

Chemical Bonding in Actinyl (V/VI) Dipyriamethyrin Complexes for the Actinide Series from Americium to Californium: A Computational Investigation

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Table S1. Bond parameters computed at PBE/ZORA-TZP level of theory (gas phase). Bond lengths (Å), bond angles (°), bond orders (B.O), charges, spin densities on the actinide and O_{yl} atoms (a.u.), and actinyl stretching vibrational frequencies (cm⁻¹) of the bare actinyl (V,VI) cations.

	An=O _{yl}	O=An=O	B.O.	Mayer B.O.	Q(An)	Q(O1/O2)	Spin on An	Spin on O _{yl}	ν_{symm} (AnO ₂ -L)	ν_{asymm} (AnO ₂ -L)
Am ^{VO} ₂ ⁺	1.745	180.0	2.059	2.105	1.648	-0.324	4.518	-0.259	845.0	963.1
Cm ^{VO} ₂ ⁺	1.767	180.0	1.718	2.005	1.816	-0.408	5.198	-0.099	808.8	907.6
Bk ^{VO} ₂ ⁺	1.775	180.0	1.674	1.930	1.972	-0.486	5.343	0.329	826.0	909.1
Cf ^{VO} ₂ ⁺	1.866	94.9	1.949	2.036	1.867	-0.433	4.966	1.017	696.8	598.7
Am ^{VI} O ₂ ²⁺	1.698	180.0	2.210	2.198	2.242	-0.121	3.452	-0.226	915.3	1053.0
Cm ^{VI} O ₂ ²⁺	1.712	180.0	2.077	2.102	1.983	0.008	4.719	-0.359	830.2	986.6
Bk ^{VI} O ₂ ²⁺	1.744	179.9	1.818	1.952	2.161	-0.080	5.033	-0.016	795.0	903.5
Cf ^{VI} O ₂ ²⁺	1.795	158.4	1.758	1.936	2.245	-0.123	4.565	0.717	697.1	706.5

Table S2. Bond parameters (i.e., bond lengths (Å) and bond angles (°)) computed at the PBE/TZP level of theory in the gas phase for the actinyl (V and VI) complexes

	An-N1	An-N2	An-N3	An-N4	An-N5	An-N6	An=O _{yl} (1)	An=O _{yl} (2)
Bond Lengths								
Am ^V O2-L ⁻	2.897	2.801	2.808	2.897	2.801	2.808	1.798	1.799
Cm ^V O2-L ⁻	2.872	2.785	2.808	2.892	2.800	2.801	1.821	1.824
Bk ^V O2-L ⁻	2.864	2.779	2.783	2.861	2.767	2.774	1.828	1.831
Cf ^V O2-L ⁻	2.802	2.691	2.698	2.831	2.780	2.802	1.817	1.820
Am ^{VI} O2-L	2.758	2.656	2.658	2.758	2.656	2.658	1.761	1.763
Cm ^{VI} O2-L	2.798	2.764	2.786	2.816	2.695	2.692	1.778	1.783
Bk ^{VI} O2-L	2.817	2.741	2.739	2.811	2.727	2.740	1.814	1.816
Cf ^{VI} O2-L	2.811	2.723	2.733	2.811	2.731	2.740	1.812	1.815
Bond Angles								
	O _{yl} =An=O _{yl}	N1-An-N2	N2-An-N3	N3-An-N4	N4-An-N5	N5-An-N6	N6-An-N1	
Am ^V O2-L ⁻	180.0	57.9	65.9	57.6	57.9	65.9	57.6	
Cm ^V O2-L ⁻	179.9	58.3	65.7	57.6	57.9	65.6	57.8	
Bk ^V O2-L ⁻	180.0	58.1	65.5	57.9	58.3	65.8	58.0	
Cf ^V O2-L ⁻	179.5	59.1	66.4	58.8	57.8	64.5	58.4	
Am ^{VI} O2-L	180.0	59.2	65.6	59.0	59.2	65.6	59.0	
Cm ^{VI} O2-L	179.9	58.7	64.1	58.0	59.0	66.5	58.8	
Bk ^{VI} O2-L	179.9	58.5	65.3	58.3	58.7	65.5	58.2	
Cf ^{VI} O2-L	180.0	58.8	65.5	58.2	58.7	65.3	58.1	

Table S3. Calculated bond lengths (Å) and bond angles (deg) of the actinyl (VI and V) complexes at the PBE/TZP level of theory using ADF and ZORA in CH₂Cl₂ solvent

	An-N1	An-N2	An-N3	An-N4	An-N5	An-N6	An=O _{yl} (1)	An=O _{yl} (2)
Bond Lengths								
Am ^V O ₂ -L ⁻	2.883	2.777	2.782	2.883	2.777	2.782	1.817	1.818
Cm ^V O ₂ -L ⁻	2.859	2.758	2.772	2.874	2.769	2.767	1.838	1.838
Bk ^V O ₂ -L ⁻	2.837	2.745	2.747	2.836	2.736	2.741	1.843	1.845
Cf ^V O ₂ -L ⁻	2.795	2.686	2.699	2.814	2.789	2.778	1.838	1.839
Am ^{VI} O ₂ -L	2.897	2.800	2.808	2.897	2.800	2.808	1.798	1.799
Cm ^{VI} O ₂ -L	2.825	2.758	2.768	2.834	2.734	2.733	1.815	1.814
Bk ^{VI} O ₂ -L	2.818	2.743	2.743	2.810	2.735	2.742	1.841	1.840
Cf ^{VI} O ₂ -L	2.817	2.734	2.739	2.818	2.739	2.745	1.832	1.830
Bond Angles								
	O _{yl} =An=O _{yl}	N1-An-N2	N2-An-N3	N3-An-N4	N4-An-N5	N5-An-N6	N6-An-N1	
Am ^V O ₂ -L ⁻	180.0	58.0	65.8	88.6	58.0	65.8	57.7	
Cm ^V O ₂ -L ⁻	180.0	58.3	65.7	57.8	58.0	65.6	58.0	
Bk ^V O ₂ -L ⁻	180.0	58.4	65.4	58.2	58.5	65.6	58.2	
Cf ^V O ₂ -L ⁻	179.7	59.2	65.9	58.8	58.2	64.2	58.4	
Am ^{VI} O ₂ -L	180.0	57.9	65.9	57.6	57.9	65.9	57.6	
Cm ^{VI} O ₂ -L	179.9	58.4	65.0	57.9	58.5	65.8	58.3	
Bk ^{VI} O ₂ -L	179.9	58.5	65.2	58.3	58.7	65.3	58.3	

Cf ^{VI} O ₂ -L	180.0	58.6	65.4	58.2	58.5	65.3	58.1
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Table S4. Mayer, Nalewajski-Mrozek and Gophinatan-Jug Bond Orders of the Actinyl (V and VI) Complexes

	Am ^V O ₂ -L ⁻	Cm ^V O ₂ -L ⁻	Bk ^V O ₂ -L ⁻	Cf ^V O ₂ -L ⁻	Am ^{VI} O ₂ -L	Cm ^{VI} O ₂ -L	Bk ^{VI} O ₂ -L	Cf ^{VI} O ₂ -L
Mayer Bond Orders								
An=O _{yl} (1)	1.843	1.732	1.679	1.652	1.862	1.717	1.642	1.664
An=O _{yl} (2)	1.860	1.753	1.693	1.670	1.887	1.743	1.656	1.682
An-N1	0.121	0.116	0.126	0.137	0.218	0.157	0.145	0.135
An-N2	0.167	0.165	0.173	0.166	0.287	0.228	0.194	0.184
An-N3	0.165	0.164	0.171	0.173	0.287	0.228	0.191	0.180
An-N4	0.121	0.120	0.125	0.148	0.218	0.161	0.144	0.134
An-N5	0.168	0.168	0.170	0.203	0.287	0.199	0.190	0.185
An-N6	0.165	0.163	0.169	0.205	0.287	0.204	0.192	0.182
Nalewajski-Mrozek Bond Orders								
An=O _{yl} (1)	2.006	2.315	1.813	1.769	1.994	2.297	1.737	1.765
An=O _{yl} (2)	2.015	2.330	1.824	1.783	2.004	2.321	1.747	1.779
An-N1	0.112	0.223	0.131	0.153	0.208	0.262	0.146	0.151
An-N2	0.157	0.293	0.181	0.182	0.280	0.360	0.195	0.206
An-N3	0.156	0.294	0.180	0.193	0.282	0.362	0.193	0.204
An-N4	0.112	0.230	0.130	0.166	0.208	0.193	0.143	0.150
An-N5	0.157	0.299	0.178	0.226	0.280	0.316	0.191	0.208
An-N6	0.156	0.291	0.178	0.231	0.282	0.323	0.193	0.206
Gophinatan-Jug Bond Order								
An=O _{yl} (1)	1.593	2.100	1.271	1.268	1.648	2.178	1.274	1.275
An=O _{yl} (2)	1.602	2.114	1.280	1.279	1.658	2.156	1.282	1.287
An-N1	0.096	0.207	0.100	0.117	0.181	0.248	0.114	0.115
An-N2	0.135	0.273	0.137	0.138	0.244	0.343	0.152	0.157
An-N3	0.134	0.274	0.138	0.147	0.245	0.346	0.151	0.155

An-N4	0.096	0.214	0.099	0.127	0.181	0.151	0.112	0.115
An-N5	0.135	0.278	0.135	0.172	0.244	0.300	0.149	0.159
An-N6	0.134	0.271	0.135	0.176	0.245	0.307	0.151	0.157

Table S5. Calculated nitrogen NPA charges on the free actinyl (V and VI) cations and ligand anion at the PBE/TZP level of theory in gas phase

	Am ^V O ₂ ⁺	Cm ^V O ₂ ⁺	Bk ^V O ₂ ⁺	Cf ^V O ₂ ⁺	Am ^{VI} O ₂ ²⁺	Cm ^{VI} O ₂ ²⁺	Bk ^{VI} O ₂ ²⁺	Cf ^{VI} O ₂ ²⁺
Q(An)	1.648	1.816	1.972	1.867	2.242	1.983	2.161	2.245
Q(O _{yl} (1/2))	-0.323	-0.408	-0.486	-0.433	-0.121	0.008	-0.080	-0.123
Free ligand anion								
N1/N4	-0.342							
N2/N5	-0.352							
N3/N6	-0.356							

Table S6. NPA charges on the actinide and donor atoms (atomic units) of the actinyl (V and VI) complexes using the PBE/ZORA-TZP level of theory in gas phase

	Am ^V O ₂ -L ⁻	Cm ^V O ₂ -L ⁻	Bk ^V O ₂ -L ⁻	Cf ^V O ₂ -L ⁻	Am ^{VI} O ₂ -L	Cm ^{VI} O ₂ -L	Bk ^{VI} O ₂ -L	Cf ^{VI} O ₂ -L
(An)	1.237	1.398	1.516	1.431	1.216	1.302	1.470	1.417
(O _{yl} (1))	-0.422	-0.515	-0.593	-0.516	-0.311	-0.316	-0.494	-0.497
(O _{yl} (2))	-0.416	-0.506	-0.584	-0.506	-0.303	-0.300	-0.483	-0.485
(N1)	-0.385	-0.384	-0.383	-0.391	-0.363	-0.393	-0.384	-0.378
(N2)	-0.421	-0.423	-0.421	-0.414	-0.401	-0.439	-0.432	-0.423
(N3)	-0.419	-0.421	-0.419	-0.414	-0.400	-0.437	-0.430	-0.422
(N4)	-0.385	-0.385	-0.382	-0.393	-0.363	-0.394	-0.384	-0.378
(N5)	-0.421	-0.424	-0.420	-0.462	-0.401	-0.434	-0.430	-0.424
(N6)	-0.419	-0.421	-0.419	-0.460	-0.400	-0.432	-0.431	-0.423

Table S7.
Populations
density of An
Bk and Cf)
actinyl (VI and

Complex	Natural Orbital Population				Spin Density on An		
	5f	6d	7s	7p	Complex ^a	Bare actinyl ^b	An ^c
Am ^V O ₂ -L ⁻	6.38	1.14	0.17	0.01	4.459	4.518	f ⁴
Cm ^V O ₂ -L ⁻	7.20	1.17	0.17	0.01	5.269	5.198	f ⁵
Bk ^V O ₂ -L ⁻	8.00	1.25	0.18	0.01	5.478	5.343	f ⁶
Cf ^V O ₂ -L ⁻	9.10	1.23	0.18	0.01	4.388	4.966	f ⁷
Am ^{VI} O ₂ -L	6.33	1.38	0.19	0.01	3.680	3.452	f ³
Cm ^{VI} O ₂ -L	7.18	1.27	0.19	0.01	4.892	4.719	f ⁴
Bk ^{VI} O ₂ -L	7.99	1.30	0.18	0.01	5.350	5.033	f ⁵
Cf ^{VI} O ₂ -L	9.10	1.25	0.18	0.01	4.395	4.565	f ⁶

Natural Orbital
and spin
(An = Am, Cm,
centers in
V) complexes

^aActinide spin densities for actinyl (V and VI) complexes. ^bActinide spin densities for bare actinyl (V and VI) cations, calculated at the same level of theory. ^cFormal f-orbital occupations corresponding to formal actinide spin densities for the actinyl (V and VI) cations.

Table S8. Natural Orbital Population and spin density of An (An = Am, Cm, Bk and Cf) centers in the bare actinyl (V and VI) cations

Actinyl Moiety	NPA Charge	Natural Orbital Population				Spin Density
	An	5f	6d	7s	7p	An
Am ^V O ₂ ⁺	-0.324	6.53	0.80	0.03	0.00	4.518
Cm ^V O ₂ ⁺	-0.408	7.36	0.81	0.03	0.00	5.198
Bk ^V O ₂ ⁺	-0.486	8.17	0.85	0.03	0.01	5.343
Cf ^V O ₂ ⁺	-0.433	9.21	0.84	0.01	0.08	4.966
Am ^{VI} O ₂ ²⁺	-0.121	6.12	0.87	0.03	0.00	3.452
Cm ^{VI} O ₂ ²⁺	0.008	7.22	0.80	0.02	0.00	4.719
Bk ^{VI} O ₂ ²⁺	-0.080	8.04	0.81	0.02	0.00	5.033
Cf ^{VI} O ₂ ²⁺	-0.123	9.05	0.71	0.02	0.00	4.565

Table S9. Calculated spin density on the actinide and donor atoms (atomic units) of the actinyl (V and VI) complexes at the PBE/ZORA-TZP level of theory in gas phase

	Am ^V O ₂ -L ⁻	Cm ^V O ₂ -L ⁻	Bk ^V O ₂ -L ⁻	Cf ^V O ₂ -L ⁻	Am ^{VI} O ₂ -L	Cm ^{VI} O ₂ -L	Bk ^{VI} O ₂ -L	Cf ^{VI} O ₂ -L
An	4.459	5.269	5.478	4.388	3.680	4.892	5.350	4.395
An ^a	V	V	VI	V	VI	VI	VI	V
O _{yl} (1)	-0.266	-0.155	0.218	0.315	-0.217	-0.241	0.141	0.322
O _{yl} (2)	-0.267	-0.160	0.216	0.316	-0.219	-0.250	0.134	0.325
(N1)	0.004	0.007	0.009	0.090	-0.033	0.007	0.011	0.023
(N2)	0.003	0.005	0.015	0.020	-0.032	0.015	0.035	0.014
(N3)	0.003	0.005	0.015	0.024	-0.033	0.016	0.034	0.011
(N4)	0.004	0.007	0.009	0.094	-0.033	0.008	0.010	0.023
(N5)	0.003	0.005	0.015	0.089	-0.032	0.015	0.032	0.015
(N6)	0.003	0.005	0.015	0.089	-0.032	0.016	0.037	0.011

^aAssigned oxidation states of the actinide based on the gas-phase optimized geometries and spin density

Table S10. Natural electron configurations of the actinide in the free actinyl (VI and V) cations

Actinyl moiety	alpha				beta			
	5f	6d	7s	7p	5f	6d	7s	7p
Am ^V O ₂ ⁺	5.49	0.43	0.02	0.00	1.04	0.36	0.02	0.00
Cm ^V O ₂ ⁺	6.25	0.43	0.02	0.00	1.11	0.37	0.01	0.00
Bk ^V O ₂ ⁺	6.75	0.43	0.01	0.00	1.42	0.42	0.01	0.00
Cf ^V O ₂ ⁺	6.96	0.52	0.06	0.01	2.24	0.32	0.02	0.00
Am ^{VI} O ₂ ²⁺	4.76	0.46	0.02	0.00	1.36	0.41	0.01	0.00
Cm ^{VI} O ₂ ²⁺	5.93	0.44	0.01	0.00	1.29	0.36	0.01	0.00
Bk ^{VI} O ₂ ²⁺	6.51	0.43	0.01	0.00	1.53	0.38	0.01	0.00
Cf ^{VI} O ₂ ²⁺	6.78	0.38	0.01	0.00	2.26	0.34	0.01	0.00

Table S11. Natural electron configurations of the actinide in actinyl (VI and V) complexes

Complex	alpha				beta			
	5f	6d	7s	7p	5f	6d	7s	7p
Am ^V O ₂ -L ⁻	5.38	0.61	0.09	0.00	1.00	0.53	0.08	0.01
Cm ^V O ₂ -L ⁻	6.19	0.62	0.09	0.00	1.01	0.55	0.08	0.01
Bk ^V O ₂ -L ⁻	6.72	0.64	0.09	0.00	1.65	0.61	0.08	0.00

Cf ^V O ₂ -L	6.74	0.62	0.09	0.00	2.36	0.62	0.09	0.00
Am ^{VI} O ₂ -L	4.96	0.73	0.10	0.01	1.37	0.65	0.09	0.01
Cm ^{VI} O ₂ -L	5.99	0.68	0.10	0.00	1.19	0.59	0.09	0.00
Bk ^{VI} O ₂ -L	6.65	0.67	0.09	0.01	1.35	0.63	0.09	0.00
Cf ^{VI} O ₂ -L	6.74	0.62	0.09	0.01	2.36	0.62	0.09	0.00

Table S12. QTAIM analysis of electron density (ρ_c) at An-N and An=O_{yl} bond critical points (e/bohr³) computed at the ZORA/PBE level of theory in gas phase.

Complex	An-N1	An-N2	An-N3	An-N4	An-N5	An-N6	An=O _{yl} (1)	An=O _{yl} (2)	N2-N3	N5-N6
Am ^V O ₂ -L	0.0238	0.0298	0.0294	0.0238	0.0298	0.0294	0.272	0.271	-	-
Cm ^V O ₂ -L	0.0234	0.0291	0.0291	0.0244	0.0300	0.0287	0.254	0.253	0.0093	0.0094
Bk ^V O ₂ -L	0.0239	0.0293	0.0291	0.0240	0.0301	0.0297	0.246	0.246	0.0096	0.0096
Cf ^V O ₂ -L	0.0266	0.0345	0.0341	0.0250	0.0268	0.0283	0.246	0.245	0.0109	0.0100
Am ^{VI} O ₂ -L	0.0323	0.0406	0.0406	0.0323	0.0406	0.0406	0.295	0.294	-	-
Cm ^{VI} O ₂ -L	0.0286	0.0311	0.0298	0.0275	0.0364	0.0366	0.281	0.277	0.0109	-
Bk ^{VI} O ₂ -L	0.0268	0.0327	0.0318	0.0265	0.0317	0.0319	0.254	0.252	0.0107	0.0106
Cf ^{VI} O ₂ -L	0.0260	0.0313	0.0308	0.0259	0.0318	0.0313	0.250	0.247	0.0106	0.0106

Table S13. Calculated bond lengths (Å) and bond angles (°) of the actinyl (V and VI) dichloride at the PBE/ZORA-TZP level of theory in CH₂Cl₂ medium.

	Am ^V O ₂ Cl ₂ ⁻	Cm ^V O ₂ Cl ₂ ⁻	Bk ^V O ₂ Cl ₂ ⁻	Cf ^V O ₂ Cl ₂ ⁻	Am ^{VI} O ₂ Cl ₂	Cm ^{VI} O ₂ Cl ₂	Bk ^{VI} O ₂ Cl ₂	Cf ^{VI} O ₂ Cl ₂
Bond Length								
An=O	1.803	1.825	1.835	1.928	1.751	1.756	1.790	1.890
An-Cl	2.720	2.683	2.634	2.610	2.592	2.585	2.580	2.492
Bond Angle								
O=An=O	178.8	177.1	170.8	105.9	179.9	177.5	180.0	92.9
Cl-An-Cl	107.2	108.6	98.6	107.2	177.4	104.0	161.3	90.4

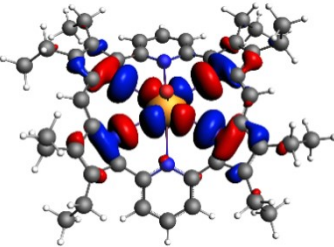
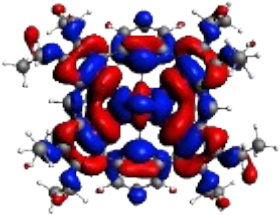
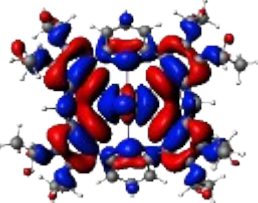
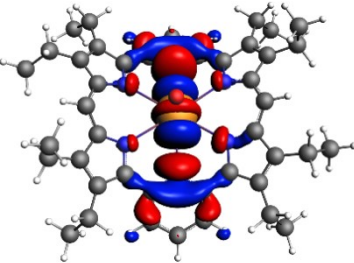
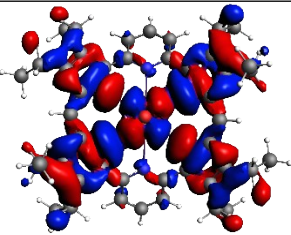
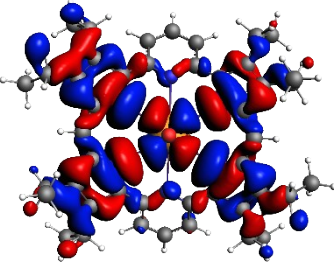
Table S14. Calculated formation energy (ΔE) for the complexation reaction (eq. 1 and eq. 2) with scalar relativity and with SOC relativity (kcal/mol)

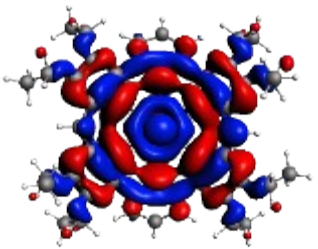
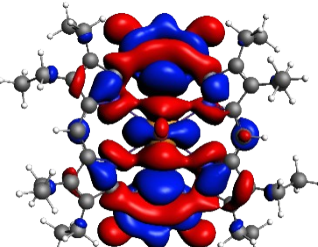
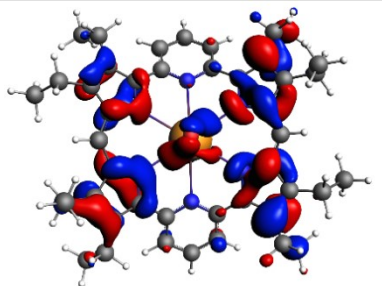
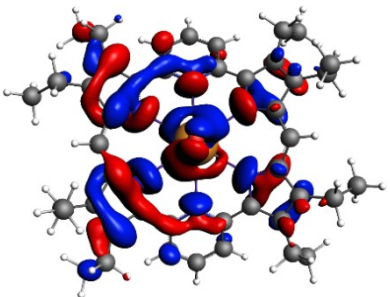
	Am ^V O ₂ -L ⁻	Cm ^V O ₂ -L ⁻	Bk ^V O ₂ -L ⁻	Cf ^V O ₂ -L ⁻	Am ^{VI} O ₂ -L	Cm ^{VI} O ₂ -L	Bk ^{VI} O ₂ -L	Cf ^{VI} O ₂ -L
Eq. 1 – ΔE - SCALAR	-294.1	-302.8	-312.8	-304.8	-619.1	-595.9	-619.0	-641.0
Eq. 1 – ΔE - SOC	-318.6	-332.2	-335.2	-364.7	-615.4	-601.2	-614.5	-628.3
Eq. 2 – ΔE - SCALAR	32.7	31.6	31.0	44.5	19.1	5.7	66.8	-98.9
Eq. 2 – ΔE - SOC	12.6	-5.1	-12.7	27.8	1.4	-14.3	-26.01	-51.2

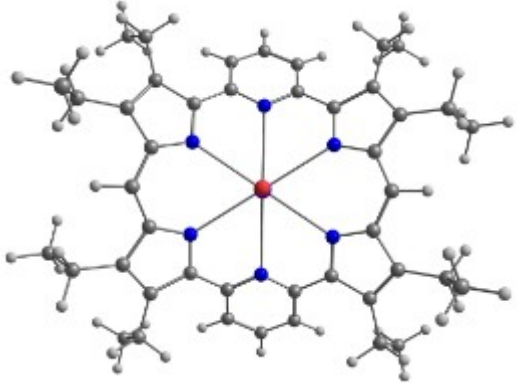
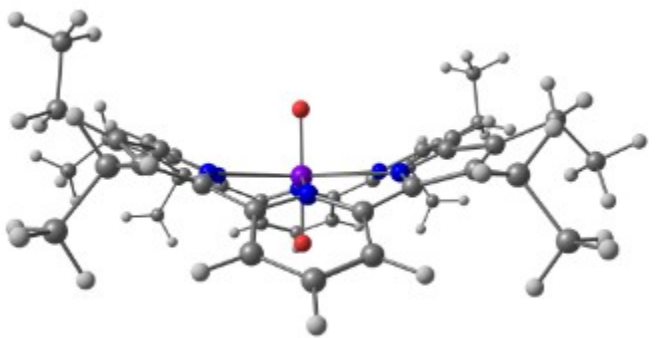
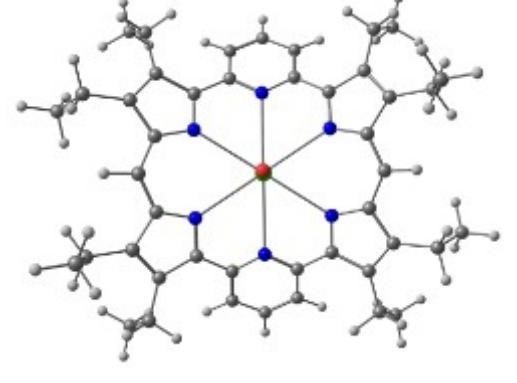
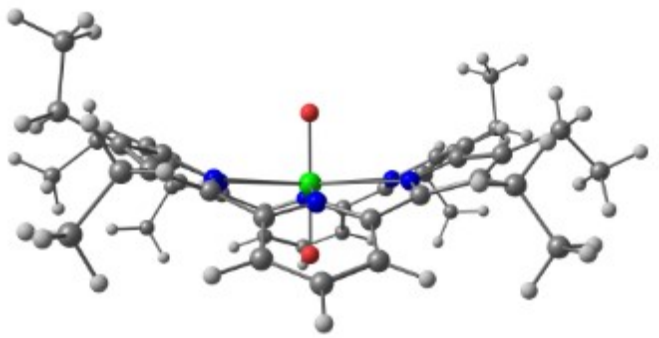
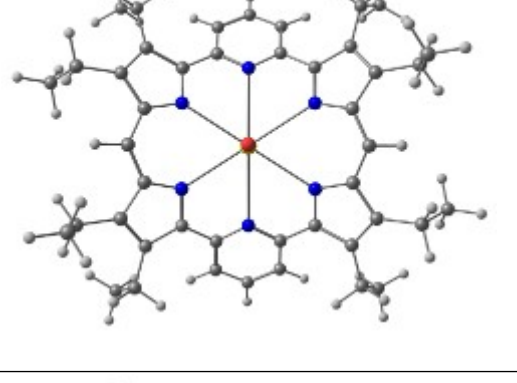
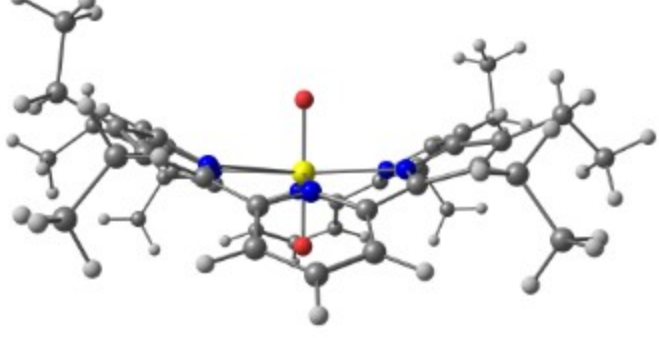
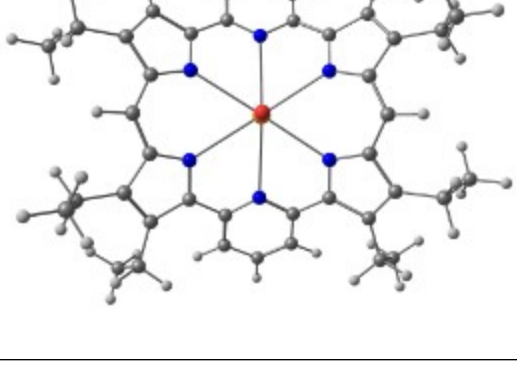
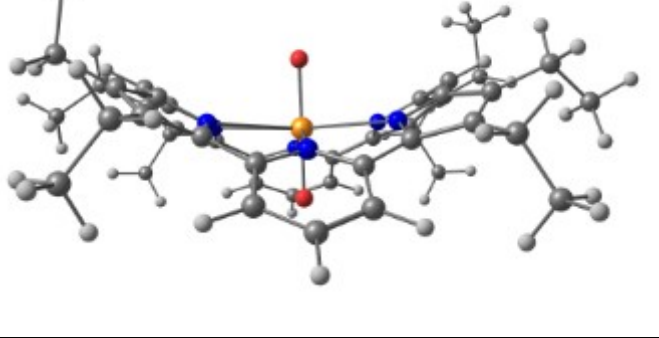
Table S15. EDA results (kcal/mol) for the actinyl (V and VI) complexes in the gas phase at the PBE/TZP level of theory

Energy Terms	Interacting Fragments			
	L ²⁻ vs. Am ^V O ₂ ⁺	L ²⁻ vs. Cm ^V O ₂ ⁺	L ²⁻ vs. Bk ^V O ₂ ⁺	L ²⁻ vs. Cf ^V O ₂ ⁺
ΔE_{int}	-305.35	-313.75	-326.64	-384.39
ΔE_{steric}	-211.68	-214.95	-220.42	-223.22
ΔE_{Pauli}	72.87 (23.86%)	74.76 (23.83%)	76.20 (23.32%)	83.18 (21.64%)
$\Delta E_{\text{el-static}}$	-284.55 (93.19%)	-289.70 (92.33%)	-296.62 (90.81%)	-306.40 (79.71%)
ΔE_{orb}	-93.67 (30.68%)	-98.81 (31.49%)	-106.22 (32.52%)	-161.17 (41.93%)
Energy Terms	Interacting Fragments			
	L ²⁻ vs. Am ^{VI} O ₂ ²⁺	L ²⁻ vs. Cm ^{VI} O ₂ ²⁺	L ²⁻ vs. Bk ^{VI} O ₂ ²⁺	L ²⁻ vs. Cf ^{VI} O ₂ ²⁺
ΔE_{int}	-584.37	-617.03	-638.41	-660.43
ΔE_{steric}	-391.01	-392.70	-396.97	-399.55
ΔE_{Pauli}	117.87 (20.17%)	90.71 (14.70%)	86.56 (13.56%)	85.31 (12.92%)
$\Delta E_{\text{el-static}}$	-508.89 (87.08%)	-483.41 (78.34%)	-483.53 (75.74%)	-484.86 (73.42%)
ΔE_{orb}	-193.35 (33.09%)	-224.33 (36.36%)	-241.44 (37.82%)	-260.87 (39.50%)

Table S16. Selected NOCVs with Energy Values ΔE_i^{orb} ($i = 1-10$) and NOCV Contour Plots of the Deformation Density^a

<i>i</i>	Orbital Type	Complex	ΔE_i^{orb} (kcal/mol)				Dominant NOCV
			Am	Cm	Bk	Cf	
1	σ	$[\text{AnO}_2\text{L}]^{-1}$	-8.150	-10.411	-14.551	-28.599	
		$[\text{AnO}_2\text{L}]^0$	21.410	-45.940	-70.841	-74.415	
2	σ	$[\text{AnO}_2\text{L}]^{-1}$	-13.701	-12.543	-15.802	-26.943	
		$[\text{AnO}_2\text{L}]^0$	-43.985	-18.785	-24.658	-32.495	
3	σ	$[\text{AnO}_2\text{L}]^{-1}$	-6.476	-13.574	-14.136	-15.075	
		$[\text{AnO}_2\text{L}]^0$	-	-22.932	-15.593	-16.902	
4	σ	$[\text{AnO}_2\text{L}]^{-1}$	-11.947	-11.831	-8.134	-12.733	
		$[\text{AnO}_2\text{L}]^0$	-16.831	-20.772	-20.472	-9.792	
5	σ	$[\text{AnO}_2\text{L}]^{-1}$	-8.723	-4.645	-5.128	-7.504	
		$[\text{AnO}_2\text{L}]^0$	-12.956	-16.890	-14.575	-14.277	
6	σ	$[\text{AnO}_2\text{L}]^{-1}$	-4.862	-3.766	-4.603	-10.274	
		$[\text{AnO}_2\text{L}]^0$	-23.006	-9.084	-6.590	-8.794	

7	σ	$[\text{AnO}_2\text{L}]^{-1}$	-4.249	-3.725	-4.127	-9.595	
		$[\text{AnO}_2\text{L}]^0$	-19.130	-8.552	-7.757	-11.571	
8	σ	$[\text{AnO}_2\text{L}]^{-1}$	-3.372	-3.332	-3.249	-8.425	
		$[\text{AnO}_2\text{L}]^0$	-12.190	-7.844	-7.129	-9.605	
9	π	$[\text{AnO}_2\text{L}]^{-1}$	-2.906	-2.772	-3.579	-3.362	
		$[\text{AnO}_2\text{L}]^0$	-3.799	-6.727	-3.287	-4.704	
10	π	$[\text{AnO}_2\text{L}]^{-1}$	-2.313	-2.835	-2.703	-3.396	
		$[\text{AnO}_2\text{L}]^0$	-5.504	-4.264	-4.703	-3.806	
ªBlue – Density inflow; Red – Density outflow (isosurface value: 0.01)							

COMPLEX	TOP VIEW	SIDE VIEW
$\text{Am}^{\text{V}}\text{O}_2\text{-L}^-$		
$\text{Cm}^{\text{V}}\text{O}_2\text{-L}^-$		
$\text{Bk}^{\text{V}}\text{O}_2\text{-L}^-$		
$\text{Cf}^{\text{V}}\text{O}_2\text{-L}^-$		

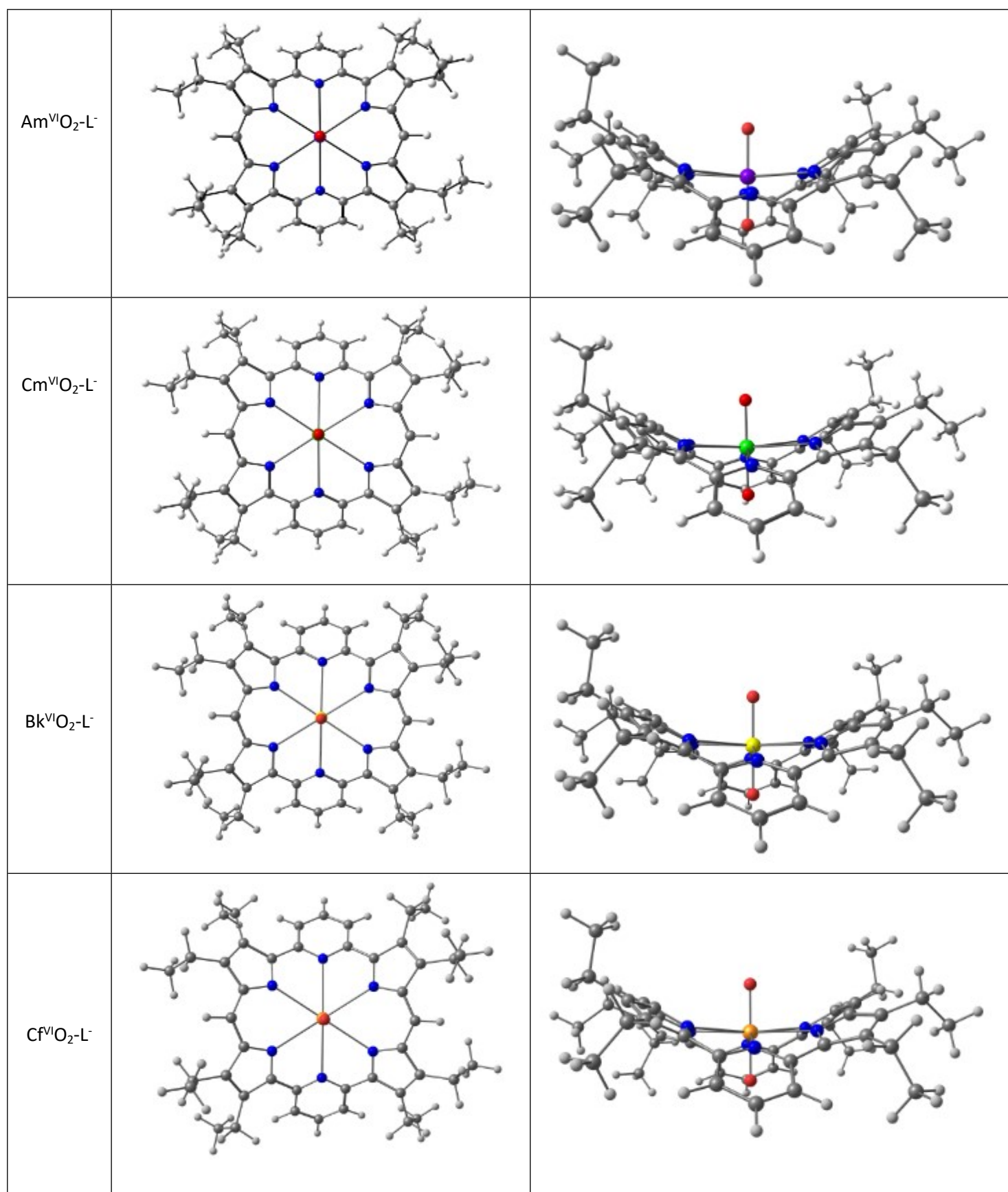
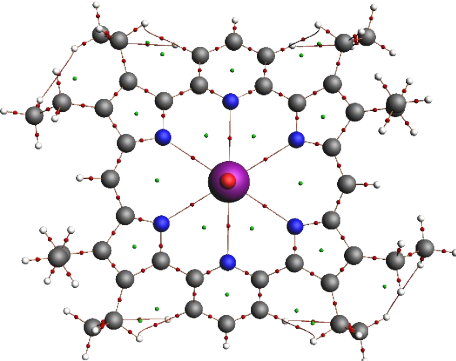
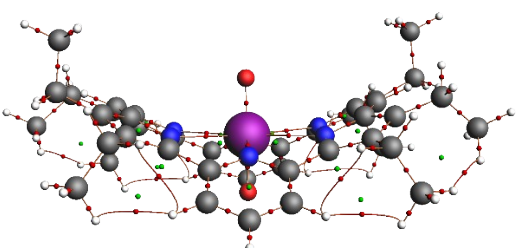
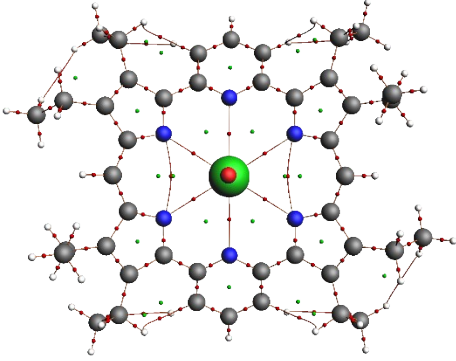
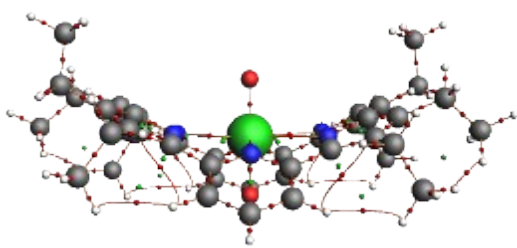
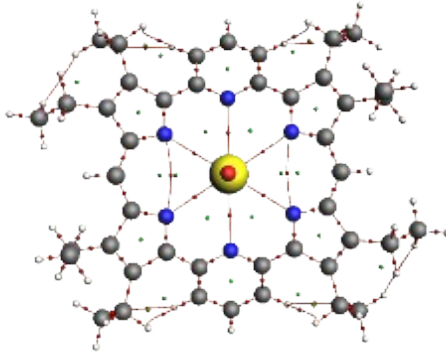
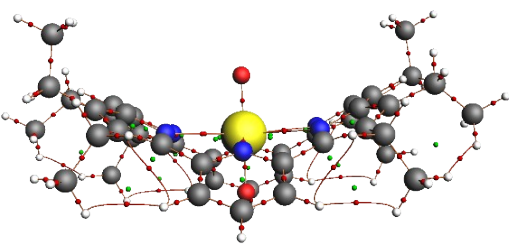
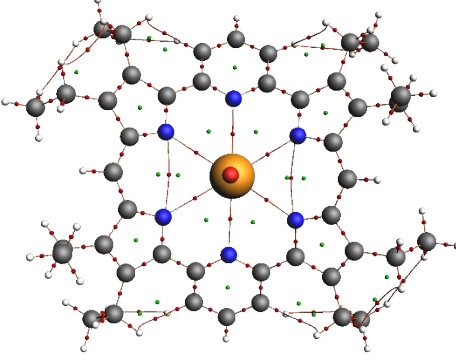
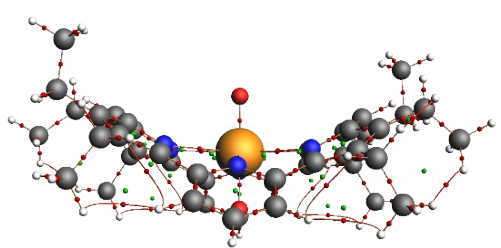


Figure S1. Optimized geometries of Pentavalent and Hexavalent Actinyl - Dipyrimethyrin complexes optimized at PBE/TZP level of theory in gas phase

Complex	Top View	Side View
AmO_2L^-		
CmO_2L^-		
BkO_2L^-		
CfO_2L^-		

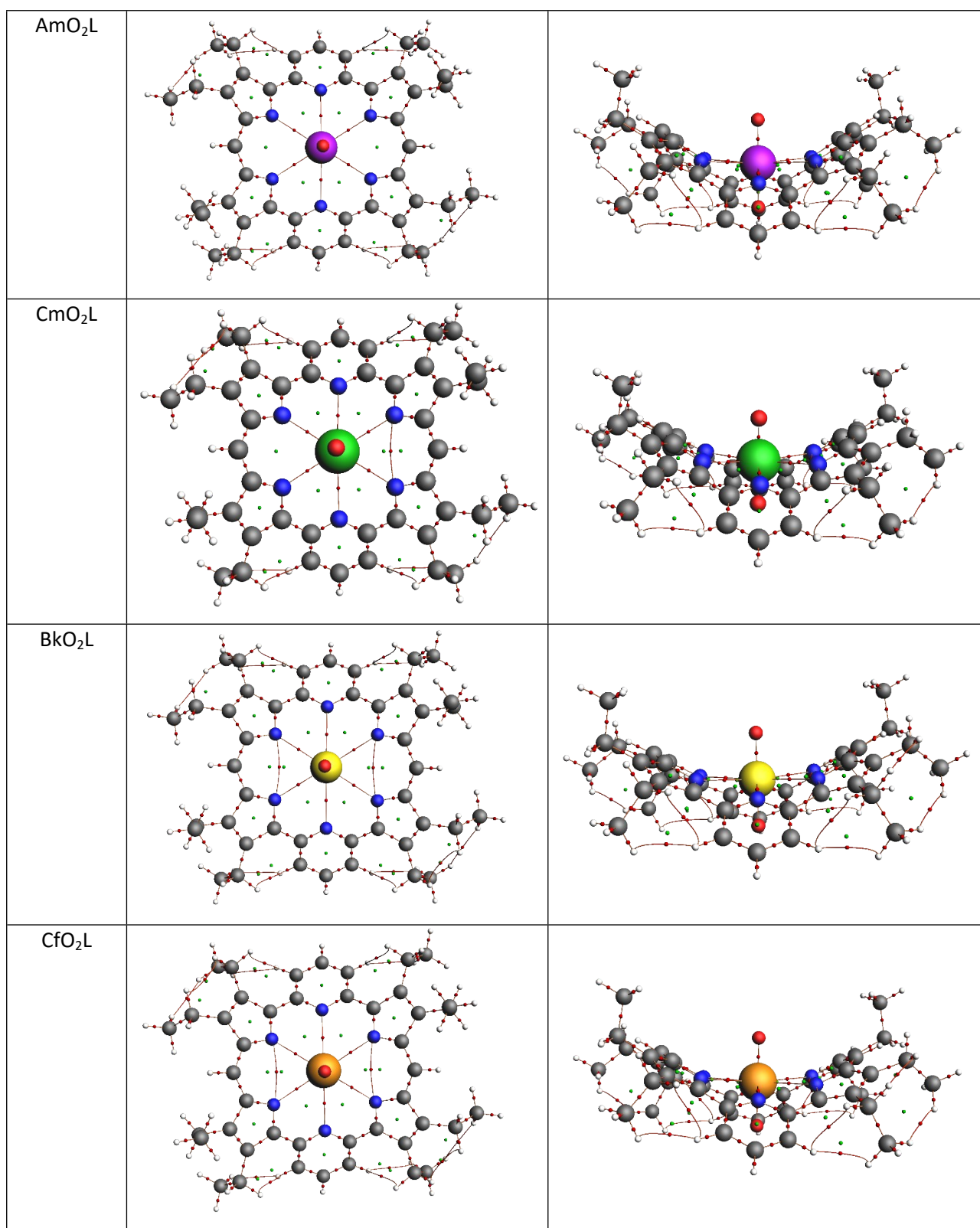


Figure S2. QTAIM molecular topology of [AnO₂L]^{-1/0} complexes, where An=Am, Cm, Bk and Cf where red dots indicate BCP and green dot indicate RCP

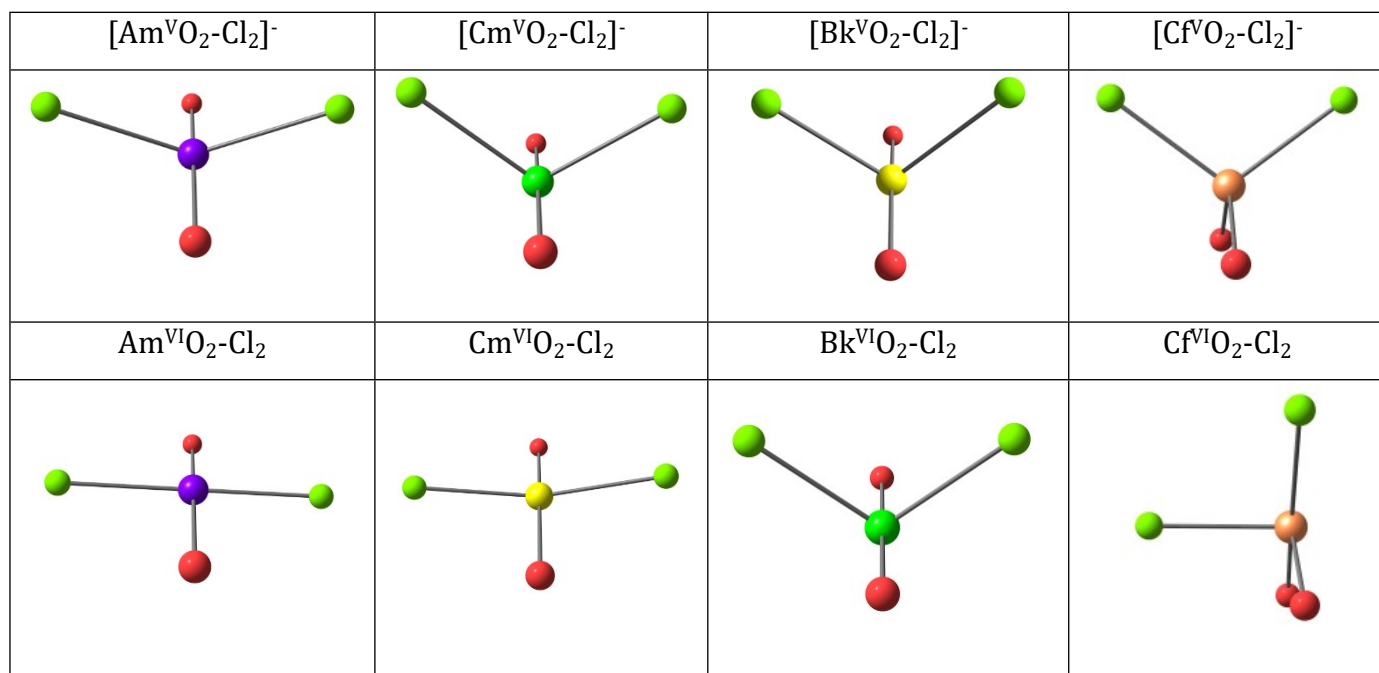


Figure S3. Optimized geometries of $[\text{An}^{\text{V}}\text{O}_2\text{-Cl}_2]^-$ and $\text{An}^{\text{VI}}\text{O}_2\text{-Cl}_2$ in CH_2Cl_2 medium.

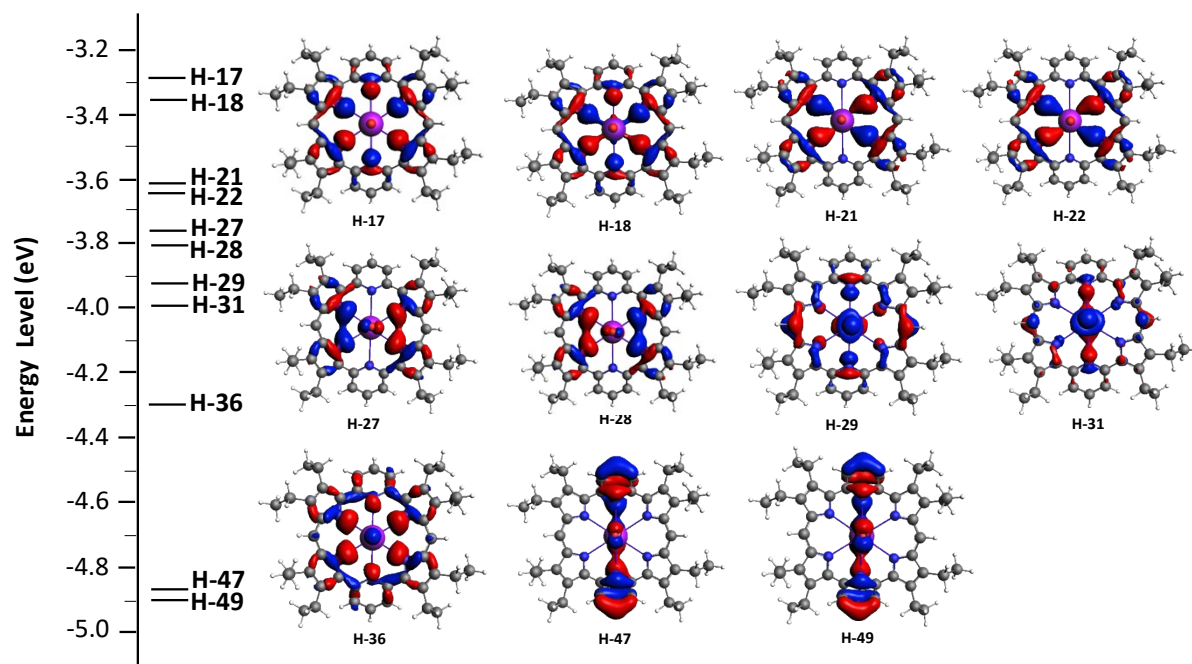


Figure S4(a). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Am}^{\text{V}}\text{O}_2\text{-L}^-$ complex (isovalue is 0.03 a.u.). H represents HOMO

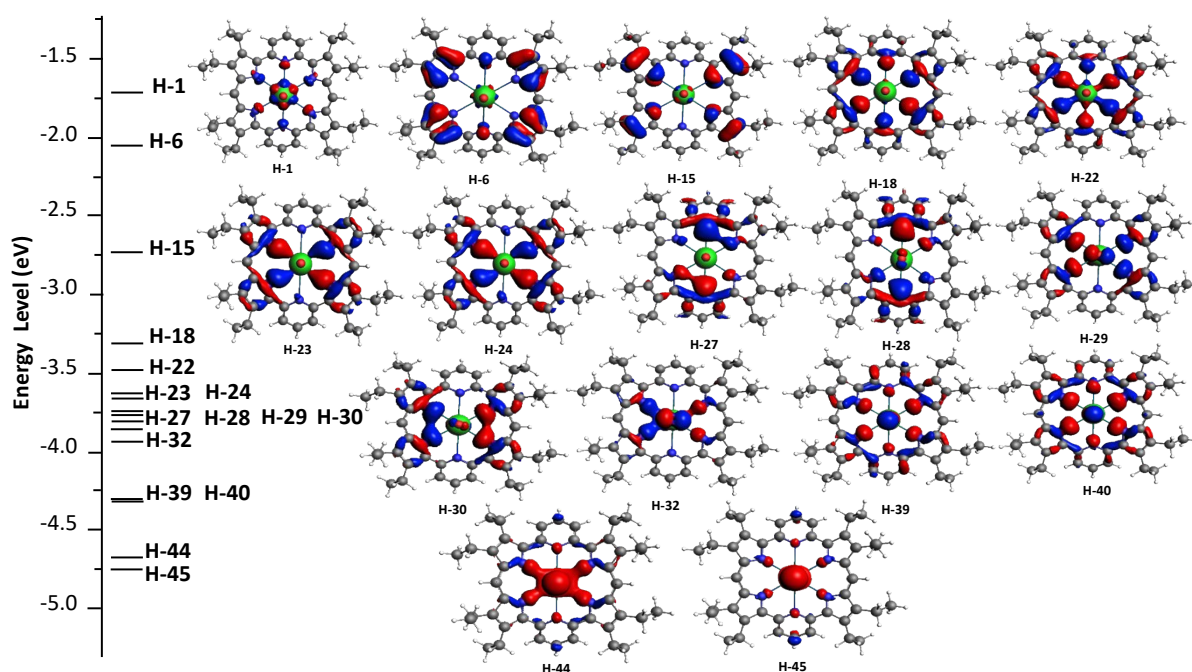


Figure S4(b). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Cm}^{\text{V}}\text{O}_2\text{-L}^-$ complex (isovalue is 0.03 a.u.). H represents HOMO

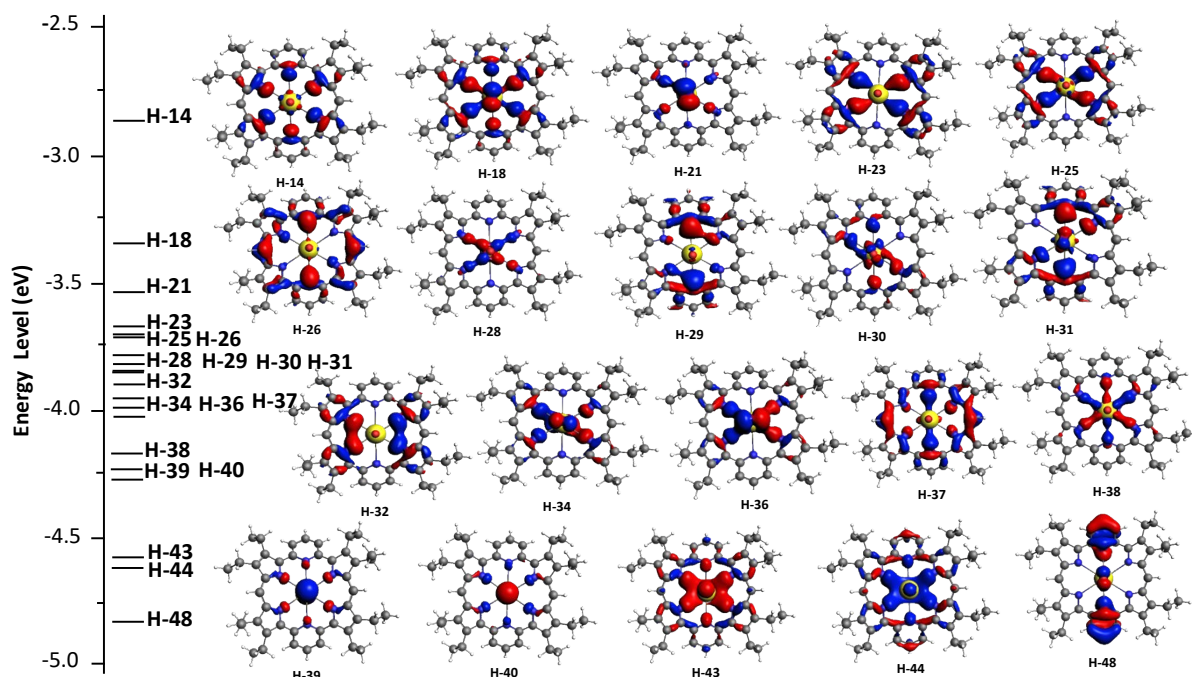


Figure S4(c). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Bk}^{\text{V}}\text{O}_2\text{-L-}$ complex (isovalue is 0.03 a.u.). H represents HOMO

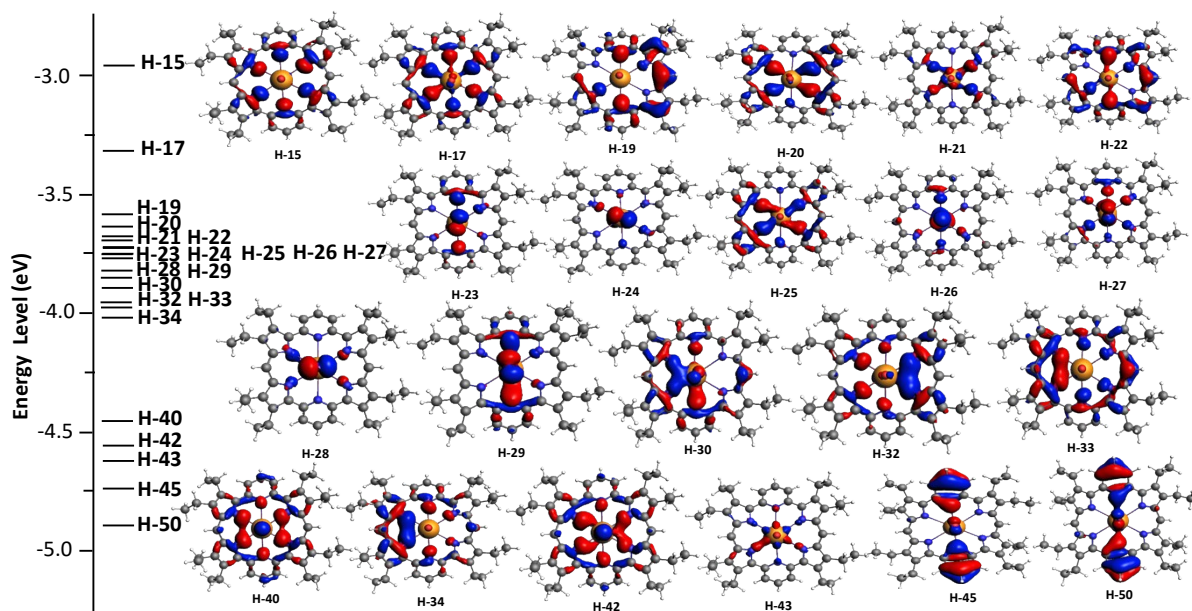


Figure S4(d). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Cf}^{\text{V}}\text{O}_2\text{-L-}$ complex (isovalue is 0.03 a.u.). H represents HOMO

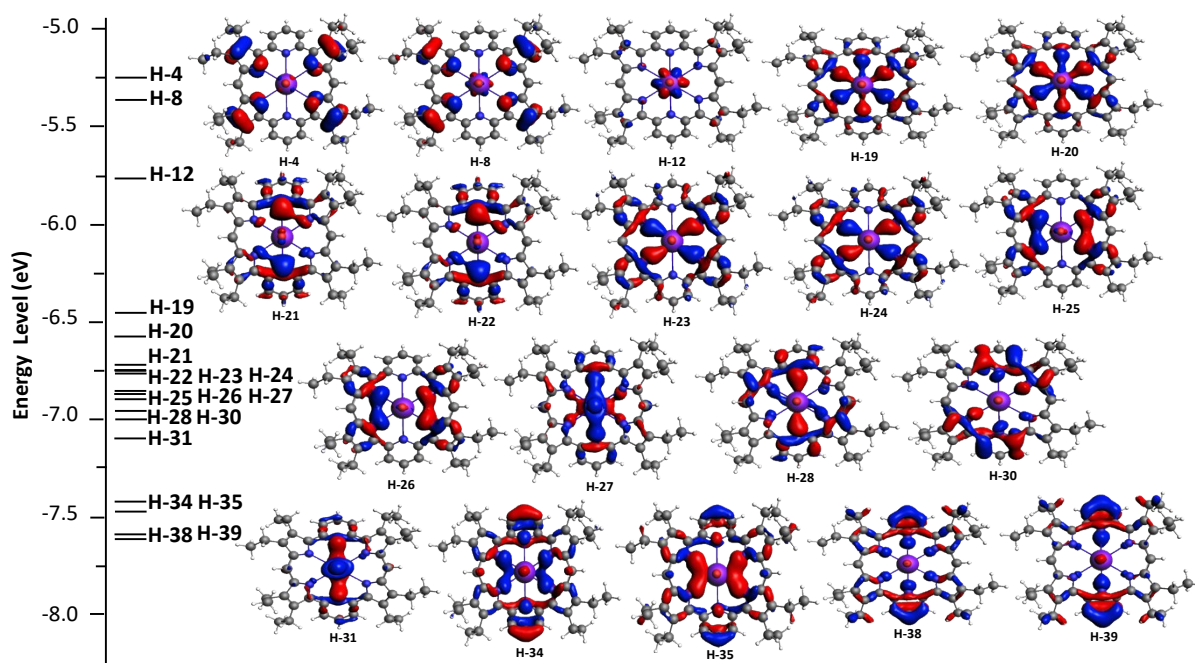


Figure S4(e). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Am}^{\text{VI}}\text{O}_2\text{-L}$ complex (isovalue is 0.03 a.u.). H represents HOMO

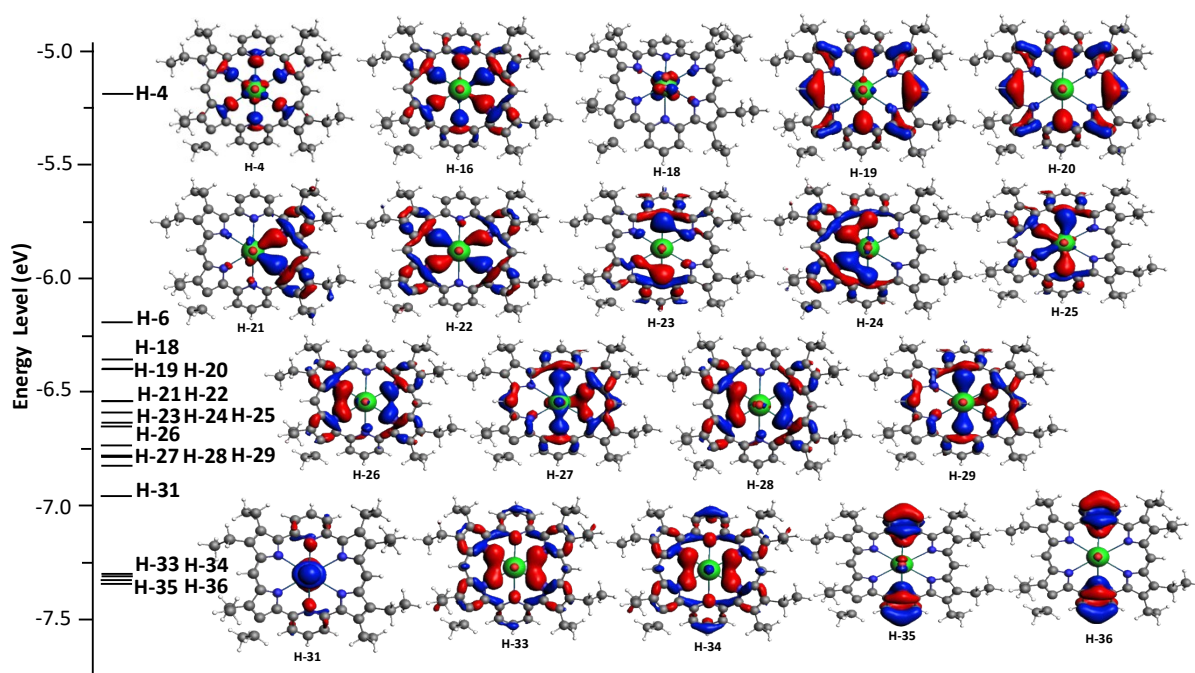


Figure S4(f). The MOs that represent the stabilizing actinyl-ligand interactions for $\text{Cm}^{\text{VI}}\text{O}_2\text{-L}$ complex (isovalue is 0.03 a.u.). H represents HOMO

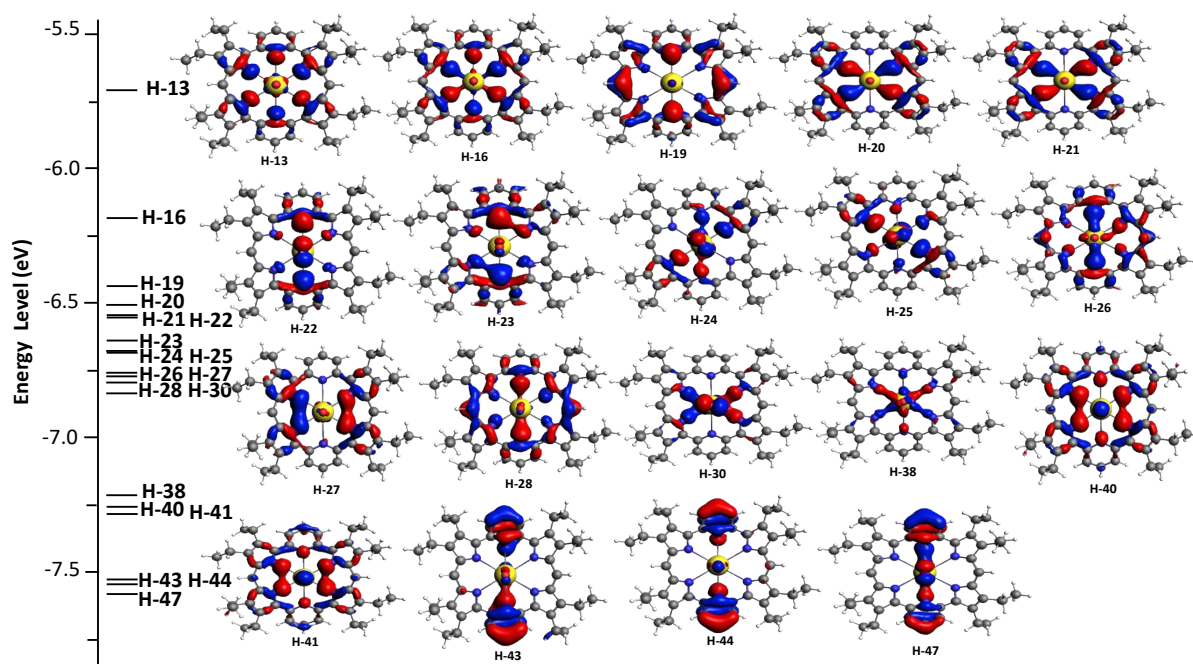


Figure S4(g). The MOs that represent the stabilizing actinyl-ligand interactions for Bk^{VI}O₂-L complex (isovalue is 0.03 a.u.). H represents HOMO

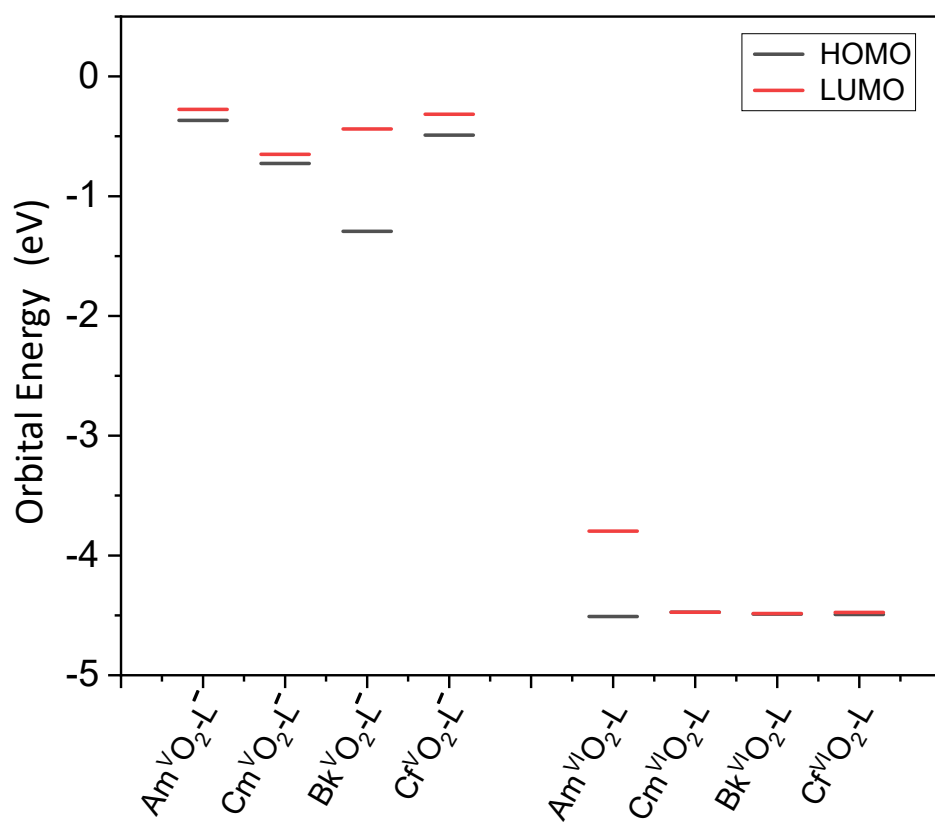


Figure S5. Molecular orbital energy (eV) observed in the $[\text{An}^{\text{V/VI}}\text{O}_2\text{-L}]^{1-/0}$ complexes