

Supplementary Material (ESI)

**To Chelate Thallium(I) - Synthesis and Evaluation of Kryptofix-
based Chelators for ^{201}Tl .**

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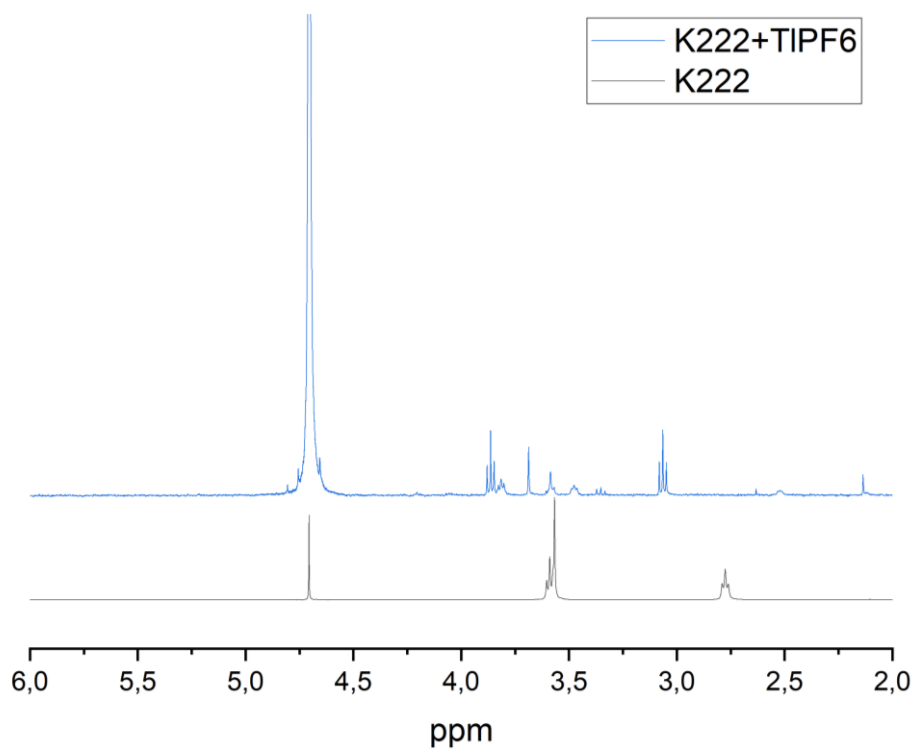


Figure S1. ¹H NMR spectrum of Kryptofix 222 before (black) and after (blue) addition of TIPF₆ in D₂O (400 MHz, 300 K)

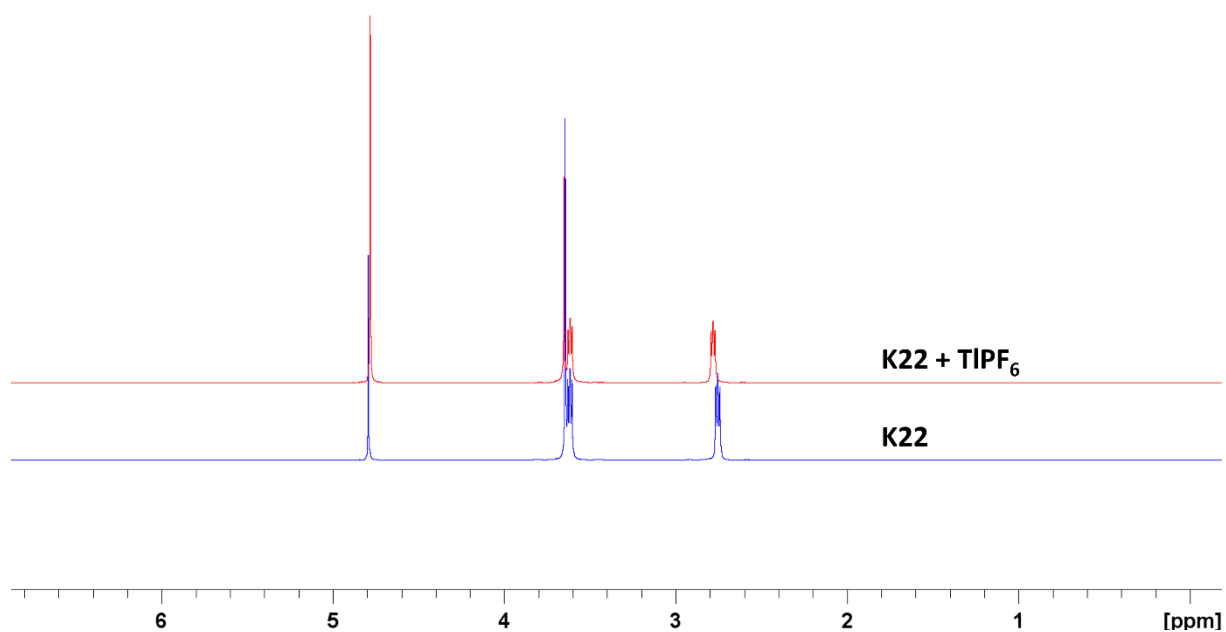


Figure S2. ¹H NMR spectrum of Kryptofix 22 before (blue) and after (red) addition of TIPF₆ in D₂O (400 MHz, 300 K).

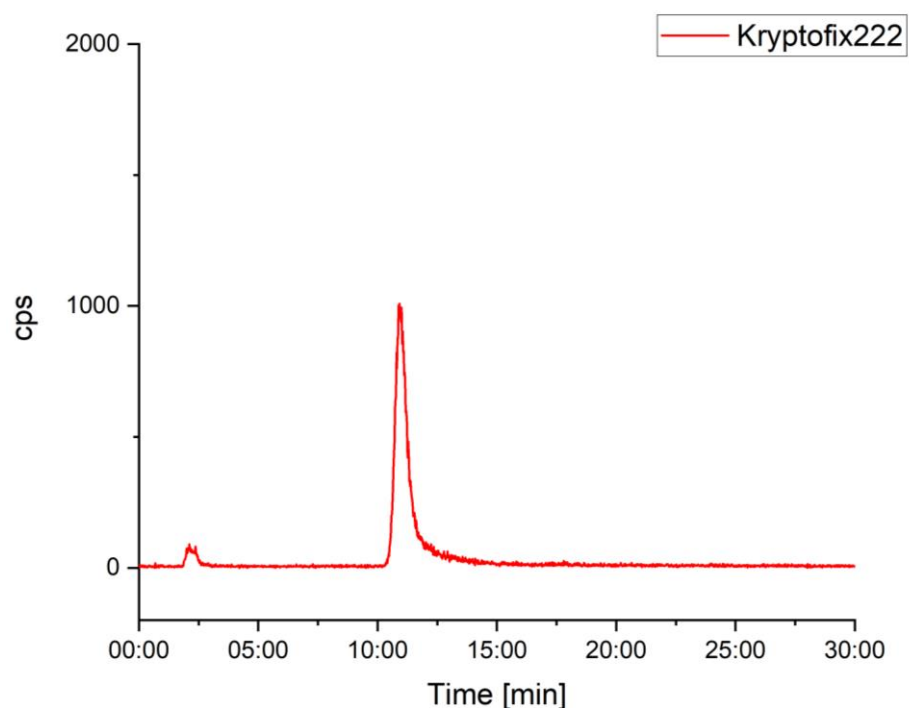


Figure S3. Radio-HPLC γ -trace after reaction of K222 with $[^{201}\text{Tl}]\text{Ti}(\text{I})\text{Cl}$.

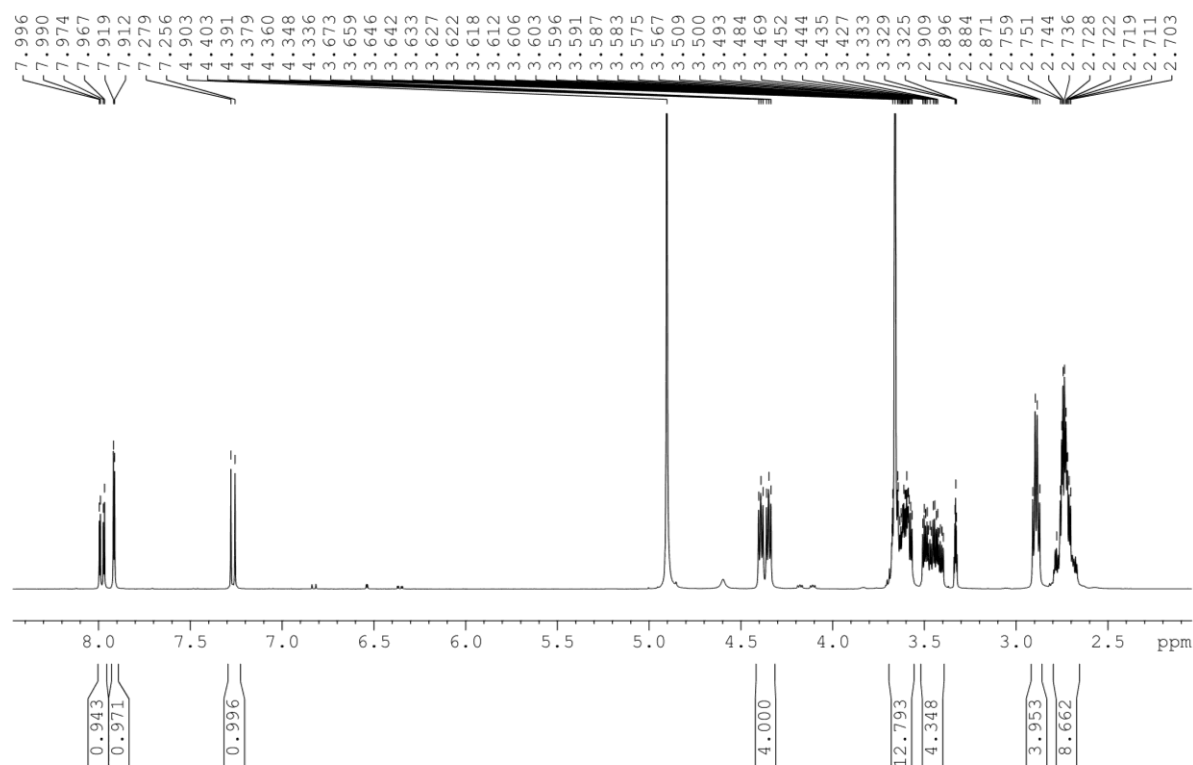


Figure S4. ^1H NMR spectrum of compound **5** in CD_3OD (400 MHz, 300 K)

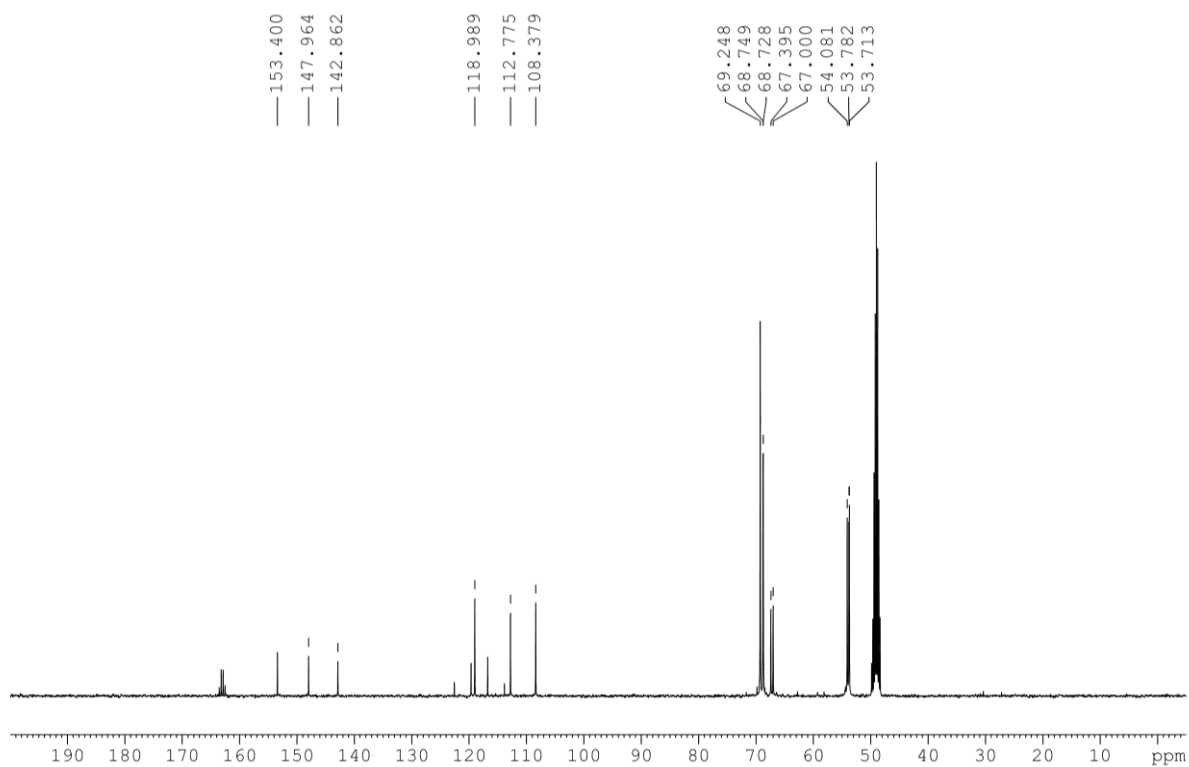


Figure S5. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound **5** in CD_3OD (101 MHz, 300 K).

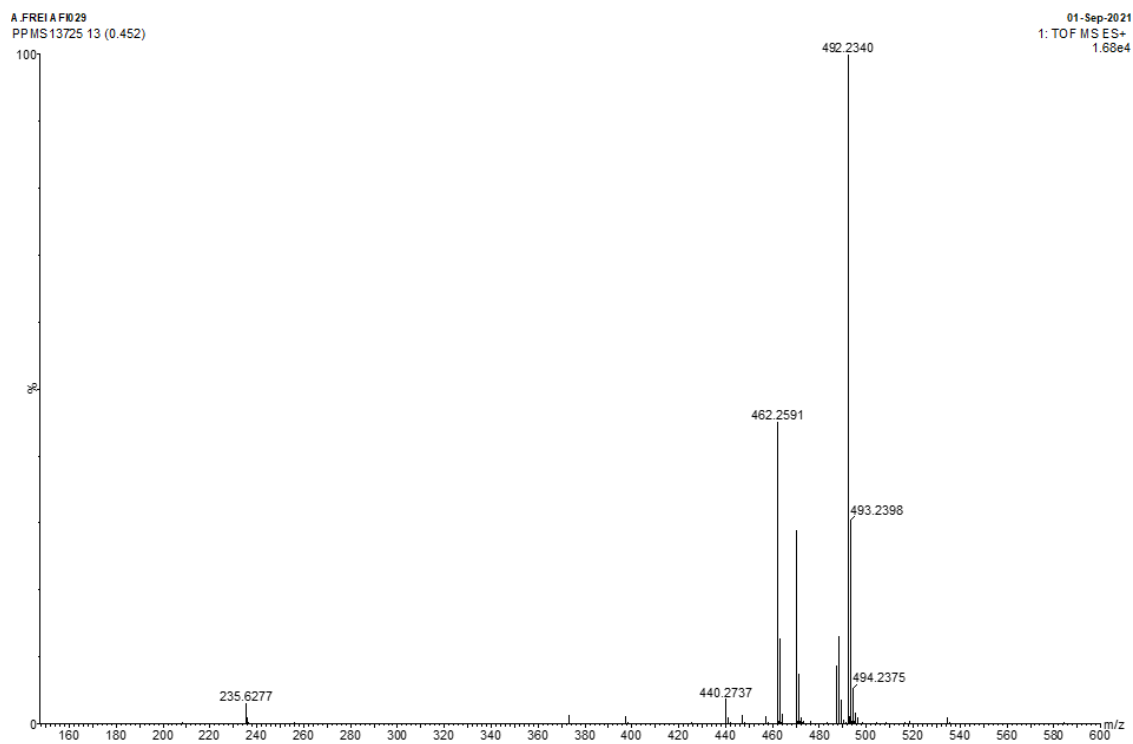


Figure S6. HR-MS spectrum of compound **5**

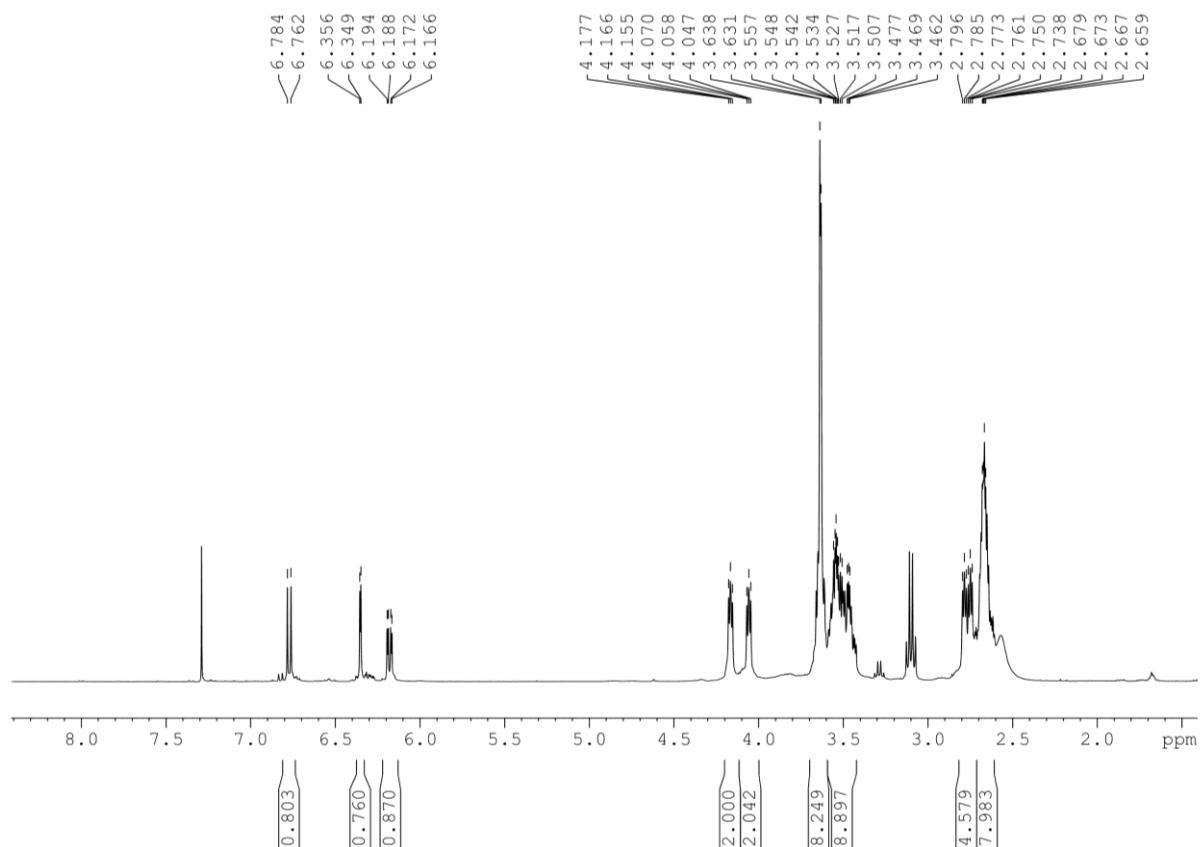


Figure S7. ^1H NMR spectrum of **K222-NH₂** in CDCl_3 (400 MHz, 300 K).

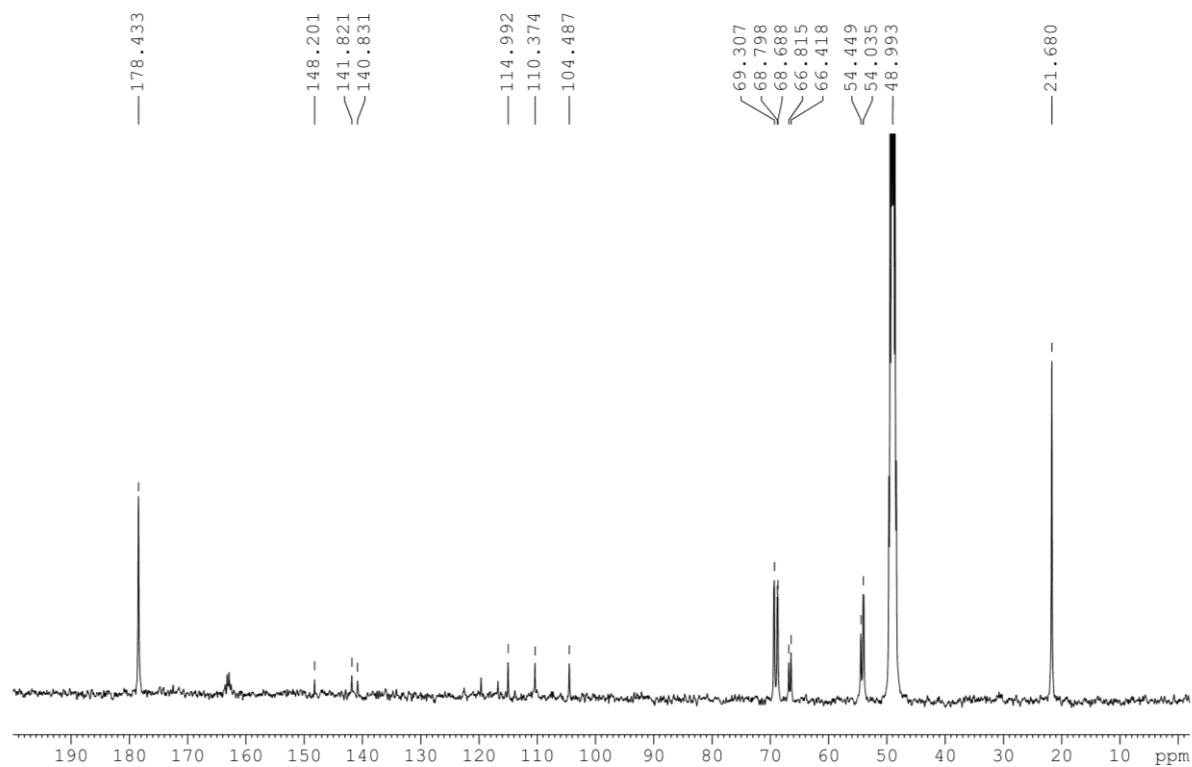


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **K222-NH₂** in CD_3OD (101 MHz, 300 K).

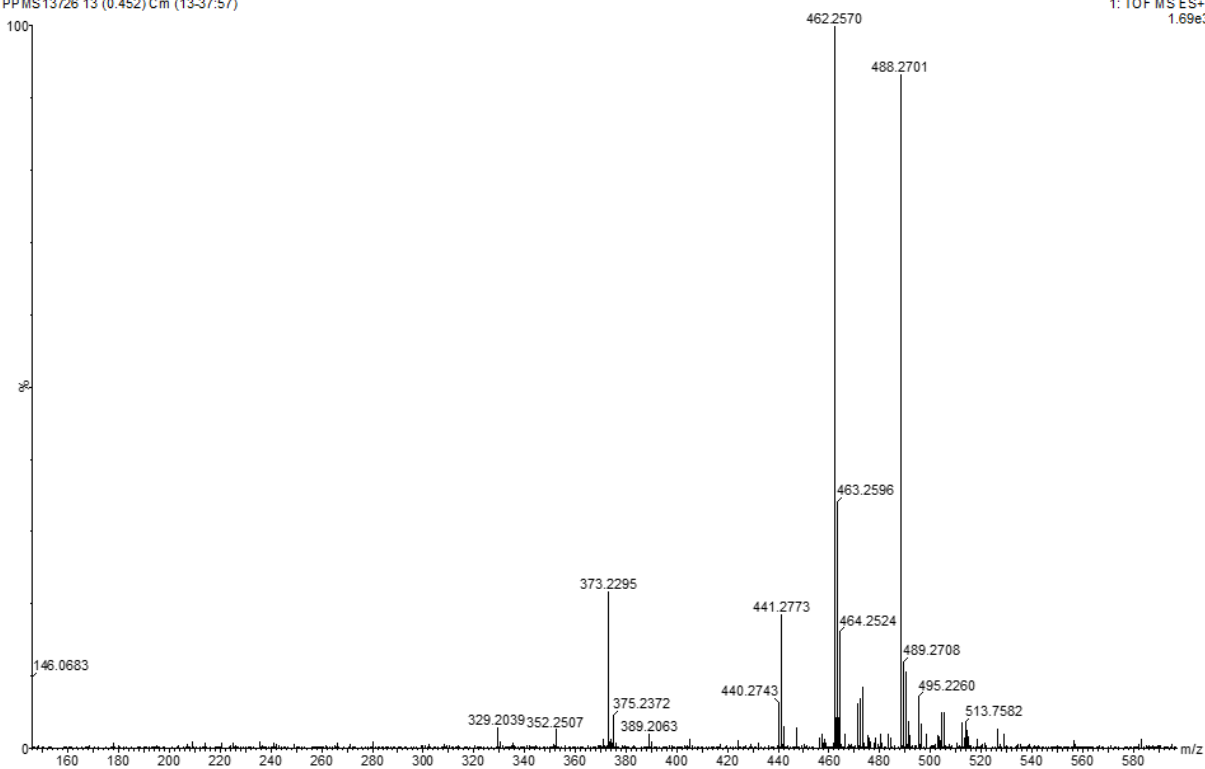


Figure S9. HR-MS spectrum of K222-NH₂

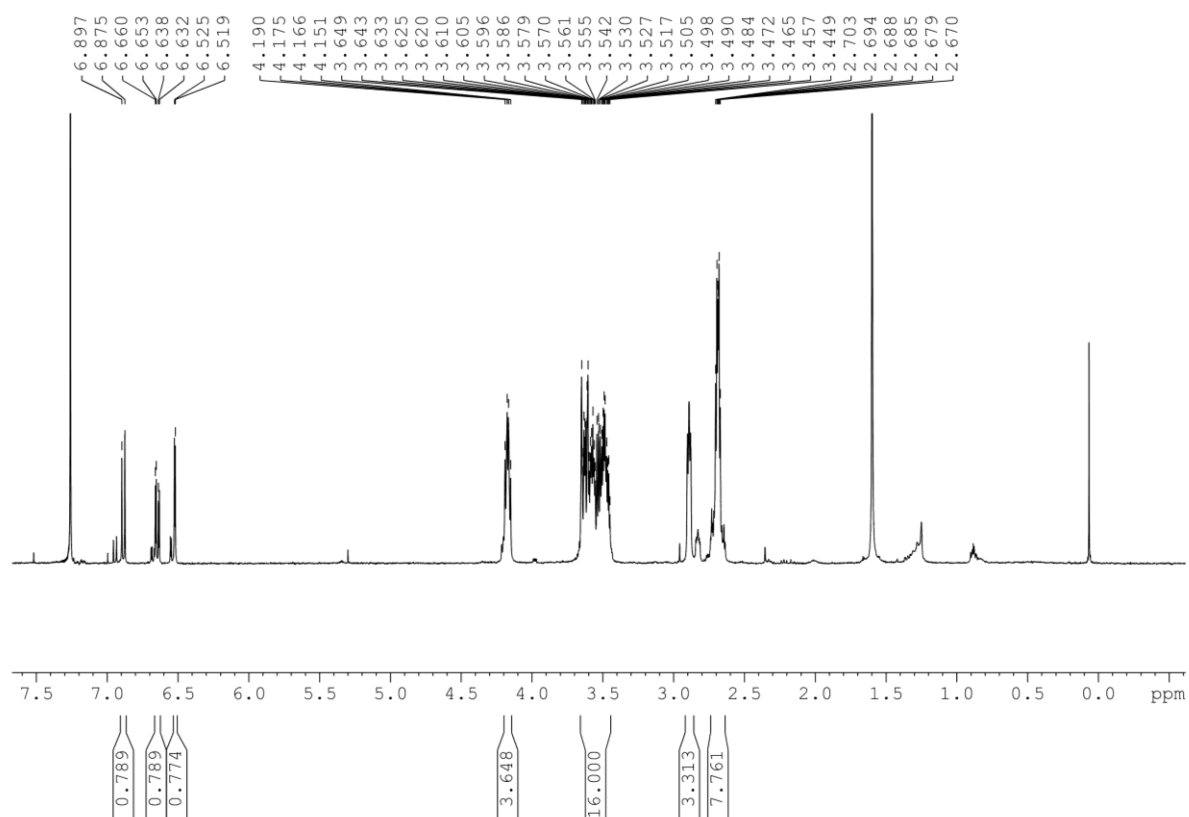


Figure S10. ¹H NMR spectrum of K222-N₃ in CDCl₃ (400 MHz, 300 K)

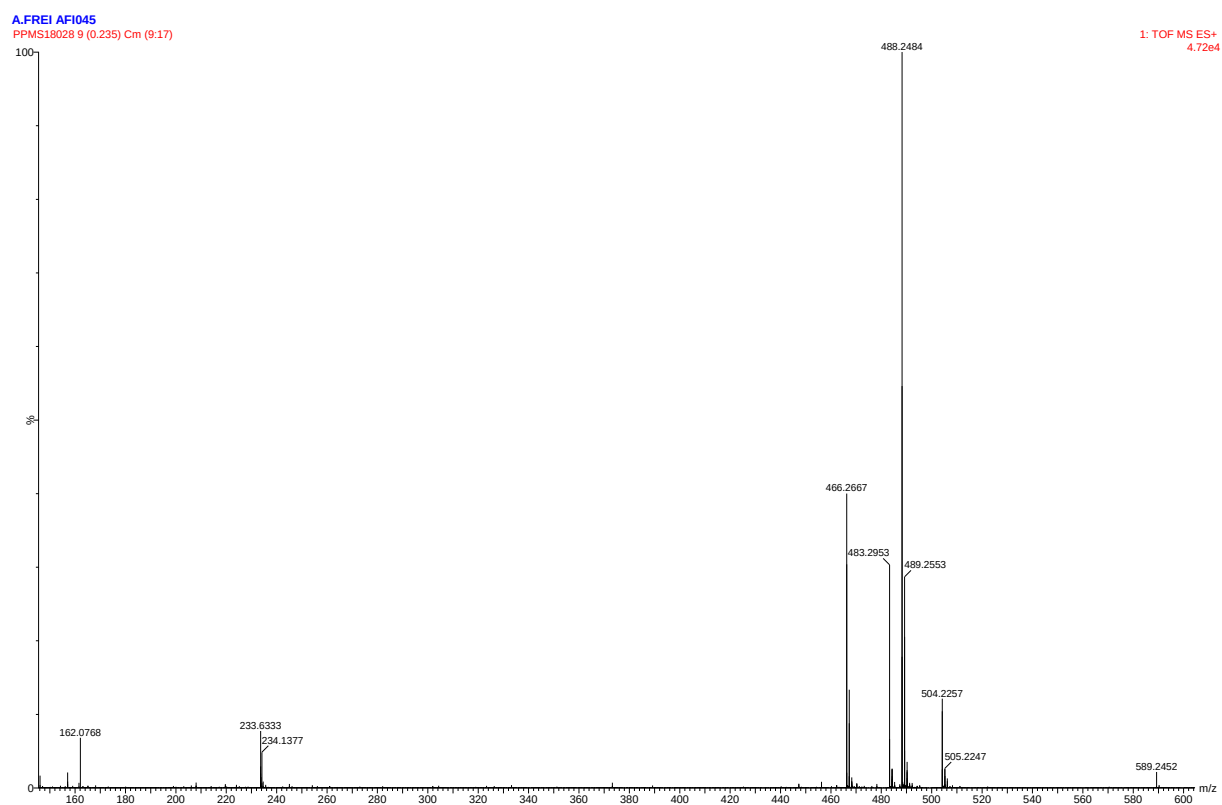


Figure S11. HR-MS spectrum of **K222-N₃**

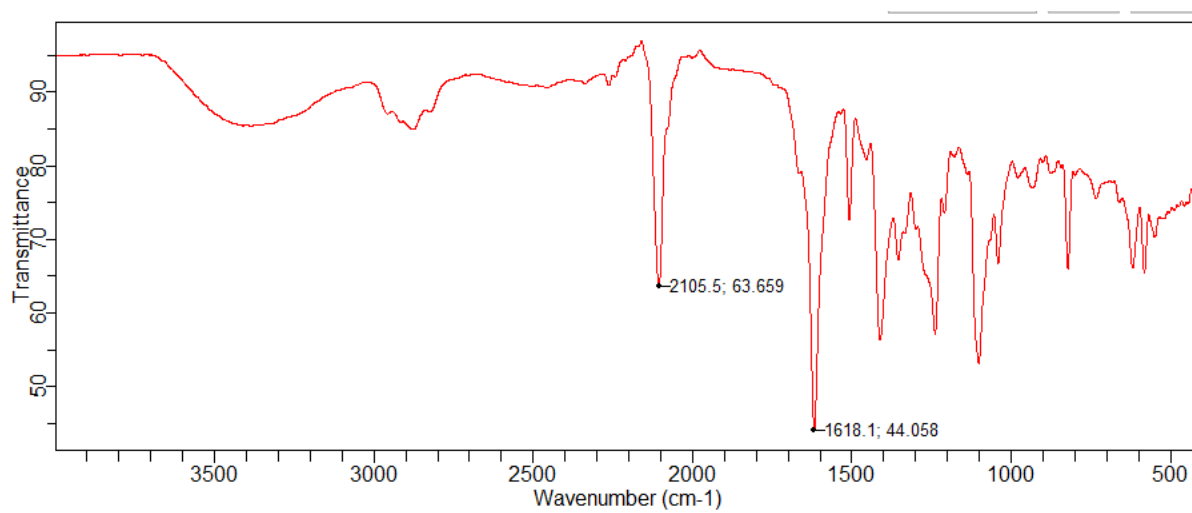


Figure S12. IR-spectrum of **K222-N₃**.

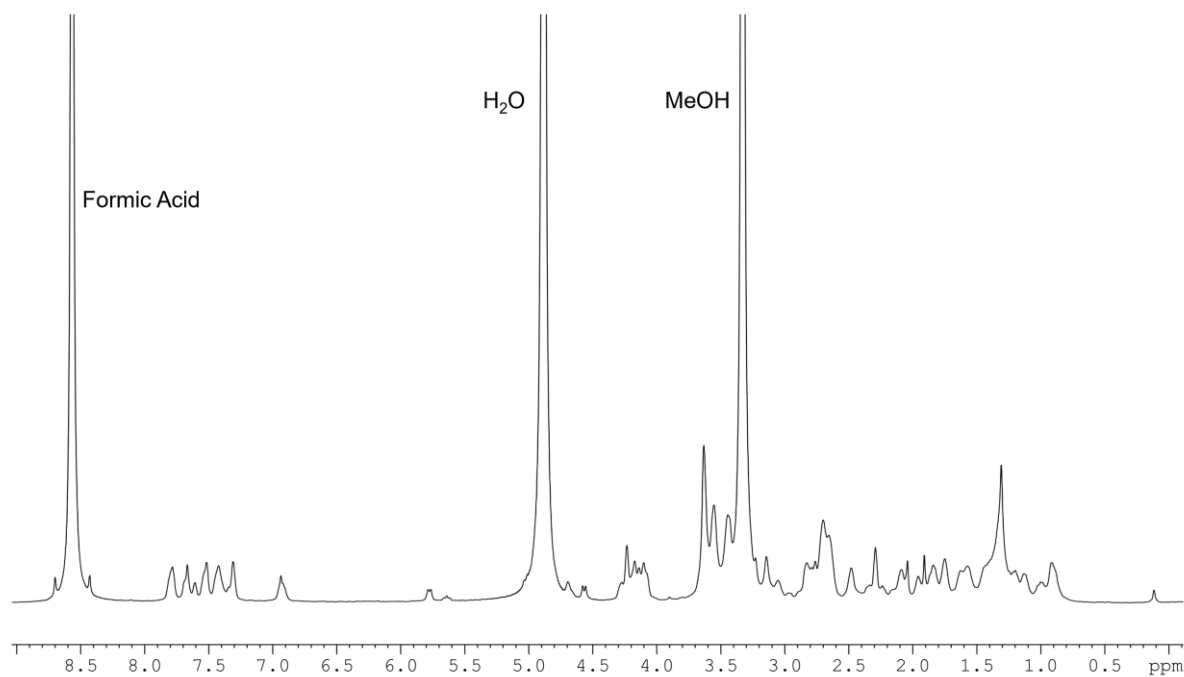


Figure S13. ¹H NMR spectrum of K222-PSMA in MeOD (400 MHz, 300 K)

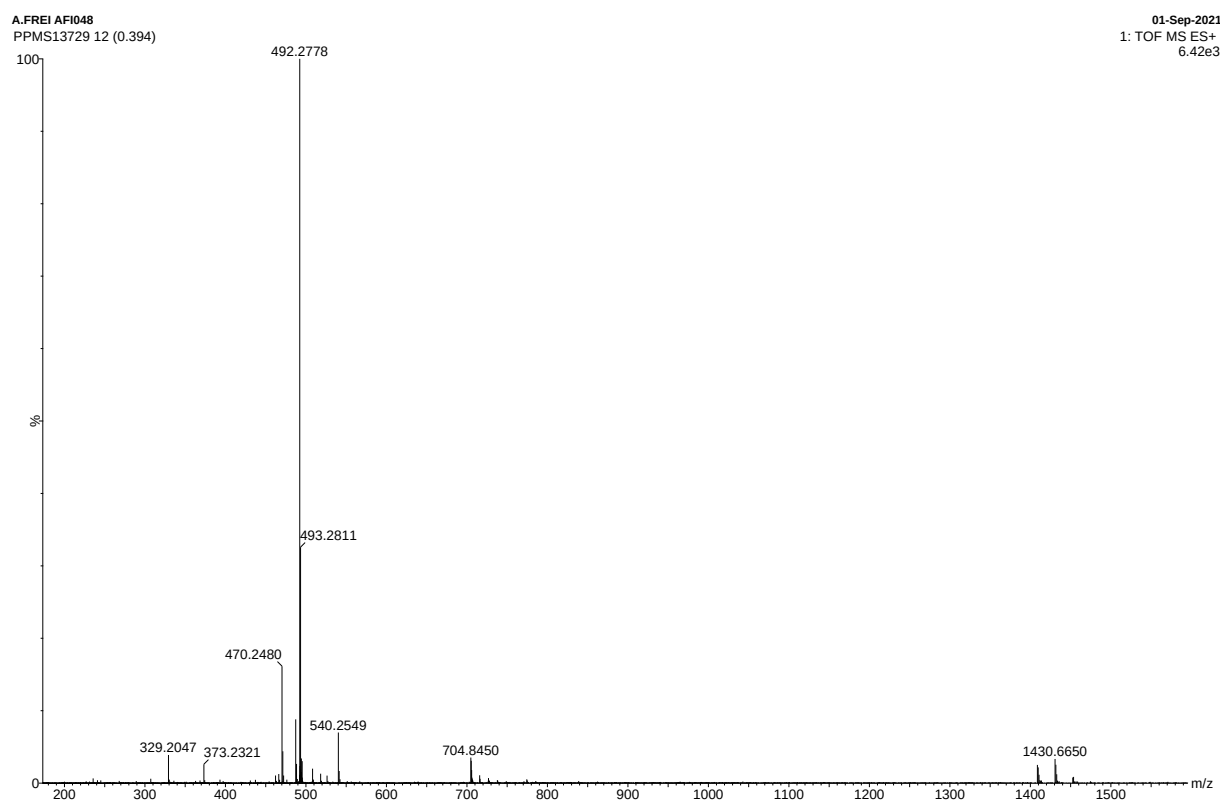


Figure S14. HR-MS spectrum of K222-PSMA (m/z = 1407.6751)

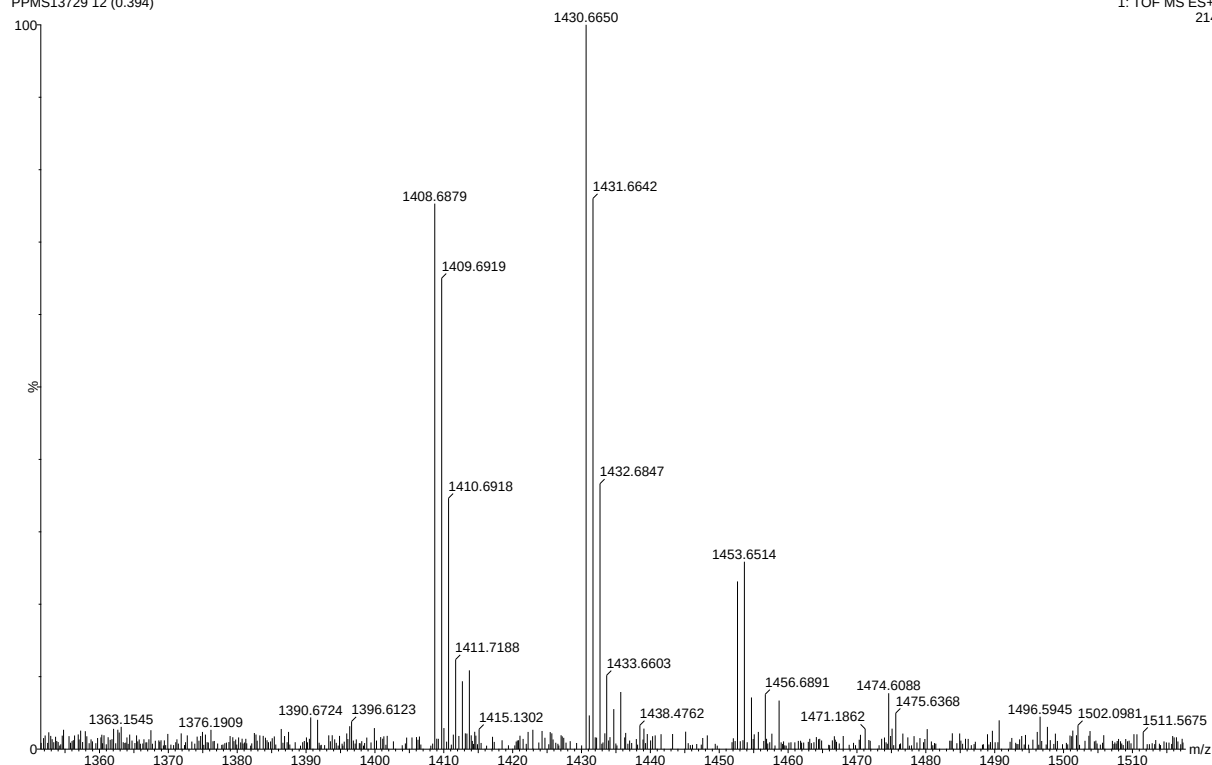


Figure S15. Detailed HR-MS spectrum of K222-PSMA ($m/z = 1407.6751$)

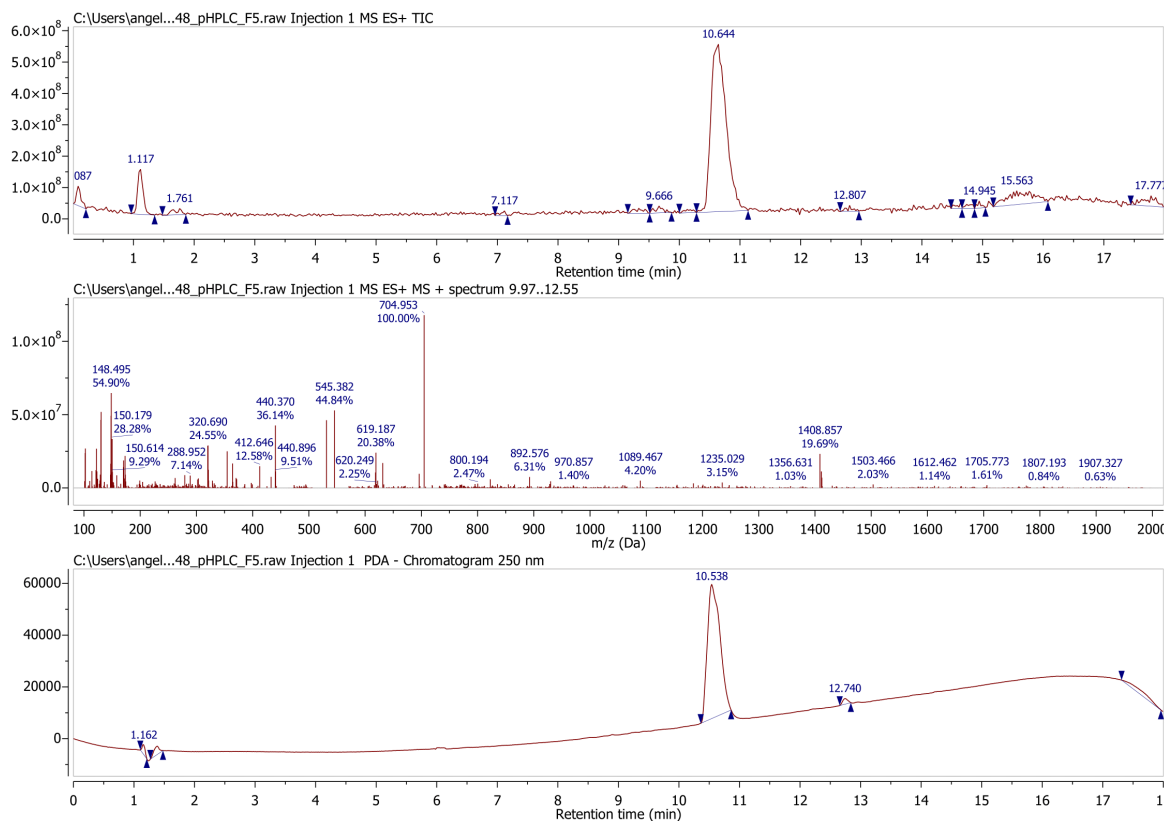


Figure S16. LC-MS trace of K222-PSMA after preparative HPLC purification.

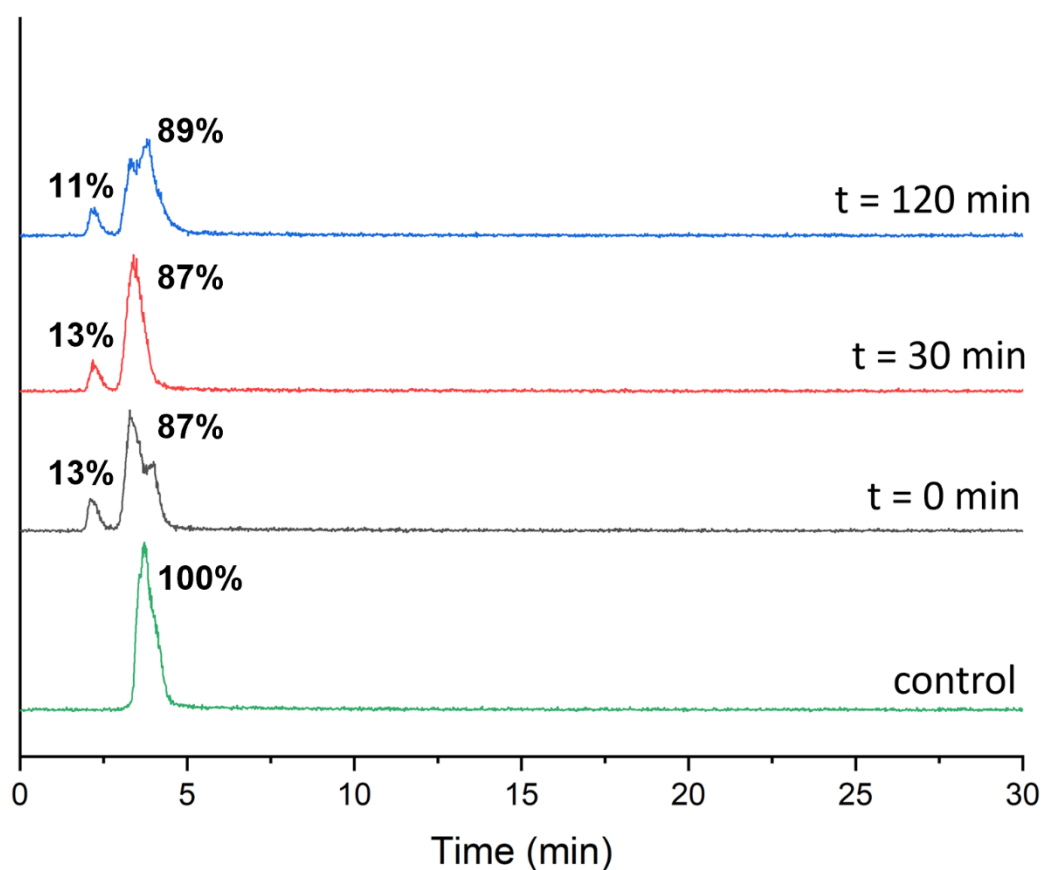


Figure S17. Stacked plot of the serum stability study of **K222-PSMA** in human serum. Radio-HPLC γ -traces are shown for three different time points and the control. Relative integrals of the first peak and the second peak are given in %.

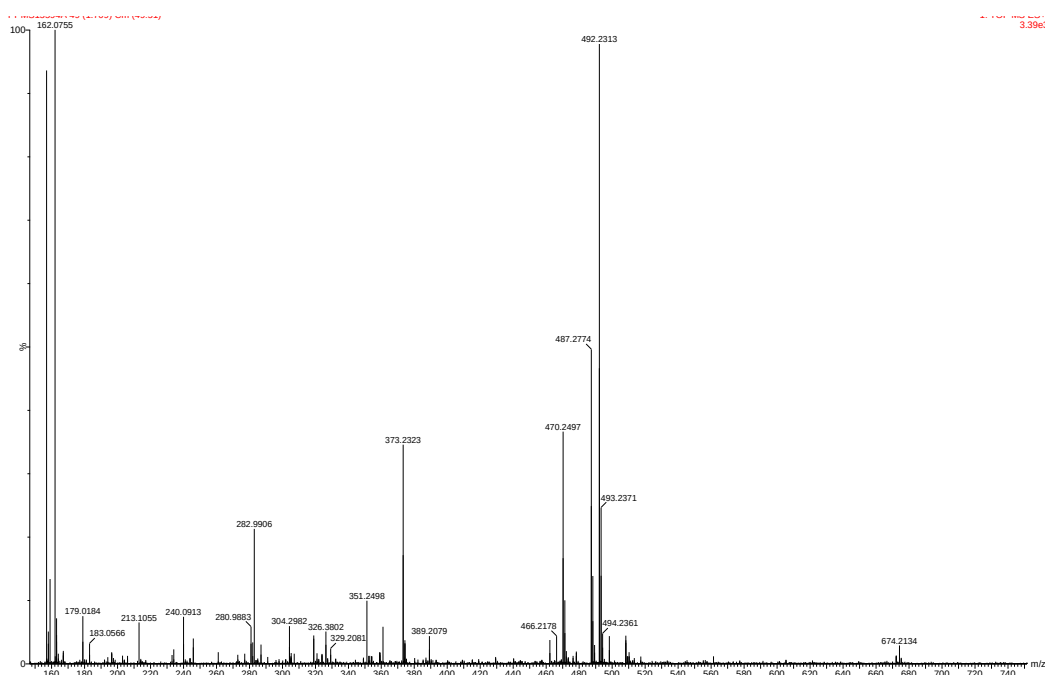


Figure S18. HR-ESI-MS spectrum of $[TI(5)]^+$

	AR	stdev
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ (positive, 1 h)	4.70%	0.23%
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ (negative, 1 h)	4.93%	0.20%
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ (positive, 10 min)	2.75%	0.15%
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ (negative, 10 min)	2.88%	0.13%
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ + PMPA (positive, 1 h)	4.97%	0.13%
$[^{201}\text{Tl}][\text{Ti}(\text{K222-PSMA})]^+$ + PMPA (negative, 1 h)	5.18%	0.08%
$[^{201}\text{Tl}]\text{TiCl}$ (positive, 1 h)	6.09%	0.90%
$[^{201}\text{Tl}]\text{TiCl}$ (negative, 1 h)	6.34%	0.14%

Table S1. Average (of three experiments) percentage added radioactivity measured for each experiment and calculated standard deviation.

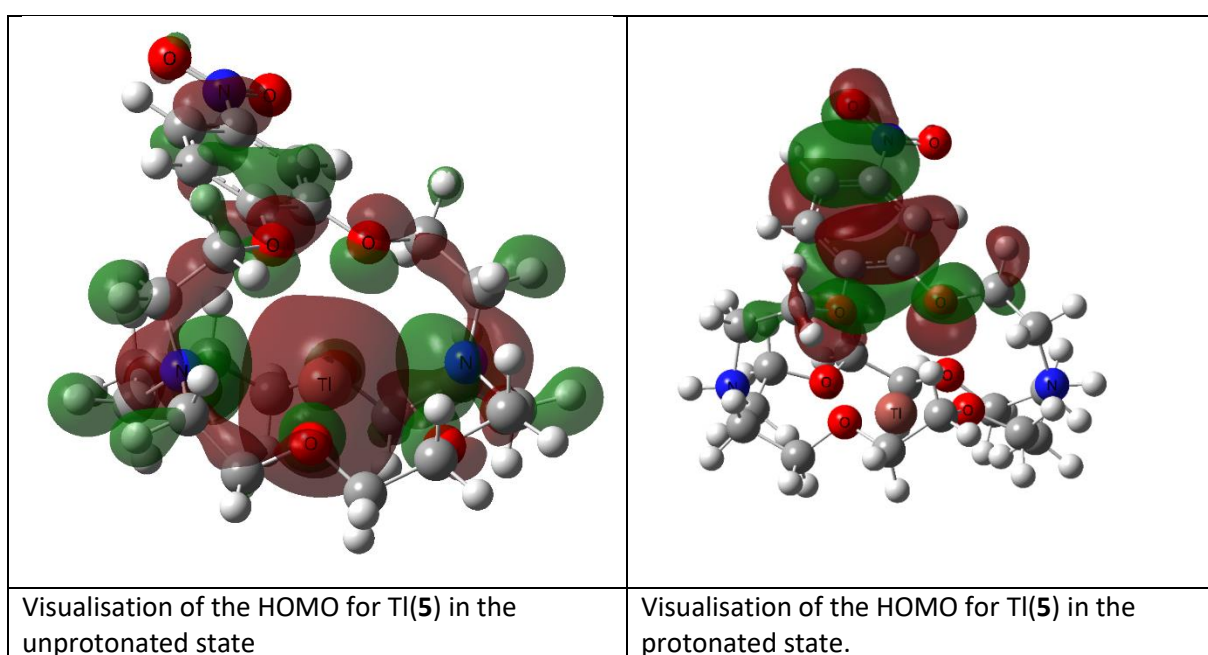


Figure S19. Comparison of the HOMO and orbital overlaps in the protonated and unprotonated K22 macrocycle

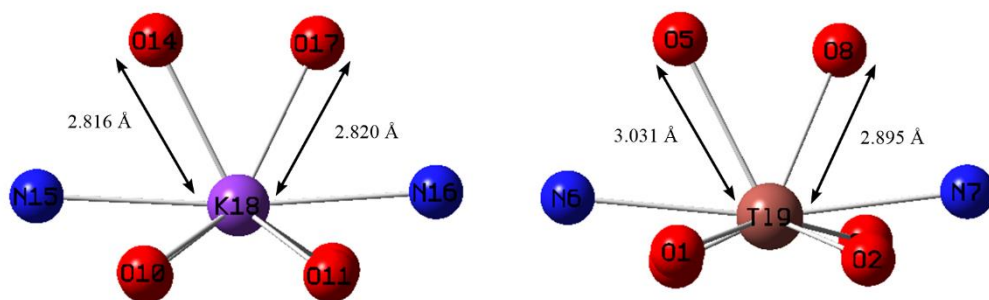


Figure S20. Visualisation of the calculated coordination environment for Ti(5)⁺ and K(5)⁺. Oxygens O14, O17, O5 and O8 are the ones belonging to the aromatic ring.

M-Crypt Complexes	Bond Lengths (Å) [σ]		
	O-M Aliphatic	O-M Aromatic	N-M
K(K222) (calcd.)	2.83 [0.023]	N/A	3.02 [0.001]
K(K222) (exp.) ^[1]	2.85 [0.025]	N/A	3.03 [0.005]
K(5)	2.79 [0.016]	2.82 [0.0015]	3.05 [0.03]
Tl(5)	2.86 [0.041]	2.96 [0.068]	3.05 [0.068]

Table S2. Comparison of coordination bond lengths between the experimental K(**K222**)⁺ structure and calculated K(**K222**)⁺, Tl(**5**)⁺ and K(**5**)⁺

References

- [1] R. Alberto, K. Ortner, N. Wheatley, R. Schibli, A. P. Schubiger, *Journal of the American Chemical Society* **2001**, 123, 3135-3136.