### **Electronic Supporting Information**

## Boron complexes of π-extended nitroxide ligands exhibiting three-state redox processes and nearinfrared-II (NIR-II) absorption properties

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### General remarks

All the manipulations were performed under a dry Ar atmosphere using Schlenk techniques or using a MBraun MB150B-G-II glovebox. Unless otherwise noted, compounds obtained from commercial suppliers (Tokyo Chemical Industries Co., Ltd., KANTO CHEMICAL Co., Inc., FUJIFILM Wako Pure Chemical Corp., or Sigma-Aldrich) were used without further purifications. Column chromatography was performed using Wako Silicagel 70 PF254, Kanto Silica Gel 60 (spherical, particle size 100-210 µm). <sup>1</sup>H, <sup>13</sup>C and <sup>19</sup>F NMR spectra were measured on a Bruker Avance-III 400 spectrometer (<sup>1</sup>H: 400 MHz, <sup>13</sup>C: 100 MHz, <sup>19</sup>F 376 MHz) and chemical shifts were reported as the delta scale in ppm relative to the residual solvent signals (<sup>1</sup>H and <sup>13</sup>C) or an external reference (CFCl<sub>3</sub>, <sup>19</sup>F). UV-vis-NIR spectra were recorded on a Shimadzu UV-3101PC spectrophotometer. Spectroelectrochemical measurements were conducted on a BAS SEC-2000-UV/vis spectrometer and a BAS Electrochemical Analyzer Model 1140A. ESR spectra were recorded at the X-band frequency using a JEOL JES-X320 spectrometer equipped with a 100 kHz magnetic-field modulation. ESR microwave power low enough to prevent the saturation and distortion of the spectrum. ESR spectra were recorded at 25 °C or -150 °C. High resolution mass spectra (HRMS) were recorded on a Bruker micrOTF or a timsTOF (IMS-QTOF) spectrometer using ESI or APPI mode. Electrochemical measurements were performed under a dry Ar atmosphere using a BAS Electrochemical Analyzer Model 1210C or a BAS Electrochemical Analyzer Model 1140A. Elemental analysis was carried out using a YANACO MT-6 microanalyzer. (C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>BOEt was prepared according to a literature.<sup>S1</sup>

Single crystal X-ray diffraction (XRD) measurements were carried out on a Bruker D8 VENTURE System (PHOTONIII 14 with I $\mu$ S Diamond) (**4a**) or a Rigaku MicroMax-007HF diffractometer equipped with a VariMax DW light source and a Pilatus P200 detector ([**5a**]SbCl<sub>6</sub> and [**5b**]SbCl<sub>6</sub>). A Mo-K $\alpha$  radiation ( $\lambda =$ 0.71073Å) that was mono-chromated by the multilayer confocal mirror was used for the measurement of the diffractions. The crystals were kept at -173 °C (**4a**) or -100 °C ([**5a**]SbCl<sub>6</sub> and [**5b**]SbCl<sub>6</sub>) while the data collections. The collected data were processes using the Bruker SAINT v8.40A (Bruker Nano, Inc., 2019) (**4a**) or CrysAlisPro ver. 1.171.41.117a program package (Rigaku Oxford Diffraction, 2021) ([**5a**]SbCl<sub>6</sub> and [**5b**]SbCl<sub>6</sub>). Using Olex2,<sup>S2</sup> the structures were solved with the SHELXT and refined with the SHELXL program packages.<sup>S3</sup> The full-matrix least-squares refinements were performed on  $F^2$ . All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were refined using the riding model. For [**5a**]SbCl<sub>6</sub>, the following level B CheckCIF alert was pointed, which may stem from the specific beamstop alignment of the Pilatus P200 detector.

PLAT910\_ALERT\_3\_B Missing # of FCF Reflection(s) Below Theta (Min). 15 Note In addition, the crystals of [**5a**]SbCl<sub>6</sub> contained two CH<sub>2</sub>Cl<sub>2</sub> molecules per one [**5a**]SbCl<sub>6</sub> unit, and one of the two CH<sub>2</sub>Cl<sub>2</sub> molecules was highly disordered over four positions (occupancy ratio = 0.30:0.30:0.20:0.20), resulting in the relatively strong residual electron density (1.310 e Å<sup>-3</sup>) around the disordered CH<sub>2</sub>Cl<sub>2</sub> molecules. In order to refine the disordered CH<sub>2</sub>Cl<sub>2</sub> molecule, DFIX and SIMU restraints were applied to the bond lengths and anisotropic parameters of this CH<sub>2</sub>Cl<sub>2</sub> molecule. These crystallographic problems may not have substantial effects on the structural features of the main part CCDC-2182062 (**4a**), 2182063 ([**5a**]SbCl<sub>6</sub>) and 2182064 ([**5b**]SbCl<sub>6</sub>) contain the supplementary crystallographic data for this paper. Density functional theory (DFT) calculations were performed using Gaussian 16, Revision C.02<sup>S4</sup> program package. CAM-B3LYP density functional with 6-31G(d) and 6-31+G(d) base were employed for the geometry optimizations and single point energy calculations, respectively. ESR *g*-tensors were calculated using ORCA  $5.0.3^{S5}$  program suite at the B3LYP/EPR-II level of theory. Natural Bond Orbital (NBO) analyses were performed using NBO 7.0.<sup>S6</sup> The computation was performed using Research Center for Computational Science, Okazaki, Japan (Project: 19-IMS-C235, 20-IMS-C151, 21-IMS-C164, and 22-IMS-C164).

High resolution mass spectrum (HRMS) measurements and elemental analyses were carried out at the Microanalytical Laboratory, Institute for Chemical Research (ICR), Kyoto University and at University of Tsukuba, respectively.

Syntheses and copies of spectra of new compounds N-(5-bromopyridin-2-yl)-N-(tert-butyl)hydroxylamine (2)

$$\begin{array}{c}
\text{Br} & \begin{array}{c}
1) \text{ BuLi} \\
2) (t-\text{BuNO})_2 (0.5 \text{ eq.}) \\
\hline
\text{toluene} \\
-78 ^{\circ}\text{C to } 25 ^{\circ}\text{C} \\
1 \end{array} \xrightarrow{\text{Br}} & \begin{array}{c}
\text{N} \\
\text{N} \\
\text{OH} \\
2 (35\%)
\end{array}$$

To a toluene solution (50 mL) of 2,5-dibromopyridine (1) (3.6 g, 15 mmol) was added n-butyl lithium (1.57 M in hexane, 11 mL, 18 mmol) at -78 °C, and the mixture was stirred for 2 h at -78 °C. This mixture was treated with 2-methyl-2-nitrosopropane dimer (1.70 g, 9.8 mmol) in toluene (20 mL) and stirred for 12 h at 25 °C. The reaction was quenched with aq. NH4Cl, and the mixture was extracted with ethyl acetate. The organic layer was dried with Na<sub>2</sub>SO<sub>4</sub>, and the solvents were evaporated. The crude product was separated by column chromatography (SiO<sub>2</sub>, eluent: hexane/ethyl acetate = 9:1, v/v) to give **2** as a yellow solid (1.3 g, 5.3 mmol, 35%). Compound **2** was rapidly oxidized under air to give the corresponding nitroxide radical, and thus satisfactory <sup>13</sup>C NMR spectrum could not be recorded due to the substantial signal broadening.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.36 (d, J = 1.6 Hz, 1H, ArH), 7.67 (dd, J = 8.7,1.6 Hz, 1H, ArH), 7.06 (d, J = 8.7 Hz, 1H, ArH), 6.91 (brs, 1H, OH), 1.25 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); HRMS (ESI): m/z calcd for C<sub>9</sub>H<sub>14</sub><sup>79</sup>BrN<sub>2</sub>O 245.0284, C<sub>9</sub>H<sub>19</sub><sup>81</sup>BrN<sub>2</sub>O 247.0264. Found: 245.0281, 247.0263 ([M+H]<sup>+</sup>).



#### <sup>1</sup>H NMR of **2** (400 MHz, CDCl<sub>3</sub>)



Aminoxide-substituted borane compound 3



To a CH<sub>2</sub>Cl<sub>2</sub> (10 mL) solution of **2** (0.40 g, 1.4 mmol) was added (C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>BOEt (0.71 g, 1.8 mmol) at 25 °C, and the mixture was stirred for 1 h. The solvent was evaporated. The crude product was separated by column chromatography (SiO<sub>2</sub>, eluent: hexane/ethyl acetate = 7:3, v/v) to give **3** as a yellow solid (0.78 g, 1.3 mmol, 94%).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.83 (brs, 1H, Ar*H*), 7.61 (dd, *J* = 9.6, 2.0 Hz, 1H, Ar*H*), 6.95 (dd, *J* = 9.6, 2.0 Hz, 1H, Ar*H*), 1.51 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>):  $\delta$  –134.92 (dd, *J* = 24.5, 9.4 Hz, 4F), -156.16 (t, *J* = 20.4 Hz, 2F), -163.15~-163.30 (m, 4F); <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  27.80 (s, C(CH<sub>3</sub>)<sub>3</sub>), 62.02 (s, C(CH<sub>3</sub>)<sub>3</sub>), 105.69 (s, C(Ar)), 108.94 (s, C(Ar)), 116.12 (br, CB), 135.70~138.56 (m, *C*F) 139.06~141.83 (m, *C*F), 139.39 (brt, *J* = 3.4 Hz, *C*(Ar)), 143.20 (s, *C*(Ar)), 146.82-149.51 (m, *C*F), 151.59 (s, *C*(Ar)); HRMS (APCI): *m*/*z* calcd for C<sub>21</sub>H<sub>13</sub>B<sup>79</sup>BrF<sub>10</sub>N<sub>2</sub>O 589.0143, C<sub>21</sub>H<sub>13</sub>B<sup>81</sup>BrF<sub>10</sub>N<sub>2</sub>O 591.0125. Found: 589.0143, 591.0124 ([M+H]<sup>+</sup>).

## <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) of **3**



<sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) of **3** 



## <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>) of **3**



### HRMS (APCI) of 3



### Aminoxide-substituted borane compound 4a



A mixture of **3** (0.30 g, 0.51 mmol), bis(4-*tert*-butylphenyl)amine (0.15 g, 0.53 mmol), Pd(OAc)<sub>2</sub> (10 mg, 0.05 mmol),  $[(t-Bu_3)PH]BF_4$  (20 mg, 0.080 mmol), *t*-BuONa(0.15 g, 1.5 mmol), and toluene (20 mL) was stirred at 100 °C for 20 h. The reaction was quenched with aq. NH<sub>4</sub>Cl, and the mixture was extracted with CHCl<sub>3</sub>. The organic layer was dried with MgSO<sub>4</sub>, and the solvents were evaporated. The crude product was separated by column chromatography (SiO<sub>2</sub>, eluent: hexane/ethyl acetate = 8:2,

v/v) to give **4a** as a yellow solid (0.32 g, 0.40 mmol, 79%). Compound **4a** was stable in the solid state but was gradually oxidized in air-saturated solutions to give purple-coloured and NMR-silent materials, which may contain the corresponding radical and oxoammonium species.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.52-7.49 (m, 2H, Ar*H*), 7.26 (d, *J* = 8.7 Hz, 4H, Ar*H*), 7.04 (d, *J* = 9.2 Hz, 1H, Ar*H*), 6.93 (d, *J* = 8.7 Hz, 4H, Ar*H*), 1.47 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 1.29 (s, 18H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>19</sup>F NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  -134.70 (dd, *J* = 24.4, 9.5 Hz, 4F), -156.97 (t, *J* = 20.1 Hz, 2F), -163.71~-163.85 (m, 4F); <sup>13</sup>C{<sup>1</sup>H} (100 MHz, CDCl<sub>3</sub>):  $\delta$  27.74 (s, C(CH<sub>3</sub>)<sub>3</sub>), 31.30 (s, C(CH<sub>3</sub>)<sub>3</sub>), 34.33 (s, C(CH<sub>3</sub>)<sub>3</sub>), 61.48 (s, C(CH<sub>3</sub>)<sub>3</sub>), 109.05 (s, C(Ar)), 122.98 (s, C(Ar)), 126.46 (s, C(Ar)), 133.16 (t, *J* = 6.7 Hz, *C*(Ar)), 135.60~138.43 (m, *C*F), 137.73 (s, *C*(Ar)), 137.80 (s, *C*(Ar)), 138.78~141.68 (m, *C*F), 143.49 (s, *C*(Ar)), 146.89 (s, *C*(Ar)), 146.95~149.51 (m, *C*F), 150.71 (s, *C*(Ar)) (the *C*B signal could not be observed probably because of the signal broadening); <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  7.54-7.51 (m, 2H, Ar*H*), 7.28 (d, *J* = 6.7 Hz, 4H, Ar*H*), 7.04 (d, *J* = 8.4 Hz, 1H, Ar*H*), 6.94 (d, *J* = 6.7 Hz, 4H, Ar*H*), 1.47 (s, 9H, C(*CH*<sub>3</sub>)<sub>3</sub>), 1.29 (s, 18H, C(*CH*<sub>3</sub>)<sub>3</sub>); <sup>19</sup>F NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta$  -135.07 (dd, *J* = 24.6, 10.0 Hz, 4F), -158.00 (t, *J* = 20.5 Hz, 2F), -164.52~-164.66 (m, 4F); UV-vis (CH<sub>2</sub>Cl<sub>2</sub>, rt):  $\lambda_{max} 424$  nm ( $\varepsilon 2.4 \times 10^3$  M<sup>-1</sup> cm<sup>-1</sup>); HRMS (APCI): m/z calcd for C<sub>41</sub>H<sub>39</sub>BF<sub>10</sub>N<sub>3</sub>O 790.3028. Found: 790.3027 ([M+H]<sup>+</sup>).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) of **4a** 



## <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) of 4a







# $^1\mathrm{H}$ NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>) of 4a



## HRMS (APCI) of 4a



Aminoxide-substituted borane compound 4b



A mixture of **3** (0.15 g, 0.25 mmol), bis(4-methoxyphenyl)amine (0.07 g, 0.28 mmol), Pd(OAc)<sub>2</sub> (7 mg, 0.03 mmol), [(*t*-Bu<sub>3</sub>)PH]BF<sub>4</sub> (0.012 g, 0.04 mmol), *t*-BuONa (0.07 g, 0.75 mmol), and toluene (5 mL) was stirred at 100 °C for 20 h. The reaction was quenched with aq. NH<sub>4</sub>Cl, and the mixture was extracted with CHCl<sub>3</sub>. The organic layer was dried with MgSO<sub>4</sub>, and the solvents were evaporated. The crude product was separated by column chromatography (SiO<sub>2</sub>, eluent: hexane/ethyl acetate = 8:2, v/v) to give **4b** as a yellow solid (0.11 g, 0.16 mmol, 62%). Compound **4b** was gradually oxidized under air even in the solid state and was rapidly oxidized in air-saturated solutions to give green-coloured and NMR-silent materials, which may contain the corresponding radical and oxoammonium species.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.46 (br, 1H, Ar*H*), 7.40 (dd, *J* = 9.6, 2.5 Hz, 1H, Ar*H*), 7.02 (dd, *J* = 1.9, 8.8 Hz, 1H, Ar*H*), 6.95 (d, *J* = 9.0 Hz, 4H, Ar*H*), 6.93~6.98 (m, 1H, Ar*H*), 6.81 (d, *J* = 9.0 Hz, 4H, Ar*H*), 3.78 (s, 6H, OC*H*<sub>3</sub>), 1.46 (s, 9H, C(C*H*<sub>3</sub>)<sub>3</sub>); <sup>19</sup>F NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  –135.02 (dd, *J* = 24.5, 9.4 Hz, 4F), -158.02 (t, *J* = 20.2 Hz, 2F), -164.55~-164.70 (m, 4F); <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  27.68 (s, C(CH<sub>3</sub>)<sub>3</sub>), 55.49 (s, C(CH<sub>3</sub>)<sub>3</sub>), 61.43 (s, OCH<sub>3</sub>), 109.24 (s, C(Ar)), 115.62 (s, C(Ar)), 125.50 (s, *C*(Ar)), 130.39 (t, *J* = 9.5 Hz, *C*(Ar)), 135.63 (s, *C*(Ar)), 135.73~138.45 (m, *C*F), 138.82 (s, *C*(Ar)), 139.24 (s, *C*(Ar)), 138.95~141.62 (m, *C*F), 146.83~149.53 (m, *C*F), 150.34 (s, *C*(Ar)), 156.44 (s, *C*(Ar)) (the *C*B signal could not be observed probably because of the signal broadening); UV-vis (CH<sub>2</sub>Cl<sub>2</sub>, rt):  $\lambda_{max}$  427 nm ( $\varepsilon$  2.3×10<sup>3</sup> M<sup>-1</sup> cm<sup>-1</sup>); HRMS (APCI): *m*/*z* calcd for C<sub>35</sub>H<sub>27</sub>BF<sub>10</sub>N<sub>3</sub>O<sub>3</sub> 738.1986. Found: 738.1984 ([M+H]<sup>+</sup>).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) of **4b** 



# <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) of **4b**



### HRMS (APCI) of 4b



Nitroxide radical-boranylium ion complex [5a]SbCl6



A mixture of **4a** (0.10 g, 0.13 mmol), [(4-BrC<sub>6</sub>H<sub>4</sub>)<sub>3</sub>N]SbCl<sub>6</sub> (0.13 g, 0.15 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was stirred at 25 °C for 1 h, and the solvent was evaporated. The residue was washed with Et<sub>2</sub>O (2 times) and re-crystalized from CH<sub>2</sub>Cl<sub>2</sub> at -10 °C to give [**5a**]SbCl<sub>6</sub> as a dark purple solid (0.095 g, 0.085 mmol). UV-vis-NIR (CH<sub>2</sub>Cl<sub>2</sub>, rt):  $\lambda_{max}$  1052 nm ( $\varepsilon$  1.7×10<sup>4</sup> M<sup>-1</sup> cm<sup>-1</sup>); ESR (X-band, CH<sub>2</sub>Cl<sub>2</sub>, rt): *g* 2.0037; HRMS (APCI): *m/z* calcd for C<sub>41</sub>H<sub>39</sub>BF<sub>10</sub>N<sub>3</sub>O 790.3028. Found: 790.3024 ([**5a**+H]<sup>+</sup>). Anal. calcd for C<sub>41</sub>H<sub>38</sub>N<sub>3</sub>BOCl<sub>6</sub>Sb: C, 43.81; H, 3.41; N, 3.74. Found: C, 43.99; H, 3.63; N, 3.84.

### HRMS (APCI) of [5a]SbCl6



Nitroxide radical-boranylium ion complex [5b]SbCl6



A mixture of **4b** (0.10 g, 0.12 mmol), [(4-BrC<sub>6</sub>H<sub>4</sub>)<sub>3</sub>N]SbCl<sub>6</sub> (0.11 g, 0.13 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was stirred at 25 °C for 1 h, and the solvent was evaporated. The residue was washed with hexane and re-crystallized from CH<sub>2</sub>Cl<sub>2</sub> at -10 °C to give [**5b**]SbCl<sub>6</sub> as a dark green solid (0.11 g, 0.10 mmol). UV-vis-NIR (CH<sub>2</sub>Cl<sub>2</sub>, rt):  $\lambda_{max}$  1081 nm ( $\varepsilon$  1.6×10<sup>4</sup> M<sup>-1</sup> cm<sup>-1</sup>); ESR (X-band, CH<sub>2</sub>Cl<sub>2</sub>, rt): *g* 2.0035; HRMS (APCI): *m*/*z* calcd for C<sub>35</sub>H<sub>27</sub>BF<sub>10</sub>N<sub>3</sub>O<sub>3</sub> 738.1986. Found: 738.1984 ([**5b**+H]<sup>+</sup>). Anal. calcd for C<sub>35</sub>H<sub>26</sub>N<sub>3</sub>BO<sub>3</sub>Cl<sub>6</sub>Sb: C, 39.22; H, 2.45; N, 3.92. Found: C, 39.34; H, 2.65; N, 4.00.

### HRMS (APCI) of [5b]SbCl6



| XRD data                                      |                                    |                                |
|---|------------------------------------|--------------------------------|
| Compound <b>4a</b>                            |                                    |                                |
| Empirical formula                             | $C_{41}H_{38}BF_{10}N_3O$          |                                |
| Formula weight                                | 789.55                             |                                |
| Temperature                                   | −173 °C                            |                                |
| Crystal system                                | Triclinic                          |                                |
| Space group                                   | <i>P</i> -1                        |                                |
| Unit cell dimensions                          | a = 9.8688(5) Å                    | <i>α</i> = 95.144(2)°.         |
|   | <i>b</i> = 10.8972(6) Å            | β=100.083(2)°.                 |
|   | c = 18.0044(9) Å                   | $\gamma = 96.151(2)^{\circ}$ . |
| Volume  | 1883.69(17) Å <sup>3</sup>         |                                |
| Z   | 2                                  |                                |
| Density (calculated)                          | $1.392 \text{ g cm}^{-3}$          |                                |
| Absorption coefficient                        | 0.118 mm <sup>-1</sup>             |                                |
| <i>F</i> (000)                                | 816                                |                                |
| Crystal size                                  | 0.220×0.200×0.150 mm <sup>3</sup>  |                                |
| $\theta$ range for data collection            | 1.891 to 27.499°.                  |                                |
| Index ranges                                  | -12<=h<=12, -14<=k<=14, -23<=l<=23 |                                |
| Reflections collected                         | 39509                              |                                |
| Independent reflections                       | 8650 [ <i>R</i> (int) = 0.0718]    |                                |
| Completeness to $\theta = 25.242^{\circ}$     | 99.9 %                             |                                |
| Absorption correction                         | Semi-empirical from equiva         | alents                         |
| Max. and min. transmission                    | 1.000 and 0.910                    |                                |
| Refinement method                             | Full-matrix least-squares or       | n F <sup>2</sup>               |
| Data / restraints / parameters                | 8650 / 0 / 514                     |                                |
| Goodness-of-fit on $F^2$                      | 1.025                              |                                |
| Final <i>R</i> indices [ $I \ge 2\sigma(I)$ ] | $R_1 = 0.0538, wR_2 = 0.1436$      |                                |
| R indices (all data)                          | $R_1 = 0.0689, wR_2 = 0.1594$      |                                |
| Largest diff. peak and hole                   | 0.585 and –0.408 e Å <sup>-3</sup> |                                |
| CCDC No.                                      | 2182062                            |                                |

| Compound [ <b>5a</b> ]SbCl6                 |   |  |
|---|---|--|
| Empirical formula                           | $C_{43}H_{42}BCl_{10}F_{10}N_3OSb$              |  |
| Formula weight                              | 1293.85   |  |
| Temperature                                 | –100 °C   |  |
| Crystal system                              | Monoclinic                                      |  |
| Space group                                 | $P2_{1}/c$                                      |  |
| Unit cell dimensions                        | a = 21.093(7) Å                                 |  |
|   | $b = 11.908(5)$ Å $\beta = 108.42(5)^{\circ}$ . |  |
|   | c = 23.298 (14)  Å                              |  |
| Volume                                      | 5552(5) Å <sup>3</sup>                          |  |
| Z   | 4   |  |
| Density (calculated)                        | 1.548 g cm <sup>-3</sup>                        |  |
| Absorption coefficient                      | 1.047 mm <sup>-1</sup>                          |  |
| <i>F</i> (000)                              | 2580.0  |  |
| Crystal size                                | 0.36×0.26×0.23 mm <sup>3</sup>                  |  |
| $\theta$ range for data collection          | 5.846 to 59.528°.                               |  |
| Index ranges                                | -28<=h<=27, -15<=k<=13, -32<=l<=32              |  |
| Reflections collected                       | 46335   |  |
| Independent reflections                     | 14171 [ $R(int) = 0.0233$ ]                     |  |
| Completeness to $\theta = 25.242^{\circ}$   | 99.8%   |  |
| Absorption correction                       | Semi-empirical from equivalents                 |  |
| Max. and min. transmission                  | 1.000 and 0.91904                               |  |
| Refinement method                           | Full-matrix least-squares on $F^2$              |  |
| Data / restraints / parameters              | 14171/210/712                                   |  |
| Goodness-of-fit on $F^2$                    | 1.044   |  |
| Final <i>R</i> indices $[I \ge 2\sigma(I)]$ | $R_1 = 0.0516$ , w $R_2 = 0.1269$               |  |
| R indices (all data)                        | $R_1 = 0.0757, wR_2 = 0.1383$                   |  |
| Largest diff. peak and hole                 | 1.307 and -0.850 e Å <sup>-3</sup>              |  |
| CCDC No                                     | 2182063   |  |
|   |   |  |



Fig. S1 Structure of [5a]SbCl<sub>6</sub> (50% probability level). Hydrogen atoms are omitted.

| Compound [5b]SbCl6                          |                                    |                                |
|---|------------------------------------|--------------------------------|
| Empirical formula                           | $C_{35}H_{26}BF_{10}N_3O_3SbCl_6$  |                                |
| Formula weight                              | 1071.85                            |                                |
| Temperature                                 | −100 °C                            |                                |
| Crystal system                              | Monoclinic                         |                                |
| Space group                                 | <i>P</i> 2 <sub>1</sub> /n         |                                |
| Unit cell dimensions                        | <i>a</i> = 18.0480(11) Å           |                                |
|   | b = 10.6917(7) Å                   | $\beta = 102.647(6)^{\circ}$ . |
|   | <i>c</i> = 21.8391(12) Å           |                                |
| Volume                                      | 4111.9(4) Å <sup>3</sup>           |                                |
| Z   | 4                                  |                                |
| Density (calculated)                        | 1.731 g cm <sup>-3</sup>           |                                |
| Absorption coefficient                      | 1.146 mm <sup>-1</sup>             |                                |
| <i>F</i> (000)                              | 2116                               |                                |
| Crystal size                                | 0.070×0.070×0.050 mm <sup>3</sup>  |                                |
| $\theta$ range for data collection          | 2.767 to 27.481°.                  |                                |
| Index ranges                                | -23<=h<=22, -13<=k<=13, -25<=l<=28 |                                |
| Reflections collected                       | 34211                              |                                |
| Independent reflections                     | 9429 [ <i>R</i> (int) = 0.0601]    |                                |
| Completeness to $\theta = 25.242^{\circ}$   | 99.9 %                             |                                |
| Absorption correction                       | Semi-empirical from equiva         | lents                          |
| Max. and min. transmission                  | 1.0000 and 0.51931                 |                                |
| Refinement method                           | Full-matrix least-squares on       | $F^2$                          |
| Data / restraints / parameters              | 9429 / 0 / 537                     |                                |
| Goodness-of-fit on $F^2$                    | 1.013                              |                                |
| Final <i>R</i> indices $[I \ge 2\sigma(I)]$ | $R_1 = 0.0513, wR_2 = 0.0841$      |                                |
| R indices (all data)                        | $R_1 = 0.0967, wR_2 = 0.0949$      |                                |
| Largest diff. peak and hole                 | 0.662 and -0.487 e Å <sup>-3</sup> |                                |
| CCDC No                                     | 2182064                            |                                |
|   |                                    |                                |



Fig. S2 Structure of [5b]SbCl<sub>6</sub> (50% probability level). Hydrogen atoms are omitted.





Fig. S3 ESR spectrum of [5a]SbCl<sub>6</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -150 °C.



Fig. S4 ESR spectrum of [5b]SbCl<sub>6</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -150 °C.



Fig. S5 ESR spectrum of [5a]SbCl<sub>6</sub> in the solid state at 25 °C.



Fig. S6 ESR spectrum of [5b]SbCl<sub>6</sub> in the solid state at 25 °C.



Fig. S7 ESR spectrum of [5a]SbCl<sub>6</sub> in the solid state at -150 °C.



Fig. S8 ESR spectrum of [5b]SbCl<sub>6</sub> in the solid state at -150 °C.

## Computational data Cartesian coordinates of **4a**

| С       | -0.6126  | -2 31438 | -1 46247 |
|---------|----------|----------|----------|
| C       | 0.303218 | -0.48229 | -0 28994 |
| C       | 0.616315 | -2 50368 | -2 12574 |
| C       | 1 518538 | -0 59159 | -0.92914 |
| н       | 0.091863 | 0.301294 | 0.423635 |
| n<br>C  | 1 653593 | -1 64324 | -1.86    |
| н       | 0.736159 | -3 28518 | -2.85817 |
| н       | 2 596197 | 1 76561  | 2.00017  |
| N       | 0.71670  | 1 30036  | 0.57265  |
| IN<br>N | 1 91447  | -1.30930 | -0.37203 |
| N O     | -1.01447 | -2.93433 | -1.03103 |
| 0       | -2.00043 | -2.33303 | -0.38288 |
| C       | -2.06543 | -4.35846 | -1.96/24 |
| C       | -1.35/81 | -4.72927 | -3.27486 |
| C       | -1.59997 | -5.25381 | -0.81498 |
| С       | -3.57717 | -4.4899  | -2.18434 |
| Н       | -1.49046 | -3.95122 | -4.03211 |
| Н       | -0.29244 | -4.92659 | -3.13958 |
| Н       | -1.79971 | -5.65152 | -3.66054 |
| Н       | -2.08249 | -4.9542  | 0.116963 |
| Н       | -1.85003 | -6.29869 | -1.0221  |
| Н       | -0.51596 | -5.18424 | -0.67703 |
| Н       | -3.80774 | -5.51564 | -2.48544 |
| Н       | -4.12835 | -4.25616 | -1.27462 |
| Н       | -3.9105  | -3.806   | -2.96904 |
| В       | -2.23247 | -1.22891 | -0.06974 |
| С       | -2.34266 | -1.26672 | 1.549153 |
| С       | -3.18666 | -0.4901  | 2.334054 |
| С       | -1.59163 | -2.20585 | 2.25197  |
| С       | -3.26959 | -0.61176 | 3.714977 |
| С       | -1.64235 | -2.36071 | 3.627377 |
| С       | -2.49072 | -1.55192 | 4.366388 |
| С       | -2.90036 | 0.066663 | -0.82313 |
| С       | -3.77086 | -0.02573 | -1.90231 |

| С | -2.56574 | 1.364952 | -0.45565 |
|---|----------|----------|----------|
| С | -4.27261 | 1.086717 | -2.57026 |
| С | -3.03786 | 2.497582 | -1.09215 |
| С | -3.903   | 2.356668 | -2.16586 |
| F | -4.17927 | -1.21218 | -2.37292 |
| F | -1.73591 | 1.571334 | 0.59009  |
| F | -3.99207 | 0.428461 | 1.782881 |
| F | -0.756   | -3.03204 | 1.594234 |
| F | -0.88961 | -3.27621 | 4.241627 |
| F | -2.55855 | -1.68236 | 5.689397 |
| F | -4.09817 | 0.164255 | 4.415466 |
| F | -2.66961 | 3.720213 | -0.68671 |
| F | -4.36912 | 3.43138  | -2.79971 |
| F | -5.10334 | 0.93638  | -3.60407 |
| Ν | 2.53252  | 0.358146 | -0.7135  |
| С | 2.117345 | 1.695515 | -0.43355 |
| С | 3.877505 | -0.03353 | -0.52109 |
| С | 2.426053 | 2.312663 | 0.776866 |
| С | 1.321969 | 2.375524 | -1.35089 |
| С | 4.914148 | 0.832889 | -0.88161 |
| С | 4.210997 | -1.27283 | 0.020782 |
| С | 1.915114 | 3.570188 | 1.064381 |
| Н | 3.050047 | 1.795656 | 1.498847 |
| С | 0.801444 | 3.625383 | -1.04258 |
| Н | 1.08789  | 1.907507 | -2.30201 |
| С | 6.234033 | 0.459232 | -0.6984  |
| Н | 4.675985 | 1.802067 | -1.30652 |
| С | 5.543907 | -1.63956 | 0.180246 |
| Н | 3.428092 | -1.95618 | 0.332653 |
| С | 1.074552 | 4.249488 | 0.175694 |
| Н | 2.154938 | 4.01383  | 2.025243 |
| Н | 0.166948 | 4.111865 | -1.77302 |
| С | 6.589677 | -0.78765 | -0.16902 |

| н | 7.009075 | 1 161441 | -0.99003 |
|---|----------|----------|----------|
|   | 1.007075 | 1.101771 | -0.77003 |
| Н | 5.753104 | -2.61405 | 0.60543  |
| С | 8.066655 | -1.15657 | 0.003307 |
| С | 8.248    | -2.55656 | 0.601175 |
| С | 8.766841 | -1.12533 | -1.36706 |
| С | 8.743088 | -0.14238 | 0.942936 |
| Н | 7.796085 | -2.63642 | 1.594971 |
| Н | 7.812077 | -3.33158 | -0.03736 |
| Н | 9.314906 | -2.77689 | 0.705433 |
| Н | 8.707965 | -0.13673 | -1.83148 |
| Н | 9.826884 | -1.37987 | -1.25946 |
| Н | 8.312428 | -1.84543 | -2.05502 |
| Н | 9.802465 | -0.38977 | 1.071704 |
| Н | 8.685221 | 0.877023 | 0.550624 |

| Н | 8.270003 | -0.14793 | 1.929996 |
|---|----------|----------|----------|
| С | 0.437234 | 5.582431 | 0.579651 |
| С | -0.30225 | 6.247897 | -0.58813 |
| С | -0.58148 | 5.301863 | 1.700833 |
| С | 1.509882 | 6.561337 | 1.085352 |
| Н | 0.365956 | 6.439795 | -1.43438 |
| Н | -1.14069 | 5.639898 | -0.93755 |
| Н | -0.70892 | 7.210125 | -0.26184 |
| Н | -0.09643 | 4.878141 | 2.585745 |
| Н | -1.08373 | 6.227567 | 2.002898 |
| Н | -1.34073 | 4.592025 | 1.359657 |
| Н | 1.045092 | 7.50917  | 1.376425 |
| Н | 2.043491 | 6.177319 | 1.959176 |
| Н | 2.249246 | 6.772038 | 0.305796 |



**Fig. S9** DFT-optimized structure of **4a**. Selected bond lengths (Å): N1-O1 1.410, C1-N1 1.363, O1-B1 1.467, N2-B1 1.599, C2-N3 1.406.

| Cartesian coordinates of <b>4b</b> |          |          |          |  |
|------------------------------------|----------|----------|----------|--|
| С                                  | -0.4726  | -2.22231 | -1.0556  |  |
| С                                  | 0.583693 | -0.33992 | -0.09987 |  |
| С                                  | 0.744836 | -2.60473 | -1.64724 |  |
| С                                  | 1.800954 | -0.64012 | -0.67694 |  |

| Н | 0.430763 | 0.538791 | 0.508387 |
|---|----------|----------|----------|
| С | 1.851289 | -1.81079 | -1.46425 |
| Н | 0.80879  | -3.48488 | -2.26631 |
| Н | 2.786996 | -2.08693 | -1.93741 |

| Ν | -0.50429 | -1.09704 | -0.32016 |
|---|----------|----------|----------|
| Ν | -1.72603 | -2.76026 | -1.17734 |
| 0 | -2.54545 | -2.14943 | -0.20165 |
| С | -2.0616  | -4.20424 | -1.2764  |
| С | -1.41046 | -4.81199 | -2.5238  |
| С | -1.61528 | -4.93338 | -0.00539 |
| С | -3.58304 | -4.28382 | -1.44417 |
| Н | -1.52289 | -4.15158 | -3.38836 |
| Н | -0.35332 | -5.0453  | -2.38183 |
| Н | -1.90987 | -5.75649 | -2.75464 |
| Н | -2.03852 | -4.44897 | 0.87666  |
| Н | -1.94552 | -5.97626 | -0.02883 |
| Н | -0.52428 | -4.92573 | 0.089955 |
| Н | -3.87278 | -5.32903 | -1.58438 |
| Н | -4.09827 | -3.88836 | -0.57003 |
| Н | -3.90083 | -3.70658 | -2.31586 |
| В | -2.03534 | -0.80409 | 0.063105 |
| С | -2.21797 | -0.48611 | 1.640359 |
| С | -2.98423 | 0.536749 | 2.186487 |
| С | -1.61337 | -1.32756 | 2.571406 |
| С | -3.13506 | 0.727666 | 3.553877 |
| С | -1.73693 | -1.17274 | 3.942254 |
| С | -2.50617 | -0.13191 | 4.437345 |
| С | -2.55271 | 0.337579 | -0.99615 |
| С | -3.4369  | 0.093263 | -2.03976 |
| С | -2.05223 | 1.633077 | -0.95098 |
| С | -3.78339 | 1.05816  | -2.98028 |
| С | -2.35789 | 2.620327 | -1.8696  |
| С | -3.238   | 2.326761 | -2.89994 |
| F | -4.01352 | -1.10499 | -2.20546 |
| F | -1.20737 | 1.983604 | 0.042839 |
| F | -3.64072 | 1.402314 | 1.40252  |
| F | -0.85432 | -2.35897 | 2.157827 |
| F | -1.12622 | -2.01008 | 4.783966 |
| F | -2.64012 | 0.036914 | 5.751095 |

| F | -3.88397 | 1.728033 | 4.020674 |
|---|----------|----------|----------|
| F | -1.81373 | 3.838745 | -1.77833 |
| F | -3.55269 | 3.255722 | -3.80121 |
| F | -4.63415 | 0.764855 | -3.9664  |
| Ν | 2.909276 | 0.20418  | -0.5338  |
| С | 2.715475 | 1.600544 | -0.31922 |
| С | 4.220231 | -0.3425  | -0.46656 |
| С | 3.415991 | 2.274037 | 0.674302 |
| С | 1.806351 | 2.320432 | -1.10404 |
| С | 5.257105 | 0.22762  | -1.21419 |
| С | 4.502549 | -1.44181 | 0.33716  |
| С | 3.226793 | 3.638014 | 0.88469  |
| Н | 4.123664 | 1.731064 | 1.291664 |
| С | 1.58223  | 3.663316 | -0.87581 |
| Н | 1.254372 | 1.812702 | -1.88811 |
| С | 6.53608  | -0.28947 | -1.15055 |
| Н | 5.047673 | 1.087136 | -1.84216 |
| С | 5.784419 | -1.98506 | 0.387591 |
| Н | 3.71211  | -1.88463 | 0.934909 |
| С | 2.299157 | 4.337315 | 0.117195 |
| Н | 3.79508  | 4.133035 | 1.662312 |
| Н | 0.844804 | 4.209939 | -1.45271 |
| С | 6.810363 | -1.40516 | -0.35351 |
| Н | 7.346753 | 0.147083 | -1.72355 |
| Н | 5.967764 | -2.84465 | 1.020228 |
| 0 | 8.096087 | -1.8432  | -0.37084 |
| 0 | 2.021669 | 5.659563 | 0.254247 |
| С | 2.692547 | 6.378516 | 1.265266 |
| Н | 2.317088 | 7.400503 | 1.20974  |
| Н | 2.478879 | 5.968813 | 2.25997  |
| Н | 3.777668 | 6.383627 | 1.104306 |
| С | 8.429511 | -2.95763 | 0.427603 |
| Н | 9.493278 | -3.13599 | 0.269373 |
| Н | 8.250931 | -2.75877 | 1.491272 |
| Н | 7.86571  | -3.84998 | 0.129392 |



**Fig. S10** DFT-optimized structure of **4b**. Selected bond lengths (Å): N1-O1 1.413, C1-N1 1.369, O1-B1 1.463, N2-B1 1.605, C2-N3 1.401.

|   | e        |          |          |
|---|----------|----------|----------|
| С | -0.58405 | -2.48737 | -1.02083 |
| С | 0.243037 | -0.50272 | -0.07717 |
| С | 0.716039 | -2.87456 | -1.42014 |
| С | 1.561434 | -0.80166 | -0.4753  |
| Н | -0.00656 | 0.405447 | 0.454014 |
| С | 1.766029 | -2.04774 | -1.13683 |
| Н | 0.879    | -3.79879 | -1.95433 |
| Н | 2.765774 | -2.32467 | -1.44732 |
| Ν | -0.75759 | -1.31522 | -0.3512  |
| Ν | -1.76865 | -3.09415 | -1.18555 |
| 0 | -2.77909 | -2.42614 | -0.61799 |
| С | -2.11947 | -4.3688  | -1.89773 |
| С | -1.74504 | -4.19339 | -3.3738  |
| С | -1.37486 | -5.52502 | -1.2223  |
| С | -3.62832 | -4.57755 | -1.77073 |
| Н | -2.26874 | -3.33154 | -3.79539 |
| Н | -0.67134 | -4.06241 | -3.52853 |
| Н | -2.05095 | -5.0859  | -3.92488 |
| Н | -1.60764 | -5.56105 | -0.1549  |

| Cartesian | со | oordinates | of | 5a |
|-----------|----|------------|----|----|
|           |    |            |    |    |

| Н | -1.7016  | -6.46389 | -1.67566 |
|---|----------|----------|----------|
| Н | -0.29078 | -5.4661  | -1.33947 |
| Н | -3.88156 | -5.49256 | -2.31111 |
| Н | -3.93392 | -4.69525 | -0.72987 |
| Н | -4.18495 | -3.74734 | -2.20644 |
| В | -2.32387 | -1.08449 | -0.08488 |
| С | -2.66978 | -1.03308 | 1.48201  |
| С | -3.56708 | -0.16205 | 2.091082 |
| С | -2.09695 | -1.97993 | 2.327222 |
| С | -3.85413 | -0.20459 | 3.449035 |
| С | -2.35352 | -2.05872 | 3.683814 |
| С | -3.24356 | -1.15674 | 4.249633 |
| С | -2.77888 | 0.140532 | -1.03988 |
| С | -3.40444 | -0.00756 | -2.27222 |
| С | -2.46067 | 1.450359 | -0.69542 |
| С | -3.726   | 1.067169 | -3.09299 |
| С | -2.7525  | 2.546488 | -1.48429 |
| С | -3.39939 | 2.353506 | -2.69687 |
| F | -3.71328 | -1.22427 | -2.75414 |

| F | -1.80994 | 1.692733 | 0.46286  |
|---|----------|----------|----------|
| F | -4.21119 | 0.767837 | 1.378914 |
| F | -1.23875 | -2.88964 | 1.816232 |
| F | -1.76486 | -2.98367 | 4.43719  |
| F | -3.50914 | -1.2093  | 5.546183 |
| F | -4.71405 | 0.656285 | 3.981575 |
| F | -2.40457 | 3.77594  | -1.09832 |
| F | -3.68378 | 3.388393 | -3.47397 |
| F | -4.32519 | 0.863244 | -4.26117 |
| Ν | 2.551986 | 0.106493 | -0.24673 |
| С | 2.184713 | 1.484322 | -0.07914 |
| С | 3.928544 | -0.2629  | -0.17555 |
| С | 2.484988 | 2.159257 | 1.100371 |
| С | 1.473433 | 2.128347 | -1.08743 |
| С | 4.887315 | 0.518227 | -0.8209  |
| С | 4.327619 | -1.37424 | 0.561622 |
| С | 2.028172 | 3.456342 | 1.27457  |
| Н | 3.048439 | 1.661435 | 1.882794 |
| С | 1.025681 | 3.427836 | -0.89573 |
| Н | 1.268717 | 1.611428 | -2.02021 |
| С | 6.223432 | 0.169995 | -0.73366 |
| Н | 4.582293 | 1.390024 | -1.38945 |
| С | 5.673789 | -1.71075 | 0.632167 |
| Н | 3.593609 | -1.96344 | 1.102764 |
| С | 1.274767 | 4.117013 | 0.293877 |
| Н | 2.247427 | 3.955408 | 2.211472 |
| Н | 0.468021 | 3.903606 | -1.692   |
| С | 6.653919 | -0.95165 | -0.0108  |

| Н | 6.950132 | 0.789661 | -1.24751 |
|---|----------|----------|----------|
| Н | 5.954247 | -2.57602 | 1.219402 |
| С | 8.145192 | -1.2915  | 0.054622 |
| С | 8.417637 | -2.54698 | 0.891026 |
| С | 8.6716   | -1.53513 | -1.37183 |
| С | 8.906219 | -0.1121  | 0.687822 |
| Н | 8.09538  | -2.4254  | 1.930098 |
| Н | 7.924097 | -3.43156 | 0.475612 |
| Н | 9.492089 | -2.74932 | 0.904659 |
| Н | 8.549372 | -0.65604 | -2.01105 |
| Н | 9.739005 | -1.77427 | -1.34041 |
| Н | 8.151793 | -2.37309 | -1.84701 |
| Н | 9.975189 | -0.34098 | 0.737862 |
| Н | 8.792769 | 0.808774 | 0.108703 |
| Н | 8.554446 | 0.083855 | 1.70548  |
| С | 0.693408 | 5.506387 | 0.5662   |
| С | 0.145901 | 6.163214 | -0.70799 |
| С | -0.46753 | 5.332259 | 1.565557 |
| С | 1.761456 | 6.436763 | 1.164801 |
| Н | 0.916728 | 6.259088 | -1.47983 |
| Н | -0.70023 | 5.610402 | -1.12481 |
| Н | -0.21144 | 7.169324 | -0.47199 |
| Н | -0.12021 | 4.916905 | 2.516831 |
| Н | -0.9362  | 6.299543 | 1.772634 |
| Н | -1.23078 | 4.661206 | 1.159237 |
| Н | 1.330317 | 7.426551 | 1.341259 |
| Н | 2.143154 | 6.076817 | 2.124008 |
| Н | 2.610499 | 6.556072 | 0.484386 |



**Fig. S11** DFT-optimized structure of **5a**. Selected bond lengths (Å): N1-O1 1.338, C1-N1 1.341, O1-B1 1.514, N2-B1 1.605, C2-N3 1.363.

| С | -0.53966 | -2.35678 | -0.65342 |
|---|----------|----------|----------|
| С | 0.524992 | -0.3928  | 0.088976 |
| С | 0.705055 | -2.90643 | -1.06065 |
| С | 1.785194 | -0.8639  | -0.30668 |
| Н | 0.386516 | 0.576058 | 0.548199 |
| С | 1.836212 | -2.16812 | -0.88443 |
| Н | 0.754197 | -3.88226 | -1.51998 |
| Н | 2.790067 | -2.56613 | -1.20799 |
| Ν | -0.56807 | -1.10689 | -0.10539 |
| Ν | -1.77012 | -2.8562  | -0.71572 |
| 0 | -2.69445 | -2.03443 | -0.16505 |
| С | -2.26584 | -4.16658 | -1.25063 |
| С | -1.9619  | -4.20666 | -2.75252 |
| С | -1.58997 | -5.30343 | -0.47674 |
| С | -3.77777 | -4.22412 | -1.03093 |
| Н | -2.43662 | -3.36012 | -3.25419 |
| Н | -0.89094 | -4.18841 | -2.97024 |
| Н | -2.3673  | -5.12997 | -3.17355 |
| Н | -1.78229 | -5.20647 | 0.59491  |

| Cartesian | coordinates | of 5b |
|-----------|-------------|-------|
|-----------|-------------|-------|

| Н | -2.00795 | -6.25512 | -0.81372 |
|---|----------|----------|----------|
| Н | -0.50992 | -5.3485  | -0.63258 |
| Н | -4.13567 | -5.17202 | -1.43972 |
| Н | -4.0324  | -4.1861  | 0.029334 |
| Н | -4.28682 | -3.40654 | -1.54108 |
| В | -2.10903 | -0.67365 | 0.078537 |
| С | -2.43889 | -0.23732 | 1.585799 |
| С | -3.22009 | 0.847399 | 1.968428 |
| С | -1.96997 | -1.03066 | 2.629749 |
| С | -3.5035  | 1.140631 | 3.295712 |
| С | -2.22714 | -0.77505 | 3.964751 |
| С | -3.00399 | 0.325264 | 4.298627 |
| С | -2.41713 | 0.36499  | -1.13315 |
| С | -3.09519 | 0.045897 | -2.30313 |
| С | -1.89706 | 1.654063 | -1.0935  |
| С | -3.26532 | 0.943021 | -3.3517  |
| С | -2.02638 | 2.574003 | -2.11679 |
| C | -2.72684 | 2.215255 | -3.26015 |
| F | -3.60889 | -1.18132 | -2.50042 |

| F | -1.19541 | 2.050043 | -0.00833 |
|---|----------|----------|----------|
| F | -3.74792 | 1.667569 | 1.053658 |
| F | -1.21606 | -2.11432 | 2.351302 |
| F | -1.74152 | -1.56697 | 4.917811 |
| F | -3.26539 | 0.593524 | 5.570007 |
| F | -4.2519  | 2.193417 | 3.608251 |
| F | -1.46953 | 3.782941 | -2.02271 |
| F | -2.86214 | 3.076462 | -4.25875 |
| F | -3.92227 | 0.5767   | -4.44753 |
| Ν | 2.898428 | -0.07167 | -0.16519 |
| С | 2.772748 | 1.345359 | -0.13227 |
| С | 4.192932 | -0.65363 | -0.04473 |
| С | 3.492239 | 2.089014 | 0.803752 |
| С | 1.920285 | 2.006191 | -1.0308  |
| С | 5.254198 | -0.15072 | -0.79518 |
| С | 4.411067 | -1.71748 | 0.841021 |
| С | 3.35404  | 3.466336 | 0.86189  |
| Н | 4.150296 | 1.584061 | 1.502195 |
| C | 1.764895 | 3.372053 | -0.96593 |
| Н | 1.385978 | 1.441438 | -1.78764 |

| С | 6.520226 | -0.70418 | -0.68082 |
|---|----------|----------|----------|
| Н | 5.08564  | 0.672337 | -1.48108 |
| С | 5.666997 | -2.27188 | 0.95659  |
| Н | 3.596377 | -2.08972 | 1.453856 |
| С | 2.481429 | 4.119116 | -0.01778 |
| Н | 3.913466 | 4.022041 | 1.603217 |
| Н | 1.08799  | 3.890949 | -1.63434 |
| С | 6.735731 | -1.77322 | 0.195721 |
| Н | 7.328134 | -0.30407 | -1.2796  |
| Н | 5.861309 | -3.08623 | 1.644987 |
| 0 | 2.266047 | 5.440019 | -0.03628 |
| 0 | 7.91513  | -2.38149 | 0.383088 |
| С | 2.95138  | 6.26887  | 0.892246 |
| Н | 2.616438 | 7.283838 | 0.685157 |
| Н | 2.693777 | 6.00239  | 1.92249  |
| Н | 4.035332 | 6.205865 | 0.750977 |
| С | 9.051215 | -1.92553 | -0.33778 |
| Н | 9.87776  | -2.55793 | -0.01799 |
| Н | 8.903702 | -2.03594 | -1.41717 |
| Н | 9.276665 | -0.88115 | -0.09826 |



Fig. S12 DFT-optimized structure of 5b. Selected bond lengths (Å): N1-O1 1.354, C1-N1 1.329, O1-B1 1.501,

## N2-B1 1.611, C2-N3 1.374.

## Cartesian coordinates of 6a

| С | -0.6519  | -2.22026 | -1.44801 |
|---|----------|----------|----------|
| С | 0.240758 | -0.39521 | -0.25438 |
| С | 0.627378 | -2.53905 | -1.9918  |
| С | 1.541382 | -0.64502 | -0.77522 |
| Н | 0.045596 | 0.404045 | 0.449606 |
| С | 1.704834 | -1.77377 | -1.64043 |
| Н | 0.741041 | -3.35577 | -2.68999 |
| Н | 2.680778 | -1.99189 | -2.05885 |
| N | -0.78184 | -1.14983 | -0.58308 |
| N | -1.83437 | -2.81332 | -1.6111  |
| 0 | -2.81728 | -2.24046 | -0.92336 |
| С | -2.21966 | -4.05442 | -2.42265 |
| С | -1.95207 | -3.73329 | -3.90514 |
| С | -1.39073 | -5.23878 | -1.89176 |
| С | -3.71582 | -4.31067 | -2.20177 |
| Н | -2.53121 | -2.86102 | -4.22241 |
| Н | -0.8956  | -3.55893 | -4.1277  |
| Н | -2.27416 | -4.59062 | -4.50307 |
| Н | -1.52536 | -5.35513 | -0.81239 |
| Н | -1.74793 | -6.14975 | -2.38019 |
| Н | -0.32251 | -5.15217 | -2.10771 |
| Н | -3.98836 | -5.17775 | -2.80959 |
| Н | -3.9418  | -4.54051 | -1.15812 |
| Н | -4.32558 | -3.46192 | -2.51732 |
| В | -2.34706 | -1.00995 | -0.14302 |
| С | -2.56587 | -1.31552 | 1.417483 |
| С | -3.4433  | -0.62305 | 2.259908 |
| С | -1.92154 | -2.41554 | 2.000837 |
| С | -3.63795 | -0.97482 | 3.59611  |
| С | -2.08146 | -2.79856 | 3.325524 |
| С | -2.95412 | -2.06704 | 4.132095 |
| С | -2.91164 | 0.356595 | -0.79513 |

| С | -3.74506 | 0.427675 | -1.91581 |
|---|----------|----------|----------|
| С | -2.53319 | 1.589409 | -0.25668 |
| С | -4.19864 | 1.637264 | -2.44712 |
| С | -2.94774 | 2.813095 | -0.76034 |
| С | -3.79711 | 2.838521 | -1.86745 |
| F | -4.14412 | -0.68808 | -2.55428 |
| F | -1.70136 | 1.61647  | 0.820798 |
| F | -4.14099 | 0.428526 | 1.81052  |
| F | -1.08363 | -3.17005 | 1.239101 |
| F | -1.42606 | -3.84944 | 3.817438 |
| F | -3.13059 | -2.40713 | 5.399681 |
| F | -4.466   | -0.27271 | 4.36147  |
| F | -2.52818 | 3.958052 | -0.20492 |
| F | -4.20046 | 3.997267 | -2.3711  |
| F | -4.99129 | 1.645118 | -3.51454 |
| Ν | 2.562507 | 0.210847 | -0.43858 |
| С | 2.250929 | 1.573214 | -0.20073 |
| С | 3.899047 | -0.23261 | -0.32886 |
| С | 2.788178 | 2.246345 | 0.917076 |
| С | 1.369608 | 2.250179 | -1.07264 |
| С | 4.952508 | 0.600726 | -0.76804 |
| С | 4.191904 | -1.49107 | 0.237783 |
| С | 2.396096 | 3.547054 | 1.173433 |
| Н | 3.454082 | 1.726161 | 1.597869 |
| С | 1.010038 | 3.560492 | -0.80662 |
| Н | 1.022271 | 1.761952 | -1.97845 |
| С | 6.258393 | 0.165597 | -0.6499  |
| Н | 4.731853 | 1.555959 | -1.23269 |
| С | 5.509899 | -1.89982 | 0.356223 |
| Н | 3.394165 | -2.11114 | 0.635599 |
| С | 1.490399 | 4.244122 | 0.331786 |
| Н | 2.7803   | 4.033858 | 2.062679 |

| Н | 0.356953 | 4.06993  | -1.50371 |
|---|----------|----------|----------|
| С | 6.584145 | -1.09239 | -0.08463 |
| Н | 7.050671 | 0.808976 | -1.01596 |
| Н | 5.712647 | -2.85489 | 0.824773 |
| С | 8.043958 | -1.52357 | 0.033155 |
| С | 8.200995 | -2.91021 | 0.684535 |
| С | 8.665377 | -1.57082 | -1.39041 |
| С | 8.808373 | -0.48015 | 0.892444 |
| Н | 7.810614 | -2.933   | 1.708334 |
| Н | 7.710162 | -3.69869 | 0.102717 |
| Н | 9.263312 | -3.16445 | 0.739447 |
| Н | 8.637834 | -0.59788 | -1.89111 |
| Н | 9.714892 | -1.8719  | -1.31232 |
| Н | 8.151318 | -2.29987 | -2.02588 |
| Н | 9.857802 | -0.78121 | 0.972139 |

| Н | 8.786349 | 0.521703 | 0.451914 |
|---|----------|----------|----------|
| Н | 8.396442 | -0.41987 | 1.905425 |
| С | 1.030527 | 5.653673 | 0.691935 |
| С | 0.215312 | 6.314626 | -0.43675 |
| С | 0.121352 | 5.52459  | 1.949735 |
| С | 2.247747 | 6.554045 | 1.019415 |
| Н | 0.792767 | 6.399928 | -1.36454 |
| Н | -0.71471 | 5.775668 | -0.64499 |
| Н | -0.06269 | 7.327837 | -0.13287 |
| Н | 0.660033 | 5.104779 | 2.80576  |
| Н | -0.23775 | 6.518993 | 2.234716 |
| Н | -0.74906 | 4.893233 | 1.74188  |
| Н | 1.892242 | 7.556215 | 1.278005 |
| Н | 2.830328 | 6.189643 | 1.871276 |
| Н | 2.917974 | 6.647124 | 0.158136 |



**Fig. S13** DFT-optimized structure of **6a**. Selected bond lengths (Å): N1-O1 1.329, C1-N1 1.333, O1-B1 1.531, N2-B1 1.632, C2-N3 1.374.

### Cartesian coordinates of 6b

| С | -0.52975 | -2.15062 | -1.12506 |
|---|----------|----------|----------|
| С | 0.560026 | -0.34775 | -0.0626  |
| С | 0.700133 | -2.6115  | -1.68177 |

| С | 1.816417 | -0.75169 | -0.58073 |
|---|----------|----------|----------|
| Н | 0.450897 | 0.507666 | 0.590985 |
| С | 1.847792 | -1.92244 | -1.40345 |

| Н | 0.721866 | -3.47076 | -2.33634 |
|---|----------|----------|----------|
| Н | 2.785783 | -2.2457  | -1.84025 |
| Ν | -0.54135 | -1.00463 | -0.35235 |
| Ν | -1.76015 | -2.6535  | -1.20745 |
| 0 | -2.67237 | -1.93909 | -0.54185 |
| С | -2.27655 | -3.9061  | -1.91697 |
| С | -2.08393 | -3.67886 | -3.42865 |
| С | -1.50118 | -5.12165 | -1.37779 |
| С | -3.7689  | -4.0488  | -1.59034 |
| Н | -2.61683 | -2.77984 | -3.7513  |
| Н | -1.03271 | -3.59419 | -3.72019 |
| Н | -2.50434 | -4.5361  | -3.96231 |
| Н | -1.59399 | -5.19247 | -0.28995 |
| Н | -1.93891 | -6.0239  | -1.81405 |
| Н | -0.44082 | -5.1171  | -1.64303 |
| Н | -4.13132 | -4.93747 | -2.11435 |
| Н | -3.94043 | -4.18886 | -0.52056 |
| Н | -4.3446  | -3.187   | -1.93139 |
| В | -2.09532 | -0.63438 | -0.00512 |
| С | -2.36551 | -0.56924 | 1.569921 |
| С | -3.15988 | 0.383746 | 2.216394 |
| С | -1.8493  | -1.57975 | 2.391724 |
| С | -3.40671 | 0.350097 | 3.589106 |
| С | -2.06591 | -1.64925 | 3.761385 |
| С | -2.85748 | -0.67079 | 4.364725 |
| С | -2.48381 | 0.61325  | -0.96631 |
| С | -3.29706 | 0.53626  | -2.1015  |
| С | -1.93709 | 1.874823 | -0.71833 |
| С | -3.56929 | 1.636645 | -2.91755 |
| С | -2.16066 | 2.98924  | -1.51335 |
| С | -2.99568 | 2.871694 | -2.62514 |
| F | -3.85323 | -0.63077 | -2.48227 |
| F | -1.1132  | 2.041902 | 0.353157 |
| F | -3.72379 | 1.386031 | 1.527715 |
| F | -1.08563 | -2.55666 | 1.835106 |
| F | -1.53484 | -2.63157 | 4.489882 |

| F | -3.08177 | -0.7104  | 5.670634 |
|---|----------|----------|----------|
| F | -4.15694 | 1.286951 | 4.161475 |
| F | -1.5645  | 4.155787 | -1.23486 |
| F | -3.22216 | 3.921261 | -3.40376 |
| F | -4.35241 | 1.502224 | -3.98409 |
| Ν | 2.942901 | -0.00029 | -0.30072 |
| С | 2.831505 | 1.391654 | -0.10503 |
| С | 4.21065  | -0.61363 | -0.21192 |
| С | 3.592601 | 2.024808 | 0.907458 |
| С | 1.939251 | 2.164162 | -0.89548 |
| С | 5.343441 | 0.029452 | -0.76273 |
| С | 4.358633 | -1.86757 | 0.433233 |
| С | 3.455795 | 3.377888 | 1.142051 |
| Н | 4.248973 | 1.433692 | 1.537046 |
| С | 1.795154 | 3.509972 | -0.66228 |
| Н | 1.397043 | 1.699473 | -1.71261 |
| С | 6.590551 | -0.56001 | -0.6867  |
| Н | 5.225172 | 0.970597 | -1.28941 |
| С | 5.599833 | -2.45063 | 0.523595 |
| Н | 3.503075 | -2.34338 | 0.902017 |
| С | 2.549669 | 4.142925 | 0.362323 |
| Н | 4.020562 | 3.839918 | 1.942649 |
| Н | 1.114266 | 4.115267 | -1.25031 |
| С | 6.738669 | -1.8112  | -0.03894 |
| Н | 7.441663 | -0.06898 | -1.14253 |
| Н | 5.745452 | -3.39155 | 1.043243 |
| 0 | 2.336391 | 5.438308 | 0.506272 |
| 0 | 7.879926 | -2.46479 | 0.09729  |
| С | 3.057492 | 6.212179 | 1.490451 |
| Н | 2.694599 | 7.231    | 1.368462 |
| Н | 2.829584 | 5.85102  | 2.497543 |
| Н | 4.132531 | 6.169718 | 1.29346  |
| С | 9.1181   | -1.91576 | -0.40427 |
| Н | 9.87881  | -2.64831 | -0.14091 |
| Н | 9.069266 | -1.80015 | -1.49103 |
| Н | 9.333544 | -0.95928 | 0.080731 |



**Fig. S14** DFT-optimized structure of **6b**. Selected bond lengths (Å): N1-O1 1.336, C1-N1 1.332, O1-B1 1.524, N2-B1 1.635, C2-N3 1.383.

| С | -3.68884 | -0.69861 | -0.24835 |
|---|----------|----------|----------|
| С | -1.53983 | -1.3227  | -0.74379 |
| С | -3.23131 | 0.450662 | 0.40795  |
| С | -0.97237 | -0.21159 | -0.10455 |
| Н | -0.89655 | -2.05776 | -1.2206  |
| С | -1.8705  | 0.68414  | 0.481284 |
| Н | -3.9038  | 1.162895 | 0.863438 |
| Н | -1.50313 | 1.566992 | 0.99222  |
| Ν | -2.8392  | -1.56042 | -0.80621 |
| Ν | -5.06497 | -1.03184 | -0.38901 |
| 0 | -5.36489 | -2.05398 | -1.11284 |
| С | -6.19442 | -0.29542 | 0.277416 |
| С | -6.00575 | -0.29594 | 1.800336 |
| С | -6.32403 | 1.119987 | -0.30345 |
| С | -7.48302 | -1.05686 | -0.04126 |
| Н | -5.94133 | -1.32395 | 2.168001 |
| Н | -5.11411 | 0.236886 | 2.134079 |
| Н | -6.87188 | 0.178829 | 2.270978 |

| Н | -6.50298 | 1.064446 | -1.38101 |
|---|----------|----------|----------|
| Н | -7.17843 | 1.624822 | 0.157235 |
| Н | -5.44379 | 1.744422 | -0.1422  |
| Н | -8.32163 | -0.51299 | 0.403936 |
| Н | -7.64308 | -1.13794 | -1.11617 |
| Н | -7.46067 | -2.06721 | 0.370201 |
| Ν | 0.418152 | -0.01395 | -0.0616  |
| С | 1.30619  | -1.12187 | -0.01674 |
| С | 0.956223 | 1.30011  | -0.04935 |
| С | 2.443698 | -1.14909 | -0.82406 |
| С | 1.074953 | -2.19682 | 0.83874  |
| С | 1.961446 | 1.645432 | 0.856212 |
| С | 0.505654 | 2.269983 | -0.94143 |
| С | 3.320729 | -2.22111 | -0.76515 |
| Н | 2.639489 | -0.32146 | -1.49765 |
| С | 1.951504 | -3.27422 | 0.8716   |
| Н | 0.200435 | -2.19183 | 1.481078 |
| С | 2.495915 | 2.923535 | 0.855362 |

## Cartesian coordinates of 11a

| Н | 2.324237 | 0.901732 | 1.557894 |
|---|----------|----------|----------|
| С | 1.037513 | 3.554561 | -0.91756 |
| Н | -0.26739 | 2.01719  | -1.65984 |
| С | 3.100976 | -3.31394 | 0.080177 |
| Н | 4.190273 | -2.20439 | -1.41369 |
| Н | 1.729615 | -4.08778 | 1.551948 |
| С | 2.047378 | 3.914827 | -0.02528 |
| Н | 3.277279 | 3.153712 | 1.573276 |
| Н | 0.65539  | 4.276731 | -1.62929 |
| С | 2.662553 | 5.317558 | 0.021062 |
| С | 2.049749 | 6.254851 | -1.02599 |
| С | 2.428837 | 5.933369 | 1.412227 |
| С | 4.17549  | 5.224412 | -0.2461  |
| Н | 2.206105 | 5.885022 | -2.04432 |
| Н | 0.974545 | 6.390449 | -0.87165 |
| Н | 2.519708 | 7.24086  | -0.95706 |
| Н | 2.878156 | 5.328479 | 2.205101 |
| Н | 2.870972 | 6.934299 | 1.465456 |

| Н | 1.358866 | 6.022507 | 1.625581 |
|---|----------|----------|----------|
| Н | 4.630733 | 6.22018  | -0.20966 |
| Н | 4.682022 | 4.599926 | 0.49532  |
| Н | 4.372412 | 4.796302 | -1.23409 |
| С | 4.054653 | -4.51431 | 0.070284 |
| С | 3.85802  | -5.41506 | 1.297112 |
| С | 3.78494  | -5.34594 | -1.1971  |
| С | 5.520082 | -4.04653 | 0.06359  |
| Н | 4.001399 | -4.86127 | 2.230644 |
| Н | 2.863998 | -5.87133 | 1.318899 |
| Н | 4.586814 | -6.23132 | 1.275028 |
| Н | 3.944331 | -4.75165 | -2.10228 |
| Н | 4.453551 | -6.21322 | -1.23873 |
| Н | 2.752766 | -5.7096  | -1.21417 |
| Н | 6.188914 | -4.913   | 0.095896 |
| Н | 5.771669 | -3.47764 | -0.83529 |
| Н | 5.737722 | -3.41771 | 0.932842 |



Fig. S15 DFT-optimized structure of 11a. Selected bond lengths (Å): N1-O1 1.288, C1-N1 1.423, C2-N3 1.405.

## Cartesian coordinates of 11b

| С | -3.15902 | -0.16605 | -0.39608 |
|---|----------|----------|----------|
| С | -1.07929 | -1.0283  | -0.82993 |
| С | -2.60692 | 0.899637 | 0.329455 |
| С | -0.41566 | -0.01993 | -0.09965 |
| Н | -0.50026 | -1.81212 | -1.31282 |
| С | -1.22884 | 0.963907 | 0.487216 |
| Н | -3.23482 | 1.679898 | 0.743644 |
| Н | -0.78053 | 1.781134 | 1.042545 |
| N | -2.39865 | -1.10621 | -0.96783 |
| N | -4.56153 | -0.27056 | -0.64452 |
| 0 | -4.92304 | -0.55998 | -1.84168 |
| С | -5.64889 | -0.07688 | 0.381647 |
| С | -5.13071 | -0.35883 | 1.80092  |
| С | -6.19997 | 1.356217 | 0.251417 |
| С | -6.75887 | -1.0908  | 0.049147 |
| Н | -4.67479 | -1.35275 | 1.860704 |
| Н | -4.39913 | 0.373925 | 2.148497 |
| Н | -5.97739 | -0.33376 | 2.4953   |
| Н | -6.508   | 1.541984 | -0.78213 |
| Н | -7.0703  | 1.490698 | 0.903697 |
| Н | -5.45335 | 2.10861  | 0.528344 |
| Н | -7.57884 | -0.97494 | 0.766065 |
| Н | -7.14064 | -0.93557 | -0.96025 |
| Н | -6.37955 | -2.11562 | 0.116381 |
| Ν | 0.98571  | -0.00208 | 0.02111  |
| С | 1.724562 | -1.22215 | 0.066889 |
| С | 1.690461 | 1.236316 | 0.088096 |
| С | 2.86732  | -1.39372 | -0.72024 |
|   |          |          |          |

| 568<br>705<br>736<br>01<br>04<br>553<br>578<br>774<br>377    |
|--|
| 705<br>36<br>01<br>04<br>553<br>378<br>974<br>377            |
| 36       01       04       553       378       974       377 |
| 01<br>04<br>553<br>378<br>974<br>377                         |
| 04<br>553<br>378<br>974<br>377                               |
| 553<br>378<br>974<br>377                                     |
| 378<br>974<br>377  |
| 974<br>377   |
| 377  |
|  |
| 2  |
| 36   |
| 02   |
| 13   |
| 718  |
| 731  |
| 3  |
| 65   |
| 35   |
| 387  |
| 55   |
| 22   |
| 07   |
| 58   |
| 65   |
| 17   |
| 39   |
| 10   |
|  |



Fig. S16 DFT-optimized structure of 11b. Selected bond lengths (Å): N1-O1 1.284, C1-N1 1.428, C2-N3 1.407.

| С | 0.195394 | 2.312463 | 0.991988 |
|---|----------|----------|----------|
| С | 0.110854 | 0.632649 | 2.612248 |
| С | 0.229828 | 3.319602 | 1.946758 |
| С | 0.157026 | 1.579896 | 3.621358 |
| Н | 0.044417 | -0.43365 | 2.791489 |
| С | 0.213302 | 2.92903  | 3.279869 |
| Н | 0.272196 | 4.362354 | 1.670789 |
| Ν | 0.130702 | 1.01207  | 1.332328 |
| Ν | 0.223576 | 2.418288 | -0.39354 |
| 0 | 0.198018 | 1.255677 | -0.97632 |
| С | 0.270906 | 3.639384 | -1.27944 |
| С | -1.01491 | 4.434879 | -1.02589 |
| С | 1.548327 | 4.418552 | -0.94883 |
| С | 0.308633 | 3.154025 | -2.72743 |
| Н | -1.8923  | 3.815686 | -1.22983 |
| Н | -1.08613 | 4.825674 | -0.00852 |
| Н | -1.03237 | 5.28884  | -1.70705 |
| Н | 2.428215 | 3.775904 | -1.03665 |
| Н | 1.648176 | 5.233347 | -1.6696  |

| Curresture coordinates of 12 |
|------------------------------|
|------------------------------|

| Н | 1.53946  | 4.865731 | 0.046796 |
|---|----------|----------|----------|
| Н | 0.314752 | 4.037972 | -3.36868 |
| Н | 1.208347 | 2.572625 | -2.93786 |
| Н | -0.5668  | 2.551925 | -2.97415 |
| В | 0.042279 | 0.090668 | 0.02556  |
| С | 1.329312 | -0.84058 | -0.147   |
| С | 1.316794 | -2.16963 | -0.56295 |
| С | 2.594428 | -0.2927  | 0.058121 |
| С | 2.478173 | -2.91348 | -0.72672 |
| С | 3.774237 | -0.9953  | -0.09515 |
| С | 3.711245 | -2.32508 | -0.49102 |
| С | -1.44164 | -0.51466 | -0.07067 |
| С | -2.46437 | -0.00391 | -0.86176 |
| С | -1.79085 | -1.59639 | 0.730536 |
| С | -3.73777 | -0.55595 | -0.89414 |
| С | -3.04503 | -2.1773  | 0.73273  |
| С | -4.02832 | -1.65119 | -0.0961  |
| F | -2.26804 | 1.082678 | -1.63304 |
| F | -0.86652 | -2.11099 | 1.567667 |

| F | 0.169572 | -2.79325 | -0.8341  |
|---|----------|----------|----------|
| F | 2.696695 | 1.007085 | 0.420399 |
| F | 4.947547 | -0.41269 | 0.121651 |
| F | 4.822684 | -3.02232 | -0.64771 |
| F | 2.412323 | -4.17889 | -1.11572 |

| F | -3.31519 | -3.21084 | 1.518064 |
|---|----------|----------|----------|
| F | -5.23878 | -2.18228 | -0.10954 |
| F | -4.67548 | -0.02978 | -1.67175 |
| Н | 0.145231 | 1.261595 | 4.656455 |
| Н | 0.2444   | 3.686257 | 4.055618 |



**Fig. S17** DFT-optimized structure of **12**. Selected bond lengths (Å): N1-O1 1.301, C1-N1 1.390, O1-B1 1.544, N2-B1 1.601.

Cartesian coordinates of 13

| С | -0.69671 | 0.232657 | 0.139845 |
|---|----------|----------|----------|
| С | -2.77775 | 0.97401  | -0.40563 |
| С | -1.17582 | -1.04078 | 0.451358 |
| С | -3.3579  | -0.26242 | -0.14603 |
| Н | -3.38793 | 1.808524 | -0.74394 |
| С | -2.53154 | -1.28752 | 0.29486  |
| Н | -0.51238 | -1.80449 | 0.836884 |
| Ν | -1.47864 | 1.224294 | -0.27363 |
| Ν | 0.677744 | 0.570261 | 0.311849 |
| 0 | 0.926823 | 1.66068  | 0.925807 |
| С | 1.842929 | -0.23837 | -0.16282 |

| С | 2.407569 | -1.02848 | 1.023024 |
|---|----------|----------|----------|
| С | 1.441075 | -1.16034 | -1.31556 |
| С | 2.891866 | 0.757563 | -0.66863 |
| Н | 2.63088  | -0.34665 | 1.847611 |
| Н | 1.701794 | -1.78384 | 1.382151 |
| Н | 3.330442 | -1.54063 | 0.734081 |
| Н | 0.960717 | -0.59543 | -2.11983 |
| Н | 2.344081 | -1.6239  | -1.72264 |
| Н | 0.768194 | -1.96372 | -1.01173 |
| Н | 3.774229 | 0.209892 | -1.01202 |
| Н | 2.496942 | 1.341828 | -1.50452 |

| Н | 3.184106 | 1.448331 | 0.121297 |
|---|----------|----------|----------|
| Н | -4.424   | -0.40973 | -0.27772 |



Cartesian coordinates of TEMPO

Fig. S18 DFT-optimized structure of 13. Selected bond lengths (Å): N1-O1 1.276, C1-N1 1.426.

| С | -1.3322  | -0.06919 | -0.02897 |
|---|----------|----------|----------|
| С | -1.24669 | 1.402189 | -0.48352 |
| С | 0        | 2.128882 | 0.024236 |
| Н | -2.16143 | 1.913105 | -0.15844 |
| Н | -1.24404 | 1.432868 | -1.58186 |
| С | 1.24669  | 1.402189 | -0.48352 |
| Н | 0        | 2.184816 | 1.120185 |
| Н | 0        | 3.16475  | -0.33635 |
| С | 1.332203 | -0.06919 | -0.02897 |
| Н | 2.161426 | 1.913105 | -0.15844 |
| Н | 1.244043 | 1.432868 | -1.58186 |
| Ν | 0        | -0.74339 | -0.20223 |
| С | 1.767158 | -0.18376 | 1.44764  |
| С | 2.338199 | -0.82465 | -0.91306 |
| С | -2.3382  | -0.82465 | -0.91306 |

| С | -1.76716 | -0.18376 | 1.44764  |
|---|----------|----------|----------|
| 0 | 0        | -2.01956 | -0.0471  |
| Н | -1.14961 | 0.427778 | 2.112554 |
| Н | -1.69316 | -1.22626 | 1.769549 |
| Н | -2.80702 | 0.144137 | 1.558857 |
| Н | -2.01861 | -0.80986 | -1.96053 |
| Н | -3.32105 | -0.34509 | -0.84419 |
| Н | -2.42649 | -1.86556 | -0.5969  |
| Н | 1.149614 | 0.427778 | 2.112554 |
| Н | 2.80702  | 0.144137 | 1.558857 |
| Н | 1.693162 | -1.22626 | 1.769549 |
| Н | 2.018612 | -0.80986 | -1.96053 |
| Н | 2.426493 | -1.86556 | -0.5969  |
| Н | 3.321047 | -0.34509 | -0.84419 |



Fig. S19 DFT-optimized structure of TEMPO. Selected bond lengths (Å): N1-O1 1.286.



**Fig. S20** UV-vis-NIR spectrum of [**5a**]SbCl<sub>6</sub> in CH<sub>2</sub>Cl<sub>2</sub> and TD-DFT assignments for the longest and second longest absorption bands.

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