

Supplementary Information

Aryl bismuth phosphinates [BiAr₂(O(O)PRR')]: structure-activity relationships for antibacterial activity and cytotoxicity

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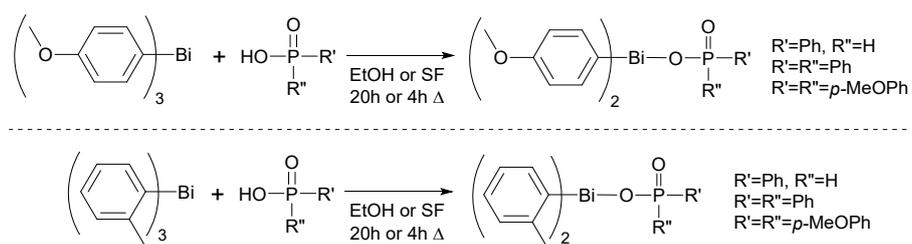
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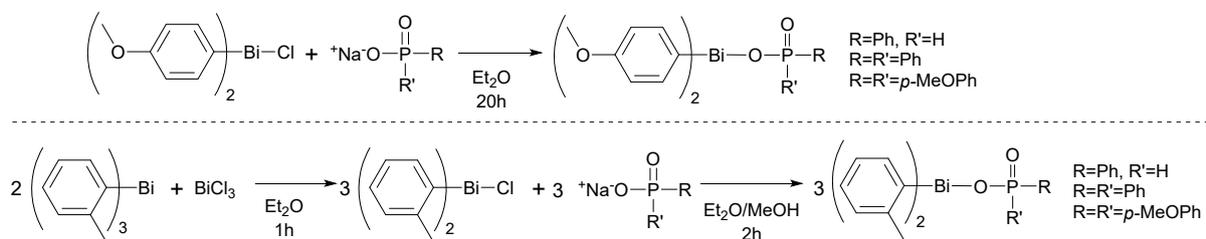
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Scheme S1 Attempted solvent mediated and solvent free routes to form diaryl bismuth *mono*-phosphinato complexes.



Scheme S2 Attempted salt metathesis route to form diaryl bismuth *mono*-phosphinato complexes.

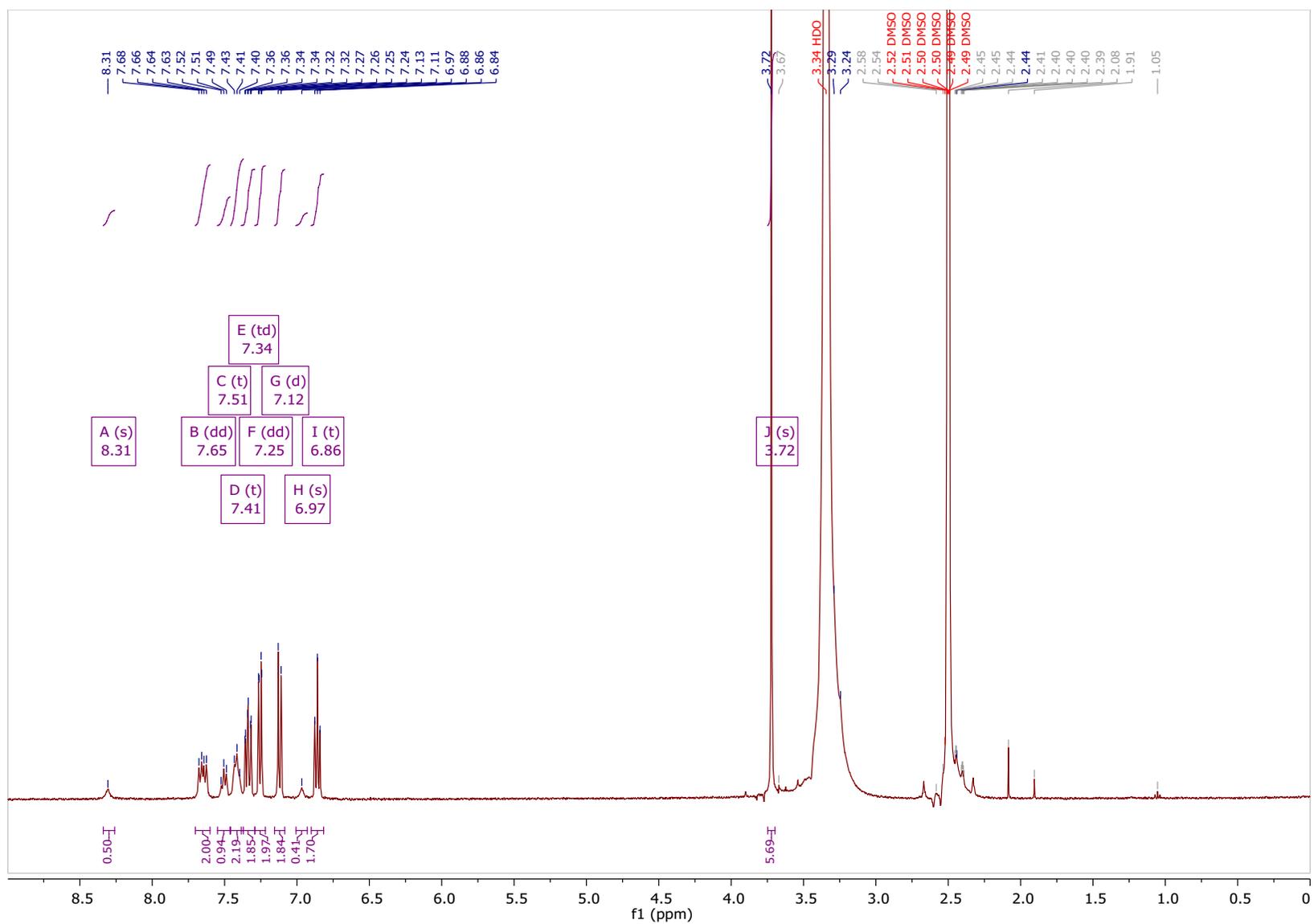


Figure S1 ¹H NMR spectrum of complex [Bi(o-MeOPh)₂(O(O)P(H)Ph)]_n **1** in d₆-DMSO.

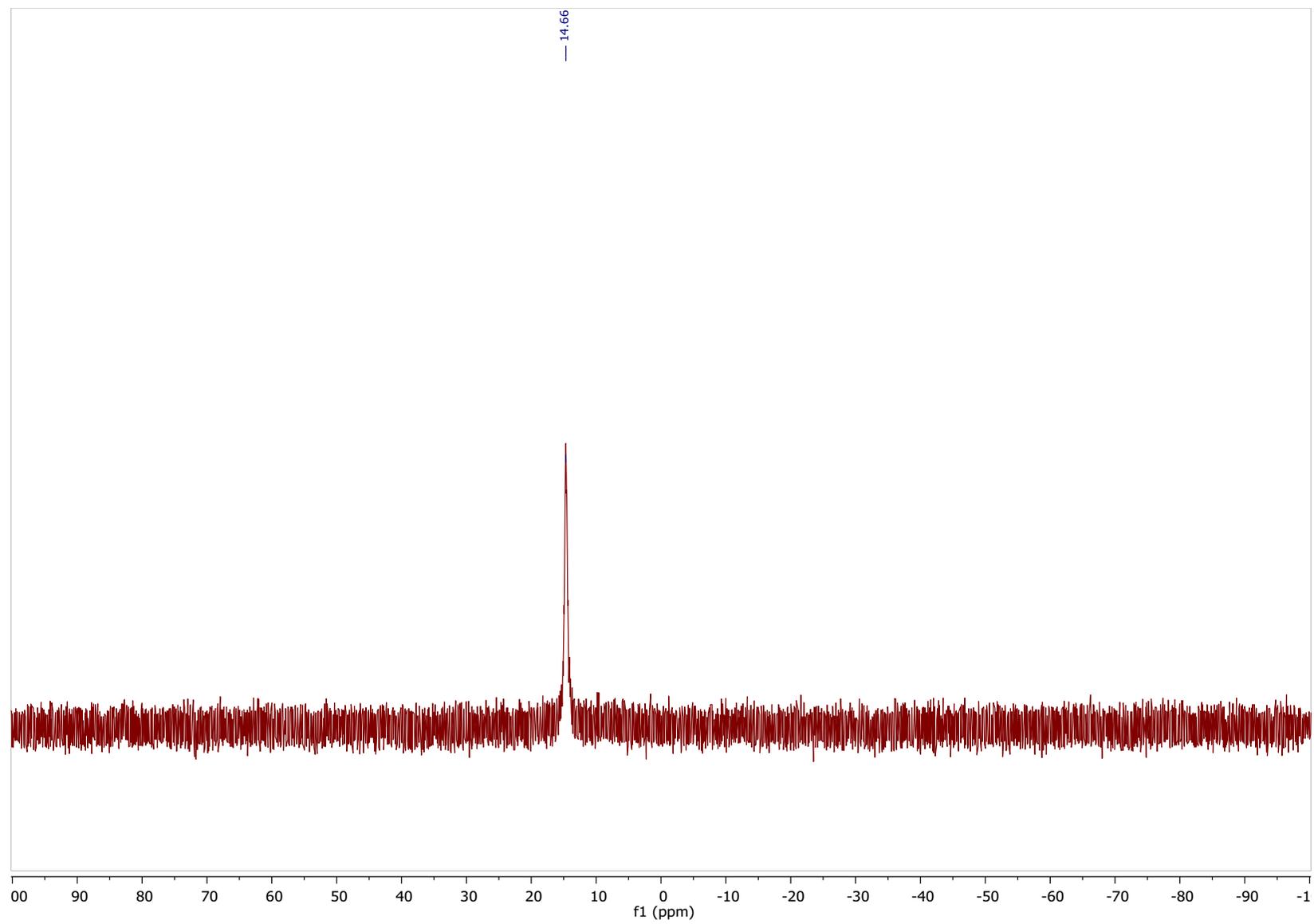


Figure S2 ^{31}P NMR spectrum of complex $[\text{Bi}(\text{o-MeOPh})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **1** in d_6 -DMSO.

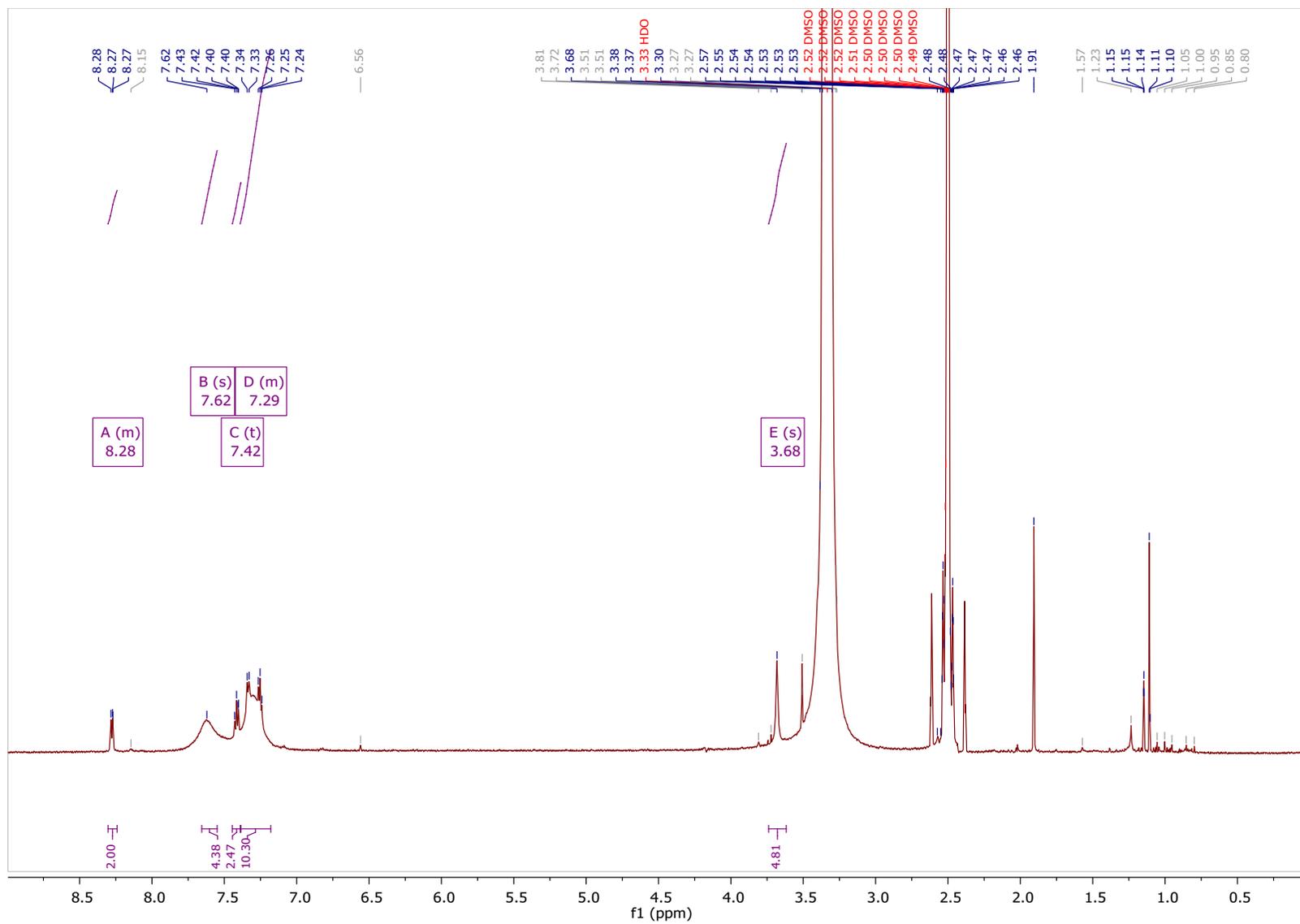


Figure S3 ¹H NMR spectrum of complex [Bi(o-MeOPh)₂(O(O)PPh₂)]_n **2** in d₆-DMSO.

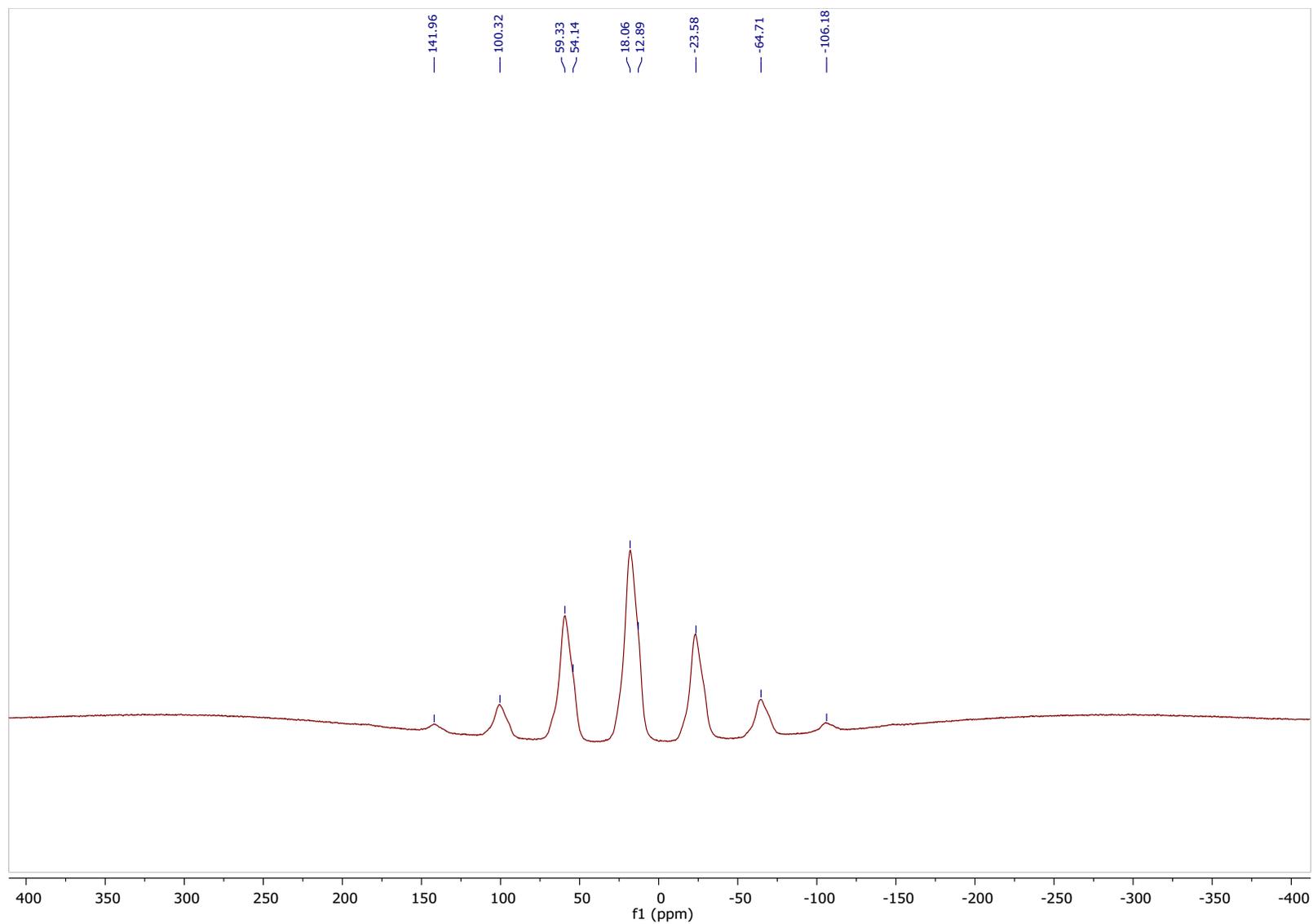


Figure S4 Solid state ^{31}P NMR spectrum of complex $[\text{Bi}(\text{o-MeOPh})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **2**.

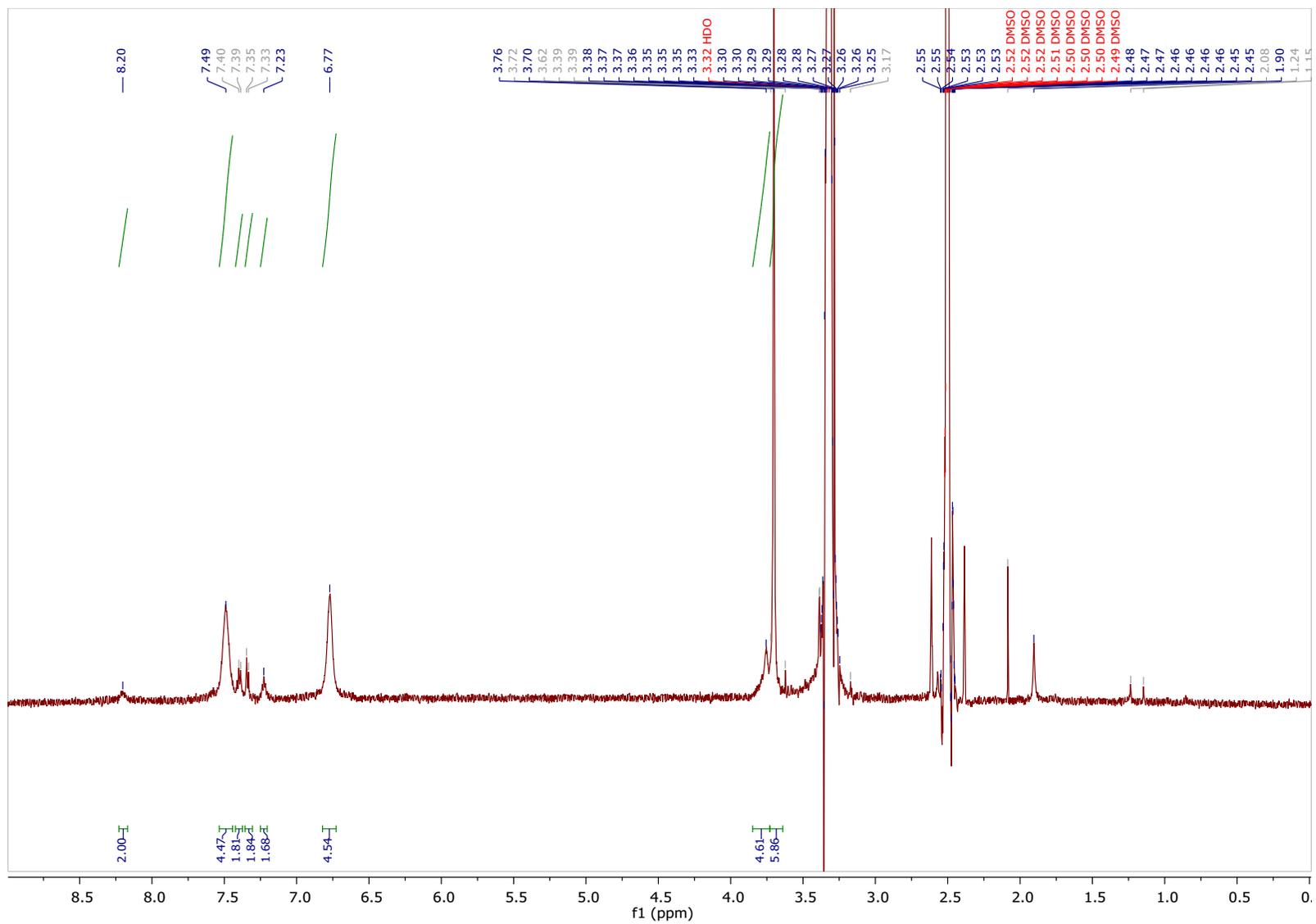


Figure S5 ^1H NMR spectrum of complex $[\text{Bi}(\text{o-MeOPh})_2(\text{O}(\text{O})\text{P}(\text{p-MeOPh})_2)]_n$ **3** in d_6 -DMSO.

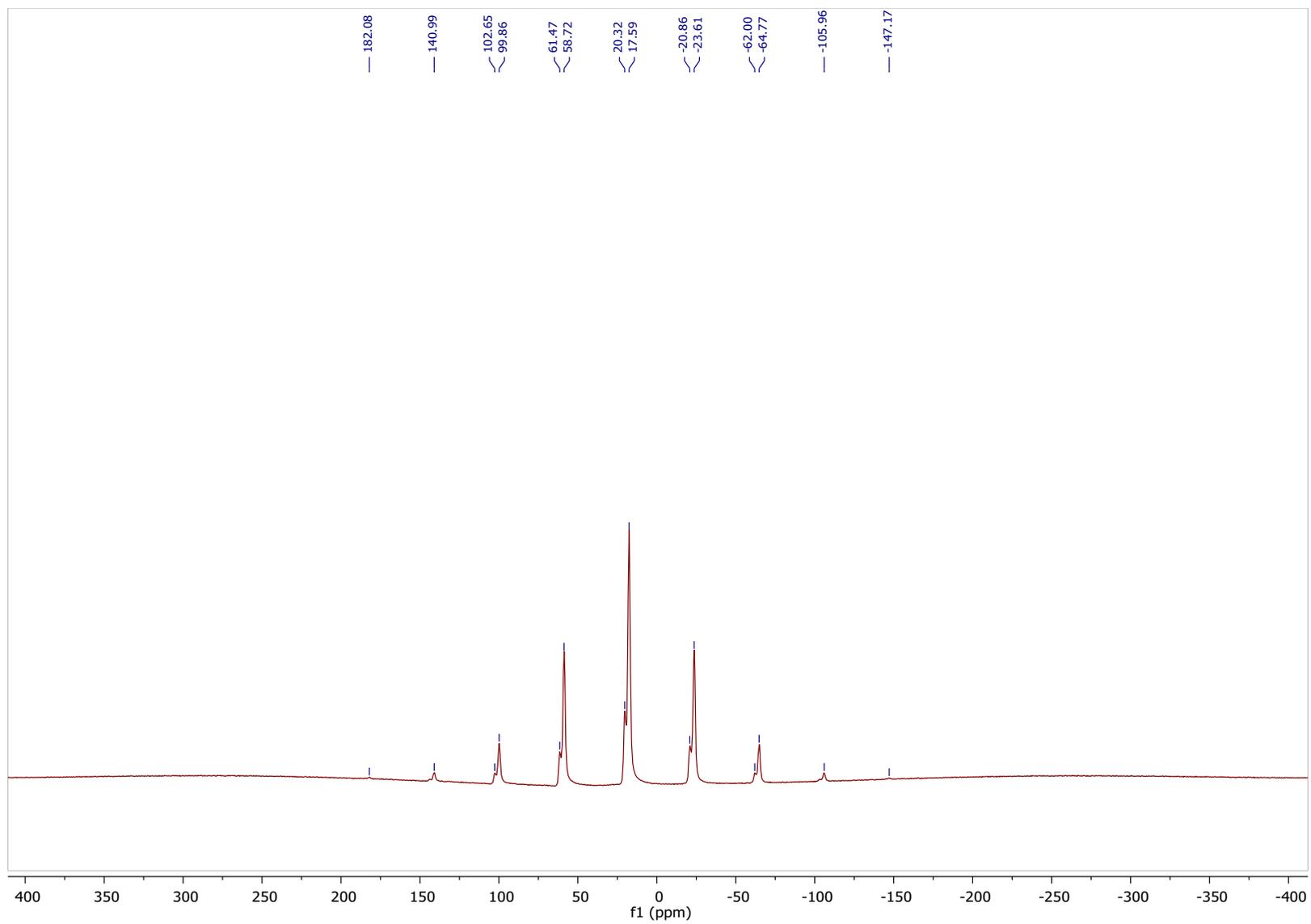


Figure S6 Solid state ^{31}P NMR spectrum of complex $[\text{Bi}(\text{o-MeOPh})_2(\text{O}(\text{O})\text{P}(\text{p-MeOPh})_2)]_n$ **3**.

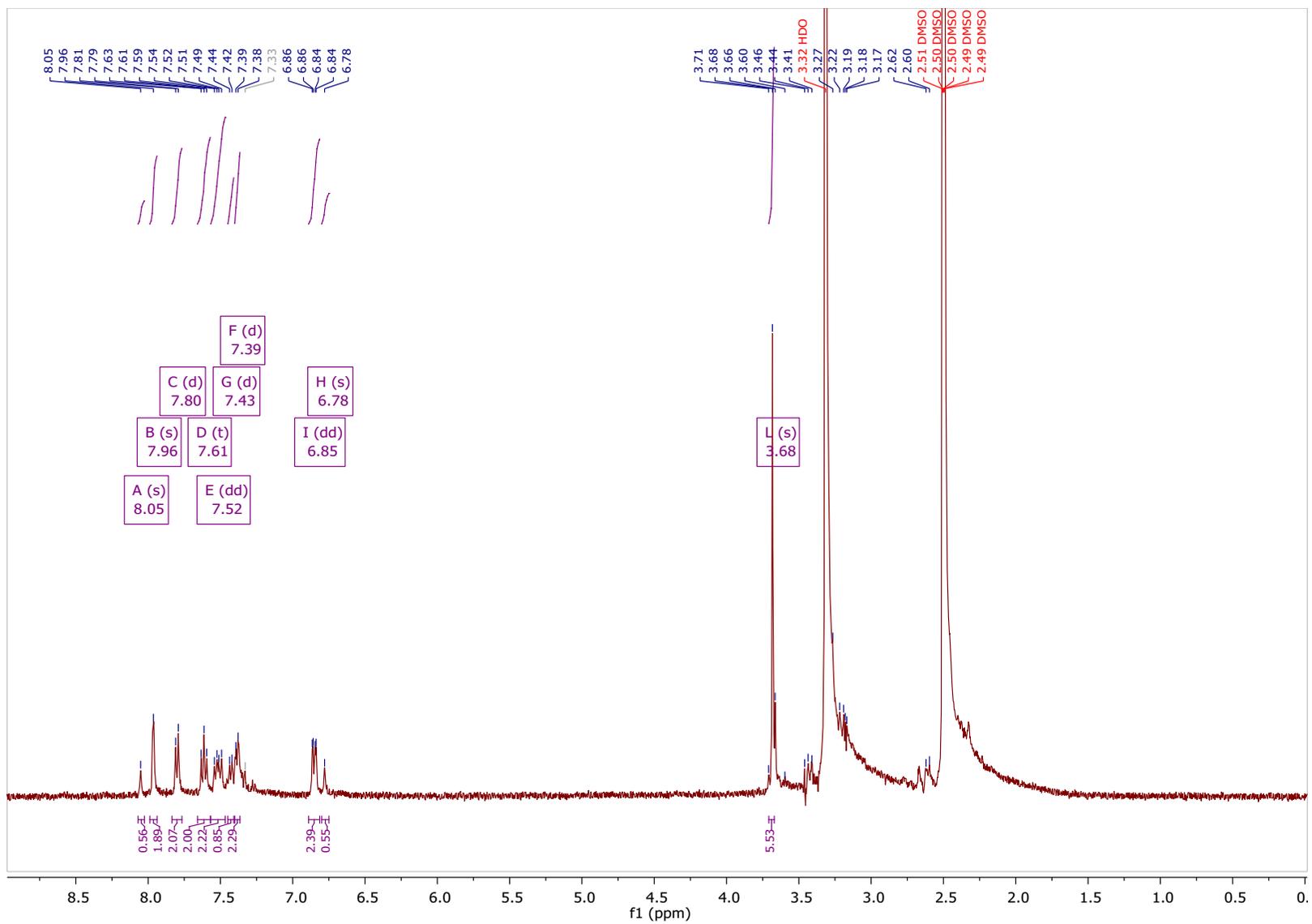


Figure S7 ¹H NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **4** in d₆-DMSO.

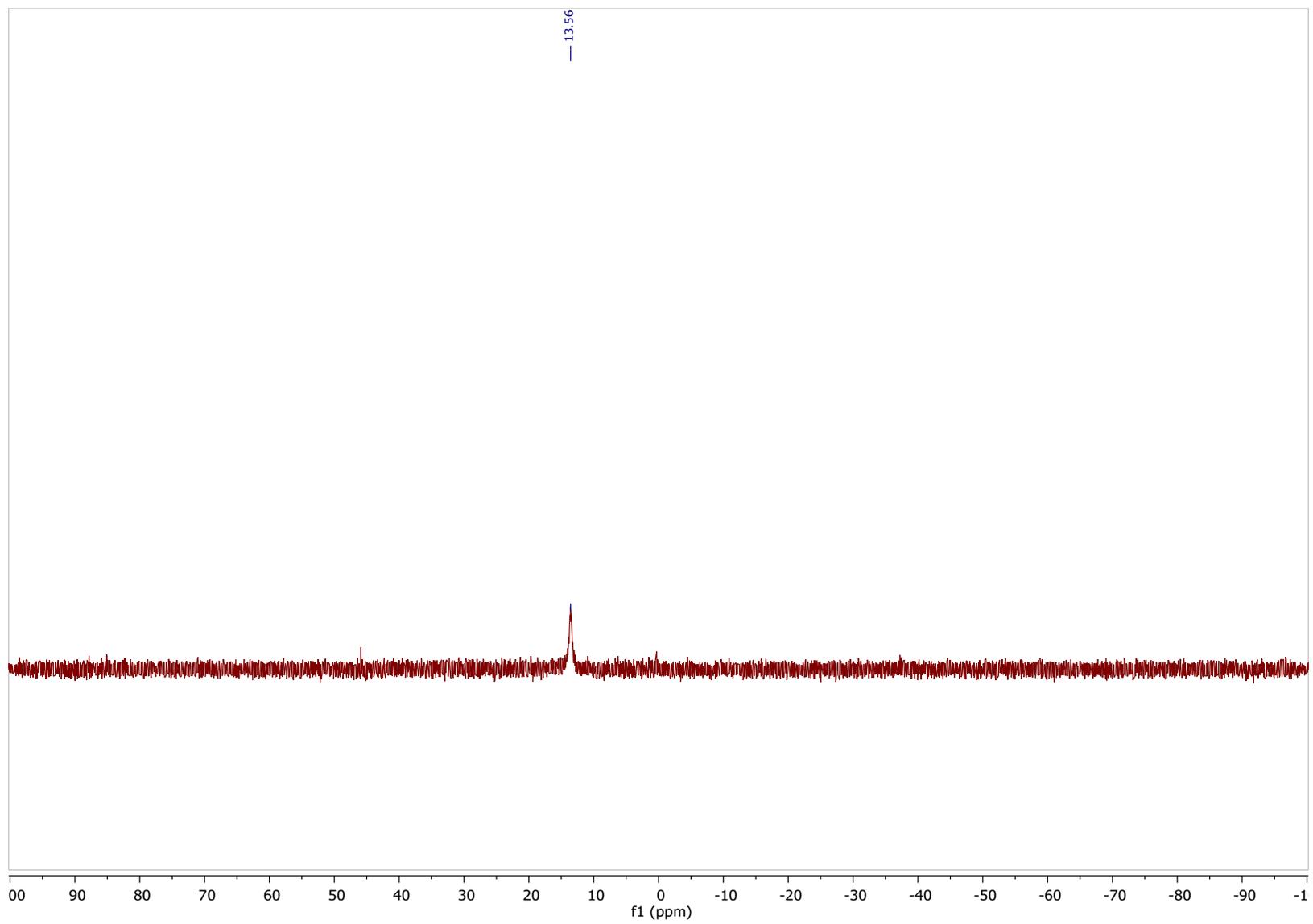


Figure S8 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **4** in d_6 -DMSO.

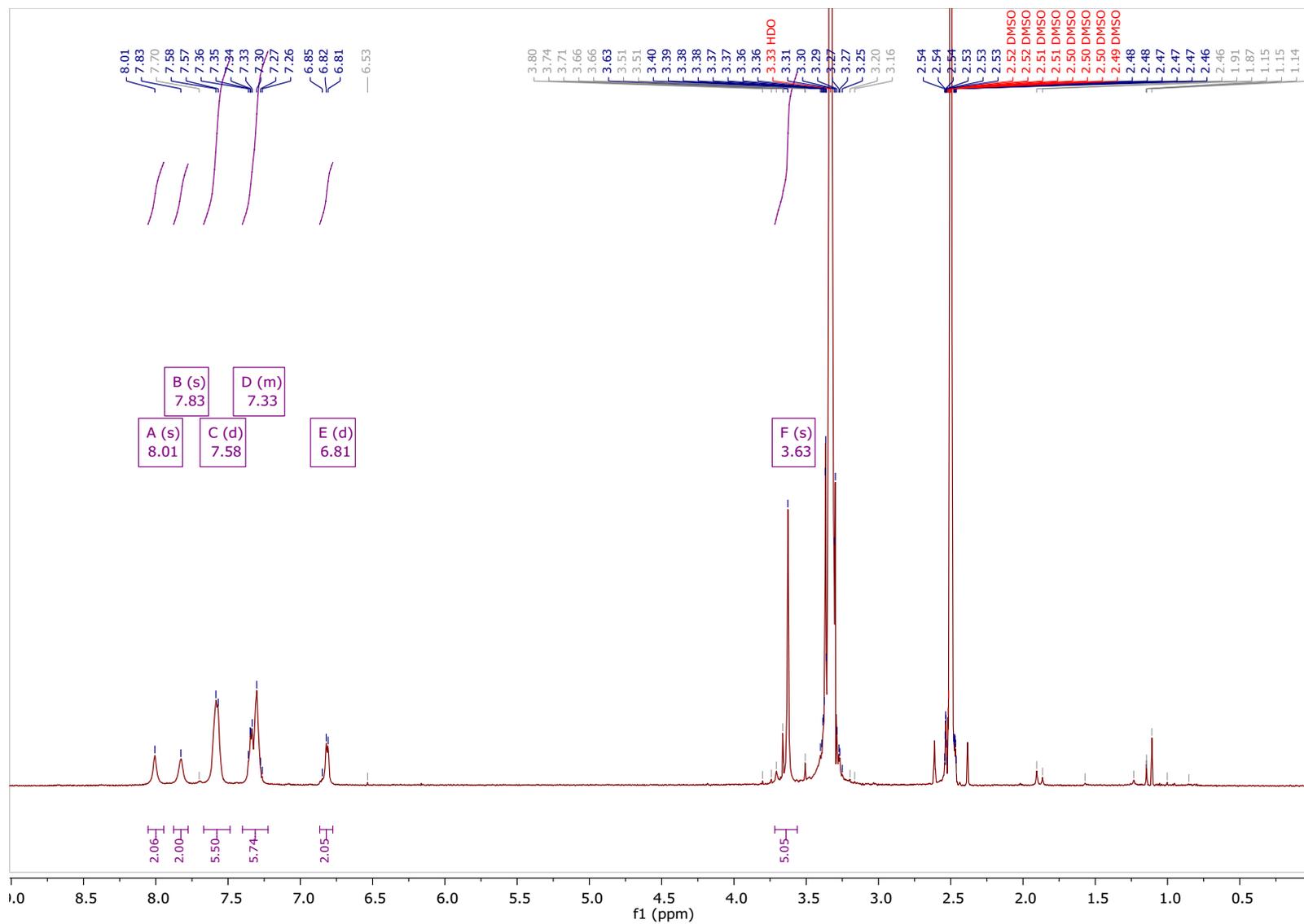


Figure S9 ¹H NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **5** in d₆-DMSO.

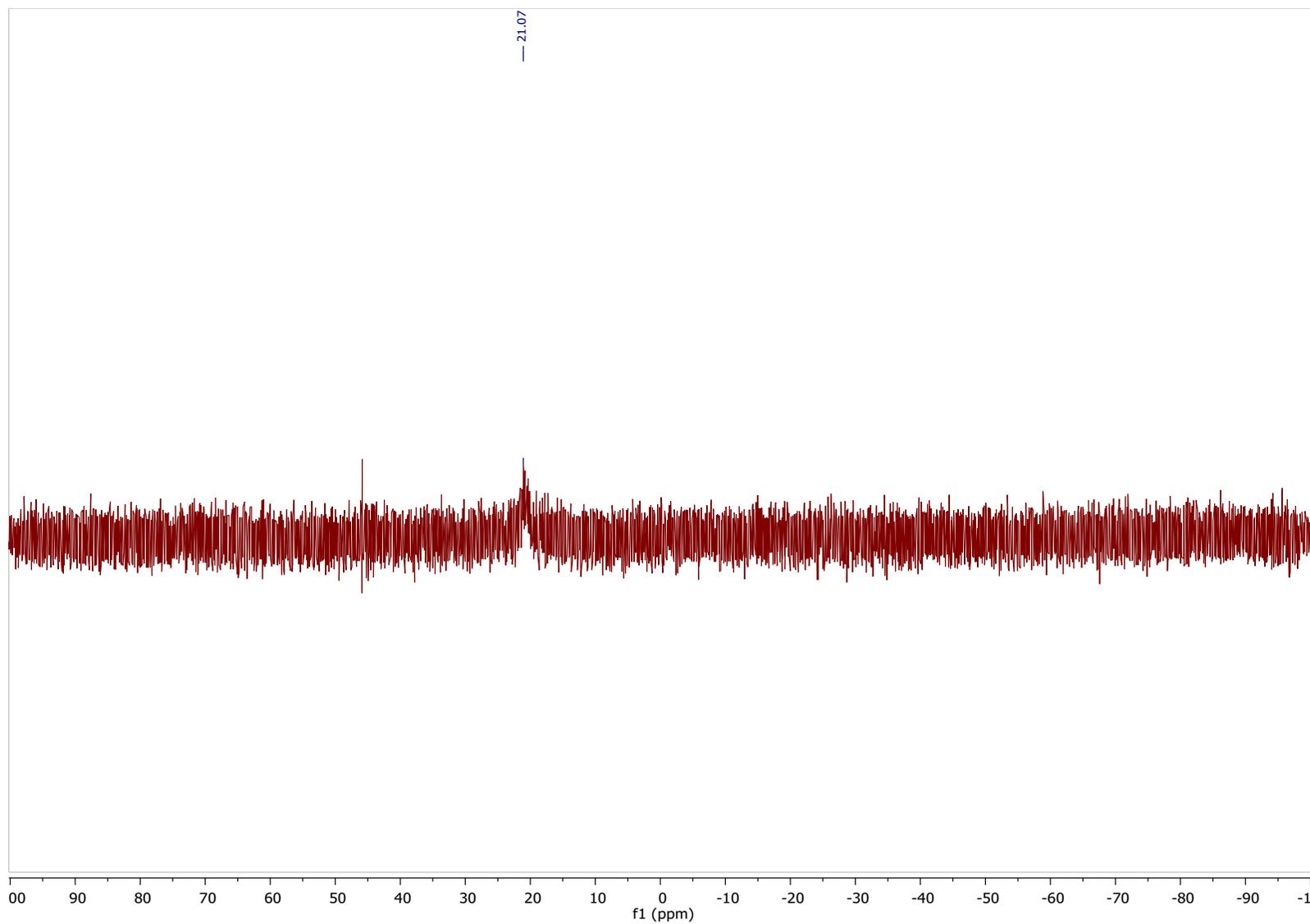


Figure S10 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **5** in $\text{d}_6\text{-DMSO}$.

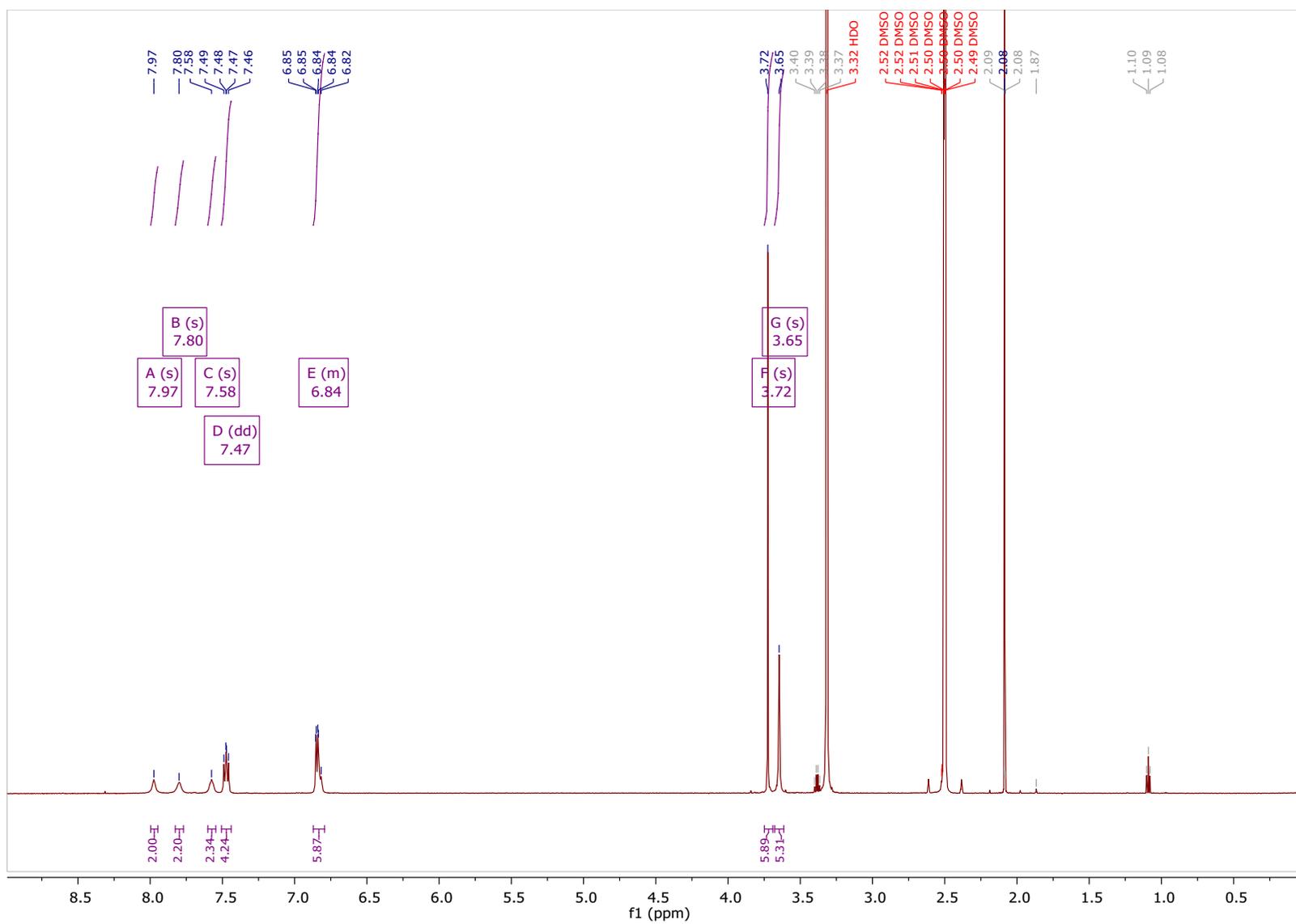


Figure S11 ^1H NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{P}(p\text{-MeOPh})_2)]_n$ **6** in d_6 -DMSO. Traces of acetone and Et_2O can be observed after washing.

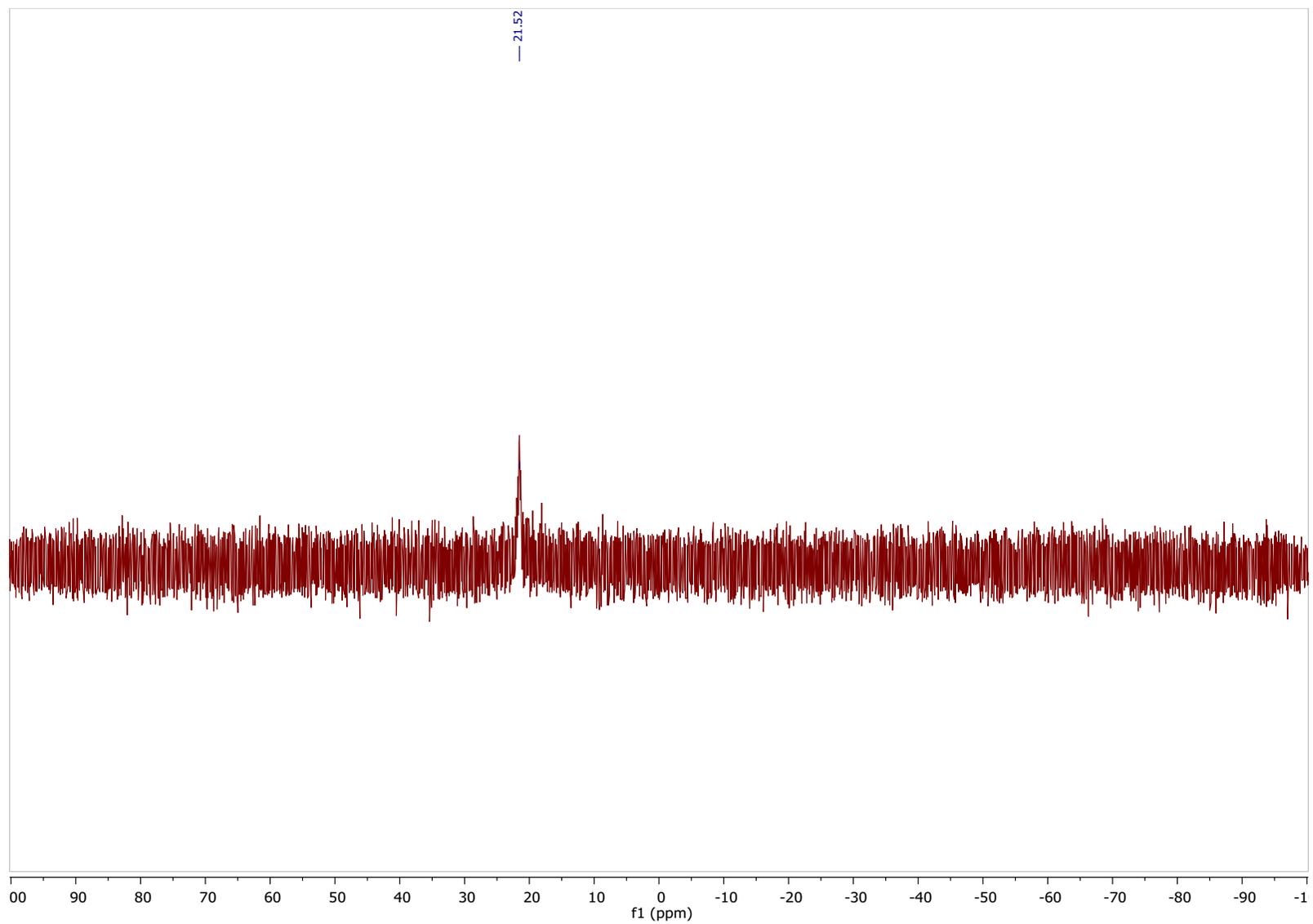


Figure S12 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-MeOPh})_2(\text{O}(\text{O})\text{P}(p\text{-MeOPh})_2)]_n$ **6** in d_6 -DMSO.

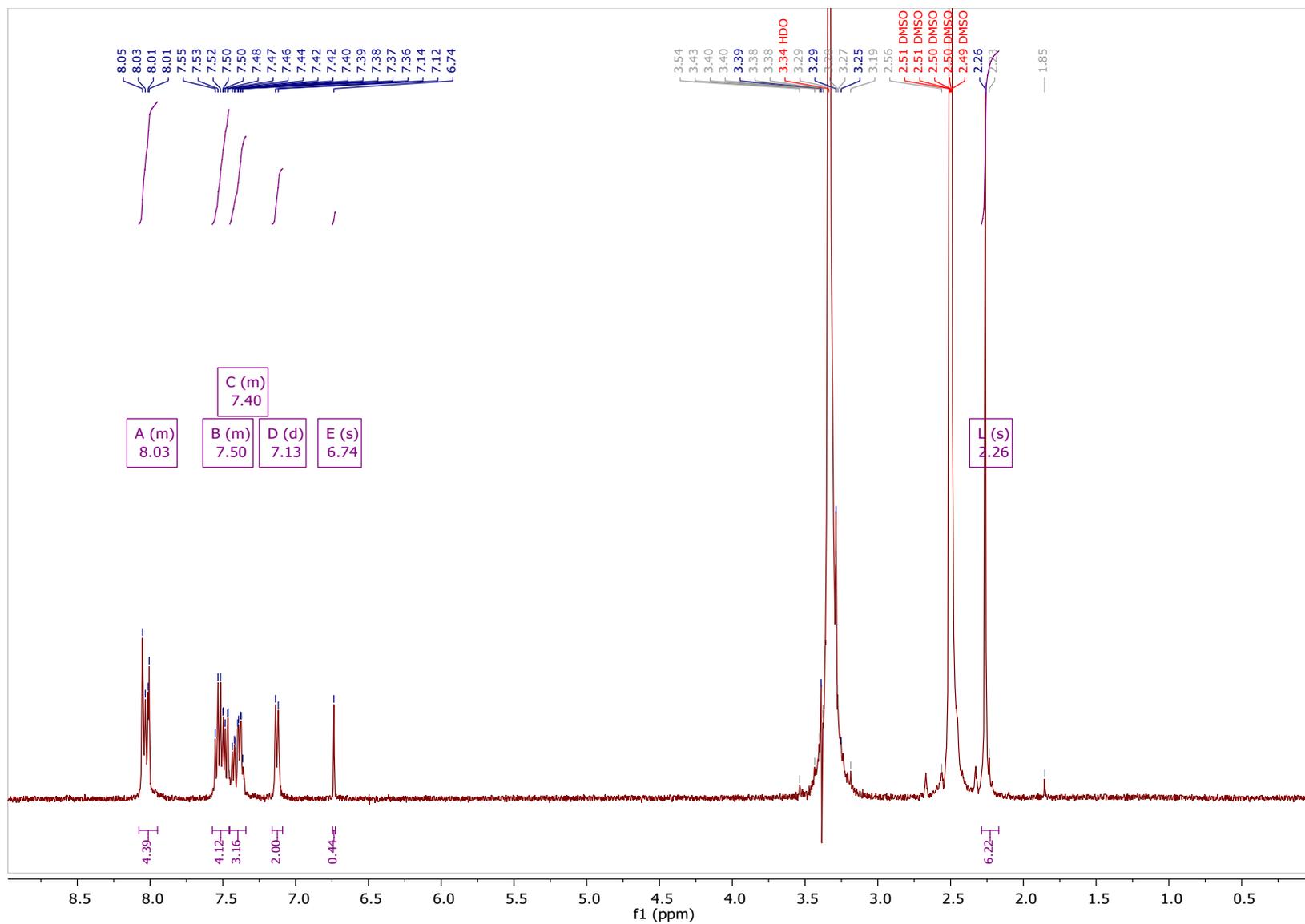


Figure S13 ^1H NMR spectrum of complex $[\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ in d_6 -DMSO.

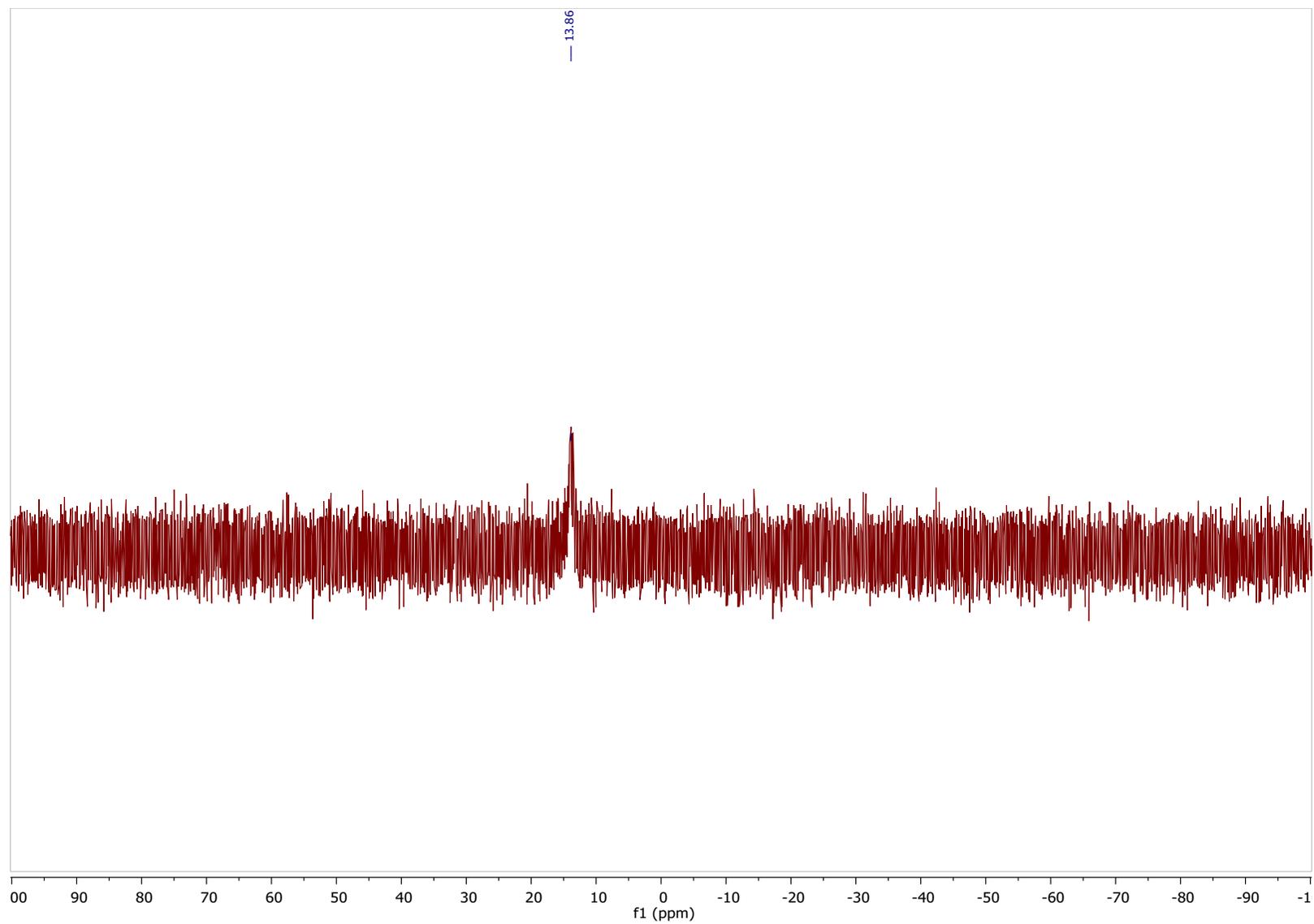


Figure S14 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **7** in $\text{d}_6\text{-DMSO}$.

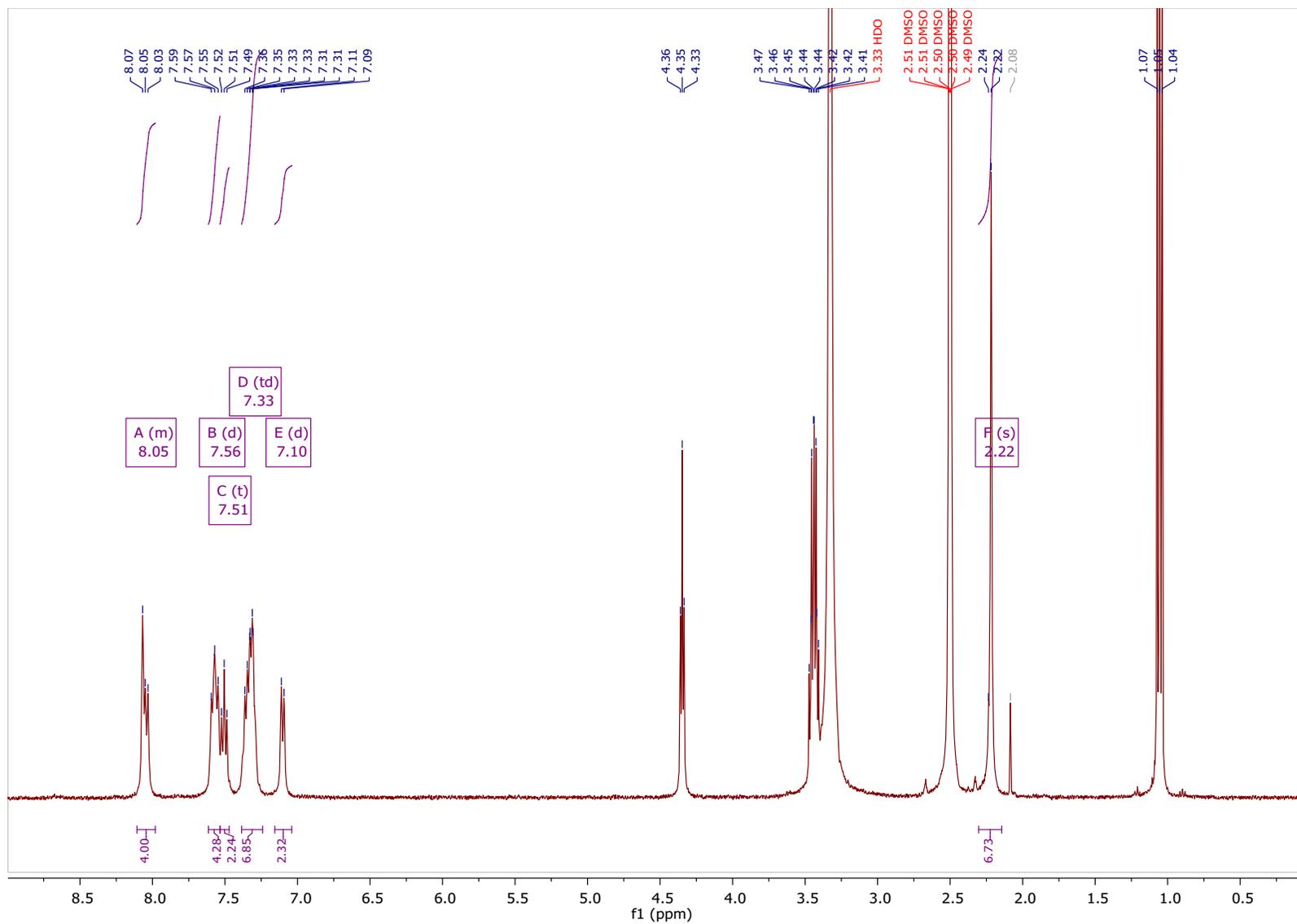


Figure S15 ¹H NMR spectrum of complex ([Bi(*m*-tol)₂(O(O)PPh₂)]·DMSO)_n **8** in d₆-DMSO. Traces of acetone and ethanol can be observed after washing.

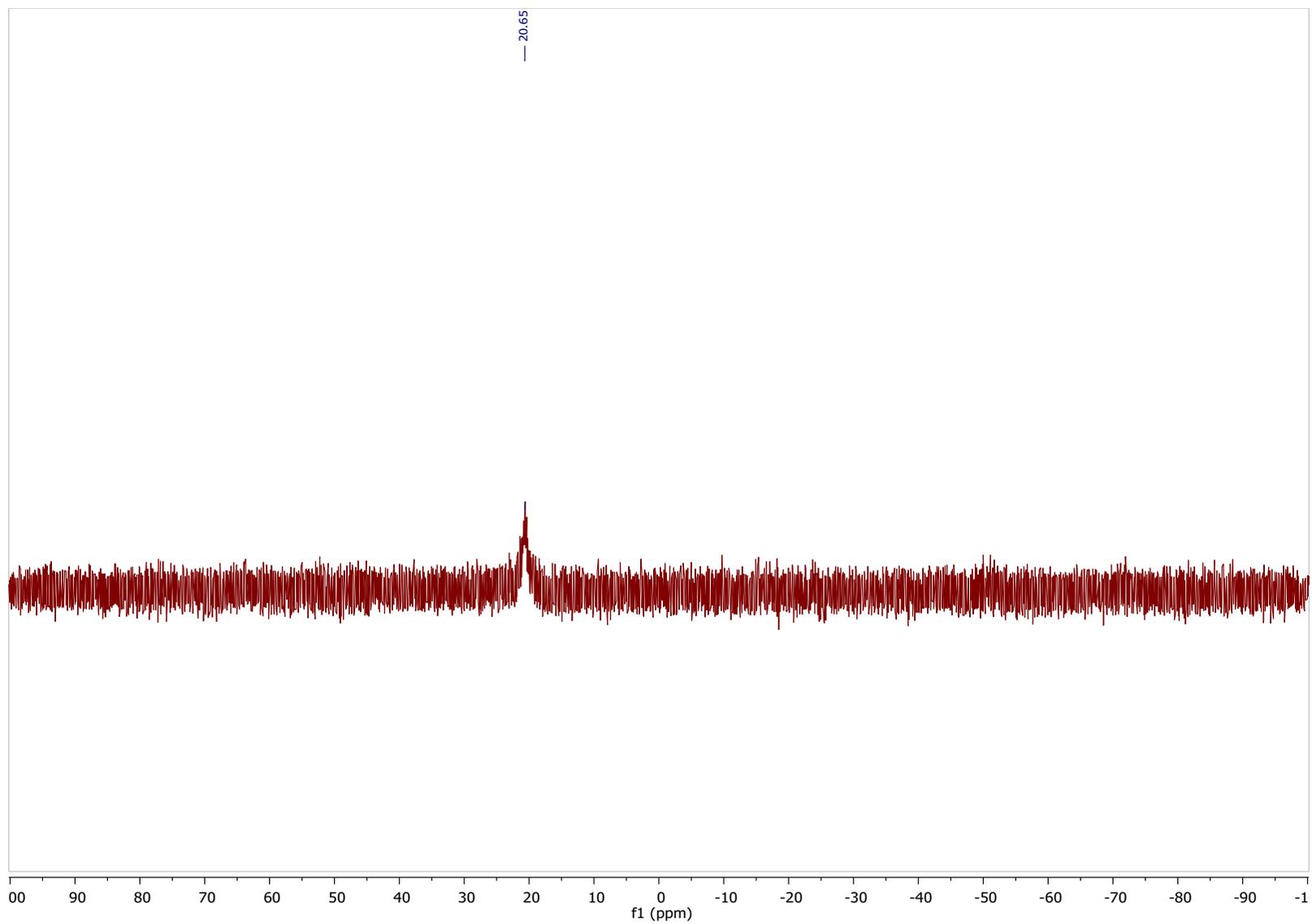


Figure S16 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **8** in d_6 -DMSO.

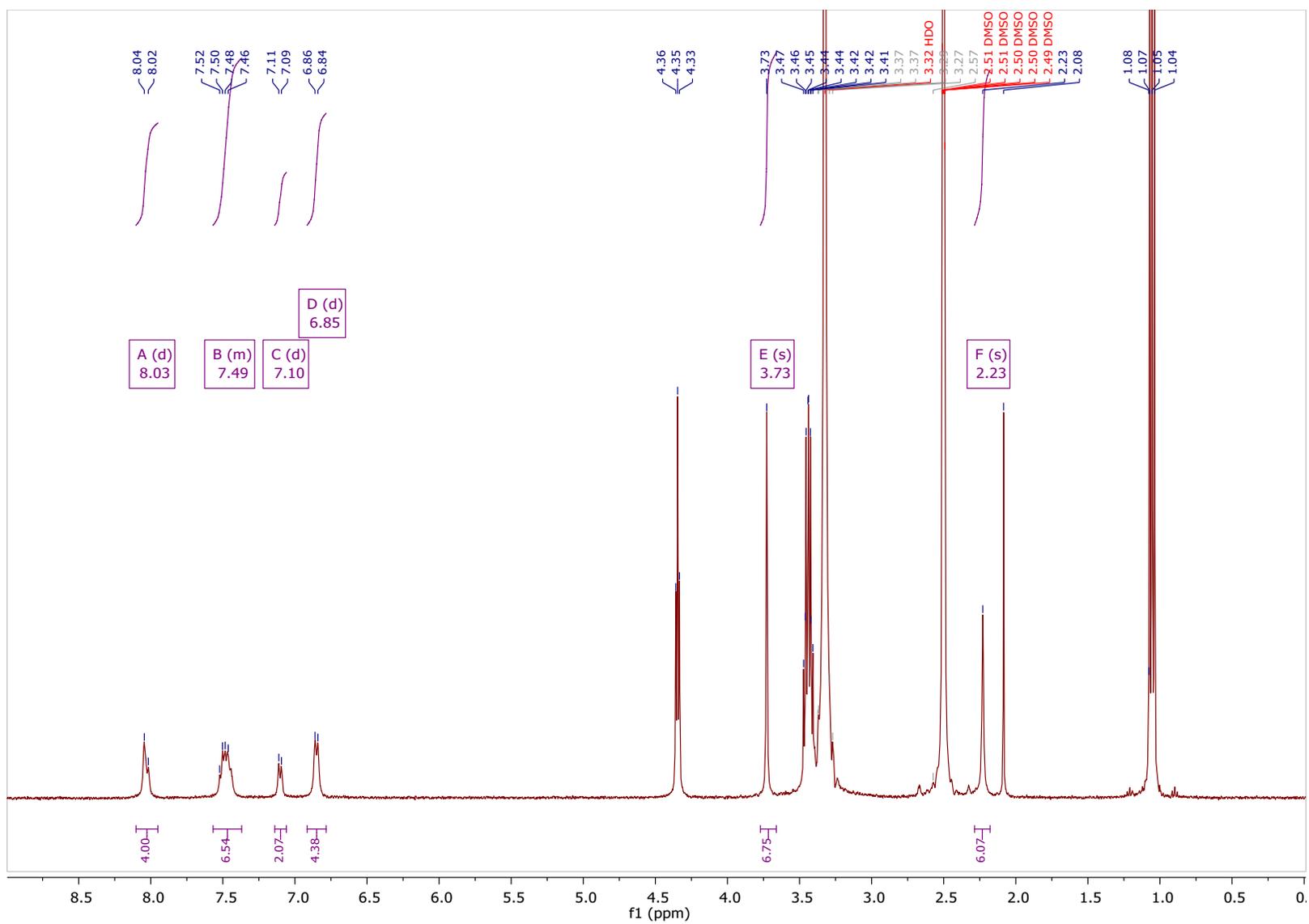


Figure S17 ^3H NMR spectrum of complex $[\text{Bi}(\text{m-tol})_2(\text{O}(\text{O})\text{P}(\text{p-MeOPh})_2)]_n$ **9** in $\text{d}_6\text{-DMSO}$. Traces of acetone and ethanol can be observed after washing.

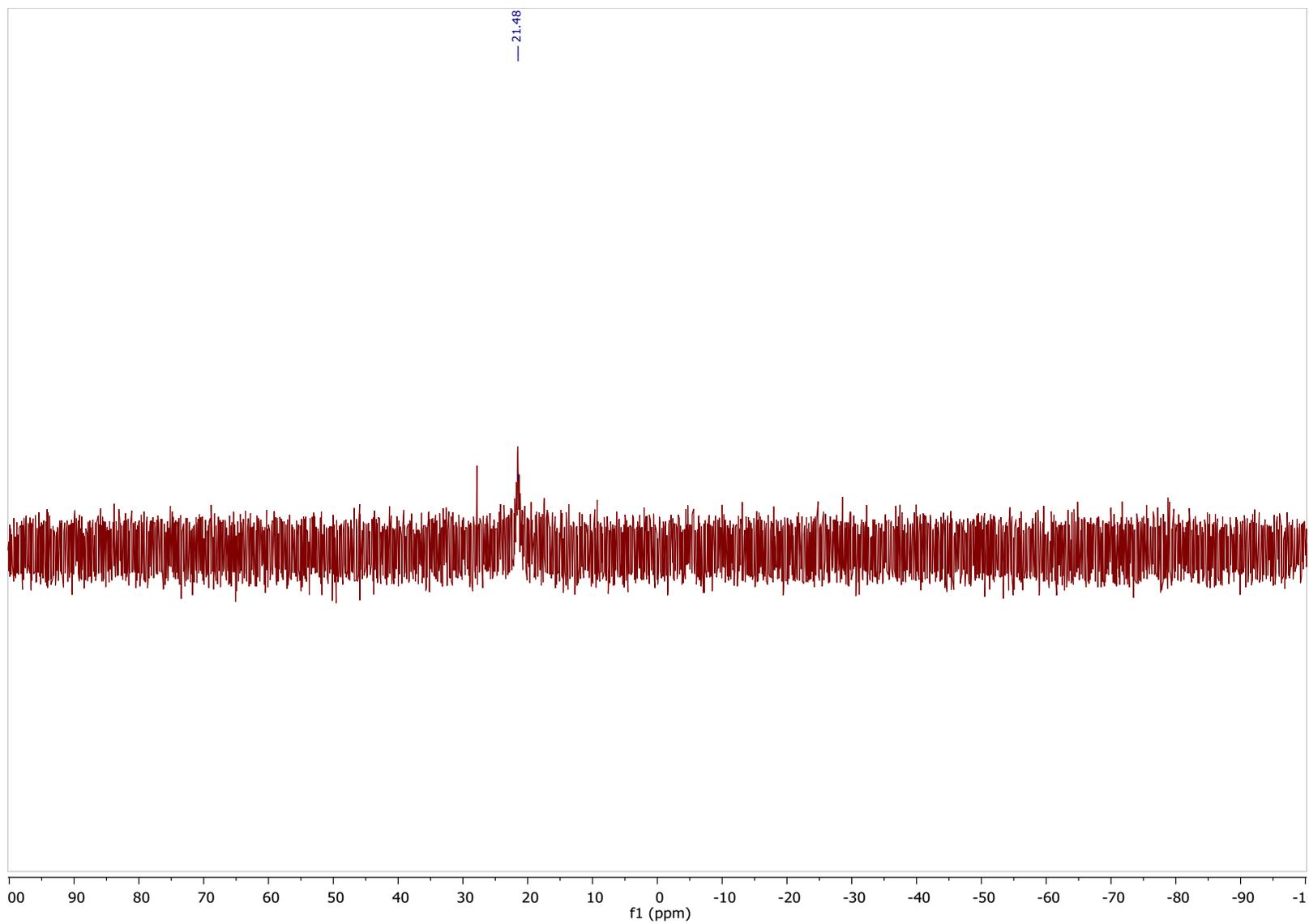


Figure S18 ^{31}P NMR spectrum of complex $[\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{P}(p\text{-MeOPh})_2)]_n$ **9** in d_6 -DMSO.

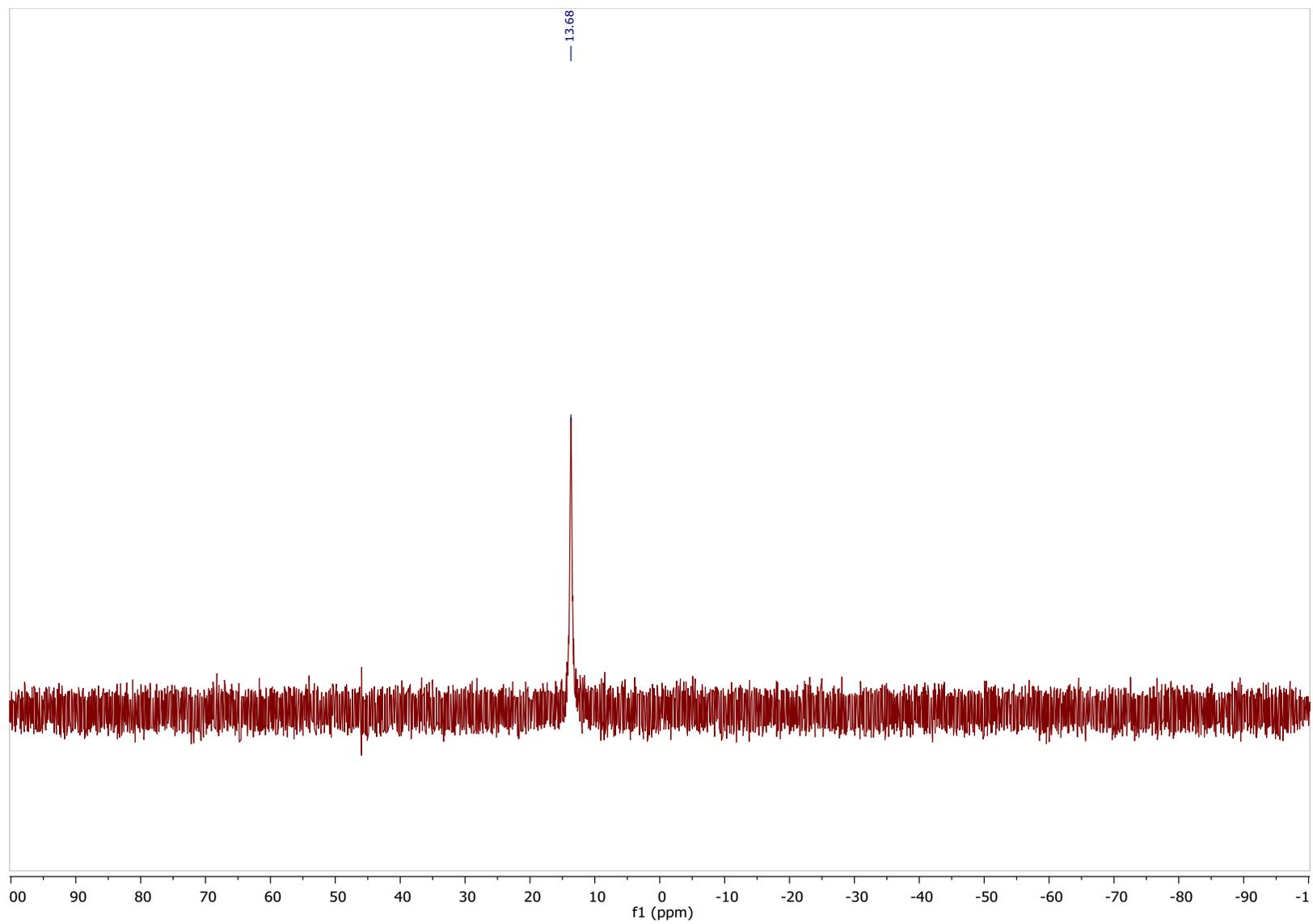


Figure S20 ^{31}P NMR spectrum of complex $[\text{Bi}(p\text{-tol})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **10** in d_6 -DMSO.

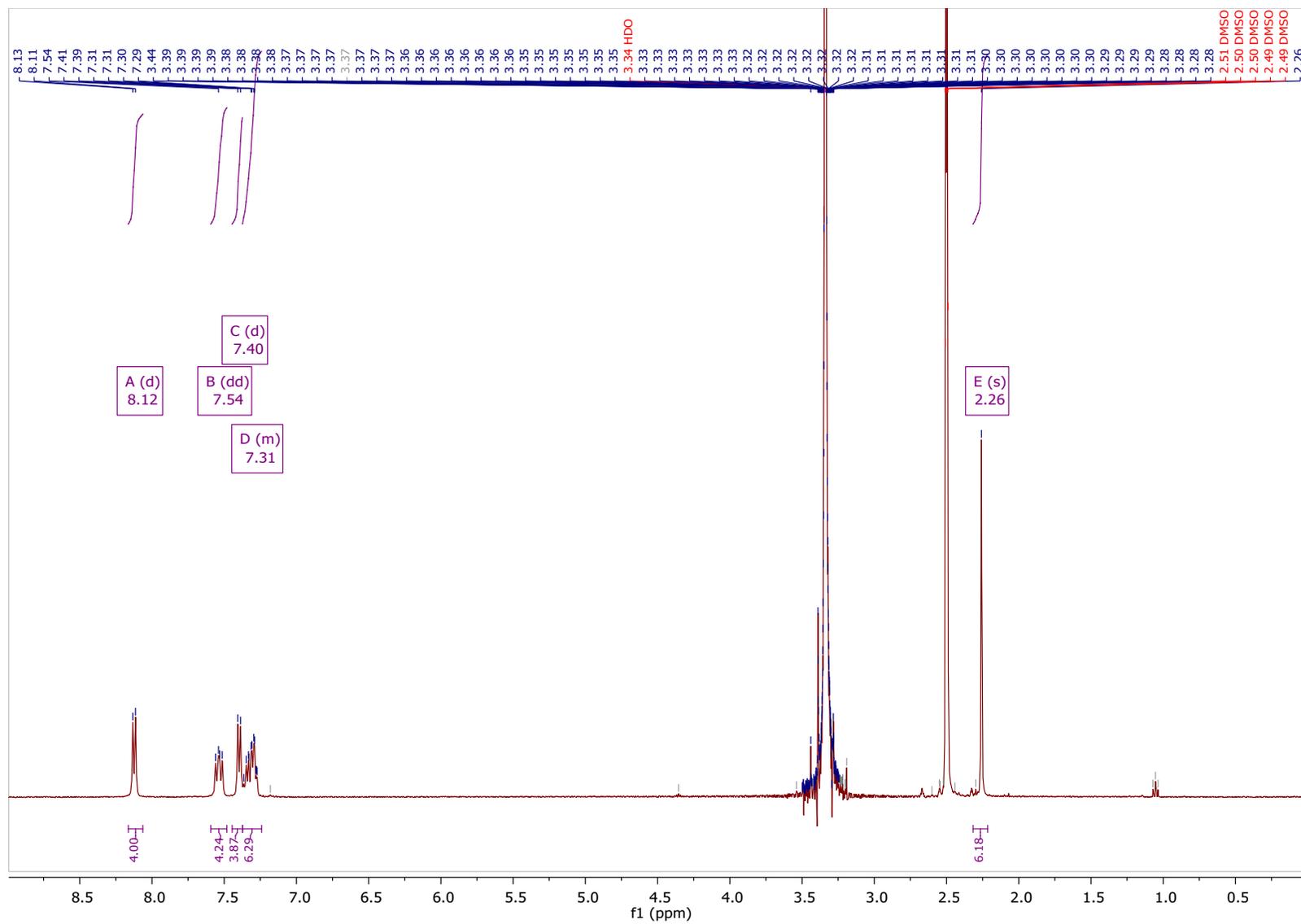


Figure S21 ¹H NMR spectrum of complex [Bi(*p*-tol)₂(O(O)PPh₂)]_n **11** in d₆-DMSO.

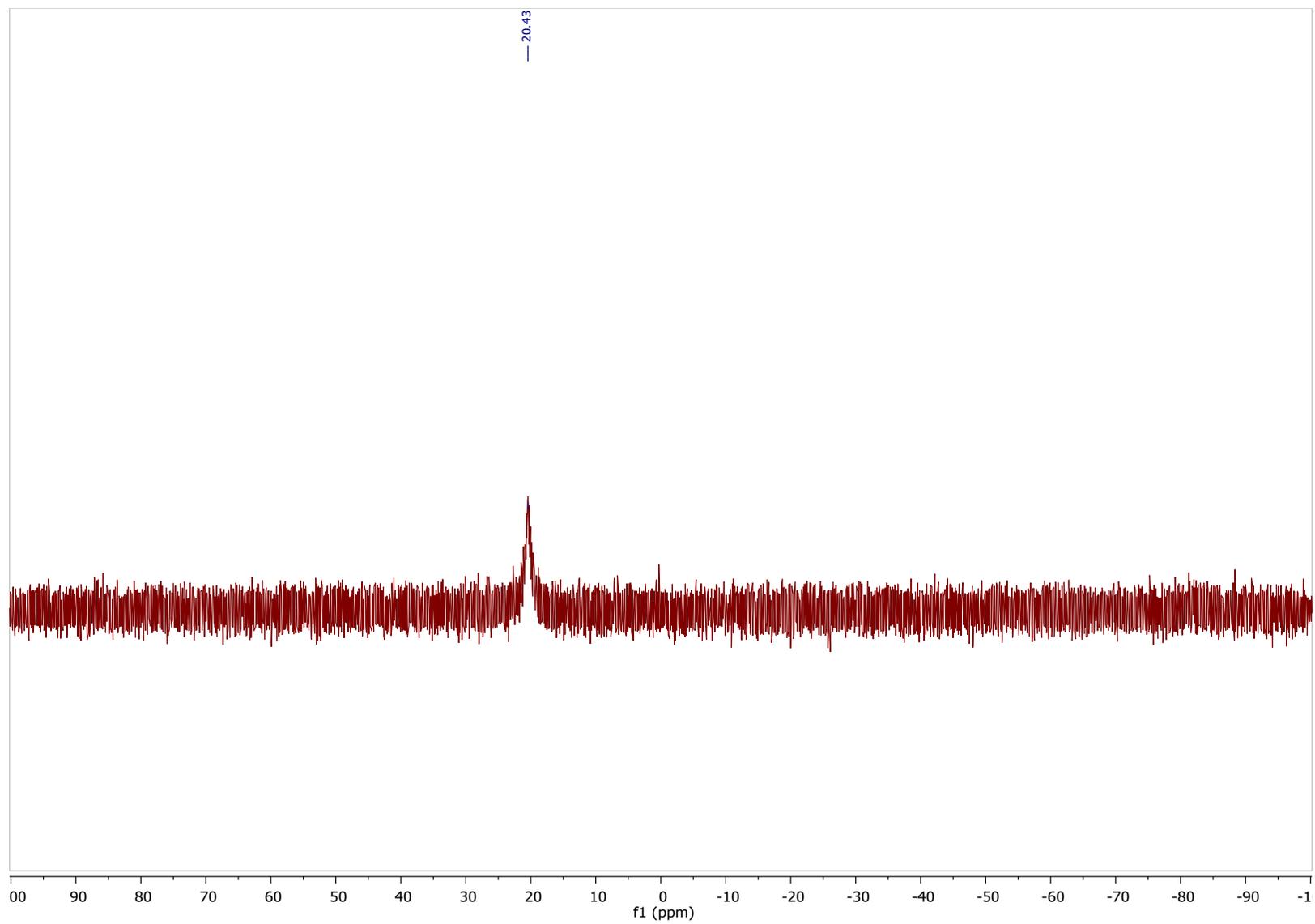


Figure S22 ^{31}P NMR spectrum of complex $[\text{Bi}(p\text{-tol})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **11** in $\text{d}_6\text{-DMSO}$.

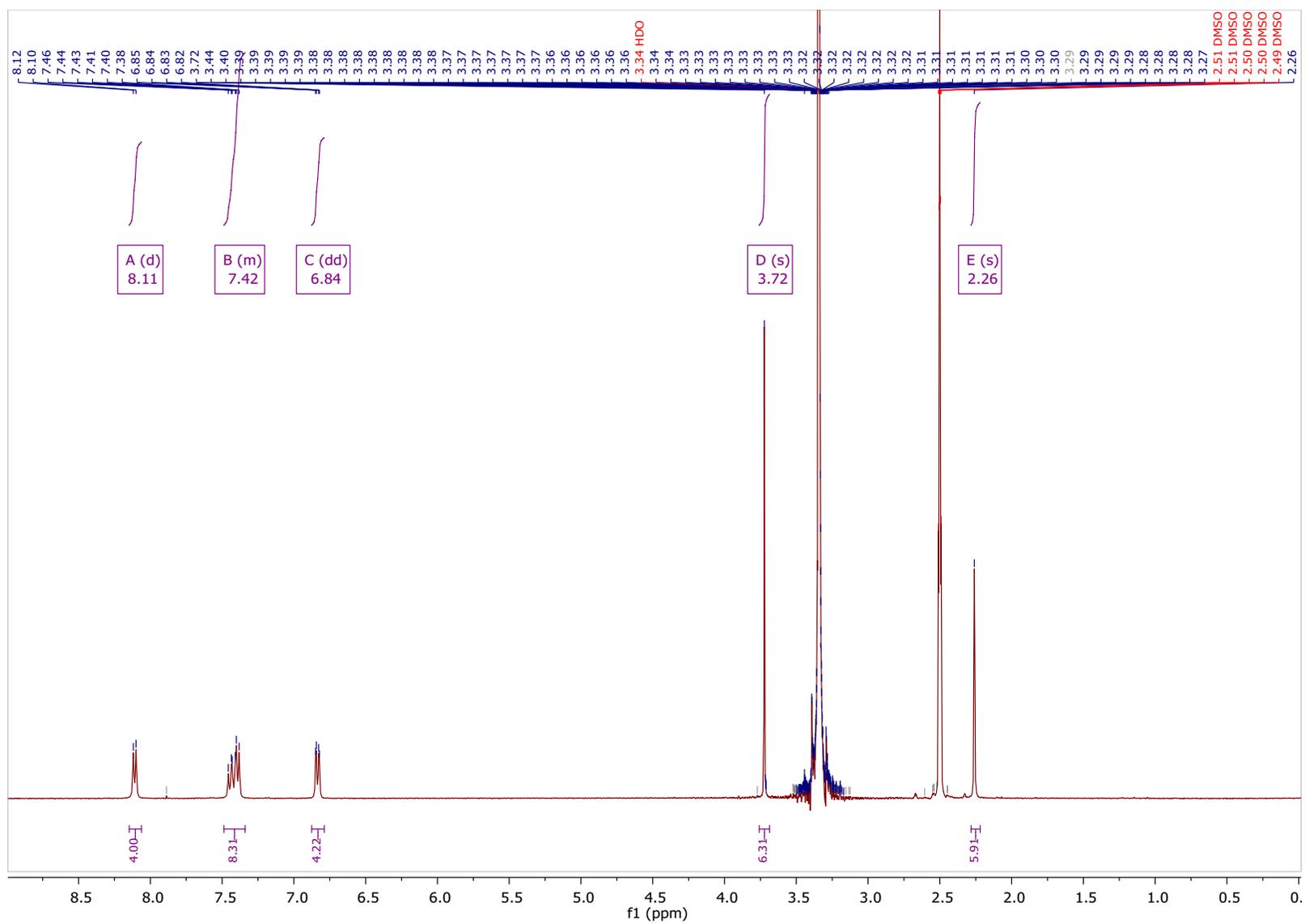


Figure S23 ^1H NMR spectrum of complex $[\text{Bi}(\text{p-tol})_2(\text{O}(\text{O})\text{P}(\text{p-MeOPh})_2)]_n$ **12** in d_6 -DMSO.

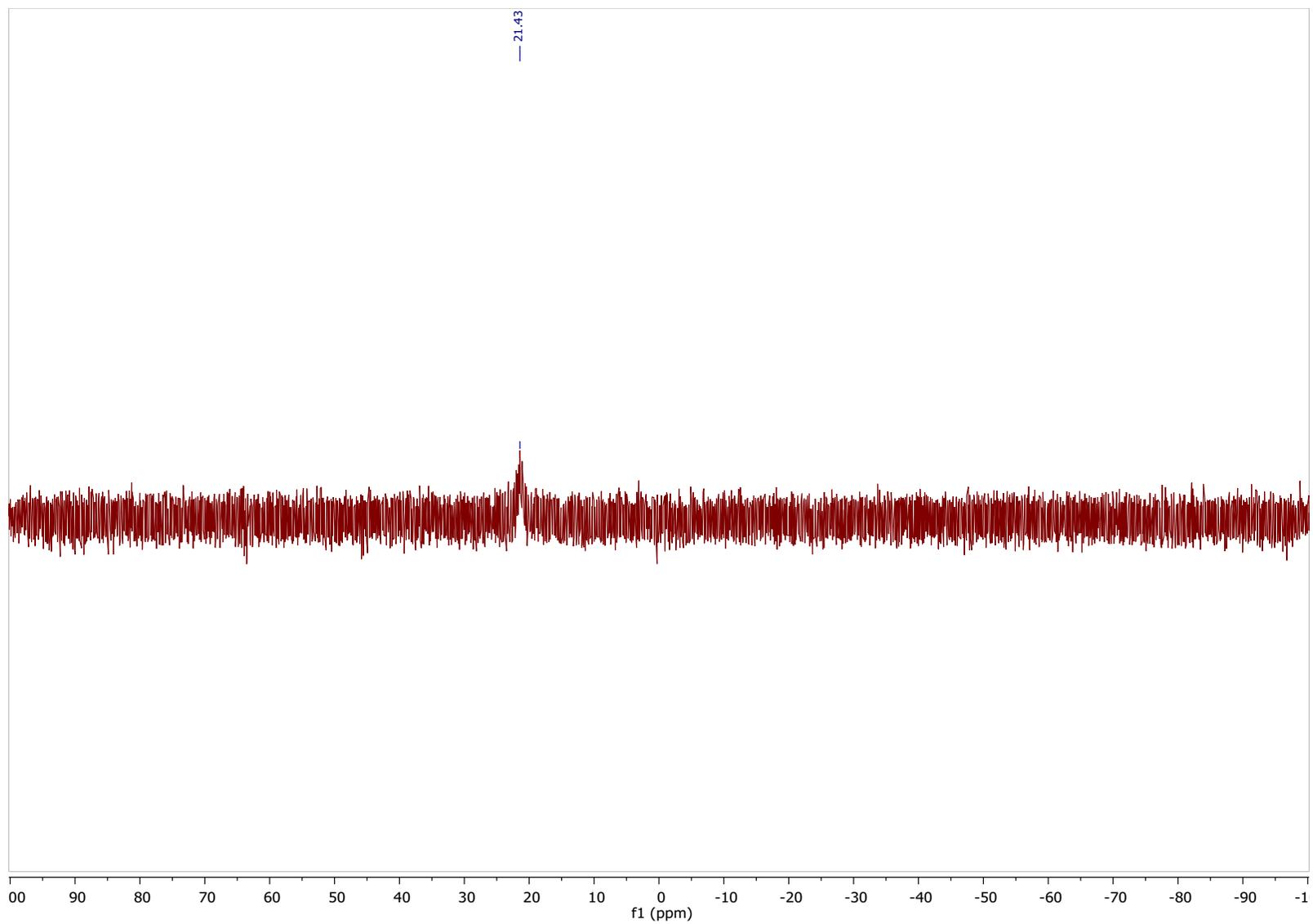
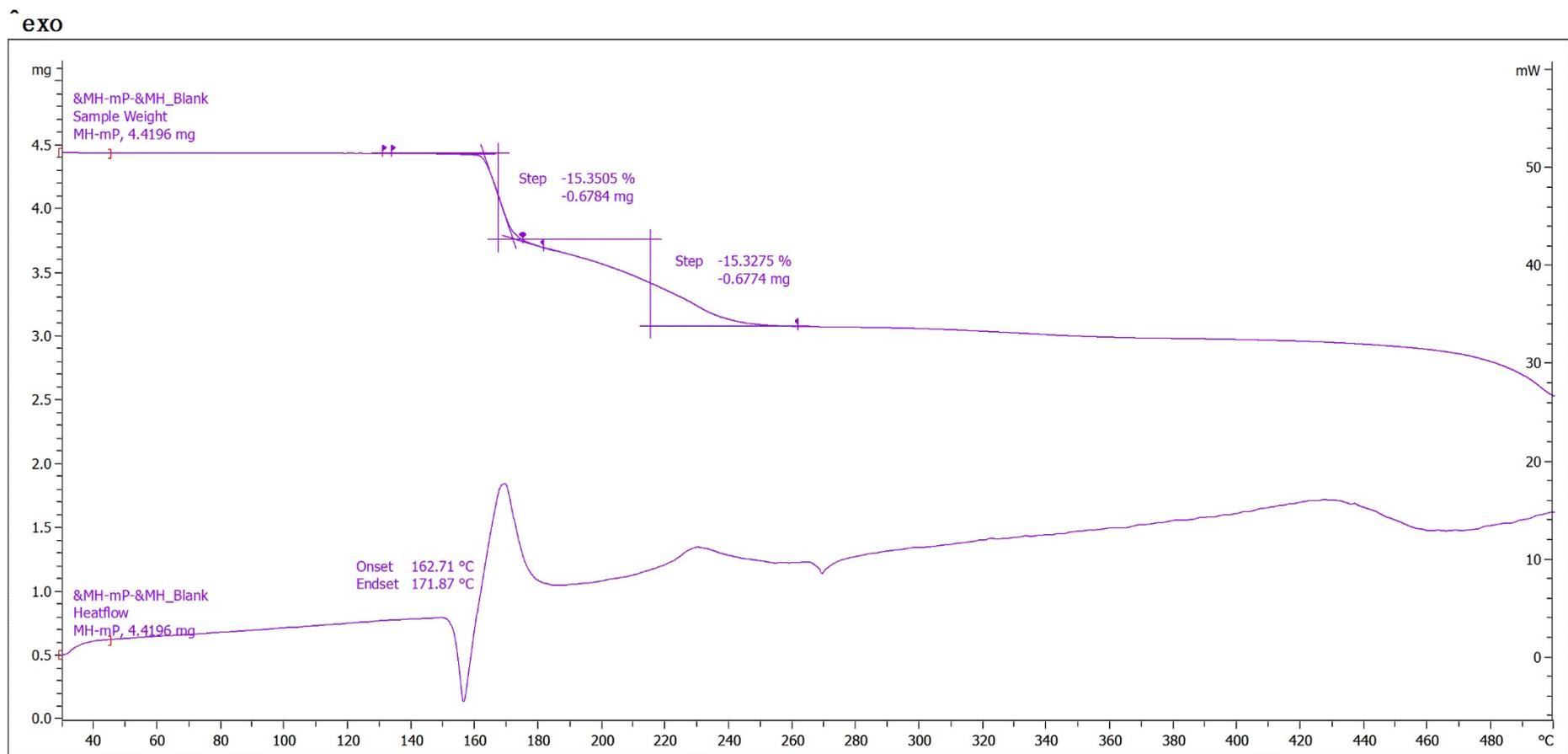


Figure S24 ^{31}P NMR spectrum of complex $[\text{Bi}(\textit{p}\text{-tol})_2(\text{O}(\text{O})\text{P}(\textit{p}\text{-MeOPh})_2)]_n$ **12** in $\text{d}_6\text{-DMSO}$.

Table S1 Characterisation of complexes **1** – **12** compared with their parent phosphinic acids including difference in asymmetric and symmetric PO₂ stretching as determined by IR, and melting temperatures as determined by DSC-TGA.

Compound		$\Delta\nu$ (cm ⁻¹)	Melting point (°C)
[Bi(<i>o</i> -MeOPh) ₂ (O(O)P(H)Ph)] _n	1	94, 105	149*
[Bi(<i>o</i> -MeOPh) ₂ (O(O)PPh ₂)] _n	2	110, 111	177*
[Bi(<i>o</i> -MeOPh) ₂ (O(O)P(<i>p</i> -MeOPh) ₂)] _n	3	107, 112	186*
[Bi(<i>m</i> -MeOPh) ₂ (O(O)P(H)Ph)] _n	4	91, 98	127
[Bi(<i>m</i> -MeOPh) ₂ (O(O)PPh ₂)] _n	5	111, 130	161*
[Bi(<i>m</i> -MeOPh) ₂ (O(O)P(<i>p</i> -MeOPh) ₂)] _n	6	92, 110	151
[Bi(<i>m</i> -tol) ₂ (O(O)P(H)Ph)] _n	7	90, 100	139
[Bi(<i>m</i> -tol) ₂ (O(O)PPh ₂)] _n	8	75, 98	152
[Bi(<i>m</i> -tol) ₂ (O(O)P(<i>p</i> -MeOPh) ₂)] _n	9	98, 99	239*
[Bi(<i>p</i> -tol) ₂ (O(O)P(H)Ph)] _n	10	108, 130	156*
[Bi(<i>p</i> -tol) ₂ (O(O)PPh ₂)] _n	11	88, 93	219*
[Bi(<i>p</i> -tol) ₂ (O(O)P(<i>p</i> -MeOPh) ₂)] _n	12	101, 108	188*
Ph(H)P(O)OH		164	
Ph ₂ P(O)OH		168	
(<i>p</i> -MeOPh) ₂ P(O)OH		177	

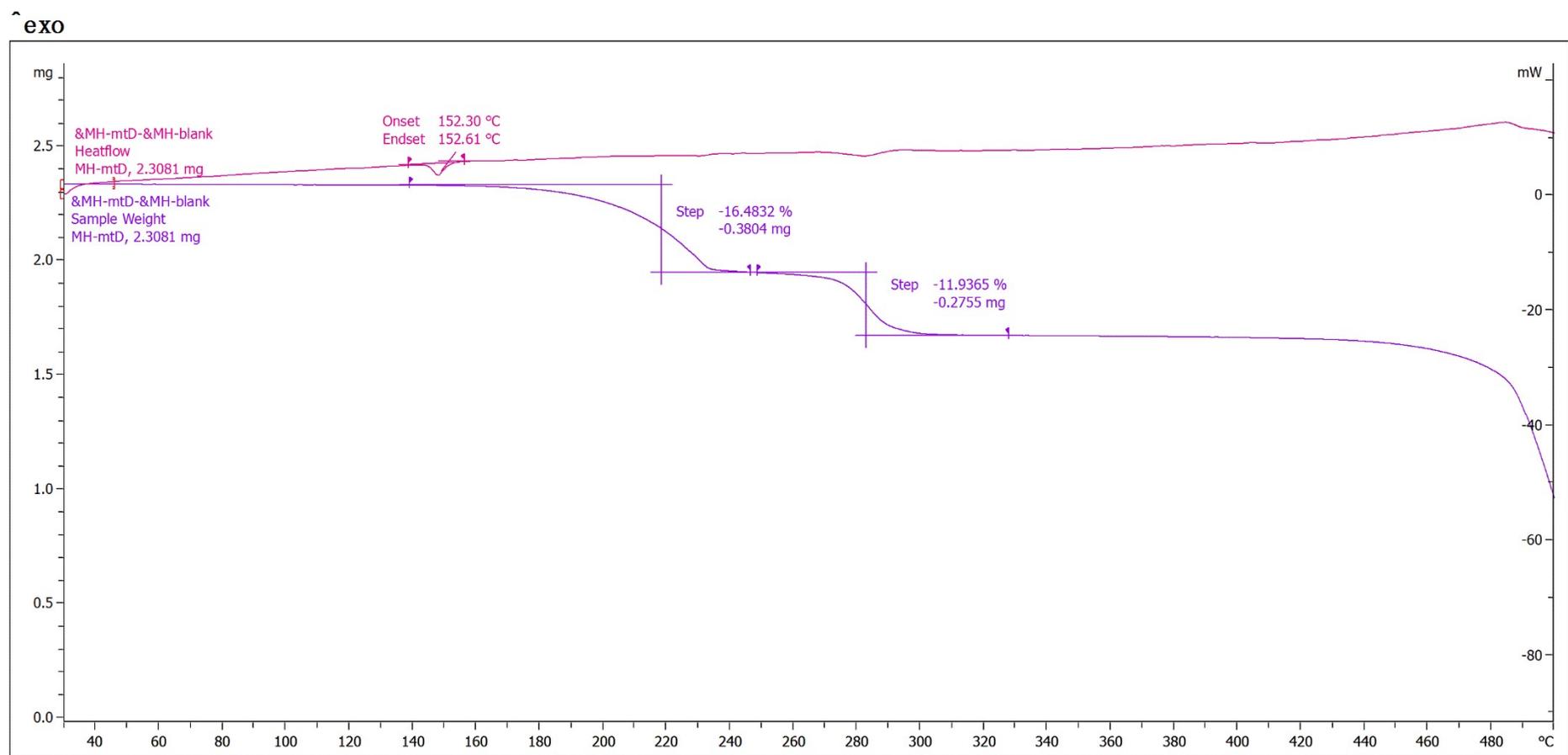
† Denotes solid-state ³¹P NMR. * Denotes decomposition point.



Lab: METTLER

STAR^e SW 10.00

Figure S25 TGA trace of complex $[\text{Bi}(m\text{-MeOPh})_2\text{O}(\text{O})\text{P}(\text{H})\text{Ph}]_n$ **4**, showing the loss of two methoxyphenyl moieties. DSC trace is shown below.



Lab: METTLER

STAR^e SW 10.00

Figure S26 TGA trace (purple) of complex $[\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **8**, showing the loss of two tolyl moieties. DSC trace is shown in pink.

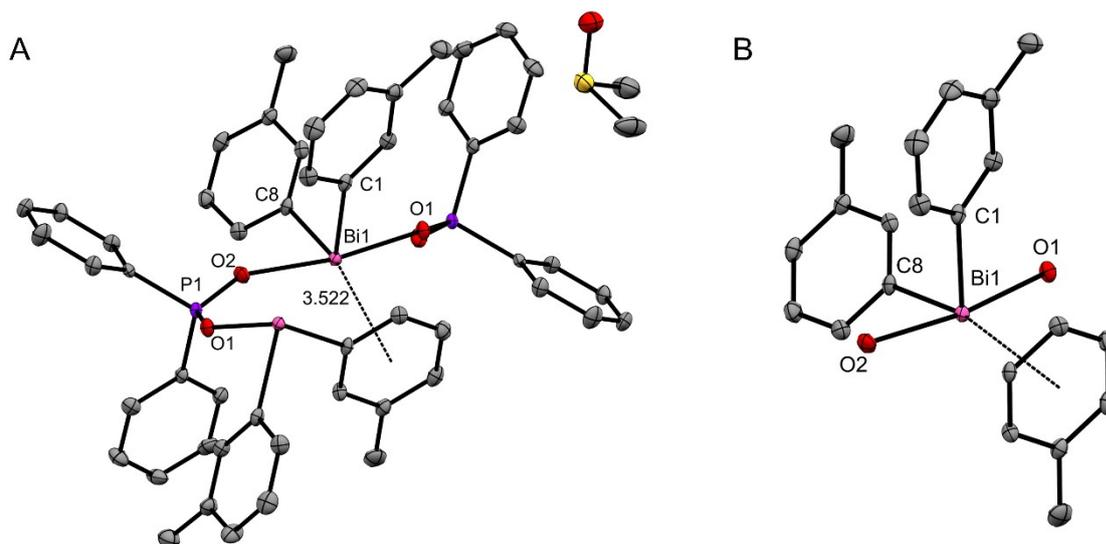


Figure S27 Polymeric structure of $([\text{Bi}(m\text{-tol})_2(\text{O}(\text{O})\text{PPh}_2)].\text{DMSO})_n$ **8** (A) with DMSO in the lattice, and geometry around the bismuth centre (B) showing the distorted square pyramidal confirmation with evidence of a sterically active lone pair. Thermal ellipsoids shown at 50 % probability with hydrogen atoms omitted for clarity. Selected bond lengths (\AA) and angles ($^\circ$): Bi1-O1 2.3238(17); Bi1-O2 2.3742(17); Bi1-C1 2.255(2); Bi1-C8 2.264(2); Bi1-arene_{centroid} 3.522; O1-Bi1-O2 172.23(6); O1-Bi1-C1 86.30(8); O1-Bi1-C8 90.07(8); O2-Bi1-C1 88.51(8); O2-Bi1-C8 84.42(7); C1-Bi1-C8 92.93(9); C8-Bi1-arene_{centroid} 169.04. Symmetry operator: $(-x + 1/2, y + 1/2, -z + 1/2)$.

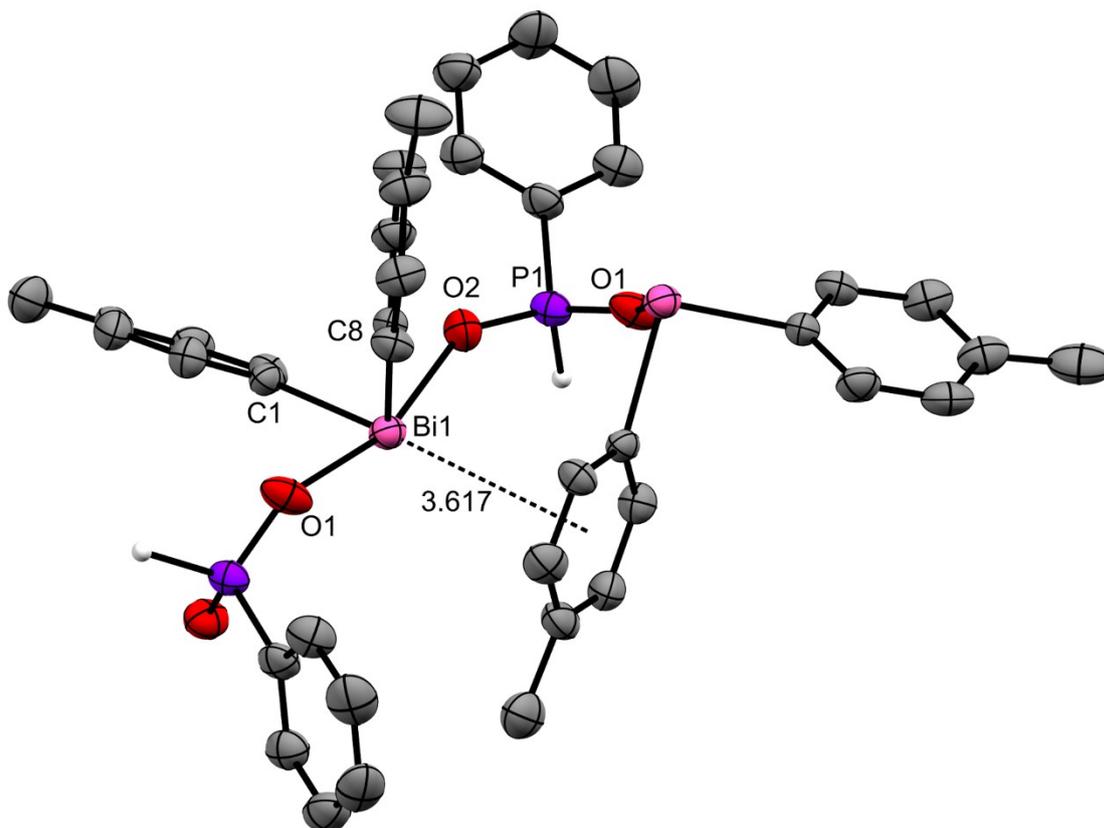


Figure S28 Molecular structure of complex $[\text{Bi}(p\text{-tol})_2(\text{O}(\text{O})\text{P}(\text{H})\text{Ph})]_n$ **10**. Thermal ellipsoids shown at 50 % probability with aromatic hydrogen atoms omitted for clarity. Selected bond lengths (\AA) and angles ($^\circ$): Bi1-O1 2.350(5); Bi1-O2 2.361(5); Bi1-C1 2.245(6); Bi1-C8 2.226(6); Bi1-arene_{centroid} 3.617; O1-Bi1-O2 162.66(18); O1-Bi1-C1 86.2(2); O1-Bi1-C8 82.19(18); O2-Bi1-C1 82.60(19); O2-Bi1-C8 85.49(18); C1-Bi1-C8 94.5(2); C1-Bi1-arene_{centroid} 171.25. Symmetry operator: $(-x + 1/2, y - 1/2, -z + 3/2)$.

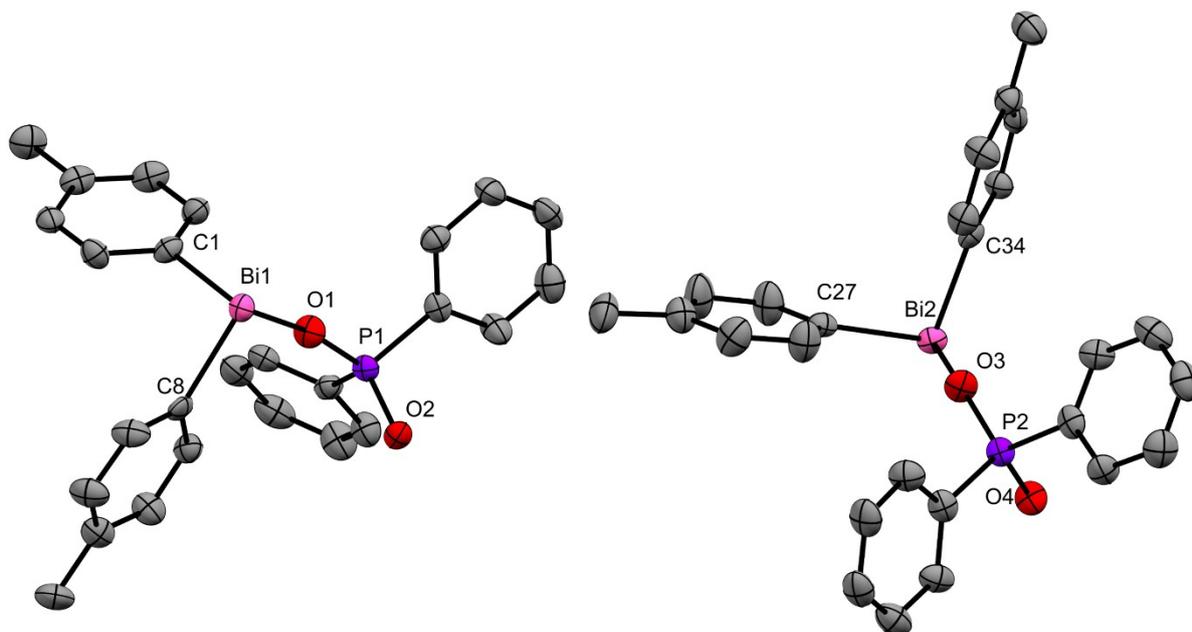


Figure S29 Molecular structures of complex $[\text{Bi}(p\text{-tol})_2(\text{O}(\text{O})\text{PPh}_2)]_n$ **11**. Thermal ellipsoids shown at 50 % probability with hydrogen atoms omitted for clarity. Selected bond lengths (\AA) and angles ($^\circ$): Bi1-O1 2.379(6); Bi1-O2 2.329(6); Bi1-C1 2.272(8); Bi1-C8 2.254(9); Bi1-arene_{centroid} 3.429; Bi2-O3 2.355(6); Bi2-O4 2.361(6); Bi2-C27 2.262(9); Bi2-C34 2.249(10); Bi1-arene_{centroid} 3.509; O1-Bi1-O2 172.1(2); O1-Bi1-C1 84.4(3); O1-Bi1-C8 90.2(3); O2-Bi1-C1 89.0(3); O2-Bi1-C8 85.5(3); C1-Bi1-C8 90.8(3); C1-Bi1-arene_{centroid} 171.16; O3-Bi2-O4 171.2(2); O3-Bi2-C27 85.8(3); O3-Bi2-C34 86.4(3); O4-Bi2-C27 87.8(3); O4-Bi2-C34 88.0(3); C27-Bi2-C34 93.2(3); C27-Bi2-arene_{centroid} 90.20. Symmetry operator: Bi1 $(-x + 1, y + 1/2, -z + 2)$ and Bi2 $(-x, y + 1/2, -z + 1)$.

Table S2 Minimum inhibitory concentrations (MICs) of triaryl bismuth reagents (BiAr_3) to inhibit $\geq 80\%$ growth of MRSA, VRE, *E. coli*, and *P. aeruginosa*, as measured by absorbance at 600 nm following overnight incubation at 37 $^\circ\text{C}$.

Compound	MIC (μM)			
	MRSA	VRE	<i>E. coli</i>	<i>P. aeruginosa</i>
$\text{Bi}(o\text{-MeOPh})_3$	> 100	> 100	> 100	> 100
$\text{Bi}(m\text{-MeOPh})_3$	> 100	> 100	> 100	> 100
$\text{Bi}(m\text{-tol})_3$	> 100	> 100	> 100	> 100
$\text{Bi}(p\text{-tol})_3$	> 50	> 50	> 50	> 50

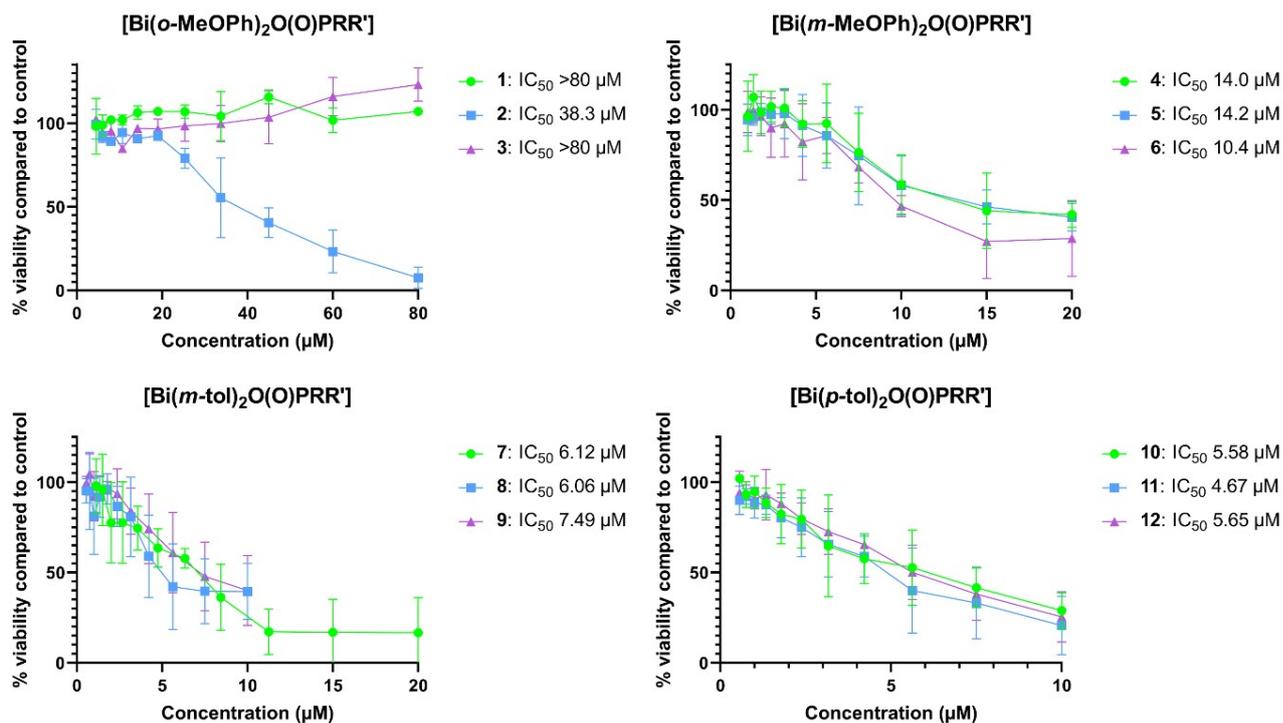


Figure S30 Cos-7 viability following 24 hour incubation at 37 °C, 5 % CO₂ with complexes 1 – 12.

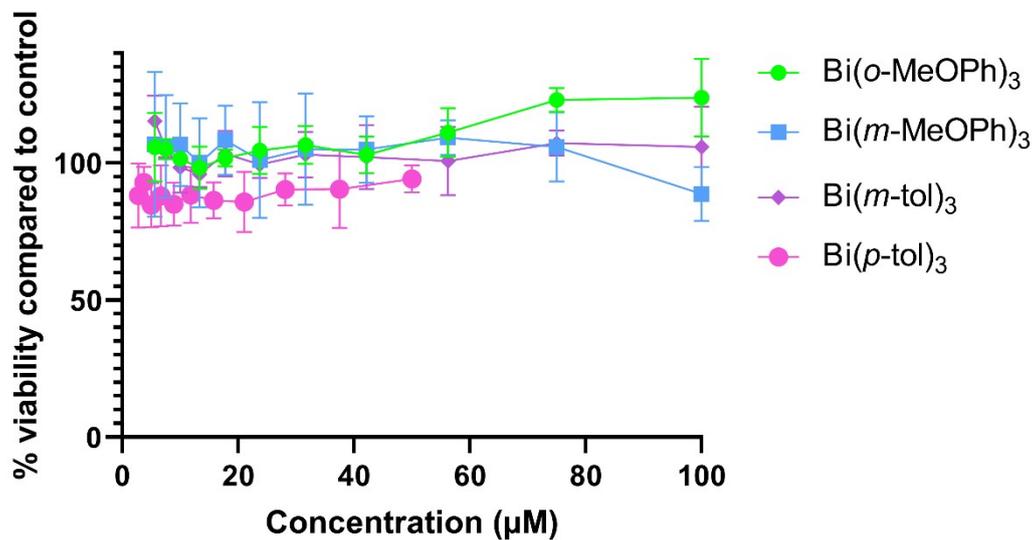


Figure S31 Cos-7 viability following 24 hour incubation at 37 °C, 5 % CO₂ with BiAr₃.

Table S3 Summary of crystallographic data for complexes **4**, **7**, **8**, **10** and **11**.

Complex	[Bi(<i>m</i> -MeOPh) ₂ O(O)P(H)Ph] 4	[Bi(<i>m</i> -tol) ₂ O(O)P(H)Ph] 7	[Bi(<i>m</i> -tol) ₂ O(O)PPh ₂] .DMSO 8	[Bi(<i>p</i> -tol) ₂ O(O)P(H)Ph] 10	[Bi(<i>p</i> -tol) ₂ O(O)PPh ₂] 11
CCDC	2130070	2130069	2130071	2130072	2130073
Formula	C ₂₀ H ₂₀ BiO ₄ P	C ₂₀ H ₂₀ BiO ₂ P	C ₂₆ H ₂₄ BiO ₂ P, C ₂ H ₆ OS	C ₂₀ H ₂₀ BiO ₂ P	C ₂₆ H ₂₄ BiO ₂ P
Formula weight	564.31	532.31	686.53	532.31	608.40
Crystal system	Orthorhombic	Monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ 2 ₁ 2 ₁	P2 ₁	P2 ₁ /n	C2/c	P2 ₁
<i>a</i> (Å)	9.4175(4)	8.7590(18)	14.2607(2)	18.7237(4)	11.7325(2)
<i>b</i> (Å)	9.4368(3)	9.4710(19)	10.1056(2)	9.2659(2)	9.3795(2)
<i>c</i> (Å)	21.9641(6)	11.418(2)	18.4960(4)	23.2109(5)	10.9611(5)
α (°)	90	90	90	90	90
β (°)	90	99.95(3)	95.6810(10)	107.783(5)	96.470(2)
γ (°)	90	90	90	90	90
Volume (Å ³)	1951.97(12)	932.9(3)	2652.42(9)	3834.50(15)	2291.97(8)
Z	4	2	4	8	4
Density (g cm ⁻³)	1.920	1.895	1.719	1.844	1.763
μ (mm ⁻¹)	9.136	9.543	6.813	18.902	7.782
F ₀₀₀	1080	508	1344	2032	1176
T (K)	123(2)	293(2)	123.00(10)	122.99(10)	123.00(10)
2 θ _{max} (°)	56.6	57.1	62.1	160.0	52.7
Reflections collected	18660	11397	34988	20042	25295
Unique reflections	4810 (R _{int} = 0.0462)	3692 (R _{int} = 0.0198)	7568 (R _{int} = 0.0455)	4101 (R _{int} = 0.0622)	9013 (R _{int} = 0.0373)
Final Goof	1.044	1.173	1.083	1.031	1.013
R ₁	0.0264	0.0184	0.0240	0.0386	0.0310
wR ₂ (all data)	0.0482	0.0475	0.0574	0.1282	0.0789
Flack parameter	-0.038(6)	-0.017(3)	-	-	-0.021(5)