

Supporting information

Copper mesh supported nickel nanowire array as catalyst for hydrogen evolution reaction in high current density water electrolysis

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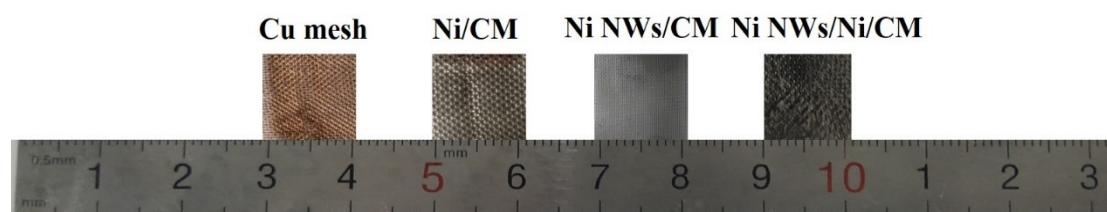


Figure S1. Picture of Cu mesh, Ni/CM, Ni NWs/CM, Ni NWs/Ni/CM.



Figure S2. Bionic model picture of Ni NWs/Ni/CM.

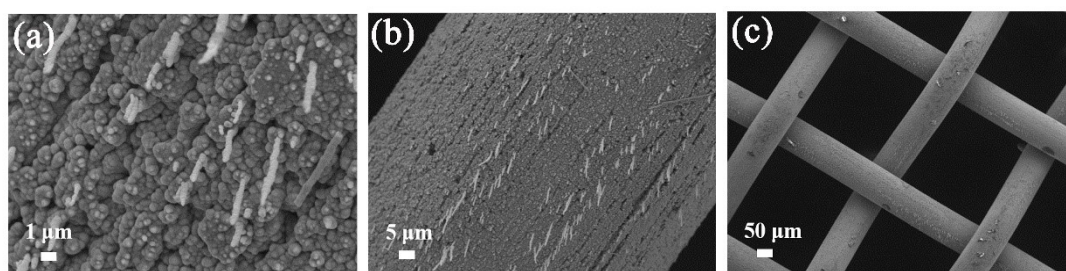


Figure S3. SEM images of Ni NWs/CM with different magnifications, respectively.

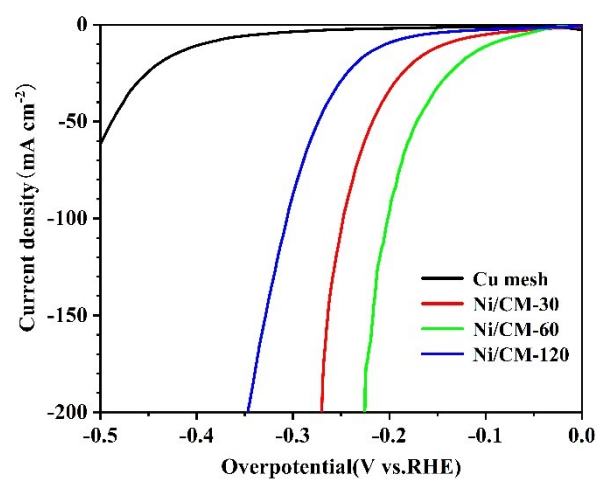


Figure S4. HER LSVs of Ni/CM-30、Ni/CM-60 and Ni/CM-120 with different deposition times, respectively.

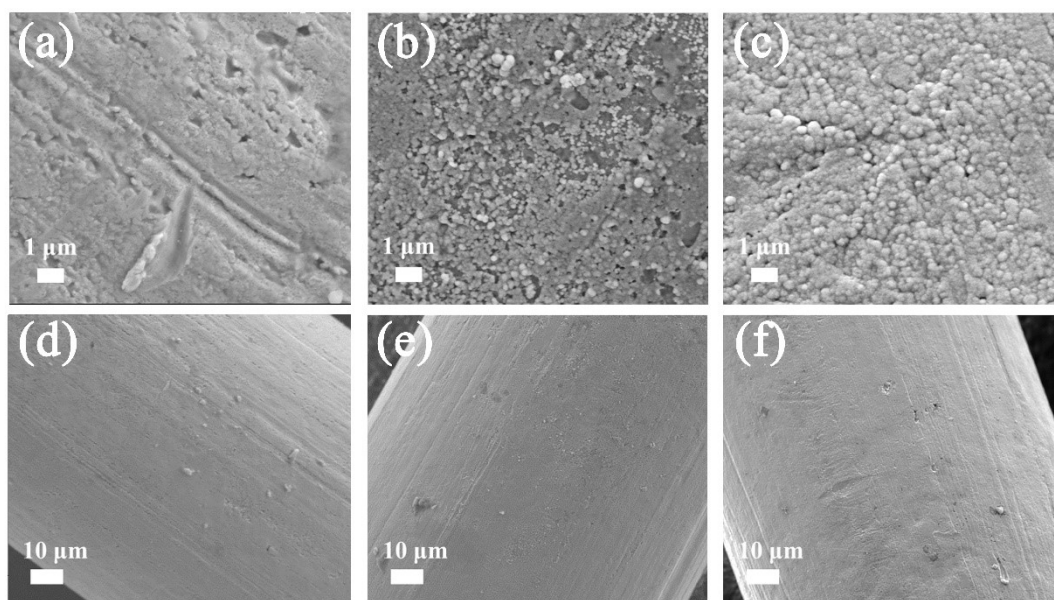


Figure S5. SEM images of Ni/CM-30、Ni/CM-60 and Ni/CM-120 with different deposition times, respectively.

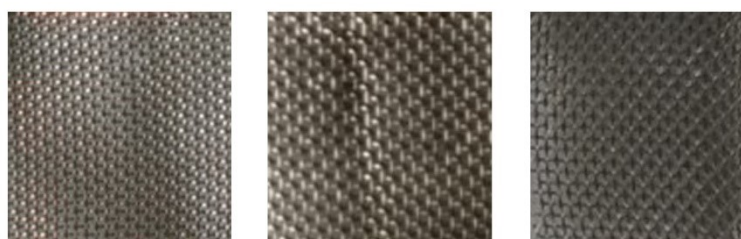


Figure S6. Picture of Ni/CM-30、Ni/CM-60 and Ni/CM-120 with different deposition times, respectively.

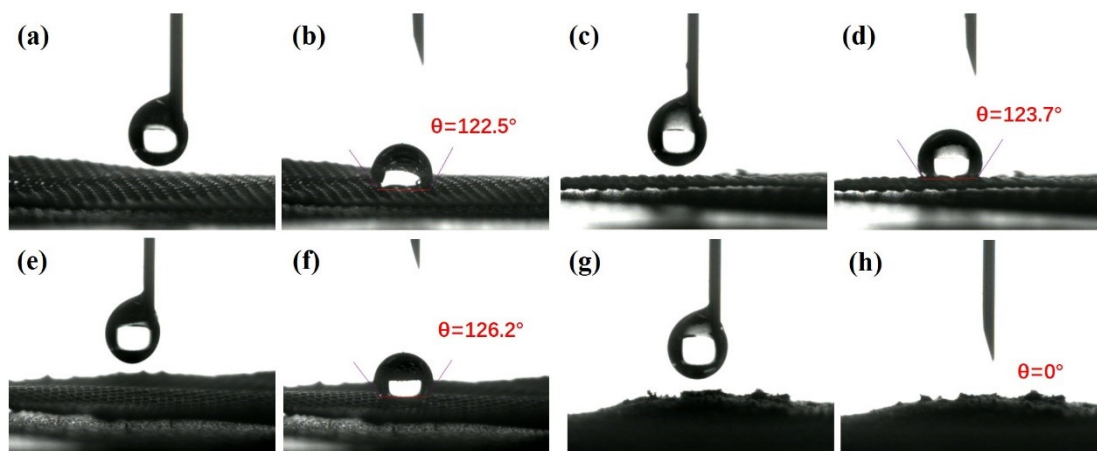


Figure S7. Contact angle measurements of the Cu Mesh (a)before (b)after; Ni/CM (c)before (d)after; Ni NWs/CM (e)before (f)after; Ni NWs/Ni/CM (g)before (h)after.

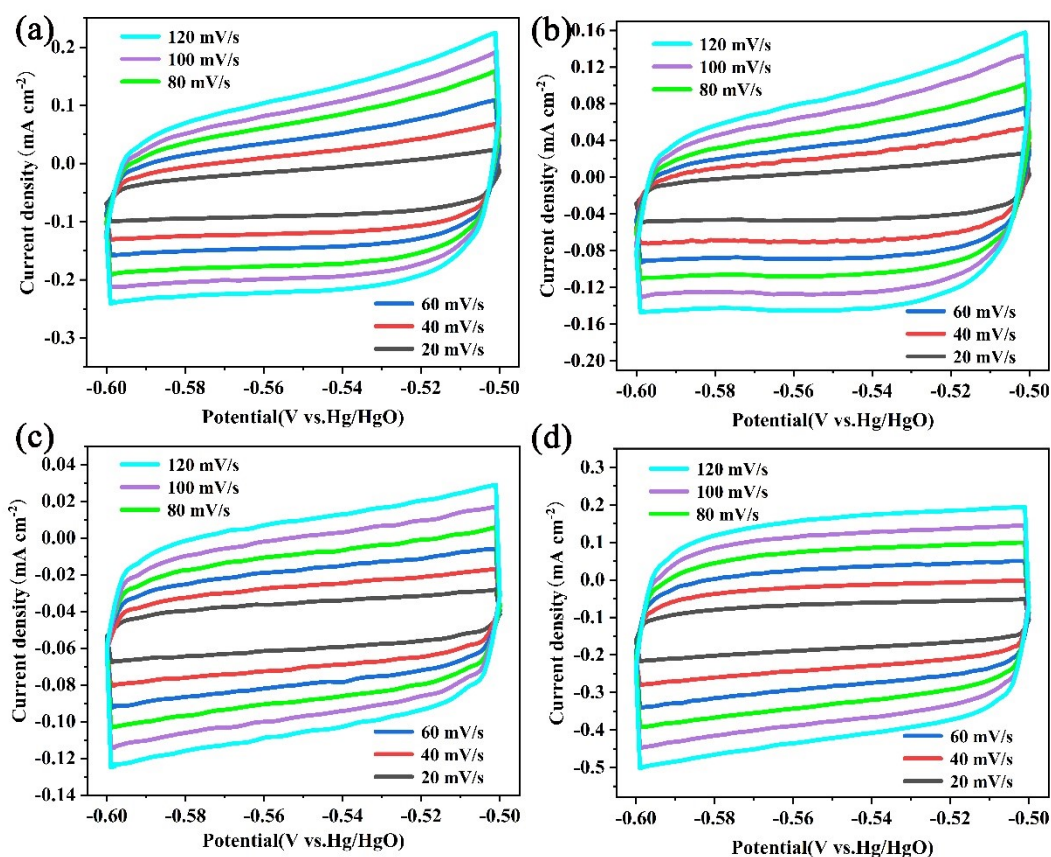


Figure S8. Cyclic Voltammetry plots of (a) Ni NWs/Ni/CM; (b) Ni/CM; (c) Ni NWs/CM; (d)Cu mesh.

Table S1. Fitting values of Pt/C, Ni NWs/E-NM, Ni NWs/NM, NM, E-NM electrochemical composition in Nyquist

Samples	R _s (ohm)	R _{ct} (ohm)
Ni NWs/Ni/CM	1.200	3.258
Ni/CM	1.231	8.815
Ni NWs/CM	1.147	11.900
Cu mesh	1.207	36.280

Table S2. HER activities of different catalysts were compared at the current density of 500 mA cm⁻².

Catalyst	Electrolyte	Substrate	$\eta_{500}(\text{mV})$	Reference
Ni NWs/Ni /CM	1 M KOH	Cu mesh	248	This work
F-CoP/Fe ₂ P/IF	1 M KOH	Iron foam	260	1
MoS ₂ /M ₂ C	1 M KOH	Ti foil	191	2
MoS ₂ /Ni ₃ S ₂ NW-NF	1 M KOH	Ni foam	182	3
Cu-m/Cu-W/NiCo-LDH	1 M KOH	Cu mesh	139	4
NiCo/NiCo-OH	1 M KOH	Ni foam	184	5
a-MoS _x /CC	0.5 M H ₂ SO ₄	carbon cloth	250	6
Ni ₂ P/NF	1 M KOH	Ni foam	250	7
Co ₄ N-CeO ₂ /GP	1 M KOH	Graphite plate	190	8

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