	C116	H124	180.00
325	H123	C115	C116	C111	180.00
326	H123	C115	C116	H124	0.00
327	C77	N126	C127	N101	0.00
328	C77	N126	C127	C128	180.00
329	N101	C127	C128	C129	180.00
330	N101	C127	C128	C133	0.00
331	N126	C127	C128	C129	0.00
332	N126	C127	C128	C133	180.00
333	C127	C128	C129	C130	180.00
334	C127	C128	C129	H142	0.00
335	C133	C128	C129	C130	0.00
336	C133	C128	C129	H142	180.00
337	C127	C128	C133	C132	180.00
338	C127	C128	C133	H145	0.00
339	C129	C128	C133	C132	0.00
340	C129	C128	C133	H145	180.00
341	C128	C129	C130	C131	0.00
342	C128	C129	C130	H143	180.00
343	H142	C129	C130	C131	180.00
344	H142	C129	C130	H143	0.00
345	C129	C130	C131	C132	0.00
346	C129	C130	C131	N134	180.00
347	H143	C130	C131	C132	180.00
348	H143	C130	C131	N134	0.00
349	C130	C131	C132	C133	0.00
350	C130	C131	C132	H144	180.00
351	N134	C131	C132	C133	180.00
352	N134	C131	C132	H144	0.00
353	C130	C131	N134	C135	0.00
354	C132	C131	N134	C135	180.00
355	C131	C132	C133	C128	0.00
356	C131	C132	C133	H145	180.00
357	H144	C132	C133	C128	180.00
358	H144	C132	C133	H145	0.00
359	C131	N134	C135	C136	180.00
360	C131	N134	C135	H150	0.00

361	N134	C135	C136	C137	0.00
362	N134	C135	C136	C141	180.00
363	H150	C135	C136	C137	180.00
364	H150	C135	C136	C141	0.00
365	C135	C136	C137	C138	180.00
366	C135	C136	C137	H147	0.00
367	C141	C136	C137	C138	0.00
368	C141	C136	C137	H147	180.00
369	C135	C136	C141	C140	180.00
370	C135	C136	C141	H149	0.00
371	C137	C136	C141	C140	0.00
372	C137	C136	C141	H149	180.00
373	C136	C137	C138	C139	0.00
374	C136	C137	C138	H146	180.00
375	H147	C137	C138	C139	180.00
376	H147	C137	C138	H146	0.00
377	C137	C138	C139	C64	180.00
378	C137	C138	C139	C140	0.00
379	H146	C138	C139	C64	0.00
380	H146	C138	C139	C140	180.00
381	C64	C139	C140	C141	180.00
382	C64	C139	C140	H148	0.00
383	C138	C139	C140	C141	0.00
384	C138	C139	C140	H148	180.00
385	C139	C140	C141	C136	0.00
386	C139	C140	C141	H149	180.00
387	H148	C140	C141	C136	180.00
388	H148	C140	C141	H149	0.00

Table S4. List of all angles present in the COF (TRITER-2)

No	Atom1	Atom2	Atom3	Angle
1	C2	N1	C27	120.74
2	N1	C2	C3	120.37
3	N1	C2	N51	119.26
4	C3	C2	N51	120.37
5	C2	C3	C4	120.64
6	C2	C3	C8	120.61
7	C4	C3	C8	118.75
8	C3	C4	C5	120.58
9	C3	C4	H17	119.42
10	C5	C4	H17	120.00
11	C4	C5	C6	120.46
12	C4	C5	H18	120.00
13	C6	C5	H18	119.54
14	C5	C6	C7	119.12

15	C5	C6	N9	121.29
16	C7	C6	N9	119.59
17	C6	C7	C8	120.47
18	C6	C7	H19	119.99
19	C8	C7	H19	119.54
20	C3	C8	C7	120.62
21	C3	C8	H20	119.38
22	C7	C8	H20	120.00
23	C6	N9	C10	120.66
24	N9	C10	C11	127.29
25	N9	C10	H25	119.99
26	C11	C10	H25	112.73
27	C10	C11	C12	123.37
28	C10	C11	C16	117.93
29	C12	C11	C16	118.70
30	C11	C12	C13	120.57
31	C11	C12	H22	119.44
32	C13	C12	H22	120.00
33	C12	C13	C14	121.54
34	C12	C13	H21	120.00
35	C14	C13	H21	118.45
36	C13	C14	C15	116.75
37	C13	C14	C89	121.62
38	C15	C14	C89	121.63
39	C14	C15	C16	121.37
40	C14	C15	H23	119.99
41	C16	C15	H23	118.64
42	C11	C16	C15	121.07
43	C11	C16	H24	118.91
44	C15	C16	H24	120.02
45	C27	N26	C52	120.74
46	N1	C27	N26	119.26
47	N1	C27	C28	120.37
48	N26	C27	C28	120.37

49	C27	C28	C29	120.64
50	C27	C28	C33	120.61
51	C29	C28	C33	118.75
52	C28	C29	C30	120.58
53	C28	C29	H42	119.42
54	C30	C29	H42	120.00
55	C29	C30	C31	120.46
56	C29	C30	H43	120.00
57	C31	C30	H43	119.54
58	C30	C31	C32	119.12
59	C30	C31	N34	121.29
60	C32	C31	N34	119.59
61	C31	C32	C33	120.47
62	C31	C32	H44	119.99
63	C33	C32	H44	119.54
64	C28	C33	C32	120.62
65	C28	C33	H45	119.38
66	C32	C33	H45	120.00
67	C31	N34	C35	120.66
68	N34	C35	C36	127.29
69	N34	C35	H50	119.99
70	C36	C35	H50	112.73
71	C35	C36	C37	123.37
72	C35	C36	C41	117.93
73	C37	C36	C41	118.70
74	C36	C37	C38	120.57
75	C36	C37	H47	119.44
76	C38	C37	H47	120.00
77	C37	C38	C39	121.54
78	C37	C38	H46	120.00
79	C39	C38	H46	118.45
80	C38	C39	C40	116.75
81	C38	C39	C114	121.62
82	C40	C39	C114	121.63

83	C39	C40	C41	121.37
84	C39	C40	H48	119.99
85	C41	C40	H48	118.64
86	C36	C41	C40	121.07
87	C36	C41	H49	118.91
88	C40	C41	H49	120.02
89	C2	N51	C52	120.74
90	N26	C52	N51	119.26
91	N26	C52	C53	120.37
92	N51	C52	C53	120.37
93	C52	C53	C54	120.64
94	C52	C53	C58	120.61
95	C54	C53	C58	118.75
96	C53	C54	C55	120.58
97	C53	C54	H67	119.42
98	C55	C54	H67	120.00
99	C54	C55	C56	120.46
100	C54	C55	H68	120.00
101	C56	C55	H68	119.54
102	C55	C56	C57	119.12
103	C55	C56	N59	121.29
104	C57	C56	N59	119.59
105	C56	C57	C58	120.47
106	C56	C57	H69	119.99
107	C58	C57	H69	119.54
108	C53	C58	C57	120.62
109	C53	C58	H70	119.38
110	C57	C58	H70	120.00
111	C56	N59	C60	120.66
112	N59	C60	C61	127.29
113	N59	C60	H75	119.99
114	C61	C60	H75	112.73
115	C60	C61	C62	123.37
116	C60	C61	C66	117.93

117	C62	C61	C66	118.70
118	C61	C62	C63	120.57
119	C61	C62	H72	119.44
120	C63	C62	H72	120.00
121	C62	C63	C64	121.54
122	C62	C63	H71	120.00
123	C64	C63	H71	118.45
124	C63	C64	C65	116.75
125	C63	C64	C139	121.62
126	C65	C64	C139	121.63
127	C64	C65	C66	121.37
128	C64	C65	H73	119.99
129	C66	C65	H73	118.64
130	C61	C66	C65	121.07
131	C61	C66	H74	118.91
132	C65	C66	H74	120.02
133	C77	N76	C102	120.74
134	N76	C77	C78	120.37
135	N76	C77	N126	119.26
136	C78	C77	N126	120.37
137	C77	C78	C79	120.64
138	C77	C78	C83	120.61
139	C79	C78	C83	118.75
140	C78	C79	C80	120.58
141	C78	C79	H92	119.42
142	C80	C79	H92	120.00
143	C79	C80	C81	120.46
144	C79	C80	H93	120.00
145	C81	C80	H93	119.54
146	C80	C81	C82	119.12
147	C80	C81	N84	121.29
148	C82	C81	N84	119.59
149	C81	C82	C83	120.47
150	C81	C82	H94	119.99

151	C83	C82	H94	119.54
152	C78	C83	C82	120.62
153	C78	C83	H95	119.38
154	C82	C83	H95	120.00
155	C81	N84	C85	120.66
156	N84	C85	C86	127.29
157	N84	C85	H100	119.99
158	C86	C85	H100	112.73
159	C85	C86	C87	123.37
160	C85	C86	C91	117.93
161	C87	C86	C91	118.70
162	C86	C87	C88	120.57
163	C86	C87	H97	119.44
164	C88	C87	H97	120.00
165	C87	C88	C89	121.54
166	C87	C88	H96	120.00
167	C89	C88	H96	118.45
168	C88	C89	C90	116.75
169	C88	C89	C14	121.62
170	C90	C89	C14	121.63
171	C89	C90	C91	121.37
172	C89	C90	H98	119.99
173	C91	C90	H98	118.64
174	C86	C91	C90	121.07
175	C86	C91	H99	118.91
176	C90	C91	H99	120.02
177	C102	N101	C127	120.74
178	N76	C102	N101	119.26
179	N76	C102	C103	120.37
180	N101	C102	C103	120.37
181	C102	C103	C104	120.64
182	C102	C103	C108	120.61
183	C104	C103	C108	118.75
184	C103	C104	C105	120.58

185	C103	C104	H117	119.42
186	C105	C104	H117	120.00
187	C104	C105	C106	120.46
188	C104	C105	H118	120.00
189	C106	C105	H118	119.54
190	C105	C106	C107	119.12
191	C105	C106	N109	121.29
192	C107	C106	N109	119.59
193	C106	C107	C108	120.47
194	C106	C107	H119	119.99
195	C108	C107	H119	119.54
196	C103	C108	C107	120.62
197	C103	C108	H120	119.38
198	C107	C108	H120	120.00
199	C106	N109	C110	120.66
200	N109	C110	C111	127.29
201	N109	C110	H125	119.99
202	C111	C110	H125	112.73
203	C110	C111	C112	123.37
204	C110	C111	C116	117.93
205	C112	C111	C116	118.70
206	C111	C112	C113	120.57
207	C111	C112	H122	119.44
208	C113	C112	H122	120.00
209	C112	C113	C114	121.54
210	C112	C113	H121	120.00
211	C114	C113	H121	118.45
212	C113	C114	C115	116.75
213	C113	C114	C39	121.62
214	C115	C114	C39	121.63
215	C114	C115	C116	121.37
216	C114	C115	H123	119.99
217	C116	C115	H123	118.64
218	C111	C116	C115	121.07

219	C111	C116	H124	118.91
220	C115	C116	H124	120.02
221	C77	N126	C127	120.74
222	N101	C127	N126	119.26
223	N101	C127	C128	120.37
224	N126	C127	C128	120.37
225	C127	C128	C129	120.64
226	C127	C128	C133	120.61
227	C129	C128	C133	118.75
228	C128	C129	C130	120.58
229	C128	C129	H142	119.42
230	C130	C129	H142	120.00
231	C129	C130	C131	120.46
232	C129	C130	H143	120.00
233	C131	C130	H143	119.54
234	C130	C131	C132	119.12
235	C130	C131	N134	121.29
236	C132	C131	N134	119.59
237	C131	C132	C133	120.47
238	C131	C132	H144	119.99
239	C133	C132	H144	119.54
240	C128	C133	C132	120.62
241	C128	C133	H145	119.38
242	C132	C133	H145	120.00
243	C131	N134	C135	120.66
244	N134	C135	C136	127.29
245	N134	C135	H150	119.99
246	C136	C135	H150	112.73
247	C135	C136	C137	123.37
248	C135	C136	C141	117.93
249	C137	C136	C141	118.70
250	C136	C137	C138	120.57
251	C136	C137	H147	119.44
252	C138	C137	H147	120.00

253	C137	C138	C139	121.54
254	C137	C138	H146	120.00
255	C139	C138	H146	118.45
256	C64	C139	C138	121.62
257	C64	C139	C140	121.63
258	C138	C139	C140	116.75
259	C139	C140	C141	121.37
260	C139	C140	H148	119.99
261	C141	C140	H148	118.64
262	C136	C141	C140	121.07
263	C136	C141	H149	118.91
264	C140	C141	H149	120.02

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