

**Shearing bridge bond in carbon nitride vesicles with enhanced hot
carrier utilization for photocatalytic hydrogen production**

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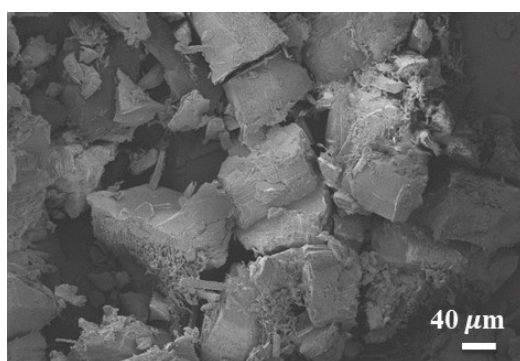


Figure S1 The SEM image of as-purchased melamine.

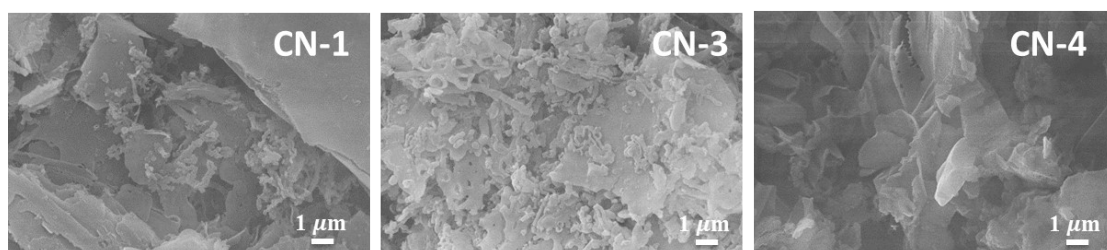


Figure S2 SEM images of other CN-x samples as indicated.

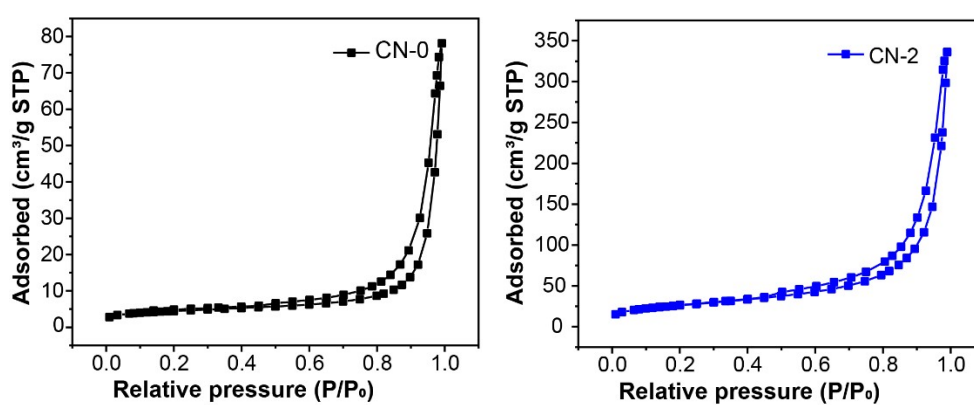


Figure S3 N_2 adsorption isothermals of CN-0 and CN-2 as indicated.

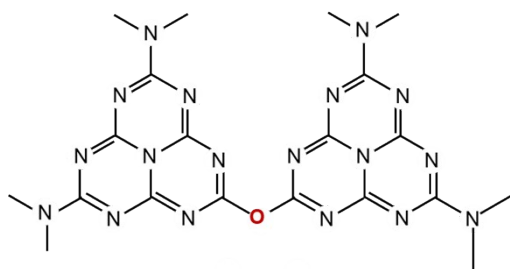


Figure S4 The corresponding bonding of C-O-C in g-C₃N₄ structure.

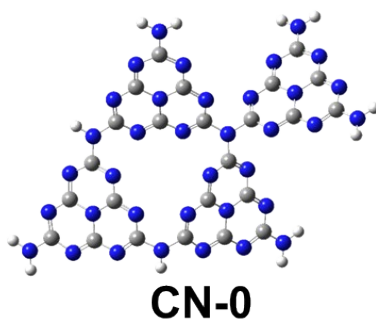
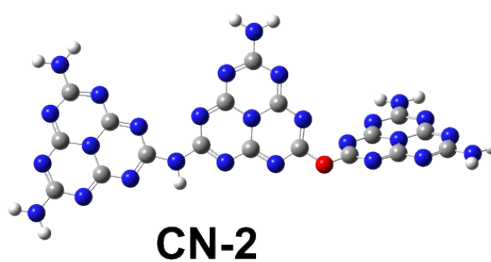
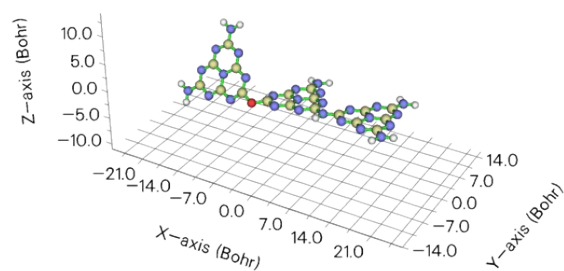
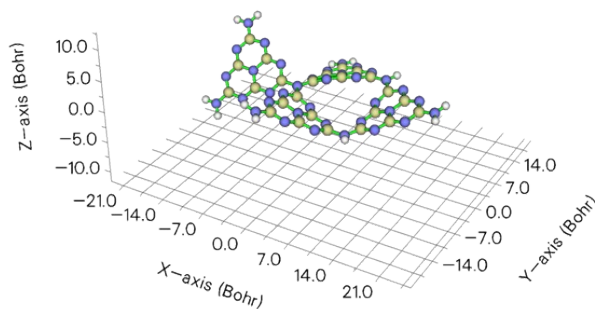


Figure S5 The optimized ground state geometry of CN-2 and CN-0 as indicated.



CN-2



CN-0

Figure S6 The illustrations of 3D analytical conformation of of CN-2 and CN-0 as indicated.

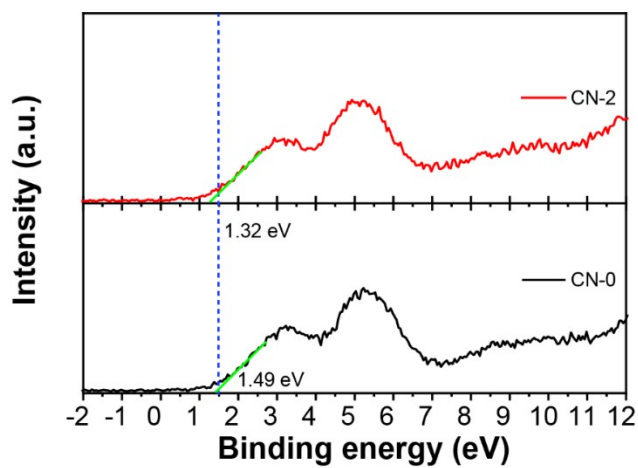


Figure S7 VB-XPS spectra of CN samples as indicated.

Table S1 Summary of reported g-C₃N₄ for H₂ evolution reaction.

Catalyst	Light source	H ₂ yield (μmol/h/g)	Experimental conditions	Reference
SCN nanosheets	λ>420 nm	111.5	1 wt% Pt 10% TEOA	¹
O-doping (g-C ₃ N ₄)	λ>420 nm	348	3 wt% Pt 10% TEOA	²
FAT-1.0	λ>420 nm	772	3 wt% Pt 10% TEOA	³
C defective-C ₃ N ₄	λ>420 nm	82.9	3 wt% Pt 10% TEOA	⁴
OH-CN3	λ>420 nm	310	1 wt% Pt 10% lactic acid	⁵
1D g-C ₃ N ₄ nanotubes	λ>420 nm	118.5	3 wt% Pt 10% TEOA	⁶
bulk P-g-C ₃ N ₄	λ>420 nm	600	1 wt% Pt 10% TEOA	⁷
C-doping C ₃ N ₄	λ>420 nm	666	1 wt% Pt 10% TEOA	⁸
CRed-AT-C ₃ N ₄	λ>420 nm	246.2	3 wt% Pt 10% TEOA	⁹
carbon nitride vesicles	λ>420 nm	608.3	1 wt% Pt 10% TEOA	This work

Supporting references

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