

Electronic Supplementary Information for
Ni₂P nanowire arrays grown on Ni foam as an efficient
monolithic cocatalyst for visible light dye-sensitized H₂
evolution

Fang Wang, Tongliang Liu, Zhaoting Liu, Zhengguo Zhang and Shixiong Min*

*School of Chemistry and Chemical Engineering, North Minzu University, Yinchuan,
750021, P. R. China.*

*Corresponding authors: sxmin@nwnu.edu.cn

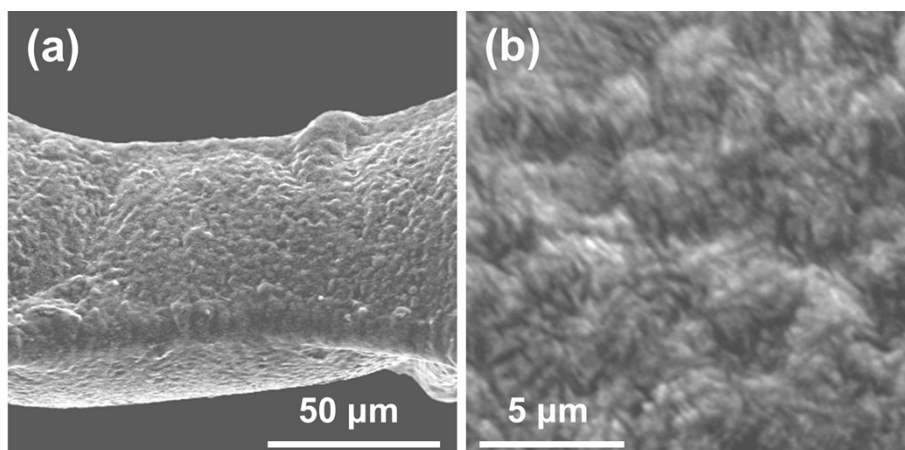


Fig. S1 SEM images of Ni₂P NL/NF.

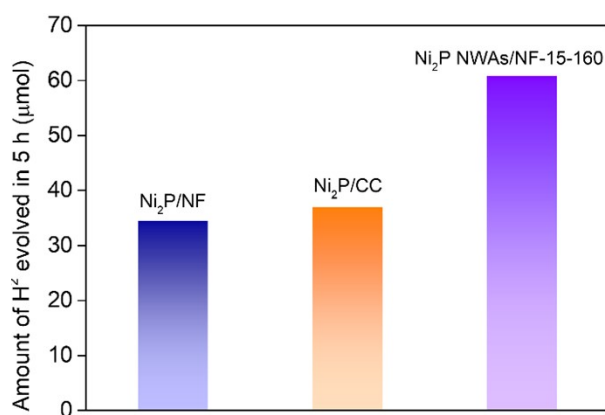


Fig. S2 Photocatalytic HER activities of Ni₂P/CC, Ni₂P/NF, and NWAs/NF-15-160 from ErB-TEOA (10 vol.%, 25 mL, pH 10) system. The Ni₂P/CC and Ni₂P/NF were prepared by drop-coating Ni₂P powder on carbon paper ($d=1.6$ cm) and NF ($d=1.6$ cm), respectively. The loading amount of Ni₂P in Ni₂P/CC and Ni₂P/NF was same as in NWAs/NF-15-160.

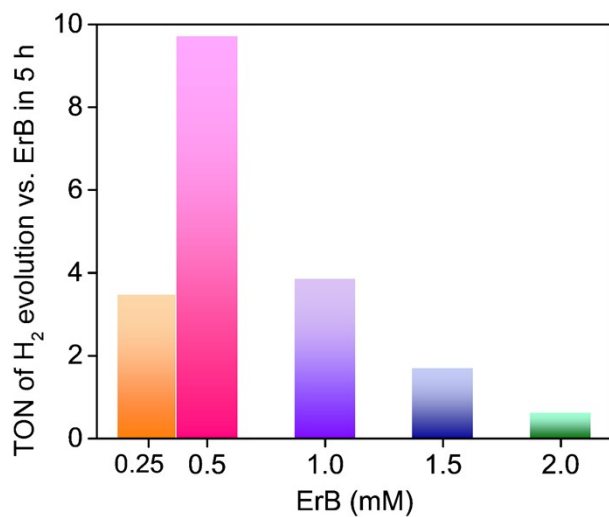


Fig. S3 TON of H₂ evolution based on ErB on Ni₂P NWAs/NF-15-160 from ErB-TEOA (10 vol.%, 25 mL, pH 10) system.

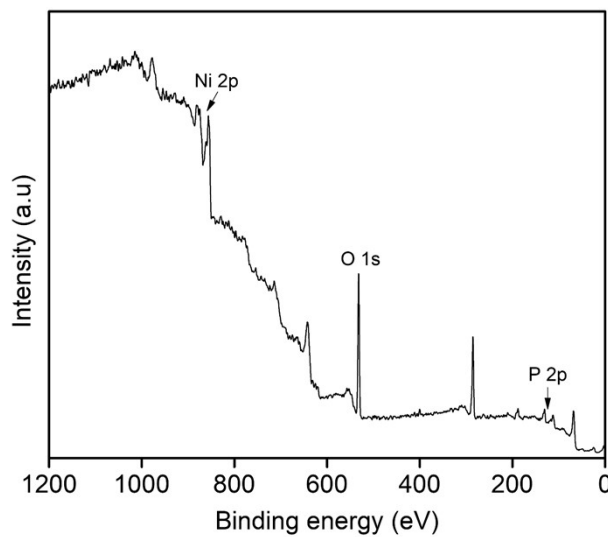


Fig. S4 XPS survey spectrum of Ni₂P NWAs/NF-15-160 after photocatalytic HER.

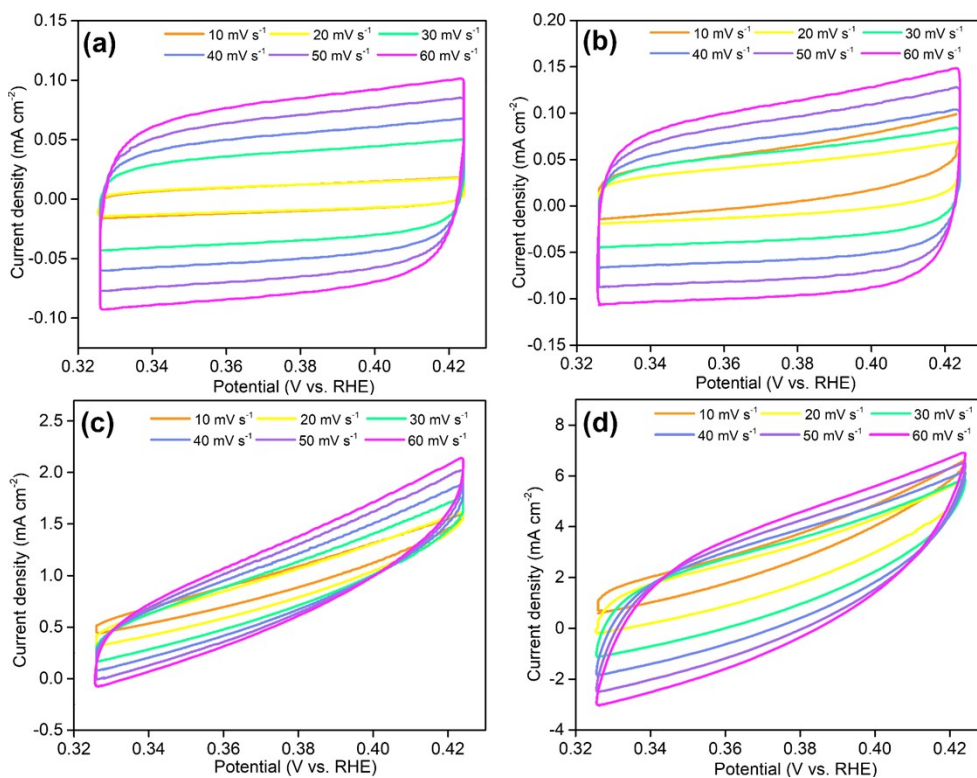


Fig. S5 Cyclic voltammograms (CV) of (a) pristine NF, (b) NF-O-15-160, (c) Ni₂P NL/NF and (d) Ni₂P NWAs/NF-15-160 at different scan rates in 0.5 M Na₂SO₄ solution.

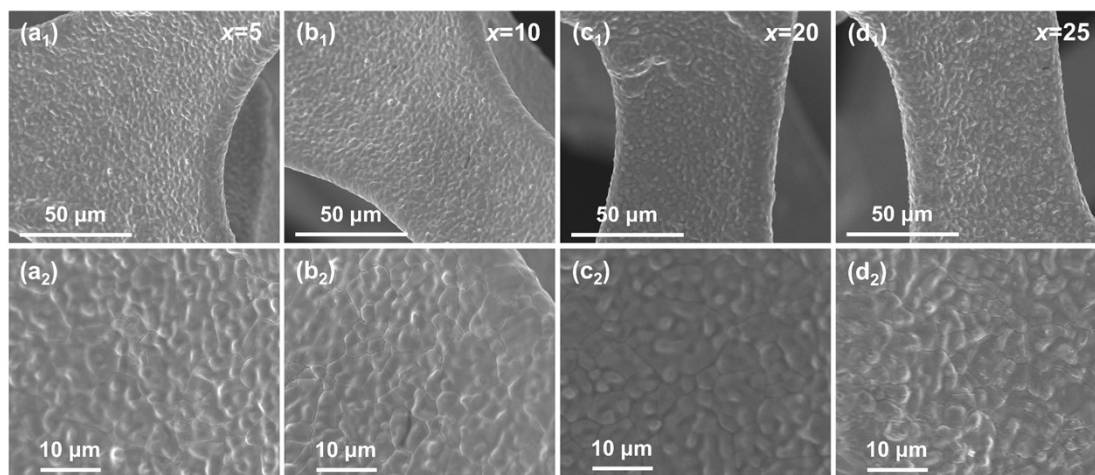


Fig. S6 SEM images of NF-O-*x*-160 samples, where the “*x*” and “160” represent the concentration of H₂O₂ and hydrothermal temperature, respectively.

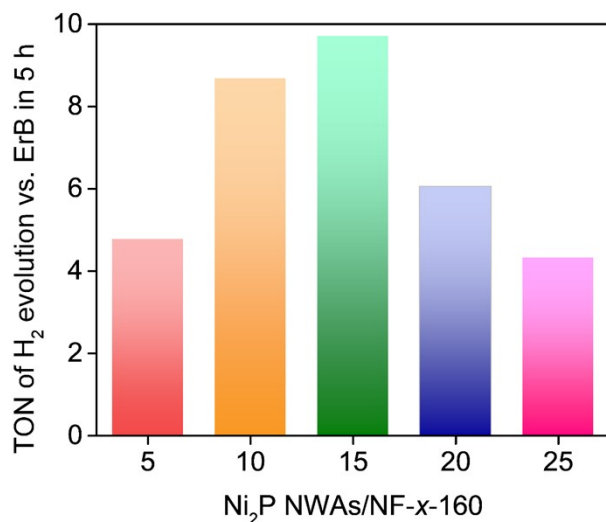


Fig. S7 TON of H₂ evolution on Ni₂P NWAs/NF-x-160 from TEOA solution (10 vol.%, 25 mL, pH 10) containing ErB (0.5 mM) under visible light irradiation.

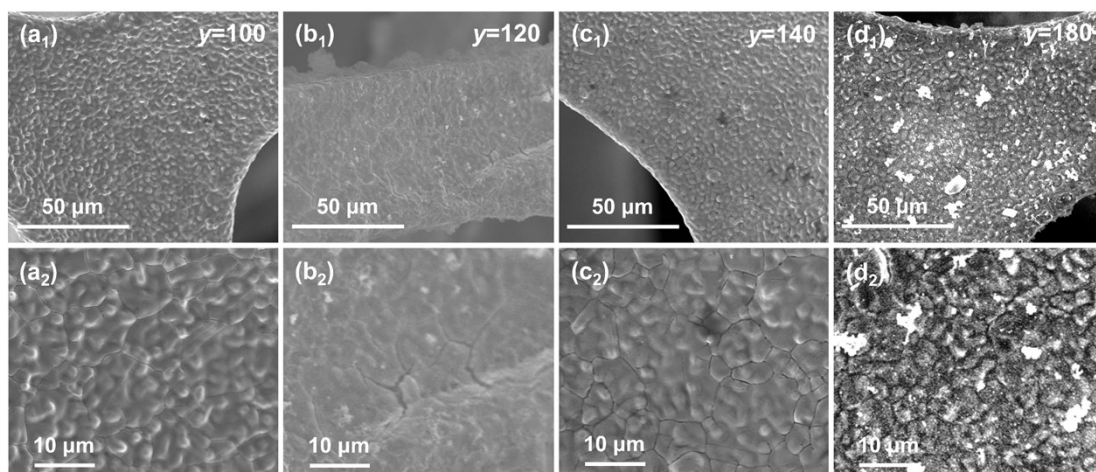


Fig. S8 SEM images of NF-O-15-y samples, where the “15” and “y” represent the concentration of H₂O₂ and hydrothermal temperature, respectively.

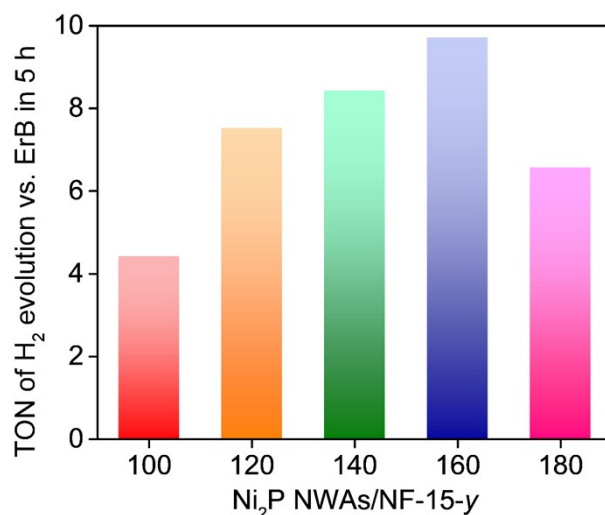


Fig. S9 TON of H₂ evolution on Ni₂P NWAs/NF-*x*-160 from TEOA solution (10 vol.%, 25 mL, pH 10) containing ErB (0.5 mM) under visible light irradiation.

Table S1 Comparison of photocatalytic HER activity of dye-sensitized Ni₂P NWAs/NF with other photocatalysts with Ni₂P as cocatalyst.

Photocatalyst	Form	Reaction conditions	Light source	H ₂ evolution rate (mmol h ⁻¹ g _{catalyst} ⁻¹)	Ref.
Ni ₂ P-CdS/g-C ₃ N ₄	Powder	0.35 M Na ₂ S and 0.25 M Na ₂ SO ₃	300 W Xe lamp (>400 nm)	44.45	1
Cd _{0.5} Zn _{0.5} S@ZnS-Ni ₂ P/g-C ₃ N ₄	Powder	0.35 M Na ₂ S and 0.25 M Na ₂ SO ₃	300 W Xe lamp (>420 nm)	55.43	2
Ni ₁₂ P ₅ /g-C ₃ N ₄	Powder	10 vol% TEOA	350 W Xe lamp (>420 nm)	0.12	3
Ni ₂ P/CdS-DETA	Powder	0.35 M Na ₂ S and 0.25 M Na ₂ SO ₃	300 W Xe lamp (>420 nm)	6.83	4
Ni ₂ P/CdS	Powder	10 vol.% Lactic acid	300 W Xe lamp (>420 nm)	1.18	5
Ni ₂ P-Cd _{0.9} Zn _{0.1} S	Powder	0.7 M Na ₂ S and 0.5 M Na ₂ SO ₃	300 W Xe lamp (>400 nm)	94	6
MoS ₂ -g-C ₃ N ₄ /Ni ₂ P	Powder	10 vol.% TEOA	300 W Xe lamp (>420 nm)	0.29	7
C-Ni ₂ P@CdS	Powder	10 vol.% TEOA	300 W Xe lamp (>420 nm)	28.39	8
Ni ₂ P/ZnIn ₂ S ₄	Powder	10 vol.% Lactic acid	300 W Xe lamp (>420 nm)	2.06	9
NiO/Ni ₂ P/g-C ₃ N ₄	Powder	10 vol.% TEOA	300 W Xe lamp (>420 nm)	0.50	10
7-Ni ₂ P/MoP@g-C ₃ N ₄	Powder	10 vol.% TEOA	300 W Xe lamp (>420 nm)	0.52	11
NiCoP/g-C ₃ N ₄	Powder	10 vol.% CH ₃ OH	300 W Xe lamp (300-700 nm)	0.16	12
Ni ₂ P/TiO ₂ /Ti ₃ C ₂ -ox	Powder	5 vol.% glycerol	35 W UV-visible lamp	16.1	13
Ti ₃ AlC ₂ /TiO ₂ /Ni ₂ P	Powder	5 vol.% glycerol	35 W UV-visible lamp	4.3	14
Ag/pCN-TiO ₂	Monolith	5 vol.% CH ₃ OH	200 W Hg lamp	8.2	15

Ag-pCN/TiO ₂	Monolith	5 vol.% glycerol	200 W Hg lamp	12.1	16
Ni ₂ P NWAs/NF	Monolith	10 vol% TEOA	10W LED (380 nm≤λ≤780 nm)	0.96	This work

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