

Supporting Information

In-site hydrogen bubble template method to prepare Ni coated metal meshes as effective bi-functional electrodes for water splitting

Yucheng Dong,^a Shan Ji^{*b}, Hui Wang,^a Vladimir Linkov^c, Rongfang Wang,^{*a}

a. State Key Laboratory Base for Eco-Chemical Engineering, College of Chemical Engineering, Qingdao University of Science and Technology, Qingdao, 266042, China.

b. College of Biological, Chemical Science and Engineering, Jiaying University, Jiaying, 314001, China.

c. South African Institute for Advanced Material Chemistry, University of the Western Cape, Cape Town, 7535, South Africa.

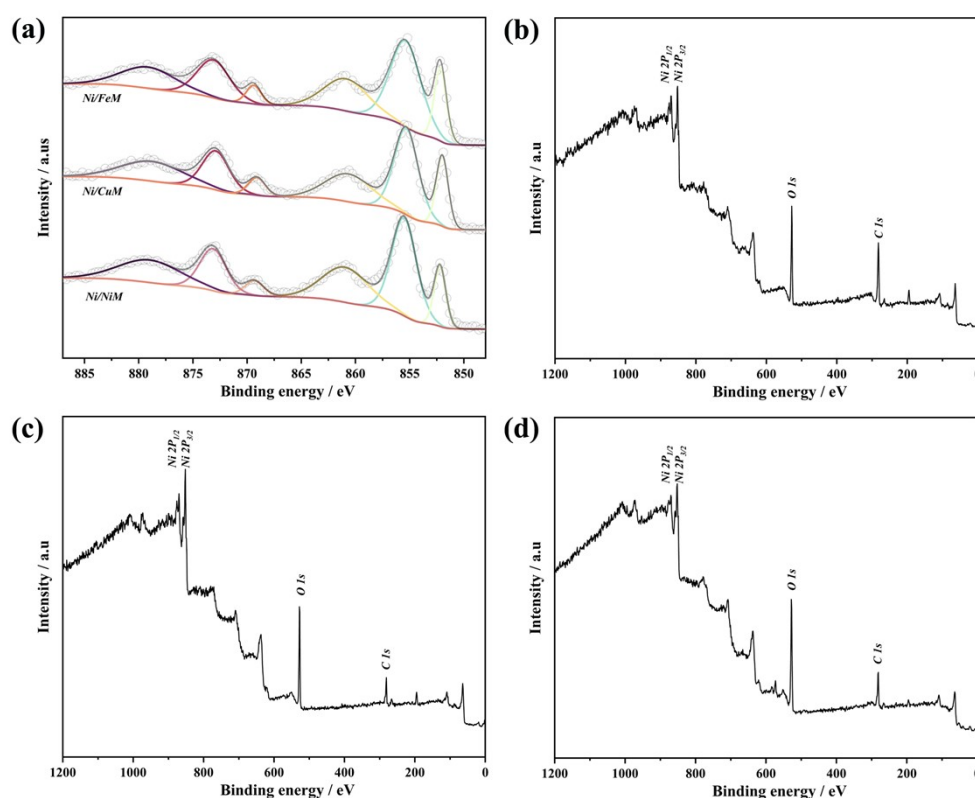


Figure S1. (a) The high resolution XPS spectra of Ni 2p of different materials; XPS survey spectra of (a) Ni/CuM, (b) Ni/NiM, (c) Ni/FeM.

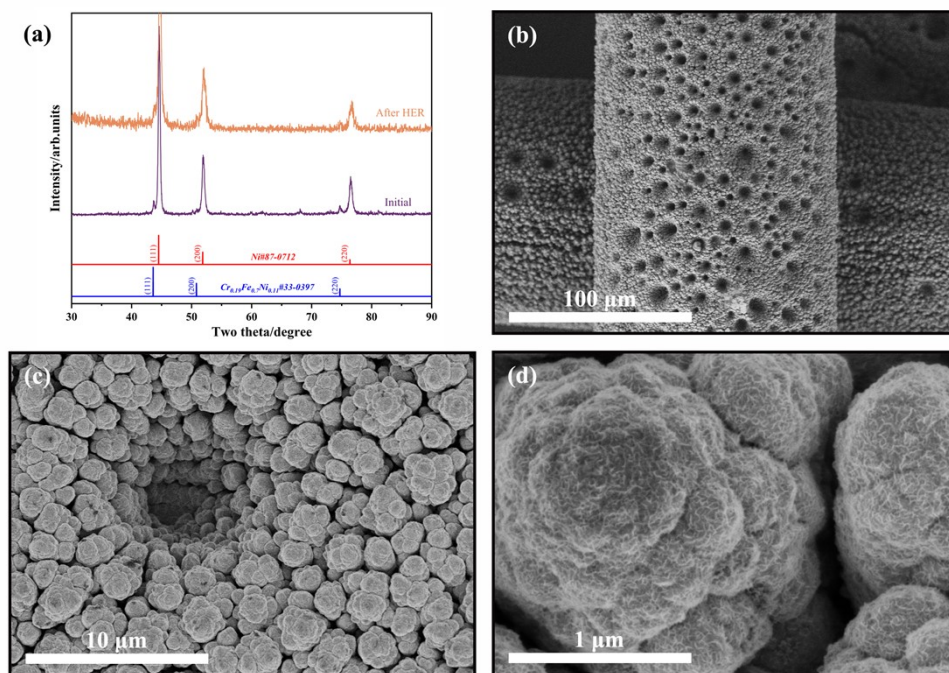


Figure S2. (a) XRD and (b-d) SEM images of after HER stability tests of Ni/FeM at 50 mA cm⁻² for 10 h.

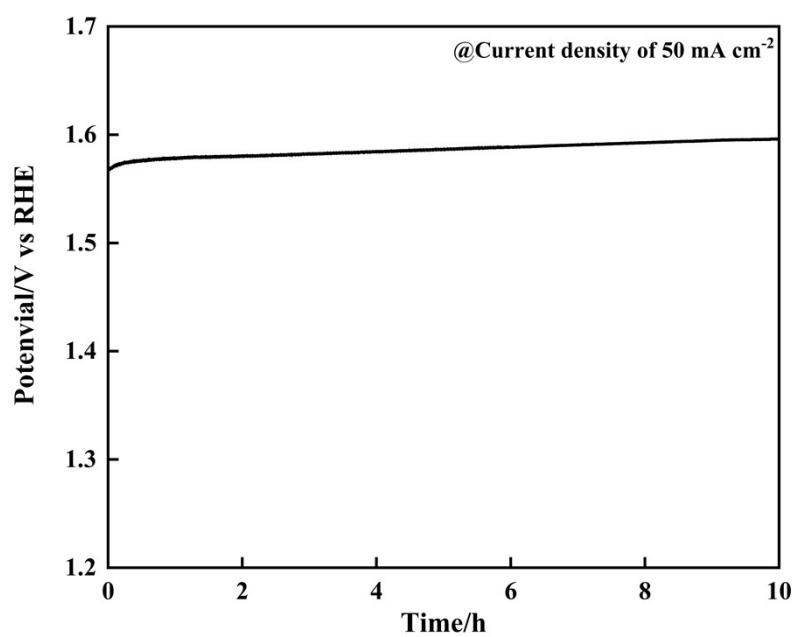


Figure S3. Long-term chronopotentiometry of Ni/FeM at 50 mA cm⁻² for 10 h.

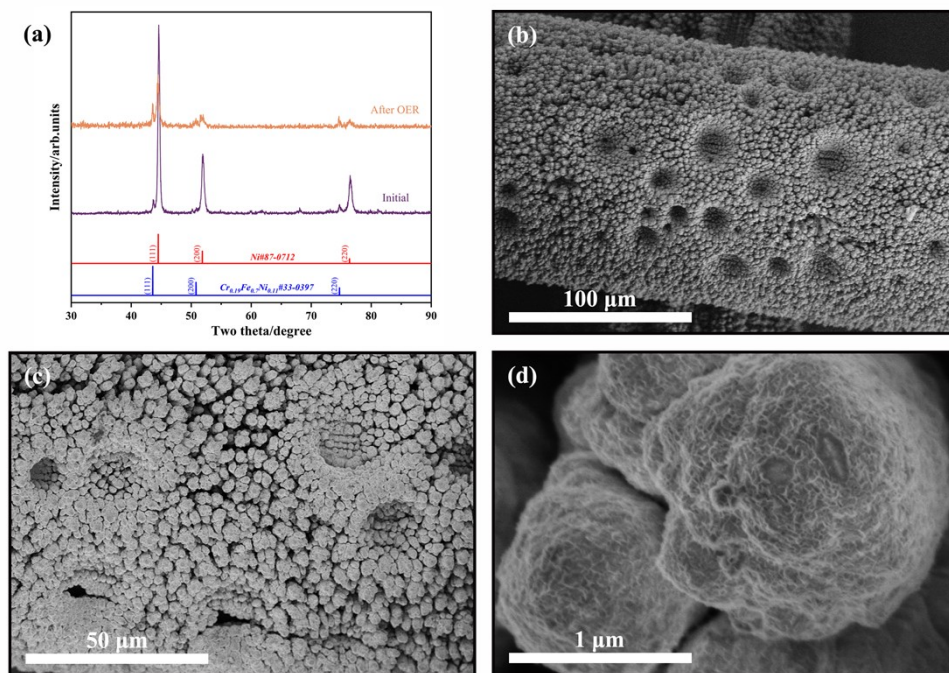


Figure S4. (a) XRD and (b-d) SEM images of after OER stability tests of Ni/FeM at 50 mA cm^{-2} for 10 h.

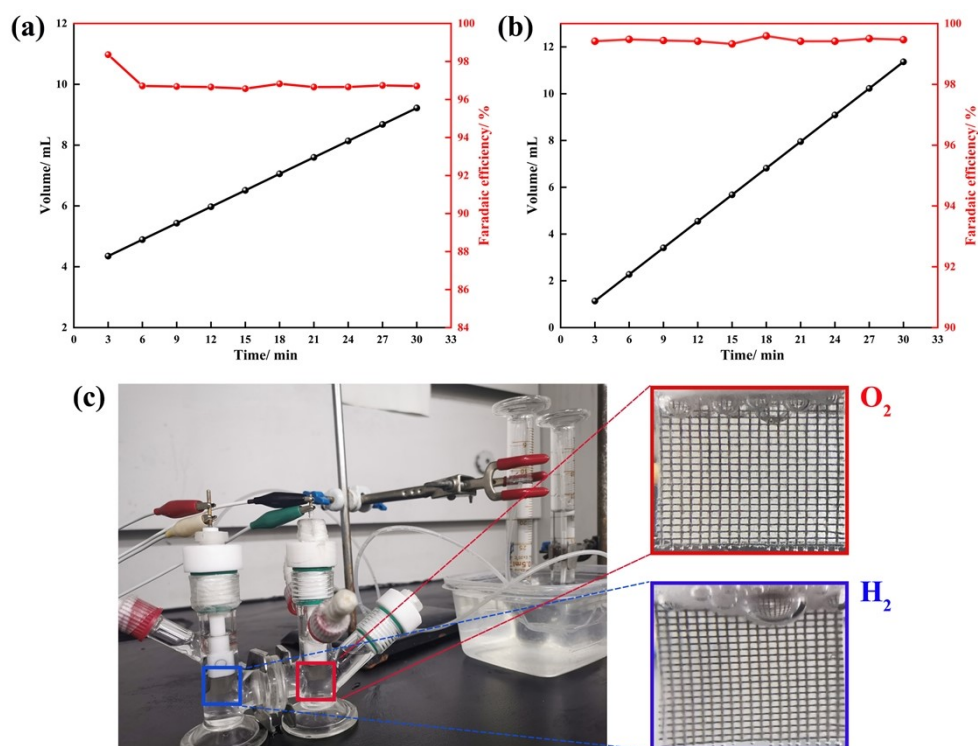


Figure S5. Gas volume versus time and corresponding Faradaic efficiency (a) O_2 , (b) H_2 ; (c) Two-electrode configuration for water splitting and gases generated on the cathode.

Table S1. Comparison of alkaline HER activities of different Ni-Based electrocatalysts.

HER catalysts	Electrolytes	η_{HER} (mV)	Ref.
Ni/FeM	1M KOH	38 (10 mA/cm ²) 113 (100 mA/cm ²)	This work
Ru-NiO/Co ₃ O ₄	1M KOH	138 mV (100 mA/cm ²)	[1] ¹
Ni-Mo ₂ C-0.67	1M KOH	151 (10 mA/cm ²) 194 (100 mA/cm ²)	[2] ²
Ni NWs/Ni/CM	1M KOH	160 (100 mA/cm ²)	[3] ³
NiCo ₂ O ₄ /NF	1M KOH	53 (10 mA/cm ²)	[4] ⁴
FeNi ₃ @FeNi LDH/NF	1M KOH	106 (10 mA/cm ²)	[5] ⁵
Ni-Ni(OH) ₂	1M KOH	72 (10 mA/cm ²) 233 (100 mA/cm ²)	[6] ⁶
Ni-GF/VC	1M KOH	128 (10 mA/cm ²)	[7] ⁷
WN-Ni(OH) ₂	1M KOH	245 (100 mA/cm ²)	[8] ⁸
Ni-Fe-Se/NF	1M KOH	85 (10 mA/cm ²)	[9] ⁹
Mo-Ni-NSs-3	1M KOH	91 (10 mA/cm ²)	[10] ¹⁰

Table S2. Comparison of alkaline OER activities of different electrocatalysts.

OER catalysts	Electrolytes	η_{OER} (mV)	Ref.
Ni/FeM	1M KOH	220 (10 mA/cm ²)	This work
CNCP	1M KOH	224 (10 mA/cm ²)	[1] ¹¹
1T-MoS ₂	1M KOH	224 (10 mA/cm ²)	[2] ¹²
NiCo-H ₂ /NF	1M KOH	331 (10 mA/cm ²)	[3] ¹³

CoNiP _x -CNF	1M KOH	269 (10 mA/cm ⁻²)	[4] ¹⁴
RuO ₂ -Fe ₂ O ₃	1M KOH	292 (10 mA/cm ⁻²)	[5] ¹⁵
Ni/MoO ₂	1M KOH	287 (10 mA/cm ⁻²)	[6] ¹⁶
FeNiO _x (OH) _y -2	1M KOH	250 (10 mA/cm ⁻²)	[7] ¹⁷
Ni SAs/Fe-NiOOH	1M KOH	269 (10 mA/cm ⁻²)	[8] ¹⁸
NiFe-MOF/IF	1M KOH	230 (10 mA/cm ⁻²)	[9] ¹⁹
Ir-Ni(OH) ₂	1M KOH	248 (10 mA/cm ⁻²)	[10] ²⁰

Reference:

1. J. Zhang, J. Lian, Q. Jiang and G. Wang, *Chemical Engineering Journal*, 2022, **439**, 135634.
2. W. Liu, X. Wang, J. Qu, X. Liu, Z. Zhang, Y. Guo, H. Yin and D. Wang, *Applied Catalysis B: Environmental*, 2022, **307**, 121201.
3. D. Gao, S. Ji, H. Wang, X. Wang, V. Linkov and R. Wang, *Dalton Transactions*, 2022, **51**, 5309-5314.
4. Y. Yu, C. Chen, Y. Liu, H. Yu, S. Li, Y. Xue, N. Cai, J. Wang and F. Yu, *The Journal of Physical Chemistry C*, 2022, **126**, 5493-5501.
5. Z. Liu, S. Li, F. Wang, M. Li and Y. Ni, *Dalton Transactions*, 2021, **50**, 6306-6314.
6. W. Zhong, W. Li, C. Yang, J. Wu, R. Zhao, M. Idrees, H. Xiang, Q. Zhang and X. Li, *Journal of Energy Chemistry*, 2021, **61**, 236-242.
7. C. Yang, R. Zhao, H. Xiang, J. Wu, W. Zhong, W. Li, Q. Zhang, N. Yang and X. Li, *Advanced Energy Materials*, 2020, **10**, 2002260.
8. C. Lv, X. Wang, L. Gao, A. Wang, S. Wang, R. Wang, X. Ning, Y. Li, D. W. Boukhvalov, Z. Huang and C. Zhang, *ACS Catalysis*, 2020, **10**, 13323-13333.
9. X. Yi, X. He, F. Yin, B. Chen, G. Li and H. Yin, *Dalton Transactions*, 2020, **49**, 6764-6775.
10. X. Lu, M. Cai, J. Huang and C. Xu, *Journal of Colloid and Interface Science*, 2020, **562**, 307-312.
11. H. Singh, R. Biswas, I. Ahmed, P. Thakur, A. Kundu, A. R. Panigrahi, B. Banerjee, K. K. Halder, J. Lahtinen, K. Mondal and K. K. Halder, *ACS Applied Materials & Interfaces*, 2022, **14**, 6570-6581.
12. G. Wang, G. Zhang, X. Ke, X. Chen, X. Chen, Y. Wang, G. Huang, J. Dong, S. Chu and M. Sui, *Small*, 2022, **18**, 2107238.
13. M. Chen, D. Liu, J. Feng, P. Zhou, L. Qiao, W. Feng, Y. Chen, K. Wei Ng, S. Wang, W. Fai Ip and H. Pan, *Chemical Engineering Journal*, 2022, **443**, 136432.
14. J. Chen, F. Huang, S. Ke, J. Shen, Y. Li, F. Zheng and S. Li, *Dalton Transactions*, 2022, **51**, 5168-5174.
15. H. Mosallaei, H. Hadadzadeh, A. Foelske, M. Sauer, H. Amiri Rudbari and O. Blacque, *Dalton Transactions*, 2022, **51**, 6314-6331.
16. S. Ni, H. Qu, H. Xing, Z. Xu, X. Zhu, M. Yuan, L. Wang, J. Yu, Y. Li, L. Yang and H. Liu, *ACS Applied Materials & Interfaces*, 2021, **13**, 17501-17510.
17. H. Liao, P. Tan, R. Dong, M. Jiang, X. Hu, L. Lu, Y. Wang, H. Liu, Y. Liu and J. Pan, *Journal of Colloid and Interface Science*, 2021, **591**, 307-313.
18. H. Liu, X. Xu, H. Xu, S. Wang, Z. Niu, Q. Jia, L. Yang, R. Cao, L. Zheng and D. Cao, *Applied Catalysis B: Environmental*, 2021, **297**, 120451.
19. Y. Zhang, J. Wang, L. Ye, M. Zhang and Y. Gong, *Dalton Transactions*, 2021, **50**, 4720-4726.
20. R. Guo, W. Shi, Wenzhu Liu, X. Yang, Y. Xie, T. Yang and J. Xiao, *Chemical Engineering Journal*, 2022, **429**, 132478.