Electronic Supplementary Information for:

Reactions and catalytic applications of a PNCNP pincer palladium hydride complex

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Experimental general information

The reactions were carried out under a nitrogen atmosphere. The solvents were dried using standard procedures and degassed before use. NMR spectra were recorded on a 600 MHz Bruker Advance spectrometer at room temperature. The chemical shifts were either referenced internally to the residual solvent peaks or standardized by outside samples. High resolution mass spectrometry (HRMS) was performed in the ESI mode by using a MicrOTOF mass spectrometer. GC-MS analyses were performed by using a SHIMADZU-(GCMS-QP2020) instrument. Elemental analysis was performed on an Elementar Vario-III analyzer. FTIR spectra were recorded on a VECTOR 22 Bruker spectrophotometer. The starting palladium pincer chloride complex [2,6-(^tBu₂PNH)₂C₆H₃]PdCl (**Pd-Cl**) was prepared according to the literature method.^{S1}

Synthesis and characterization of [2,6-(^tBu₂PNH)₂C₆H₃]PdNO₃ (Pd-NO₃)

A mixture of complex **Pd-Cl** (537 mg, 1.0 mmol), AgNO₃ (255 mg, 1.5 mmol) and THF (30 mL) was stirred at room temperature for 4 h. Solvent was evaporated under vacuum and the residue was extracted with CH₂Cl₂ (15 mL × 3). CH₂Cl₂ was removed from the combined extraction and a light grey crystalline solid of complex **Pd-NO₃** was obtained (509 mg, 90% yield). ¹H NMR (600 MHz, CDCl₃, δ): 6.75 (t, 1H, *J*_{H-H} = 7.8 Hz, Ar*H*), 6.10 (d, 2H, *J*_{H-H} = 7.8 Hz, Ar*H*), 4.15 (s, 2H, N*H*), 1.37–1.39 (m, 36H, C(C*H*₃)₃). ¹³C{¹H} NMR (151 MHz, CDCl₃, δ): 158.28 (t, *J*_{C-P} = 11.1 Hz, Ar*C*), 127.48 (s, Ar*C*), 116.74 (s, Ar*C*), 102.10 (t, *J*_{C-P} = 7.8 Hz, Ar*C*), 38.13 (t, *J*_{C-P} = 8.7 Hz, *C*(CH₃)₃), 28.11 (t, *J*_{C-P} = 3.8 Hz, C(*CH*₃)₃). ³¹P{¹H} NMR (243 MHz, CDCl₃, δ): 119.48 (s). Anal. calcd for C₂₂H₄₁N₃O₃P₂Pd + H⁺ [M + H]⁺ 564.1737, found 564.1739.



Scheme S1 Synthesis of complex Pd-NO₃



Fig. S1 ¹H NMR spectrum of complex Pd-NO₃ (600 MHz, CDCl₃)



Fig. S2 $^{13}C{^{1}H}$ NMR spectrum of complex Pd-NO₃ (151 MHz, CDCl₃)



Fig. S3 $^{31}P{^{1}H}$ NMR spectrum of complex Pd-NO₃ (243 MHz, CDCl₃)



Fig. S4 Thermal ellipsoid plots of complex **Pd-NO**₃ at the 50% probability level (for clarity, hydrogen atoms and the co-crystallized THF molecule have been omitted and the *tert*-butyl groups have been simplified). Selected bond lengths (Å) and angles (°): Pd1-C1, 1.996(3); Pd1-P1, 2.3084(6); Pd1-P2, 2.3296(6); Pd1-O1, 2.1415(18); Pd1 ··O3, 3.122(2); N3-O1, 1.279(3); N3-O3, 1.247(3); N3-O2, 1.224(3); P1-Pd1-P2, 162.70(2); C1-Pd1-O1, 170.32(9); O3-N3-O1, 119.2(2); O2-N3-O1, 118.1(2); O2-N3-O3, 122.7(2).

Synthesis and characterization of [2,6-(^tBu₂PNH)₂C₆H₃]PdH (1)

Route (I): A suspension of complex **Pd-Cl** (537 mg, 1.0 mmol) and LiAlH₄ (76 mg, 2.0 mmol) in THF (30 mL) was stirred at room temperature for 1 h. Solvent was evaporated under vacuum and the residue was extracted with toluene (15 mL \times 3). Toluene was then removed from the combined extraction solutions and a white crystalline solid of [2,6-(^tBu₂PNH)₂C₆H₃]PdH (1) was obtained (407 mg, 81% yield).

Route (II): A mixture of complex **Pd-NO₃** (282 mg, 0.5 mmol), KOCH₃ (700 mg, 10 mmol) and THF (20 mL) was sonicated at room temperature for 5 h. The resulting suspension was centrifuged and the solvent was evaporated under reduced pressure. The solid residue was extracted with *n*-hexane (15 mL \times 3). The combined extraction solution was concentrated and recrystallized. Complex [2,6-(^{*t*}Bu₂PNH)₂C₆H₃]PdH (**1**) was obtained in 85% yield.

¹H NMR (600 MHz, benzene- d_6 , δ): 7.10 (t, 1H, $J_{\text{H-H}} = 7.7$ Hz, Ar*H*), 6.40 (dd, 2H, $J_{\text{H-H}} = 7.7$, 0.9 Hz, Ar*H*), 3.94 (s, 2H, N*H*), 1.19–1.22 (m, 36H, C(CH₃)₃), -2.84 (t, 1H, $J_{\text{H-P}} = 19.6$ Hz, Pd*H*). ¹³C{¹H} NMR (151 MHz, benzene- d_6 , δ): 157.72 (t, $J_{\text{C-P}} = 11.7$ Hz, Ar*C*), 140.47 (t, $J_{\text{C-P}} = 3.8$ Hz, Ar*C*), 126.85 (s, Ar*C*), 101.23 (t, $J_{\text{C-P}} = 7.4$ Hz, Ar*C*), 36.76 (t, $J_{\text{C-P}} = 9.5$ Hz, *C*(CH₃)₃), 28.86 (t, $J_{\text{C-P}} = 4.5$ Hz, C(CH₃)₃). ³¹P{¹H} NMR (243 MHz, benzene- d_6 , δ): 141.37 (s). Anal. calcd for C₂₂H₄₂N₂P₂Pd: C, 52.54; H, 8.42. Found: C, 52.67; H, 8.55. HRMS (ESI): m/z calculated for C₂₂H₄₂N₂P₂Pd [M] 502.1853, found 502.1811. Selected FTIR absorptions (KBr disc, cm⁻¹): 3367 (m), 1718 (s).



Fig. S5 ³¹P{¹H} NMR spectrum the reaction mixture of complex Pd-NO₃, KOCH₃ and THF recorded 3 h after mixing the reactants (Route II).



Fig. S6 ¹H NMR spectrum of complex $1 (600 \text{ MHz}, C_6D_6)$



Fig. S7 $^{13}C\{^{1}H\}$ NMR spectrum of complex 1 (151 MHz, $C_{6}D_{6})$



Fig. S8 ${}^{31}P{}^{1}H$ NMR spectrum of complex 1 (243 MHz, C₆D₆)



Fig. S9 FTIR spectrum of complex 1 (KBr disc)

Reaction of [2,6-(^tBu₂PNH)₂C₆H₃]PdH (1) with methanol

EXP (I): Complex **1** (0.01 mmol) was dissolved in 0.5 mL of benzene- d_6 in a NMR tube, and then 50 equiv. of methanol was added. The NMR tube was sealed and the reaction was monitored by ¹H and ³¹P{¹H} NMR spectra at room temperature.

EXP (II): Complex **1** (0.01 mmol) was dissolved in 0.5 mL of methanol/methanol- d_4 (1:4) in a NMR tube. The NMR tube was sealed and the reaction was monitored by ¹H NMR spectra at room temperature.



Fig. S10 ${}^{31}P{}^{1}H$ NMR spectra of the reaction of complex 1 with methanol (243 MHz, C₆D₆, EXP I).



Fig. S11 ¹H NMR spectrum of the reaction of complex **1** with methanol (600 MHz, methanol- d_4). Spectrum recorded 5 h after adding complex **1** to methanol/methanol- d_4 (1:4) in a NMR tube (EXP II).

Reaction of [2,6-(^tBu₂PNH)₂C₆H₃]PdH (1) with H₂O

Complex 1 (0.01 mmol) was dissolved in 0.5 mL of benzene- d_6 in a NMR tube and H₂O (0.50 mmol) was added. The NMR tube was sealed and kept at room temperature and the reaction was monitored by NMR spectra.



Fig. S12 ³¹P{¹H} NMR spectrum of the reaction of complex **1** with water (243 MHz, C_6D_6). Bottom: spectrum recorded 24 h after mixing complex **1** with 50 equiv. of water; top: spectrum recorded 96 h after mixing complex **1** with 50 equiv. of water.

Synthesis and characterization of [2,6-(^tBu₂PNH)₂C₆H₃]PdOH (2)

A mixture of complex **Pd-NO**₃ (282 mg, 0.5 mmol), NaOH (500 mg, 12.5 mmol) and THF (20 mL) was sonicated at room temperature for 5 h. The resulting suspension was centrifuged and the solvent was evaporated under reduced pressure. The solid residue was extracted with *n*-hexane (15 mL × 3). The combined extraction solution was concentrated and recrystallized. Complex $[2,6-({}^{t}Bu_{2}PNH)_{2}C_{6}H_{3}]PdOH$ (**2**) was obtained as a white solid (190 mg, 73% yield). ¹H NMR (600 MHz, benzene-*d*₆, δ): 6.98 (t, 1H, *J*_{H-H} = 7.7 Hz, Ar*H*), 6.20 (d, 2H, *J*_{H-H} = 7.7 Hz, Ar*H*), 3.90 (s, 2H, N*H*), 1.28–1.30 (m, 36H, C(C*H*₃)₃), -1.35 (s, br., 1H, O*H*). ¹³C{¹H} NMR (151 MHz, benzene-*d*₆, δ): 158.49 (t, *J*_{C-P} = 12.2 Hz, Ar*C*), 126.24 (s, Ar*C*), 124.06 (s, Ar*C*), 102.13 (t, *J*_{C-P} = 6.7 Hz, Ar*C*), 37.69 (t, *J*_{C-P} = 8.4 Hz, C(CH₃)₃), 28.32 (t, *J*_{C-P} = 3.6 Hz, C(CH₃)₃). ³¹P{¹H} NMR (243 MHz, benzene-*d*₆, δ): 114.67 (s). HRMS (ESI): *m*/*z* calculated for C₂₂H₄₂N₂OP₂Pd + H⁺ [M + H]⁺ 519.1880, found 519.1839; *m*/*z* calculated for C₂₂H₄₂N₂OP₂Pd]⁺ [M]⁺ 518.1802, found 518.1807.



Fig. S13 ¹H NMR spectrum of complex **2** (600 MHz, C_6D_6)



Fig. S15 ${}^{31}P{}^{1}H$ NMR spectrum of complex **2** (243 MHz, C₆D₆)

Synthesis and characterization of [2,6-(^tBu₂PNH)₂C₆H₃]PdNHC(O)CH₃ (3)

Route (1): Complex **1** (100 mg, 0.20 mmol) and acetamide (21 μ L, 0.40 mmol) were dissolved in 20 mL of THF. The solution was refluxed for 48 h. The reaction mixture was slowly cooled to room temperature and [2,6-(^tBu₂PNH)₂C₆H₃]PdNHC(O)CH₃ (**3**) precipitated from the solution as solid product (80 mg, 71% yield).

Route (II): Complex **1** (20 mg, 0.04 mmol) was dissolved in acetonitrile (2 mL), and then H₂O (4 μ L, 0.20 mmol) was added. The mixture was kept at room temperature without stirring overnight. Complex **3** precipitated as a dark yellow crystalline solid (9 mg, 40% yield).

Route (III): Complex **2** (0.02 mmol) was dissolved in 0.5 mL of benzene- d_6 in a NMR tube and acetonitrile (5 μ L, 0.10 mmol) was added. The NMR tube was heated to 80 °C and the reaction was monitored by ³¹P{¹H} NMR spectrum. After 15 h, complex **2** was cleanly converted into complex **3**. The NMR tube was slowly cooled to room temperature in 24 h. Complex **3** precipitated from the solution as a crystalline solid (10 mg, 89% yield).

¹H NMR (600 MHz, DMSO- d_6 , δ): 6.50 (t, 1H, $J_{\text{H-H}} = 7.8$ Hz, Ar*H*), 6.30 (s, 2H, PN*H*), 6.04 (d, 2H, $J_{\text{H-H}} = 7.8$ Hz, Ar*H*), 1.72 (s, 3H, CH₃), 1.29–1.31 (m, 36H, C(CH₃)₃). ³¹P{¹H} NMR (243 MHz, DMSO- d_6 , δ): 117.20 (s). HRMS (ESI): m/z calculated for C₂₄H₄₃N₃OP₂Pd + H⁺ [M + H]⁺ 560.2146, found 560.2102; m/z calculated for C₂₄H₄₃N₃OP₂Pd + [M]⁺ 559.2067, found 559.2058.



Fig. S16 ¹H NMR spectrum of complex **3** (600 MHz, DMSO- d_6).



Fig. S17 ${}^{31}P{}^{1}H$ NMR spectrum of complex **3** (243 MHz, DMSO- d_6).

X-ray crystal structure determination

The crystal structures of complexes **1**, **2**, **3**, and **Pd-NO₃** were determined on a SuperNova diffractometer (Oxford Diffraction). The intensity data were collected at 150 or 170 K using monochromated Cu-K α radiation ($\lambda = 1.54184$ Å). The frames were integrated with the Bruker APEX2 software package using a narrow-frame algorithm. The data were corrected for decay, Lorentz, and polarization effects and absorption and beam corrections based on the multi-scan technique. The structures were solved by a combination of direct methods in SHELXTL and difference Fourier techniques and refined by full-matrix least-squares procedures interfaced with the program OLEX2.^{S2,S3} Non-hydrogen atoms were refined with anisotropic displacement parameters. The H-atoms were either located or calculated and subsequently treated with a riding model. The crystals suitable for X-ray diffraction analysis were obtained in benzene solution for **1**, THF/*n*-hexane solution for **Pd-NO₃** and **2**, acetonitrile/H₂O solution for **3**, respectively. Complex **Pd-NO₃** co-crystallized with a THF molecule, complex **2** co-crystallized with three molecules of H₂O, and complex **3** co-crystallized with two molecules of CH₃CN. Summaries of the crystallographic data and structure refinement for these complexes are provided in the Tables S1 and S2.

Complex	1	2 3H ₂ O
CCDC number	2191732	2191734
Empirical formula	$C_{22}H_{42}N_2P_2Pd$	$C_{22}H_{48}N_2O_4P_2Pd$
Formula weight	502.91	572.96
Temp, K	150.00(10)	149.99(10)
Crystal system	orthorhombic	orthorhombic
Space group	Pbca	Pbca
<i>a</i> , Å	11.4139(2)	16.9062(2)
b, Å	15.3554(3)	14.9710(2)
<i>c</i> , Å	28.0400(5)	21.5306(3)
α()	90	90
β ()	90	90
γ()	90	90
Volume, Å ³	4914.43(16)	5449.45(12)
Ζ	8	8
$d_{\rm calc}$, g cm ⁻³	1.359	1.397
λ, Å	1.54184	1.54184
μ , mm ⁻¹	7.376	6.831
No. of data collected	12651	14725
No. of unique data	4655	5210
<i>R</i> _{int}	0.0428	0.0282
Goodness-of-fit on F^2	1.067	1.047
$R_1, \mathrm{w}R_2 (I > 2\sigma(I))$	0.0455, 0.1153	0.0312, 0.0778
R_1 , w R_2 (all data)	0.0541, 0.1222	0.0366, 0.0809

Table S1. Summary of crystal data and structure refinement for complexes 1 and 2

Complex	3 2CH ₃ CN	Pd-NO ₃ THF		
CCDC number	2191735	2191733		
Empirical formula	$C_{28}H_{51}N_5OP_2Pd$	$C_{26}H_{49}N_3O_4P_2Pd$		
Formula weight	642.07	636.02		
Temp, K	170.00(10)	149.99(10)		
Crystal system	monoclinic	monoclinic		
Space group	$P2_1/c$	$P2_1/n$		
<i>a</i> , Å	11.71590(10)	11.34860(10)		
b, Å	14.45290(10)	22.9353(2)		
<i>c</i> , Å	19.6929(2)	11.59380(10)		
α()	90	90		
β()	95.1230(10)	95.3590(10)		
γ(⁹	90	90		
Volume, Å ³	3321.25(5)	3004.48(5)		
Z	4	4		
$d_{\rm calc}$, g cm ⁻³	1.284	1.406		
λ, Å	1.54184	1.54184		
μ , mm ⁻¹	5.622	6.266		
No. of data collected	14789	13423		
No. of unique data	6346	5747		
R _{int}	0.0265	0.0323		
Goodness-of-fit on F^2	1.081	1.035		
$R_1, \mathrm{w}R_2 (I > 2\sigma(I))$	0.0335, 0.0865	0.0351, 0.0889		
R_1 , w R_2 (all data)	0.0371, 0.0885	0.0386, 0.0913		

Table S2. Summary of crystal data and structure refinement for complexes 3 and Pd-NO₃

General procedure for the catalytic reactions

Nitrile (1.0 mmol), THF (1.0 mL), water (1.0 mL), and complex **1** (5.0 mg, 0.01 mmol) were mixed in a reaction tube. The tube was then sealed and the reaction mixture was stirred at 80 °C. The reaction was monitored by GC-MS and stopped at the required time or until the nitrile substrate disappeared completely. THF and water were then removed under reduced pressure. The resulting residue was extracted with CH_2Cl_2 (3 × 5 mL) and filtered. Removal of CH_2Cl_2 from the combined extraction solutions produced the crude amide product which was washed with cold diethyl ether or *n*-hexane and dried in vacuo. The isolated amide products were characterized by ¹H, ¹³C{¹H} NMR spectra and high resolution mass spectrometry (HRMS). The NMR spectra are presented in Fig. S18–S67.

Characterization of the isolated amide products

O _______NH₂ (4a)

White solid (46 mg, 78% yield), M.P. 80–83 °C. ¹H NMR (600 MHz, CD₃OD, δ): 1.96 (s, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 176.39 (CO), 22.08 (CH₃). HRMS (ESI): *m*/*z* calculated for C₂H₆NO [M + H]⁺ 60.0444; Found 60.0445. These spectral data correspond to previously reported data.^{S4}





White solid (75 mg, 86% yield), M.P. 115–116 °C. ¹H NMR (600 MHz, CD₃OD, δ): 2.20 (t, J = 7.8 Hz, 2H, CH₂), 1.66 (m, 2H, CH₂), 0.98 (t, J = 7.8 Hz, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 179.12 (CO), 38.42 (CH₂), 20.22 (CH₂), 13.97 (CH₃). HRMS (ESI): m/z calculated for C₄H₁₀NO [M + H]⁺ 88.0757; Found 88.0764. These spectral data correspond to previously reported data.^{S7}



Fig. S21¹³C{¹H} NMR spectrum of the isolated **4b** (151 MHz, CD₃OD)



White solid (73 mg, 72% yield), M.P. 104–106 °C. ¹H NMR (600 MHz, CD₃OD, δ): 2.22 (t, *J* = 7.2 Hz, 2H, CH₂), 1.61 (m, 2H, CH₂), 1.39 (m, 2H, CH₂), 0.96 (t, *J* = 7.2 Hz, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 179.31 (CO), 36.26 (CH₂), 29.05 (CH₂), 23.37 (CH₂), 14.10 (CH₃). HRMS (ESI): *m*/*z* calculated for C₅H₁₂NO [M + H]⁺ 102.0913; Found 102.0915. These spectral data correspond to previously reported data.^{S7}





White solid (115 mg, 95% yield), M.P. 128–130 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.88 (d, J = 7.8 Hz, 2H, Ar*H*), 7.54 (t, J = 7.8 Hz, 1H, Ar*H*), 7.46 (t, J = 7.8 Hz, 2H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 172.41 (*C*O), 134.94 (Ar*C*), 132.90 (Ar*C*), 129.50 (Ar*C*), 128.62 (Ar*C*). HRMS (ESI): m/z calculated for C₇H₈NO [M + H]⁺ 122.0600; Found 122.0601. These spectral data correspond to previously reported data.^{S4}



Fig. S25 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4d (151 MHz, CD₃OD)



White solid (149 mg, 98% yield), M.P. 164–168 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.86 (d, *J* = 9 Hz, 2H, Ar*H*), 6.98 (d, *J* = 7.8 Hz, 2H, Ar*H*), 3.85 (s, 3H, C*H*₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 172.02 (*CO*), 164.12 (Ar*C*), 130.60 (Ar*C*), 126.90 (Ar*C*), 114.66 (Ar*C*), 55.89 (*C*H₃). HRMS (ESI): *m*/*z* calculated for C₈H₁₀NO₂ [M + H]⁺ 152.0706; Found 152.0705. These spectral data correspond to previously reported data.^{S4}



Fig. S27 $^{13}C\{^1H\}$ NMR spectrum of the isolated 4e (151 MHz, CD₃OD)



White solid (130 mg, 96% yield), M.P. 161–163 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.78 (d, J = 8.4 Hz, 2H, Ar*H*), 7.28 (d, J = 8.4 Hz, 2H, Ar*H*), 2.39 (s, 3H, C*H*₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 172.40 (CO), 143.63 (ArC), 132.04 (ArC), 130.09 (ArC), 128.70 (ArC), 21.43 (CH₃). HRMS (ESI): m/z calculated for C₈H₁₀NO [M + H]⁺ 136.0758; Found 136.0754. These spectral data correspond to previously reported data.^{S4}



Fig. S29 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4f (151 MHz, CD₃OD)



White solid (130 mg, 96% yield), M.P. 94–96 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.69 (s, 1H, Ar*H*), 7.66 (d, *J* = 6.6 Hz, 2H, Ar*H*), 7.31 (m, 2H, Ar*H*), 2.35 (s, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 172.56 (CO), 139.39 (Ar*C*), 134.80 (Ar*C*), 133.52 (Ar*C*), 129.36 (Ar*C*), 129.14 (Ar*C*), 125.71 (Ar*C*), 21.34 (CH₃). HRMS (ESI): *m*/*z* calculated for C₈H₁₀NO [M + H]⁺ 136.0757; Found 136.0759. These spectral data correspond to previously reported data.^{S7}





White solid (34 mg, 26% yield), M.P. 145–148 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.42 (d, *J* = 7.8 Hz, 1H, Ar*H*), 7.34 (t, *J* = 7.8 Hz, 1H, Ar*H*), 7.26 (d, *J* = 7.8 Hz, 1H, Ar*H*), 7.23 (t, *J* = 7.8 Hz, 1H, Ar*H*), 2.45 (s, 3H, C*H*₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 175.69 (CO), 137.43 (ArC), 136.76 (ArC), 131.81 (ArC), 130.97 (ArC), 128.08 (ArC), 126.69 (ArC), 19.84 (CH₃). HRMS (ESI): *m*/*z* calculated for C₈H₁₀NO [M + H]⁺ 136.0757; Found 136.0754. These spectral data correspond to previously reported data.^{S4}



Fig. S33 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4h (151 MHz, CD₃OD)



White solid (131 mg, 94% yield), M.P. 154–156 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.92–7.94 (m, 2H, Ar*H*), 7.15–7.18 (m, 2H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 171.14 (*C*O), 166.29 (d, ¹*J*_{C-F} = 251 Hz, Ar*C*), 131.20–131.34 (m, Ar*C*), 116.28 (d, ²*J*_{C-F} = 22 Hz, Ar*C*). ¹⁹F{¹H} NMR (565 MHz, CD₃OD, δ): -106.21. HRMS (ESI): *m*/*z* calculated for C₇H₇FNO [M + H]⁺ 140.0506; Found 140.0509. These spectral data correspond to previously reported data.^{S6}



Fig. S35 ¹³C{¹H} NMR spectrum of the isolated **4i** (151 MHz, CD₃OD)



White solid (127 mg, 82% yield), M.P. 173–175 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.86 (d, *J* = 8.4 Hz, 2H, Ar*H*), 7.46 (d, *J* = 9 Hz, 2H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 171.09 (*C*O), 138.96 (Ar*C*), 133.60 (Ar*C*), 130.33 (Ar*C*), 129.64 (Ar*C*). HRMS (ESI): *m*/*z* calculated for C₇H₇ClNO [M + H]⁺ 156.0211; Found 156.0210. These spectral data correspond to previously reported data.^{S5}



Fig. S37 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4j (151 MHz, CD₃OD)



White solid (101 mg, 65% yield), M.P. 132–134 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.90 (s, 1H, Ar*H*), 7.81 (d, *J* = 7.8 Hz, 1H, Ar*H*), 7.55 (d, *J* = 7.8 Hz, 1H, Ar*H*), 7.45 (t, *J* = 7.8 Hz, 1H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 170.66 (CO), 136.97 (ArC), 135.56 (ArC), 132.74 (ArC), 131.10 (ArC), 128.78 (ArC), 126.96 (ArC). HRMS (ESI): *m*/*z* calculated for C₇H₇ClNO [M + H]⁺ 155.0211; Found 155.0212. These spectral data correspond to previously reported data.^{S4}



Fig. S39 ¹³C{¹H} NMR spectrum of the isolated **4k** (151 MHz, CD₃OD)



White solid (127 mg, 82% yield), M.P. 142–145 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.52 (d, *J* = 7.2 Hz, 1H, Ar*H*), 7.48 (d, *J* = 7.8 Hz, 1H, Ar*H*), 7.38 (t, *J* = 7.8 Hz, 1H, Ar*H*), 7.45 (t, *J* = 7.8 Hz, 1H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 172.34 (*C*O), 137.32 (Ar*C*), 132.17 (Ar*C*), 131.76 (Ar*C*), 131.09 (Ar*C*), 129.92 (Ar*C*), 128.05 (Ar*C*). HRMS (ESI): *m*/*z* calculated for C₇H₇ClNO [M + H]⁺ 156.0211; Found 156.0211. These spectral data correspond to previously reported data.^{S4}



Fig. S41 $^{13}C{^{1}H}$ NMR spectrum of the isolated **4l** (151 MHz, CD₃OD)



White solid (192 mg, 96% yield), M.P. 190–193 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.79 (d, *J* = 7.8 Hz, 2H, Ar*H*), 7.64 (d, *J* = 7.8 Hz, 2H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 171.21 (*CO*), 134.11 (Ar*C*), 132.73 (Ar*C*), 130.51 (Ar*C*), 127.35 (Ar*C*). HRMS (ESI): *m*/*z* calculated for C₇H₇BrNO [M + H]⁺ 199.9706; Found 199.9701. These spectral data correspond to previously reported data.^{S4}



Fig. S43 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4m (151 MHz, CD₃OD)



White solid (166 mg, 88% yield), M.P. 183–186 °C. ¹H NMR (600 MHz, CD₃OD, δ): 8.05 (d, *J* = 8.4 Hz, 2H, Ar*H*), 7.77 (d, *J* = 8.4 Hz, 2H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 170.82 (*C*O), 138.79 (Ar*C*), 134.26 (q, ²*J*_{C-F}= 33 Hz, ArC), 129.38 (Ar*C*), 126.46 (q, ³*J*_{C-F}= 4 Hz, Ar*C*), 125.27 (q, ¹*J*_{C-F}= 272 Hz, *C*F₃). ¹⁹F{¹H} NMR (565 MHz, CD₃OD, δ): -64.48. HRMS (ESI): *m*/*z* calculated for C₈H₇F₃NO [M + H]⁺ 190.0474; Found 190.0467. These spectral data correspond to previously reported data.^{S5}



Fig. S45 ${}^{13}C{}^{1}H$ NMR spectrum of the isolated **4n** (151 MHz, CD₃OD)



White solid (121 mg, 73% yield), M.P. 175–178 °C. ¹H NMR (600 MHz, CD₃OD, δ): 8.09 (dd, J = 8.2, 1.0 Hz, 1H, Ar*H*), 7.78 (td, J = 7.5, 1.2 Hz, 1H, Ar*H*), 7.70 (td, J = 8.1, 1.4 Hz, 1H, Ar*H*), 7.65 (dd, J = 7.5, 1.4 Hz, 1H, Ar*H*). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 171.78 (CO), 148.18 (ArC), 134.72 (ArC), 138.84 (ArC), 131.81 (ArC), 129.94 (ArC), 125.40 (ArC). HRMS (ESI): m/z calculated for C₇H₇N₂O₃ [M + H]⁺ 168.0529; Found 168.0526. These spectral data correspond to previously reported data.^{S6}







Yellow solid (155 mg, 95% yield), M.P. 188–192 °C. ¹H NMR (600 MHz, DMSO- d_6 , δ): 8.15 (s, br, 1H, NH), 7.98–8.02 (m, 4H, ArH), 7.57 ppm (s, br, 1H, NH), 2.61 (s, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, DMSO- d_6 , δ): 197.74 (NH₂CO), 167.17 (CH₃CO), 138.68 (ArC), 138.12 (ArC), 128.10 (ArC), 127.79 (ArC), 26.95 (CH₃). HRMS (ESI): m/z calculated for C₉H₁₀NO₂ [M + H]⁺ 164.0706; Found 164.0707. These spectral data correspond to previously reported data.^{S4}



Fig. S49 ${}^{13}C{}^{1}H$ NMR spectrum of the isolated **4p** (151 MHz, DMSO-*d*₆)



White solid (126 mg, 93% yield), M.P. 156–158 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.31–7.32 (m, 4H, Ar*H*), 7.24–7.26 (m, 1H, Ar*H*), 3.52 (s, 2H, C*H*₂). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 176.96 (CO), 136.91 (ArC), 130.14 (ArC), 129.56 (ArC), 127.89 (ArC), 43.41 (CH₂). HRMS (ESI): *m*/*z* calculated for C₈H₁₀NO [M + H]⁺ 136.0757; Found 136.0756. These spectral data correspond to previously reported data.^{S5}





White solid (150 mg, 98% yield), M.P. 158–159 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.30–7.33 (m, 2H, Ar*H*), 7.03–7.05 (m, 2H, Ar*H*), 3.50 (s, 2H, C*H*₂). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 176.69 (CO), 163.31 (d, ¹*J*_{C-F} = 243.7 Hz, Ar*C*), 132.89 (d, ³*J*_{C-F} = 3.2 Hz, Ar*C*), 131.91 (d, ⁴*J*_{C-F} = 3.2 Hz, Ar*C*), 116.12 (d, ²*J*_{C-F} = 21.6 Hz, Ar*C*), 42.39 (CH₂). ¹⁹F{¹H} NMR (565 MHz, CD₃OD, δ): -118.33. HRMS (ESI): *m*/*z* calculated for C₈H₉NO [M + H]⁺ 154.0603; Found 154.0605. These spectral data correspond to previously reported data.^{S7}



Fig. S53 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4r (151 MHz, CD₃OD)



White solid (128 mg, 86% yield), M.P. 184–186 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.19 (d, *J* = 7.8 Hz, 2H, Ar*H*), 7.13 (d, *J* = 7.8 Hz, 2H, Ar*H*), 3.47 (s, 2H, CH₂), 2.32 (s, 3H, CH₃). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 177.23 (CO), 137.58 (Ar*C*), 133.79 (Ar*C*), 130.18 (Ar*C*), 130.03 (Ar*C*), 43.02 (CH₂), 21.08 (CH₃). HRMS (ESI): *m*/*z* calculated for C₉H₁₂NO [M + H]⁺ 150.0913; Found 150.0910. These spectral data correspond to previously reported data.⁵⁸





White solid (106 mg, 83% yield), M.P. 184–185 °C. ¹H NMR (600 MHz, CD₃OD, δ): 2.22 (tt, *J* = 12.0, 3.4 Hz, 1H, CH), 1.91–1.75 (m, 4H, CH₂), 1.71 (dd, *J* = 7.6, 6.0 Hz, 1H, CH₂), 1.45 (dt, *J* = 21.6, 6.3 Hz, 2H, CH₂), 1.38–1.29 (m, 2H, CH₂), 1.26 (ddt, *J* = 12.6, 6.4, 2.9 Hz, 1H, CH₂). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 182.24 (CO), 45.92 (CH), 30.74 (CH₂), 26.91 (CH₂), 26.81 (CH₂). HRMS (ESI): *m*/*z* calculated for C₇H₁₄NO [M + H]⁺ 127.0992; Found 127.0993. These spectral data correspond to previously reported data.⁸⁵





White solid (94 mg, 77% yield), M.P. 109–110 °C. ¹H NMR (600 MHz, CD₃OD, δ): 8.64 (d, J = 4.7 Hz, 1H), 8.12 (d, J = 7.8 Hz, 1H), 7.96 (t, J = 7.7 Hz, 1H), 7.61–7.49 (m, 1H). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 169.28 (CO), 151.03 (ArC), 149.83 (ArC), 138.66 (ArC), 127.80 (ArC), 123.24 (ArC). HRMS (ESI): m/z calculated for C₆H₇N₂O [M + H]⁺ 123.0553; Found 123.0555. These spectral data correspond to previously reported data.^{S4}



Fig. S59¹³C{¹H} NMR spectrum of the isolated 4u (151 MHz, CD₃OD)



White solid (120 mg, 94% yield), M.P. 181–183 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.72 (d, J = 3.7 Hz, 1H), 7.66 (d, J = 5.0 Hz, 1H), 7.13 (t, J = 4.3 Hz, 1H). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 166.63 (CO), 139.85 (ArC), 132.15 (ArC), 130.54 (ArC), 128.82 (ArC). HRMS (ESI): m/z calculated for C₅H₆NOS [M + H]⁺ 128.0165; Found 128.0160. These spectral data correspond to previously reported data.^{S5}



Fig. S61 $^{13}C\{^1H\}$ NMR spectrum of the isolated 4v (151 MHz, CD₃OD)



White solid (96 mg, 86% yield), M.P. 140–142 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.66 (d, J = 1.6 Hz, 1H), 7.16 (d, J = 3.5 Hz, 1H), 6.58 (dd, J = 3.3, 1.6 Hz, 1H). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 162.89 (CO), 148.81 (ArC), 146.46 (ArC), 115.82 (ArC), 112.97 (ArC). HRMS (ESI): m/z calculated for C₅H₆NO₂ [M + H]⁺ 112.0393; Found 112.0394. These spectral data correspond to previously reported data.^{S5}



Fig. S63 $^{13}C\{^{1}H\}$ NMR spectrum of the isolated 4w (151 MHz, CD₃OD)



White solid (113 mg, 77% yield), M.P. 149–151 °C. ¹H NMR (600 MHz, CD₃OD, δ): 7.65–7.48 (m, 3H), 7.46–7.29 (m, 3H), 6.66 (d, J = 15.9 Hz, 1H). ¹³C{¹H} NMR (151 MHz, CD₃OD, δ): 170.92 (*C*O), 142.69, 136.14, 130.92, 129.93, 128.88, 121.40. HRMS (ESI): m/z calculated for C₉H₁₀NO [M + H]⁺ 148.0757; Found 148.0758. These spectral data correspond to previously reported data.^{S5}



Fig. S65 $^{13}C{^{1}H}$ NMR spectrum of the isolated 4x (151 MHz, CD₃OD)



White solid (156 mg, 91% yield), M.P. 191–192 °C. ¹H NMR (600 MHz, DMSO- d_6 , δ): 8.49 (s, 1H, Ar*H*), 8.14 (s, br, 1H, N*H*), 8.04–7.92 (m, 4H, Ar*H*), 7.66–7.54 (m, 2H, Ar*H*), 7.47 (s, br, 1H, N*H*). ¹³C{¹H} NMR (151 MHz, DMSO- d_6 , δ): 167.96 (CO), 134.16 (ArC), 132.15 (ArC), 131.65 (ArC), 127.81 (ArC), 127.76 (ArC), 127.58 (ArC), 127.56 (ArC), 126.64 (ArC), 124.39 (ArC). HRMS (ESI): m/z calculated for C₁₁H₁₀NO [M + H]⁺ 173.0835; Found 173.0832. These spectral data correspond to previously reported data.^{S5}



Fig. S67 ¹³C $\{^{1}H\}$ NMR spectrum of the isolated **4y** (151 MHz, DMSO- d_{6})

Reference

- 1 D. Benito-Garagorri, V. Bocokić, K. Mereiter and K. Kirchner, *Organometallics*, 2006, **25**, 3817–3823.
- 2 G. M. Sheldrick, Acta Crystallogr., Sect. A: Found. Crystallogr., 2008, 64, 112–122.
- O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard and H. Puschmann, J. Appl. Crystallogr., 2009, 42, 339–341.
- 4 E. Tom ás-Mendivil, F. J. Su árez, J. D éza and V. Cadierno, *Chem. Commun.*, 2014, **50**, 9661–9664.
- 5 X. Wen, J. He, H. Xi, Q. Zheng and W. Liu, Asian J. Org. Chem., 2022, 11, e202100781.
- 6 W. Zhan, L. Ji, Z.-M. Ge, X. Wang and R.-T. Li, *Tetrahedron*, 2018, 74, 1527–1532.
- 7 Q.-Q. Zhou, Y.-Q. Zou, S. Kar, Y. Diskin-Posner, Y. Ben-David and D. Milstein, ACS Catal., 2021, 11, 10239– 10245.
- 8 J. C. Bab ón, M. A. Esteruelas, A. M. L ópez and E. O ñate, *Inorg. Chem.*, 2021, **60**, 7284–7296.