

Supplementary information

Chemical deactivation of titanosilicates catalysts caused by propylene oxide in the HPPO process

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Supplementary information caption

Fig. S1 XRD patterns of Ti-MWW samples. (1): Ti-MWW, (2): Ti-MWW-PO, (3): Ti-MWW-PG, (4): Ti-MWW-MME, (5): Ti-MWW-C₃⁼.

Fig. S2 SEM images of Ti-MWW samples.

Fig. S3 TEM images of Ti-MWW samples.

Fig. S4 UV-Vis spectra of Ti-MWW samples.

Fig. S5 Epoxidation of different olefins over Ti-MWW and Ti-MWW-PO.

Fig. S6 TGA curves of Ti-MWW-PO samples with different PO contents. Ti-MWW-PO-I-IV were treated by PO of different masses with 0.02 g, 0.04 g, 0.08 g, 0.10 g.

Fig. S7 Conversion (A) and concentration (B) of 1-hexene with under different Ti-MWW samples.

Fig. S8 The 1-hexene adsorption experiment of Ti-MWW-IV.

Fig. S9 Plots of apparent activation energy in the 1-hexene epoxidation catalyzed by Ti-MWW and Ti-MWW-PO.

Fig. S10 XPS spectra of Ti-MWW.

Fig. S11 ¹³C solid-state MAS NMR spectrum of Ti-MWW-PO.

Fig. S12 In-situ IR spectra of TS-1 catalyst recorded in He at 323 K (3 min, 10 min, and 25 min) after a period of 15 min of PO adsorption, (A): 4000–2500 cm⁻¹, (B): 1800–1500 cm⁻¹.

Fig. S13 UV-vis spectra of TS-1 (a), TS-1^s (b) and the corresponding deconvolution band at 203 nm and 235 nm. IR spectra in the hydroxyl stretching evacuated at 723 K of TS-1 samples in the range 3000–4000 cm⁻¹ (c) ((1): TS-1, (2): TS-1^s).

Fig. S14 The lifetime in the epoxidation of propene over formed-Ti-MWW and formed-Ti-MWW-KCl.

Table S1 Catalytic performance of Ti-MWW_{Bt}-K samples for the epoxidation of 1-hexene.

Table S2 Lost weight of Ti-MWW-PO with different amounts of PO treatment.

Table S3 The initial reaction rate of Ti-MWW samples.

Table S4 Catalytic performance for the epoxidation of 1-hexene with H₂O₂.

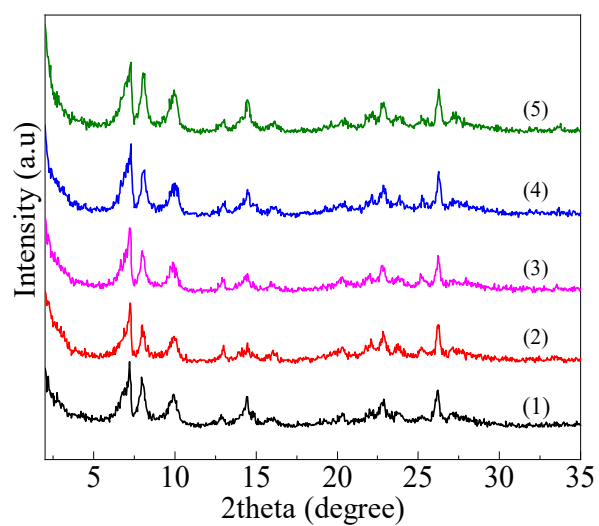


Fig. S1 XRD patterns of Ti-MWW samples. (1): Ti-MWW, (2): Ti-MWW-PO, (3): Ti-MWW-PG, (4): Ti-MWW-MME, (5): Ti-MWW-C₃=

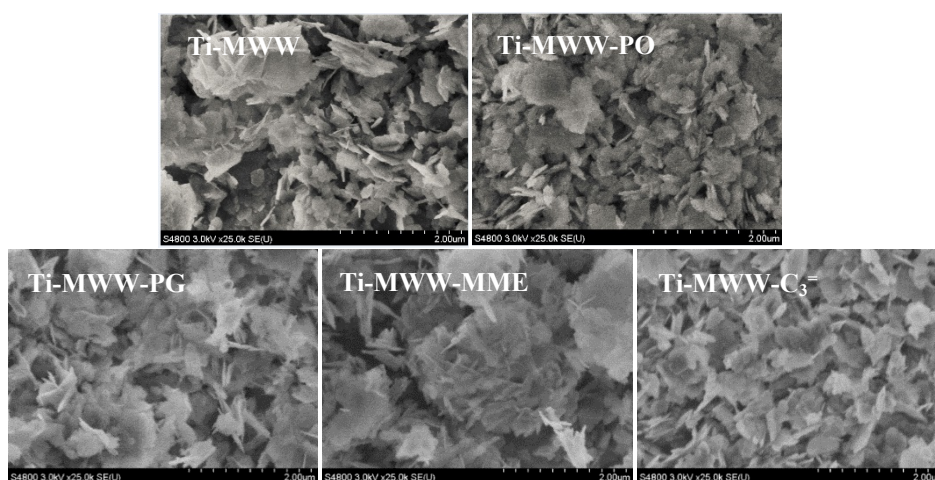


Fig. S2 SEM images of Ti-MWW samples.

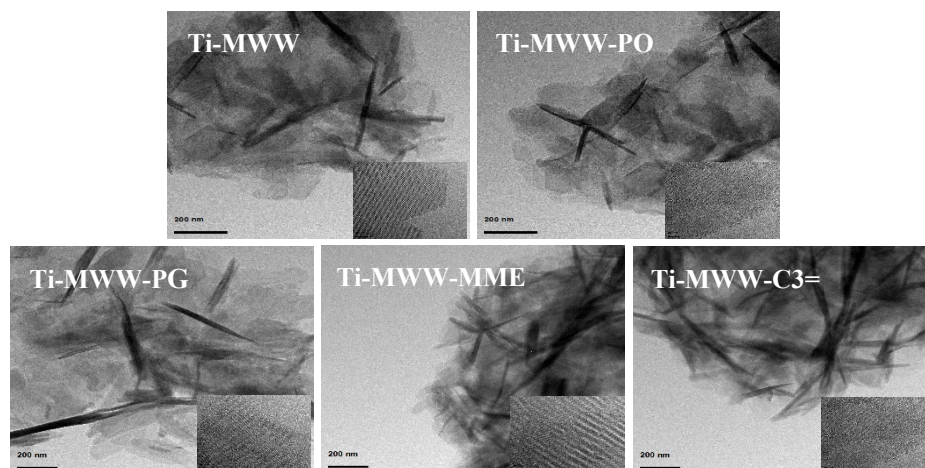


Fig. S3 TEM images of Ti-MWW samples.

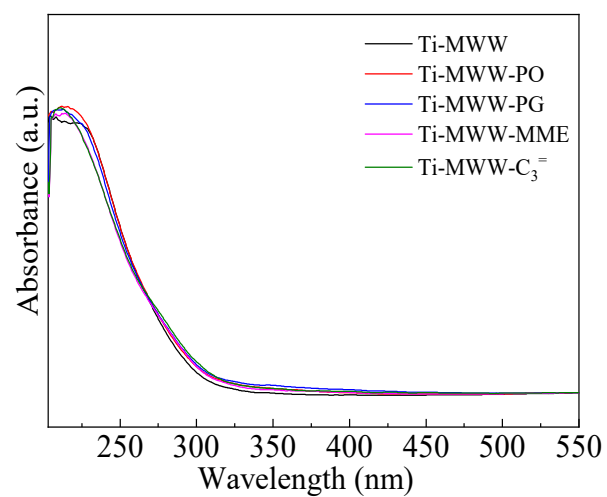


Fig. S4 UV-Vis spectra of Ti-MWW samples.

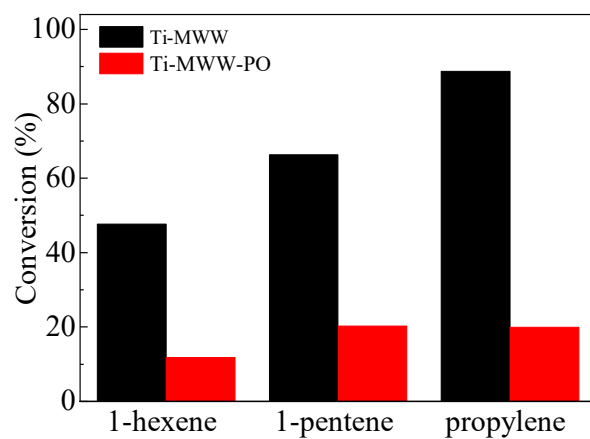


Fig. S5 Epoxidation of different olefins over Ti-MWW and Ti-MWW-PO.

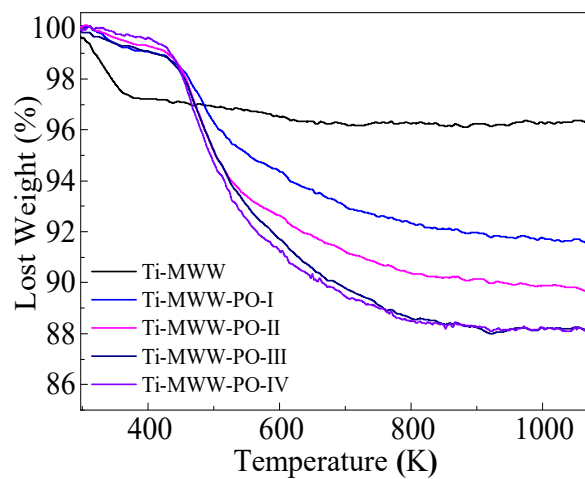


Fig. S6 TGA curves of Ti-MWW-PO samples with different PO contents. Ti-MWW-PO-I-IV were treated by PO of different masses with 0.02 g, 0.04 g, 0.08 g, 0.10 g.

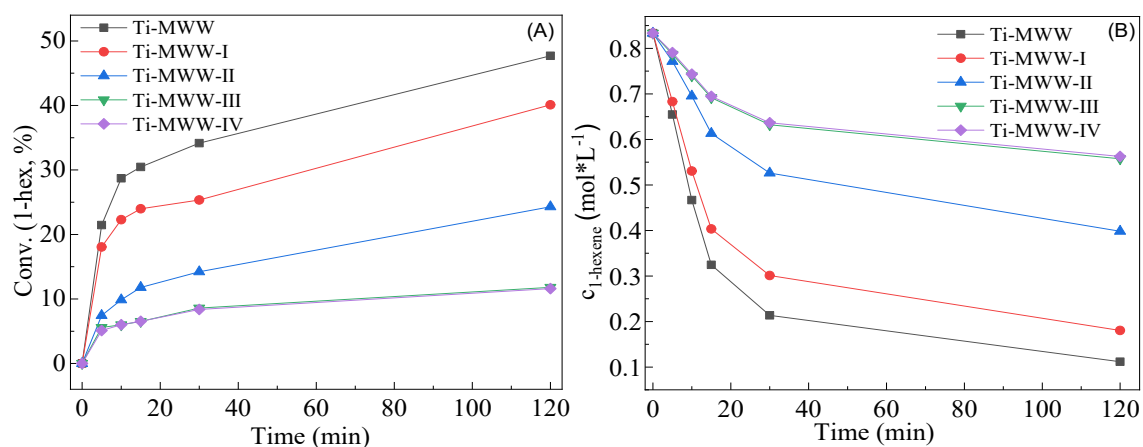


Fig. S7 Conversion (A) and concentration (B) of 1-hexene with under different Ti-MWW samples. Ti-MWW were post-treated by the different amount of PO (I: 0.02 g, II: 0.04 g, III: 0.08 g, IV: 0.10 g.). Reaction conditions: catalyst 0.05 g, 1-hexene 10 mmol, H_2O_2 10 mmol, CH_3CN 10 mL, time 2 h, reaction temperature 60 °C.

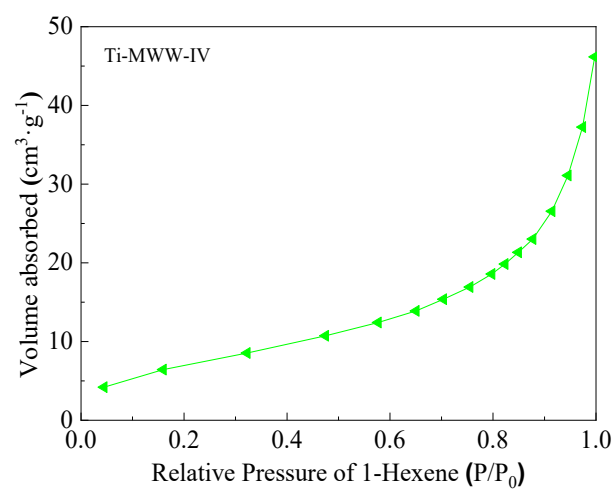


Fig. S8 The 1-hexene adsorption experiment of Ti-MWW-IV.

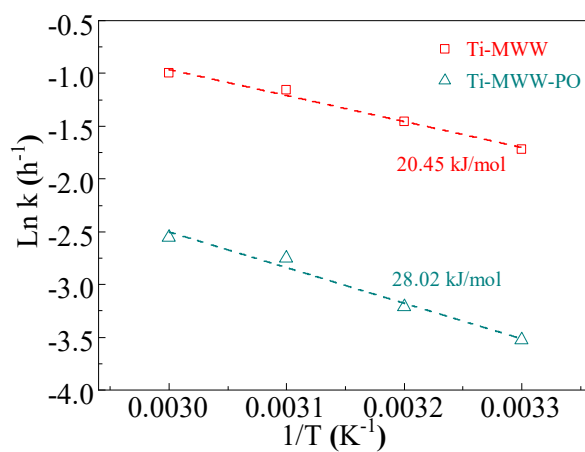


Fig. S9 Plots of apparent activation energy in the 1-hexene epoxidation catalyzed by Ti-MWW and Ti-MWW-PO.

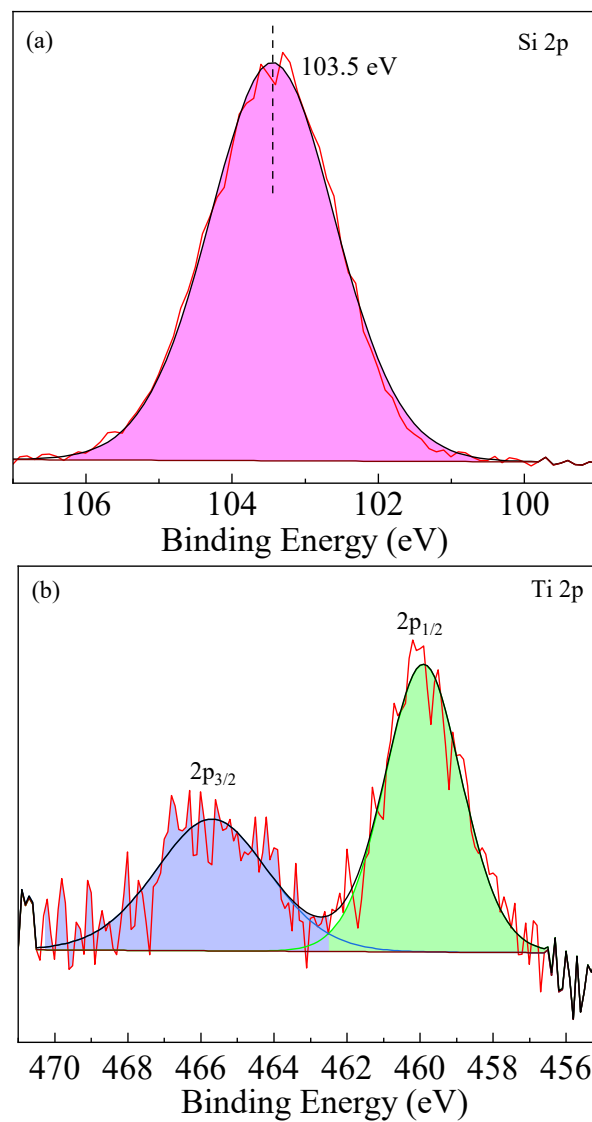


Fig. S10 XPS spectra of Ti-MWW.

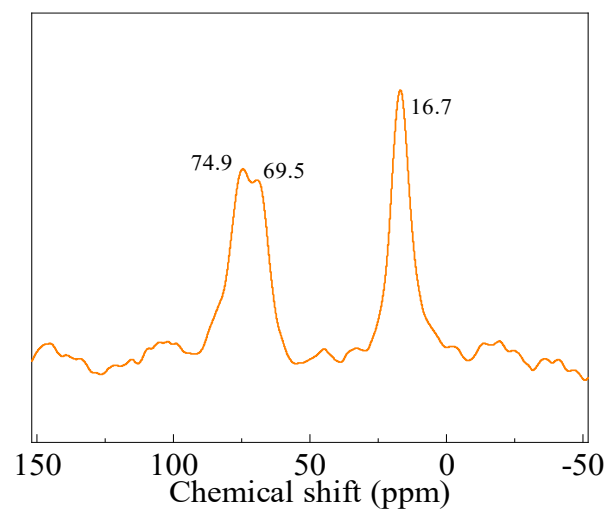


Fig. S11 ^{13}C solid-state MAS NMR spectrum of Ti-MWW-PO.

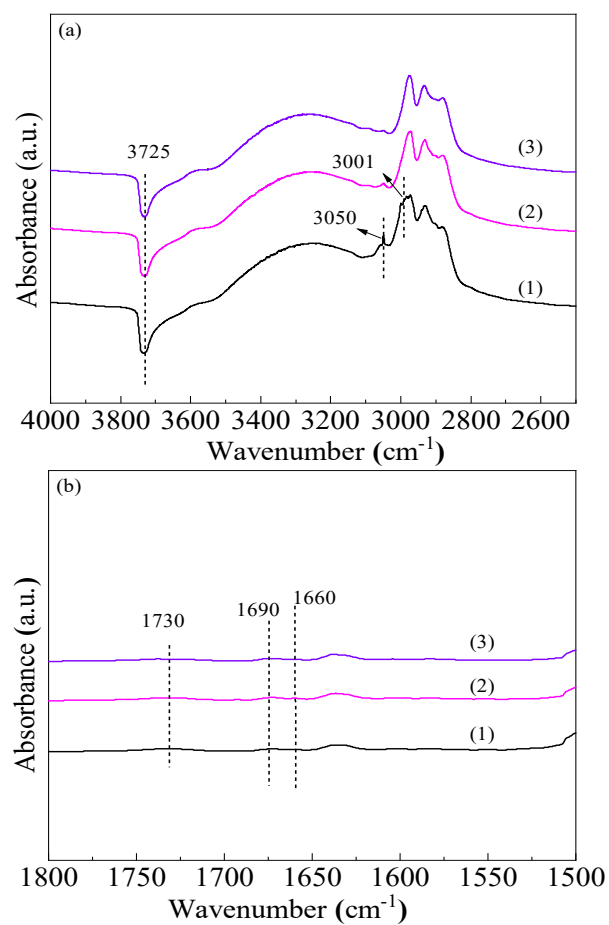


Fig. S12 In-situ IR spectra of TS-1 catalyst recorded in He at 323 K (3 min, 10 min, and 25 min) after a period of 15 min of PO adsorption, (A): 4000–2500 cm^{-1} , (B): 1800–1500 cm^{-1} .

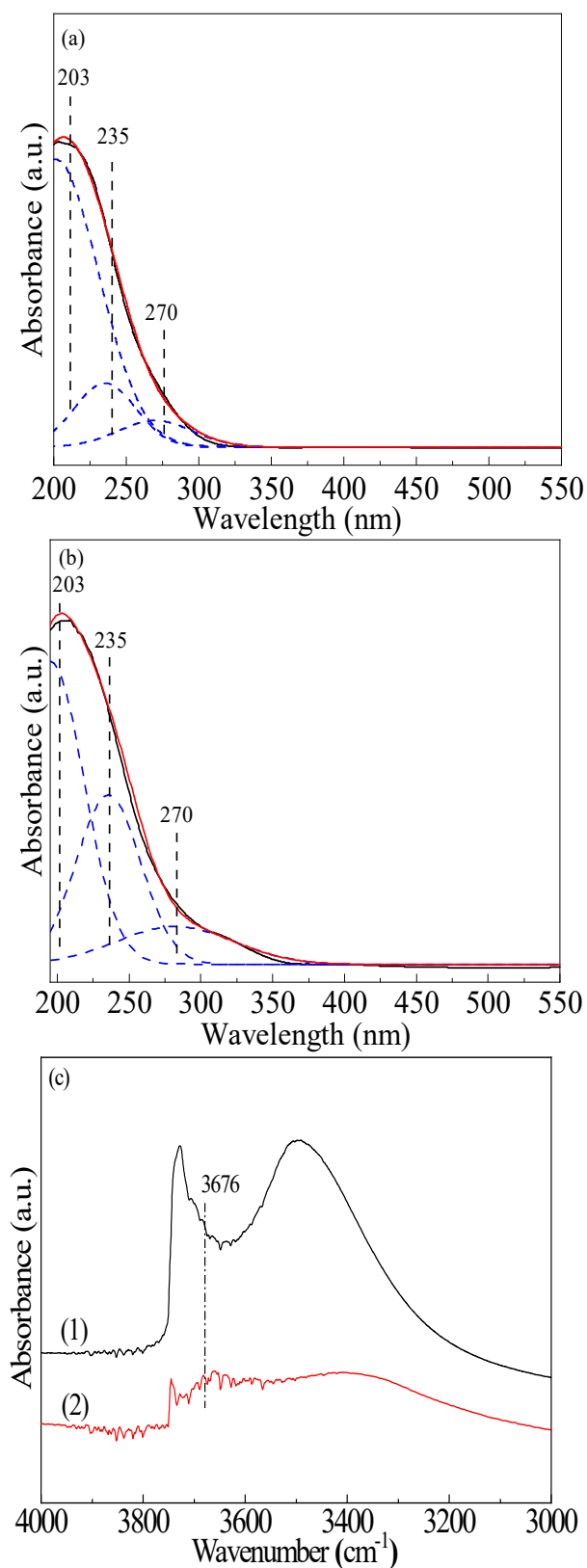


Fig. S13 UV-vis spectra of TS-1 (a), TS-1^s (b) and the corresponding deconvolution band at 203 nm and 235 nm. IR spectra in the hydroxyl stretching evacuated at 723 K of TS-1 samples in the range 3000–4000 cm⁻¹ (c)((1): TS-1, (2): TS-1^s).

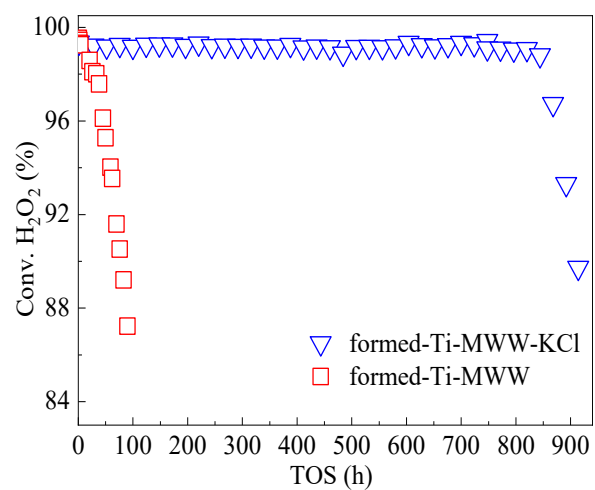


Fig. S14 The lifetime in the epoxidation of propene over formed-Ti-MWW and formed-Ti-MWW-KCl.

Table S1 Catalytic performance of Ti-MWW_{Bf}-K samples for the epoxidation of 1-hexene.

Samples	^a TON	^b Conv.(1-hex.)/%	^b Conv.(H ₂ O ₂)/%	^b Sel.(Epo.)/%	^b Eff.(H ₂ O ₂)/%	^c K/Ti
Ti-MWW _{Bf} -K(A)	281	75.5	85.5	100	88.3	0.39
Ti-MWW _{Bf} -K(B)	46	12.3	20.4	100	60.4	1.91

^a Turnover number (TON) of converted 1-hexene per Ti site in mol per mol Ti. ^b Reaction conditions: catalyst 0.05 g, 1-hexene 10 mmol, H₂O₂ 10 mmol, CH₃CN 10 mL, time 2 h, reaction temperature 60 °C. ^c Detected by ICP.

Table S2 Lost weight of Ti-MWW-PO with different amounts of PO treatment.

^a Sample	^b 298~373 K/%	^b 373~923 K/%	^b Total weight loss/%
Ti-MWW	2.42	1.07	3.49
Ti-MWW-PO-I	0.77	7.11	7.88
Ti-MWW-PO-II	0.54	9.49	10.0
Ti-MWW-PO-III	0.45	11.35	11.8
Ti-MWW-PO-IV	0.43	11.47	11.9

^a Ti-MWW-PO-I-IV were treated by PO of different masses with 0.02 g (I), 0.04 g (II), 0.08 g (III), and 0.10 g (IV). ^b Calculated by TG curves (Fig. S5).

Table S3 The initial reaction rate of Ti-MWW samples.

Sample	^a $r_0(\text{epo})$ mol/(L*g*min)
Ti-MWW	0.686
Ti-MWW-I	0.576
Ti-MWW-II	0.294
Ti-MWW-III	0.190
Ti-MWW-IV	0.186

^a In addition, the initial reaction rates ($r_0(\text{epo})$) were calculated from tangential slopes at reaction time $(t) = 0$ using exponential curve fitting.

Table S4 Catalytic performance for the epoxidation of 1-hexene with H₂O₂.

Samples	^a Conv.(1-hex.)/%	^a Conv.(H ₂ O ₂)/%	^a Sel.(Epo)/%	^a Eff.(H ₂ O ₂)/%	^b TON
TS-1-80	13.1	16.1	95.3	75.6	127
TS-1-80-PO	10.6	17.1	97.2	62.0	103
^c TS-1 ^s	23.0	29.6	95.0	74.5	224
^c TS-1 ^s -PO	16.6	29.5	98.2	55.1	162

^a Reaction conditions: catalyst 0.05 g, 1-hexene 10 mmol, H₂O₂ 10 mmol, CH₃CN 10 mL, time 2 h, reaction temperature 60 °C. ^b Turnover number (TON) of converted 1-hexene per Ti site in mol per mol Ti. ^c TS-1^s was treated with ethylamine (EA) at 443 K.