Supporting Information

Formation of NiFe-MOF nanosheets on Fe foam to achieve advanced

electrocatalytic oxygen evolution

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Figure S1. XRD images of NiFe-MOF/FF-0.25, NiFe-MOF/FF-0.5, NiFe-MOF/FF-0.75, NiFe-MOF/FF-1.0, Fe-MOF/FF and Ni-MOF.



Figure S2. SEM images of a) Ni-MOF, b) Fe-MOF/FF.



Figure S3. high-resolution XPS spectra of NiFe-MOF/FF: a) C 1s, b) O 1s.



Figure S4. SEM images of a) NiFe-MOF/FF-0.25, b) NiFe-MOF/FF-0.5, c) NiFe-MOF/FF-0.75, and d) NiFe-MOF/FF-1.0.



Figure S5. a) Polarization curves and b) Electrochemical impedance spectra (EIS) of NiFe-MOF/FF-0.25, NiFe-MOF/FF-0.5, NiFe-MOF/FF-0.75, and NiFe-MOF/FF-1.0.



Figure S6. a) The Cdl value of NiFe-MOF/FF-0.5, Ni-MOF /FF, Fe-MOF /FF and Fe Foam. Cyclic voltammetry curves of b) NiFe-MOF/FF, c) Ni-MOF /FF, d) Fe-MOF /FF and e) Fe Foam.

	Fe molar%	Ni molar%
	7.16	2.74
NiFe-MOF/FF-0.25	7.15	3.76
NiFe-MOF/FF-0.5	6.32	4.94
NiFe-MOF/FF-0.75	1.97	7.75
NiFe-MOF/FF-1.0	1.22	11.29
Fe-MOF /FF	100	0
Ni-MOF	0	100

 Table S1. The Fe/Ni atomic ratios in NiFe-MOF/FF samples by EDS.

Catalyst	Electrolyte	Mass loading (mg cm ⁻²)	Over potential (V) @ 10 mA cm ⁻²	Ref.
NiFe-MOF/FF-0.5	1 M KOH	_	0.216 V at 50 mA cm^{-2}	This work
MIL-53(FeNi)/NF	1 M KOH	2.63	$0.233 \text{ at } 50 \text{ mA cm}^{-2}$	[1]
NiCo-C/NF	1 M KOH	_	0.26	[2]
FCN-MOF/NF	1 M KOH	_	0.196	[3]
FeCo-NCNFs-800	0.1 M KOH	_	0.45	[4]
Co-Fe-BDC	1 M KOH	0.5	0.295	[5]
NiCo-BDC BMNSs	1 M KOH	_	0.23	[6]
CoNi1@C	1 M KOH	_	0.276	[7]
NiFe-MOF/NF	1 M KOH	_	$0.225 \text{ at } 50 \text{ mA cm}^{-2}$	[8]
NiFe MOF/OM- NFH	1 М КОН	0.4	0.27	[9]
IrO ₂	1 M KOH	_	0.4	[7]
RuO ₂	0.1 M KOH	_	0.316	[9]

Table S2. Comparisons of OER performance of NiFe-MOF/FF-0.5 with bimetallic MOF and precious metal catalysts.

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