## **Supporting Information:**

Construction of Unique Heterojunction Photoanodes through *in-situ* Quasi-Epitaxial Growth of FeVO<sub>4</sub> on Fe<sub>2</sub>O<sub>3</sub> Nanorod Arrays for Enhanced Photoelectrochemical Performance

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Figure S1. Color diagram of the photoanodes.

Figure S1 is the color schematic diagram of the prisitine  $Fe_2O_3$  photoanode (a), FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode with 7 mM vanadyl acetylacetonate (b) and FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode with 20 mM vanadyl acetylacetonate (c) . It can be seen that as the concentration of vanadyl acetylacetonate increases, the color of the photoelectrode also changes. Compared with the prisitine Fe<sub>2</sub>O<sub>3</sub> photoanode, the FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode with 20 mM vanadyl acetylacetonate has shown obvious color deepening.



Figure S2. EDS elemental mapping of the FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode.



Figure S3. TEM image of the FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode.

	Area of $Fe^{2+}2p_{3/2}$	Area of $Fe^{2+}2p_{1/2}$	Sum
FeVO <sub>4</sub> -Fe <sub>2</sub> O <sub>3</sub>	12106.05	5186.25	17292.3
photoanode			
Fe <sub>2</sub> O <sub>3</sub> photoanode	14262.23	3864.48	18126.71

Table S1. The  $Fe^{2+}elemental \ content \ of \ the \ photoanodes$ 



**Figure S4**. LSV curves of FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode of various concentrations of vanadyl acetylacetonate.

wavelength	400	435	450	475	500	550	600	650
	nm	nm	nm	nm	nm	nm	nm	nm
IPCE of the FeVO <sub>4</sub> -	22.67	15.51	11.03	6.76	4.93	2.43	0.95	0.67
Fe <sub>2</sub> O <sub>3</sub> photoanode (%)								
IPCE of the pristine	9.26	6.39	4.54	2.78	2.07	0.92	0.54	0.46
Fe <sub>2</sub> O <sub>3</sub> photoanode (%)								

 Table S2. The IPCE of the photoanodes



**Figure S5.** LSV curves of pristine  $Fe_2O_3$  photoanode and  $FeVO_4$ - $Fe_2O_3$  photoanode with and without hole scavenger.



Figure S6. EIS equivalent circuit diagram.

photoanode	<b>Rs</b> (Ω)	Rct ( $\Omega$ )
Fe <sub>2</sub> O <sub>3</sub>	11.01	329.9
FeVO <sub>4</sub> -Fe <sub>2</sub> O <sub>3</sub>	10.82	239.1

Figure S7. EIS parameters of pristine Fe<sub>2</sub>O<sub>3</sub> photoanode and FeVO<sub>4</sub>-Fe<sub>2</sub>O<sub>3</sub> photoanode.



Figure S8. SEM images of the  $FeVO_4$ - $Fe_2O_3$  photoanode (a) before and (b) after the stability test.



Figure S9. XRD image of the  $FeVO_4$ - $Fe_2O_3$  photoanode before and after the stability test.



Figure S10. (a) UV-vis diffuse reflectance spectra of pristine  $Fe_2O_3$  and  $FeVO_4$ - $Fe_2O_3$  photoanodes, (b) the Tauc plot of pristine  $Fe_2O_3$  photoanode.

The UV-vis results show the absorption curve of the  $FeVO_4$ - $Fe_2O_3$  photoanode is slightly improved in the whole wavelength range, compared with the curve intensity of the  $Fe_2O_3$  photoanode (Figure 5a). While the absorptive intensity and shift degree do not be improved significantly, but still demonstrate the effect of  $FeVO_4$ , and the insignificant lifting effect is attributed to the thinness of the  $FeVO_4$ . Formation of  $FeVO_4$ - $Fe_2O_3$  heterojunction could enable the prepared photoanode to utilize visible light efficaciously. The band gap could be calculated according to the following equation:

$$(\alpha h v)^n = A(h v - E_q) \tag{S1}$$

Where  $\alpha$ , hv, A, and  $E_{gare}$  the absorption coefficient, discrete photo energy, a constant and bandgap energy. The n value of Fe<sub>2</sub>O<sub>3</sub> and FeVO<sub>4</sub> are 2.<sup>1, 2</sup> The calculated result is displayed in Figure 5b, the band gap energies of pristine Fe<sub>2</sub>O<sub>3</sub> photoanode is about 2.2 eV, which is consistent with theoretical results.<sup>3</sup>



Figure S11. (a)MT result and (b)Tauc plot of pristine FeVO<sub>4</sub> photoanode.

The pristine FeVO<sub>4</sub> film was prepared on the basis of previously reported methods.<sup>4</sup> The preparation of FeOOH is the same as this work, the original FeVO<sub>4</sub> photoanode is prepared by drop casting and annealing process. First, an excess dimethyl sulfoxide (DMSO) solution containing 0.1 M vanadium acetylacetonate (VO (acac)2) is evenly coated on the FEOOH film. After drying on a plate heater at 75°C, it is placed in a muffle furnace, annealed in air at 550°C for 5 hours, and then cooled naturally. The obtained composite membrane was placed in a 0.1 M NaOH solution for 2 h to dissolve the excess vanadium oxide, then washed several times with distilled water, and dried at 60 °C. After that, a FeVO<sub>4</sub> photoanode was obtained.

- 1. J. Ma, Q. Wang, L. Li, X. Zong, H. Sun, R. Tao and X. Fan, *Journal of Colloid and Interface Science*, 2021, **602**, 32-42.
- 2. A. Chachvalvutikul, J. Jakmunee, S. Thongtem, S. Kittiwachana and S. Kaowphong, *Applied Surface Science*, 2019, **475**, 175-184.
- 3. G. Yang, S. Li, X. Wang, B. Ding, Y. Li, H. Lin, D. Tang, X. Ren, Q. Wang and S. Luo, *Applied Catalysis B: Environmental*, 2021, **297**, 120268.
- 4. Q. Zeng, X. Fu, S. Chang, Q. Zhang, Z. Xiong, Y. Liu, G. Peng and M. Li, *Journal of Colloid and Interface Science*, 2021, **604**, 562-567.