

## Supplementary Information

### Ionic Liquid-Mediated ZnO Shape-Controlled Nanocrystal-Supported Au Catalysts: Highly Stable Materials for aldehyde Oxidative esterification

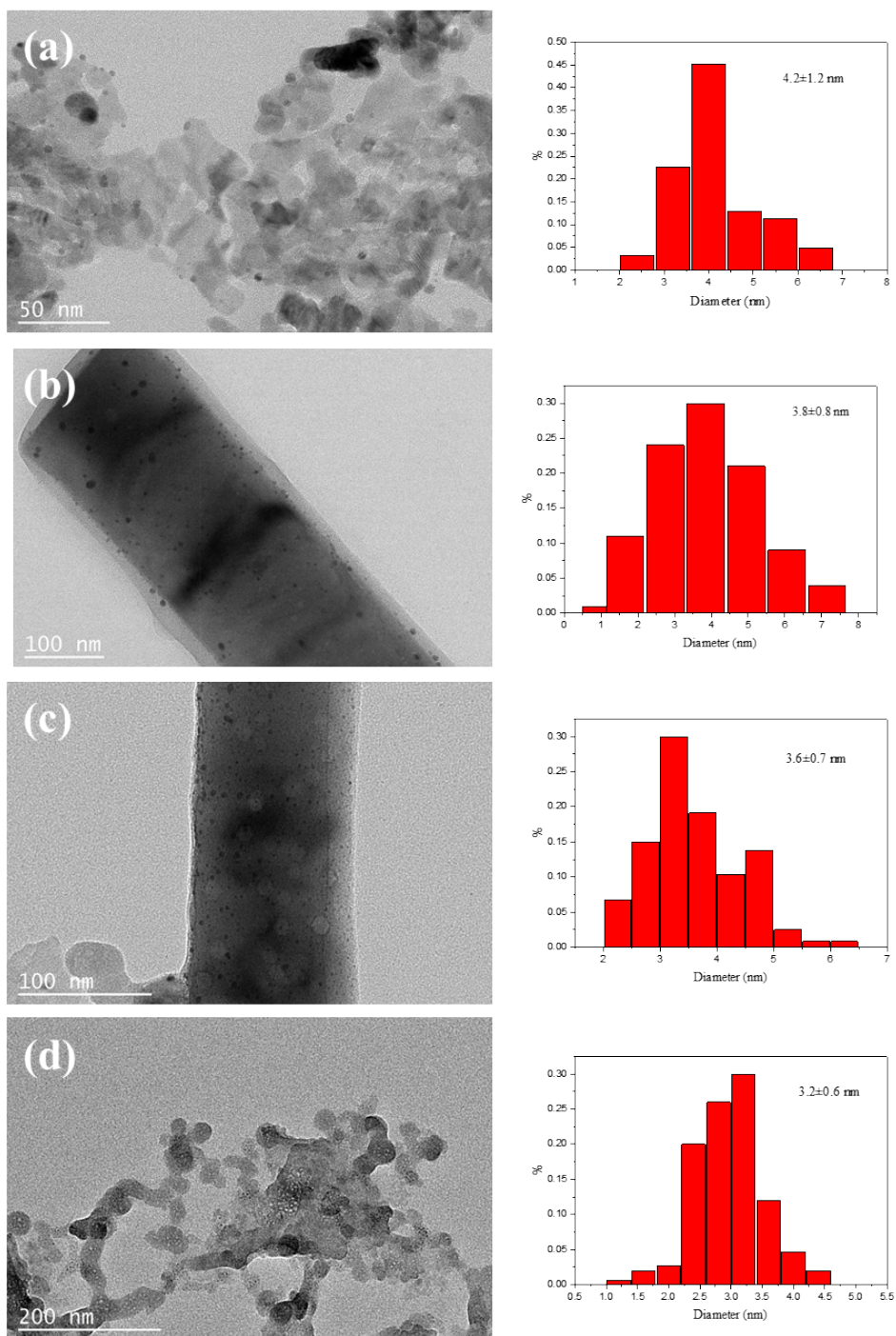
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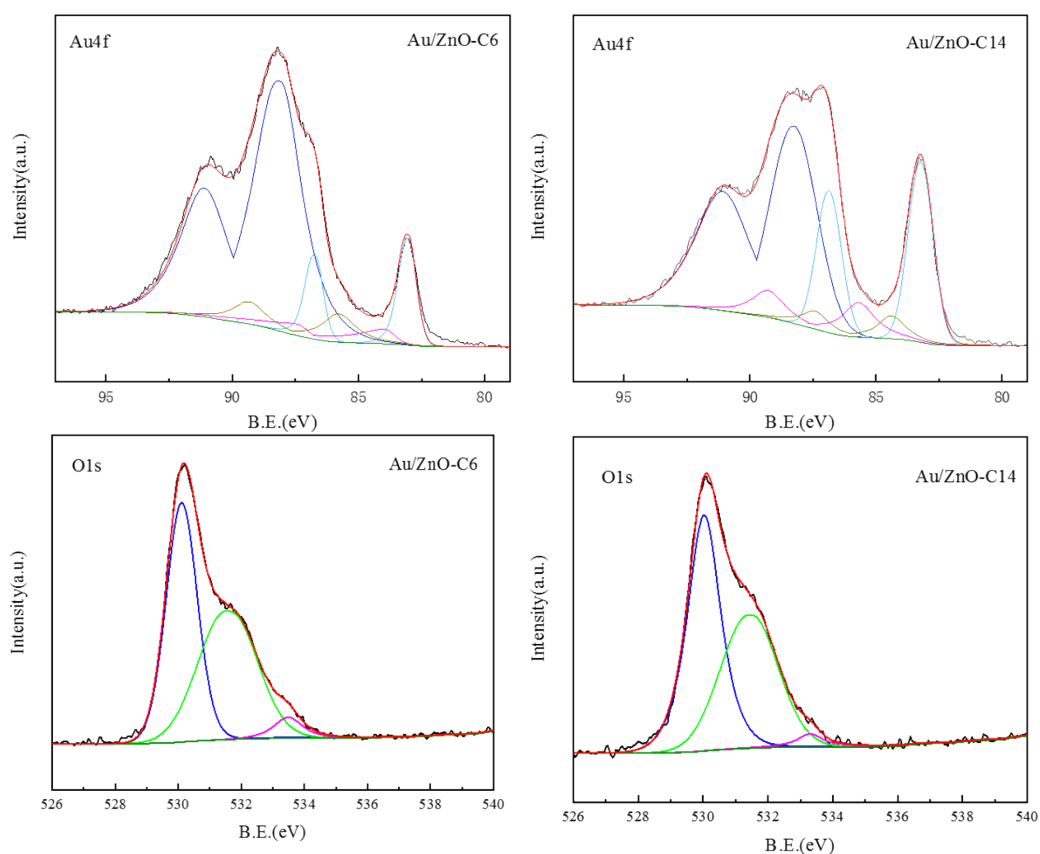
<sup>b</sup> School of Resources and Environmental Engineering, Shandong University of Technology, Zibo 255049, P.R. China

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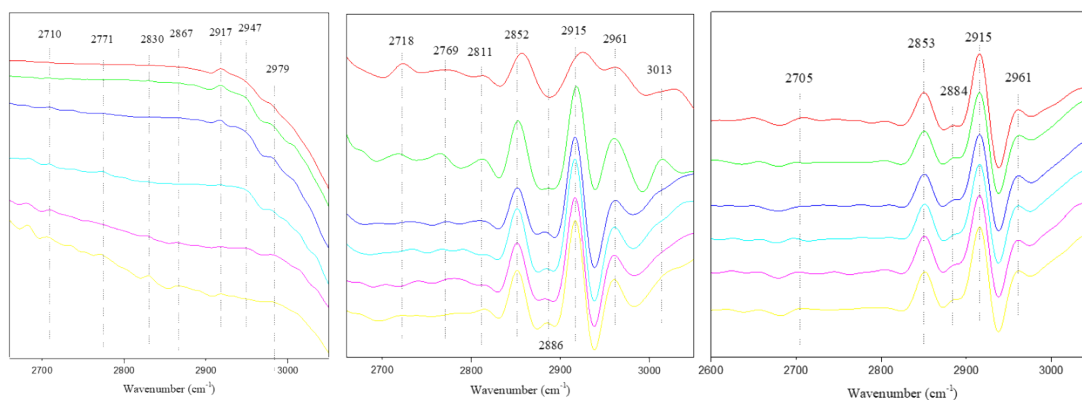
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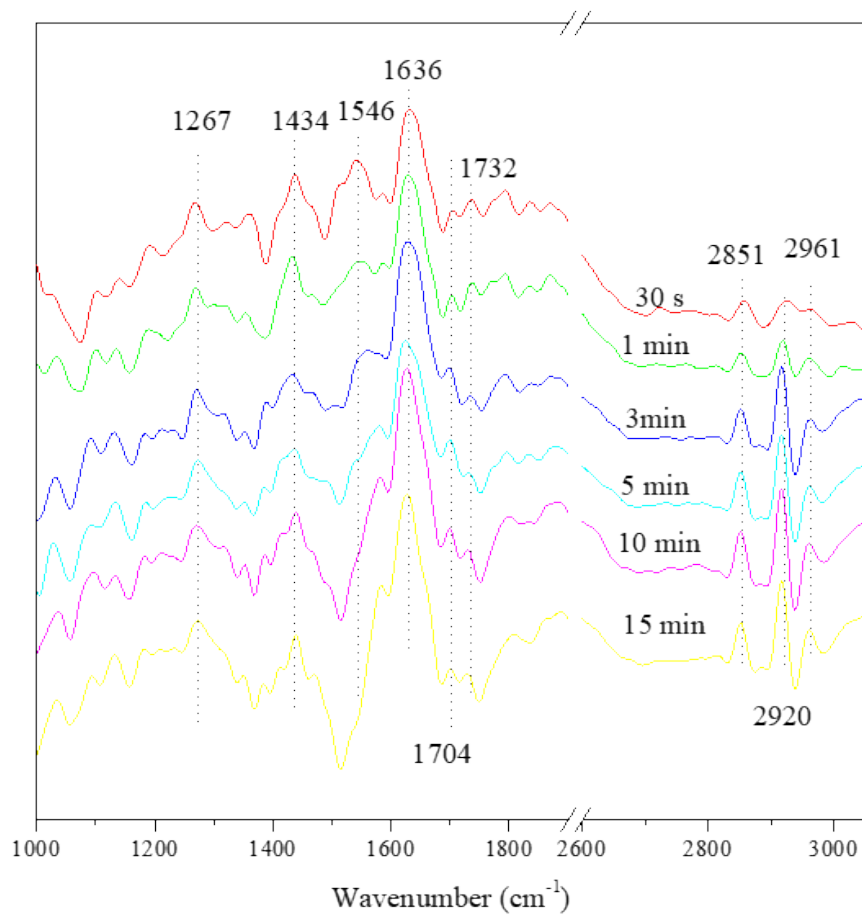
**Fig. S1.** The TEM images of catalysts and relevant Au particles size distribution: (a) Au/ZnO-C6; (b) Au/ZnO-C10; (c) Au/ZnO-C14; (d) Au/ZnO-CTAB.



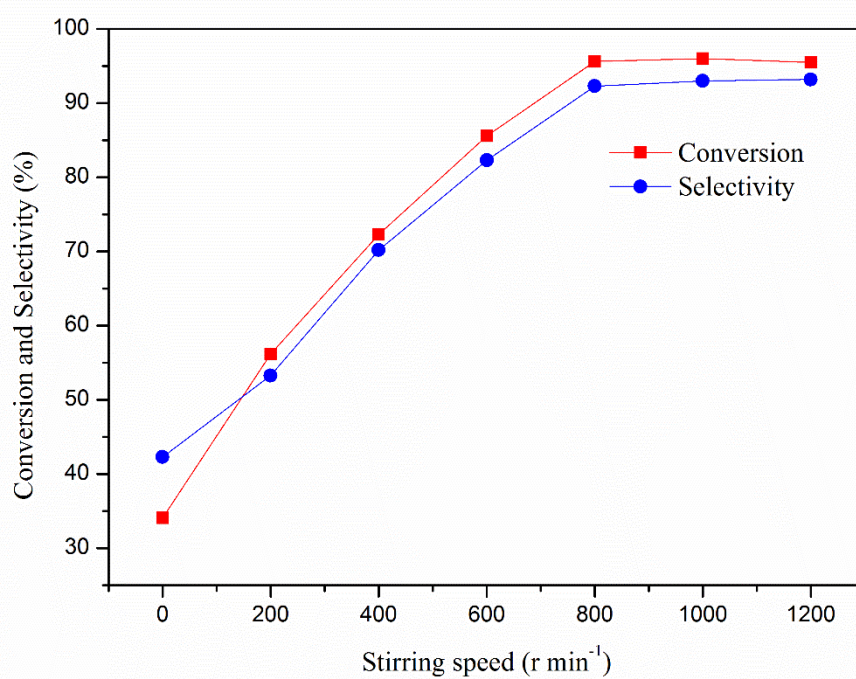
**Fig. S2.** XPS spectra of Au/ZnO-C6 and Au/ZnO-C14



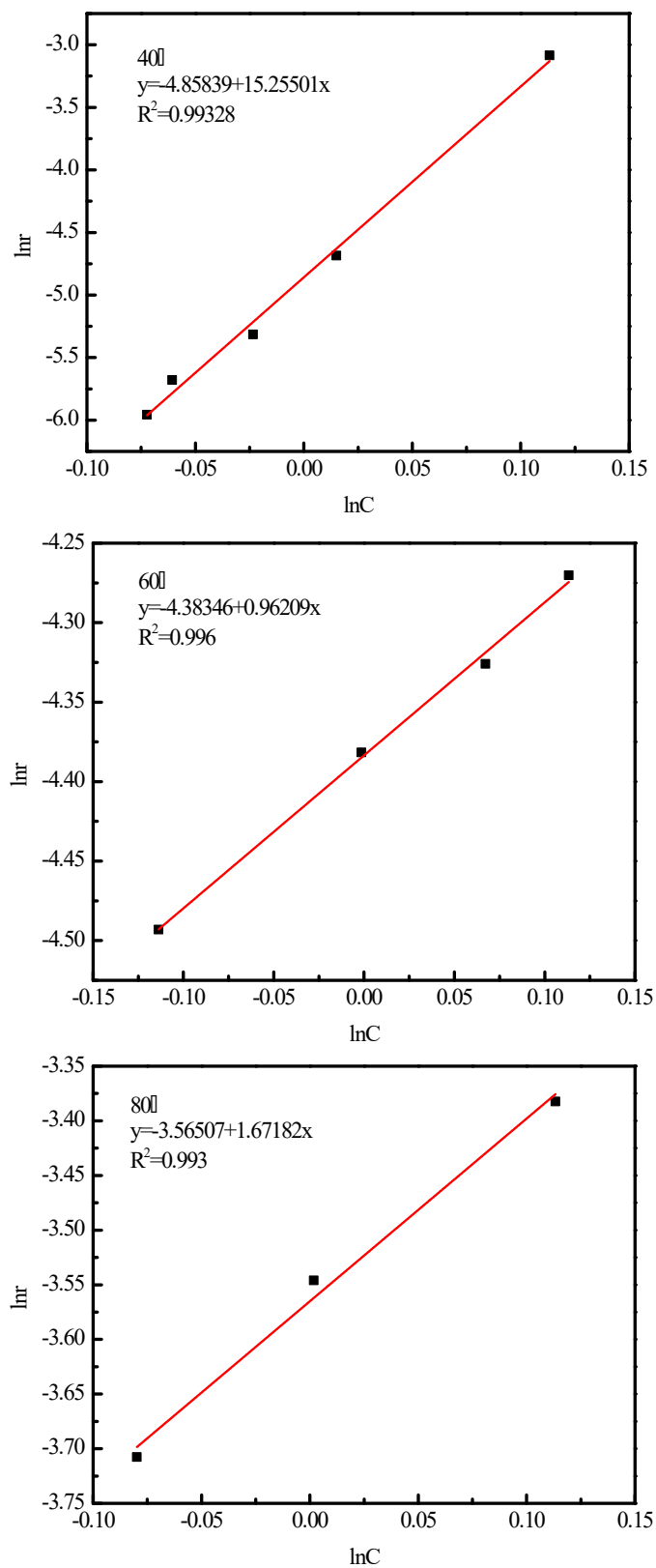
**Fig. S3.** In-situ FT-IR spectra of MAL and MeOH in presence of  $O_2$  at different temperatures over time on Au/ZnO-C10 in the region of 2700-3100  $cm^{-1}$ : (a) 40  $^{\circ}C$ ; (b) 60  $^{\circ}C$ ; (c) 80  $^{\circ}C$ .



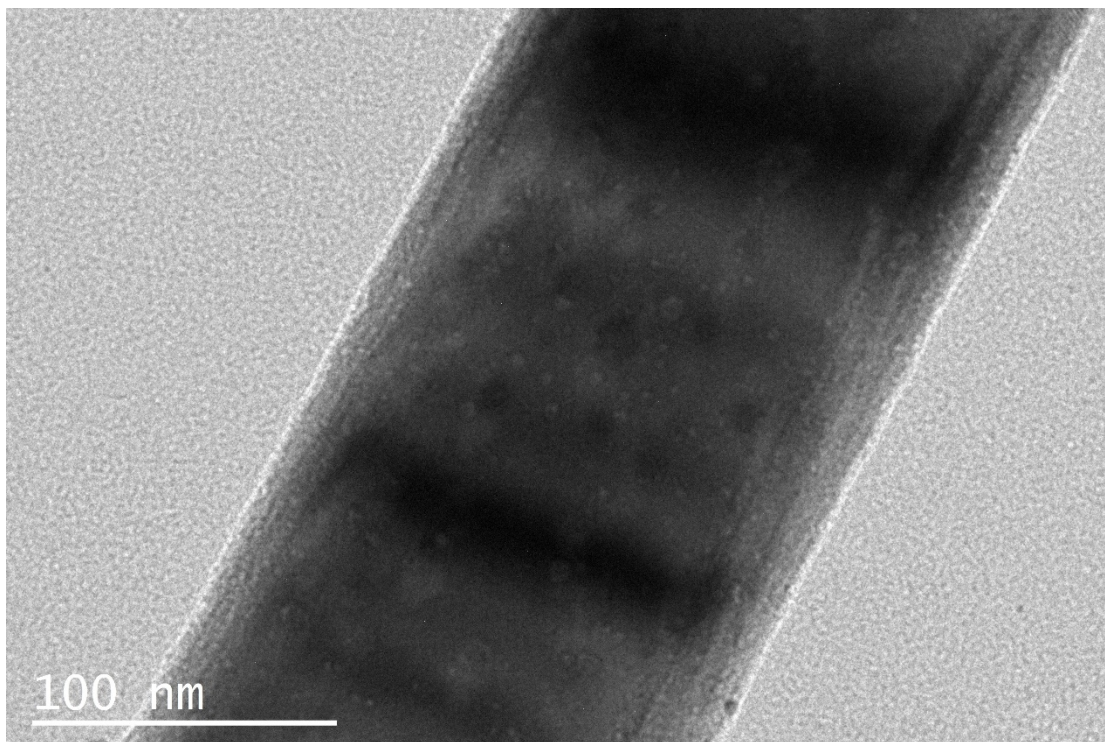
**Fig. S4.** In-situ FT-IR spectra of MAL and MeOH in absence of O<sub>2</sub> at different temperatures at 10 min on Au/ZnO-C10



**Fig. S5.** The effect of stirring speed



**Fig. S6.** Fitting curves between  $\ln r$  and  $\ln(CMAL)$  at 313 K, 333 K and 353 K when the conversion is  $< 15\%$ .



**Fig. S7.** TEM image of Au/ZnO-C10 after reaction

**Table S1** XPS analysis for Au/ZnO catalysts

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<b>Catalyst</b>	<b>O<sub>I</sub></b>		<b>O<sub>II</sub></b>		<b>O<sub>III</sub></b>	
	<b>BE(eV)</b>	<b>%</b>	<b>BE(eV)</b>	<b>%</b>	<b>BE(eV)</b>	<b>%</b>
<b>Au/ZnO-C6</b>	<b>530.11</b>	<b>48.72</b>	<b>531.51</b>	<b>47.03</b>	<b>533.48</b>	<b>4.25</b>
<b>Au/ZnO-C10</b>	<b>529.99</b>	<b>45.43</b>	<b>530.68</b>	<b>38.01</b>	<b>531.72</b>	<b>16.56</b>
<b>Au/ZnO-C14</b>	<b>530.03</b>	<b>51.64</b>	<b>531.40</b>	<b>45.09</b>	<b>533.29</b>	<b>3.27</b>
<b>Au/ZnO-CTAB</b>	<b>529.92</b>	<b>53.71</b>	<b>531.35</b>	<b>42.42</b>	<b>533.08</b>	<b>3.87</b>

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