Supporting Information

Aqua-friendly organometallic Ir-Pt complexes: pH responsive AIPE guided imaging of bacterial cells

Sakira Tabassum Borah,¹ Bishnu Das,¹ Prakash Biswas,² Amirul I Mallik^{2*} and Parna Gupta^{1*}

Indian Institute of Science Education and Research Kolkata, Mohanpur, West Bengal 741246, India

¹Department of Chemical Sciences

²Department of Biological Sciences

Table of Contents

Figure S1: ESI-MS spectra of L, Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt

Figure S2: NMR spectra of L, Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt

A.¹H NMR Spectra

B.¹³C NMR Spectra

C.¹⁹5Pt NMR Spectra

Figure S3: Geometry optimized structure of Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt

Figure S4: The overlaid absorption spectral profile of experimental and theoretical data

Table S1: Frontier orbital diagram of complexes Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt

Figure S5: Temperature dependent (290 – 80 K) emission spectra of Ir1 -Ir3, Ir1Pt -Ir3Pt in solution and solid state

Figure S6: Concentration dependent AIE of Ir1Pt – Ir3Pt

Figure S7: Toxicity of Ir1Pt, Ir2Pt & Ir3Pt on *E. coli* (DH5a)

Figure S8: Agar well diffusion assay using

A. mononuclear

B. dinuclear complexes

Figure S1: ESI-MS spectra















Figure S2: NMR spectra

A. ¹H NMR spectra















B. ¹³C NMR spectra















C. ¹⁹⁵Pt NMR spectra





Figure S3: Geometry optimized structure of Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt



Figure S4: The overlaid absorption spectral profile of experimental and theoretical data

Table S1: Frontier orbital diagram of complexes Ir1, Ir2, Ir3, Ir1Pt, Ir2Pt and Ir3Pt

Absorption			
lr1	Calculated $\lambda_{Abs}(nm, force constant)$		
	518.8492 f = 0.1111 HOMO-1->LUMO		
	450.6713 f = 0.0198 HOMO-2->LUMO		
	416.7816 f = 0.0857 HOMO-4->LUMO		
	332.2103 f = 0.058 HOMO-3->LUMO+1		
	300.4803 f = 0.1221 HOMO-1->LUMO+4		
	270.1946 f = 0.2821 HOMO-2->LUMO+6		

lr2	Calculated $\lambda_{Abs}(nm, force constant)$	
	518.047 f = 0.1099 HOMO-1->LUMO	
	420.6419 f = 0.086 HOMO-5->LUMO	
	405.1109 f = 0.0478 HOMO->LUMO+1	
	300.0295 f = 0.1301 HOMO-1->LUMO+4	
	HOMO-1->LUMO+5	
	275.1108 f = 0.2161 HOMO-13->LUMO	
	HOMO-2->LUMO+6	

lr3	Calculated $\lambda_{Abs}(nm, force)$		
	constant)		
	520.5483	÷	6
	HOMO-1->LUMO		I
		2.0	2.5
	426.9722		
	f = 0.042		· 1.
	HOMO->L+1		1. A
		25	
	328.3045	~	
	f = 0.2012 H-4->1+2	. I	
	H-4->L+1	۶ م	ę
		°	·
			~ ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
	282 2024		
	f = 0.1124	÷	6
	HOMO-13->LUMO	e de ser se en el	and the state
			27
		20	
Ir1Pt	Calculated $\lambda_{Abs}(nm, force)$		
	constant)		
	523.4051		
	f = 0.1176	the states	in a start of
	HUMU-1->LUMO+1		
			a state of the second sec
			the second second
		-	

	452.398 f = 0.0854 HOMO-4->LUMO	
	415.4409 f = 0.085 HOMO-5->LUMO+1	
	406 f = 0.0441 HOMO>L+3	
	320.8452 f = 0.3298 HOMO-13->LUMO	
	270.3124 f = 0.289 HOMO-2->LUMO+10	
lr2Pt	Calculated $\lambda_{Abs}(nm, force constant)$	
	522.6108 f = 0.1205 HOMO-1->LUMO+1	
	452.5466 f = 0.0888 HOMO-4->LUMO	

	405.5217 f = 0.0465 HOMO->LUMO+3	
	353.7655 f = 0.0151 HOMO-8->LUMO+1	
	275.7288 f = 0.2522 HOMO-2->LUMO+10	
	HOMO-17->LUMO+1	
lr3Pt	Calculated $\lambda_{\text{Abs}}(\text{nm, force}$ constant)	
	525.1342 f = 0.1131 HOMO-1->LUMO+1	
	452.2 f = 0.0889 HOMO-4->LUMO	
	429.7844 f = 0.0747 HOMO-5->LUMO+1	

320.7373 f = 0.303 HOMO-14->LUMO	
252.3749 f = 0.5829 HOMO-18->LUMO+2	

Figure S5: Temperature dependent (290 – 80 K) emission spectra of

A: Ir1 -Ir3 in Acetonitrile: MeTHF (λ_{ex} = 380 - 400 nm)



B: Ir1Pt -Ir3Pt in Acetonitrile: MeTHF (λ_{ex} = 380 - 400 nm)





C: Ir1 -Ir3 in solid state (Thin



D: Ir1Pt -Ir3Pt in solid state (λ_{ex} = 500 nm, Thin film on glass)



Figure S6: Concentration dependent AIE of Ir1Pt – Ir3Pt



Figure S7: Toxicity of Ir1Pt, Ir2Pt & Ir3Pt on *E. coli* (DH5α)



Figure S8: Agar well diffusion assay using A. mononuclear B. dinuclear complexes





29