Electronic Supplementary Information

Coordination mode and stability of the tetrahydroborate ligand in group 10

metal pincer complexes

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Fig. S1 ³¹P{¹H} NMR spectrum of **Pt-POCOP** (162 Hz, THF with THF- d_8)



Fig. S2 ¹¹B NMR spectrum of Pt-POCOP (193 Hz, THF)





Fig. S4 ¹³C{¹H} NMR spectrum of **Ni-PBP** (151 Hz, THF- d_8)



Fig. S6¹¹B NMR spectrum of Ni-PBP (193 MHz, C₆D₆)



Fig. S7 FTIR spectrum of Ni-PBP (KBr disc)



Fig. S8 ¹H NMR spectrum of Pt-PBP (600 MHz, C_6D_6)



Fig. S9 ${}^{13}C{}^{1}H$ NMR spectrum of **Pt-PBP** (151 Hz, C₆D₆)



Fig. S10 ${}^{31}P{}^{1}H$ NMR spectrum of **Pt-PBP** (243 MHz, C₆D₆)



Fig. S11 ¹¹B NMR spectrum of Pt-PBP (193 MHz, CDCl₃)



Fig. S12 FTIR spectrum of Pt-PBP (KBr disc)







Fig. S14 ${}^{13}C{}^{1}H$ NMR spectrum of Pt-PBP' (151 Hz, C₆D₆)



Fig. S16 ¹¹B NMR spectrum of Pt-PBP' (193 MHz, C_6D_6)



Fig. S17 ³¹P{¹H} NMR spectrum of the reaction of $[2,6-({}^{t}Bu_{2}PO)_{2}C_{6}H_{3}]$ PtCl with BH₃ THF (243 MHz, THF with THF-*d*₈). Spectrum recorded 5 min after adding BH₃ THF to the solution of the chloride complex at room temperature.



Fig. S18 ³¹P{¹H} NMR spectrum of the interaction of **Pt-PBP** with CS₂ (243 MHz, THF- d_8). (a) Spectrum recorded immediately after **Pt-PBP** was mixed with 10 equiv. of CS₂ in THF- d_8 . (b) Spectrum recorded after the NMR tube was heated at 50°C for 12 h.



Fig. S19 NPA charges of the nickel centers in different nickel complexes