Supporting information for

## Use of heterometallic alkali metal-magnesium aryloxides in ring-opening

#### polymerization of cyclic esters

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# Crystallographic Data for Compounds 4-8.

Crystal	4	5	<b>6</b> ·(C <sub>4</sub> H <sub>8</sub> O)	<b>7</b> ·2(C <sub>4</sub> H <sub>8</sub> O)	8
Chemical formula	C54H54Li2Mg2O18	C <sub>27</sub> H <sub>27</sub> KMgO <sub>9</sub>	C118H118AlMg6Na4O50	C86H98Mg4Na2O30	$C_{40}H_{56}Mg_4O_{20}$
Formula Mass	1053.47	558.89	2600.92	1754.86	954.08
Crystal system	Monoclinic	Triclinic	Monoclinic	Monoclinic	Tetragonal
Space group	$P2_{1}/n$	<i>P</i> 1	C2/c	$P2_{1}/n$	$I4_{1}/a$
a/Å	11.669 (2)	6.610 (2)	58.522 (18)	17.285 (3)	15.850 (3)
b/Å	17.376 (3)	12.874 (4)	15.832 (5)	13.750 (3)	
$c/{ m \AA}$	13.155 (2)	16.166 (5)	27.991 (12)	18.986 (4)	18.016 (3)
a/°		97.48 (4)			
$eta / ^{\circ}$	92.27 (2)	90.45 (4)	108.43 (5)	106.54 (2)	
γ/°		97.24 (4)			
Unit cell volume/Å3	2665.2 (8)	1352.7 (7)	24604 (17)	4325.7 (16)	4526.0 (19)
Temperature/K	100(2)	100(2)	100(2)	100(2)	100(2)
Ζ	2	2	8	2	4
Radiation type	ΜοΚα	ΜοΚα	ΜοΚα	ΜοΚα	ΜοΚα
Absorption coefficient, $\mu/\text{mm}^{-1}$	0.12	0.27	0.15	0.14	0.16
No. of reflections measured	27574	14609	54054	41380	5798
No. of independent reflections	5802	14609	24832	10372	2436
No. of observed reflections	observed reflections		8570	6235	1946
$(I > 2\sigma(I))$	5127	9709	8370		
Rint	0.0390	-	0.0948	0.0801	0.0784
Final $R_I$ values $(I > 2\sigma(I))$	0.0422	0.0727	0.0782	0.0480	0.0775
Final $wR(F^2)$ values $(I > 2\sigma(I))$	0.1122	0.1594	0.1564	0.1118	0.1980
Final <i>R</i> <sup>1</sup> values (all data)	0.0477	0.1130	0.2219	0.0855	0.0916
Final $wR(F^2)$ values (all data)	0.1188	0.1594	0.2427	0.1215	0.2126
Goodness of fit on $F^2$	1.032	0.97	0.98	0.87	1.09
Δρmax/eÅ <sup>-3</sup>	0.35	0.51	0.82	0.49	0.77
Δpmin/eÅ <sup>-3</sup>	-0.25	-0.57	-0.44	-0.51	-0.67

## Table S1. Crystal and data collection parameters for compounds 4-8.



**Figure S1**. Molecular structure of  $[Mg_2Li_2(MesalO)_6]$  (1) with displacement ellipsoids drawn at the 25% probability level. Hydrogen atoms are omitted for clarity [symmetry code: (i) -x+1, -y+1, -z+1]. Reprinted (adapted) with permission from *Macromolecules* 2021, 54, 5, 2449–2465. Copyright 2021 American Chemical Society.



**Figure S2**. Molecular structure of  $[Mg_2Na_2(MesalO)_6(THF)_y]$  (2) for y = 2 or 4 with displacement ellipsoids drawn at the 25% probability level. The hydrogen atoms are omitted for clarity [symmetry code: (i) -x+1, -y+1, -z+1; (ii) -x, -y, -z]. Reprinted (adapted) with permission from *Macromolecules* 2021, 54, 5, 2449–2465. Copyright 2021 American Chemical Society.



**Figure S3.** Molecular structure of  $[Mg_2K_2(MesalO)_6(THF)_4]$  (**3**) with displacement ellipsoids drawn at the 25% probability level. The hydrogen atoms are omitted for clarity [symmetry code: (i) -x+1, -y+1, -z+1]. Reprinted (adapted) with permission from *Macromolecules* 2021, 54, 5, 2449–2465. Copyright 2021 American Chemical Society.



Figure S4. <sup>1</sup>H NMR spectrum of 4 in THF-d<sub>8</sub>.



Figure S6. <sup>7</sup>Li NMR spectrum of 4 in THF-d<sub>8</sub>.







Figure S8. <sup>1</sup>H NMR spectrum of 4 in THF-d<sub>8</sub> at 50 °C.



Figure S10. <sup>1</sup>H NMR spectrum of 5 in THF-d<sub>8</sub>.





**Figure S12**. <sup>1</sup>H-DOSY NMR spectrum of **5** in THF-d<sub>8</sub>. \* - assigned the residues of MesalO ligands.







Figure S14. <sup>1</sup>H NMR spectrum of 6 in THF-d<sub>8</sub>.





Figure S16. FTIR-ATR spectrum of 6.



Figure S17. <sup>1</sup>H NMR spectrum of 7 in THF-d<sub>8</sub>.



Figure S18.  ${}^{13}C{}^{1}H$  NMR spectrum of 7 in THF-d<sub>8</sub>.



Figure S19. FTIR-ATR spectrum of 7.



Figure S20. <sup>1</sup>H NMR spectrum of 8 in THF-d<sub>8</sub>.



Figure S21. <sup>13</sup>C NMR spectrum of 8 in THF-d<sub>8</sub>.



Figure S22. FTIR-ATR spectrum of 8.



**Figure S23**. <sup>1</sup>H-DOSY NMR spectrum of the mixture of **1** and 4 equiv. of cetyl alcohol in THF-d<sub>8</sub>.



Figure S24. <sup>1</sup>H-DOSY NMR spectrum of the mixture of 2 and 4 equiv. of cetyl alcohol in THF- $d_8$ .

compound	$\log(D_{\rm x,norm})$	Fw (g/mol)	Fw <sub>calc</sub> (g/mol)	r <sub>x-ray</sub> (Å)	r <sub>H</sub> (Å)	Δ Fw (%)
1	-9.168	969	944	6.82	8.42	3
2	-9.216	1290	1214	7.23	9.16	6
3	-9.236	1322	1354	7.24	9.49	2
1	-9.234	969	1060	6.82	8.75	9
2	-9.232	1290	1264	7.23	9.28	2
3	-9.236	1322	1302	7.24	9.37	2
5	-9.258	1406	1519	6.86	9.86	7

Table S2. Formula weights (Fws) and the hydrodynamic radii  $(r_H)$  of 1-3 and 5 estimated from Stokes-Einstein Gierer-Wirtz method.

**Table S3.** Diffusion coefficients and estimated formula weight for 1-3 and 6-7 determine using calibration plots.

compound	$\log(D_{\rm x})$	Fw (g/mol)	Fw <sub>calc</sub> (g/mol)	Δ Fw (%)
1	-9.200	969	1156	19
2	-9.230	1290	1273	2
3	-9.290	1322	1516	15
7	-9.233	1610	2035	26
	-9.238 ( <b>2</b> )	1290	1107	14
6	-9.333 (7)	1610	1702	6
	-9.436	2556	2710	6



**Figure S25**. <sup>1</sup>H DOSY NMR spectrum of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene, and **1** in THF-d<sub>8</sub>.



**Figure S26**. <sup>1</sup>H DOSY NMR spectrum of a mixture of anthracene,  $\alpha$ , $\alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene, and **2** in THF-d<sub>8</sub>.



**Figure S27**. <sup>1</sup>H DOSY NMR spectrum of a mixture of anthracene,  $\alpha$ , $\alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene, and **3** in THF-d<sub>8</sub>.



**Figure S28**. Plot of diffusion coefficient (log*D*) versus formula weight (log Fw) of a mixture of anthracene,  $\alpha$ , $\alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene and **1** in THF-d<sub>8</sub>.



**Figure S29**. Plot of diffusion coefficient (log*D*) versus formula weight (log Fw) of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene and **2** in THF-d<sub>8</sub>.



**Figure S30**. Plot of diffusion coefficient (log*D*) versus formula weight (log Fw) of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene and **3** in THF-d<sub>8</sub>.



Figure S31. <sup>1</sup>H-DOSY NMR spectrum in THF-d<sub>8</sub> of the PLLA synthesized using 1 and ROH.



Figure S32. <sup>1</sup>H-DOSY NMR spectrum in THF-d<sub>8</sub> of the PLLA synthesized using 2 and ROH.



**Figure S33**. <sup>1</sup>H DOSY NMR spectrum of a mixture of anthracene,  $\alpha$ , $\alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene, and **6** in THF-d<sub>8</sub>.



**Figure S34**. <sup>1</sup>H DOSY NMR spectrum of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene, and **7** in THF-d<sub>8</sub>.



**Figure S35**. Plot of diffusion coefficient (log*D*) versus formula weight (log Fw) of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene and **6** in THF-d<sub>8</sub>.



**Figure S36**. Plot of diffusion coefficient (log*D*) versus formula weight (log Fw) of a mixture of anthracene,  $\alpha, \alpha$ '-dibromo-o-xylene, 1,2,4,5-tetrakis(bromomethyl)benzene and **7** in THF-d<sub>8</sub>.