Aggregation of gold(I) complexes: phosphorescence vs singlet oxygen production

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Supporting Information



R = H (P1), Cl (P2), Br (P3)

Figure S1. Chemical structure of $P{CH_2-1-N(H)naphthyl}_3$ phosphane.





Figure S4. ¹H-NMR spectra of 2 in DMSO- d_6 .



Figure S5. ³¹P-NMR spectra of 2 in DMSO- d_6 .



Figure S6. ¹H-NMR spectra of 3 in DMSO- d_6 .



Figure S7. ³¹P-NMR spectra of 3 in DMSO- d_6 .



Figure S8. ¹H-NMR spectra of 4 in DMSO- d_6 .



Figure S9. ³¹P-NMR spectra of 4 in DMSO-*d*₆.



Figure S10. ¹H-NMR spectra of 5 in DMSO-*d*₆.



Figure S11. ³¹P-NMR spectra of 5 in DMSO-*d*₆.



Figure S12. ¹H-NMR spectra of 6 in DMSO- d_6 .



Figure S13. ³¹P-NMR spectra of 6 in DMSO-*d*₆.



Figure S14. Mass spectra of 1.



Figure S15. Mass spectra of 2.



Figure S16. Mass spectra of 3.



Figure S17. Mass spectra of 4.



Figure S18. Mass spectra of 5.



Figure S19. Mass spectra of 6.



Figure S20. Representation of the packing of 5 showing the short contacts.



Figure S21. Absorption spectra of phosphanes P1-3.



Figure S22. Emission spectra of phosphanes P1-3.



Figure S23. Emission spectra of 4 in different solvents.



Figure S24. Emission spectra of phosphanes P1-3 at 77K.



Figure S25. Emission spectra of 4 in different concentrations at 77K.



Figure S26. Absorption (left) and emission (right) spectra of complex 1 in DMSO:water mixtures.





Figure S28. Absorption (left) and emission (right) spectra of complex P1 in DMSO:water mixtures.



Figure S29. Absorption (left) and emission (right) spectra of complex P2 in DMSO:water mixtures.



Figure S30. Absorption (left) and emission (right) spectra of complex P3 in DMSO:water mixtures.



Figure S31. Absorption (left) and emission (right) spectra of complex 4 in DMSO:water mixtures.



Figure S32. Absorption (left) and emission (right) spectra of complex 6 in DMSO:water mixtures.



Figure S33. Emission spectra of phosphanes P2 and gold(I) complexes 2 at 77 K in DMSO/water mixture (10:90).



Figure S34. Dynamic light scattering (DLS) of P1.



Figure S35. Dynamic light scattering (DLS) of P2.



Figure S36. Dynamic light scattering (DLS) of P3.



Figure S37. Dynamic light scattering (DLS) of 1.



Figure S38. Dynamic light scattering (DLS) of 2.



Figure S39. Dynamic light scattering (DLS) of 3.



Figure S40. Dynamic light scattering (DLS) of 4.



Figure S41. Dynamic light scattering (DLS) of 5.



Figure S42. Dynamic light scattering (DLS) of 6.



Figure S43. Absorption and emission spectra of phosphanes P1-3 and gold(I) complexes 1-6 in SDS solutions.



Figure S44. Emission spectra of phosphanes P1-3 and gold(I) complexes 1-6 in alginate.



Figure S45. Singlet oxygen spectra for 1H-phenal-1-one (PN) and gold(I) complexes 1-6 in dichloromethane upon excitation on the absorption maxima.



Figure S46. Singlet oxygen spectra for gold(I) complexes 1 and 2 in alginate matrixes upon excitation on the absorption maxima.

Table S1. Crystal data and structure for 5			
Compound	5		
Formula	$C_{70}H_{70}AuCl_7N_6O_4P_2$		
Crystal size, nm	0.4 x 0.34 x 0.12		
Fw	1566.37		
Temp., K	170		
Wavelength, Å	0.71073		
Crystal system	Monoclinic		
Space group	C 1 2/c 1		
a, Å	20.0915(6)		
b, Å	14.5770(4)		
c, Å	24.3861(7)		
α, °	90		
β, °	107.1290(10)		
γ, ^o	90		
Volume, Å ³	6825.3(3)		
Z	4		
D _{calc} , mg m ⁻³	1.524		
Abs. coef., mm ⁻¹	2.530		
F(000)	3168		
θ range for data coll, °	1.748 to 28.282		
Reflns coll./independent	16013/8486		

Data/restraint/parameters	8486/150/412
GOF on F ²	1.278
Final R index $(I > 2\sigma(I))$	$R_1 = 0.0465$ $wR_2 = 0.1200$
R index (all data)	$R_1 = 0.0569$ $wR_2 = 0.1244$
Peak and hole, e Å ⁻³	1.319 and -1.140
CCDC	2165920

Table S2. Phosphorescent quantum yield, lifetime, radiative and non-radiative rateconstants for gold(I) 1-6 in dimethyl sulfoxide at 77K.

Compound	$\Phi_{ ext{Ph}}$	$ au_{Ph}$ (ms)	k _r (ms ⁻¹)	k _{nr} (ms ⁻¹)
1	0.41	2.3	0.178	0.256
2	0.22	2.6	0.084	0.300
3	0.10	1.9	0.052	0.473
4	0.53	3.1	0.170	0.151
5	0.56	2.6	0.215	0.169
6	0.17	1.9	0.089	0.436

Table S3. Emission maxima of phosphanes P1-3 and gold(I) complexes 1-6 inDMSO, PBS, alginate and SDS.

Compound	Emission spectra	Emission spectra	Emission spectra
	(DMSO)	(alginate)	(SDS)
P1	430	422	428
P2	446	440	440

P3	446	434	433
1	420	449	432
2	431	453	434
3	428	448	437
4	426	431	430
5	439	453	439
6	437	448	434