# Electronic Supporting Information (ESI)

## Phosphorus guiding palladium: [4+4] metallomacrocyclic Pd<sup>"</sup> complex and self-assembly of heterometallic Pd<sup>"</sup>/Zn<sup>"</sup> gridtype complex

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Figure S2.  $^{13}C{^{1}H}$  NMR spectrum of 2 in CD<sub>3</sub>CN at 25 °C.



Figure S3.  ${}^{19}F{}^{1}H$  NMR spectrum of 2 in CD<sub>3</sub>CN at 25 °C.







Figure S5. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of **2** in CD<sub>3</sub>CN at 25 °C.



Figure S6.  $^{1}H^{-13}C$  HMBC spectrum of 2 in CD<sub>3</sub>CN at 25 °C.



Figure S7.  $^{1}H^{-13}C$  HSQC spectrum of 2 in CD<sub>3</sub>CN at 25 °C.



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Figure S9. <sup>1</sup>H NMR spectrum of **3** in CD<sub>3</sub>CN at 25 °C.



Figure S10.  $^{13}\text{C}\{^{1}\text{H}\}$  NMR spectrum of 3 in CD\_3CN at 25 °C.



Figure S11.  $^{19}\text{F}\{^{1}\text{H}\}$  NMR spectrum of 3 in CD\_3CN at 25 °C.



Figure S12.  ${}^{31}P{}^{1}H$  NMR spectrum of 3 in CD<sub>3</sub>CN at 25 °C.



Figure S13. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 3 in CD<sub>3</sub>CN at 25 °C.



Figure S14.  $^{1}H$ - $^{13}C$  HMBC spectrum of 3 in CD<sub>3</sub>CN at 25 °C.



Figure S15.  $^{1}H^{-13}C$  HSQC spectrum of **3** in CD<sub>3</sub>CN at 25 °C.



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Figure S17. <sup>1</sup>H-<sup>1</sup>H NOESY spectrum of **3** in CD<sub>2</sub>Cl<sub>2</sub> at 25 °C.



Figure S18. <sup>1</sup>H NMR spectrum of 4 in CD<sub>3</sub>CN at 25 °C.





Figure S20. <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of 4 in CD<sub>3</sub>CN at 25 °C.







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Figure S27. <sup>1</sup>H-<sup>1</sup>H NOESY spectrum of 4 in CD<sub>3</sub>CN at 25 °C.



Figure S28. <sup>1</sup>H NMR spectrum of 5 in CD<sub>3</sub>CN at 25 °C.



Figure S29.  $^{13}C{^{1}H}$  NMR spectrum of 5 in CD<sub>3</sub>CN at 25 °C.



Figure S30.  $^{19}\text{F}\{^{1}\text{H}\}$  NMR spectrum of 5 in CD\_3CN at 25 °C.



Figure S31.  $^{31}P\{^{1}H\}$  NMR spectrum of 5 in CD<sub>3</sub>CN at 25 °C.







Figure S33.  $^{1}H^{-13}C$  HMBC spectrum of 5 in CD<sub>3</sub>CN at 25 °C.



Figure S34.  $^{1}H^{-13}C$  HSQC spectrum of 5 in CD<sub>3</sub>CN at 25 °C.



Figure S35. <sup>1</sup>H-<sup>1</sup>H NOESY spectrum of 5 in CD<sub>3</sub>CN at 25 °C.







Figure S39. <sup>31</sup>P{<sup>1</sup>H} NMR spectrum of 6 in CD<sub>3</sub>CN at 25 °C.



Figure S40. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 6 in CD<sub>3</sub>CN at 25 °C.



Figure S41. <sup>1</sup>H-<sup>13</sup>C HMBC spectrum of 6 in CD<sub>3</sub>CN at 25 °C.



Figure S42. <sup>1</sup>H-<sup>13</sup>C HSQC spectrum of 6 in CD<sub>3</sub>CN at 25 °C.



Figure S43. <sup>1</sup>H-<sup>1</sup>H NOESY spectrum of 6 in CD<sub>3</sub>CN at 25 °C.



Figure S44. <sup>1</sup>H NMR spectrum of **7** in CD<sub>3</sub>CN at 25 °C.



Figure S45.  $^{13}C{^1H}$  NMR spectrum of 7 in CD<sub>3</sub>CN at 25 °C.



Figure S46.  ${}^{19}F{}^{1}H$  NMR spectrum of 7 in CD<sub>3</sub>CN at 25 °C.



Figure S47.  ${}^{31}P{}^{1}H$  NMR spectrum of 7 in CD<sub>3</sub>CN at 25 °C.







Figure S49. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 7 in CD<sub>3</sub>CN at 25 °C.



Figure S50.  $^{1}H$ - $^{13}C$  HMBC spectrum of 7 in CD<sub>3</sub>CN at 25 °C.



Figure S51. <sup>1</sup>H-<sup>13</sup>C HSQC spectrum of 7 in CD<sub>3</sub>CN at 25 °C.



Figure S52. <sup>1</sup>H-<sup>1</sup>H NOESY spectrum of 7 in CD<sub>3</sub>CN at 25 °C.

## 7. VT NMR Spectra ( ${}^{1}H$ , ${}^{31}P{}^{1}H$ ) of **4** and **7**



## VT<sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN) of 4

**Figure S53**. VT <sup>1</sup>H NMR spectra of **4** in CD<sub>3</sub>CN from 25 °C to -25 °C.



#### VT<sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN) of 4

Figure S54. VT  $^{1}$ H NMR spectra of 4 (signal at 3.00 ppm at 25 °C) in CD<sub>3</sub>CN from 25 °C to –25 °C.

#### VT <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN) of 4



Figure S55. VT  $^{1}$ H NMR spectra of 4 (signal at 7.36 ppm 25 °C) in CD<sub>3</sub>CN from 25 °C to –25 °C.



VT <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN) of 4

Figure S56. VT  $^{1}$ H NMR spectra of 4 (signal at 8.60 ppm at 25 °C) in CD<sub>3</sub>CN from 25 °C to -25 °C.

## $\underline{VT^{31}P{}^{1}H}NMR$ (162 MHz, $\underline{CD_{3}CN}$ ) of 4



Figure S57. VT  $^{31}P\{^{1}H\}$  NMR spectra of 4 in CD<sub>3</sub>CN from 25 °C to –25 °C.

## VT<sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN) of 7





### <u>VT <sup>31</sup>P{<sup>1</sup>H}NMR (162 MHz, CD<sub>3</sub>CN) of 7</u>



Figure S59. VT  ${}^{31}P{}^{1}H$  NMR spectra of 7 in CD<sub>3</sub>CN from 25 °C to -5 °C.

#### 8. Reaction NMR Spectra



Scheme S1. <sup>1</sup>H and <sup>31</sup>P{<sup>1</sup>H} NMR spectra of the reaction mixture of complex 4 with  $[Pd(CH_3CN)_4](OTf)_2$ showing the formation of 3 and an unidentified species (X).

Reactions to form 7



Figure S60. <sup>1</sup>H and <sup>31</sup>P{<sup>1</sup>H} NMR spectra of the formation of 7 based on different building blocks.

#### Disassembly and reassembly of 7



Scheme S2.  $^1\text{H}$  and  $^{31}\text{P}\{^1\text{H}\}$  NMR spectra of the disassembly and reassembly of 7.

9. <sup>1</sup>H DOSY NMR Spectra of Complexes 2–7



**Figure S61**. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Pd_2(1)(CH_3CN)_2](OTf)_4$  (2) (8.64 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.



Figure S62. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Pd(1)]_4(OTf)_8$  (3) (10.32 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.



Figure S63. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Pd(1)_2](OTf)_2$  (4) (9.17 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.



Figure S64. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Zn(1)_2](OTf)_2$  (5) (8.83 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.



**Figure S65**. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Pd_2Zn(1)_2](OTf)_6$  (6) (8.90 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.



Figure S66. Plot of <sup>1</sup>H DOSY NMR experiment for determination of the diffusion coefficient of  $[Pd_2Zn_2(1)_4](OTf)_8$  (7) (9.23 mmol·L<sup>-1</sup>, 25 °C) in CD<sub>3</sub>CN.

**Table S1**. Diffusion coefficients *D* for complexes **2–7** obtained by <sup>1</sup>H DOSY NMR spectroscopy (CD<sub>3</sub>CN, 25 °C, concentration as indicated) by fitting the individual peaks' integral decay *vs*. the gradient strength applied. For comparability between the

Compound	Concentration (mmol · L <sup>-1</sup> )	Compound diffusion coefficient D [Complex] (10	<i>D</i> [TMS] <sup>J<sup>-10</sup> m<sup>2</sup>· s<sup>-1</sup>)</sup>	D [CD₂HCN]
$[Pd_2(1)(CH_3CN)_2](OTf)_4(2)$	8.64	7.70	28.3	36.0
[Pd( <b>1</b> )] <sub>4</sub> (OTf) <sub>8</sub> ( <b>3</b> )	10.32	4.51	26.9	34.4
[Pd( <b>1</b> ) <sub>2</sub> ](OTf) <sub>2</sub> ( <b>4</b> )	9.17	7.12	28.0	35.7
[Zn(1) <sub>2</sub> ](OTf) <sub>2</sub> (5)	8.83	7.18	28.1	35.6
[Pd <sub>2</sub> Zn( <b>1</b> ) <sub>2</sub> ](OTf) <sub>6</sub> ( <b>6</b> )	8.90	6.21	27.7	35.1
[Pd <sub>2</sub> Zn <sub>2</sub> ( <b>1</b> ) <sub>2</sub> ](OTf) <sub>8</sub> ( <b>7</b> )	9.23	4.79	26.8	33.6

individual measurements, diffusion coefficients *D* for CD<sub>2</sub>HCN and tetramethylsilane (TMS) are also listed.

#### 10. Mass Spectrometry



Figure S67. HR-ESI+ mass spectrum of  $[Pd_2(1)(CH_3CN)_2](OTf)_2$  (2) in  $CH_3CN$ .



Figure S68. HR-ESI+ mass spectrum of  $[Pd(1)]_4(OTf)_8$  (3) in CH<sub>3</sub>CN.



Figure S69. HR-ESI+ mass spectrum of  $[Pd(1)_2](OTf)_2$  (4) in CH<sub>3</sub>CN.



Figure S70. HR-ESI+ mass spectrum of  $[Zn(1)_2](OTf)_2$  (5) in CH<sub>3</sub>CN.



700 750 800 850 900 950 1000 1050 1100 1150 1200 1250 1300 1350 1400 1450 1500 1650 1700 1750 1800 1850 1900 1950 2000 2050 2100 215 m/z

**Figure S71**. HR-ESI+ mass spectrum of  $[Pd_2Zn(1)_2](OTf)_6$  (6) in CH<sub>3</sub>CN.



Figure S72. HR-ESI+ mass spectrum of  $[Pd_2Zn_2(1)_4](OTf)_8$  (7) in CH<sub>3</sub>CN.

## 11. UV/Vis Spectroscopy

Compound	Solvent	Concentration (• 10 <sup>-5</sup> mol • L <sup>-1</sup> )	Wavelength λ (nm)	Absorbance (a.u.)	Molar extinction coefficient ε (L · cm <sup>-1</sup> · mol <sup>-1</sup> )
Ligand ( <b>1</b> )	$CH_2CI_2$	2.9	230	1.366	27100
			286	0.857	17000
			318	1.128	22400
			346	1.095	21700
[Pd <sub>2</sub> ( <b>1</b> )(CH <sub>3</sub> CN) <sub>2</sub> ](OTf) <sub>4</sub> ( <b>2</b> )	CH₃CN	2.6	253	0.843	32200
			<b>293</b> <sup>1</sup>	0.526	20100
			<b>397</b> <sup>1</sup>	0.802	30700
			416	1.006	38500
[Pd( <b>1</b> )] <sub>4</sub> (OTf) <sub>8</sub> ( <b>3</b> )	CH₃CN	2.0	235	1.422	72400
			276	0.816	41600
			331	1.074	54700
			373	1.026	52300
[Pd( <b>1</b> ) <sub>2</sub> ](OTf) <sub>2</sub> ( <b>4</b> )	CH₃CN	2.3	230	0.970	42600
			285	0.536	23500
			331	0.743	32600
			358 <sup>1</sup>	0.607	26700
			<b>390</b> <sup>1</sup>	0.313	13700
[Zn( <b>1</b> ) <sub>2</sub> ](OTf) <sub>2</sub> ( <b>5</b> )	CH₃CN	2.5	270	0.676	27000
			373	1.380	55100
[Pd <sub>2</sub> Zn( <b>1</b> ) <sub>2</sub> ](OTf) <sub>6</sub> ( <b>6</b> )	CH₃CN	3.7	234	1.104	29600
			273 <sup>1</sup>	0.734	19700
			333 <sup>1</sup>	0.715	19100
			382	1.082	29000
[Pd <sub>2</sub> Zn <sub>2</sub> ( <b>1</b> ) <sub>4</sub> ](OTf) <sub>8</sub> ( <b>7</b> )	CH₃CN	0.7	231	1.029	14100
			266	0.744	10200
			332 <sup>1</sup>	0.682	18200
			375	1.328	93500

 Table S2. UV/Vis data of compounds 1–8.

<sup>1</sup> Shoulder.



Figure S73. UV/Vis spectra of ligand (1) (in  $CH_2CI_2$ ) and complexes 2–4 (in  $CH_3CN$ ).



Figure S74. UV/Vis spectra of ligand (1) (in CH<sub>2</sub>Cl<sub>2</sub>) and complexes 5–7 (in CH<sub>3</sub>CN).

#### 12. Additional Information on X-Ray Diffraction Analyses

Table S3.	Summary	of cr	ystallogr	aphic data.
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Compound	<b>2</b> ·2CH₃CN	<b>3</b> ·4CH₃CN	4.500.01	<b>7</b> ⋅6CH <sub>3</sub> CN	
		$\cdot CH_2 CI_2$ <sup>[1]</sup>	4-3CH <sub>2</sub> Cl <sub>2</sub>	·6CH <sub>2</sub> Cl <sub>2</sub>	
Empirical formula	$C_{43}H_{40}F_{12}N_{11}$	$C_{141}H_{124}CI_2F_{24}$	$C_{69}H_{66}CI_{10}F_{6}$	$C_{150}H_{142}CI_{12}F_{24}$	
	$O_{12}PPd_2S_4$	$N_{32}O_{24}P_4Pd_4S_8$	$N_{14}O_6P_2PdS_2$	$N_{34}O_{24}P_4Pd_2S_8Zn_2$	
Formula weight	1502.87	3983.57	1888.31	4410.27	
<i>Т/</i> К	130	130	130	180	
λ/Å	0.71073	0.71073	0.71073	1.54186	
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic	
Space group	C2/c	P21/c	<i>P</i> 2 <sub>1</sub> /n	P21/c	
a/Å	23.1345(3)	26.4899(3)	13.9015(1)	18.2836(3)	
b/Å	19.3453(2)	21.7388(3)	38.1339(4)	25.6217(3)	
<i>c/</i> Å	50.1762(6)	32.7073(3)	14.9542(2)	39.2969(6)	
α/°	90	90	90	90	
в/°	94.366(1)	95.369(1)	98.185(1)	96.946(1)	
γ/°	90	90	90	90	
V∕/ų	22390.9(5)	18752.1(4)	7846.7(2)	18273.8(5)	
Ζ	16	4	4	4	
$ ho_{calcd.}/g\cdot cm^{-3}$	1.783	1.411	1.598	1.603	
μ/mm <sup>-1</sup>	0.929	0.619	0.746	5.513	
F(000)	12000	8032	3832	8928	
Crystal size/mm <sup>3</sup>	0.050 x 0.150 x	0.250 x 0.400 x	0.300 x 0.300 x	0.050 x 0.200 x	
	0.150	0.400	0.400	0.250	
Colour and shape	Yellow plate	Yellow prism	Orange prism	Yellow plate	
θ <sub>max</sub> /°	27.344	27.367	28.033	68.000	
GooF on F <sup>2</sup>	1.055	1.025	1.092	1.023	
R <sub>int</sub> /%	5.59	6.89	4.62	5.64	
$R_1/wR2(l>2\sigma)/\%$	5.99/13.36	5.58/13.81	6.11/13.69	9.26/24.68	
$R_1/wR2$ (all data)/%	7.93/14.28	8.82/15.38	7.87/14.58	12.18/27.92	
Largest diff.	1 662/-0 708	1 225/_1 505	1 212/ 0 075	1 660/ 1 172	
peak/hole/e∙Å <sup>–3</sup>	1.003/-0.708	1.323/-1.333	1.213/-0.073	1.000/-1.175	

<sup>[1]</sup> The large unit cell of  $3 \cdot 4CH_3CN \cdot CH_2Cl_2$  contained a high number of solvent molecules and a highly disordered triflate in the asymmetric unit, which could not be resolved sufficiently with a reasonable accuracy. A significant unidentified high number of dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>) and acetonitrile molecules (CH<sub>3</sub>CN) was removed by the SQUEEZE procedure. Consequently, the exact amount of lattice solvent molecules was not determinable.



**Figure S75**. Molecular structure of **3**·4CH<sub>3</sub>CN·CH<sub>2</sub>Cl<sub>2</sub> and intermetallic Pd···Pd distances in the rectangular macrocycle (triflate anions, solvent molecules and hydrogen atoms are omitted, carbon atoms are drawn as wireframes and **3** includes non-symmetry related A–D labels for clarity; thermal ellipsoids are set at the 50% probability level).



**Figure S76**. Molecular structure of **7**·6CH<sub>3</sub>CN·6CH<sub>2</sub>Cl<sub>2</sub>and intermetallic Pd···Pd distances in the rectangular macrocycle (triflate anions, solvent molecules and hydrogen atoms are omitted, and carbon atoms are drawn as wireframes for clarity; thermal ellipsoids are set at the 50% probability level).